Syntheses of Materials. 1,2-bis(diphenylarseno)ethylene (23) and cis-1,2bis(diphenylphosphino)ethane (24), were purchased from either Aldrich or Alfa and recrystallized $3 \times$ from butanol before use. 4,4'-diphenyl-2,2'-bipyridine (21) and Bathophenanthroline (22) were purchased from GFS Chemicals. 21 was $\sim 50 \%$ pure and was purified by repeated washings and recrylstalization from DMF to yield 2.3 g out of 15 g purchased. The following chemicals were used as received. 2,2'-Bipyridine (26) was purchased from either Alfa or Aldrich. Hydrogen peroxide $35 \%$ in water, acetyl bromide, tert-butyl lithium in hexane and 1.6 M butyl lithium in hexane were purchased from Aldrich. Potassium hexachloroosmiate, fuming sulfuric acid, $30 \% \mathrm{SO}_{3}$, and fuming $90 \%$ nitric acid were purchased from Alfa.
cis-1,2-vinylenebis(diphenylarsine) (25)(Fig. 3): The procedure used was a modification of a previous method resulting in an improved yield of product. ${ }^{32}$ Diphenylarsine (Organometalics, $25.00 \mathrm{~g}, 108.6 \mathrm{mmol}$ ) was used as received and added to 400 mL of freshly dried (sodium/benzophenone) THF. The solution was stirred under nitrogen and cooled to $-78{ }^{\circ} \mathrm{C}$ using an acetone/dry ice bath. To this solution was added n-butyl lithium ( 1.6 M in hexane, $1.05 \mathrm{eq}, 114.1 \mathrm{mmol}$ ). The solution was allowed to stir for 1 h . The acetone bath was then removed and cis-dichloroethylene (TCI-America, 10.66 g , 110.0 mmol ) was added. The solution was allowed to slowly warm to $18.5^{\circ} \mathrm{C}$ and react overnight. Water was added and the THF was removed by rotary evaporation under vacuum at $40^{\circ} \mathrm{C}$. The water was removed by filtration and the solid material was washed with large amounts of de-ionized water. The sample was dried under vacuum and then recrystalized three times from butanol. Yield: $23.97 \mathrm{~g}(91 \%) .{ }^{1} \mathrm{H}$ NMR (DMSO): 7.63 (s,

2H), $7.35(20 H)$. Elemental anal. calc.: C, $64.48 \%$; H, $4.58 \%$. found: C, $64.10 \%$; H, $4.28 \%$.

4,4'-dibromo-2, 2'-bipyridine (Fig. 4): The procedure used was an adaptation and combination of the methods reported by Haginiwa ${ }^{33}$ and Chase ${ }^{34}$.

