Supporting Information

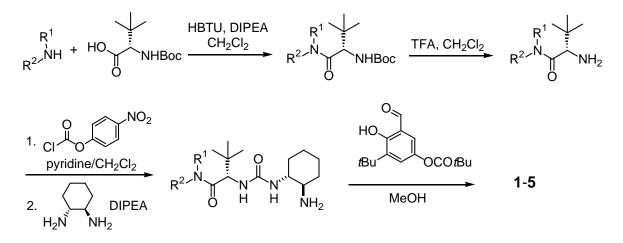
Structure-Based Analysis and Optimization of a Highly Enantioselective Catalyst for the Strecker Reaction

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General: Boc-L-*tert*-leucine was purchased from Fluka, O-Benztriazole-1-N,N,N',N'tetraethyluronium hexafluorophosphate (HBTU) from Advanced ChemTech, (*S*)-*N*-*tert*-Butoxycarbonyl-2-amino-3-methyl-3-phenylbutyric acid *tert*-butylamine salt was purchased from ChiroTech; unless stated otherwise, all other chemicals were purchased from Aldrich or Alfa Aesar and used without purification. (*R*,*R*)-1,2-Diaminocyclohexane was resolved by literature methods.¹ Imine substrates and Strecker adducts were prepared according to published procedures.² ¹⁵*N*-Benzylamine for the synthesis of isotopically labeled ¹⁵*N*-2,2-dimethylpropylidene benzylamine was prepared in two steps from ¹⁵*N*amonium chloride according to literature procedure.³ 2-Hydroxy-5-pivaloyloxy-3-*tert*butylbenzaldehyde was prepared according to published procedure.²

General Procedure for the Preparation of the Urea Catalysts (Illustrated for 1)⁴



Coupling of Boc-L-tert-*leucine with benzylamine, followed by deprotection*: A 1000-mL round bottom flask equipped with a stirbar was charged with 5.00 g (21.6 mmol) of Boc-L-*tert*-leucine. Dichloromethane (170 mL) and HBTU (8.21 g, 1.0 eq.) were added with stirring. After 2 min, DIPEA (7.55 mL, 2 eq.) and benzylamine (2.37

mL, 1.0 eq.) were added sequentially and the reaction was stirred for 90 min. The mixture was combined with dichloromethane (250 mL) and water (250 mL) and the organic layer was separated, washed three times with 1N hydrochloric acid (250 mL), and dried over sodium sulfate. Solvents were removed in vacuo to afford crude Boc-protected amide as colorless oil. The oil was dissolved in dichloromethane (110 mL); then trifluoroacetic acid (25 mL, 15 eq.) was added in one portion and the reaction was stirred at rt for 1 hour. The reaction mixture was then cooled to 0°C and a 20% aqueous solution of sodium carbonate (250 mL) was added slowly. The resulting biphasic mixture was transferred to a separatory funnel, diluted with chloroform (140 mL), and the organic and aqueous layers were separated. The organic layer was washed with a 20% aqueous solution of sodium carbonate (250 mL). The combined aqueous layers were washed with chloroform (3 x 150 mL). All organic phases were combined, dried over sodium sulfate and concentrated to afford a mixture of product and tetramethylurea as a white solid (4.71 g, 21.4 mmol, 99% over two steps based on crude mass and ¹H NMR analysis). The mixture was carried on to the next step without further purification. The spectral properties are as follows: mp 53-54 °C: IR (KBr) 3303, 1650 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.33 (m, 5H), 7.05 (s, 1H), 4.45 (d, J = 0.9 Hz, 1H), 4.43 (d, J = 0.9 Hz, 1H), 3.14 (s, 1H), 1.41 (s, 2H), 1.01 (s, 9H); ¹³C NMR {¹H} (100 MHz, CDCl₃) δ 173.4, 138.5, 128.5, 127.8, 127.3, 64.3, 43.0, 34.1, 26.7; HRMS (M + H) calcd 221.1654, obsd 221.1658.

Carbamate and urea formation: A 500mL-round-bottom flask equipped with a stir bar was flame-dried and charged with the entire amount of crude amine obtained from the previous step (4.71 g, 21.4 mmol) dissolved in freshly distilled dichloromethane (50 mL). Freshly distilled pyridine (3.49 mL, 2 equiv.) was added via syringe to the stirred solution; after 2 min, 4-nitrophenylchloroformate (4.44 g, 1.02 equiv.) was added in one portion. After the reaction was stirred for 10 min, (R,R)-1,2-diaminocyclohexane (7.40 g, 3 equiv.) was added in one portion, followed by addition of DIPEA (4.2 mL, 1.1 equiv.) via syringe, and the reaction mixture was stirred for an additional 10 min. The resulting mixture was then combined with dichloromethane (500 mL) and 0.5 M sodium hydroxide solution (120 mL). The organic layer was separated, washed with another portion of 0.5 M sodium hydroxide solution (120 mL), and dried over sodium sulfate. The organic layer was concentrated to afford viscous oil, which was suspended in hexanes (500 mL). The resulting mixture was allowed to stand for 30 min, and then filtered, with the collected solids then washed with (3 x 125 mL) hexanes. The product was obtained as a white powder (6.25 g, 17.3 mmol, 82% yield over 2 steps) with no impurities detectable by ¹H NMR analysis (for some urea catalysts, the amine product was purified by flash chromatography on silica gel; eluent: 2M solution of amonia in methanol/dichloromethane = 1/9): IR (thin film) 3284, 2934, 2858, 1631, 1555 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.28 (m, 5H), 7.08 (s, 1 H), 6.11 (s, 1H), 5.31 (s, 1H), 4.48 $(dd, J_1 = 14.9 Hz, J_2 = 6.1 Hz, 1H), 4.26 (dd, J_1 = 14.9 Hz, J_2 = 5.1 Hz, 1H), 4.20 (d, J = 14.9 Hz, J_2 = 5.1 Hz, 1H)$ 8.8 Hz, 1H), 3.20 (m, 1H), 2.31 (m, 1H), 1.98 (d, J = 11.7 Hz, 1H), 1.85 (m, 2H), 1.68 (d, J = 11.2 Hz, 2H), 1.16 (m, 5H), 1.03 (s, 9H); ¹³C NMR {¹H} (100 MHz, CDCl₃) δ 172.7, 159.0, 138.5, 128.5, 127.5, 127.1, 61.3, 57.0, 55.1, 43.1, 35.0, 34.7, 33.4, 27.1, 25.3, 25.1.

Schiff base formation: A 1L-round-bottom flask equipped with a stirbar was charged with 6.25 g of amine prepared in the previous step and anhydrous methanol (40 mL) was added with stirring. Once the solution became homogeneous, sodium sulfate

(10 g) was added. In a separate flask, 2-hydroxy-5-pivaloyloxy-3-*tert*-butylbenzaldehyde (4.73 g, 0.98 eq.) was dissolved in anhydrous methanol (40 mL), then transferred to the reaction mixture. An additional 30 mL of methanol was used to effect quantitative transfer of the aldehyde into the reaction mixture. The reaction mixture was stirred for 90 min, then concentrated under reduced pressure with the sodium sulfate still present. The resulting mixture was combined with hexanes (250 mL) and filtered through a Buchner funnel, and the solids were rinsed with hexanes (250 mL). The filtrate was concentrated under reduced pressure to yield 10.55 g of 1 as a yellow solid (17.0 mmol, 98% yield, 80% overall yield from Boc-L-tert-leucine): IR (KBr) 3309, 2960, 1752, 1684, 1550, 1437, 1270, 1150, 1116 cm⁻¹; ¹H NMR (400 MHz, C₆D₆) δ 14.32 (s, 1H), 8.08 (s, 1H), 7.23 (d, J = 2.6 Hz, 1H), 7.12 (m, 2H), 7.08 (m, 2H), 7.02 (d, J = 2.6 Hz, 1H), 7.00 (t, J = 2.6 Hz, 2H), 7 7.1 Hz, 1H), 6.67 (m, 1H), 5.63 (m, 1H), 4.59 (m, 1H), 4.37 (dd, J = 14.8, 6.6 Hz, 1H), 4.29 (d, J = 9.1 Hz, 1H), 3.86 (dd, J = 14.8, 4.6 Hz, 1H), 3.43 (m, 1H), 3.15 (m, 1H), 1.95 (m, 1H), 1.68- 1.0 (m, 7H), 1.51 (s, 9H), 1.30 (s, 9H), 1.05 (s, 9H); ${}^{13}C$ NMR { ${}^{1}H$ } (100 MHz, CDCl₃) δ 177.4, 172.0, 164.3, 158.1, 157.7, 141.7, 138.5, 130.1, 128.4, 127.4, 127.1, 122.6, 121.2, 118.1, 70.3, 61.5, 54.0, 43.1, 38.9, 34.8, 34.7, 31.5, 29.1, 27.1, 26.7, 24.2, 23.6, 22.6; HRMS (ES) (M)⁺ calcd 621.4016, obsd 621.3986.

