

## Terms & Conditions

Electronic Supporting Information files are available without a subscription to ACS Web Editions. The American Chemical Society holds a copyright ownership interest in any copyrightable Supporting Information. Files available from the ACS website may be downloaded for personal use only. Users are not otherwise permitted to reproduce, republish, redistribute, or sell any Supporting Information from the ACS website, either in whole or in part, in either machine-readable form or any other form without permission from the American Chemical Society. For permission to reproduce, republish and redistribute this material, requesters must process their own requests via the RightsLink permission system. Information about how to use the RightsLink permission system can be found at <http://pubs.acs.org/page/copyright/permissions.html>



ACS Publications

MOST TRUSTED. MOST CITED. MOST READ.

Copyright © 1997 American Chemical Society

**Supporting information to:**

**Structures and Dynamics of Ln(III) Complexes of Sugar-Based DTPA-bis(amides) in Aqueous Solution; A Multinuclear NMR Study.**

Hendrik Lammers, Frédéric Maton, Dirk Pubanz, Martijn W. van Laren, Herman van Bekkum, André E. Merbach, Robert N. Muller, and Joop A. Peters

**Table S1:** Measured  $^{13}\text{C}$  chemical shifts of 0.10 M [Nd(DTPA-bisamide)(H<sub>2</sub>O)] complexes in D<sub>2</sub>O at 80 °C and 100.6 MHz.

**Table S2A-C:** Measured longitudinal and transverse relaxation times and chemical shifts of the acidified water reference ( $T_{1A}$ ,  $T_{2A}$  and  $\Delta\omega_A$ ) and of the paramagnetic solutions ( $T_1$ ,  $T_2$  and  $\Delta\omega$ ), at molar ratios of bound water  $P_m$ . The calculated values of reduced relaxation rates and chemical shifts ( $1/T_{1r}$ ,  $T_2r$  and  $\Delta\omega_r$ ) of the variable temperature study on the [Gd(DTPA-bisamide)(H<sub>2</sub>O)] complexes are also tabulated. The variable temperature  $^{17}\text{O}$  NMR measurements were mainly carried out on a Varian VXR-400 S spectrometer. Several measurements, denoted with an asterisk, were recorded using a Bruker AM-400 spectrometer. All measurements were carried out at 9.4 T.

**Table S3A-C:** Measured transverse relaxation times of the [Gd(DTPA-bisamide)(H<sub>2</sub>O)] complexes ( $T_2$ ) and of the acidified water (0.9 M perchloric acid) reference ( $1/T_{2A}$ ) and the calculated reduced relaxation rates ( $1/T_{2r}$ ) at variable pressure. The reduced relaxation times were calculated taking into account an activation volume of +1.09 cm<sup>3</sup>mol<sup>-1</sup> for the reference. The measurements were recorded on a Bruker AM-400 spectrometer at 9.4 T.

**Table S4:** Comparison of water exchange rates at temperature T,  $k_{ex}^T$  (VT) as calculated from eq 17 using the parameters obtained from the variable temperature data, with the corresponding water exchange rates at zero pressure and temperature T,  $k_0^T$  (VP), as obtained from simultaneous least squares fits of the variable pressure data using eqs 8, 12, 15, 19, 21, 22 and 27.

**Table S5A-C:** Longitudinal proton relaxation rates ( $T_1$ ) for the [Gd(DTPA-bisamide)(H<sub>2</sub>O)] complexes at several temperatures T. The data at 0.01 MHz to 50 MHz were obtained on an IBM Research Relaxometer using the field cycling method. The spin-lattice proton relaxation rates at 200 MHz and 400 MHz were measured on a Bruker MSL-200 spectrometer and a Varian VXR-400 S spectrometer, respectively.

**Figure S1:** NMRD profiles for the complexes [Gd(DTPA-BPDA)(H<sub>2</sub>O)], [Gd(DTPA-BGLUCA)(H<sub>2</sub>O)] and [Gd(DTPA-BENGALAA)(H<sub>2</sub>O)]. The curves are calculated with the parameters obtained from the simultaneous least-squares fittings (see text). The bottom lines in each box represent the outer sphere contribution to the relaxivity.

