

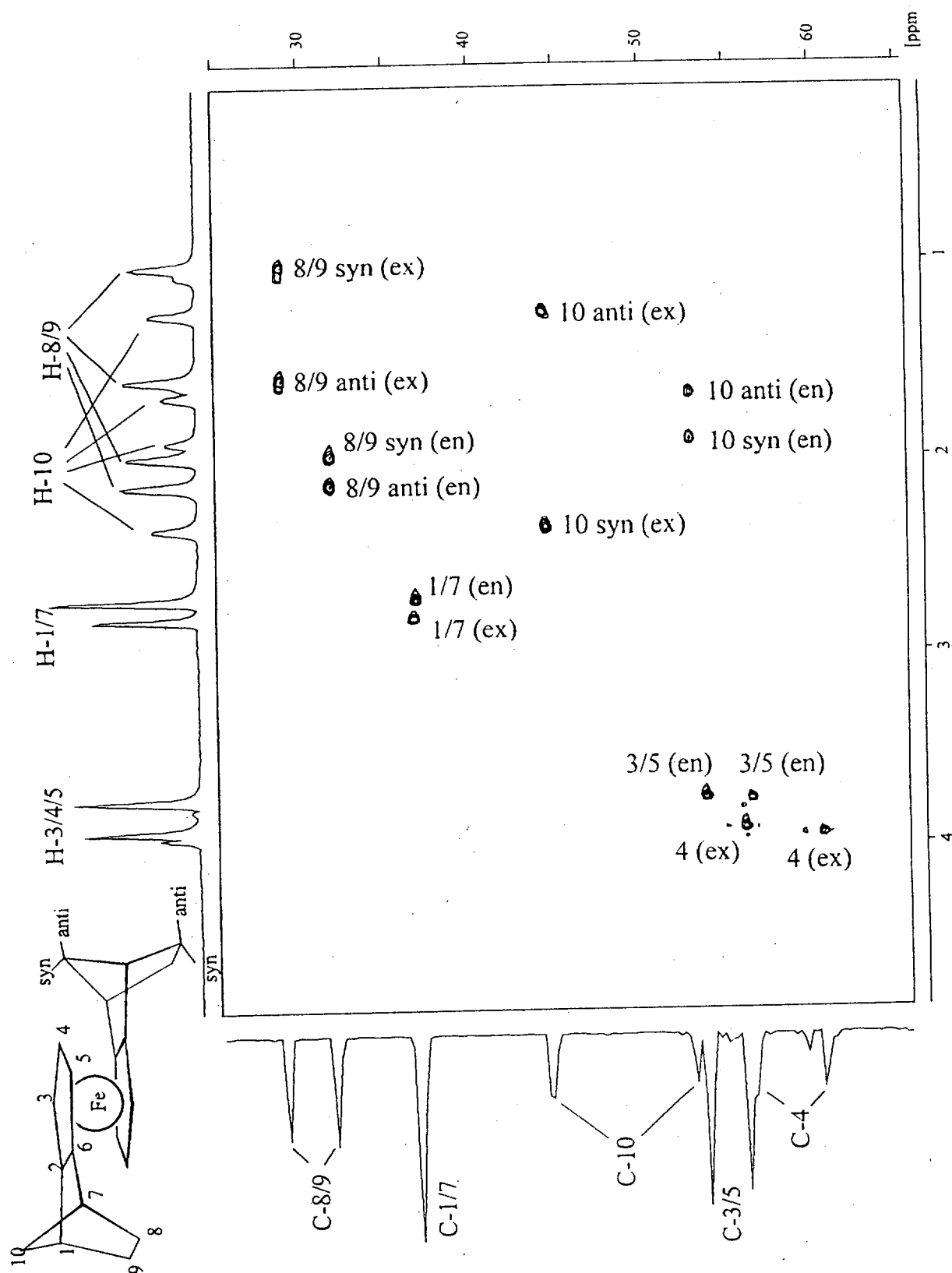


Figure S1. HMQC NMR spectrum of **3b** (600.14 MHz, C_6D_6 , 300 K). (ex) and (en) = *exo*- and *endo*-bound ligand, respectively.

