

## Amidinatogermylene Metal Complexes as Homogeneous Catalysts in Alcoholic Media

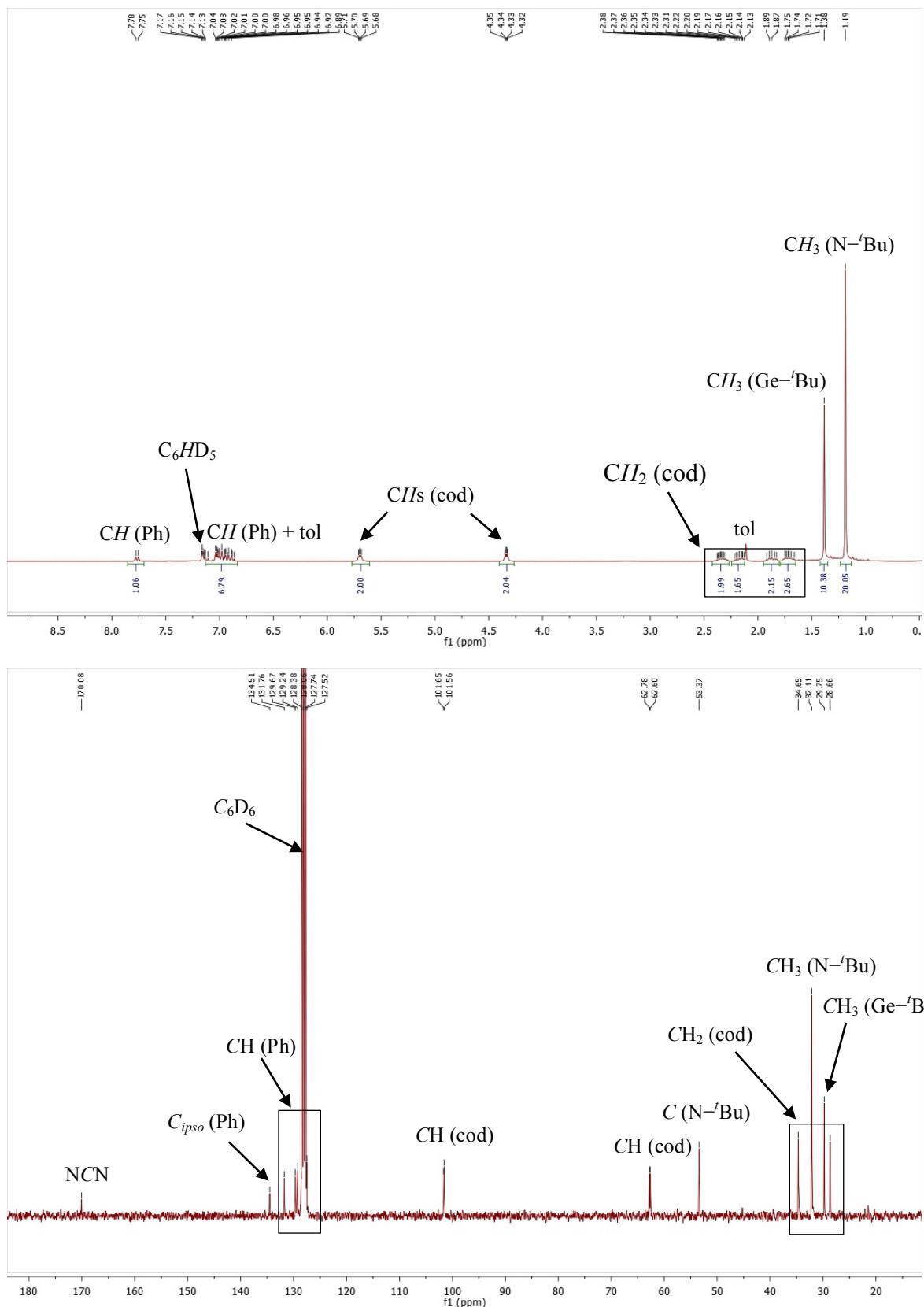
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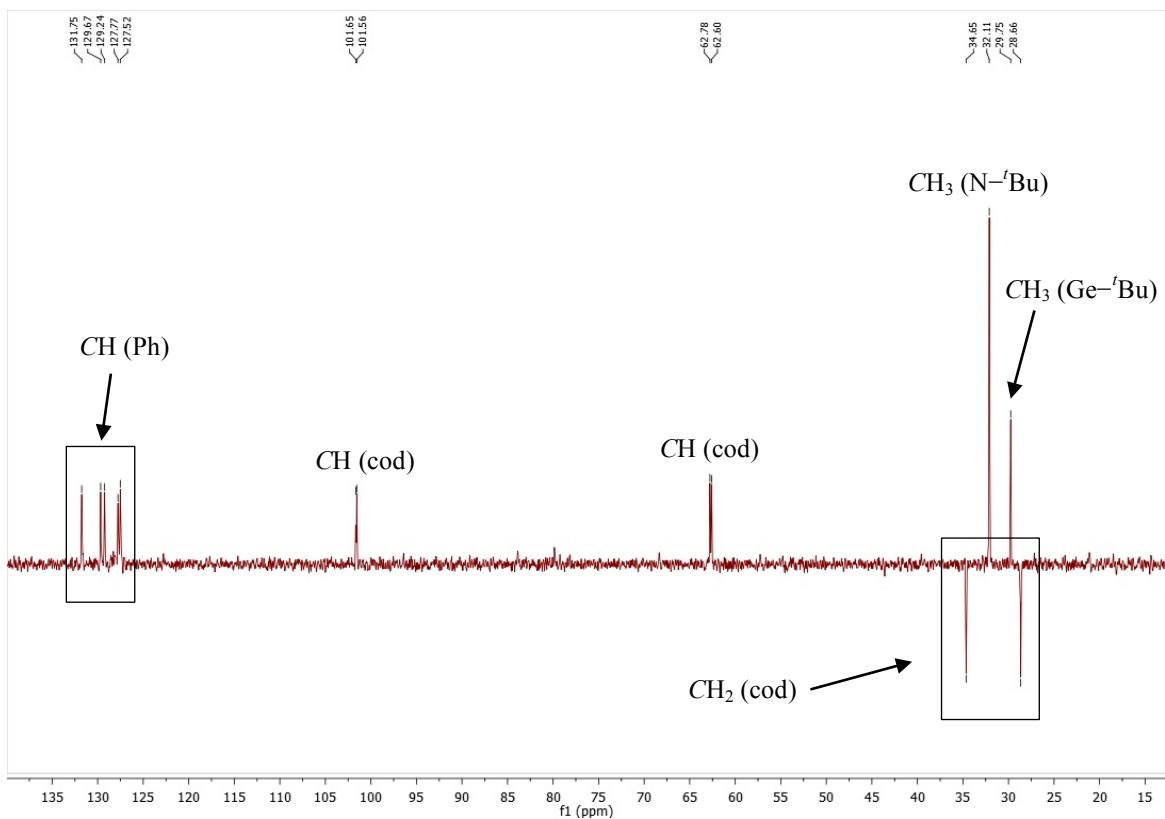
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**Table S1.** Crystal, measurement and refinement data for the compounds studied by XRD

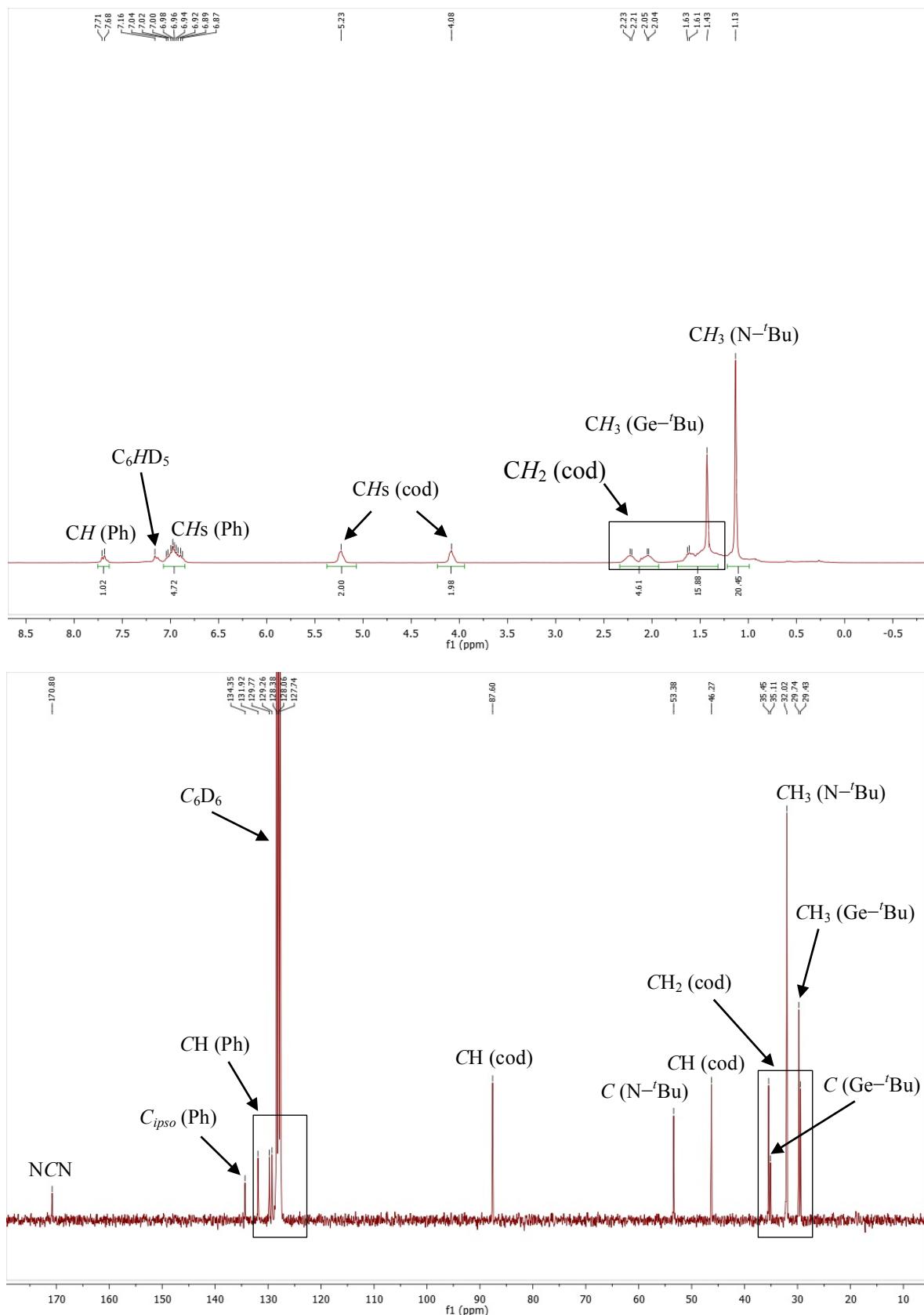
	<b>2</b>	<b>5</b>	<b>6</b>
formula	C <sub>27</sub> H <sub>44</sub> ClGeN <sub>2</sub> Rh	C <sub>29</sub> H <sub>46</sub> Cl <sub>2</sub> GeN <sub>2</sub> Ru	2(C <sub>29</sub> H <sub>47</sub> Cl <sub>2</sub> GeIrN <sub>2</sub> )