2: Catalyst was prepared in 50% overall yield according to the general procedure for the synthesis of urea catalysts. The spectral properties are as follows: IR (thin film) 3368, 1750, 1633, 1550cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 14.22 (s, 1H), 8.22 (s, 1H), 7.24 (m, 3H), 7.14 (m, 2H), 6.94 (d, *J* = 6 Hz, 1H), 6.76 (d, *J* = 6 Hz, 1H), 5.38 (d, *J* = 9 Hz, 1H), 5.33 (d, *J* = 8 Hz, 1H), 4.81 (m, 1H), 4.70 (m, 1H), 4.38 (d, *J* = 15 Hz, 0.3H), 4.10 (d, *J* = 15 Hz, 0.7H), 3.34 (m, 1H), 3.17 (m, 1H), 2.91 (s, 0.7x3H), 2.79 (s, 0.3x3H), 1.99 (d, *J* = 12Hz, 1H), 1.81 (d, 12Hz, 1H), 1.69 (m, 2H), 1.21-1.50 (m, 4H), 1.40 (s, 0.7x9H), 1.39 (s, 0.3x9H), 1.34 (s, 0.7x9H), 1.33 (s, 0.3x9H), 0.91 (s, 0.7x9H), 0.86 (s, 0.3x9H); ¹³C NMR {¹H} (100 MHz, CDCl₃) δ 177.4, 173.01, 164.1, 158.2, 141.8, 138.6, 136.9, 129.9, 128.2, 128.1, 127.8, 127.3, 126.4, 122.6, 121.2, 118.2, 72.0, 69.0, 55.2, 49.2, 38.9, 36.4, 36.2, 34.9, 33.2, 31.5, 29.2, 27.2, 24.2, 23.8; MS (MH)⁺ 635.6.

3: Catalyst was prepared in 34% overall yield according to the general procedure for the synthesis of urea catalysts. The spectral properties are as follows: IR (thin film) 3370, 1748, 1632, 1550, 1438 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 13.78 (s, 1H), 8.27 (s, 1H), 7.26 (s, 6H), 6.94 (s, 4H), 6.94 (d, *J*= 2.6 Hz, 1H), 6.77 (s, 1H), 5.07 (d, *J*= 14.6 Hz, 1H), 4.88 (d, *J*= 9.2 Hz, 1H), 4.77 (d, *J*= 16.1 Hz, 1H), 4.59 (s, 1H), 4.26 (d, *J*= 16.1 Hz, 1H), 3.90 (d, *J*= 14.6 Hz, 1H), 3.41 (s, 1H), 3.23 (s, 1H), 2.07 (s, 1H), 1.88 (m, 1H), 1.79-1.60 (m, 4H), 1.55-1.36 (m, 2H), 1.39 (s, 9H), 1.33 (s, 9H), 0.89 (s, 9H); ¹³C NMR {¹H} (100 MHz, CDCl₃) δ 177.4, 173.3, 164.3, 158.2, 157.0, 128.7, 128.7, 128.1, 127.8, 127.7, 127.5, 122.6, 121.2, 118.2, 70.3, 55.2, 54.3, 50.8, 47.7, 38.9, 36.0, 34.9, 33.0, 31.1, 29.2, 27.1, 26.9, 26.6, 24.4, 23.5; MS (TOF) m/z (M+H) calcd 711.4, obs 711.6.

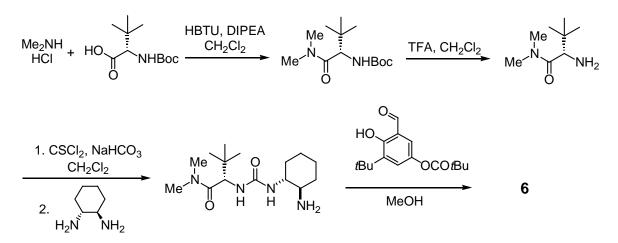
4: *Coupling with Boc-L*-tert-*leucine* was performed using dimethylamine hydrochloride (1equiv) and DIPEA (3 equiv).

Boc deprotection: A 25-mL, round-bottomed flask equipped with a stirbar was charged with N-(tert-Butoxycarbonyl)-L-tert-Leucine N,N-dimethylamide (2.0 mmol). A

4 M solution of hydrogen chloride in 1,4-dioxane (5 mL) was added at room temperature with stirring. After 2 hours, solvents were removed in vacuo. The product was used as hydrochloric salt in the subsequent step without further purification. The subsequent steps were performed according to the general procedure for urea catalysts (3 equivalents of pyridine were used for the urea formation). The product was isolated as a yellow solid in 16% overall yield. The spectral properties are as follows: IR (thin film) 3400, 1750, 1633, 1557, 1437 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 13.72 (s, 1H), 8.27 (s, 1H), 6.94 (d, *J*= 2.9 Hz, 1H), 6.79 (d, *J*= 2.6 Hz, 1H), 5.06 (s, 1H), 4.65 (d, *J*= 9.2 Hz, 1H), 4.35 (s, 1H), 3.55 (s, 1H), 3.13 (m, 1H), 2.99 (s, 3H), 2.82 (s, 3H), 2.12-2.07 (m, 1H), 1.92-1.84 (m, 1H), 1.82-1.62 (m, 4H), 1.44-1.34 (m, 2H), 1.40 (s, 9H), 1.34 (s, 9H), 0.89 (s, 9H); ¹³C NMR {¹H} (100 MHz, CDCl₃) δ 177.4, 172.7, 164.0, 158.2, 157.2, 141.7, 138.5, 122.5, 121.1, 118.2, 71.3, 54.9, 54.1, 38.9, 38.2, 35.6, 35.5, 34.8, 33.2, 31.6, 29.2, 27.1, 26.4, 24.6, 23.8; MS (TOF) m/z (M+H) calcd 559.4, obs 559.5.

5: The catalyst was prepared in 31% overall yield from the Boc-amino acid salt (3 equiv. of DIPEA used for the initial coupling), using the general synthetic protocol for the urea catalysts. The spectral properties of the yellow product are as follows: IR (thin film) 2934, 1751, 1632, 1549, 1439 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 13.72 (s, 1H), 8.31 (s, 1H), 7.37 (d, *J*= 7.3 Hz, 2H), 7.26 (t, *J*= 6.7 Hz, 2H), 7.18 (t, *J*= 7.3 Hz, 1H), 6.93 (d, *J*= 2.7 Hz, 1H), 6.81 (s, 1H), 5.24 (s, 1H), 4.86 (d, *J*= 7.9 Hz, 1H), 4.41 (s, 1H), 3.56 (m, 1H), 3.19 (m, 1H), 2.57 (s, 3H), 3.09 (s, 3H), 1.89 (d, *J*= 11.3 Hz, 1H), 1.77 (m, 3H), 1.67 (d, *J*= 13.7 Hz, 1H), 1.48-1.33 (m, 3H), 1.39 (s, 9H), 1.35 (s, 3H), 1.33 (s, 12H); ¹³C NMR {¹H} (100 MHz, CDCl₃) δ 177.4, 171.7, 164.1, 158.2, 157.1, 145.9, 141.7, 138.6, 127.9, 126.8, 126.4, 122.6, 121.1, 118.2, 71.4, 56.1, 54.3, 42.5, 38.9, 36.7, 35.1, 34.9, 33.3, 31.6, 29.2, 27.1, 26.4, 24.7, 23.8, 22.4; MS (TOF) m/z (M+H) calcd 621.4, obs 621.3.

Synthesis of catalyst 6:

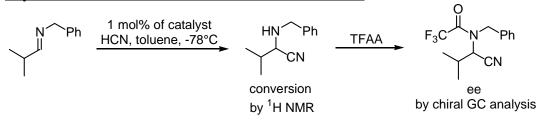


Coupling of Boc-L-tert-*leucine with benzylamine, followed by deprotection* was performed according to the procedure described for the preparation of **4**.

Thiourea formation:⁵ A 500mL-round bottom flask equipped with a stir bar was charged with 1.86 g (9.54 mmol) of crude amine hydrochloride from the deprotection step. To this mixture dichloromethane (50 mL), and saturated aqueous solution of sodium bicarbonate was added. The biphasic mixture is cooled to 0°C and neat thiophosgene (0.80 mL, 1.1 equiv.) was added via syringe with vigorous stirring. The reaction mixture was vigorously stirred at 0°C for additional 30 min, the organic layer was separated. dried over sodium sulfate, and concentrated in vacuo to afford isothiocynate used immediately without purification. The crude isothiocyanate was dissolved in freshly distilled dichloromethane (30 mL) and (R,R)-1,2-diaminocyclohexane (1.20 g, 1.1 equiv.) was added in one portion. The reaction mixture was allowed to stir at room temperature for 30 min and concentrated in vacuo. Crude product was purified by flash chromatography on silica gel (Eluent: 2M solution of ammonia in methanol/dichloromethane = 1/9, stain with ninhydrine) to afford 2.13 g (75% overall from Boc-L-tert-leucine) of pure amine.

