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J4584-1

**Collection of X-ray Diffraction Data.** A colorless crystal of approximate dimensions  $0.20 \times 0.20 \times 0.21$  mm was mounted on the Siemens P4 diffractometer. Determination of Laue symmetry, crystal class, unit cell parameters and the crystal's orientation matrix were carried out according to standard procedures<sup>1</sup>. Low temperature (158 K) intensity data were collected via a  $2\theta$ -omega scan technique with  $\text{MoK}\alpha$  radiation.

All 1807 data were corrected for absorption<sup>2</sup> and for Lorentz and polarization effects and placed on an approximately absolute scale. The crystal class is triclinic and the space group is the centrosymmetric  $\text{P}\bar{1}$ .

**Solution and Refinement of the Crystal Structure.** All crystallographic calculations were carried out using the UCI modified version of the UCLA Crystallographic Computing Package<sup>3</sup> and the SHELXTL<sup>4</sup> program. The analytical scattering factors<sup>5</sup> for neutral atoms were used throughout the analysis. The structure was solved by direct methods and refined by full-matrix least-squares techniques (SHELXTL). The structure is polymeric. Each  $\text{Ce}_2\text{Cl}_2$  unit is located about an inversion center. Refinement of the model led to convergence with  $wR2 = 0.0507$  and  $\text{GOF} = 1.09$  for 100 parameters refined against all 1650 unique data ( $R1 = 0.021$  for those 1584 data with  $F_o > 4.0\sigma(F_o)$ ).

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References.

1. XSCAnS Version 2.10, Siemens Analytical X-Ray Instruments, Inc.; Madison, WI 1990-1995.
2. XPREP program (reference 4).
3. UCLA Crystallographic Computing Package, University of California Los Angeles 1981, C. Strouse; personal communication.
4. Sheldrick, G. M., Siemens Analytical X-Ray Instruments, Inc.; Madison, WI 1990-1995.
5. International Tables for X-Ray Crystallography 1992, Vol. C., Dordrecht: Kluwer Academic Publishers.

The thermal ellipsoid plot is shown at the 50% probability level.

Table 1. Crystal data and structure refinement for 1.

|                           |                         |
|---------------------------|-------------------------|
| Formula                   | $C_4H_{10}CeCl_3O_2$    |
| fw                        | 336.59                  |
| a                         | 6.691(2) Å              |
| b                         | 7.433(2) Å              |
| c                         | 10.092(2) Å             |
| $\alpha$                  | 84.46(2)°               |
| $\beta$                   | 76.72(2)°               |
| $\gamma$                  | 74.76(3)°               |
| V                         | 471.0(2) Å <sup>3</sup> |
| Z                         | 2                       |
| Space group               | $P\bar{1}$              |
| T                         | 158 K                   |
| $\lambda$                 | 0.71073 Å               |
| $\rho$ (calcd)            | 2.374 g/cm <sup>3</sup> |
| $\mu$                     | 5.627mm <sup>-1</sup>   |
| R1, wR2[I>2 $\sigma$ (I)] | 0.0206, 0.0499          |
| R1, wR2(all data)         | 0.0224, 0.0507          |

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$$R1 = \frac{\sum ||F_o| - |F_c||}{\sum |F_o|}$$

$$wR2 = \left[ \frac{\sum [w(F_o^2 - F_c^2)^2]}{\sum [w(F_o^2)^2]} \right]^{1/2}$$

Crystal data.

|                        |                             |
|------------------------|-----------------------------|
| Empirical formula      | $C_4H_{10}CeCl_3O_2$        |
| Formula weight         | 336.59                      |
| Crystal Description    | Prism                       |
| Crystal Color          | Colorless                   |
| Crystal size           | 0.21 x 0.20 x 0.20 mm       |
| Crystal system         | Triclinic                   |
| Space group            | $P\bar{1}$                  |
| Unit cell dimensions:  | $a = 6.691(2) \text{ \AA}$  |
|                        | $b = 7.433(2) \text{ \AA}$  |
|                        | $c = 10.092(2) \text{ \AA}$ |
|                        | $\alpha = 84.46(2)^\circ$   |
|                        | $\beta = 76.72(2)^\circ$    |
|                        | $\gamma = 74.76(3)^\circ$   |
| Volume                 | $471.0(2) \text{ \AA}^3$    |
| Z                      | 2                           |
| Density (calculated)   | $2.374 \text{ Mg/m}^3$      |
| Absorption coefficient | $5.627 \text{ mm}^{-1}$     |
| F(000)                 | 318                         |

J4584-4

Data Collection and Reduction.

