

Supporting Information for
Reactions of Methane Monooxygenase Intermediate Q with Derivatized Methanes

Edna A. Ambundo,[†] Richard A. Friesner,[‡] and Stephen J. Lippard^{*,†}

Department of Chemistry, Massachusetts Institute of Technology, Cambridge, Massachusetts 02139 and Department of Chemistry and Center for Biomolecular Simulation, Columbia University, New York, New York 10027

Received April 17, 2002

List of Tables

Table S1. Activation Parameters for the Decay of Q in the Presence of Substrates

List of Figures

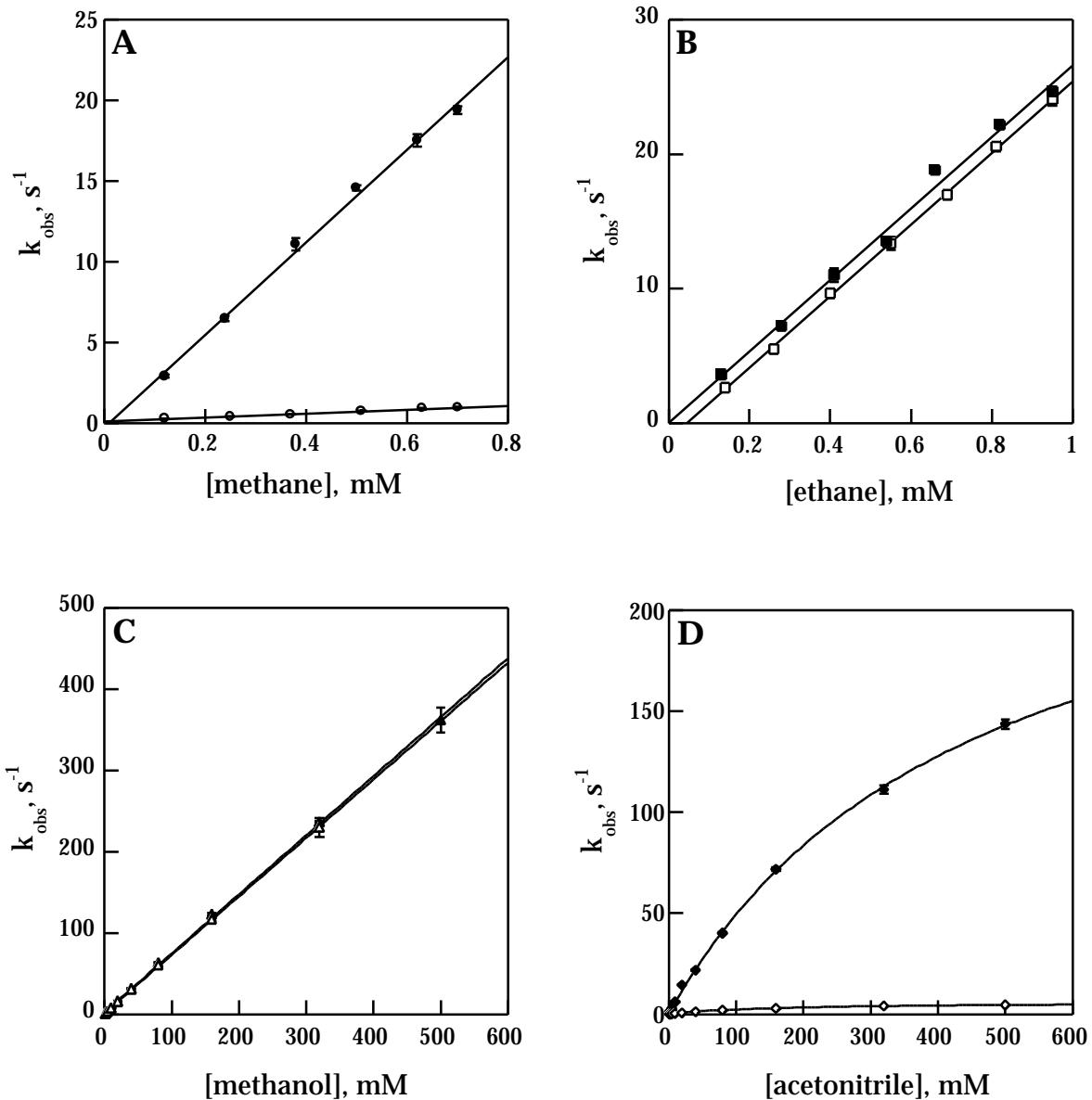
Figure S1. Plot of k_{obs} versus [substrate] for the following substrates: (A) methane, CH₄ (●), CD₄ (○), (B) ethane, C₂H₆ (■), C₂D₆(□), (C) methanol, CH₃OH (▲), CD₃OH(△) and (D) acetonitrile, CH₃CN (◆), CD₃CN (◇).

