

Table 13. Bond Angles [deg] for **7c**•SnCl₄.

bond angle	[deg]	bond angle	[deg]
C(19)-C(18)-H(18A)	111.2	H(14A)-C(14)-H(14B)	109.5
N(5)-C(18)-H(18A)	111.2	N(4)-C(14)-H(14C)	109.5
C(17)-C(18)-H(18A)	111.2	H(14A)-C(14)-H(14C)	109.5
N(6)-C(19)-C(18)	108.1(11)	H(14B)-C(14)-H(14C)	109.5
N(6)-C(19)-C(20)	102.2(12)	C(13)-N(4)-C(14)	115.1(11)
C(18)-C(19)-C(20)	117.2(14)	C(13)-N(4)-P(2)	125.7(10)
N(6)-C(19)-H(19A)	109.6	C(14)-N(4)-P(2)	119.1(9)
C(18)-C(19)-H(19A)	109.6	N(4)-C(13)-C(12)	113.0(12)
C(20)-C(19)-H(19A)	109.6	N(4)-C(13)-H(13A)	109.0
C(21)-C(20)-C(19)	103.1(14)	C(12)-C(13)-H(13A)	109.0
C(21)-C(20)-H(20A)	111.1	N(4)-C(13)-H(13B)	109.0
C(19)-C(20)-H(20A)	111.1	C(12)-C(13)-H(13B)	109.0
C(21)-C(20)-H(20B)	111.1	H(13A)-C(13)-H(13B)	107.8
C(19)-C(20)-H(20B)	111.1	C(11)-C(12)-C(13)	112.6(16)
H(20A)-C(20)-H(20B)	109.1	C(11)-C(12)-H(12A)	109.1
C(20)-C(21)-C(22)	109.3(16)	C(13)-C(12)-H(12A)	109.1
C(20)-C(21)-H(21A)	109.8	C(11)-C(12)-H(12B)	109.1
C(22)-C(21)-H(21A)	109.8	C(13)-C(12)-H(12B)	109.1
C(20)-C(21)-H(21B)	109.8	H(12A)-C(12)-H(12B)	107.8
C(22)-C(21)-H(21B)	109.8	Cl(7)-C(23)-Cl(5)	108.2(9)
H(21A)-C(21)-H(21B)	108.3	Cl(7)-C(23)-Cl(6)	108.4(9)
N(6)-C(22)-C(21)	100.0(13)	Cl(5)-C(23)-Cl(6)	108.1(9)
N(6)-C(22)-H(22A)	111.8	Cl(7)-C(23)-H(23)	110.7
C(21)-C(22)-H(22A)	111.8	Cl(5)-C(23)-H(23)	110.7
N(6)-C(22)-H(22B)	111.8	Cl(6)-C(23)-H(23)	110.7
C(21)-C(22)-H(22B)	111.8	Cl(10)-C(24)-Cl(9)	108.3(9)
H(22A)-C(22)-H(22B)	109.5	Cl(10)-C(24)-Cl(8)	108.2(9)
C(22)-N(6)-C(19)	115.7(12)	Cl(9)-C(24)-Cl(8)	108.9(9)
C(22)-N(6)-P(2)	130.8(11)	Cl(10)-C(24)-H(24)	110.5
C(19)-N(6)-P(2)	109.0(9)	Cl(9)-C(24)-H(24)	110.5
N(4)-C(14)-H(14A)	109.5	Cl(8)-C(24)-H(24)	110.5
N(4)-C(14)-H(14B)	109.5	Cl(12)-C(25)-Cl(11)	108.7(9)

Table 13. Bond Angles [deg] for **7c**•SnCl₄.

bond angle	[deg]	bond angle	[deg]
Cl(12)-C(25)-Cl(13)	109.1(9)	Cl(14)-C(26)-Cl(15)	107.1(8)
Cl(11)-C(25)-Cl(13)	108.2(8)	Cl(16)-C(26)-Cl(15)	107.1(9)
Cl(12)-C(25)-H(25)	110.3	Cl(14)-C(26)-H(26)	111.6
Cl(11)-C(25)-H(25)	110.3	Cl(16)-C(26)-H(26)	111.6
Cl(13)-C(25)-H(25)	110.3	Cl(15)-C(26)-H(26)	111.6
Cl(14)-C(26)-Cl(16)	107.5(9)		

Table 14. Torsional angles [deg] for **7c**•SnCl₄.