**Table S1.**

Assign.	DTPA-BEA	DTPA-BPDA	DTPA-BLUCA	DTPA-BENGALAA	DTPA-BPA*
C=O	186.54	186.54	187.04	187.15	186.38
	182.75	183.32	183.23	183.04	182.95
	180.18	176.55	176.46	177.65	177.04
	178.97	173.48	174.70	176.01	175.99
	172.78	171.24	171.35	173.54	172.68
	170.56	168.29	168.07	170.70	170.84
	167.70	164.63	164.06	167.48	168.07
	164.04				164.70
CH <sub>3</sub> COOH	70.20	67.55	67.69	62.39	70.67
	66.78	62.58	66.15	57.44	67.24
	65.26	60.56	63.17	55.79	61.18
	62.70	56.62	60.64		55.57
	55.23		59.28		
			56.22		
			55.31		
N <sup>1</sup> CH <sub>2</sub> CH <sub>2</sub> N <sup>3</sup>	32.52	33.31	32.62	32.35	33.11
N <sup>1</sup> CH <sub>2</sub> CH <sub>2</sub> N <sup>3</sup>	24.09	25.34	24.29	24.08	25.08
C- $\alpha$ (C- $\beta$ )			40.11 (1) <sup>b</sup> (48.37)		
			39.32 (3) (46.70)		
			37.78 (3) (46.49)		
			36.79 (1) (46.22)		
C-1	44.88 (1) <sup>b</sup>	45.48 (2) <sup>b</sup>	45.64 (4) <sup>b</sup>	52.41	45.02 (Propyl)
	44.45 (2)	45.00 (2)	45.11 (2)	51.33	44.47 (Propyl)
	43.81 (2)	43.81 (2)	44.93 (2)		43.37 (Propyl)
	42.25 (1)	43.06 (1)	44.10 (2)		42.58 (Propyl)
			43.94 (2)		
			43.46 (1)		
			43.24 (1)		
			43.04 (1)		

\* reference 20; <sup>b</sup> ratio of the signal intensities

**Table S2A.** [Gd(DTPA-BPDA)(H<sub>2</sub>O)] P<sub>m</sub> = 2.79 10<sup>-3</sup>

T (K)	10 <sup>3</sup> K (T)	T <sub>1A</sub> (s)	T <sub>1</sub> (s)	ln(1/T <sub>1r</sub> )	T <sub>2A</sub> (s)	T <sub>2</sub> (s)	ln(1/T <sub>2r</sub> )	v <sub>A</sub> (s <sup>-1</sup> )	v (s <sup>-1</sup> )	Δω <sub>r</sub> (rad s <sup>-1</sup> )
271*	3.69				2.85E-03	1.84E-03	11.15	720	690	-6.78E+04
283	3.53	4.45E-03	2.90E-03	10.67	4.09E-03	1.76E-03	11.66	208	201	-1.58E+04
302	3.31	7.76E-03	4.76E-03	10.28	6.17E-03	7.69E-04	12.92	161	153	-1.80E+04
321	3.12	1.13E-02	7.51E-03	9.68	8.38E-03	3.25E-04	13.87	117	71	-1.04E+04
323	3.10	1.19E-02	7.89E-03	9.64	8.71E-03	3.54E-04	13.79	602	530	-1.62E+05
338	2.96	1.65E-02	1.08E-02	9.35	9.02E-03	2.36E-04	14.21	69	-114	-4.12E+05
358	2.79	2.24E-02	1.42E-02	9.13	1.44E-02	2.50E-04	14.16	24	-195	-4.93E+05
375	2.67	2.72E-02	1.86E-02	8.17	1.47E-02	3.27E-04	13.75	-21	-302	-6.33E+05

**Table S2B.** [Gd(DTPA-GLUCA)(H<sub>2</sub>O)] P<sub>m</sub> = 3.44 10<sup>-3</sup>

T (K)	10 <sup>3</sup> K (T)	T <sub>1A</sub> (s)	T <sub>1</sub> (s)	ln(1/T <sub>1r</sub> )	T <sub>2A</sub> (s)	T <sub>2</sub> (s)	ln(1/T <sub>2r</sub> )	v <sub>A</sub> (s <sup>-1</sup> )	v (s <sup>-1</sup> )	Δω <sub>r</sub> (rad s <sup>-1</sup> )
270*	3.70							720	702	-3.29E+04
273*	3.67							710	686	-4.38E+04
283	3.53	4.29E-03	2.37E-03	10.92	3.95E-03	1.35E-03	11.87			
287*	3.49							701	694	-1.28E+04
302	3.31	7.82E-03	3.62E-03	10.67	6.03E-03	5.62E-04	13.06			
322	3.11	1.20E-02	5.87E-03	10.13	7.81E-03	2.35E-04	14.00			
324*	3.09	1.19E-02	5.99E-03	10.09				603	494	-1.99E+05
341	2.93	1.57E-02	8.58E-03	9.64	9.33E-03	1.73E-04	14.32	82	-57	-2.54E+05
360	2.78	2.23E-02	1.20E-02	9.32	1.11E-02	2.00E-04	14.17	42	-233	-5.02E+05
374	2.67	2.67E-02	1.48E-02	9.08	1.32E-02	2.64E-04	13.89	76	-257	-6.08E+05
393	2.54	3.34E-02	1.89E-02	8.80	1.23E-02	4.53E-04	13.33	-47	-304	-4.69E+05

