

Analytical Data of T-n-T (n = 11 and 12), A-n-A (n = 11 and 12), and T-n-A (n = 11 and 12)

N, N'-bis[3-(2,4-dihydroxy-5-methylpyrimidine-1-yl)propionyl]1, 11-diaminoundecane (T-11-T). This bolaamphiphile was synthesized in the same way as T-10-T. mp 200.0–204.0; TLC (silica gel, 1:1 CHCl₃/MeOH) R_f = 0.9; ¹HNMR (600 MHz, DMSO-*d*₆, 25°C): δ 1.21–1.34 (m, 18H, –CH₂–), 1.71 (s, 6H, CH₃-5), 2.41 (t, *J* = 6.6 Hz, 4H, –CH₂CH₂CONH–), 2.99 (dt, *J* = 6.6, 6.6 and 7.2 Hz, 4H, –CONHCH₂–), 3.80 (t, *J* = 6.6 Hz, 4H, –CH₂CH₂CONH–), 7.37 (s, 2H, H-6), 7.94 (t, *J* = 5.4 Hz, 2H, –CONH–), 11.2 (s, 2H, NH-3). Anal. Calcd for C₂₇H₄₂N₆O₆·H₂O: C, 57.43; H, 7.85; N, 14.88. Found: C, 57.03; H, 7.57; N, 14.73.

N, N'-bis[3-(6-aminopurine-9-yl)propionyl]1, 11-diaminoundecane (A-11-A). This bolaamphiphile was synthesized in the same way as A-10-A. mp 224.0–230.0; TLC (silica gel, 2:1 CHCl₃/MeOH) R_f = 0.6; ¹HNMR (600 MHz, DMSO-*d*₆, 25°C) δ 1.12–1.31 (m, 18H, –CH₂–), 2.65 (t, *J* = 6.6 Hz, 4H, –CH₂CH₂CONH–), 2.97 (dt, *J* = 6.0, 6.0 and 7.2 Hz, 4H, –CONHCH₂–), 4.33 (t, *J* = 6.6 Hz, 4H, –CH₂CH₂CONH–), 7.20 (s, 4H, NH₂-6), 7.88 (t, *J* = 5.4 Hz, 2H, –CONH–), 7.98 (s, 2H, H-8), 8.13 (s, 2H, H-2). Anal. Calcd for C₂₇H₄₀N₁₂O₂·H₂O: C, 55.65; H, 7.26; N, 28.84. Found: C, 55.40; H, 7.33; N, 28.59.

N, N'-bis[3-(2,4-dihydroxy-5-methylpyrimidine-1-yl)propionyl]1, 12-diaminododecane (T-12-T). This bolaamphiphile was also synthesized in the same way as T-10-T. mp 211.8–214.4; TLC (silica gel, 1:1 CHCl₃/MeOH) R_f = 0.9; ¹HNMR (600 MHz, DMSO-*d*₆, 25°C): δ 1.22–1.34 (m, 18H, –CH₂–), 1.71 (s, 6H, CH₃-5), 2.41 (t, *J* = 6.6 Hz, 4H, –CH₂CH₂CONH–), 3.00 (dt, *J* = 6.6, 6.6 and 7.2 Hz, 4H, –CONHCH₂–), 3.81 (t, *J* = 6.6 Hz, 4H, –CH₂CH₂CONH–), 7.37 (s, 2H, H-6), 7.94 (t, *J* = 5.4 Hz, 2H, –CONH–), 11.2 (s, 2H, NH-3). Anal. Calcd for C₂₈H₄₄N₆O₆: C, 59.98; H, 7.91; N, 14.99. Found: C, 59.98; H, 7.93; N, 14.78.

***N, N'*-bis[3-(6-aminopurine-9-yl)propionyl]1, 12-diaminododecane (A-12-A).**

This bolaamphiphile was also synthesized in the same way as A-10-A. mp 228.9–233.4; TLC (silica gel, 2:1 CHCl₃/methanol) R_f = 0.4; ¹H-NMR (600 MHz, DMSO-*d*₆, °C) δ 1.20–1.30 (m, 18H, –CH₂–), 2.65 (t, *J* = 6.6 Hz, 4H, –CH₂CH₂CONH–), 2.97 (dt, *J* = 6.0, 6.0 and 7.2 Hz, 4H, –CONHCH₂–), 4.33 (t, *J* = 6.6 Hz, 4H, –CH₂CH₂CONH–), 7.19 (s, 4H, NH₂-6), 7.88 (t, *J* = 5.4 Hz, 2H, –CONH–), 7.98 (s, 2H, H-8), 8.13 (s, 2H, H-2). Anal. Calcd for C₂₈H₄₂N₁₂O₂·1/2H₂O: C, 57.22; H, 7.37; N, 28.60. Found: C, 57.42; H, 7.32; N, 28.64.

***N*-[3-(2,4-dihydroxy-5-methylpyrimidine-1-yl)propionyl], *N'*-[3-(6-aminopurine-9-yl)propionyl]1,11-diaminoalkane (T-11-A).** This

bolaamphiphile was synthesized in the same way as T-10-A. mp 204.0–207.0; TLC (silica gel, 1:1 CHCl₃/MeOH) R_f = 0.7; ¹H-NMR (600 MHz, DMSO-*d*₆, 25°C): δ 1.20–1.35 (m, 18H, –CH₂–), 1.71 (s, 3H, Thy-CH₃-5), 2.41 (t, *J* = 6.6 Hz, 2H, Thy-(CH₂)₂CONHCH₂CH₂–), 2.65 [t, *J* = 6.6 Hz, 2H, Ade-(CH₂)₂CONHCH₂CH₂–], 3.00 (dt, *J* = 6.2, 3.7, 6.2, and 6.6 Hz, 4H, –CONHCH₂–), 3.81 (t, *J* = 6.6 Hz, 2H, Thy-CH₂CH₂CONH–), 4.33 (t, *J* = 6.6 Hz, 2H, Ade-CH₂CH₂CONH–), 7.16 (s, 2H, Ade-NH₂-6), 7.36 (s, 1H, Thy-H-6), 7.86 [t, *J* = 5.5 Hz, 1H, Ade-(CH₂)₂CONH–], 7.92 [t, *J* = 5.4 Hz, 1H, Thy-(CH₂)₂CONH–], 7.98 (s, 1H, Ade-H-8), 8.13 (s, 1H, Ade-H-2), 11.2 (s, 1H, Thy-NH-3). Anal. Calcd for C₂₇H₄₁N₉O₄: C, 58.36; H, 7.44; N, 22.69. Found: C, 58.55; H, 7.56; N, 22.35.