torsional angle	[deg]	torsional angle	[deg]
O(2)-Sn(1)-O(1)-P(1)	127.0(16)	C(7)-C(8)-N(2)-P(1)	-127.7(16)
Cl(4)-Sn(1)-O(1)-P(1)	-49.3(15)	C(4)-C(5)-N(2)-C(8)	-127.3(17)
Cl(2)-Sn(1)-O(1)-P(1)	42.0(15)	C(6)-C(5)-N(2)-C(8)	-3.5(18)
Cl(1)-Sn(1)-O(1)-P(1)	-141.6(15)	C(4)-C(5)-N(2)-P(1)	27.5(16)
Sn(1)-O(1)-P(1)-N(3)	-109.9(15)	C(6)-C(5)-N(2)-P(1)	151.3(12)
Sn(1)-O(1)-P(1)-N(1)	127.2(15)	O(1)-P(1)-N(2)-C(8)	-109.6(15)
Sn(1)-O(1)-P(1)-N(2)	17.0(18)	N(3)-P(1)-N(2)-C(8)	15.8(17)
O(1)-P(1)-N(1)-C(1)	13.7(14)	N(1)-P(1)-N(2)-C(8)	133.0(15)
N(3)-P(1)-N(1)-C(1)	-107.0(13)	O(1)-P(1)-N(2)-C(5)	100.5(11)
N(2)-P(1)-N(1)-C(1)	136.3(12)	N(3)-P(1)-N(2)-C(5)	-134.1(11)
O(1)-P(1)-N(1)-C(4)	-121.5(11)	N(1)-P(1)-N(2)-C(5)	-16.9(11)
N(3)-P(1)-N(1)-C(4)	117.8(11)	O(1)-P(1)-N(3)-C(10)	19.4(16)
N(2)-P(1)-N(1)-C(4)	1.1(12)	N(1)-P(1)-N(3)-C(10)	141.3(14)
C(4)-N(1)-C(1)-C(2)	26.6(16)	N(2)-P(1)-N(3)-C(10)	-110.7(14)
P(1)-N(1)-C(1)-C(2)	-108.8(13)	O(1)-P(1)-N(3)-C(9)	-163.5(15)
N(1)-C(1)-C(2)-C(3)	-36.8(17)	N(1)-P(1)-N(3)-C(9)	-41.7(18)
C(1)-C(2)-C(3)-C(4)	32.3(18)	N(2)-P(1)-N(3)-C(9)	66.3(17)
C(1)-N(1)-C(4)-C(5)	-125.4(14)	C(9)-N(3)-C(10)-C(11)	70(2)
P(1)-N(1)-C(4)-C(5)	14.8(16)	P(1)-N(3)-C(10)-C(11)	-112.7(17)
C(1)-N(1)-C(4)-C(3)	-7.6(16)	N(3)-C(10)-C(11)-C(12)	39(2)
P(1)-N(1)-C(4)-C(3)	132.6(12)	O(1)-Sn(1)-O(2)-P(2)	122.9(18)
C(2)-C(3)-C(4)-N(1)	-14.9(17)	Cl(2)-Sn(1)-O(2)-P(2)	-146.6(17)
C(2)-C(3)-C(4)-C(5)	99.4(16)	Cl(3)-Sn(1)-O(2)-P(2)	-53.3(17)
N(1)-C(4)-C(5)-N(2)	-26.0(17)	Cl(1)-Sn(1)-O(2)-P(2)	37.0(17)
C(3)-C(4)-C(5)-N(2)	-138.1(13)	Sn(1)-O(2)-P(2)-N(4)	-105.5(17)
N(1)-C(4)-C(5)-C(6)	-141.1(14)	Sn(1)-O(2)-P(2)-N(5)	131.2(17)
C(3)-C(4)-C(5)-C(6)	106.8(16)	Sn(1)-O(2)-P(2)-N(6)	20(2)
N(2)-C(5)-C(6)-C(7)	-17.1(18)	O(2)-P(2)-N(5)-C(15)	12.9(14)
C(4)-C(5)-C(6)-C(7)	101(2)	N(4)-P(2)-N(5)-C(15)	-106.7(13)
C(5)-C(6)-C(7)-C(8)	31(2)	N(6)-P(2)-N(5)-C(15)	135.8(13)
C(6)-C(7)-C(8)-N(2)	-32(2)	O(2)-P(2)-N(5)-C(18)	-120.1(11)
C(7)-C(8)-N(2)-C(5)	20.8(19)	N(4)-P(2)-N(5)-C(18)	120.2(11)