JU584-5

|                                    |  |
|------------------------------------|--|
| Diffractionmeter type              | Siemens P4   |
| Radiation source                   | normal-focus sealed tube                                 |
| Monochromator type                 | graphite   |
| Radiation                          | MoK $\alpha$   |
| Wavelength                         | 0.71073 Å  |
| Temperature                        | 158 K  |
| Cell measurement reflections       | 35 ( $4.69^\circ < \theta < 15.68^\circ$ )               |
| Intensity measurement method       | $2\theta/\omega$ scans                                   |
| $\theta$ range for data collection | 2.07 to 25.00 $^\circ$                                   |
| Limiting indices                   | $-2 \leq h \leq 7, -8 \leq k \leq 8, -12 \leq l \leq 11$ |
| Scan width (in $\omega$ )          | 1.2 $^\circ$ plus $\alpha_1, \alpha_2$ separation        |
| Scan speed (in $\omega$ )          | 3.0 $^\circ$ /min  |
| Reflections collected              | 1807   |
| Independent reflections            | 1650 ( $R_{\text{int}} = 0.0248$ )                       |
| Observed reflections               | 1584 ( $I > 2\sigma(I)$ )                                |
| Number of standards                | 2  |
| Interval between standards         | 98   |
| Decay of standards                 | no decay   |
| Absorption correction              | Semi-empirical from $\psi$ -scans                        |
| Max. and min. transmission         | 0.4904 and 0.3882  |

Unit cell refinement and data collection were carried out by use of the Siemens XSCANs system, Version 2.10b. Data were processed with a local version of CARESS (R.W. Broach et al.), which employs a modified version of the Lehman-Larsen algorithm to obtain integrated intensities and standard deviations from the measured 96-step peak profiles.

Structure Solution and Refinement.

J4584-6

|  |  |
|--|--|
| Structure solution method              | direct methods   |
| Refinement method                      | Full-matrix least-squares on $F^2$                       |
| Scattering Factor Source               | International Tables Vol C<br>Tables 4.2.6.8 and 6.1.1.4 |
| Final refinement: Data                 | 1650   |
| Restraints                             | 0  |
| Parameters                             | 100  |
| Final R indices [ $I > 2\sigma(I)$ ]   | R1 = 0.0206, wR2 = 0.0499                                |
| R indices (all data)                   | R1 = 0.0224, wR2 = 0.0507                                |
| Goodness-of-fit on $F^2$               | 1.085  |
| Extinction coefficient                 | 0.0082(8)  |
| Largest diff. peak and hole            | 0.859 and $-1.306 \text{ e}\text{\AA}^{-3}$              |
| Maximum $\Delta/\sigma$ in final cycle | < 0.001  |
| Mean $\Delta/\sigma$ in final cycle    | < 0.001  |

Final weighting scheme:

$$\text{calc } w = 1 / [\sigma^2(F_o^2) + (0.0260P)^2 + 1.4804P]$$

$$\text{where } P = (F_o^2 + 2F_c^2) / 3$$

R-factor definitions:

$$R1 = \sum ||F_o| - |F_c|| / \sum |F_o|$$

$$wR2 = [\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]]^{1/2}$$

Structure solution, refinement, and generation of figures and tables were carried out by use of Version 5.03 of the Siemens SHELXTL program package.

Table 2. Crystal data and structure refinement for 1.

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|                                      |  |
|--------------------------------------|--|
| Empirical formula                    | $C_4H_{10}CeCl_3O_2$   |
| Formula weight                       | 336.59   |
| Temperature                          | 158 K  |
| Wavelength                           | 0.71073 Å  |
| Crystal system                       | Triclinic  |
| Space group                          | $P\bar{1}$   |
| Unit cell dimensions                 | $a = 6.691(2)$ Å $\alpha = 84.46(2)^\circ$<br>$b = 7.433(2)$ Å $\beta = 76.72(2)^\circ$<br>$c = 10.092(2)$ Å $\gamma = 74.76(3)^\circ$ |
| Volume, Z                            | $471.0(2)$ Å <sup>3</sup> , 2  |
| Density (calculated)                 | $2.374$ Mg/m <sup>3</sup>  |
| Absorption coefficient               | $5.627$ mm <sup>-1</sup>   |
| F(000)                               | 318  |
| Crystal size                         | 0.21 x 0.20 x 0.20 mm  |
| $\theta$ range for data collection   | 2.07 to $25.00^\circ$  |
| Limiting indices                     | $-2 \leq h \leq 7$ , $-8 \leq k \leq 8$ , $-12 \leq l \leq 11$   |
| Reflections collected                | 1807   |
| Independent reflections              | 1650 ( $R_{int} = 0.0248$ )  |
| Absorption correction                | Semi-empirical from psi-scans  |
| Max. and min. transmission           | 0.4904 and 0.3882  |
| Refinement method                    | Full-matrix least-squares on $F^2$   |
| Data / restraints / parameters       | 1650 / 0 / 100   |
| Goodness-of-fit on $F^2$             | 1.085  |
| Final R indices [ $I > 2\sigma(I)$ ] | $R1 = 0.0206$ , $wR2 = 0.0499$   |
| R indices (all data)                 | $R1 = 0.0224$ , $wR2 = 0.0507$   |
| Largest diff. peak and hole          | 0.859 and $-1.306$ eÅ <sup>-3</sup>  |