***N*-[3-(2,4-dihydroxy-5-methylpyrimidine-1-yl)propionyl], *N'*-[3-(6-aminopurine-9-yl)propionyl]1,12-diaminoalkane (T-12-A).** This

bolaamphiphile was also synthesized in the same way as T-10-A. mp 228.0–228.6; TLC (silica gel, 1:1 CHCl₃/MeOH) R_f = 0.7; ¹H-NMR (600 MHz, DMSO-*d*₆, 25°C): δ 1.20–1.32 (m, 20H, –CH₂–), 1.71 (s, 3H, Thy-CH₃-5), 2.41 (t, *J* = 6.6 Hz, 2H, Thy-(CH₂)₂CONHCH₂CH₂–), 2.65 [t, *J* = 6.6 Hz, 2H, Ade-

$(\text{CH}_2)_2\text{CONHCH}_2\text{CH}_2^-$], 2.98 (dt, $J = 6.2, 3.7, 6.2$, and 6.6 Hz, 4H, $-\text{CONHCH}_2^-$), 3.81 (t, $J = 6.6$ Hz, 2H, Thy- $\text{CH}_2\text{CH}_2\text{CONH}-$), 4.33 (t, $J = 6.6$ Hz, 2H, Ade- $\text{CH}_2\text{CH}_2\text{CONH}-$), 7.16 (s, 2H, Ade-NH₂-6), 7.36 (s, 1H, Thy-H-6), 7.86 [t, $J = 5.5$ Hz, 1H, Ade- $(\text{CH}_2)_2\text{CONH}-$], 7.92 [t, $J = 5.4$ Hz, 1H, Thy- $(\text{CH}_2)_2\text{CONH}-$], 7.98 (s, 1H, Ade-H-8), 8.13 (s, 1H, Ade-H-2), 11.2 (s, 1H, Thy-NH-3). Anal. Calcd for $\text{C}_{28}\text{H}_{43}\text{N}_9\text{O}_4$: C, 59.03; H, 7.61; N, 22.13. Found: C, 59.22; H, 7.56; N, 22.30.

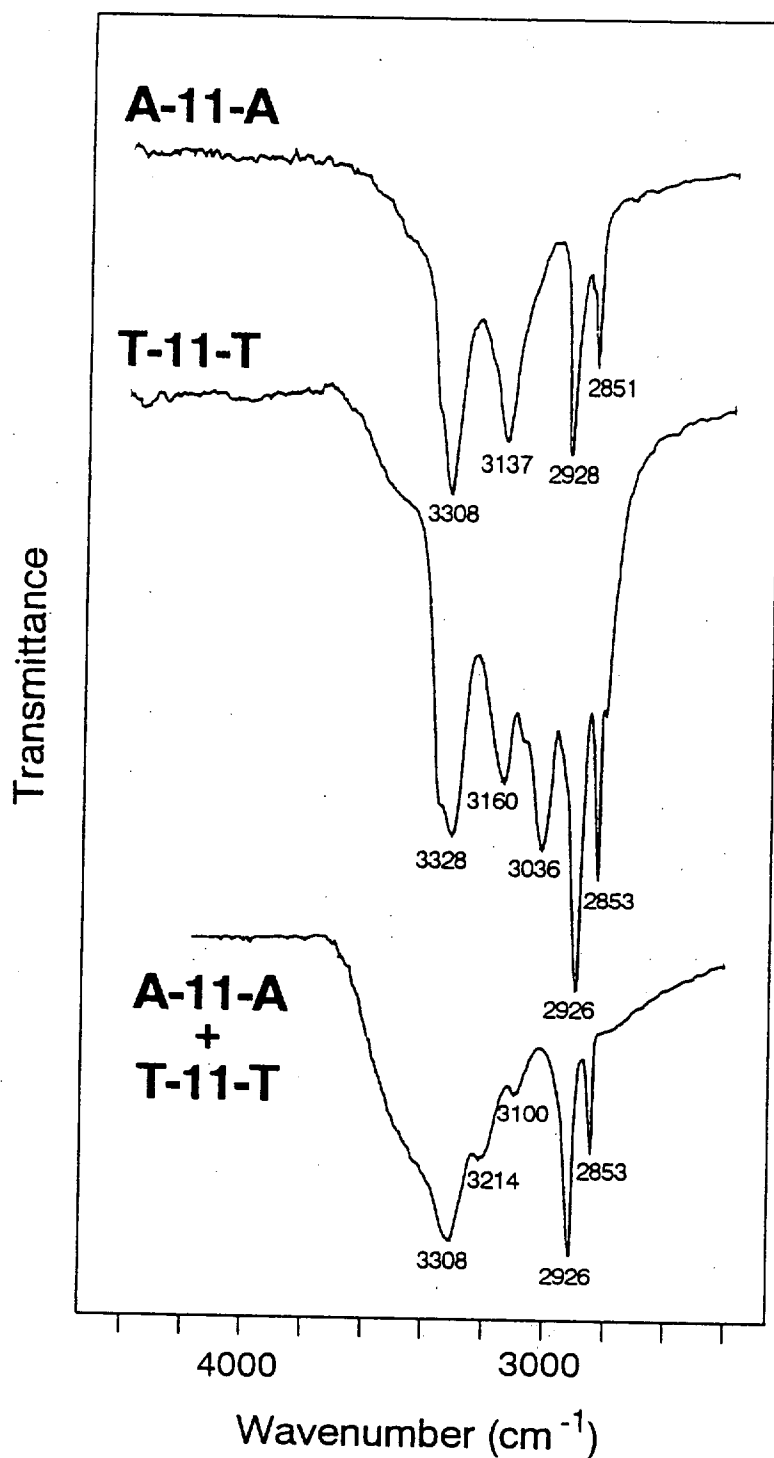


Fig. S-1. FT-IR Spectra (2700 –4400 cm^{-1}) of **A-11-A**, **T-11-T**, and their 1:1 Self-(hetero)-Assembly