fw	607.59	667.3	1518.75
cryst syst	Monoclinic	monoclinic	Monoclinic
space group	<i>P</i> 21/c	<i>P</i> 21/n	<i>P</i> 21/c
<i>a</i> , Å	16.7872(2)	9.2077(1),	24.1428(2)
<i>b</i> , Å	12.2820(2)	20.0025(4)	9.3753(1)
<i>c</i> , Å	13.9392(2)	16.7333(3)	27.3016(2)
$\alpha$ , deg	90	90	90
$\beta$ , deg	106.677(2)	90.978(1)	95.134(1)
$\gamma$ , deg	90	90	90
<i>V</i> , Å <sup>3</sup>	2753.10(7)	3081.44(9)	6154.8(1)
Z	4	4	4
<i>F</i> (000)	1256	1376	3024
<i>D</i> <sub>calcd</sub> , g cm <sup>-3</sup>	1.466	1.438	1.639
$\mu$ , mm <sup>-1</sup>	1.805 (Mo K $\alpha$ )	6.900 (Cu K $\alpha$ )	11.185 (Cu K $\alpha$ )
cryst size, mm	0.19 x 0.13 x 0.10	0.12 x 0.10 x 0.08	0.26 x 0.16 x 0.09
<i>T</i> , K	120.0(1)	150(2)	155(2)
$\theta$ range, deg	3.32 to 31.50	3.44 to 69.00	3.25 to 68.99
min./max. <i>h</i> , <i>k</i> , <i>l</i>	-23/24, -17/18, -20/19	-10/11, -23/23, -14/19	-29/27, -8/11, -32/32
no. collected reflns	40344	14539	29901
no. unique reflns	8550	5640	11357
no. reflns with <i>I</i> >2 $\sigma$ ( <i>I</i> )	7168	5094	10591