Schiff base formation was performed according to the general procedure for the preparation of urea catalysts (1.00 equivalents of aldehyde used). Catalyst **6** was isolated as a yellow solid in 75% overall yield (form Boc-L-*tert*-leucine). The spectral properties are as follows: IR (thin film) 3293, 1750, 1630, 1535cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 13.55 (s, 1H), 8.23 (s, 1H), 6.92 (d, *J* = 8 Hz, 1H), 6.85 (d, *J* = 8 Hz, 1H), 6.55 (s, 1H), 6.44 (d, *J* = 7 Hz, 1H), 5.56 (d, *J* = 9 Hz, 1H), 3.83 (m, 1H), 3.18 (s, 3H), 3.11 (td, *J* = 2, 8Hz, 1H), 2.92 (s, 3H), 2.09 (m, 1H), 1.88 (m, 1H), 1.71 (m, 3H), 1.25-1.44 (m, 3H), 1.40 (s, 9H), 1.34 (s, 9H), 0.90 (s, 9H); ¹³C NMR {¹H} (100 MHz, CDCl₃) δ 177.6, 172.0, 165.0, 157.9, 141.9, 135.5, 123.0, 121.6, 118.2, 60.4, 38.9, 38.4, 36.0, 35.6, 34.9, 29.2, 27.2, 26.6; MS (MH)⁺ 575.6.

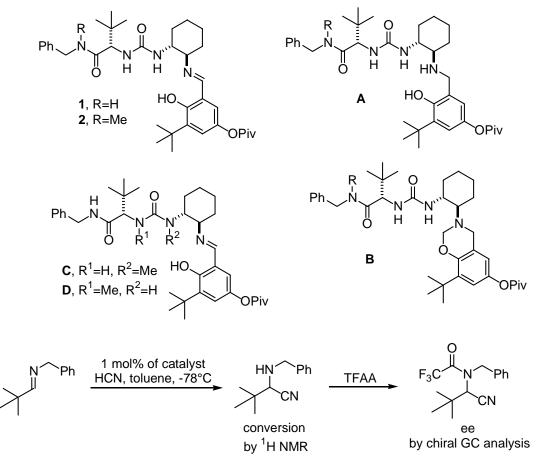
Asymmetric Strecker reaction of model imine substrate:



A flame-dried 10mL-round-bottom flask equipped with a stir bar was charged with 0.003 mmol of a catalyst (0.01 equiv.), 2.5 mL of toluene, and imine (0.3 mmol). The reaction was stirred at ambient temperature until catalyst completely dissolved and then cooled below -70 °C by means of a constant temperature bath. A flame-dried 5mL-recovery-flask was equipped with a stir bar and charged with freshly distilled toluene (0.5 mL) and 50 µL TMSCN (1.25 equiv.). The mixture was cooled to 0°C; to this solution 15 µL of anhydrous methanol (1.25 equiv.) was added via syringe; the solution was allowed to stir at 0°C for 2h and then added to reaction flask containing catalyst and substrate by a syringe addition at -78°C. After 15h, a sample was transferred into a precooled vial via a precooled syringe and the solvents were removed under reduced pressure at low temperature to ensure no reaction progress during this operation. The conversion of the Strecker reaction was determined by ¹H NMR (product/imine). To the NMR sample, an

excess of TFAA (approx. 5 equiv.) was added at room temperature in one portion and the resulting trifluoroacetamide of the Strecker adduct was analyzed by chiral GC ananlysis.²

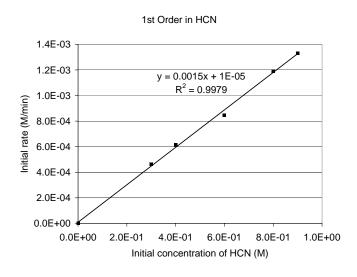
<u>Identification of the active site of the catalyst:</u> Several derivatives of 1 were prepared and tested as catalyst in the asymmetric Strecker reaction of N-(2,2-dimethylpropylidene) benzylamine; the screen was conducted according to the general procedure given for N-isobutylene benzylamine. The results and comments are summarized in the following Table:

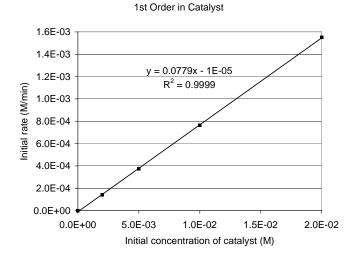


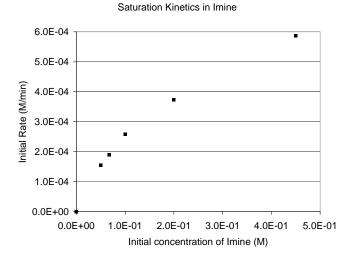
Entry	Catalyst	$\operatorname{Conv.}^{a}(\%)$	Ee (%)	Comments
1	1	> 99	95.6	
2	2	> 99	96.4	2 even better catalyst than $1 \rightarrow$ amide H not active site
3	Α	> 99	85	Ee declines from 1 (not surprise: weaker H-bond, etc.)
4	В	> 99	88	B better catalyst than $\mathbf{A} \rightarrow$ phenol H not active site
5	С	83	13	C and D are significantly worse catalysts than $1 \rightarrow$
6	D	46	27	\rightarrow urea H(s) are important for both rate and ee; might be the active site
				DE LIE ACTIVE SILE

a: in 15h at -78°C

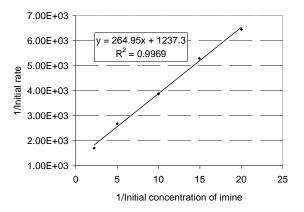
Kinetics, General: Reaction kinetics were investigated using an ASI 1000 React-IRTM instrument equipped with a silicon probe. The probe was dried by heating with a heat gun (<200°C) and allowed to cool under nitrogen atmosphere prior to each run to ensure reproducible results. Rate dependence on the concentration of each reagent was investigated under pseudo-constant concentration of remaining reagents by monitoring the change of the initial rate (10% conversion) as a function of investigated reagent concentration. The Strecker reaction was conducted by the means of the general protocol given for the testing of catalyst derivatives. The observed dependence on the concentration of each reagent is given bellow (Charts 1-3). Saturation kinetics in imine substrate was confirmed by Lineweaver Burk plot (Chart 4):









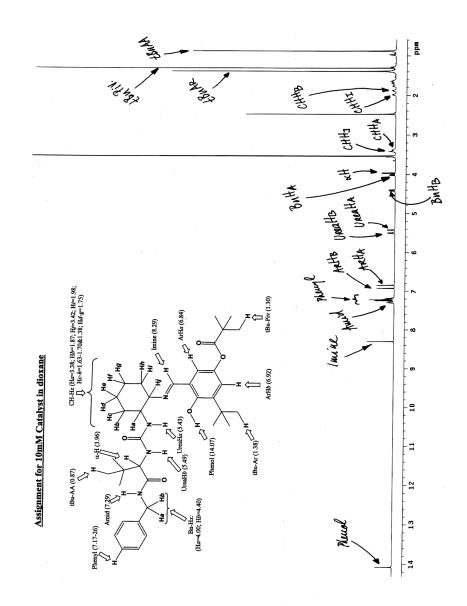


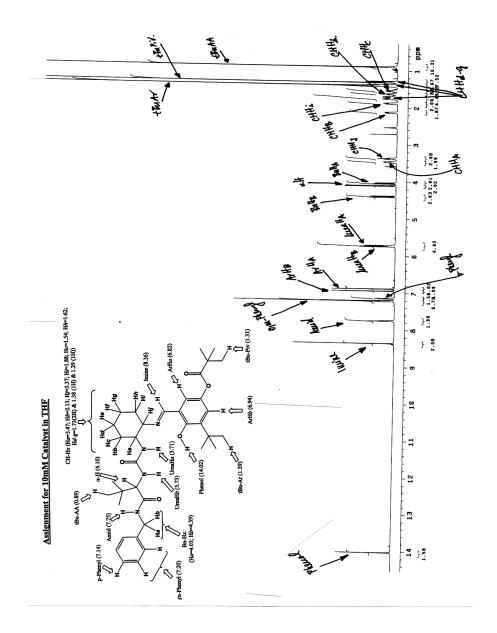
NMR Spectroscopy, General: All experiments were performed using instruments equipped with a Brucker magnet and Varian software. Experiments at room temperature (20°C, regulated) were performed using INOVA 600 MHz instrument. Low temperature (regulated) experiments were performed using INOVA 500 MHz, or MERCURY 400 MHz instruments equipped with an external thermostat filled with liquid nitrogen. All NMR solvents were purchased from Cambridge Isotope Laboratories and used as received. Prior to the ROESY and NOE experiments, the NMR sample was degassed with nitrogen for 10 min.