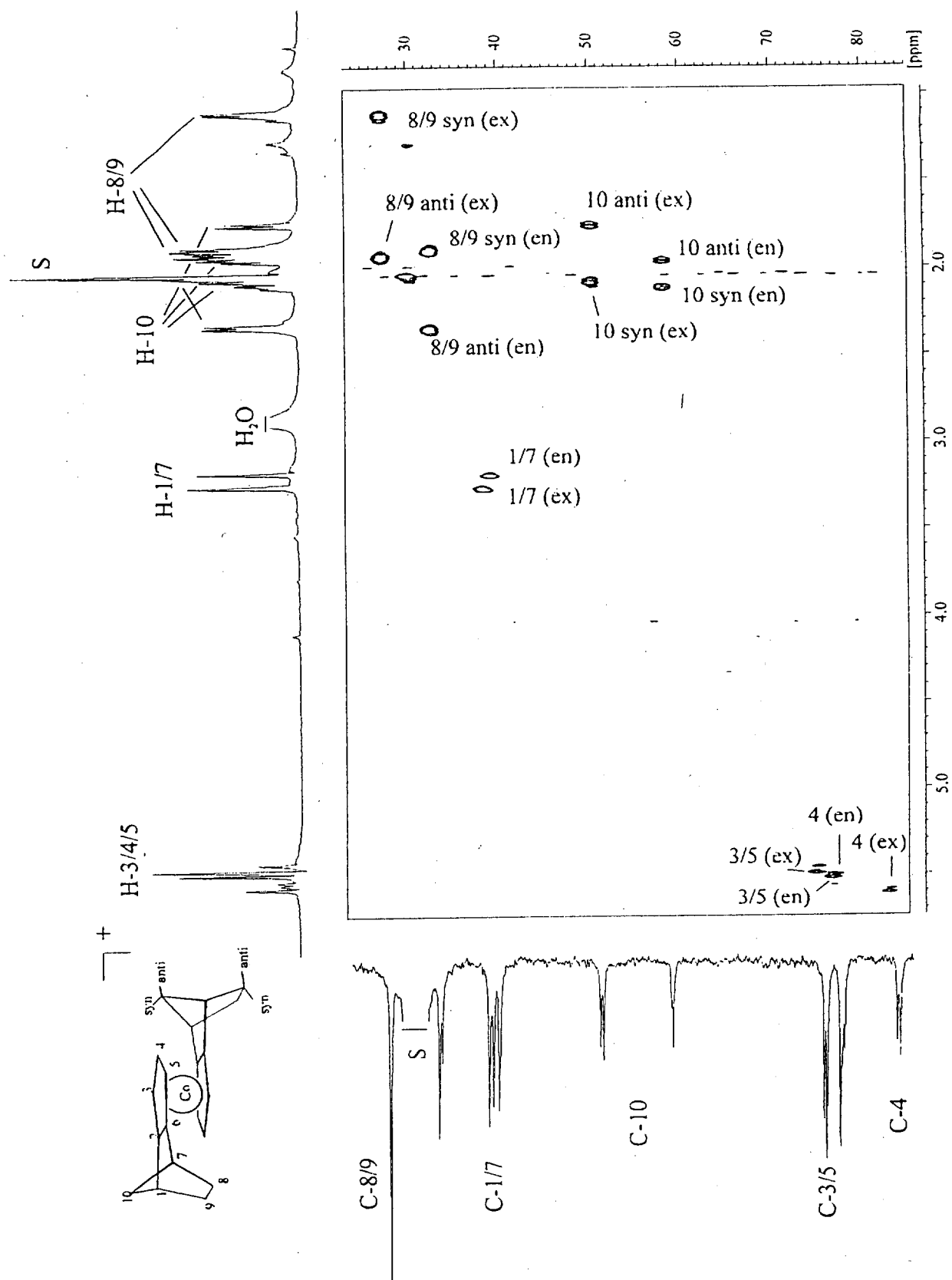


Figure S2. HMQC NMR spectrum of $6a^+ PF_6^-$ (600.14 MHz, $acetone-d_6$, 300 K). S = solvent, (ex) and (en) = *exo*- and *endo*-bound ligand, respectively.

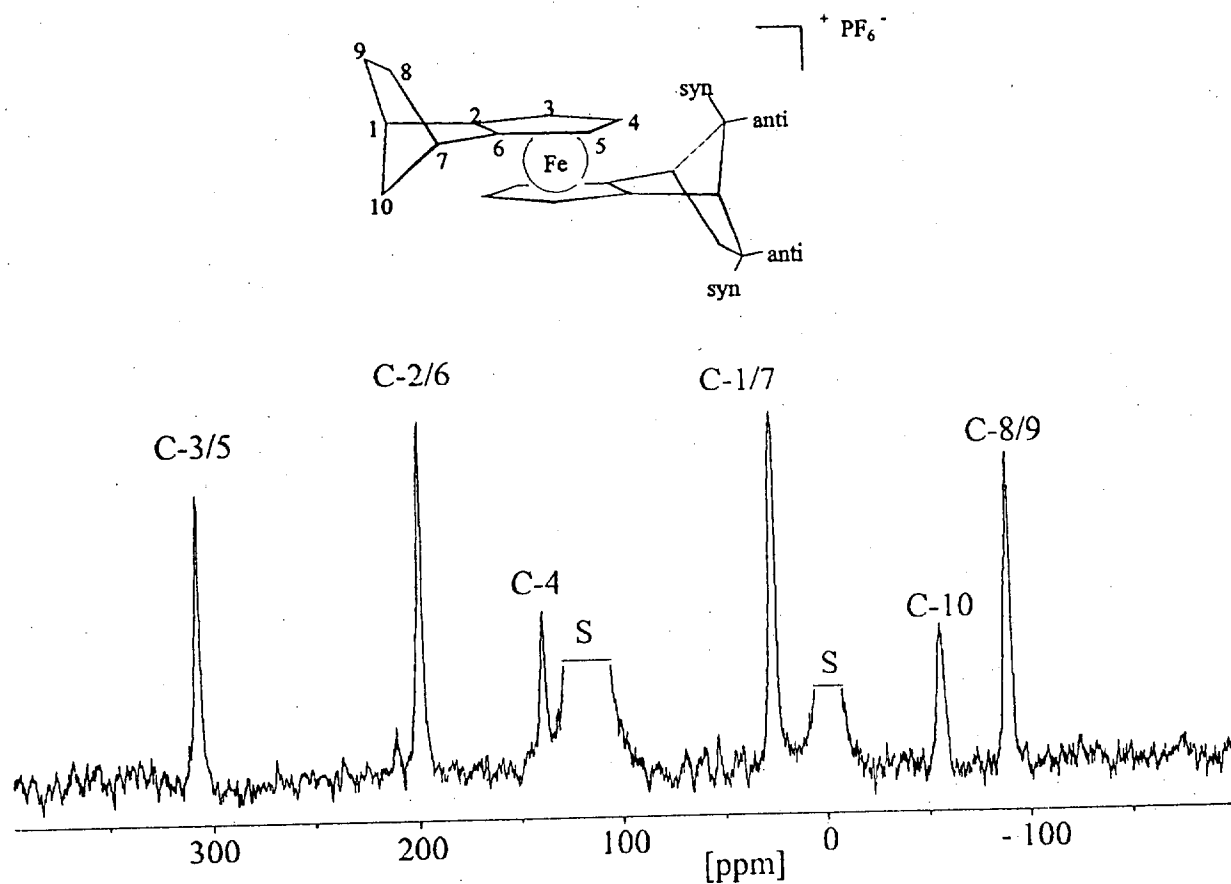


Figure S3. ^{13}C NMR spectrum of $3a^+\text{PF}_6^-$ in CD_3CN at 305 K. S = solvent.

