# SUPPORTING INFORMATION

# Probing Distinct Fullerene Formation Processes from Carbon Precursors of Different Sizes and Structures

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#### **Experimental Section**

Atomic Force Microscopy (AFM) and Field Emission Scanning Electron Microscopy (FE-SEM). For analysis of multi-layered graphene (MLG), MLG samples dispersed in a solution of water and 2-propanol (50:50, v/v) were deposited onto a silicon wafer. AFM experiments were performed at the National Institute for Nanomaterials Technology (NIST; Pohang, Korea) with a VEECO Dimension 3100 atomic force microscope. FE-SEM experiments were performed at NIST using a JEOL JSM-7401F scanning electron microscope.

UV/Vis spectroscopy. The solutions of triphenylene (Tri) and  $C_{60}$  were prepared by dissolving them in toluene. Then, the UV/Vis absorption of prepared solutions were analyzed using Shimadzu UV-1800 (Shimadzu, Japan) spectrophotometer.

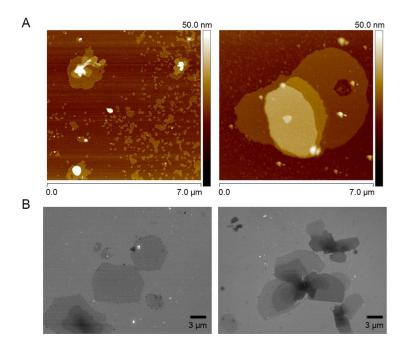
#### Discussion

#### Penetration depth of 355 nm laser in multi-walled carbon nanotubes.

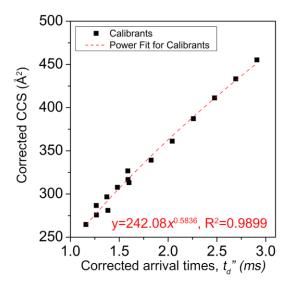
The intensity of the 355 nm laser as a function of the depth of graphite is estimated by equation (1).

$$\frac{I}{I_0} = e^{-\alpha z} \qquad (1)$$

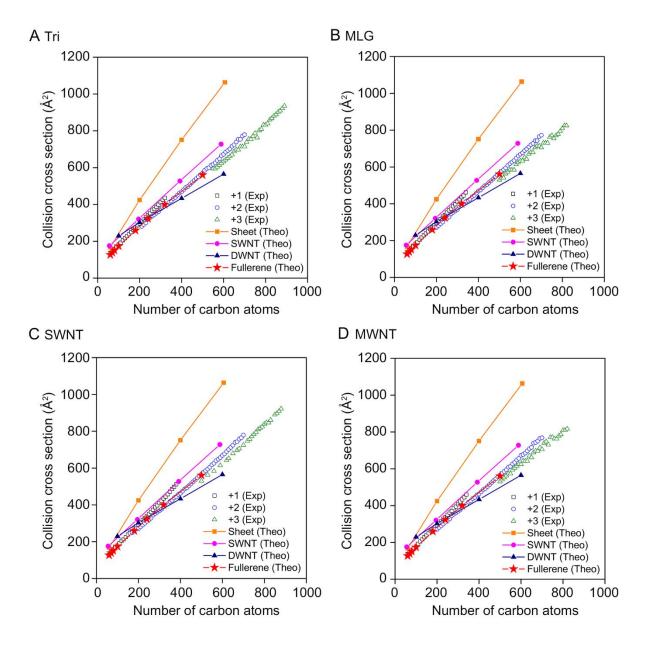
where  $\alpha = \frac{4\pi k}{\lambda}$ ,  $\lambda=355$  nm, and k=1.3, which is the extinction coefficient of graphite<sup>1</sup>. The z values at  $\frac{1}{I_0} = \frac{1}{e}$  is defined as the penetration depth of laser. This equation was previously shown to provide good estimate of laser penetration depth in graphite, similar to the experimental values<sup>2</sup>. Based on equation (1), the penetration depth of the 355 nm laser through graphite was calculated to be ~22 nm. Because the interval length between the layers in multi-walled carbon nanotubes (MWNT) (0.34 nm) is same as graphite, we assumed that the extinction coefficient of MWNT is equal to that of graphite.



**Figure S1.** (A) AFM images, and (B) FE-SEM images of MLG used in this study. Thickness of each MLG sheet was ~5 nm, indicating that it is multi-layered. The lateral dimensions were in the order of micrometers.