Table 14. Torsional angles [deg] for **7c**•SnCl₄.

torsional angle	[deg]	torsional angle	[deg]
N(6)-P(2)-N(5)-C(18)	2.7(12)	C(18)-C(19)-N(6)-C(22)	-132.4(17)
C(18)-N(5)-C(15)-C(16)	26.9(18)	C(20)-C(19)-N(6)-C(22)	-8(2)
P(2)-N(5)-C(15)-C(16)	-106.5(14)	C(18)-C(19)-N(6)-P(2)	26.6(16)
N(5)-C(15)-C(16)-C(17)	-38.6(19)	C(20)-C(19)-N(6)-P(2)	150.8(12)
C(15)-C(16)-C(17)-C(18)	35(2)	O(2)-P(2)-N(6)-C(22)	-104.1(18)
C(15)-N(5)-C(18)-C(19)	-125.3(15)	N(4)-P(2)-N(6)-C(22)	19.0(19)
P(2)-N(5)-C(18)-C(19)	12.5(16)	N(5)-P(2)-N(6)-C(22)	137.5(18)
C(15)-N(5)-C(18)-C(17)	-6.5(17)	O(2)-P(2)-N(6)-C(19)	101.1(11)
P(2)-N(5)-C(18)-C(17)	131.4(12)	N(4)-P(2)-N(6)-C(19)	-135.7(10)
C(16)-C(17)-C(18)-C(19)	98.2(18)	N(5)-P(2)-N(6)-C(19)	-17.3(12)
C(16)-C(17)-C(18)-N(5)	-16.9(18)	O(2)-P(2)-N(4)-C(13)	14.5(16)
N(5)-C(18)-C(19)-N(6)	-24.1(17)	N(5)-P(2)-N(4)-C(13)	136.7(14)
C(17)-C(18)-C(19)-N(6)	-137.0(13)	N(6)-P(2)-N(4)-C(13)	-114.3(14)
N(5)-C(18)-C(19)-C(20)	-138.9(14)	O(2)-P(2)-N(4)-C(14)	-169.0(15)
C(17)-C(18)-C(19)-C(20)	108.2(16)	N(5)-P(2)-N(4)-C(14)	-46.8(17)
N(6)-C(19)-C(20)-C(21)	23.9(19)	N(6)-P(2)-N(4)-C(14)	62.2(17)
C(18)-C(19)-C(20)-C(21)	141.8(18)	C(14)-N(4)-C(13)-C(12)	63(2)
C(19)-C(20)-C(21)-C(22)	-32(2)	P(2)-N(4)-C(13)-C(12)	-120.3(15)
C(20)-C(21)-C(22)-N(6)	27(2)	C(10)-C(11)-C(12)-C(13)	68(2)
C(21)-C(22)-N(6)-C(19)	-10(2)	N(4)-C(13)-C(12)-C(11)	53(2)
C(21)-C(22)-N(6)-P(2)	-163.5(16)		

Symmetry transformations used to generate equivalent atoms:

Crystal Preparation and Structure Refinement Summary for 7d•SnCl₄

To a solution of **7d** (50 mg, 0.10 mmol) in 2.0 mL of CH₂Cl₂ at rt under nitrogen was added SnCl₄ (12 mL, 0.10 mmol, 1.0 equiv) with stirring and the mixture was filtered through glass wool to obtain a clear solution. The solution was then added to a small test tube (15 X 50 mm). The test tube was placed in a vessel (40 X 80 mm) which contained 2.0 mL of pentane. Slow diffusion of pentane at rt provided the single crystals suitable for X-ray analysis.

The data crystal was mounted using oil (Paratone-N, Exxon) to a thin glass fiber with the (0 0 1) scattering planes roughly normal to the spindles axis. Systematic conditions suggested the unambiguous space group. Structure was solved by direct methods (Sheldrick, 1998). Two-fold symmetry along the b-axis was imposed on the host molecule. A Fourier map less contributions from the host molecule revealed a water solvate disordered in five positions about the b-axis two-fold. Anisotropic displacement parameters for the disordered solvate atom positions were restrained to have similar symmetric values with an effective standard deviation of 0.02 Å. No other restrains were imposed on the model. H atoms for the water solvate never surfaced and were not included in the proposed model. Methyl H atom positions, R-CH₃, were optimized by rotation about R-C bonds with idealized C-H, R-H and H-H distances. Remaining H atoms for the host molecule were included as fixed idealized contributors. H atom U's were assigned as 1.2 times U_{eq} of adjacent non-H atoms. The space group choice was confirmed by successful convergence of the full-matrix least-squares refinement on F² (Sheldrick, 1998). The highest peaks in the final difference Fourier map were in the vicinity of Sn and Cl atoms; the final map had no other significant features. A final analysis of variance between observed and calculated structure factors showed no dependence on amplitude or resolution.