4,4'-di(biphenyl)-2, 2'-bipyridine (15): 4,4'dibromo-2,2'-bipyridine ( $2.000 \mathrm{~g}, 6.37 \mathrm{mmol}$ ), and biphenyl-4-boronic acid neopentyl glycol ester ( $4.238 \mathrm{~g}, 15.92 \mathrm{mmol}$ ) were dissolved into 80 mL DMF (Aldrich). The solution was stirred for 1 h under nitrogen and tetrakis(triphenylphosphine) $\operatorname{Pd}(0)(0.191 \mathrm{mmol})$ was added. The solution was heated to $60{ }^{\circ} \mathrm{C}$ and 10 mL deoxygenated saturated potassium carbonate aqueous solution was added drop wise over a period of 5 min . The solution was then warmed to $85-95^{\circ} \mathrm{C}$. The coupling was allowed to proceed under these conditions for 24 h . The solution was warmed to $115{ }^{\circ} \mathrm{C}$ and the reaction was allowed to proceed for an additional 5 h . The solution was cooled to room temperature and poured into 500 mL of 0.25 M KOH solution. The precipitate was collected by vacuum filtration and washed with water, allowed to dry, and then washed with methylene chloride. The methylene chloride layer was a dark brown while the precipitate was a light gray which gave a mass of 461 . The gray powder was suspended into 250 mL of xylene (Fisher Scientific) and refluxed for 10 minutes and filtered. Small colorless crystals ( 150 mg ) of 4,4'-bis(p-biphenyl)-2, $\mathbf{2}^{\prime}$ bipyridine formed in the mother liquid. This process was repeated 10 times to yield 1.5 g of material ( $51 \%$, remainder is the mono-substituted material, which was soluble in organic solvents such as methylene chloride). 4,4'-bis(p-biphenyl)-2,2'-bipyridine was very insoluble in organic solvents at room temperature. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{DCCl}_{3}\right)$ as a $\mathrm{Ru}(\mathrm{II})(\text { Heptafluorobutyrate })_{2,}\left(\mathrm{Ru}(\mathrm{II})(\mathrm{HFB})_{2}\right)$ complex: $9.08(\mathrm{~d}, 6 \mathrm{H}), 7.97(\mathrm{~m}, 18 \mathrm{H}), 7.80$
$(\mathrm{m}, 18 \mathrm{H}), 7.61(\mathrm{~m}, 12 \mathrm{H}), 7.45(\mathrm{~m}, 18 \mathrm{H})$. Mass spectrometry (m/z): 461.6. Elemental analysis calculated: C, 88.67; H, 5.25; N, 6.08, found C, 89.06; H, 5.32; N, 6.21. 4,4'-bis(diphenylether)-2, ${ }^{\prime}$ '-bipyridine(16) was made following the same procedure as 15: Yield 0.95 g (30.4\%). 1H NMR (DMSO): 8.750 (d, 2H), 8.631 (m, 4H), 7.904 (m, $4 \mathrm{H}), 7.800(\mathrm{~m}, 4 \mathrm{H}), 7.459(\mathrm{t}, 2 \mathrm{H}), 7.147(\mathrm{~m}, 8 \mathrm{H})$. Mass spectroscopy $(\mathrm{m} / \mathrm{z}): 493.2$. Elemental analysis calculated: C, 82.91; H, 4.91; N, 5.69, found: C, 83.09; H, 4.99; N, 5.77.

4,7-Bis-(p-methoxyphenyl)-1,10-phenanthroline (17) and 4,7-Bis-(p-bromophenyl)-1,10phenanthroline(18): These compounds have been reported previously. ${ }^{17}$ Our results essentially matched theirs. They report $\mathbf{1 8}$ as a tan solid, we recrystalized 3X from DMF to produce a colorless crystalline solid.

The following phenanthroline derivatives were synthesized from $\mathbf{1 8}$ 4,7-Bis-(4'-phenoxy-biphenyl-4-yl)-1,10-phenanthroline (19) was made following the same procedure as 15. Elemental analysis: calculated: C, $86.20 ; \mathrm{H}, 4.82 ; \mathrm{N}, 4.19$; found: C, 86.32; H, 4.98; N, 4.28. 1HNMR (DMSO): 9.20 (d 2H), 8.88-7.65 (m 18H), 7.45 (t 4 H), 7.25-7.06 (m 8H). Recrystalized from benzene to give a colorless crystalline solid. 4,7-Bis-(4-naphthalen-2-yl-phenyl)-1,10-phenanthroline (20) was made following the same procedure as 15. Elemental analysis, calculated: C, 90.38 ; H, 4.83; N, 4.79; found: C, 90.45; H, 4.86; N, 4.81. 1HNMR (DCCb ): 9.29 (d 2H), 8.14 (s 2H), 8.03-7.75 (m 14 H), 7.72-7.63 (m 6H), 7.57-7.46 (m 4H). Recrystalized from DMF to give colorless crystalline solid.

Os(II) bis(4,4'-diphenyl-2, 2'-bipyridine) Dichloride (Fig. 5): The procedure used was a modification of a previous report. ${ }^{19}$ Potassium hexachloroosmiate, ( $\left.1 \mathrm{~g}, 2.078 \mathrm{mmol}\right)$, and

21, ( $1.40 \mathrm{~g}, 4.54 \mathrm{mmol}$ ), were suspended in DMF ( 25 mL ), purged with nitrogen for 1 h , and refluxed under a nitrogen atmosphere for 3 h . The solution was cooled to $18.5{ }^{\circ} \mathrm{C}$ and filtered. Much of the product, a mixture of $\left[\mathrm{Os}(\mathbf{2 1})_{2} \mathrm{Cl}_{2}\right]$ and $\left[\mathrm{Os}(\mathbf{2 1})_{2} \mathrm{Cl}_{2}\right]^{+}$ precipitated out as it was only partially soluble in DMF ( 0.8 g of Os product per 60 mL DMF). Extractions were filtered through a fine frit and performed repeatedly until the filtrate, KCl , was only a light brown color.