Catalyst 1:

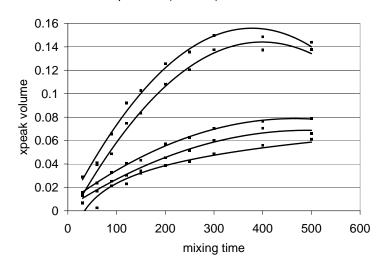
Assignment of H-signals in relevant solvents (THF and dioxane) for ¹H NMR was accomplished by COSY connectivity experiments; tBuPiv, tBuAr, ArHa, and ArHb were assign based on their NOE interactions (NOESY, ROESY):

10mM **1** in d_8 -dioxane:



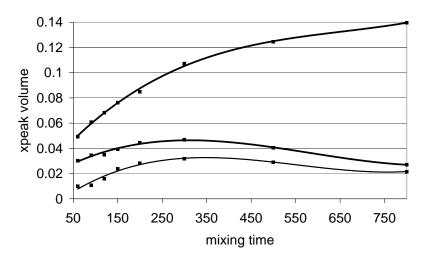


Determination of the correct mixing time for 2D-ROESY experiments was accomplished by monitoring of NOE build-up curves (mix=30-600ms) for significant xpeaks in ROESY. Charts below show selected xpeaks for 10mM 1 in d_8 -dioxane (determined as 120 ms) and 50mM 1 in d_8 -THF (determined as 160ms; unchanged when imine substrate was present); other conditions: i600, d1=2s, t=20°C (regulated), nt=4, ni=350, sw=9000.

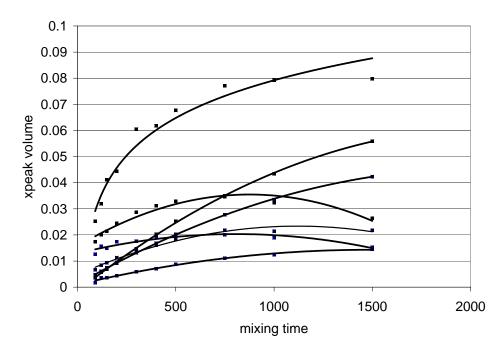


NOE build up curves (ROESY): 10 mM 1 in dioxane





Determination of the correct mixing time for NOESY, and NOESY1D experiments was accomplished by monitoring NOE build up curves for significant xpeaks in NOESY experiment. Chart below shows selected xpeaks for 50mM **1** in d_8 -THF (determined as 160ms; unchanged if imine substrate was present); other conditions: i600, d1=2s, t=20°C (regulated), nt=4, ni=350, sw=9000.



NOE build up curves (NOESY): 50mM 1 in THF

ROESY Experiment, 50mM 1 in d₈-*THF*, 20°C (regulated), i600, d1=1s, ni=1K, sw=9000, nt=16, mix=160ms; distance in Å, compare to calculations (MM2):

in-10, inix-100ins, distance in 11, compare to calculations (11112).							
Poak 2	Xpeak	distance, Å	distance, Å				
reak 2	volume	(ROESY)	(computation)				
Imine	0.18	3.8	3.9				
UreaHa	0.09	4.2	4.3				
CHHa	0.074	4.4	4.0				
tBuAr	0.351	3.4	4.0				
ArHa	2.26	define as 2.46	(law of cosine)				
UreaHa	0.242	3.6	3.6				
CHHa	0.714	3.0	2.7				
СННј	4.15	2.2	2.4				
tBuAA	0.577	3.1	3.5				
α-H	4.13	2.2	2.2				
tBuAA	1.71	2.6	3.7				
tBuAr	4.61	2.2	2.1				
tBuAr	3.57	2.3	2.4				
UreaHa	3.325	2.3	2.3				
tBuAA	3.04	2.3	2.4				
	Peak 2 Imine UreaHa CHHa tBuAr ArHa UreaHa CHHj tBuAA α-H tBuAA tBuAr tBuAr UreaHa	Xpeak Xpeak Peak 2 volume Imine 0.18 UreaHa 0.09 CHHa 0.074 tBuAr 0.351 ArHa 2.26 UreaHa 0.242 CHHa 0.714 CHHj 4.15 tBuAA 0.577 α-H 4.13 tBuAA 1.71 tBuAr 3.57 UreaHa 3.325	$\begin{array}{c c c c c c c c c c c c c c c c c c c $				

UreaHa	CHHa	1.021	2.8	2.8
UreaHa	СННј	0.959	2.8	2.9
α-H	tBuAA	5.16	2.1	2.3
BnHb	BnHa	3.685	mix w/ cosy	2.2

ROESY Experiment, 10mM 1 *in* d_8 *-dioxane,* 20°C (regulated), i600, d1=1s, ni=1K, sw=9000, nt=16, mix=120ms:

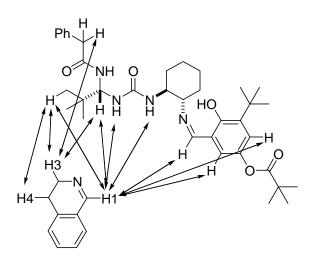
Peak 1	Peak 2	ROESY xpeak volume
Phenol (14.07)	tBu-Ar (1.38)	0.74519854
Phenol (14.07)	СН-На (3.38)	1.684133
Phenol (14.07)	Urea-Ha (5.43)	0.200816
Phenol (14.07)	Imine (8.29)	0.97271824
Imine (8.29)	tBu-AA (0.87)	1.7759821
Imine (8.29)	CH-Hb (1.87)	1.97931078
Imine (8.29)	СН-На (3.38)	7.86
Imine (8.29)	СН-Нј (3.42)	18.64
Imine (8.29)	Urea-Ha (5.43)	1.442701
Imine (8.29)	Ar-Ha (6.84)	14.717
Imine (8.29)	Ar-Hb (6.92)	5.18
Amid (7.29)	tBu-AA (0.87)	8.62042714
Amid (7.29)	a-H (3.96)	26.14315296
Amid (7.29)	Bn-Hb (4.4)	3.20317565
Phenyl (7.17-7.26)	tBu-AA (0.87)	6.48191987
Ar-Hb (6.92)	tBu-AA (0.87)	2.44275541
Ar-Hb (6.92)	tBu-Ar (1.38)	37.73484386
Ar-Ha (6.84)	tBu-AA (0.87)	4.06189233
Ar-Ha (6.84)	tBu-Ar (1.38)	15.2834654
Urea-Hb (5.49)	tBu-AA (0.87)	15.27080143
Urea-Hb (5.49)	a-H (3.96)	3.77627491
Urea-Hb (5.49)	Urea-Ha (5.43)	15
Urea-Ha (5.43)	tBu-AA (0.87)	2.8687825
Urea-Ha (5.43)	CH-Hc-h (1.63)/ CH-Hc-h (1.70)	3.69794199
Urea-Ha (5.43)	СН-На (3.38)	7.04332542
Urea-Ha (5.43)	СН-Нј (3.42)	6.04332542
a-H (3.96)	tBu-AA (0.87)	22.81397889
СН-Нј (3.42)	tBu-Ar (1.38) or CH	7.24282364
СН-Нј (3.42)	CH-Hb (1.87)	5.1718353
СН-Нј (3.42)	CH-Hi (1.98)	0.5
CH-Ha (3.38)	tBu-Ar (1.38) or CH	5.23395427
CH-Ha (3.38)	CH-Hb (1.87)	0.5000001
CH-Ha (3.38)	CH-Hi (1.98)	5.21282502
L		

CH-Hd-g (1.75)	tBu-Ar (1.38) or CH	53.46867114
CH-Hc-h (1.70)	tBu-Ar (1.38) or CH	49.25715698

NOESY1D Experiments, 70mM 1 in d_8 -*THF*, 20°C (regulated), i600, d1=1s, ni=1K, sw=9000, nt=4000-7000, mix=300ms (presence of 3,4-dihydroisoquinoline did not have an influence on the magnitude of intramolecular xpeaks; same relative volumes observed):

Iradiated	Phenol	Imine	Amide	Phenyl	ArHb	ArHa	UreaHb	UreaHa	α-H	BnHb	BnHa	СННа	СННј	CHHb	CHHi	tBuAr	tBuPiv	tBuAA
Phenol		0.20														0.99		
Imine	0.19					1.28		0.10				0.24	1.50					0.17
Amide				0.38			0.01		1.66									
ArHb		0.09														4.15	0.13	
ArHa	0.01	0.99														0.05	0.09	0.14
UreaHb		0.10																1.73
UreaHa		0.27										0.86	0.71					
α-Η			1.09															2.11
BnHb				0.41												0.03		0.11
BnHa				0.46	0.03											0.03		0.05
CHHa	0.05	0.25						0.64						0.80				
CHHj	0.04	1.67						0.42							0.82			
CHHb		0.02						0.29				1.14	0.24					
CHHi	0.01	0.27										0.22	1.07					
tBuAr	0.06				0.99													
tBuPiv					0.01	0.01										0.14		0.16
tBuAA		0.05	0.12	0.10		0.01	0.34		0.69	0.12	0.15					0.14	0.16	