$$R1 = \sum ||F_o| - |F_c|| / \sum |F_o|$$

$$wR2 = [\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]]^{1/2}$$

J4584-8

Table 3. Atomic coordinates [ $\times 10^4$ ] and equivalent isotropic displacement parameters [ $\text{\AA}^2 \times 10^3$ ] for 1.  $U(\text{eq})$  is defined as one third of the trace of the orthogonalized  $U_{ij}$  tensor.

|       | x         | y        | z        | U(eq) |
|-------|-----------|----------|----------|-------|
| Ce(1) | -2839(1)  | 7051(1)  | 9398(1)  | 9(1)  |
| Cl(1) | -3989(2)  | 9886(1)  | 11452(1) | 13(1) |
| Cl(2) | -3272(2)  | 3470(1)  | 8879(1)  | 15(1) |
| Cl(3) | -1420(2)  | 4934(1)  | 11631(1) | 14(1) |
| O(1)  | -2431(5)  | 7019(4)  | 6847(3)  | 16(1) |
| O(2)  | -622(5)   | 9322(5)  | 8591(3)  | 17(1) |
| C(1)  | -1227(8)  | 8103(7)  | 5847(4)  | 23(1) |
| C(2)  | -1586(8)  | 7747(8)  | 4479(5)  | 28(1) |
| C(3)  | -3721(11) | 7283(12) | 4854(6)  | 54(2) |
| C(4)  | -3898(9)  | 6460(9)  | 6203(5)  | 37(1) |

Table 4. Bond lengths [Å] and angles [°] for 1.

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|                       |            |                       |            |
|-----------------------|------------|-----------------------|------------|
| Ce(1)-O(2)            | 2.492(3)   | Ce(1)-O(1)            | 2.528(3)   |
| Ce(1)-Cl(3)           | 2.8232(12) | Ce(1)-Cl(3)#1         | 2.8465(14) |
| Ce(1)-Cl(2)           | 2.8606(12) | Ce(1)-Cl(2)#2         | 2.8747(14) |
| Ce(1)-Cl(1)#3         | 2.8883(14) | Ce(1)-Cl(1)           | 2.9304(13) |
| Cl(1)-Ce(1)#3         | 2.8883(14) | Cl(2)-Ce(1)#2         | 2.8747(14) |
| Cl(3)-Ce(1)#1         | 2.8465(14) |                       |            |
| O(1)-C(4)             | 1.455(6)   | O(1)-C(1)             | 1.458(5)   |
| C(1)-C(2)             | 1.514(6)   | C(2)-C(3)             | 1.513(8)   |
| C(3)-C(4)             | 1.429(8)   |                       |            |
| O(2)-Ce(1)-O(1)       | 78.76(10)  | O(2)-Ce(1)-Cl(3)      | 108.17(9)  |
| O(1)-Ce(1)-Cl(3)      | 142.47(7)  | O(2)-Ce(1)-Cl(3)#1    | 70.79(9)   |
| O(1)-Ce(1)-Cl(3)#1    | 74.03(7)   | Cl(3)-Ce(1)-Cl(3)#1   | 73.90(4)   |
| O(2)-Ce(1)-Cl(2)      | 142.59(8)  | O(1)-Ce(1)-Cl(2)      | 74.42(7)   |
| Cl(3)-Ce(1)-Cl(2)     | 79.83(4)   | Cl(3)#1-Ce(1)-Cl(2)   | 77.03(4)   |
| O(2)-Ce(1)-Cl(2)#2    | 144.64(8)  | O(1)-Ce(1)-Cl(2)#2    | 118.35(7)  |
| Cl(3)-Ce(1)-Cl(2)#2   | 77.75(4)   | Cl(3)#1-Ce(1)-Cl(2)#2 | 141.23(3)  |
| Cl(2)-Ce(1)-Cl(2)#2   | 72.35(4)   | O(2)-Ce(1)-Cl(1)#3    | 81.00(9)   |
| O(1)-Ce(1)-Cl(1)#3    | 70.91(7)   | Cl(3)-Ce(1)-Cl(1)#3   | 145.81(3)  |
| Cl(3)#1-Ce(1)-Cl(1)#3 | 138.42(3)  | Cl(2)-Ce(1)-Cl(1)#3   | 113.30(4)  |
| Cl(2)#2-Ce(1)-Cl(1)#3 | 76.87(4)   | O(2)-Ce(1)-Cl(1)      | 72.22(8)   |
| O(1)-Ce(1)-Cl(1)      | 136.63(7)  | Cl(3)-Ce(1)-Cl(1)     | 78.38(3)   |
| Cl(3)#1-Ce(1)-Cl(1)   | 122.98(4)  | Cl(2)-Ce(1)-Cl(1)     | 143.92(3)  |
| Cl(2)#2-Ce(1)-Cl(1)   | 75.15(4)   | Cl(1)#3-Ce(1)-Cl(1)   | 73.24(4)   |
| Ce(1)#3-Cl(1)-Ce(1)   | 106.76(4)  | Ce(1)-Cl(2)-Ce(1)#2   | 107.65(4)  |
| Ce(1)-Cl(3)-Ce(1)#1   | 106.10(4)  |                       |            |
| C(4)-O(1)-C(1)        | 108.4(3)   | C(4)-O(1)-Ce(1)       | 123.6(3)   |
| C(1)-O(1)-Ce(1)       | 125.0(2)   | O(1)-C(1)-C(2)        | 105.7(4)   |
| C(3)-C(2)-C(1)        | 102.6(4)   | C(4)-C(3)-C(2)        | 106.4(5)   |
| C(3)-C(4)-O(1)        | 108.2(4)   |                       |            |