**Table S2C.** [Gd(DTPA-BENGALAA)(H<sub>2</sub>O)] P<sub>m</sub> = 3.68 10<sup>-3</sup>

T (K)	10 <sup>3</sup> K (T)	T <sub>1A</sub> (s)	T <sub>1</sub> (s)	ln(1/T <sub>1B</sub> )	T <sub>2A</sub> (s)	T <sub>2</sub> (s)	ln(1/T <sub>2B</sub> )	v <sub>A</sub> (s <sup>-1</sup> )	v (s <sup>-1</sup> )	Δω <sub>r</sub> (rad s <sup>-1</sup> )
273*	3.67							-1584	-1619	-5.98E+04
279	3.59	3.73E-03	2.12E-03	10.92	3.67E-03	1.81E-03	11.24			
287*	3.49							-1590	-1615	-4.27E+04
288	3.47	5.09E-03	2.54E-03	10.89	4.48E-03	1.70E-03	11.50			
297	3.36	6.68E-03	3.03E-03	10.80	5.16E-03	1.26E-03	12.00			
311*	3.22							-1601	-1625	-4.10E+04
312	3.21	9.28E-03	4.08E-03	10.53	6.20E-03	6.16E-04	12.89			
314*	3.19							-1613	-1638	-4.27E+04
323*	3.10							-1620	-1669	-8.37E+04
326	3.07	1.19E-02	5.26E-03	10.27	7.43E-03	3.05E-04	13.66			
340	2.94	1.68E-02	7.04E-03	10.02	8.45E-03	2.08E-04	14.06	48	-9	-9.61E+04
359	2.79	2.23E-02	9.66E-03	9.68	6.86E-03	1.94E-04	14.12	2	-182	-3.15E+05
368	2.72	2.53E-02	1.07E-02	9.59	1.32E-02	2.09E-04	14.06	-21	-315	-5.03E+05
377	2.65	2.76E-02	1.18E-02	9.48	9.12E-03	2.30E-04	13.96	-38	-344	-5.23E+05
391	2.56	3.46E-02	1.57E-02	9.15	5.83E-03	3.33E-04	13.55	-70	-310	-4.09E+05

**Table S3A.** [Gd(DTPA-BPDA)(H<sub>2</sub>O)] P<sub>m</sub> = 2.79 10<sup>-3</sup> T = 302.6 K

P (MPa)	T <sub>2</sub> (s)	T <sub>2A</sub> (s)	1/T <sub>2r</sub> (s-1)	ln(1/T <sub>2r</sub> )
0.60	6.95E-04	8.46E-03	4.73E+05	13.07
4.00	7.17E-04	8.48E-03	4.58E+05	13.03
25.00	7.63E-04	8.55E-03	4.28E+05	12.97
50.00	7.96E-04	8.65E-03	4.09E+05	12.92
75.00	8.59E-04	8.74E-03	3.76E+05	12.84
101.00	8.90E-04	8.84E-03	3.62E+05	12.80
124.00	9.32E-04	8.93E-03	3.45E+05	12.75
152.00	1.01E-03	9.04E-03	3.16E+05	12.66
172.50	1.06E-03	9.12E-03	2.99E+05	12.61
201.20	1.12E-03	9.23E-03	2.81E+05	12.55

**Table S3B.** [Gd(DTPA-BGLUCA)(H<sub>2</sub>O)] P<sub>m</sub> = 3.44 10<sup>-3</sup> T = 302.5 K

P (MPa)	T <sub>2</sub> (s)	T <sub>2A</sub> (s)	1/T <sub>2r</sub> (s-1)	ln(1/T <sub>2r</sub> )
0.60	5.30E-04	8.44E-03	5.14E+05	13.15
4.00	5.30E-04	8.46E-03	5.14E+05	13.15
26.00	5.73E-04	8.54E-03	4.73E+05	13.07
51.50	6.12E-04	8.63E-03	4.41E+05	13.00
78.00	6.51E-04	8.73E-03	4.13E+05	12.93
99.00	6.78E-04	8.81E-03	3.96E+05	12.89
126.00	7.19E-04	8.92E-03	3.72E+05	12.83
151.00	7.46E-04	9.01E-03	3.57E+05	12.79
173.00	8.08E-04	9.10E-03	3.28E+05	12.70
198.00	8.38E-04	9.20E-03	3.15E+05	12.66