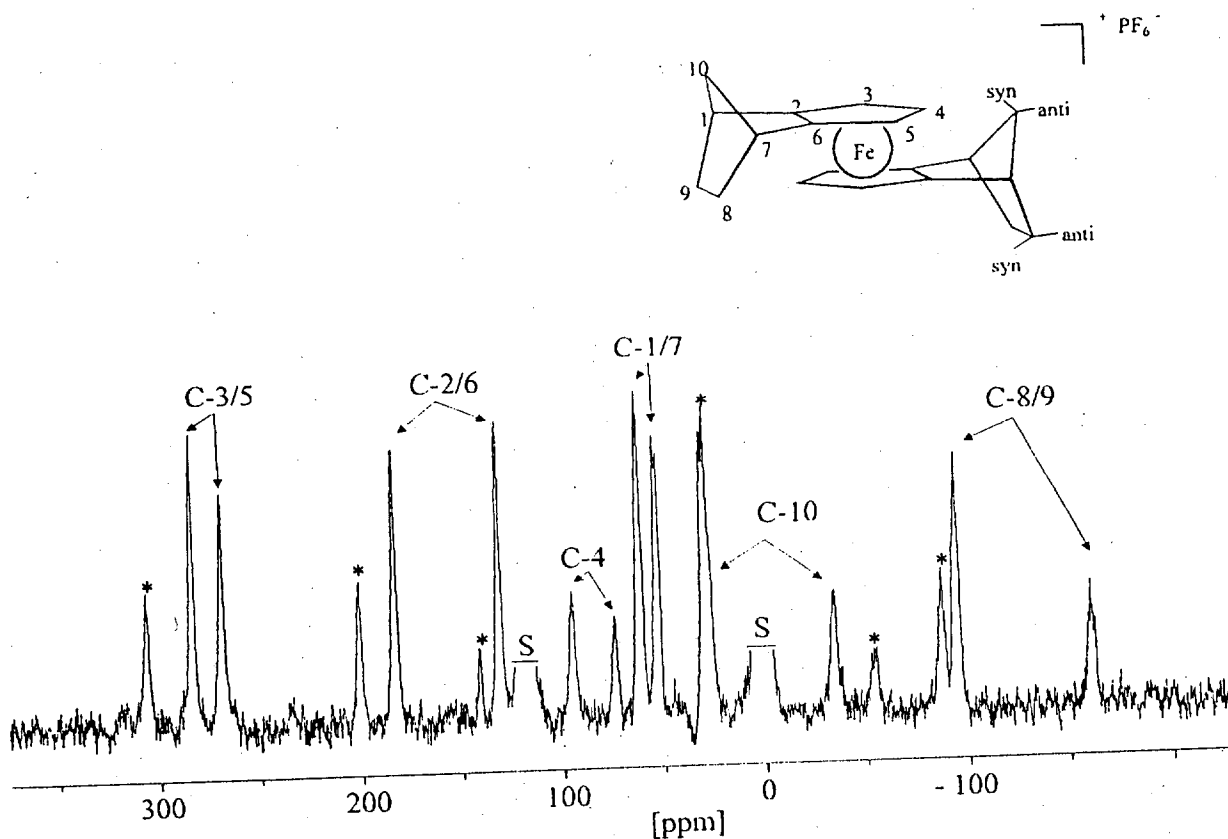


Figure S4: ^{13}C NMR spectrum of $3b^+\text{PF}_6^-$ in CD_3CN at 305 K contaminated with $3a^+\text{PF}_6^-$ (starred signals). S = solvent.