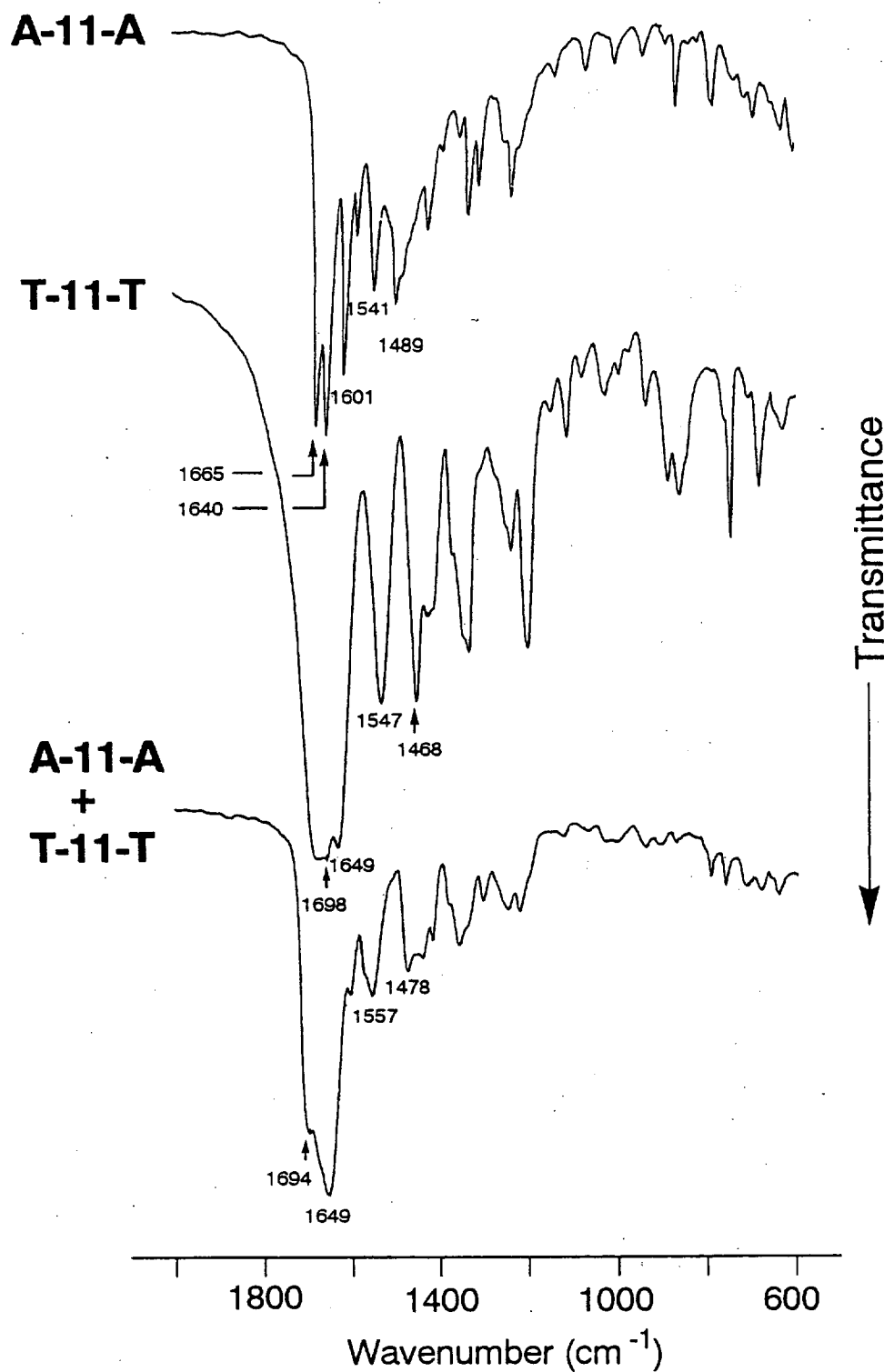


Fig. S-2. FT-IR Spectra (600 –1800 cm^{-1}) of **A-11-A**, **T-11-T**, and their 1:1 Self-(hetero)-Assembly