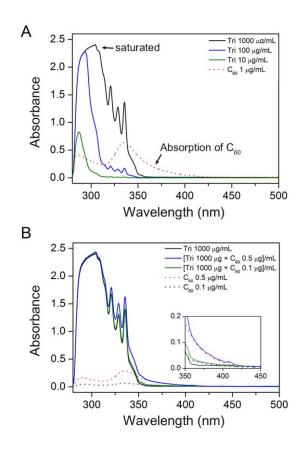
**Figure S2.** Collision cross-section (CCS) calibration curve obtained using poly-DL-alanine as calibrants.



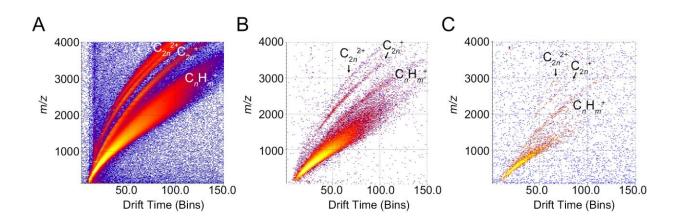
**Figure S3.** Correlation between the number of carbon atoms and collision cross-section (CCS) areas of carbon clusters from (A) triphenylene (Tri), (B) multi-layered graphene (MLG), (C) single-walled carbon nanotube (SWNT), and (D) multi-walled carbon nanotube (MWNT). The peaks in the spectra which have ratios, S/N>3, are considered. Experimental CCS values of carbon clusters were compared with the theoretical CCS values of various models of +1 charge state. Detailed description of the models is provided in Figure 2B.

**Table S1.** Theoretical CCS of the structural models presented in Figure 2. The CCS values ( $Å^2$ ) of the models were obtained for +1 and +2 charge state. The models with +2 charge show slightly increased theoretical CCS values. Due to insignificant difference between the values, we adopt the theoretical CCS values of the +1 charge state for plotting Figures 2 and S3.

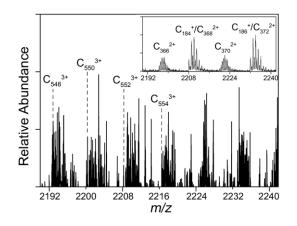
MODEL		+1	+2
Sheet	C <sub>60</sub>	167.64	170.72
	C <sub>200</sub>	424.33	425.21
	C <sub>400</sub>	751.07	752.28
	C <sub>600</sub>	1063.40	1063.80
SWNT	C <sub>56</sub>	175.60	178.29
	C <sub>195</sub>	320.81	322.46
	C <sub>392</sub>	527.21	528.76
	C <sub>588</sub>	727.77	728.46
	C <sub>100</sub>	229.59	232.23
DWNT	C <sub>200</sub>	302.57	304.96
	C <sub>400</sub>	433.36	434.54
	C <sub>600</sub>	564.99	566.00
Fullerene	C <sub>60</sub>	126.73	132.00
	C <sub>70</sub>	138.76	144.33
	C <sub>80</sub>	151.33	156.74
	C <sub>100</sub>	174.13	172.79
	C <sub>180</sub>	259.15	266.27
	C <sub>240</sub>	323.33	320.79
	C <sub>320</sub>	399.70	400.58
	C <sub>500</sub>	560.61	561.52



**Figure S4.** UV/Vis absorption spectra of triphenylene (Tri) precursor,  $C_{60}$  fullerene, and [Tri+ $C_{60}$ ] mixture for the examination of possible innate  $C_{60}$  constituents in Tri precursors. (A) The absorption spectra of Tri precursor and  $C_{60}$ .  $C_{60}$  has absorption at the wavelength range of 370-420 nm as previously reported by other groups,<sup>3,4</sup> whereas Tri precursor shows no absorption at this range. (B) The comparison of absorption spectra of Tri,  $C_{60}$ , and [Tri+ $C_{60}$ ] mixture. The same absorbance of  $C_{60}$  and [Tri+ $C_{60}$ ] at the wavelength range of 370-420 nm indicates that no  $C_{60}$  is included in the original Tri precursor.



**Figure S5.** IM-MS spectra of ions generated from (A) triphenylene (Tri), (B) anthracene (Ath), and (C) phenanthrene (Phn). These IM-MS spectra show high abundance of hydrocarbons  $(C_nH_m^+)$  and traces of carbon clusters  $(C_{2n}^+ \text{ and } C_{2n}^{2^+})$ .



**Figure S6.** Mobility-separated mass spectrum of +3 charged carbon clusters from LDI mass spectrum of multi-layer graphene (MLG). Inset shows the original mass spectrum over identical m/z range.

### References

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