no. params/restraints	298/0	328/0	659/0
GOF (on <i>F</i> <sup>2</sup> )	1.040	1.036	1.042
<i>R</i> <sub>1</sub> (on <i>F</i> , <i>I</i> >2 $\sigma$ ( <i>I</i> ))	0.031	0.043	0.024
<i>wR</i> <sub>2</sub> (on <i>F</i> <sup>2</sup> , all data)	0.063	0.108	0.059
min./max. $\Delta\rho$ , e Å <sup>-3</sup>	-0.489/0.597	-1.476/2.145	-1.411/0.948
CCDC dep. no.	1480090	1480091	1480092



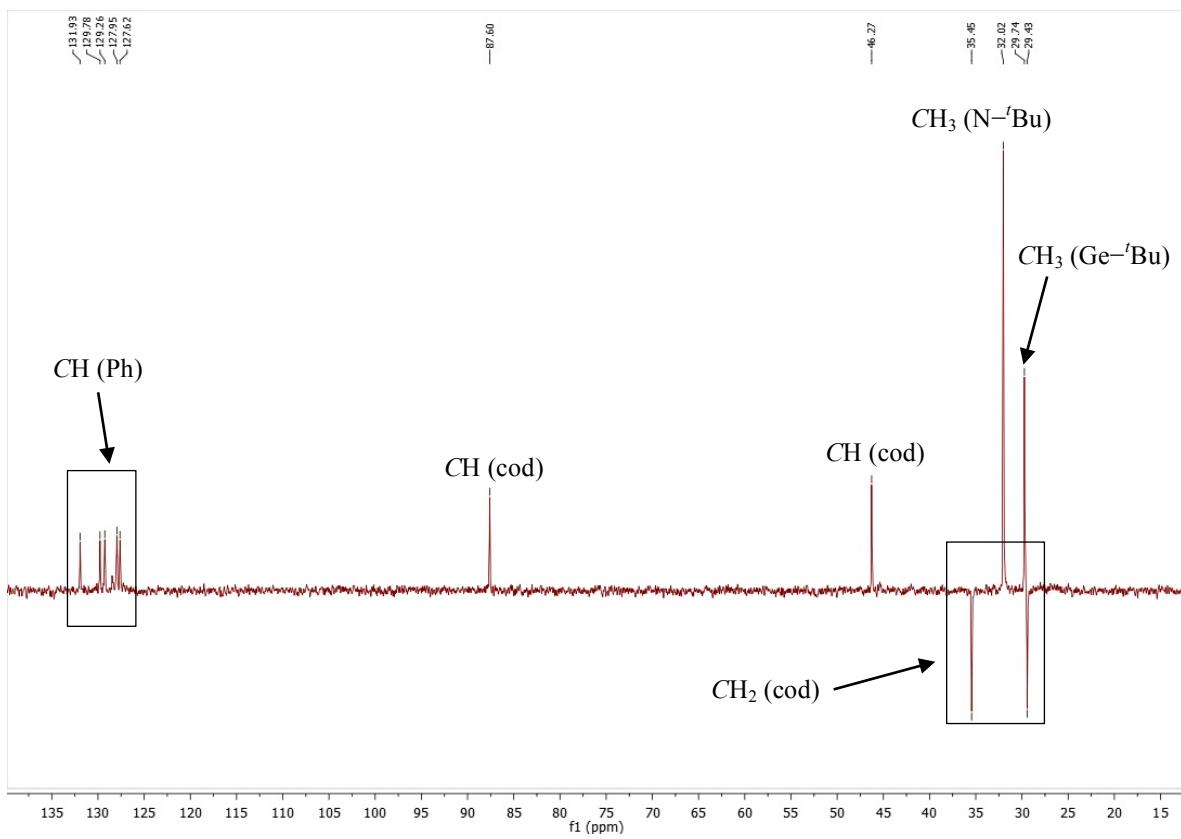
**Figure S1.**  $^1\text{H}$  (top; 300.1 MHz) and  $^{13}\text{C}\{\text{H}\}$  (bottom; 75.5 MHz) NMR spectra of **2** ( $\text{C}_6\text{D}_6$ , 25 °C).