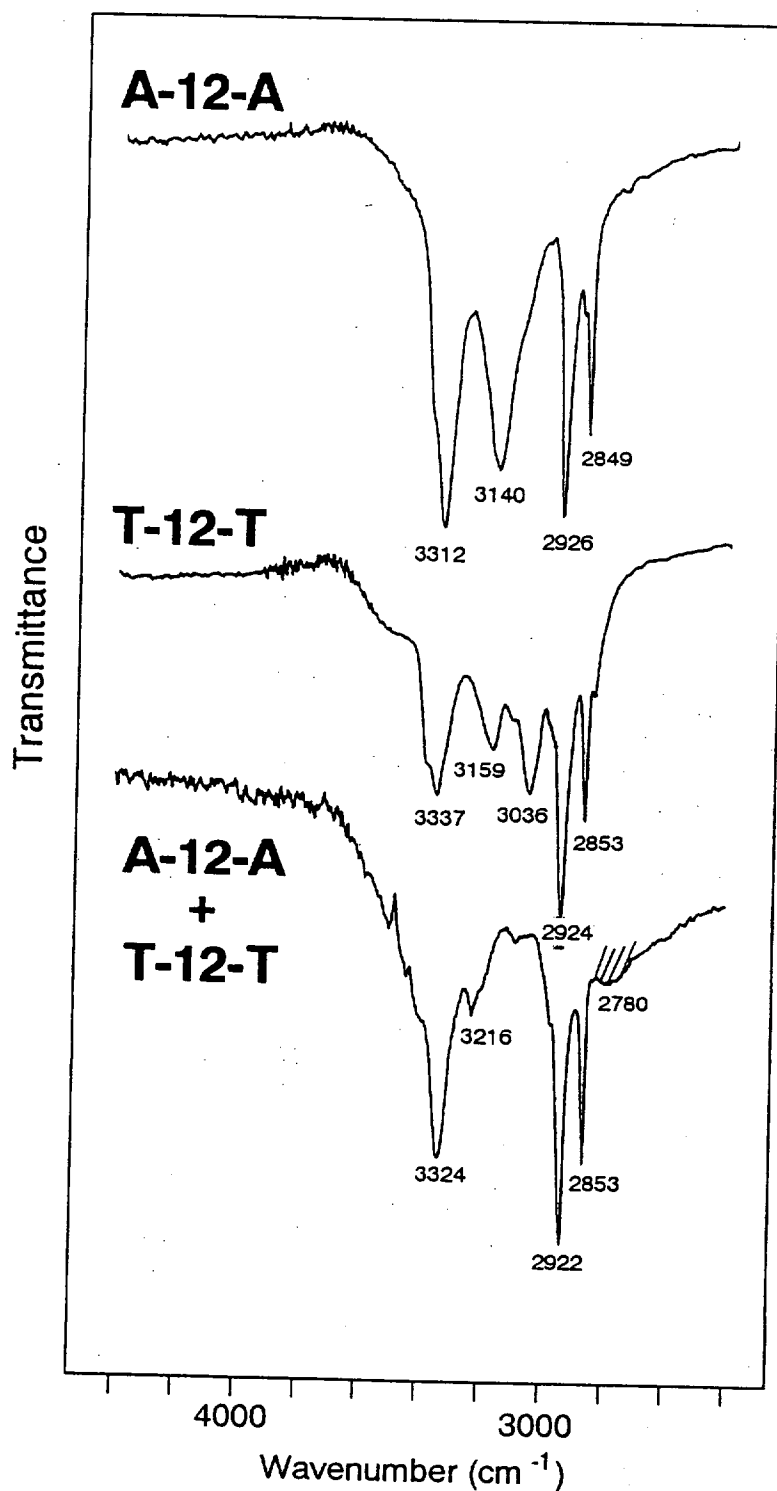


Fig. S-3. FT-IR Spectra (2700 –4400 cm^{-1}) of **A-12-A**, **T-12-T**, and their 1:1 Self-(hetero)-Assembly

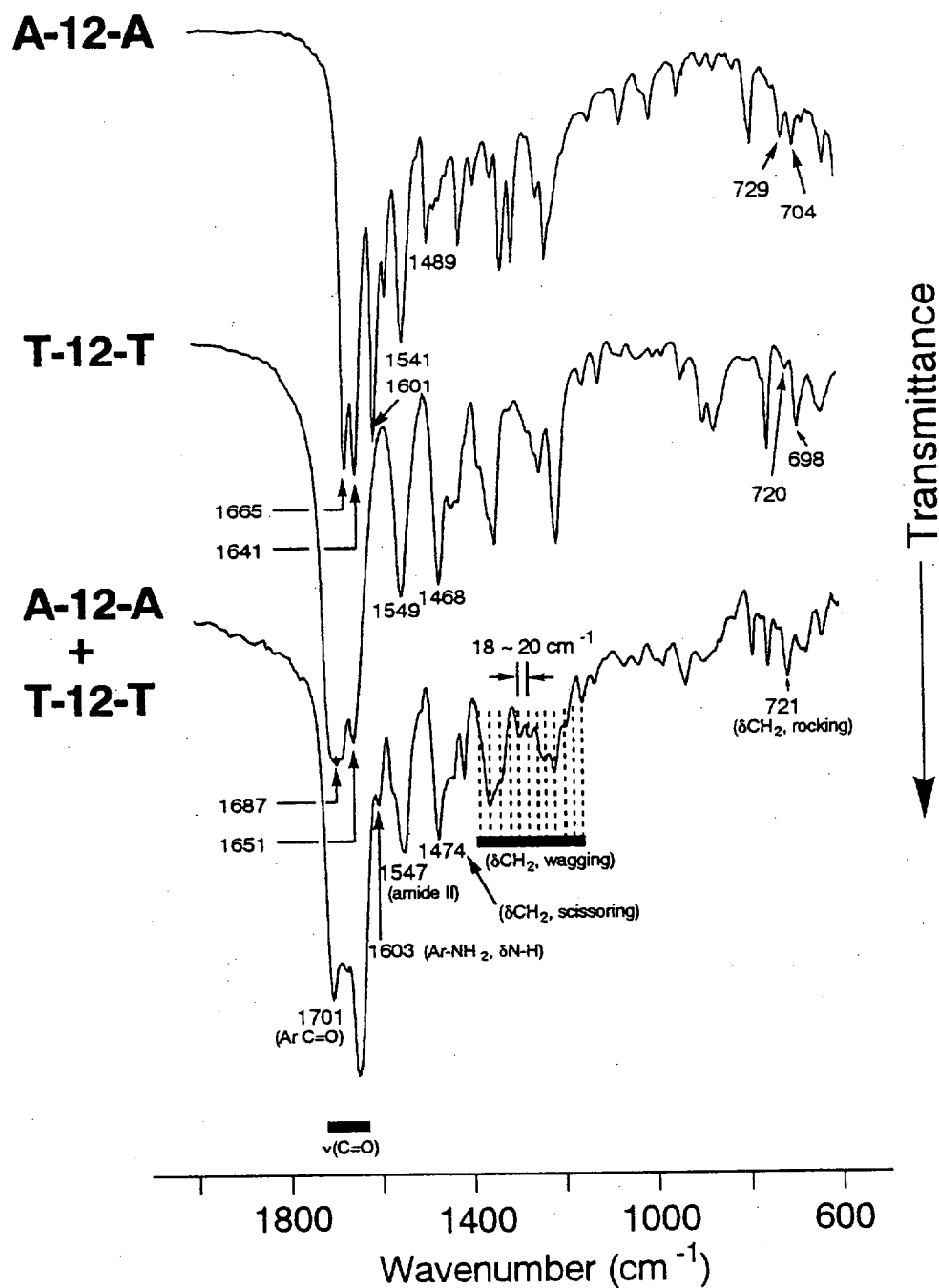


Fig. S-4. FT-IR Spectra (600 –1800 cm^{-1}) of A-12-A, T-12-T, and their 1:1 Self-(hetero)-Assembly

