Supporting Information

IscA, an Alternate Scaffold for Fe-S Cluster Biosynthesis[†]

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Figure S1. Mössbauer spectra of control samples.

Figure S2. Mössbauer spectrum of a sample frozen at a reaction time of 4 hrs during the reaction of NifS-catalyzed assembly of Fe-S cluster on ^{Nif}IscA.

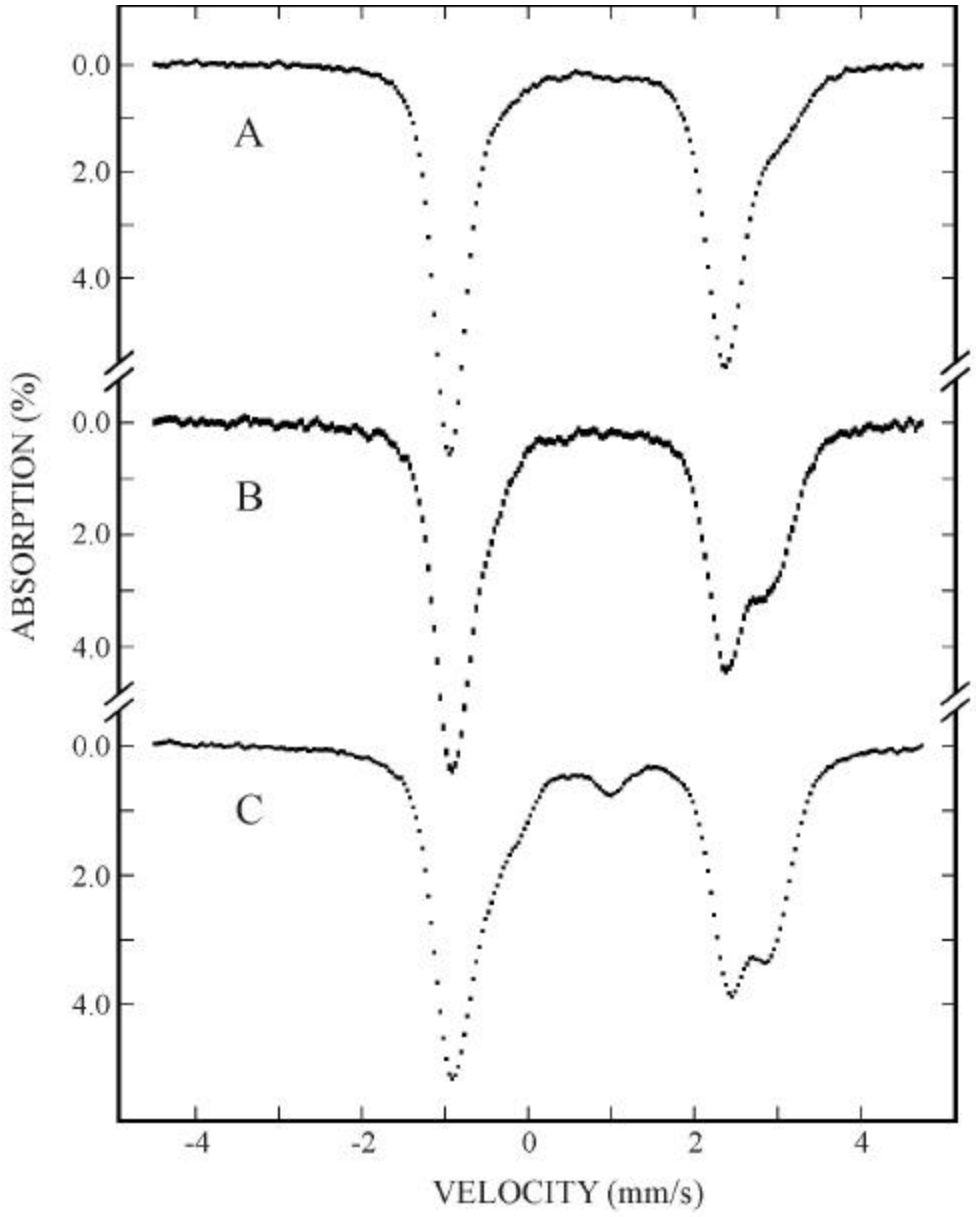

Figure S1. Mössbauer spectra of control samples containing (A) ferrous sulfate and cysteine, (B) ferrous sulfate, cysteine and ^{Nif}IscA, and (C) ferrous sulfate, cysteine, and NifS. These spectra were recorded at 4.2 K in a parallel field of 50 mT. They are dominated by the quadrupole doublets of the three Fe(II) species mentioned in the text. The small peak at ~1 mm/s in C coincides with the high-energy line of the quadrupole doublet of the $[4Fe\text{-}4S]^{2+}$.

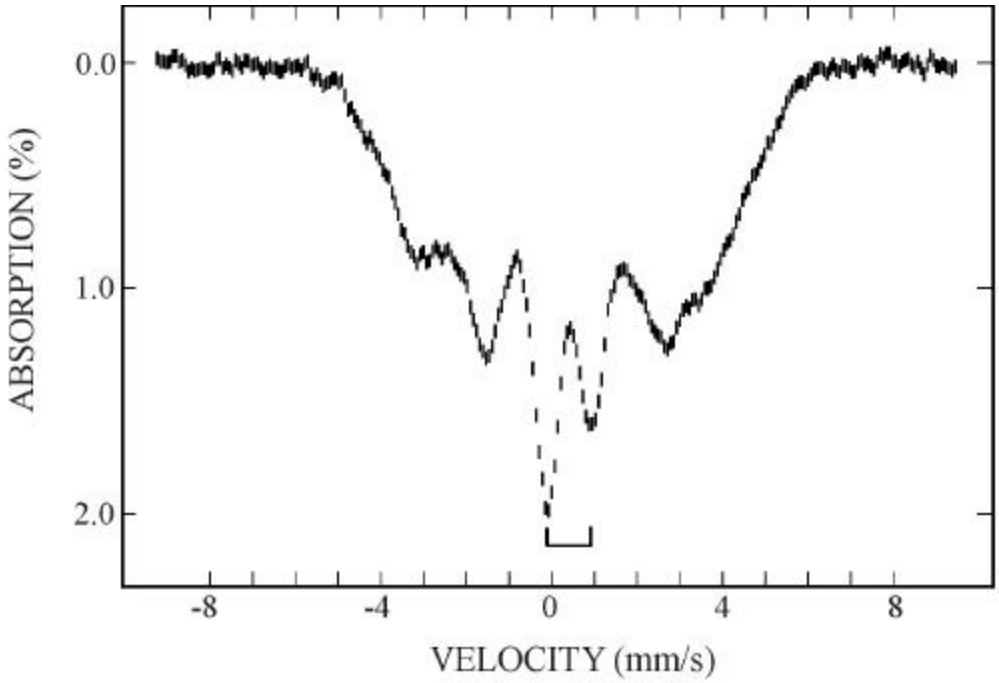

Figure S2. Mössbauer spectrum of a sample frozen at a reaction time of 4 hrs during the reaction of NifS-catalyzed assembly of Fe-S cluster on ^{Nif}IscA. The spectrum was recorded at 4.2 K in a parallel applied field of 50 mT. The bracket marks the peak positions of the quadrupole doublet associated with the $[4Fe\text{-}4S]^{2+}$ cluster. The majority of the absorption, however, is broad and featureless and can be attributed to polymeric iron sulfides in solution.