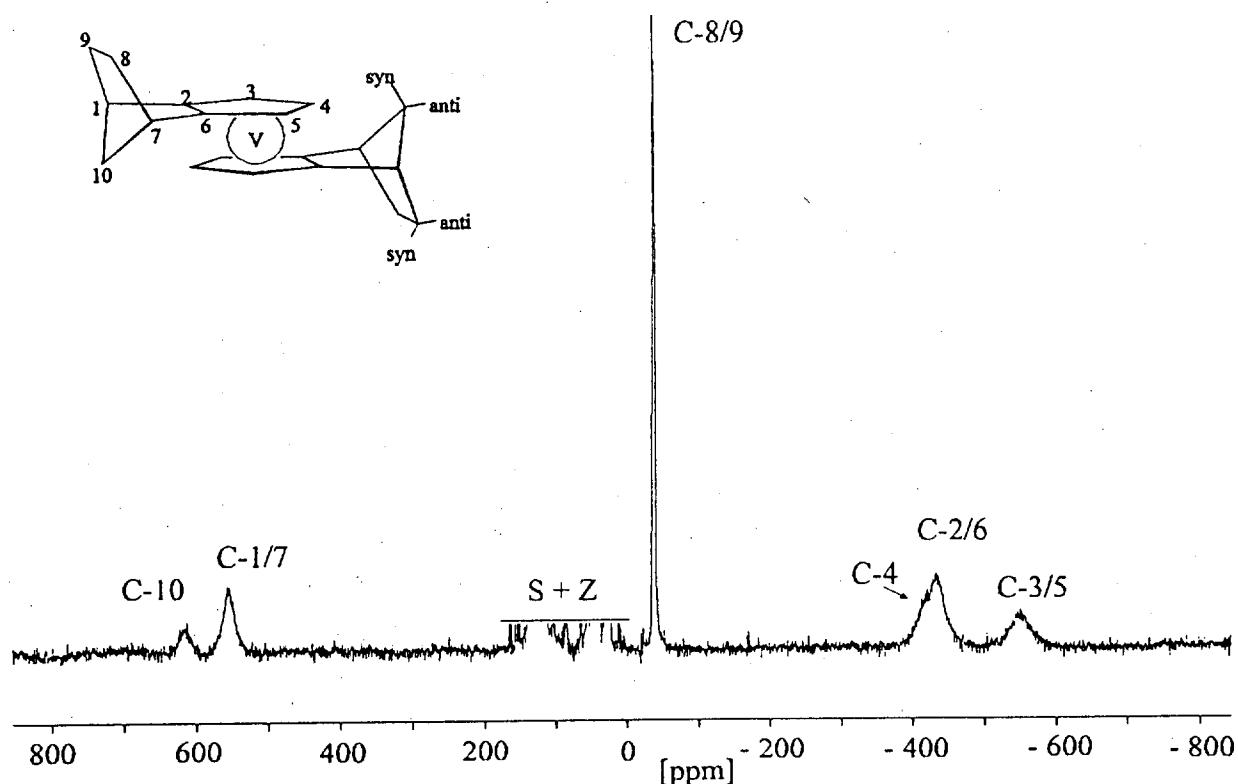


Figure S5. ^{13}C NMR spectrum of **4a** in $\text{toluene-}d_8$ at 345 K. S = solvent, Z = diamagnetic impurities.

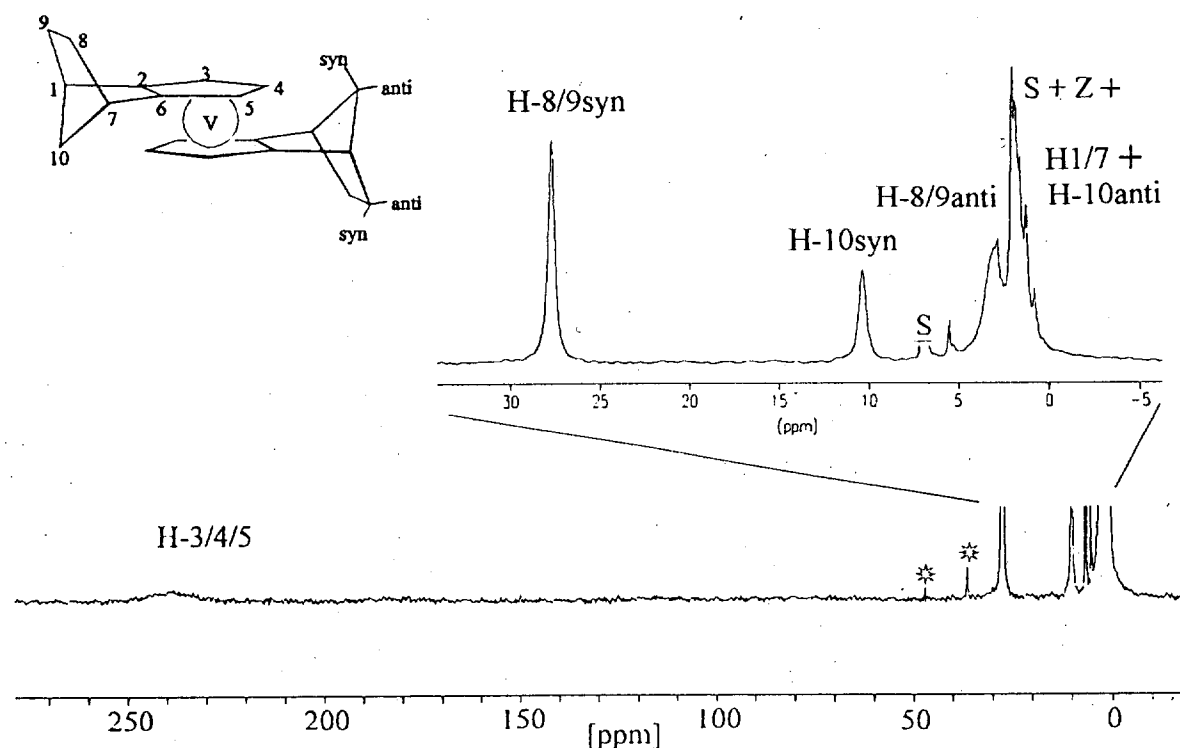


Figure S6. ^1H NMR spectrum of **4a** in $\text{toluene-}d_8$ at 345 K. S = solvent, Z and * = dia- and paramagnetic impurities, respectively.

