# Magnetic Bistability in a Discrete Organic Radical

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#### **Experimental Section**

General. All manipulations were carried out under an Ar or N<sub>2</sub> atmosphere by using standard Schlenk or glove box techniques. Solvents were dried prior to use. Tetrabis(4-methoxyphenyl)-2,3,5,6-tetramethylbenzene-1,4-diamine 1 was synthesized by a method similar to that for the preparation of 1,5-bis(di-p-methoxyphenylamino)naphthalene.<sup>S1</sup> Ag[Al(OR<sub>Me</sub>)<sub>4</sub>] was prepared by the published procedure.<sup>S2</sup> 1,4-dibromotetramethylbenzene, di-p-tolylamine and AgSbF<sub>6</sub> were purchased from Alfa Aesar and used upon arrival. Cyclic voltammetry was performed on an IM6ex electrochemical workstation in CH<sub>2</sub>Cl<sub>2</sub> at the scan rate of 100 mV/s with platinum as the working and counter electrodes, as solvent Ag/Ag<sup>+</sup> as the reference electrode and 0.2 M  $^{n}Bu_{4}NPF_{6}$  as the supporting electrolyte. The  $^{1}H$ NMR and <sup>13</sup>C NMR spectra were recorded in solution of CDCl<sub>3</sub>using a Bruker Ultra Shield 300 MHz spectrometer in ppm downfield from internal standard Me<sub>4</sub>Si. Element analyses were performed at Shanghai Institute of Organic Chemistry, the Chinese Academy of Sciences. EPR spectra were obtained using Bruker EMX-10/12 X-band variable-temperature apparatus and were simulated with the software of WINEPR SimFonia. Magnetic measurements were performed using a Quantum Design SQUID VSM magnetometer with a field of 0.1 T. X-ray crystal structures were obtained by using Bruker D8 CMOS detectors. Crystal data and structure refinement for  $1^{2+2}$ [Al(OR<sub>Me</sub>)<sub>4</sub>]<sup>-</sup> are listed in Table S1. CCDC 1479502 - 1479503 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data request/cif.

Synthesis of neutral 1. A mixture of 1,4-dibromotetramethylbenzene (2.89 g, 9.9 mmol) , di-*p*-methoxyphenylamine(5.00 g, 21.80 mmol), Pd(OAc)<sub>2</sub> (0.05 g, 0.20 mmol), tri-*tert*-butylphosphine (0.20 mmol), sodium *tert*-butoxide (2.6 g, 27.7 mmol) in toluene (100 mL) was heated under nitrogen at 100°C for 10 h. Upon cooling to room temperature, the reaction solution was taken up with Et<sub>2</sub>O and washed with brine. The organic layer was dried with Na<sub>2</sub>SO<sub>4</sub>. After evaporation of the solvent, purification by column chromatography on silica gel using CHCl<sub>3</sub> afforded 1 (3.60 g, 63.82 %) as white solid; m.p. 217-219°C; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  6.91 (d, 8H), 6.79 (d, 8H), 3,80 (s, 12H), 2.01 (s, 12H); <sup>13</sup>C NMR (75MHz, CDCl<sub>3</sub>)  $\delta$  153.2, 140.1, 135.5, 120.2, 114.6, 55.4, 15.4; UV-Vis (CH<sub>2</sub>Cl<sub>2</sub>):  $\lambda_{max} = 296$  nm.

Synthesis of Dication Salt  $1^{2+} \cdot 2[Al(OR_{Me})_4]^-$ . A solution of  $AgSbF_6$  (0.181 g, 0.52 mmol) in  $CH_2Cl_2$  (15 mL) was added dropwise to the mixture of **1** (0.148 g, 0.25 mmol) and  $Li[Al(OR_{Me})_4]$  (0.397 g, 0.52 mmol) in  $CH_2Cl_2$  (30 ml). The resultant solution was stirred overnight at room temperature and then filtered to remove the gray precipitate (Ag

metal and LiSbF<sub>6</sub>). The filtrate was concentrated and stored at around -30 °C for one day to afford blue crystals. Yield: 0.326 g, 61.97 %; m.p. 250-253 °C. UV-Vis (CH<sub>2</sub>Cl<sub>2</sub>):  $\lambda_{max} = 807$  nm; elemental analysis calcd (%) for C<sub>70</sub>H<sub>64</sub>Al<sub>2</sub>F<sub>48</sub>N<sub>2</sub>O<sub>12</sub>: C 40.21, H 3.08, N 1.34; Found: C 39.72, H 3.05, N 1.47.