Intermolecular xpeaks between catalyst 1 and 3,4-dihydroisoquinoline. The xpeaks were observed in NOESY1D experiment, 20°C (regulated), nt=4000-12000, d1=0.1-1s, sw=9000, 70mM 1 and 330mM in 3,4-dihydroisoquinoline in d_8 -THF. Since correct mixing time was not established for catalyst/substrate complex, reported xpeak volumes should be considered as qualitative, not quantitative measure of the distance. Following Figure shows important intermolecular xpeaks; integrated values are given in the Table:



Intermolecular xpeak volume:

1		
H1	H3	H4
0.01		
0.05	0.10	
0.02		
0.07		
0.12		
0.10		
0.07	0.08	
	0.10	
0.01	0.03	
0.01		
0.03	0.02	
0.12	0.13	0.18
	H1 0.01 0.05 0.02 0.07 0.12 0.10 0.07 0.01 0.01 0.01 0.03	0.01 0.05 0.10 0.05 0.10 0.02 0.07 0.12 0.10 0.10 0.07 0.08 0.07 0.08 0.10 0.07 0.03 0.02 0.01 0.03 0.02 0.03 0.02 0.02

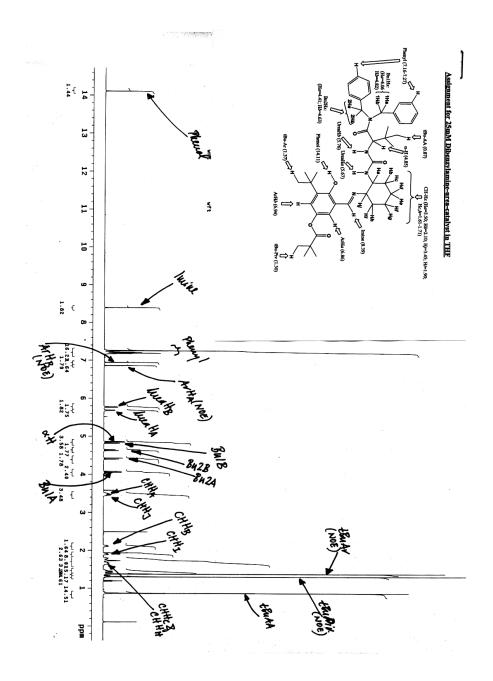
Intermolecular xpeaks between catalyst 1 and a Z-Imine. The xpeaks were observed in NOESY1D experiment, 20°C (regulated), nt=12000-25000, d1=0.1-1s, sw=9000, 100mM 1 and 90mM imine (overall concentration of both isomers) in d_8 -THF. The intensity of the intermolecular xpeaks was low and a quantitative integration was not possible, therefore qualitative comparison is given instead; four significant xpeaks are shown in the following figure:

-	Catalyst Peak	Xpeak to αMe (Z-imine)	
-	i ean		
	Phenol	W	
	Imine	S	N [×]
	Amide		H H
	ArHb	S	MeO
	ArHa	S	
	UreaHb	m	
	UreaHa	m	
	α-H	m	, ⊢ H H O I
	BnHb		
	BnHa		
	CHHa		
	CHHj		NH /
	tBuAr		
	tBuPiv		НŅ́О
	tBuAA	S	Ph
		····P	

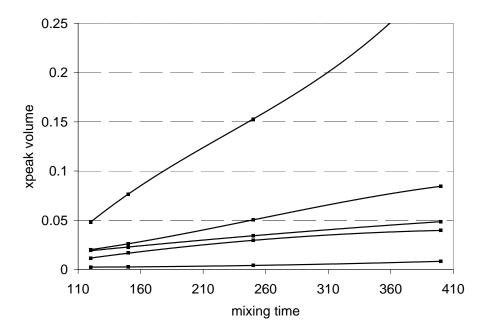
w=weak, m=medium, s=strong

Catalyst 3:

Assignment of H-signals for 25mM 3 in d₈-THF was accomplished by COSY connectivity experiments. tBuPiv, tBuAA, ArHa, and ArHb were assign based on NOE xpeaks (ROESY):



Determination of the correct mixing time for 2D-ROESY experiments was accomplished by monitoring the NOE build up curves (mix=60-400ms) for significant xpeaks in 2D-ROESY. Chart shows selected xpeaks for 25mM **3** in d_8 -THF (determined as 120ms); other conditions: i600, d1=1.5s, t=20°C (regulated), nt=4, ni=350, sw=9000.



NOE build up curves (ROESY) for 25mM 3 in THF

ROESY Experiment, 25mM 3 in d_8 -*THF*, 20°C (regulated), i600, d1=1.7s, ni=1.5K, sw=9000, nt=16, mix=120ms; distance calculated based on the relative xpeak volume related to the standard distance in Å; comments included for an easier data analysis:

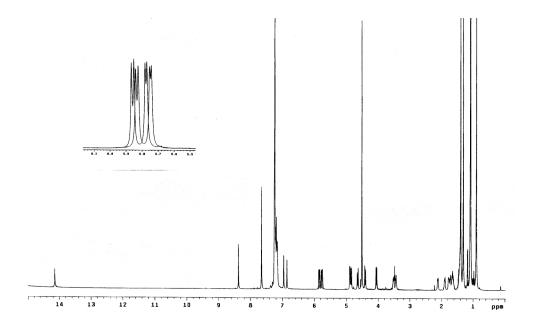
	i to the standard di	stallee III A, coll	ments menu		all easier uata allarysis.
Entry	Peak 1	Peak 2	Xpeak vol.	Dist.	Comments
1	Phenol (14.11)	CHHj (3.45)	half of 0.402	nd	overlap
2	Phenol (14.11)	CHHa (3.50)	half of 0.402	nd	overlap
3	Phenol (14.11)	Imine (8.39)	0.1953	3.5	trivial
4	Imine (8.39)	CHHc&CHHh	0.212	3.4	trivial
		(1.61-1.71)			
5	Imine (8.39)	CHHi (1.90)	0.304	3.2	sets imine-CH ring
6	Imine (8.39)	CHHj (3.45)	2.53	2.3	conformation
7	Imine (8.39)	CHHa (3.50)	1.21 ^a	nd	(Entries 5-8)
8	Imine (8.39)	UreaHa (5.67)	0.251	3.3	imine and urea: same face
9	Imine (8.39)	ArHa (6.86)	1.485	2.5	Standard for distance, 2.46A
10	Imine (8.39)	ArHb (6.94)	0.8395	2.7	trivial
11	Phenyl (7.15-7.27)	tBuAA (0.87)	2.099	2.3	very strong> hindered face
12	Phenyl (7.15-7.27)	Bn1Ha (4.06)	1.027	2.6	trivial
13	Phenyl (7.15-7.27)	Bn2Ha (4.41)	1.244	2.5	trivial
14	Phenyl (7.15-7.27)	Bn2Hb (4.63)	0.6665	2.8	trivial

15	Phenyl (7.15-7.27)	α-H (4.85)	1.544	2.4	very strong> hindered face
16	ArHb (6.94)	tBuAr (1.37)	4.917	2.0	trivial
17	ArHa (6.86)	tBuAr (1.37)	2.262	2.3	trivial
18	Urea Hb (5.76)	tBuAA (0.87)	2.4833	2.3	trivial
19	Urea Hb (5.76)	α-H (4.85)	0.4996	3.0	trivial
20	Urea Hb (5.76)	UreaHa (5.67)	0.617 ^b	nd	trivial
21	UreaHa (5.67)	CHHc&CHHh (1.61-1.71)	0.5958	2.9	sets Urea-CH ring
22	UreaHa (5.67)	CHHb (2.10)	0.2221	3.4	conformation
23	UreaHa (5.67)	CHHj (3.45)	half of 1.8166	nd	(Entries 21-24)
24	UreaHa (5.67)	CHHa (3.50)	half of 1.8166	nd	
25	α-H (4.85)	tBuAA (0.87)	3.887	2.1	trivial
26	α-H (4.85)	Bn2Ha (4.41)	2.002	2.3	confirms rigidity and potential
27	α-H (4.85)	Bn2Hb (4.63)	3.636	2.1	hindrance as origin of enantios.
28	Bn1Hb (4.82)	Bn1Ha (4.06)	9.0699 ^c	nd	trivial
29	Bn2Hb (4.63)	tBuAA (0.87)	1.731	2.4	very strong> hindered face
30	Bn2Hb (4.63)	Bn1Ha (4.06)	0.3832	3.1	weak>rigid conf., hind. face
31	Bn2Hb (4.63)	Bn2Ha (4.41)	(-) 5.6384 ^c	nd	trivial
32	Bn2Ha (4.41)	tBuAA (0.87)	1.276	2.5	very strong> hindered face
33	CHHa (3.50)	CHHb (2.10)	1.2187	2.5	trivial
34	CHHj (3.45)	CHHi (1.90)	1.2007	2.6	trivial

a: volume magnified by overlap with entry 6; *b*: volume reduced by close proximity to the diagonal peak; *c*: mixed with cosy.