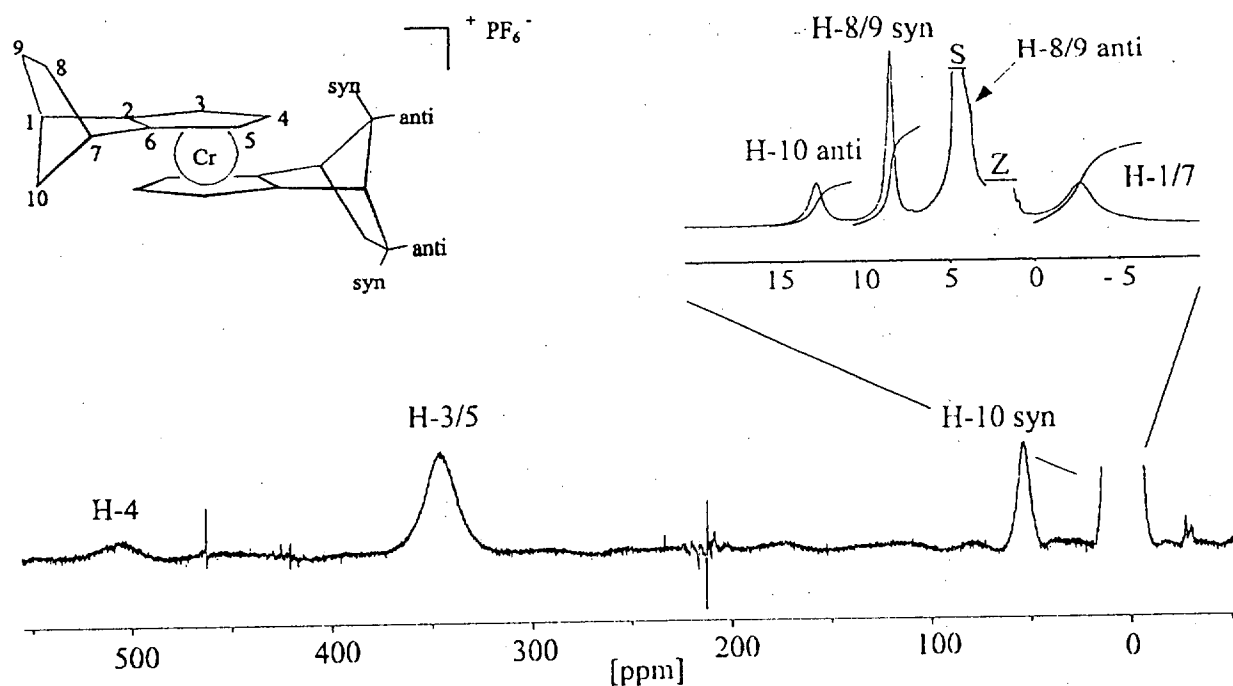


Figure S7. ^1H NMR spectrum of $5a^+\text{PF}_6^-$ in CD_3NO_2 at 297 K. S = solvent, Z = diamagnetic impurities.

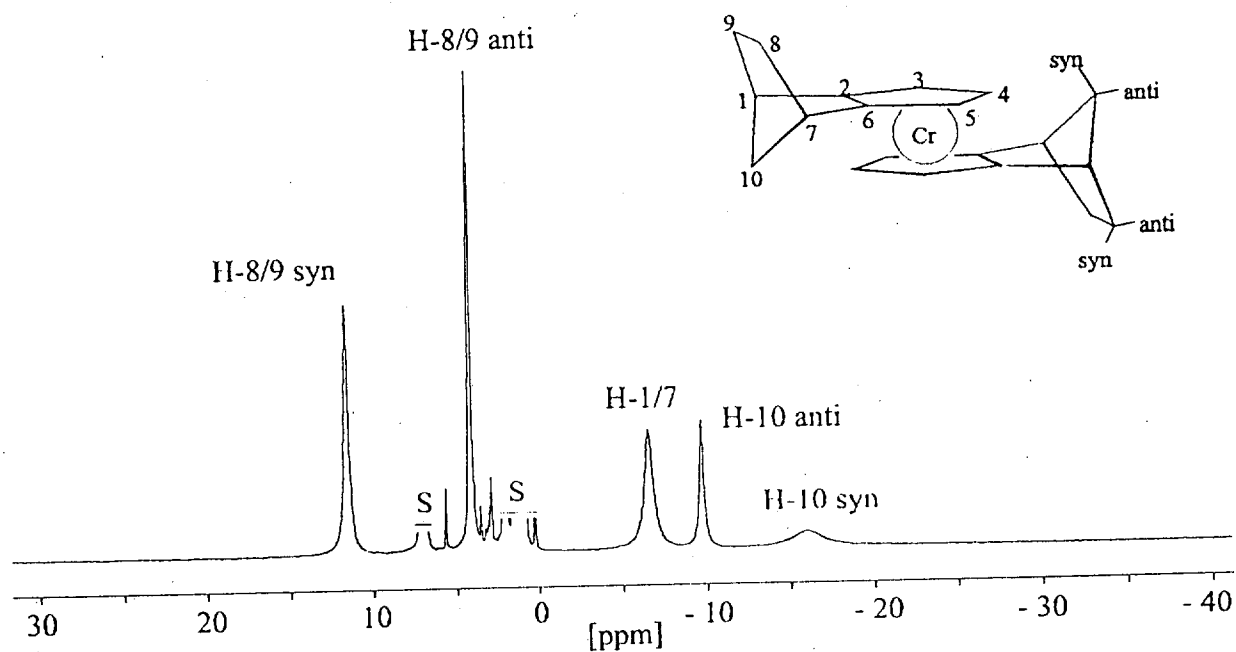


Figure S8. ^1H NMR spectrum of $5a$ in $\text{toluene-}d_8$ at 305 K. S = solvent.