	100 K	200 K
formula	$C_{72}H_{68}Al_2F_{48}Cl_4N_2O_{12}\\$	$C_{72}H_{68}Al_2F_{48}Cl_4N_2O_{12}\\$
Mr [g mol <sup>-1</sup> ]	2261.04	2261.04
crystal system	Monoclinic	Monoclinic
space group	<i>P</i> 2(1)/c	<i>P</i> 2(1)/c
Ζ	2	2
Temp. (K)	100(2)	200(2)
$\mu(\text{mm}^{-1})$	0.310	0.302
<i>a</i> (Å)	10.0924(7)	10.2354(5)
<i>b</i> (Å)	22.1142(15)	22.4215(11)
<i>c</i> (Å)	20.0895(14)	20.0452(10)
α (°)		
β (°)	92.8778(19)	92.5728(16)
γ (°)		
V [Å <sup>3</sup> ]	4478.0(5)	4595.6(4)
<i>R</i> 1 ( <i>I</i> > $2\sigma(I)$ )	0.0492	0.0643
wR2 (all data)	0.1222	0.1740

**Table S1.** Crystal data and structure refinement for  $1^{2+} \cdot 2[Al(OR_{Me})_4]^{-} \cdot 2CH_2Cl_2$ .



**Figure S1.** The cyclic voltammogram of **1** in  $CH_2Cl_2$  at room temperature with  $Ag/Ag^+$  as the reference electrode and 0.2 M  $^nBu_4NPF_6$  as the supporting electrolyte. Scan rate: 100 mV/s.



**Figure S2.** Temperature-dependent plots of  $\chi_M T$  for the crystals of  $\mathbf{1}^{2+}$  recorded in the sweep mode at different temperature sweep rates (1, 5 and 10 K min<sup>-1</sup>).



**Figure S3.** Temperature-dependent plots of  $\chi_M T$  for the crystals of  $\mathbf{1}^{2+}$  recorded in the sweep mode at the sweep rate of 1 K min<sup>-1</sup> and the fitting plots (in red) via the Bleaney-Bowers equation.



**Figure S4**. Plot of normalized magnetization,  $M/M_{sat}$  vs magnetic field, H of  $\mathbf{1}^{2+}$  at T = 2 K.



**Figure S5.** (a) The powder EPR spectrum of  $1^{2+} \cdot 2[Al(OR_{Me})_4]^-$  at 100 K; (b) The simulated EPR spectrum with  $A_y$  (N) = 0.69 mT and E = 1.20 mT. The central line due to the mono-radical impurity is simulated with g = 2.0036.

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**Figure S6.** (a) The powder EPR spectrum of  $1^{2+} \cdot 2[Al(OR_{Me})_4]^-$  at 200 K; (b) The simulated EPR spectrum. The central line due to the mono-radical impurity is simulated with g = 2.0037.



Figure S7. Plot of the product (*IT*) of the intensity for the  $\Delta m_s = 2$  resonance and the temperature (*T*) vs. *T*. The values of *IT* are obtained by numerical double integration of the  $\Delta m_s = 2$  region.



Figure S8. (a) Absorption spectrum of  $1 \times 10^{-4}$  M  $\mathbf{1}^{2+}$  in CH<sub>2</sub>Cl<sub>2</sub> at 25°C; (b) Absorption spectrum  $\mathbf{1}^{2+}$  suspension in hexane upon sonication.

### Quantum chemical calculations

All calculations were performed with the Gaussian 09 program suite.<sup>S3</sup> The symmetry-broken approach was applied for open-shell singlet calculations and spin contamination errors were corrected by approximate spin-projection method. All the geometry optimizations were carried out at the (U)B3LYP/6-31G(d) level of theory. The obtained stationary points were characterized by frequency calculations. Single point calculations using the HT and LT X-ray data of  $1^{2+}$  performed at the UB3LYP/6-311G(d,p) level.

## Coordinates for optimized geometries.

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Ν	-0.082472000	-0.026626000	2.810960000
С	0.255428000	-5.214420000	6.654642000
Н	1.344005000	-5.104503000	6.614866000
Н	-0.003235000	-6.248773000	6.876091000
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С	0.530005000	-1.433900000	4.729758000
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С	-0.122231000	-1.259570000	3.488452000
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# **1**<sup>2+</sup>-OS

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**1**<sup>2+</sup>-T

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С	1.144088000	-0.435372000	0.728012000
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0	-0.076624000	-4.763999000	-5.622754000

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