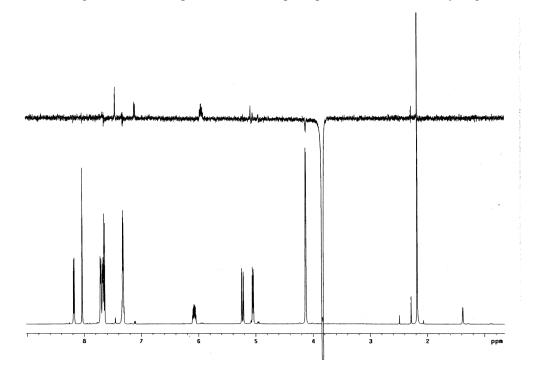
Isotopically labeled imine experiment:

Direct evidence of binding by ¹H NMR: Following Figure represents two ¹H NMR of the mixture of catalyst **3** with 2,2-dimethylpropylidene benzylamine in d_8 -THF printed on the top of each other. Both samples contain the same amount of **3** (0.162M), and the same amount of imine (0.668M). In case of sample 1, ¹⁴N-imine was used (natural abundance); In case of sample 2, ¹⁵N-imine was used (content of ¹⁵N > 99%). Frequency (δ) of none of the protons of **1** has changed ($\Delta \delta < 1$ Hz) as a function of the isotope content of the imine nitrogen, except for 2 signals: both urea-hydrogens ($\Delta \delta = 16.9$ Hz, and 16.1Hz). Urea region of catalyst **3** is enlarged in the offset. Since the only difference between the two samples is the content of ¹⁵N in the imine substrate, this result represented a direct evidence of imine nitrogen binding to the UreaH's portion of the catalyst.



Equilibrium between E-and Z-stereoisomers of imine substrate by NMR

The sample of α -methylnaphtylidene allylamine that coexists as a mixture of both stereoisomers was irradiated (saturated) with the frequency of the Z-isomer allylic proton in NOESY1D experiment in d_{12} -cyclohexane (top spectrum). Exchanging allylic protons of E-isomer ($\delta = 4.14$ ppm) and Z-isomer ($\delta = 3.84$ ppm) had the same phase (opposite phase to the NOE xpeaks) as they underwent energy transfer due to chemical exchange. The ¹H NMR spectrum of the sample is provided below for clarity (Figure follows).



Computation and Modeling, General: Calculations were performed using standard computational methods. Theoretical calculations on the simplified system were performed using Gausian 98;⁶ method B3LYP with many available basics sets. Reported data are consistent for all applied basic sets; data shown in Figure 2 of the text relate to 6-31G (d, p) basic set in the gas phase. The strength of the hydrogen bond was calculated as a difference of the energy of the complex and each individual component with dummy-atoms replacing other relevant fragments present in the complex. Calculations of the energy minimum of **1** were performed on Spartan with semi-empirical base, MM2, solvent-free as well as in hexadecane matrix. The 3D pictures of catalyst and catalyst/substrate complex were generated using Spartan and Gaussian 98 and modified in MOLMOL⁷ (see the text of the paper for relevant graphics).

Selected example of the resulting Z-matrix for imine-catalyst complex (N-ethylidene
methylamine and N,N'-dimethylthiourea; Figure 1a of the main text):
0 1

0 1				
С				
Ν	1	B1		
С	2	B2 1	A1	
Н	2	B3 1	A2 3	D1
Ν	2	B4 1	A3 3	D2
С	5	B5 2	A4 1	D3
Ν	6	B6 5	A5 2	D4
Н	7	B7 6	A6 5	D5
S	6	B8 5	A7 2	D6
С	7	B9 6	A8 5	D7
С	5	B10 2	A9 1	D8
Н	1	B11 2	A10 3	D9
Н	1	B12 2	A11 3	D10
Н	1	B13 2	A12 3	D11
Н	3	B14 2	A13 1	D12
Н	11	B15 5	A14 2	D13
Н	11	B16 5	A15 2	D14
Н	11	B17 5	A16 2	D15
Н	10	B18 7	A17 6	D16
Н	10	B19 7	A18 6	D17
Н	10	B20 7	A19 6	D18
С	3	B21 2	A20 1	D19
Н	22	B22 3	A21 2	D20
Н	22	B23 3	A22 2	D21
Н	22	B24 3	A23 2	D22
B1 B2 B3 B4 B5 B6 B7 B8 B9 B10 B11 B12	1.4557 1.2762 2.2397 3.1821 1.3606 1.3607 1.0150 1.6892 1.4497 1.449 1.095 1.095	237 701 07 597 724 077 246 707 708 109		

D12	1 005522
B13	1.095523
B14	1.095273
B15	1.093754
B16	1.095671
B17	1.093286
B18	
	1.093715
B19	1.093289
B20	1.095708
B21	1.504538
B22	1.096302
B23	1.096289
B24	1.089971
A1	122.568110
A2	125.590662
A3	132.352264
A4	101.892446
A5	113.793423
A6	117.226997
A7	123.099027
A8	123.695380
A9	134.262370
A10	108.509599
	108.503048
A11	
A12	115.769535
A13	114.856042
A14	111.033669
A15	111.775786
A16	108.033520
A17	111.016624
A18	108.035103
A19	111.789957
A20	130.784654
A21	109.221987
A22	109.229639
A23	114.402737
D1	145.373132
D2	151.123449
D3	114.066421
D4	-6.217759
D5	-4.120913
D6	173.088171
D7	-177.574987
D8	-70.440674
D9	-122.004795
D10	122.363105
D11	0.180566
	-179.995330
D12	
D13	-116.416000
D14	123.524546
D15	3.368130
D16	-58.144599
D10 D17	-177.922986
D18	61.914645
D19	0.008921
D20	121.933831
D21	-122.013128

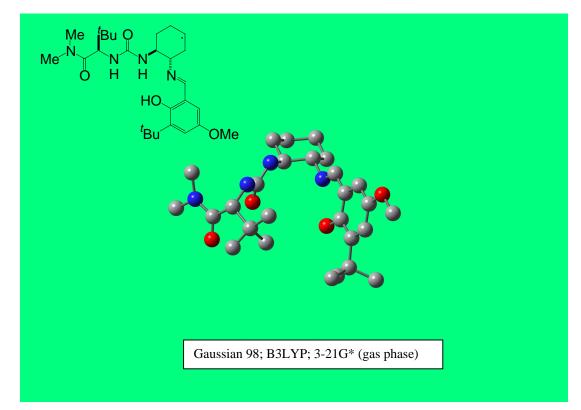
D22	-0.032016
1 2 1.0 2 2 3 2.0 3 15 1.0 4 5 1.0 5 6 1.5 2 6 7 1.5 9 7 8 1.0 2 8 9 10 19 1.0	12 1.0 13 1.0 14 1.0 22 1.0 11 1.0 9 1.0
17 18 19 20 21 22 23 1.0 23 24	0 24 1.0 25 1.0
25	

Selecte	ed example	e of the result	ting Z-matrix	for imine-	Strecker adduct complex (2-
methy	lamino pro	opionitrile and	d N.N'-dimet	hylthiourea	a; Figure 2b in the main text):
0 1		I · · · · ·		J	., 8
С					
Н	1	B1			
Ν	1	B2 2	A1		
С	3	B3 1	A2 2	D1	
Ν	4	B4 3	A3 1	D2	
Η	5	B5 4	A4 3	D3	
S	4	B6 3	A5 1	D4	
С	5	B7 4	A6 3	D5	
С	3	B8 1	A7 2	D6	
Η	1	B9 2	A8 3	D7	
Η	1	B10 2	A9 3	D8	
Η	1	B11 2	A10 3	D9	
Н	9	B12 3	A11 1	D10	
Η	9	B13 3	A12 1	D11	
Η	9	B14 3	A13 1	D12	
Η	8	B15 5	A14 4	D13	
Η	8	B16 5	A15 4	D14	
Η	8	B17 5	A16 4	D15	
С	1	B18 2	A17 3	D16	
Н	19	B19 1	A18 2	D17	
Н	19	B20 1	A19 2	D18	
Η	19	B21 1	A20 2	D19	
С	19	B22 1	A21 2	D20	
Н	23	B23 19	A22 1	D21	

N H C N	1 25 23 27	B24 2 B25 1 B26 19 B27 23	A23 3 A24 2 A25 1 A26 19	D22 D23 D24 D25
	21	D27 23	A20 17	D25
B1	3.01522			
B2	3.98258			
В3 В4	1.35874 1.38005			
Б4 В5	1.01209			
B6	1.68141			
B7	1.45701			
B8	1.45236			
B9	1.09233			
B10	1.0925	83		
B11	1.0992			
B12	1.0923			
B13	1.0930			
B14	1.0965			
B15	1.0932			
B16 B17	1.0965 1.0906			
B17 B18	3.0737			
B10 B19	1.0921			
B20	1.0924			
B21	1.0923			
B22	1.5355	76		
B23	1.0962			
B24	1.4721			
B25	1.0182			
B26	1.4847			
B27	1.1614			
A1 A2	5.51292 131.2826			
A2 A3	113.1972			
A4	114.4556			
A5	123.8717			
A6	122.0645	513		
A7	101.1184	426		
A8	83.6506			
A9	89.8531			
A10	152.593			
A11	110.366			
A12 A13	108.053 112.276			
A13 A14	108.047			
A15	112.514			
A16	109.760			
A17	62.7803			
A18	96.4049	969		
A19	154.142	282		
A20	68.0463			
A21	53.7047			
A22	107.803			
A23	39.641			
A24	107.966	420		