All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Sheldrick, G. M. SHELX-97-2. **1998**, Program for crystal structure solution and refinement. Institute für anorg. Chemie, Göttingen, Germany.

Bruker AXS, Inc. **1998**, Madison, Wisconsin, USA.

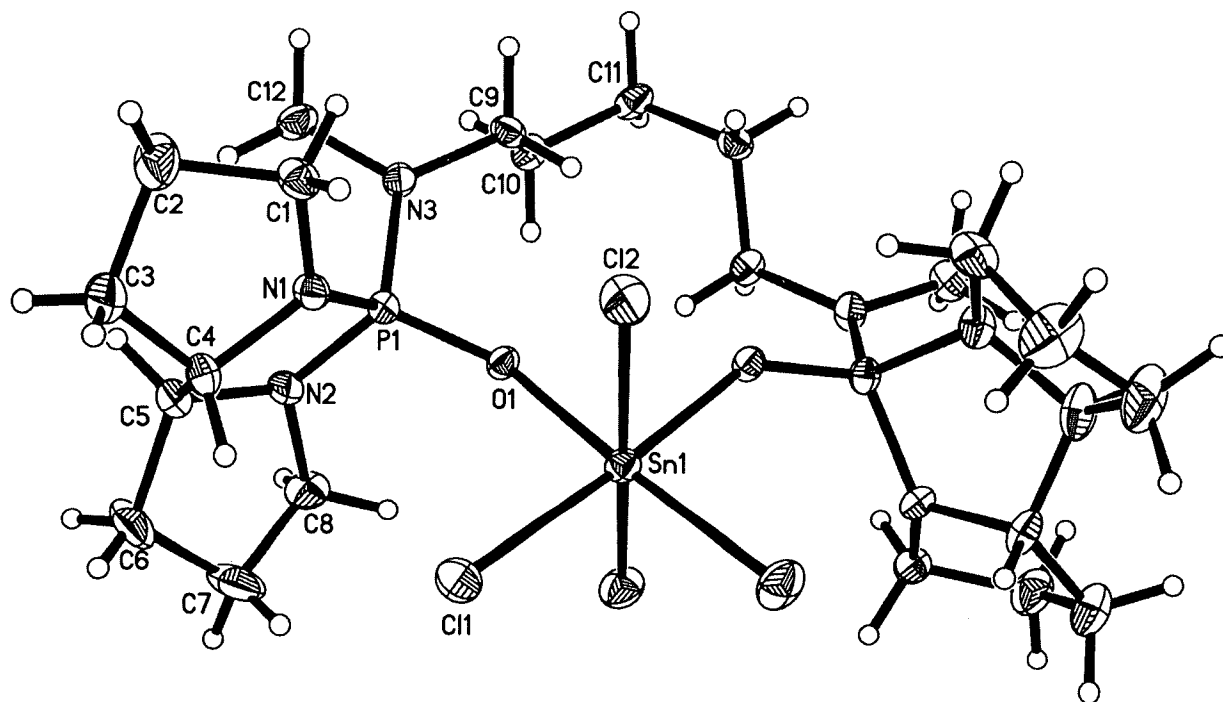


Figure 3. ORTEP representation 7d·SnCl₄ (35% ellipsoids)

Table 15. Crystal data and structure refinement for **7d**•SnCl₄.