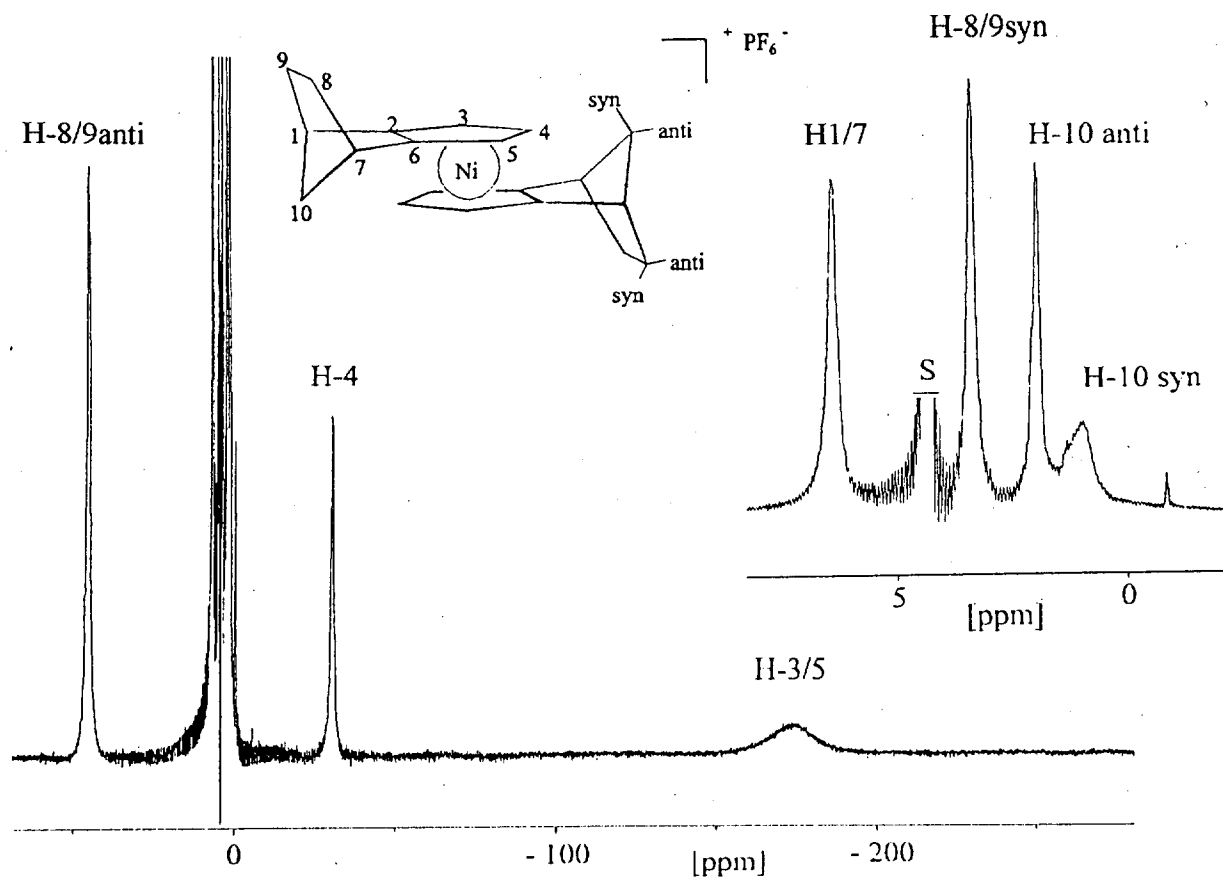


Figure S9. ^1H NMR spectrum of $7a^+ \text{PF}_6^-$ in CD_3NO_2 at 305 K. S = solvent.

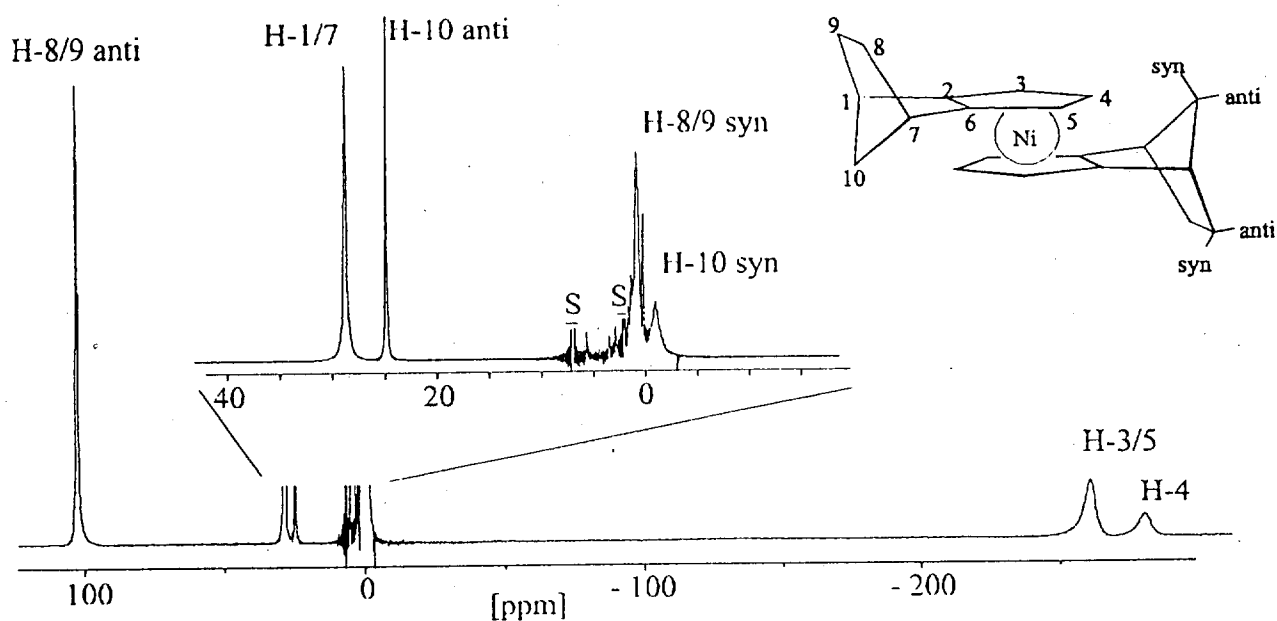


Figure S10. ^1H NMR spectrum of $7a$ in $\text{toluene-}d_8$ at 298 K. S = solvent