A25 110.905524 A26 177.230410 D1 35.971619 D2 -28.271400 D3 -25.964968 D4 151.056784 D5 -172.277881 D6 -168.998991 D7 18.902969 D8 -88.919579 D9 137.675719 D10 146.593476 D11 27.194684 D12 -93.495556 D13 -177.415610 D14 61.540940 D15 -58.412203 D16 85.971445 D17 -29.522778 D18 136.293714 D19 -137.815655 D20 79.665522 D21 -145.725867 D22 151.264711 D23 118.891813 D24 97.026468 D25 -132.139643
$\begin{array}{c} 1 \ 10 \ 1.0 \ 11 \ 1.0 \ 12 \ 1.0 \ 25 \ 1.0 \\ 2 \ 3 \ 1.0 \\ 3 \ 4 \ 1.5 \ 9 \ 1.0 \\ 4 \ 5 \ 1.5 \ 7 \ 1.0 \\ 5 \ 6 \ 1.0 \ 8 \ 1.0 \\ 6 \\ 7 \\ 8 \ 16 \ 1.0 \ 17 \ 1.0 \ 18 \ 1.0 \\ 9 \ 13 \ 1.0 \ 14 \ 1.0 \ 15 \ 1.0 \\ 10 \\ 11 \\ 12 \\ 13 \\ 14 \\ 15 \\ 16 \\ 17 \\ 18 \\ 19 \ 20 \ 1.0 \ 21 \ 1.0 \ 22 \ 1.0 \ 23 \ 1.0 \\ 20 \\ 21 \\ 22 \\ 23 \ 24 \ 1.0 \ 25 \ 1.0 \ 27 \ 1.0 \\ 24 \\ 25 \ 26 \ 1.0 \\ 26 \\ 27 \ 28 \ 3.0 \\ 28 \end{array}$

Optimization to the energy minimum of a catalyst **6** derivative using Gaussian 98; method B3LYP; basis set 3-21G* (gas phase) provided the same structure that was observed by NMR spectroscopy in solution (Figure 1a of the text). Following is the figure (all hydrogens omitted for clarity) of the calculated structure and the resulting Z-matrix (including hydrogens).



Symbolic Z-1 0 1	matr	ix:					
N							
С	1	B1					
0	2	B2	1	A1			
С	2	B3	1	A2	3	D1	0
С	4	B 4	2	A3	1	D2	0
Ν	4	B5	2	A4	1	D3	0
С	5	B6	4	A5	2	D4	0
С	5	B7	4	A6		D5	0
С	5	B8	4	A7		D6	0
С	6	B9	4	A8	2	D7	0
0	10	B10	6		4	D8	0
Ν	10	B11	6		4	D9	0
С	12	B12	10		6) 0
Н	6	B13	4	A12	2	D11	0
Н	12	B14	1				
С	13	B15	12			0 D1	
С	13	B16	12			0 D1	
С	16	B17	1.			2 D1	
С	13	B18	12			0 D1	6 0
С	16	B19	1.			2 D1	
Ν	20	B20	1	6 A19	1	3 D1	8 0

С	21 B21	20 A20	16 D19 0
C	21 B21 22 B22	20 A20 21 A21	10 D19 0 20 D20 0
C	22 B22 23 B23	21 A21 22 A22	20 D20 0 21 D21 0
C	23 B23 24 B24	22 A22 23 A23	
C			
		24 A24	
C	26 B26	25 A25	24 D24 0
C	27 B27	26 A26	25 D25 0
0	25 B28	24 A27	23 D26 0
0	28 B29	27 A28	26 D27 0
С	27 B30	26 A29	25 D28 0
С	31 B31	27 A30	26 D29 0
С	31 B32	27 A31	26 D30 0
Н	33 B33	31 A32	27 D31 0
Н	4 B34	2 A33	1 D32 0
Н	7 B35	5 A34	4 D33 0
Н	7 B36	5 A35	4 D34 0
Н	7 B37	5 A36	4 D35 0
Н	8 B38	5 A37	4 D36 0
Н	8 B39	5 A38	4 D37 0
Н	8 B40	5 A39	4 D38 0
Н	9 B41	5 A40	4 D39 0
H	9 B42	5 A41	4 D40 0
Н	9 B43	5 A42	4 D41 0
Н	18 B44	16 A43	13 D42 0
H	18 B45	16 A44	13 D42 0 13 D43 0
H	10 D45 19 B46	10 A44 13 A45	13 D43 0 12 D44 0
H	19 B40 19 B47	13 A45 13 A46	12 D44 0 12 D45 0
Н	19 B47 20 B48	15 A40 16 A47	12 D45 0 13 D46 0
H			20 D47 0
H	24 B50	23 A49	22 D48 0
Н	26 B51	25 A50	24 D49 0
H	30 B52	28 A51	27 D50 0
Н	32 B53	31 A52	27 D51 0
Н	32 B54	31 A53	27 D52 0
H	33 B55	31 A54	27 D53 0
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С	31 B58	27 A57	26 D56 0
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Н	17 B64	13 A63	12 D62 0
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Н	68 B69	1 A68	2 D67 0
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C	1 B71	2 A70	3 D69 0
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H	72 B72 72 B73	1 A72	2 D70 0 2 D71 0
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C	29 B75	25 A74	24 D72 0
Н	76 B76	29 A74 29 A75	24 D73 0 25 D74 0
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Н	76	B78	29	A77	25	D76
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B5		.47848				
B6		.54778				
B7		.54895				
B8		.55018				
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B17		.54479				
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B19		.54554				
B20		.47683				
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B28		.3935				
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B48 B49		.10475				
B49 B50		.08328				
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B51 B52		.0804				
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D 33	1					

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B55	1.09279
B56	1.09692
B57	1.0961
B58	1.55391
B59	1.09689
B60	1.09564
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B62	1.094
B63	1.09803
B64	1.09624
B65	1.09681
B66	1.09584
B67	1.47366
B68	1.09641
	1.08928
B69	
B70	1.09786
B71	1.46721
B72	1.09937
B73	1.08652
B74	1.09772
B75	1.45621
B76	1.09765
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B78	1.09065
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A 4	111 00022
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A4	111.00832
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	113.23394
A5 A6	113.23394 108.2498
A5 A6 A7	113.23394 108.2498 108.23234
A5 A6	113.23394 108.2498 108.23234 121.15722
A5 A6 A7 A8	113.23394 108.2498 108.23234 121.15722
A5 A6 A7 A8 A9	113.23394 108.2498 108.23234 121.15722 123.14424
A5 A6 A7 A8 A9 A10	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773
A5 A6 A7 A8 A9 A10	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773
A5 A6 A7 A8 A9 A10 A11	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522
A5 A6 A7 A8 A9 A10 A11 A12	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609
A5 A6 A7 A8 A9 A10 A11	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522
A5 A6 A7 A8 A9 A10 A11 A12 A13	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246
A5 A6 A7 A8 A9 A10 A11 A12 A13	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971 90.31198
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971 90.31198 112.36749
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971 90.31198
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971 90.31198 112.36749 35.10283
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971 90.31198 112.36749 35.10283 110.03427
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971 90.31198 112.36749 35.10283
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19 A20	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971 90.31198 112.36749 35.10283 110.03427 119.82911
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19 A20 A21	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971 90.31198 112.36749 35.10283 110.03427 119.82911 122.75859
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19 A20 A21 A22	$\begin{array}{c} 113.23394\\ 108.2498\\ 108.23234\\ 121.15722\\ 123.14424\\ 114.12773\\ 119.17522\\ 116.14609\\ 121.48891\\ 146.10246\\ 146.98971\\ 90.31198\\ 112.36749\\ 35.10283\\ 110.03427\\ 119.82911\\ 122.75859\\ 118.95156\end{array}$
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19 A20 A21	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971 90.31198 112.36749 35.10283 110.03427 119.82911 122.75859
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19 A20 A21 A22 A23	$\begin{array}{c} 113.23394\\ 108.2498\\ 108.23234\\ 121.15722\\ 123.14424\\ 114.12773\\ 119.17522\\ 116.14609\\ 121.48891\\ 146.10246\\ 146.98971\\ 90.31198\\ 112.36749\\ 35.10283\\ 110.03427\\ 119.82911\\ 122.75859\\ 118.95156\\ 120.00438\\ \end{array}$
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19 A20 A21 A22 A23 A24	$\begin{array}{c} 113.23394\\ 108.2498\\ 108.23234\\ 121.15722\\ 123.14424\\ 114.12773\\ 119.17522\\ 116.14609\\ 121.48891\\ 146.10246\\ 146.98971\\ 90.31198\\ 112.36749\\ 35.10283\\ 110.03427\\ 119.82911\\ 122.75859\\ 118.95156\\ 120.00438\\ 118.63279 \end{array}$
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19 A20 A21 A22 A23	$\begin{array}{c} 113.23394\\ 108.2498\\ 108.23234\\ 121.15722\\ 123.14424\\ 114.12773\\ 119.17522\\ 116.14609\\ 121.48891\\ 146.10246\\ 146.98971\\ 90.31198\\ 112.36749\\ 35.10283\\ 110.03427\\ 119.82911\\ 122.75859\\ 118.95156\\ 120.00438\\ \end{array}$
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A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19 A20 A21 A22 A23 A24 A25 A26	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971 90.31198 112.36749 35.10283 110.03427 119.82911 122.75859 118.95156 120.00438 118.63279 123.49792 117.77938
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19 A20 A21 A22 A23 A24 A25	$\begin{array}{c} 113.23394\\ 108.2498\\ 108.23234\\ 121.15722\\ 123.14424\\ 114.12773\\ 119.17522\\ 116.14609\\ 121.48891\\ 146.10246\\ 146.98971\\ 90.31198\\ 112.36749\\ 35.10283\\ 110.03427\\ 119.82911\\ 122.75859\\ 118.95156\\ 120.00438\\ 118.63279\\ 123.49792\end{array}$
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19 A20 A21 A22 A23 A24 A25 A26 A27	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971 90.31198 112.36749 35.10283 110.03427 119.82911 122.75859 118.95156 120.00438 118.63279 123.49792 117.77938 116.41702
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19 A20 A21 A22 A23 A24 A25 A26 A27 A28	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971 90.31198 112.36749 35.10283 110.03427 119.82911 122.75859 118.95156 120.00438 118.63279 123.49792 117.77938 116.41702 121.9508
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19 A20 A21 A22 A23 A24 A25 A26 A27 A28 A29	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971 90.31198 112.36749 35.10283 110.03427 119.82911 122.75859 118.95156 120.00438 118.63279 123.49792 117.77938 116.41702 121.9508 117.12898
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19 A20 A21 A22 A23 A24 A25 A26 A27 A28	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971 90.31198 112.36749 35.10283 110.03427 119.82911 122.75859 118.95156 120.00438 118.63279 123.49792 117.77938 116.41702 121.9508
A5 A6 A7 A8 A9 A10 A11 A12 A13 A14 A15 A16 A17 A18 A19 A20 A21 A22 A23 A24 A25 A26 A27 A28 A29	113.23394 108.2498 108.23234 121.15722 123.14424 114.12773 119.17522 116.14609 121.48891 146.10246 146.98971 90.31198 112.36749 35.10283 110.03427 119.82911 122.75859 118.95156 120.00438 118.63279 123.49792 117.77938 116.41702 121.9508 117.12898