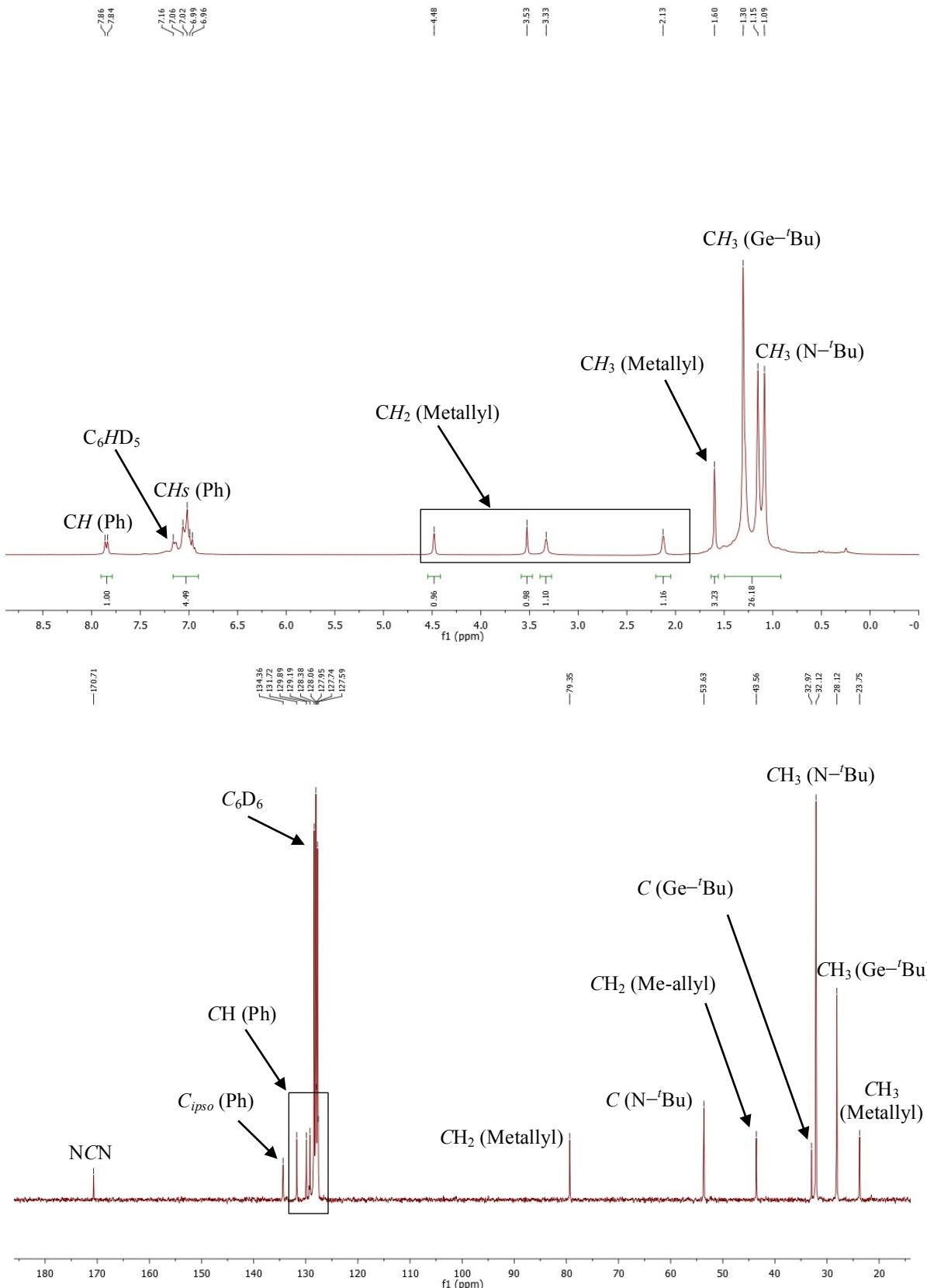
**Figure S2.**  $^{13}\text{C}\{^1\text{H}\}$ -DEPT 135 (75.5 MHz) NMR spectrum of **2** ( $\text{C}_6\text{D}_6$ , 25 °C).