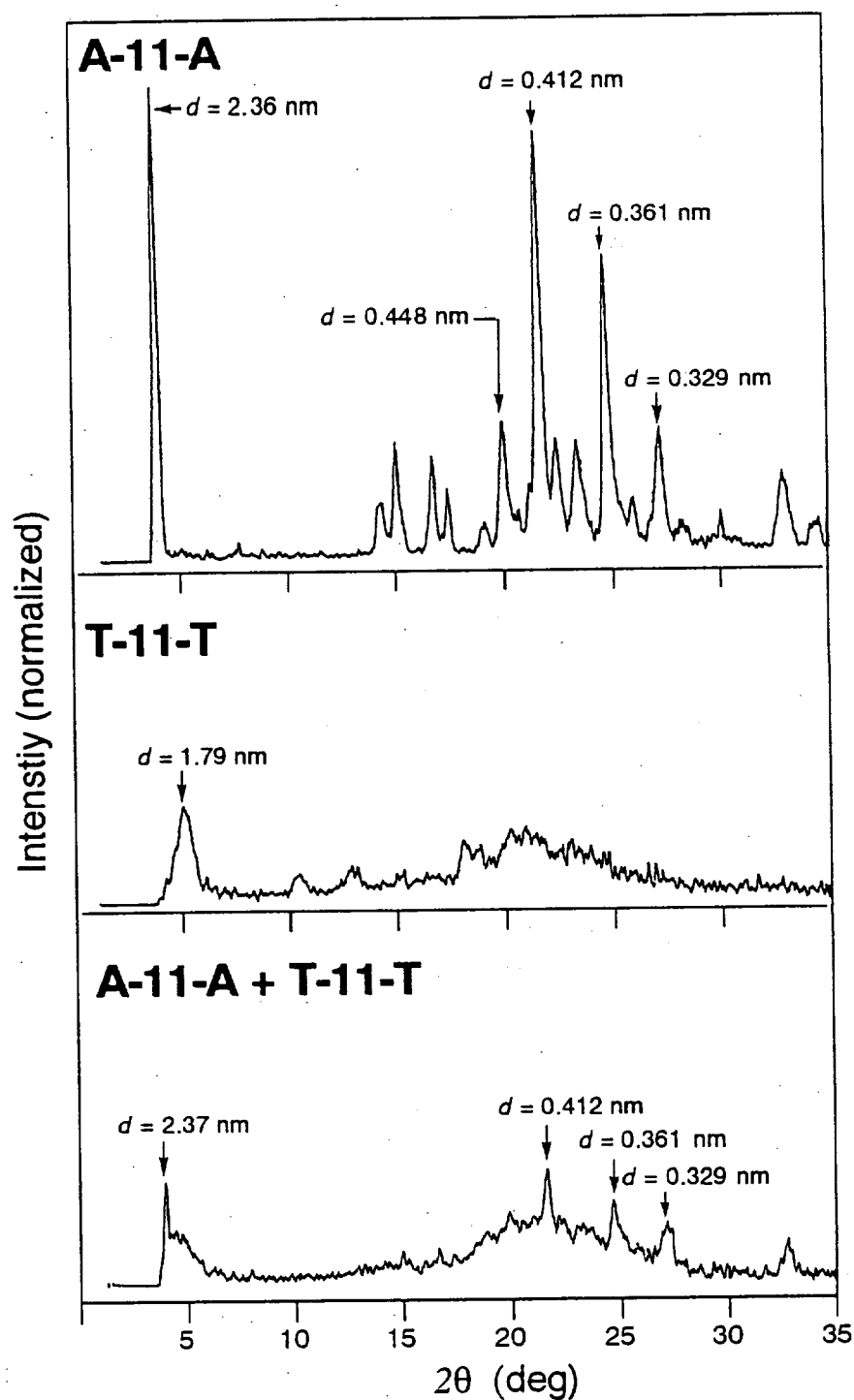


Fig. S-5. XRD diagrams of **A-11-A**, **T-11-T**, and their 1:1 Self-(hetero)-Assembly