**Table S3C.** [Gd(DTPA-BENGALAA)(H<sub>2</sub>O)] P<sub>m</sub> = 3.67 10<sup>-3</sup> T = 311.9 K

P (MPa)	T <sub>2</sub> (s)	T <sub>2A</sub> (s)	1/T <sub>2r</sub> (s-1)	ln(1/T <sub>2r</sub> )
0.60	5.86E-04	1.04E-02	4.39E+05	12.99
0.60	6.00E-04	1.04E-02	4.28E+05	12.97
28.00	6.30E-04	1.05E-02	4.07E+05	12.92
50.00	6.56E-04	1.07E-02	3.90E+05	12.87
77.00	7.03E-04	1.08E-02	3.62E+05	12.80
100.00	7.19E-04	1.09E-02	3.54E+05	12.78
127.00	7.68E-04	1.10E-02	3.30E+05	12.71
150.00	7.81E-04	1.11E-02	3.24E+05	12.69
174.50	8.36E-04	1.12E-02	3.02E+05	12.62
201.00	8.52E-04	1.14E-02	2.96E+05	12.60

**Table S4.**

Complex	$T$ (K)	$k_{ex}^T$ (VT) <sup>a</sup> ( $10^5 s^{-1}$ )	$k_0^T$ (VP) <sup>b</sup> ( $10^5 s^{-1}$ )
[Gd(DTPA-BPDA)(H <sub>2</sub> O)]	302.6	4.7	5.1
[Gd(DTPA-BGLUCA)(H <sub>2</sub> O)]	302.5	5.1	5.7
[Gd(DTPA-BENGALAA)(H <sub>2</sub> O)]	311.9	4.9	4.8

<sup>a</sup> values obtained by calculation from eq 17.

<sup>b</sup> exchange rate at zero pressure and temperature  $T$ .

**Table S5A.** [Gd(DTPA-BPDA)(H<sub>2</sub>O)] c = 1mM pH = 7.37

$\omega_1$ (MHz)	$T_1$ (s)				
	5 °C	15 °C	25 °C	37 °C	45 °C
0.01	6.397	5.994	6.344	6.472	6.313
0.02	6.428	5.832	6.356	6.528	6.263
0.04	6.393	5.971	6.157	6.489	6.265
0.100	6.478	5.929	6.227	6.552	6.364
0.200	6.357	5.904	6.248	6.591	6.246
0.400	6.329	5.930	6.233	6.455	6.417
1.000	6.338	5.791	6.202	6.390	6.271
2.000	5.908	5.712	6.055	6.199	6.013
4.000	5.600	5.359	5.710	5.812	5.741
10.000	4.825	4.524	4.967	4.828	4.616
20.000	4.541	3.949	4.432	4.373	3.857
30.000	4.133	4.020	4.198	3.938	3.539
200.00	3.915	3.824	3.893	3.800	3.436
400.00	3.582	3.576	4.002	3.687	3.286

**Table S5B.** [Gd(DTPA-BGLUCA)(H<sub>2</sub>O)] c = 1mM pH = 7.35

<i>ω<sub>1</sub></i> (MHz)	<i>T<sub>1</sub></i> (s)				
	5 °C	15 °C	25 °C	37 °C	45 °C
0.01	7.012	6.438	6.839	6.949	6.753
0.02	6.886	6.434	6.879	6.846	6.755
0.04	6.879	6.429	6.890	6.865	6.847
0.100	6.882	6.357	7.005	6.832	6.755
0.200	6.896	6.321	6.922	6.948	6.745
0.400	6.805	6.432	6.923	6.829	6.644
1.000	6.746	6.250	6.880	6.741	6.705
2.000	6.429	6.092	6.570	6.640	6.479
4.000	5.820	5.740	6.262	6.152	6.018
10.000	5.169	4.892	5.275	5.189	5.175
20.000	4.705	4.498	4.724	4.563	4.396
30.000	4.826	4.226	4.544	4.517	4.187
200.00	4.398	4.180	4.412	4.479	4.268
400.00	3.869	3.787	4.430	4.293	3.834

**Table S5C.** [Gd(DTPA-BENGALAA)(H<sub>2</sub>O)] c = 1mM pH = 6.88

$\omega_1$ (MHz)	$T_1$ (s)				
	5 °C	15 °C	25 °C	37 °C	45 °C
0.01	6.727	6.041	6.088	6.558	6.462
0.02	6.547	5.977	6.115	6.661	6.699
0.04	6.729	5.861	6.235	6.534	6.596
0.100	6.694	5.978	6.092	6.515	6.714
0.200	6.590	5.908	6.038	6.564	6.689
0.400	6.663	5.903	6.041	6.489	6.585
1.000	6.300	5.803	5.917	6.216	6.488
2.000	6.013	5.585	5.718	6.114	6.181
4.000	5.726	5.182	5.503	5.744	5.938
10.000	5.102	4.523	4.857	5.126	5.300
20.000	4.822	4.264	4.437	4.856	4.638
30.000	4.769	4.143	4.264	4.611	4.652
200.00	4.316	4.088	4.176	4.449	4.496
400.00	4.003	3.929	4.017	4.170	4.077

Figure S1