Calculation of the Dipolar Signal Shifts

The calculations are based on the work of Kurland and McGarvey.²⁴ For a given spin multiplicity and temperature T the dipolar shift δ^{dip}_T depends on the g factors $g_{\parallel}$ and $g_{\perp}$ and on the geometric factor of the nucleus under study, $G(r, \theta) = (3 \cos^2 \theta - 1)/r^3$, where r is the length of metal-nucleus vector and θ is the angle between that vector and the magnetic axis which is assumed to coincide with the Cp-M-Cp axis. The geometric factors were taken from the positional parameters which were obtained as a by-product of the EHMO calculations and which are listed in Table S1. When more than one unpaired electron is present, δ^{dip} also depends on the zero-field splitting D . The latter term is included in the proportionality constant of the formula for the various cases:

3a⁺PF₆⁻ and 3b⁺PF₆⁻ :

$S = 1/2$, $g_{\parallel} = 3.9561$, $g_{\perp} = 2.0560$, $T = 305$ K

$$\delta^{\text{dip}}_{305} = 1.9446 \times 10^{-3} \quad G(r, \theta) \quad (\text{S1})$$

7a⁺PF₆⁻ :

$S = 1/2$, $g_{\parallel} = 1.768\text{--}1.878^{\text{S1}}$, $g_{\perp} = 1.865\text{--}2.024^{\text{S1}}$, $T = 305$ K;

within the g -factor limits one gets

$$\delta^{\text{dip}}_{305} = -1.0526 \times 10^{-4} \quad G(r, \theta) \text{ and} \quad (\text{S2a})$$

$$\delta^{\text{dip}}_{305} = -6.5272 \times 10^{-5} \quad G(r, \theta) \quad (\text{S2b})$$

(S1) Zoller, L.; Moser, E.; Ammeter, J.H. *J. Phys. Chem.* **1986**, *90*, 6632.

7a:

$$S = 1, g_{\parallel} = 2.0023^{S_2}, g_{\perp} = 2.11^{S_2}, D = 33.6 \text{ cm}^{-1 S_2},$$

$$T = 305 \text{ K and } 379 \text{ K}$$

$$\delta_{305}^{\text{dip}} = -3.507 \times 10^{-4} \quad G(r, \theta) \quad (\text{S3})$$

$$\delta_{379}^{\text{dip}} = -2.587 \times 10^{-4} \quad G(r, \theta) \quad (\text{S4})$$

5a:

$$S = 1, g_{\parallel} = 2.33^{S_3}, g_{\perp} = 2.77^{S_3}, D = 15.1 \text{ cm}^{-1 S_3}, T = 305 \text{ K}$$

$$\delta_{305}^{\text{dip}} = -9.191 \times 10^{-4} \quad G(r, \theta) \quad (\text{S5})$$

5a⁺PF₆⁻:

$$S = 3/2, g_{\parallel} = 1.911, g_{\perp} = 1.7985, D = 1.461 \text{ cm}^{-1}, T = 297 \text{ and } 305 \text{ K}$$

$$\delta_{279}^{\text{dip}} = 3.387 \times 10^{-4} \quad G(r, \theta) \quad (\text{S6})$$

$$\delta_{305}^{\text{dip}} = 3.304 \times 10^{-4} \quad G(r, \theta) \quad (\text{S7})$$

4a:

$$S = 3/2, g_{\parallel} = 1.9790, g_{\perp} = 1.8627, D = 2.7 \text{ cm}^{-1 S_4}, T = 345 \text{ and } 365 \text{ K}$$

$$\delta_{345}^{\text{dip}} = 2.979 \times 10^{-4} \quad G(r, \theta) \quad (\text{S8})$$

$$\delta_{365}^{\text{dip}} = 2.835 \times 10^{-4} \quad G(r, \theta) \quad (\text{S9})$$

(S2) Baltzer, P.; Furrer, A.; Hulliger, J.; Stebler, A. *Inorg. Chem.* **1988**, 27, 1543.

(S3) Worst case assumed from the data given in Desai V.P.; König, E.; Kanellakopulos, B. *J. Chem. Phys.* **1983**, 78, 6299. König E.; Schnakig, R.; Kremer, S.; Kanellakopulos, B.; Klenze, R. *Chem. Phys.* **1978**, 27, 331.

(S4) Prins, R.; van Voorst, J.D.W. *Chem. Phys.* **1968**, 49, 4665.

Besides $G(r, \theta)$ Table S1 contains the paramagnetic signal shifts δ^{para}_T at the experimental temperature T (obtained after subtracting the signal shifts of the diamagnetic standards from the experimental signal shifts δ^{exp} of Table 2), and the dipolar signal shift δ^{dip}_T (obtained from eqs S1-S9). Subtraction of δ^{dip}_T from δ^{para}_T and conversion to the standard temperature of 298 K according to the Curie law gave the contact shifts δ^{con} listed in Table 2).