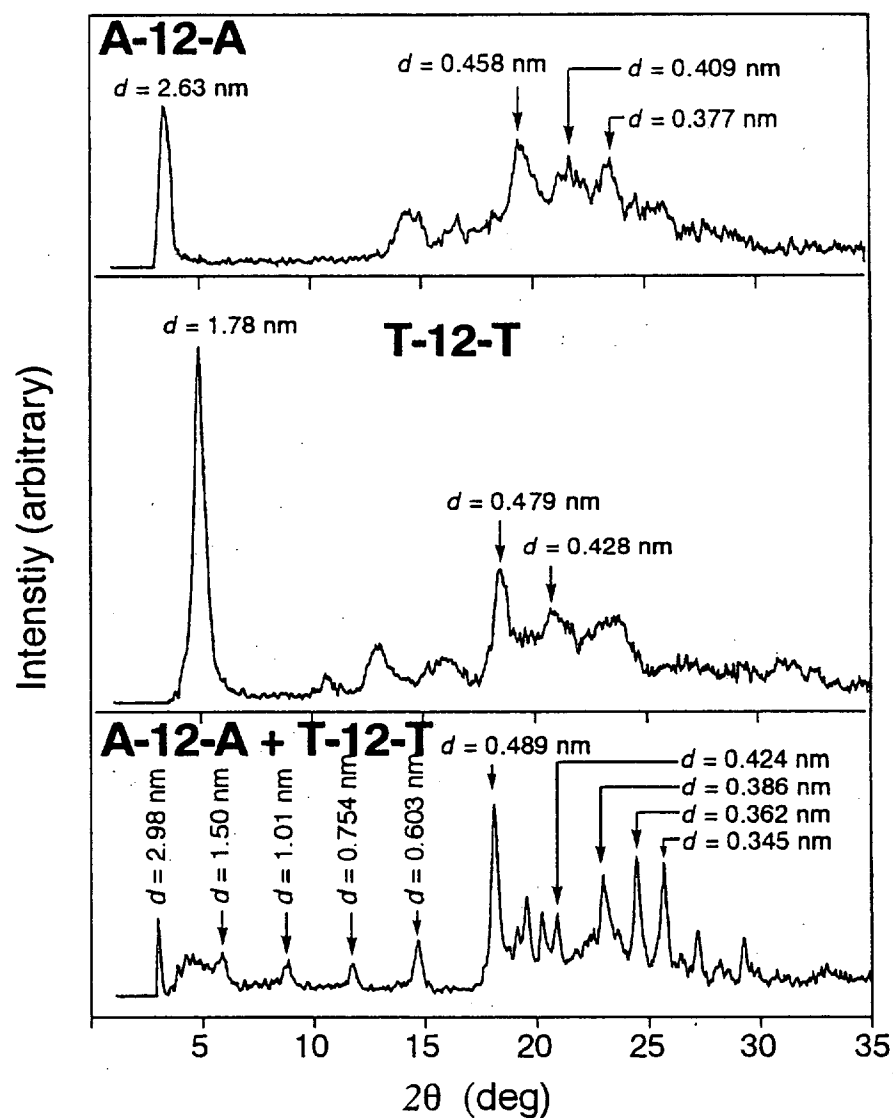


Fig. S-6. XRD diagrams of **A-12-A**, **T-12-T**, and their 1:1 Self-(hetero)-Assembly

Table S-7. Self-Assembling Morphologies^a of **T-n-T** (n = 11 and 12), **A-n-A** (n = 11 and 12), and the Equimolar Mixtures of **T-n-T/A-n-A** in Ethanol Solutions

n	Homo-Assembly		1:1 Hetero-Assembly
	T-n-T	A-n-A	T-n-T/A-n-A
11	amorphous gel (200.0–204.0) ^b	crystalline powder (224.0–230.0) ^b	amorphous gel (207.0–209.0) ^b
12	amorphous gel (211.8–214.4) ^b	crystalline powder (228.9–233.4) ^b	crystalline solid (228.2–235.1) ^b
			T-11-T/A-12-A
			crystalline solid
			T-12-T/A-11-A
			amorphous gel

^aObserved using light microscopy.

^bMelting point of the dried sample (°C).

Table S-8. FT-IR Absorption Bands (400-1800 cm^{-1} region) and Their Plausible Assignment for the Homo- and 1:1 Hetero-Assemblies Formed from T-11-T, T-12-T, A-11-A, and A-12-A^a

Assignment	Homo-Assembly			1:1 Hetero-Assembly		
	T-n-T		A-n-A	T-n-T/A-n-A		n = 12
	n = 11	n = 12		n = 11	n = 12	
Aromatic C=O str.	1698	1687	-	-	-	1701
Amide C=O str. (amide I)	1649	1651	1665	1665	1665	1647
NH ₂ scissoring plus ring C-H str.	-	-	1640 ^b	1640 ^b	1641 ^b	-
Amide N-H def. (amide II)	1547	1549	1603	1603	1601	1603
CH ₂ scissoring def.	1468	1468	1541	1541	1541	1547
CH ₂ rocking def.	721	720	1489 ^c	1489 ^c	1489 ^c	1474
			723	723	729	721
			704	704	704	

^aSelf-assembling was carried out in ethanol solutions.^bHydrogen-bonded NH₂ scissoring plus ring C-H stretching.^cBroad band.

Table S-9. X-ray Diffraction Data for the Homo- and 1:1 Hetero-Assemblies Formed from T-11-T, T-12-T, A-11-A, and A-12-A^a

	Homo-Assembly						1:1 Hetero-Assembly	
	T-n-T			A-n-A			T-n-T/A-n-A	
	n = 11	n = 12	n = 12	n = 11	n = 12	n = 12	n = 11	n = 12
Long-range ordering (nm) and the order of diffraction (parentheses)	1.79 (2)	1.78 (2)	1.78 (2)	2.36 (1)	2.63 (1)	2.37 (1)	2.98 (1)	1.50 (2)
	0.84 (4)	0.84 (4)					1.01 (3)	0.754 (4)
							0.603 (5)	
Long period (nm) ^b	3.58	3.56	2.36	2.63	2.37	3.01		
Short-range ordering (nm)	c	0.479	0.448	0.458	0.489			
		0.428	0.412	0.409	0.424			
		0.396	0.361	0.381	0.386			
		0.360	0.329		0.362			
Possible Packing Mode ^d	Dh	T//	O'⊥	O⊥	Dh	T//		

^aSelf-assembling was carried out in ethanol solutions. ^bObtained by reflection method. ^cNo well-defined peaks were obtained because of the broadening. ^dT//: triclinic subcell, Dh: distorted hexagonal packing, O⊥ and O'⊥: orthorhombic subcell

Table S-10. Molar Absorptivity (ϵ) and Absorption Maxima (λ_{max}) of T-11-T, T-12-T, A-11-A, A-12-A, and the Their 1:1 Mixtures in Ethanol Solutions and Their Hypochromicity.

Nucleobase Derivative	ϵ (λ_{max} , nm)	ϵ_{R} (λ_{max} , nm)	H(%)
T-11-T	10600 (269)	7300 ^a (269)	27 ^b
T-12-T	13800 (269)	7300 ^a (269)	5.0 ^b
A-11-A	14400 (261)	12600 ^c (261)	43 ^b
A-12-A	12500 (261)	12600 ^c (261)	51 ^b
1:1 T-11-T/A-11-A	13850 (264)	25000 ^d	45 ^e
1:1 T-12-T/A-12-A	15000 (264)	26300 ^f	43

^aThe molar absorptivity of thymine. ^bEvaluated according to the following equation: $H(\%) = 100 \times (1 - \epsilon/2\epsilon_{\text{R}})$. ^cThe molar absorptivity of adenine. ^dSummation of the molar absorptivity of T-11-T and A-11-A. ^eEvaluated according to the following equation: $H(\%) = 100 \times (1 - \epsilon/\epsilon_{\text{R}})$. ^fSummation of the molar absorptivity of T-12-T and A-12-A.