A32	111.63737
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A34	108.32793
A35	109.40145
	110 (2105
A36	112.63195
A37	111.32942
A38	109.19633
A39	111.19055
A40	110.77232
A41	109.45774
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A43	109.24995
A44	109.6469
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A45	109.32544
-	
A46	108.55464
A47	109.06709
A48	121.22349
A49	120 00212
-	120.99213
A50	118.65008
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A54	111.12335
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A56	109.09485
A57	108.38429
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A67	110.01124
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A69	110.48987
A70	127.17575
A71	110.77438
A72	110.32411
A73	109.28869
A74	117.90664
A75	111.90703
A76	111.90925
A77	104.98726
D1	-174.97216
D2	
D_{2}	172.92094
D3	-57.19889
D3 D4	-57.19889 59.89443
D3 D4 D5	-57.19889 59.89443 -179.02921
D3 D4	-57.19889 59.89443
D3 D4 D5 D6	-57.19889 59.89443 -179.02921 -62.01234
D3 D4 D5 D6 D7	-57.19889 59.89443 -179.02921 -62.01234 -43.70959
D3 D4 D5 D6 D7	-57.19889 59.89443 -179.02921 -62.01234 -43.70959
D3 D4 D5 D6 D7 D8	-57.19889 59.89443 -179.02921 -62.01234 -43.70959 3.33987
D3 D4 D5 D6 D7	-57.19889 59.89443 -179.02921 -62.01234 -43.70959
D3 D4 D5 D6 D7 D8	-57.19889 59.89443 -179.02921 -62.01234 -43.70959 3.33987

D11	158.16273
D12	-6.73402
D13	-115.98529
D14	112.87055
D15	-134.24379
D16	115.71943
D17	-7.16314
D18	120.09487
D19	114.57599
D20	179.88701
D21	178.35154
D22	-179.67675
D23	1.10892
D24	-0.48953
D25	-1.28705
D26	-179.23457
D27	-177.33593
D28	-179.95149
D29	-50.84911
D30	-170.8978
D31	-62.63932
D32	56.80967
D33	-173.62152
D34	-54.23142
D35	67.57077
D36	56.97602
D37	176.23957
D38	-65.31227
D39	-56.82477
D40	-175.9341
D41	64.0879
D42	93.0285
D43	-149.37133
D44	61.94659
D45	-55.59575
D46	-119.5129
D47	-0.38606
D48	0.31102
D49	179.53152
D50	-176.48374
D51	61.13493
D52	-60.08847
D53	57.84253
D54	177.09801
D55	-179.2681
D56	68.2336
D57	-177.92236
D58	-57.68025
D59	62.70808
D60	-3.05731
D61	-117.3486
D62	6.99653
	115.55696
D63	
D64	-6.28774
D65	-4.32683
D66	-122.99338

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170.83295
-107.78505
14.62921
133.33393
-179.81204
-61.34278
61.34813
-179.99249

Z-matrix for the catalyst-imine complex (includ	ing hydrogens); Figure 2b-c of the main
text:	

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С						
Ν	1	1.455100				
С	2	1.307659	1	120.326986		
0	3	1.217380	2	124.951408	1	0.150436
С	3	1.537574	2	113.113208	1	-179.982473
С	5	1.547259	3	111.304327	2	121.863903
Ν	5	1.464335	3	110.818912	2	-115.964654
С	6	1.537108	5	110.094633	3	61.342520
С	6	1.530865	5	109.663057	3	-178.541730
С	6	1.533359	5	109.590417	3	-58.870713
С	7	1.315610	5	120.314565	3	-10.053245
0	11	1.215877	7	120.721550	5	-0.005474
Ν	11	1.319129	7	119.034418	5	-179.990229
С	13	1.466249	11	122.751257	7	179.840369
Н	7	1.011999	5	119.597747	3	169.883662
Ν	13	2.904829	11	100.906104	7	-0.019423
С	14	2.503556	13	144.236575	11	-125.366467
С	14	2.493638	13	144.759766	11	115.671826
С	18	1.531964	14	90.072562	13	124.005363
С	14	1.529160	13	109.292019	11	115.369939
С	14	1.529334	13	109.066819	11	-124.952615
С	1	1.436670	2	112.630728	3	176.434088
С	22	1.405807	1	120.034799	2	179.997704
С	23	1.404255	22	119.919026	1	179.993126
С	24	1.404425	23	120.015986	22	-0.009202
С	25	1.404770	24	120.046423	23	-0.081602
С	22	1.403141	1	119.944134	2	-0.069812
Ν	21	1.431013	14	108.705660	13	60.147439
С	28	1.339587	21	119.989678	14	-119.783391
С	29	1.320516	28	122.036228	21	179.957090
С	30	1.404100	29	120.429102	28	179.974290
С	31	1.405305	30	119.940692	29	179.950946
С	32	1.404137	31	120.067753	30	0.112073
С	33	1.403600	32	119.956263	31	-0.108519
С	34	1.404311	33	120.046419	32	0.101346
0	32	1.392087	31	119.942920	30	179.960959
0	35	1.391273	34	120.038204	33	179.995640
С	36	1.315207	32	106.044677	31	-179.949661
0	38	1.207858	36	119.918747	32	0.134995
С	38	1.448154	36	118.521426	32	-179.912822
С	40	1.544513	38	107.343896	36	-179.762586

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$\begin{array}{cccccccccccccccccccccccccccccccccccc$					
H 13 1.013096 11 118.574677 7 -0.038520 C 16 1.471633 13 117.445992 11 -101.842503 C 48 1.530919 16 119.998372 13 154.970391 C 51 2.771308 16 153.003071 13 138.431560 C 51 1.387154 48 119.979739 16 -0.081275 C 51 1.387154 48 119.957641 16 179.957536 C 53 1.385803 51 59.990805 54 0.017075 C 50 1.531473 16 109.436884 13 25.604244 C 58 1.332800 50 119.960560 16 -179.702030 H 1 B1 2 A1 3 D1 H 8 B3 6 A3 5 D3 H 8 B4 6 A4 5 D4 H 8 B5 6 A5 D6 <td></td> <td></td> <td></td> <td></td> <td></td>					
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Н	57	B58 53	A58 51	D58
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D59	0.297970
D60	-0.000001
D61	180.000000

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⁽⁵⁾ The procedure for the thiourea formation is based on: Wenzel, A.; Jacobsen, E.N. *Unpublished results*.

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