Table S1 Relevant Data^a for the Calculation of the Dipolar Signal Shifts

nucleus and position ^b	4a		3a⁺ / 3b⁺, exo-ligand		
	$\delta_{\text{T}}^{\text{para c}}$	$G(r, \theta)$	$\delta_{\text{T}}^{\text{dip c}}$	$\delta_{305}^{\text{para}}$	$\delta_{305}^{\text{dip}}$
H3/5	231.2	0.0134	3.8	22.6 ^d	6.7
H4	231.0	0.0131	3.7	24.1 ^d	10.9
H1/7	-1.8	-0.0041	-1.2	14.8	-1.9
H8/9-syn	26.7	0.0100	2.8	23.5	23.9
H8/9-anti	1.4	0.0034	1.0	6.7	5.8
H10-syn	8.1	-0.0041	-8.7	-63.9	-76.2
H10-anti	-0.4	-0.0305	-1.8	-17.5	-12.9
C2/6	-527.0	0.0971	28.9	84.0	191.8
C3/5	-607.5	0.0940	28.0	232.8	213.3
C4	-484.8	0.0988	29.5	56.8	232.8
C1/7	516.7	0.0048	1.4	-10.8	1.3
C8/9	-67.1	0.0086	2.6	-115.6	17.2
C10	569.2	-0.0117	-3.5	5.4	-28.8

Table S1 continued

nucleus and position ^b	3b ⁺ , endo-ligand				5a			
	$\delta_{305}^{\text{para}}$	$G(r, \theta)$	$\delta_{305}^{\text{dip}}$	$\delta_{305}^{\text{para}}$	$G(r, \theta)$	$\delta_{305}^{\text{dip}}$	$\delta_{305}^{\text{para}}$	$\delta_{305}^{\text{dip}}$
H3/5	23-26 ^d	0.0034	6.7	294.5	0.0115	-10.6		
H4	23-26 ^d	0.0056	10.9	399.1	0.0110	-10.1		
H1/7	-14.0	-0.0010	-1.9	-9.4	-0.0052	4.8		
H8/9-syn	-42.2	-0.0385	-74.8	10.5	0.0103	-9.5		
H8/9-anti	19.1	-0.0111	-21.6	2.4	0.0033	-3.0		
H10-syn	23.6	0.0134	26.0	-18.4	-0.0316	29.0		
H10-anti	5.6	0.0001	0.2	-11.2	-0.0071	6.5		
C2/6	25.2	0.0986	191.8	-474.4	0.1025	-94.2		
C3/5	216.2	0.1097	213.3	-262.9	0.0989	-90.9		
C4	18.5	0.1197	232.8	-585.8	0.1041	-95.7		
C1/7	25.9	0.0007	1.3	384.7	0.0034	-3.1		
C8/9	-191.4	0.0241	46.8	-181.7	0.0086	-7.9		
C10	25.2	0.0065	12.6	105.8	-0.0135	12.4		

Table S1 continued

nucleus and position ^b	5a ⁺			7a		
	$\delta_T^{\text{para e}}$	$G(r, \theta)$	$\delta_T^{\text{dip e}}$	$\delta_T^{\text{para f}}$	$G(r, \theta)$	$\delta_T^{\text{dip f}}$
H3/5	334.5	0.0115	3.9	-264.5	0.0106	-3.7
H4	504.4	0.0114	3.9	-284.0	0.0100	-3.5
H1/7	-6.1	-0.0051	-1.7	26.0	-0.0057	2.0
H8/9-syn	2.8	0.0127	4.3	-0.2	0.0104	-3.7
H8/9-anti	6.5	0.0039	1.3	100.8	0.0033	-1.2
H10-syn	51.5	-0.0318	-10.8	-3.3	-0.0322	11.3
H10-anti	11.1	-0.0066	-2.2	23.5	-0.0073	2.6
C2/6	-414.6	0.1029	34.0	1075 ^g	0.1044	-27.0
C3/5	-328.8	0.1008	33.3	1075 ^g	0.1020	-26.4
C4	-451.7	0.1064	36.0	1075 ^g	0.1060	-27.4
C1/7	414.3	0.0048	1.6	-366.5	0.0028	-1.0
C8/9	-418.8	0.0098	3.2	1372.7	0.0086	-3.0
C10	271.7	0.0114	3.8	508.3	-0.0142	5.0

Table S1 continued

nucleus and position ^b	7a ⁺		
	$\delta^{\text{para}}_{305}$	$G(r, \theta)$	$\delta^{\text{dip}}_{305}$ ^h
H3/5	-179.8	0.0086	-0.8
H4	-36.4	0.0081	-0.7
H1/7	3.2	-0.0064	0.6
H8/9-syn	2.3	0.0131	-1.1
H8/9-anti	43.3	0.0039	-0.3
H10-syn	-1.3	-0.0330	2.9
H10-anti	0.2	-0.0072	0.6
C2/6	239.4	0.1079	-9.4
C3/5	477.2	0.1052	-9.1
C4	-161.2	0.1111	-9.7
C1/7	-115.7	0.0032	-0.3
C8/9	116.9	-0.0133	0.9
C10	17.5	0.0099	-1.2

^a Shifts in ppm, $G(r, \theta)$ in $\text{\AA}^{-3}$.^b Numbering see Figure 5. ^c ^1H and ^{13}C NMR data at 365 and 345 K, respectively. ^d For $3b^+\text{PF}_6^-$ a broad feature between 27 and 30 ppm was detected. ^e ^1H and ^{13}C NMR data at 297 and 305 K, respectively. ^f ^1H and ^{13}C NMR data at 298 and 379 K, respectively. ^g Broad feature. ^h Calculated with a mean value of eqs S2a and S2b.