Empirical formula	C23 H46 Cl4 N6 O3 P2 Sn	
Formula weight	777.09	
Temperature	193(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	C 2 2 21	
Unit cell dimensions	a = 14.898(3) Å	_alpha = 90°.
	b = 18.010(3) Å	_beta = 90°.
	c = 12.420(2) Å	_gamma = 90°.
Volume	3332.6(10) Å ³	
Z	4	
Density (calculated)	1.549 Mg/m ³	
Absorption coefficient	1.218 mm ⁻¹	
F(000)	1592	
Crystal size	0.26 x 0.06 x 0.03 mm ³	
Theta range for data collection	1.77 to 28.28°.	
Index ranges	-19<=h<=18, -23<=k<=23, -16<=l<=16	
Reflections collected	15641	
Independent reflections	4023 [R(int) = 0.0806]	
Completeness to theta = 28.28°	98.6 %	
Absorption correction	Integration	
Max. and min. transmission	0.9680 and 0.8267	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	4023 / 25 / 200	
Goodness-of-fit on F ²	0.920	
Final R indices [I>2sigma(I)]	R1 = 0.0398, wR2 = 0.0612	
R indices (all data)	R1 = 0.0766, wR2 = 0.0679	
Absolute structure parameter	0.00(3)	
Largest diff. peak and hole	0.437 and -0.420 e.Å ⁻³	

Table 16. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$). U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	U(eq)
Sn(1)	5000	3206(1)	2500	23(1)
Cl(1)	3862(1)	2283(1)	2218(1)	41(1)
Cl(2)	4698(1)	3316(1)	4380(1)	38(1)
P(1)	3102(1)	4256(1)	2453(1)	22(1)
O(1)	4072(2)	4066(1)	2255(2)	23(1)
N(1)	2517(2)	3743(2)	3269(3)	24(1)
N(2)	2458(2)	4118(2)	1400(3)	29(1)
N(3)	3095(2)	5110(2)	2874(3)	26(1)
C(1)	2292(3)	3875(3)	4388(3)	31(1)
C(2)	1280(3)	3808(3)	4410(4)	46(1)
C(3)	1068(3)	3232(3)	3565(3)	43(1)
C(4)	1824(3)	3289(2)	2720(3)	33(1)
C(5)	1621(3)	3707(3)	1682(4)	40(1)
C(6)	1441(3)	3226(4)	680(4)	57(2)
C(7)	2304(4)	3206(3)	80(4)	57(2)
C(8)	2760(3)	3934(3)	310(3)	36(1)
C(9)	3901(3)	5541(2)	3121(3)	25(1)
C(10)	4174(3)	6063(2)	2218(3)	27(1)
C(11)	5000	6526(3)	2500	32(1)
C(12)	2239(3)	5523(2)	2833(3)	32(1)
O(2)	500(20)	5170(20)	540(30)	170(17)
O(3)	70(30)	5664(17)	1830(30)	139(9)
O(4)	0	5350(30)	2500	118(11)

Table 17. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **7d**•SnCl₄. The anisotropic displacement factor exponent takes the form: $-2p^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$.

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
Sn(1)	24(1)	18(1)	27(1)	0	3(1)	0
Cl(1)	38(1)	27(1)	60(1)	-6(1)	2(1)	-8(1)
Cl(2)	40(1)	48(1)	27(1)	3(1)	6(1)	-7(1)
P(1)	20(1)	22(1)	25(1)	-4(1)	-1(1)	0(1)
O(1)	19(2)	18(1)	32(2)	-1(1)	3(1)	1(1)
N(1)	25(2)	23(2)	23(2)	-2(2)	4(2)	-6(2)
N(2)	27(2)	36(2)	23(2)	-8(2)	-1(2)	-2(2)
N(3)	22(2)	18(2)	38(2)	-4(2)	-2(2)	2(2)
C(1)	34(3)	35(3)	23(2)	0(2)	-1(2)	-7(2)
C(2)	36(3)	62(4)	39(3)	3(3)	8(2)	-10(3)
C(3)	31(3)	53(3)	45(3)	-9(3)	7(2)	-9(3)
C(4)	24(2)	32(2)	43(3)	-7(2)	3(2)	-10(2)
C(5)	27(3)	50(3)	41(3)	-19(3)	-1(2)	-5(3)
C(6)	48(3)	82(4)	40(3)	-25(4)	-4(2)	-28(4)
C(7)	76(4)	52(3)	44(3)	-27(3)	5(3)	-19(4)
C(8)	43(3)	51(3)	15(2)	-3(2)	1(2)	6(3)
C(9)	27(3)	22(2)	26(2)	-6(2)	0(2)	-5(2)
C(10)	29(2)	25(2)	28(3)	4(2)	1(2)	4(2)
C(11)	30(3)	19(3)	46(3)	0	14(4)	0
C(12)	33(3)	26(2)	36(3)	-4(2)	2(2)	9(2)
O(2)	180(20)	150(20)	180(20)	15(17)	1(16)	-1(17)
O(3)	115(12)	159(15)	143(16)	10(12)	-9(15)	21(13)
O(4)	84(13)	125(17)	145(18)	0	3(16)	0