2 g of sodium dithionite (Alfa) was added and dissolved in 400 mL cooled de-ionized water $\left(0-3{ }^{\circ} \mathrm{C}\right)$. The DMF solution of the Os compound was added drop wise to the sodium dithionite solution. Instantly a dark purple precipitate of $\mathrm{Os}(\mathrm{II})$ bis(4,4'-diphenyl-2,2'-bipyridine) dichloride formed and was collected using vacuum filtration and washed repeatedly with water. This was done repeatedly with each extraction. Yield: $92 \%$. The bathophenathroline complex had increased solubility and allowed the use of a silica flash column with methylene chloride mobile phase for purification.
[Os(II) bis(4,4'-diphenyl-2,2'-bipyridine) 1,2-bis(diphenylarseno)ethane] ${ }^{2+}$ (1) (Fig. 6) :
Osmium bis $(\mathbf{2 1})$ dichloride $(0.5 \mathrm{~g}, 0.570 \mathrm{mmol})$ and $23(0.291 \mathrm{~g}, 0.600 \mathrm{mmol})$ were added to 10 mL of a mixture of 2-(2-ethoxyethoxy)ethanol and glycerol (75:25 by volume). The solution was purged with nitrogen for 1 h and then the solution was refluxed for 1.5 h . After cooling, the solution was added to water and the appropriate counter ion, as the sodium salt of HFB, triflate (Tf), tosylate (Ts), or chloride was added. The precipitate was collected by filtration, washed with water and dried under vacuum at $40{ }^{\circ} \mathrm{C}$. The dried precipitate was dissolved in methylene chloride ( 20 mL per g of complex) and precipitated with hexane. This was repeated three times to get a bright red product. Yield: $90 \%$. Mass spectrometry (m/2z): 647.1. Isotope Pattern: 645.1, 645.6,
646.1, 646.6, 647.1, 647.6, and 648.1. Fragmentation peak at $m / 2 z=478.1(m / 2 z-308-$ 28), which corresponds to the fragmentation of a bpy ligand and the ethane bridge on the arsine. Elemental analysis, calculated: (HFB counter ions). C, 54.49; H, 3.28; N, 3.26, found C, 54.65; H, 3.36; N, 3.30.
$\left[O s(I I) \quad\left(4,4^{\prime} \text {-diphenyl-2,2'-bipyridine }\right)_{2} \quad \text { cis-1,2-bis(diphenylphosphino)ethylene }\right]^{2+}$
$(H F B)_{2}$ (2) (Fig 6): was made following the same procedure as $\mathbf{1}$ yielding an orange product. Yield: $85 \%$. Mass spectrometry $(\mathrm{m} / 2 \mathrm{z}): 602.7$. Isotope pattern $\mathrm{m} / 2 \mathrm{z}: 599.7$, $600.2,600.7,601.2,601.7,602.2,602.7,603.2$. Elemental analysis, calculated: C, $57.49 \%$; H, $3.34 \%$; N, $3.44 \%$. Found: C, 58.00; H, 3.41; N, 3.33.
$\left[O s(I I)\left(4,4^{\prime} \text {-diphenyl-2,2'-bipyridine }\right)_{2} \quad \text { cis-1,2-vinylenebis(diphenylarsine) }\right]^{2+}(T f)_{2}(\mathbf{3})$ (Fig 6): was made following the same procedure as 1 yielding a red product. Yield: $94 \%$. Mass spectrometry $(\mathrm{m} / 2 \mathrm{z}): 646.1$. Isotope pattern $(\mathrm{m} / 2 \mathrm{z}): 643.6,644.1,644.6,645.1$, 645.6, 646.1, 646.6, 647.1. Elemental analysis, calculated: C, $54.41 \%$; H, 3.42\%; N, $3.52 \%$. Found C, $54.51 ;$ H, 3.48; N, 3.36.