Figure S2. Eyring plots for the decay of Q in the presence of substrates. Substrates tested were CH₄ (●), CH₃CN (■), CD₃CN (□), CH₃OH (▲), CD₃OH (△), CH₃NO₂ (◆), CD₃NO₂ (◇) and C₂H₆ (×).

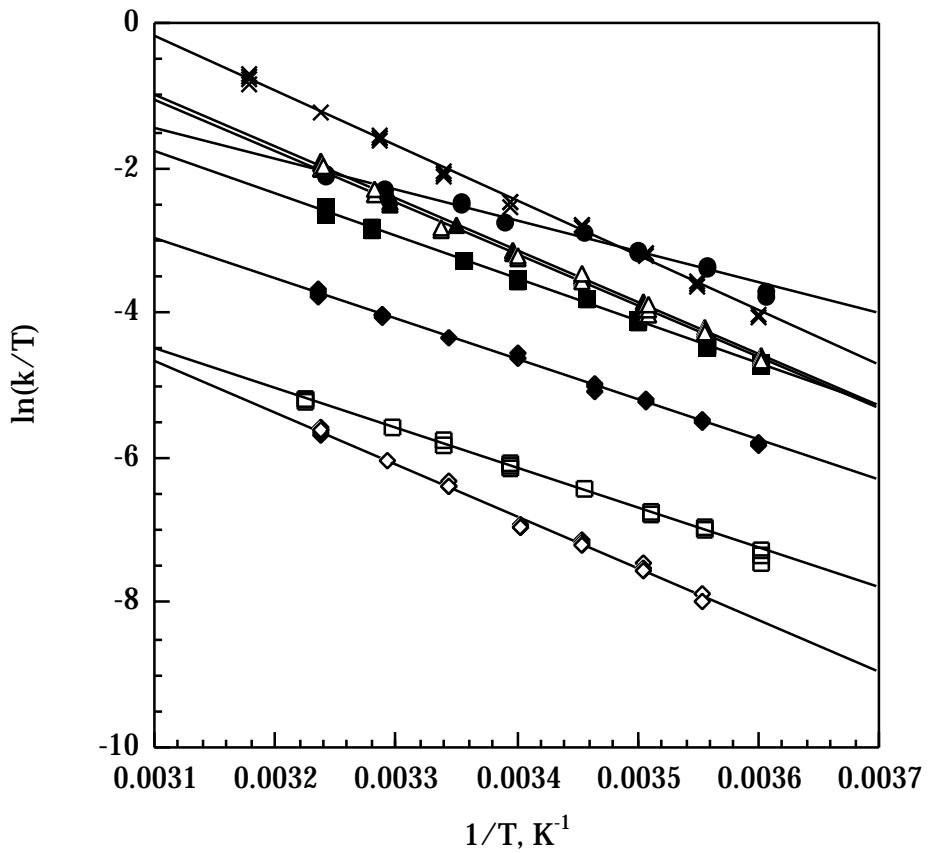
Table S1. Activation Parameters for the Decay of Q in the Presence of Substrates

Substrates	H^\ddagger , kcal/mol	S^\ddagger , cal/mol·K
CH ₄ – 0.7 mM	8.5 ± 0.3	-24 ± 1
C ₂ H ₆ – 0.95 mM	15.1 ± 0.2	-0.9 ± 0.5
C ₂ D ₆ – 0.95 mM	13.3 ± 0.1	-7.3 ± 0.5
CH ₃ CN – 15.3 mM ^a	11.7 ± 0.1	-14.5 ± 0.4
CD ₃ CN – 15.3 mM ^a	11.1 ± 0.2	-21.8 ± 0.6
CH ₃ NO ₂ – 20.0 mM ^a	11.2 ± 0.1	-18.5 ± 0.5
CD ₃ NO ₂ – 20.0 mM ^a	14.3 ± 0.2	-12.2 ± 0.8
CH ₃ OH – 15.1 mM	14.3 ± 0.2	-4.8 ± 0.6
CD ₃ OH – 15.3 mM	14.1 ± 0.2	-5.6 ± 0.7

^aReactions with these substrates exhibit saturation kinetics.



Ambundo, Friesner and Lippard, Figure S1.



Ambundo, Friesner and Lippard, Figure S2.