Table 18. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$).

	x	y	z	U(eq)
H(1A)	2487	4375	4619	37
H(1B)	2575	3499	4861	37
H(2A)	1069	3645	5128	55
H(2B)	994	4289	4232	55
H(3A)	477	3334	3231	51
H(3B)	1054	2730	3888	51
H(4)	2067	2784	2552	39
H(5)	1115	4063	1797	47
H(6A)	1255	2719	892	68
H(6B)	962	3449	232	68
H(7A)	2194	3149	-701	69
H(7B)	2679	2786	330	69
H(8A)	3421	3882	283	44
H(8B)	2572	4320	-212	44
H(9A)	4402	5195	3270	30
H(9B)	3793	5835	3782	30
H(10A)	3668	6401	2055	33
H(10B)	4301	5769	1563	33
H(11)	5147	6850	1881	38
H(12A)	2155	5796	3508	38
H(12B)	1741	5173	2736	38
H(12C)	2251	5873	2230	38

Table 19. Bond lengths [Å] for **7d**•SnCl₄.

bond	[Å]	bond	[Å]
Sn(1)-O(1)#1	2.098(2)	C(4)-C(5)	1.524(6)
Sn(1)-O(1)	2.098(2)	C(4)-H(4)	1.0000
Sn(1)-Cl(2)	2.3863(11)	C(5)-C(6)	1.540(6)
Sn(1)-Cl(2)#1	2.3863(11)	C(5)-H(5)	1.0000
Sn(1)-Cl(1)#1	2.4008(11)	C(6)-C(7)	1.485(6)
Sn(1)-Cl(1)	2.4008(11)	C(6)-H(6A)	0.9900
P(1)-O(1)	1.505(3)	C(6)-H(6B)	0.9900
P(1)-N(3)	1.623(3)	C(7)-C(8)	1.506(6)
P(1)-N(1)	1.626(3)	C(7)-H(7A)	0.9900
P(1)-N(2)	1.641(4)	C(7)-H(7B)	0.9900
N(1)-C(1)	1.449(5)	C(8)-H(8A)	0.9900
N(1)-C(4)	1.483(5)	C(8)-H(8B)	0.9900
N(2)-C(8)	1.464(5)	C(9)-C(10)	1.519(5)
N(2)-C(5)	1.492(6)	C(9)-H(9A)	0.9900
N(3)-C(9)	1.463(5)	C(9)-H(9B)	0.9900
N(3)-C(12)	1.478(5)	C(10)-C(11)	1.526(4)
C(1)-C(2)	1.513(6)	C(10)-H(10A)	0.9900
C(1)-H(1A)	0.9900	C(10)-H(10B)	0.9900
C(1)-H(1B)	0.9900	C(11)-C(10)#1	1.526(4)
C(2)-C(3)	1.509(6)	C(11)-H(11)	0.9900
C(2)-H(2A)	0.9900	C(12)-H(12A)	0.9800
C(2)-H(2B)	0.9900	C(12)-H(12B)	0.9800
C(3)-C(4)	1.544(5)	C(12)-H(12C)	0.9800
C(3)-H(3A)	0.9900	O(2)-O(2)#2	1.48(7)
C(3)-H(3B)	0.9900	O(3)-O(3)#3	1.68(8)

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,y,-z+1/2 #2 x,-y+1,-z #3 -x,y,-z+1/2

Table 20. Bond Angles [deg] for **7d**•SnCl₄:

bond angle	[deg]	bond angle	[deg]
O(1)#1-Sn(1)-O(1)	84.87(13)	N(1)-C(1)-H(1A)	111.1
O(1)#1-Sn(1)-Cl(2)	85.48(8)	C(2)-C(1)-H(1A)	111.1
O(1)-Sn(1)-Cl(2)	87.49(8)	N(1)-C(1)-H(1B)	111.1
O(1)#1-Sn(1)-Cl(2)#1	87.49(8)	C(2)-C(1)-H(1B)	111.1
O(1)-Sn(1)-Cl(2)#1	85.48(8)	H(1A)-C(1)-H(1B)	109.0
Cl(2)-Sn(1)-Cl(2)#1	170.48(6)	C(3)-C(2)-C(1)	104.5(4)
O(1)#1-Sn(1)-Cl(1)#1	91.40(7)	C(3)-C(2)-H(2A)	110.8
O(1)-Sn(1)-Cl(1)#1	176.23(7)	C(1)-C(2)-H(2A)	110.8
Cl(2)-Sn(1)-Cl(1)#1	92.75(4)	C(3)-C(2)-H(2B)	110.8
Cl(2)#1-Sn(1)-Cl(1)#1	93.84(4)	C(1)-C(2)-H(2B)	110.8
O(1)#1-Sn(1)-Cl(1)	176.23(7)	H(2A)-C(2)-H(2B)	108.9
O(1)-Sn(1)-Cl(1)	91.40(7)	C(2)-C(3)-C(4)	106.0(4)
Cl(2)-Sn(1)-Cl(1)	93.84(4)	C(2)-C(3)-H(3A)	110.5
Cl(2)#1-Sn(1)-Cl(1)	92.75(4)	C(4)-C(3)-H(3A)	110.5
Cl(1)#1-Sn(1)-Cl(1)	92.33(6)	C(2)-C(3)-H(3B)	110.5
O(1)-P(1)-N(3)	105.91(15)	C(4)-C(3)-H(3B)	110.5
O(1)-P(1)-N(1)	119.16(17)	H(3A)-C(3)-H(3B)	108.7
N(3)-P(1)-N(1)	109.48(18)	N(1)-C(4)-C(5)	104.7(3)
O(1)-P(1)-N(2)	113.41(17)	N(1)-C(4)-C(3)	103.3(3)
N(3)-P(1)-N(2)	113.34(18)	C(5)-C(4)-C(3)	117.6(4)
N(1)-P(1)-N(2)	95.58(18)	N(1)-C(4)-H(4)	110.2
P(1)-O(1)-Sn(1)	140.99(15)	C(5)-C(4)-H(4)	110.2
C(1)-N(1)-C(4)	111.8(3)	C(3)-C(4)-H(4)	110.2
C(1)-N(1)-P(1)	129.0(3)	N(2)-C(5)-C(4)	106.2(3)
C(4)-N(1)-P(1)	113.6(3)	N(2)-C(5)-C(6)	103.6(4)
C(8)-N(2)-C(5)	111.2(4)	C(4)-C(5)-C(6)	116.1(4)
C(8)-N(2)-P(1)	126.2(3)	N(2)-C(5)-H(5)	110.2
C(5)-N(2)-P(1)	112.1(3)	C(4)-C(5)-H(5)	110.2
C(9)-N(3)-C(12)	116.6(3)	C(6)-C(5)-H(5)	110.2
C(9)-N(3)-P(1)	124.4(3)	C(7)-C(6)-C(5)	105.6(4)
C(12)-N(3)-P(1)	118.1(3)	C(7)-C(6)-H(6A)	110.6
N(1)-C(1)-C(2)	103.5(3)	C(5)-C(6)-H(6A)	110.6

Table 20. Bond Angles [deg] for **7d**•SnCl₄.

bond angle	[deg]	bond angle	[deg]
C(7)-C(6)-H(6B)	110.6	N(3)-C(9)-H(9B)	108.9
C(5)-C(6)-H(6B)	110.6	C(10)-C(9)-H(9B)	108.9
H(6A)-C(6)-H(6B)	108.8	H(9A)-C(9)-H(9B)	107.8
C(6)-C(7)-C(8)	105.9(4)	C(9)-C(10)-C(11)	112.6(3)
C(6)-C(7)-H(7A)	110.6	C(9)-C(10)-H(10A)	109.1
C(8)-C(7)-H(7A)	110.6	C(11)-C(10)-H(10A)	109.1
C(6)-C(7)-H(7B)	110.6	C(9)-C(10)-H(10B)	109.1
C(8)-C(7)-H(7B)	110.6	C(11)-C(10)-H(10B)	109.1
H(7A)-C(7)-H(7B)	108.7	H(10A)-C(10)-H(10B)	107.8
N(2)-C(8)-C(7)	103.5(4)	C(10)-C(11)-C(10)#1	113.9(4)
N(2)-C(8)-H(8A)	111.1	C(10)-C(11)-H(11)	108.8
C(7)-C(8)-H(8A)	111.1	C(10)#1-C(11)-H(11)	108.8
N(2)-C(8)-H(8B)	111.1	N(3)-C(12)-H(12A)	109.5
C(7)-C(8)-H(8B)	111.1	N(3)-C(12)-H(12B)	109.5
H(8A)-C(8)-H(8B)	109.0	H(12A)-C(12)-H(12B)	109.5
N(3)-C(9)-C(10)	113.2(3)	N(3)-C(12)-H(12C)	109.5
N(3)-C(9)-H(9A)	108.9	H(12A)-C(12)-H(12C)	109.5
C(10)-C(9)-H(9A)	108.9	H(12B)-C(12)-H(12C)	109.5