$[\text { Os(II) (4,4'-bis(p-diphenylether)-2,2'-bipyridine })_{2}$ cis-1,2vinylenebis(diphenylarsine) $]^{2+}(T f)_{2}(4)$ (Fig 6): was made following the same procedure as 1 yielding a red product. Yield: $90 \%$ Mass spectrometry ( $\mathrm{m} / 2 \mathrm{z}$ ): 830.2. Isotope Pattern for 830.2 peak: 828.2, 828.7, 829.2 829.7, 830.2, 830.7, 831.2. Elemental analysis, calculated: C, $58.89 ; \mathrm{H}, 3.60 ; \mathrm{N}, 2.86$, found C, $59.01 ; \mathrm{H}, 3.59 ; \mathrm{N}, 2.88$.
$[\text { Os(II) (4,4'-bis(p-biphenyl)-2,2'-bipyridine })_{2} \quad$ 1,2-bis(diphenylarseno)ethane $]^{2+}(\mathrm{HFB})_{2}$ (5) (Fig 6): was made following the same procedure as $\mathbf{1}$ yielding a brick red product. Yield: $90 \%$ Mass spectrometry $(\mathrm{m} / 2 \mathrm{z}): 646.1$. Isotope pattern $(\mathrm{m} / 2 \mathrm{z}): 643.6,644.1$,
644.6, 645.1, 645.6, 646.1, 646.6, 647.1. Elemental analysis, calculated: C, 60.54; H, 3.59; N, 2.77, found C; 59.92; H, 3.34; N, 2.70.
$[\text { Os(II) (4,4'-bis(p-biphenyl)-2,2'-bipyridine })_{2} \quad$ cis-1,2-vinylenebis(diphenylarsine) $]^{2+}$ $(T f)_{2}(\mathbf{6})($ Fig 6): was made following the same procedure as $\mathbf{1}$ a red product. Yield: $89 \%$ Mass spectrometry $(\mathrm{m} / 2 \mathrm{z}): 798.2$. Isotope pattern (m/2z): 795.7, 796.2, 796.7, 797.2, 797.7, 798.2, 798.7, and 799.2. Elemental analysis, calculated: C, 60.88; H, 3.73; N, 2.96, found: C, 61.01, H, 4.02; N, 2.85
$[O s(I I) \text { (bathophenanthroline) })_{2}$ cis-1,2-bis(diphenylphosphino)ethylene $]^{2+}(T f)_{2}(7)$ (Fig. 6): was made following the same procedure as $\mathbf{1}$ yielding a yellow-orange product. Yield: $92 \%$. Mass spectrometry $(\mathrm{m} / 2 \mathrm{z}): 626.2$. Isotope pattern $(\mathrm{m} / 2 \mathrm{z}): 623.7,624.2,624.7$, 625.2, 625.7, 626.2, 626.7, 627.2, and 627.7, no fragmentation. Elemental analysis, calculated: C, $58.91 \%$; H, $3.51 \%$; N, $3.62 \%$. Found C, $58.78 ;$ H, 3.52; N, 3.66. $[O s(I I) \text { (bathophenanthroline) })_{2}$ cis-1,2-vinylenebis(diphenylarsine) $]^{2+}(T s)_{2}(\mathbf{8})$ (Fig 6): was made following the same procedure as $\mathbf{1}$ yielding a red-orange product. Yield: $90 \%$. Mass spectrometry (m/2z): 670.1. Isotope pattern (m/2z): 667.6, 668.1, 668.6, 669.1, 669.6, 670.1, 670.6, 671.1, no fragmentation. Elemental analysis: calculated: C, $55.75 \%$; H, $3.32 \%$; N, $3.42 \%$. Found C, $56.05 ;$ H, 3.34 ; N, 3.44 .
[Osmium(II) (4,7-Bis-(p-methoxyphenyl)-1,10-phenanthroline) $2_{2}$ 1,2bis(diphenylarseno)ethane $]^{2+}(T s)_{2}(\mathbf{9})$ (Fig 6): was made following the same procedure as 1 yielding a brick red product. Yield: $89 \%$. Mass spectrometry $\mathrm{m} / 2 \mathrm{z}=731.2$. Isotope pattern: 728.7, 729.2, 729.7, 730.2, 730.7, 731.2, 731.7, 732.2, and 732.7. Fragment at $\mathrm{m} / 2 \mathrm{z}-7.5$ for loss of a methyl group. Elemental analysis calc. C, 61.26; H, 4.36; N, 3.11; found C, 61.36; H, 4.41; N, 3.05.