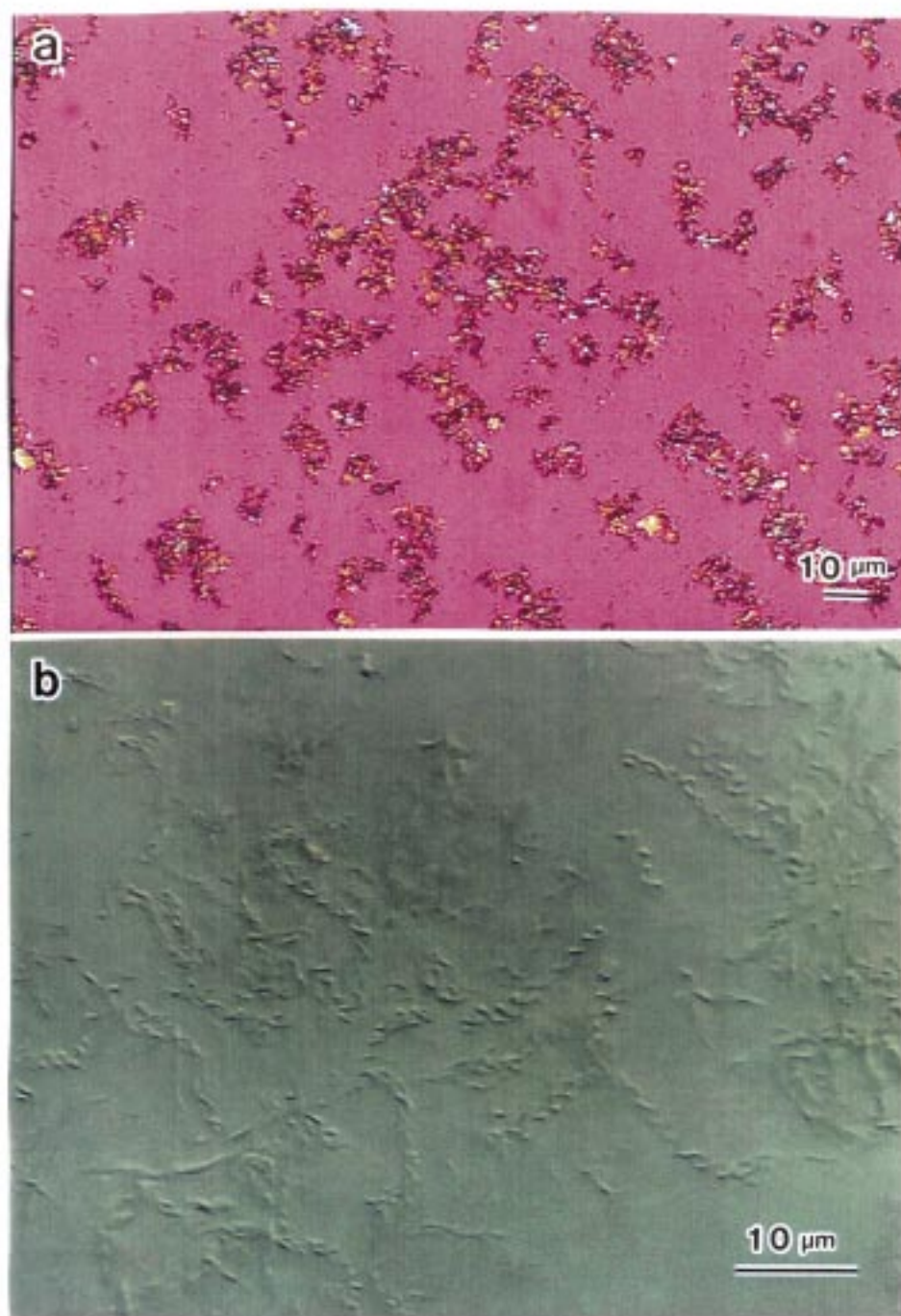


Fig. S-12. (a) Microcrystalline Solids Formed from **A-10-A** and (b) a Swarm of Double-Helical Ropes from **T-10-T** Observed Using (a) Polarized and (b) Dark-Field Light Microscopy (at 25°C in 10%-Ethanolc Aqueous Solutions).