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,y,-z+1/2 #2 x,-y+1,-z #3 -x,y,-z+1/2

Table 21. Torsional angles [deg] for **7d**•SnCl₄.

torsional angle	[deg]	torsional angle	[deg]
N(3)-P(1)-O(1)-Sn(1)	137.1(2)	P(1)-N(1)-C(1)-C(2)	125.9(4)
N(1)-P(1)-O(1)-Sn(1)	13.3(3)	N(1)-C(1)-C(2)-C(3)	33.0(5)
N(2)-P(1)-O(1)-Sn(1)	-98.0(3)	C(1)-C(2)-C(3)-C(4)	-29.0(5)
O(1)#1-Sn(1)-O(1)-P(1)	-144.5(3)	C(1)-N(1)-C(4)-C(5)	131.1(4)
Cl(2)-Sn(1)-O(1)-P(1)	-58.9(2)	P(1)-N(1)-C(4)-C(5)	-24.9(4)
Cl(2)#1-Sn(1)-O(1)-P(1)	127.6(3)	C(1)-N(1)-C(4)-C(3)	7.4(5)
Cl(1)#1-Sn(1)-O(1)-P(1)	-152.7(11)	P(1)-N(1)-C(4)-C(3)	-148.6(3)
Cl(1)-Sn(1)-O(1)-P(1)	34.9(2)	C(2)-C(3)-C(4)-N(1)	13.8(5)
O(1)-P(1)-N(1)-C(1)	99.2(4)	C(2)-C(3)-C(4)-C(5)	-101.0(5)
N(3)-P(1)-N(1)-C(1)	-22.9(4)	C(8)-N(2)-C(5)-C(4)	124.6(4)
N(2)-P(1)-N(1)-C(1)	-140.1(4)	P(1)-N(2)-C(5)-C(4)	-22.8(4)
O(1)-P(1)-N(1)-C(4)	-109.9(3)	C(8)-N(2)-C(5)-C(6)	1.8(5)
N(3)-P(1)-N(1)-C(4)	128.1(3)	P(1)-N(2)-C(5)-C(6)	-145.6(3)
N(2)-P(1)-N(1)-C(4)	10.9(3)	N(1)-C(4)-C(5)-N(2)	28.3(4)
O(1)-P(1)-N(2)-C(8)	-8.6(4)	C(3)-C(4)-C(5)-N(2)	142.3(4)
N(3)-P(1)-N(2)-C(8)	112.2(4)	N(1)-C(4)-C(5)-C(6)	142.8(4)
N(1)-P(1)-N(2)-C(8)	-133.8(4)	C(3)-C(4)-C(5)-C(6)	-103.2(5)
O(1)-P(1)-N(2)-C(5)	132.8(3)	N(2)-C(5)-C(6)-C(7)	17.9(6)
N(3)-P(1)-N(2)-C(5)	-106.3(3)	C(4)-C(5)-C(6)-C(7)	-98.1(5)
N(1)-P(1)-N(2)-C(5)	7.7(3)	C(5)-C(6)-C(7)-C(8)	-30.8(6)
O(1)-P(1)-N(3)-C(9)	-6.8(4)	C(5)-N(2)-C(8)-C(7)	-20.3(5)
N(1)-P(1)-N(3)-C(9)	122.9(3)	P(1)-N(2)-C(8)-C(7)	121.5(4)
N(2)-P(1)-N(3)-C(9)	-131.7(3)	C(6)-C(7)-C(8)-N(2)	31.3(5)
O(1)-P(1)-N(3)-C(12)	162.5(3)	C(12)-N(3)-C(9)-C(10)	-69.9(4)
N(1)-P(1)-N(3)-C(12)	-67.9(3)	P(1)-N(3)-C(9)-C(10)	99.4(4)
N(2)-P(1)-N(3)-C(12)	37.5(3)	N(3)-C(9)-C(10)-C(11)	178.3(3)
C(4)-N(1)-C(1)-C(2)	-25.4(5)	C(9)-C(10)-C(11)-C(10)#1	58.1(3)

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,y,-z+1/2 #2 x,-y+1,-z #3 -x,y,-z+1/2

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