Symmetry transformations used to generate equivalent atoms:

#1 -x, -y+1, -z+2

#2 -x-1, -y+1, -z+2

#3 -x-1, -y+2, -z+2

**Table 5. Anisotropic displacement parameters [ $\text{\AA}^2 \times 10^3$ ] for 1.**

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The anisotropic displacement factor exponent takes the form:

$$-2\pi^2 [ (ha^*)^2 U_{11} + \dots + 2hka^* b^* U_{12} ]$$

|       | U11   | U22    | U33   | U23   | U13    | U12    |
|-------|-------|--------|-------|-------|--------|--------|
| Ce(1) | 11(1) | 9(1)   | 5(1)  | 2(1)  | -1(1)  | -3(1)  |
| Cl(1) | 17(1) | 13(1)  | 10(1) | 2(1)  | -4(1)  | -2(1)  |
| Cl(2) | 17(1) | 12(1)  | 15(1) | -1(1) | 3(1)   | -6(1)  |
| Cl(3) | 14(1) | 17(1)  | 7(1)  | 3(1)  | 0(1)   | -1(1)  |
| O(1)  | 18(2) | 24(2)  | 8(1)  | 2(1)  | -1(1)  | -9(1)  |
| O(2)  | 14(2) | 11(2)  | 25(2) | -1(1) | 1(1)   | -5(1)  |
| C(1)  | 31(3) | 34(3)  | 10(2) | 9(2)  | -3(2)  | -20(2) |
| C(2)  | 37(3) | 40(3)  | 9(2)  | 6(2)  | -4(2)  | -16(2) |
| C(3)  | 64(4) | 101(6) | 18(3) | 14(3) | -15(3) | -59(4) |
| C(4)  | 35(3) | 73(4)  | 17(3) | 2(3)  | -8(2)  | -36(3) |

Table 6. Hydrogen coordinates ( $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for 1.

J4584-11

|       | x         | y         | z        | U(eq)  |
|-------|-----------|-----------|----------|--------|
| H(1A) | 299(8)    | 7692(7)   | 5866(4)  | 28     |
| H(1B) | -1738(8)  | 9447(7)   | 6038(4)  | 28     |
| H(2A) | -1626(8)  | 8870(8)   | 3862(5)  | 34     |
| H(2B) | -467(8)   | 6691(8)   | 4038(5)  | 34     |
| H(3A) | -4875(11) | 8428(12)  | 4828(6)  | 64     |
| H(3B) | -3809(11) | 6397(12)  | 4210(6)  | 64     |
| H(4A) | -3548(9)  | 5084(9)   | 6167(5)  | 45     |
| H(4B) | -5367(9)  | 6890(9)   | 6733(5)  | 45     |
| H(1)  | -1163(79) | 10362(78) | 8692(50) | 12(13) |
| H(2)  | 571(109)  | 9265(88)  | 8559(66) | 41(19) |