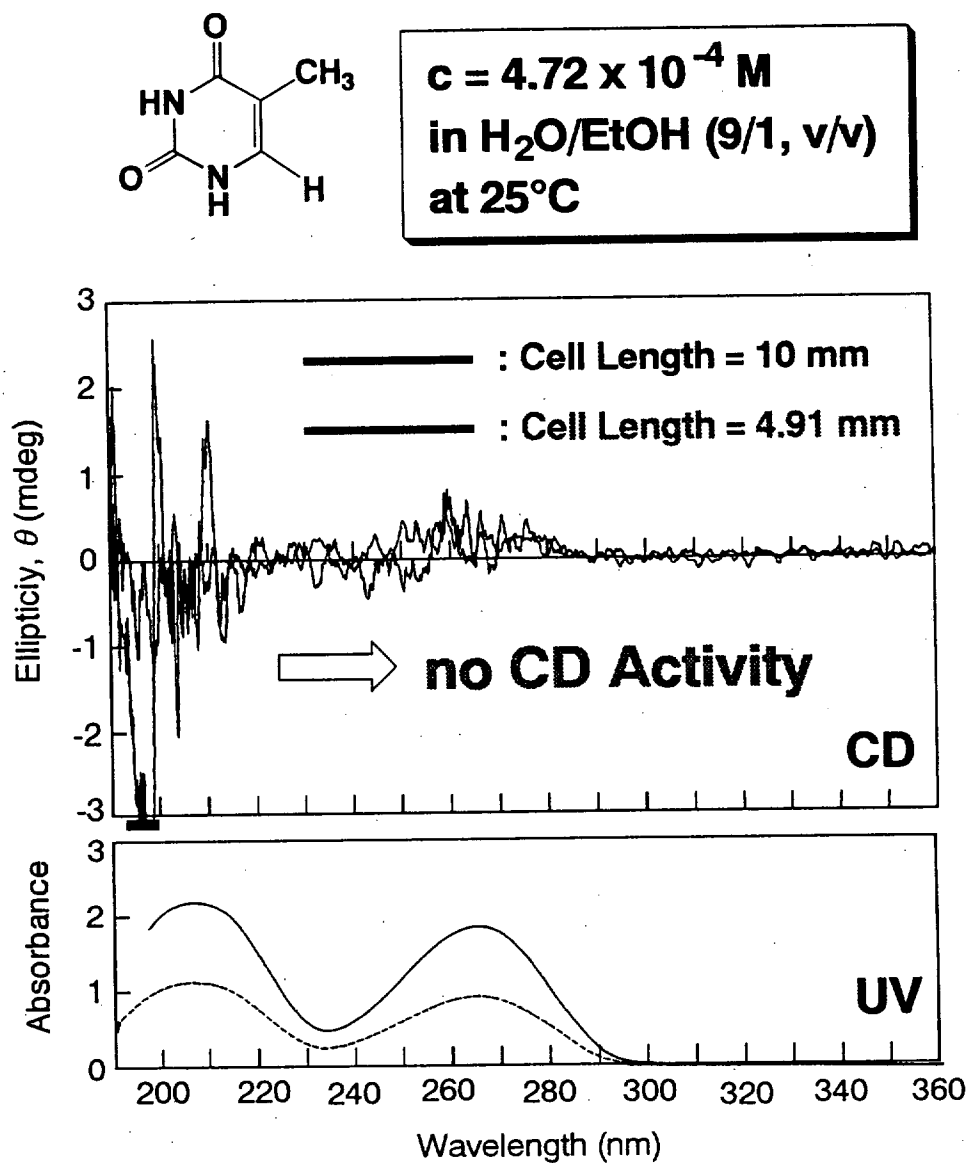
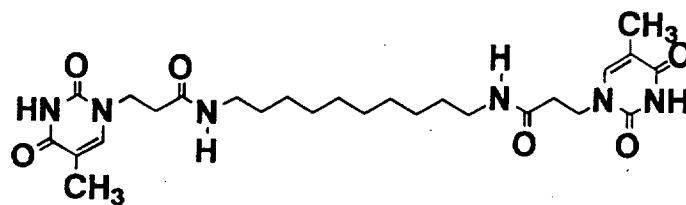


Fig. S-13. Circular Dichroism and Ultraviolet Spectra of Thymine (JASCO J-725)



$$c = 9.4 \times 10^{-5} \text{ M}$$

in H₂O/EtOH (9/1, v/v)

25°C $\Rightarrow$ 70°C $\Rightarrow$ 25°C $\Rightarrow$ 70°C $\Rightarrow$ 25°C

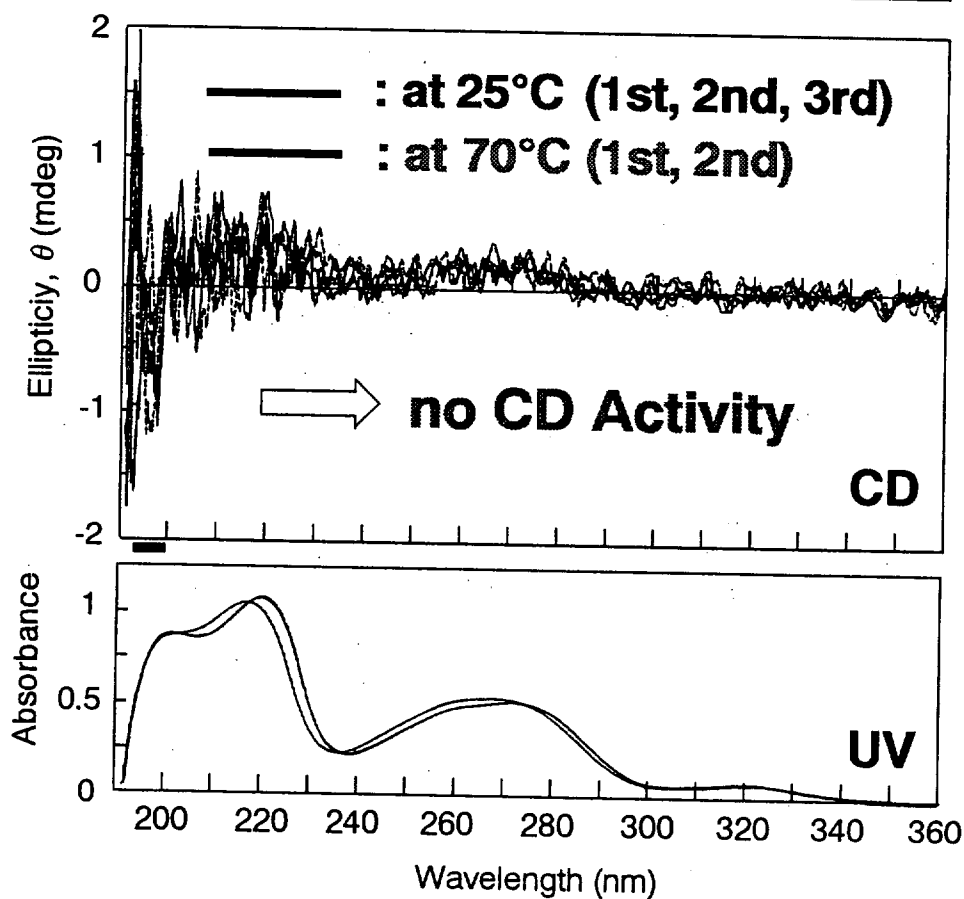


Fig. S-14. Circular Dichroism and Ultraviolet Spectra of Thymine (JASCO J-725)