

Inorganic Chemistry

including bioinorganic chemistry

Inorg. Chem., 1996, 35(24), 6974-6980, DOI: [10.1021/ic960792r](https://doi.org/10.1021/ic960792r)

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Table S1. Electrochemical Data for $\{[(bpy)_2Ru(bpm)]_2MCl_2\}^{5+}$ where $M = Rh^{III}$ or Ir^{III} , $bpy = 2,2'$ -bipyridine and $bpm = 2,2'$ -bipyrimidine in 0.1 M Bu_4NPF_6 in CH_3CN ^a

Compound	$E_{1/2}$ (V)	Assignment
$\{[(bpy)_2Ru(bpm)]_2RhCl_2\}^{5+}$	+1.70	$2Ru^{II}/2Ru^{III}$
	-0.13	$bpm, bpm/bpm^-, bpm$
	-0.26	$bpm^-, bpm/bpm^-, bpm^-$
	-0.78 (E_p^a)	Rh^{III}/Rh^I
	-1.19	$2Ru(bpy)_2/2Ru(bpy^-(bpy))^b$
	-1.45	$2Ru(bpy^-(bpy))/2Ru(bpy^-(bpy))^b$
$\{[(bpy)_2Ru(bpm)]_2IrCl_2\}^{5+}$	+1.70	$2Ru^{II}/2Ru^{III}$
	-0.08	$bpm, bpm/bpm^-, bpm$
	-0.21	$bpm^-, bpm/bpm^-, bpm^-$
	-0.90	$bpm^-, bpm^-/bpm^{2-}, bpm^-$
	-1.07	$bpm^{2-}, bpm^-/bpm^{2-}, bpm^{2-}$
	-1.60	$2Ru(bpy)_2/2Ru(bpy^-(bpy))^b$
	-1.70	Ir^{III}/Ir^I^b
	-1.75	$2Ru(bpy^-(bpy))/2Ru(bpy^-(bpy))^b$

^a Potentials reported vs. Ag/AgCl (0.29 V vs. NHE) at a scan rate of 200 mV/s.

^b Resolution of these waves is difficult and reduction in this potential region leads to adsorption onto the electrode surface.

Figure S1. Electronic Absorption Spectra for $\{[(bpy)_2Ru(bpm)]_2MCl_2\}^{5+}$ in acetonitrile at room temperature where $M = Rh^{III}$ (—) or Ir^{III} (— —), $bpy = 2,2'$ -bipyridine and $bpm = 2,2'$ -bipyrimidine.