bis(diphenylphosphino)ethylene $]^{2+}(T s)_{2}$ (10) (Fig 6): was made following the same procedure as 1 yielding a yellow-orange product. Yield: 95\%. Mass Spectrometry $\mathrm{m} / 2 \mathrm{z}$ $=686.2$. Isotope pattern: 683.7, 684.2, 684.7, 685.2, 685.7, 686.2, 686.2, 687.2, and 687.7. Fragment at $\mathrm{m} / 2 \mathrm{z}-7.5$ for the fragmentation of a methyl group. Elemental analysis calc. C, 64.47 ; H, 4.47; N, 3.27; found C, 64.57; H, 4.65; N, 3.25.
[Osmium(II) (4,7-Bis-(p-methoxyphenyl)-1,10-phenanthroline $)_{2}$ cis-1,2vinylenebis(diphenylarsine) $]^{2+}(T s)_{2}(\mathbf{1 1}) \quad$ (Fig 6): was made following the same procedure as $\mathbf{1}$ yielding a red product. Yield: $92 \%$ Mass Spectrometry m/2z: 730.1. Isotope Pattern: 727.6, 728.1, 728.6, 729.1, 729.6, 730.1, 730.6, 731.1, and 731.6. Fragment appears at $\mathrm{m} / 2 \mathrm{z}-7.5$ for loss of a methyl group. Elemental analysis calc.: C, 61.33; H, 4.25; N, 3.11; found: C, 61.64; H, 4.33, N, 3.12.
[Osmium(II) (4,7-Bis-(p-bromophenyl)-1,10-phenanthroline) $)_{2}$ cis-1,2vinylenebis(diphenylarsine) $]^{2+}(T s)_{2}(\mathbf{1 2 )}$ (Fig 6): was made following the same procedure as $\mathbf{1}$ yielding a dark red product. Yield: $89 \%$ Mass spectrometry $\mathrm{m} / 2 \mathrm{z}=827.9$, isotope pattern 824.4, 824.9, 825.4, 825.9, 826.4, 826.9, 827.4, 827.9, 828.4, 828.9, 829.4, 829.9. Elemental analysis: calculated: C, 51.76; H, 3.31; N, 2.87, Found: C, 52.01; H, 3.38; N, 2.88 .
[Osmium(II) (4,7-Bis-(4'-phenoxy-biphenyl-4-yl)-1,10-phenanthroline) $2_{2}$ cis-1,2vinylenebis(diphenylarsine) $]^{2+}(T s)_{2}(\mathbf{1 3})$ (Fig 6): was made following the same procedure as 1 yielding a red product. Yield: $93 \%$ Mass Spectrometry $m / 2 z=1006.1$, isotope pattern 1004.2, 1004.7, 1005.2, 1005.7, 1006.2, 1006.7, 1007.2, 1007.7. Elemental analysis, calculated: C, 69.38; H, 4.28; N, 2.38; Found: C, 69.07; H, 4.36; N, 2.44.
[Osmium(II) (4,7-Bis-(4-naphthalen-2-yl-phenyl)-1,10-phenanthroline) ${ }_{2}$ cis-1,2vinylenebis(diphenylarsine) $]^{2+}(T s)_{2}(\mathbf{1 4})$ (Fig 6): was made following the same procedure as $\mathbf{1}$ yielding a red product. Yield: $88 \%$ Mass Spectrometry $\mathrm{m} / 2 \mathrm{z}=922.7$, isotope
pattern 920.2, 920.7, 921.2, 921.7, 922.2, 922.7, 923.2, 923.7. Elemental analysis, calculated: C, 70.25 ; H, 4.33; N, 2.56, Found: C, 70.56; H, 4.32; N, 2.48.


Absorbance and Emission of [Osmium Bis(4,4'-diphenyl-2,2'-bipyridine) cis-1,2-
Bis(diphenylphosphino)ethylene)] Dichloride in Methanol


Absorption and Emssion of [Osmium (4,4'-di(diphenylether)-2,2'-bipyridine) ${ }_{2}$ cis-1,2vinylenebis(diphenylarsine)] Tosylate ${ }_{2}$ in Methylene Chloride


Wavelength ( nm )