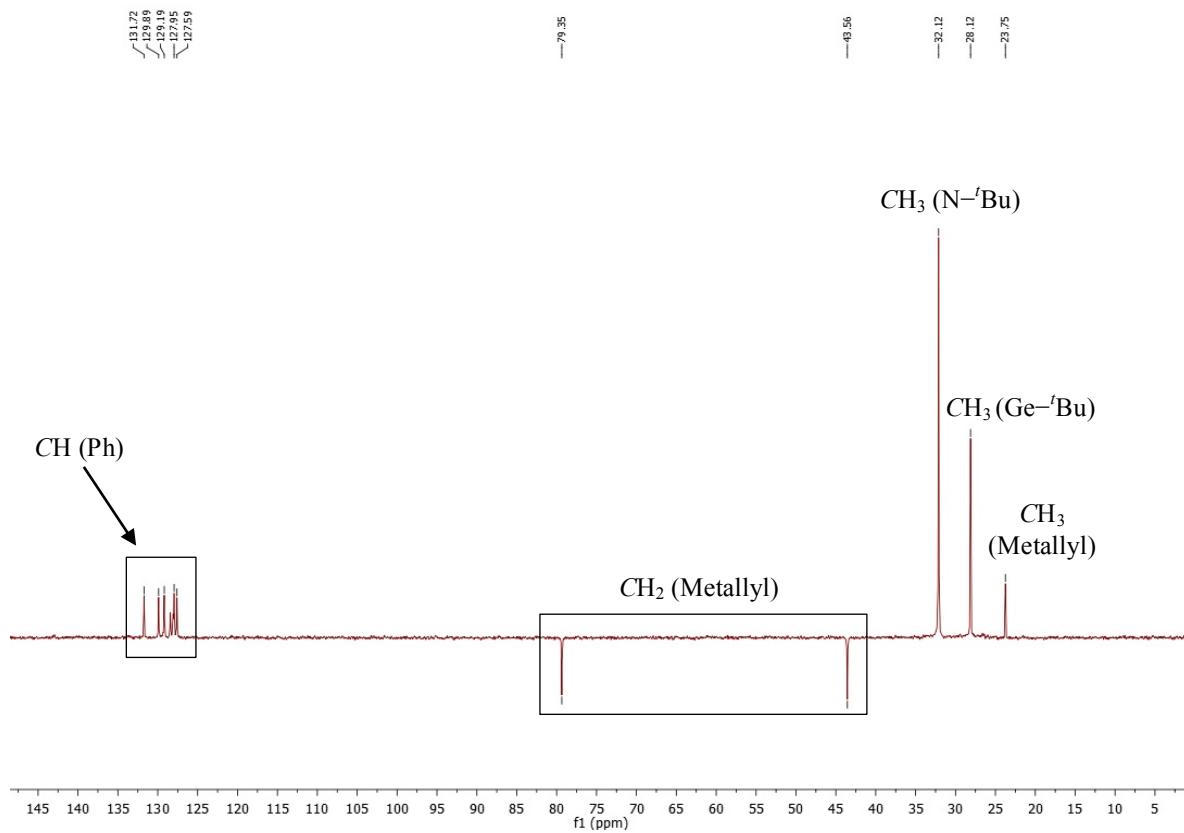
**Figure S3.**  $^1\text{H}$  (top; 300.1 MHz) and  $^{13}\text{C}\{\text{H}\}$  (bottom; 75.5 MHz) NMR spectra of **3** ( $\text{C}_6\text{D}_6$ , 25 °C).



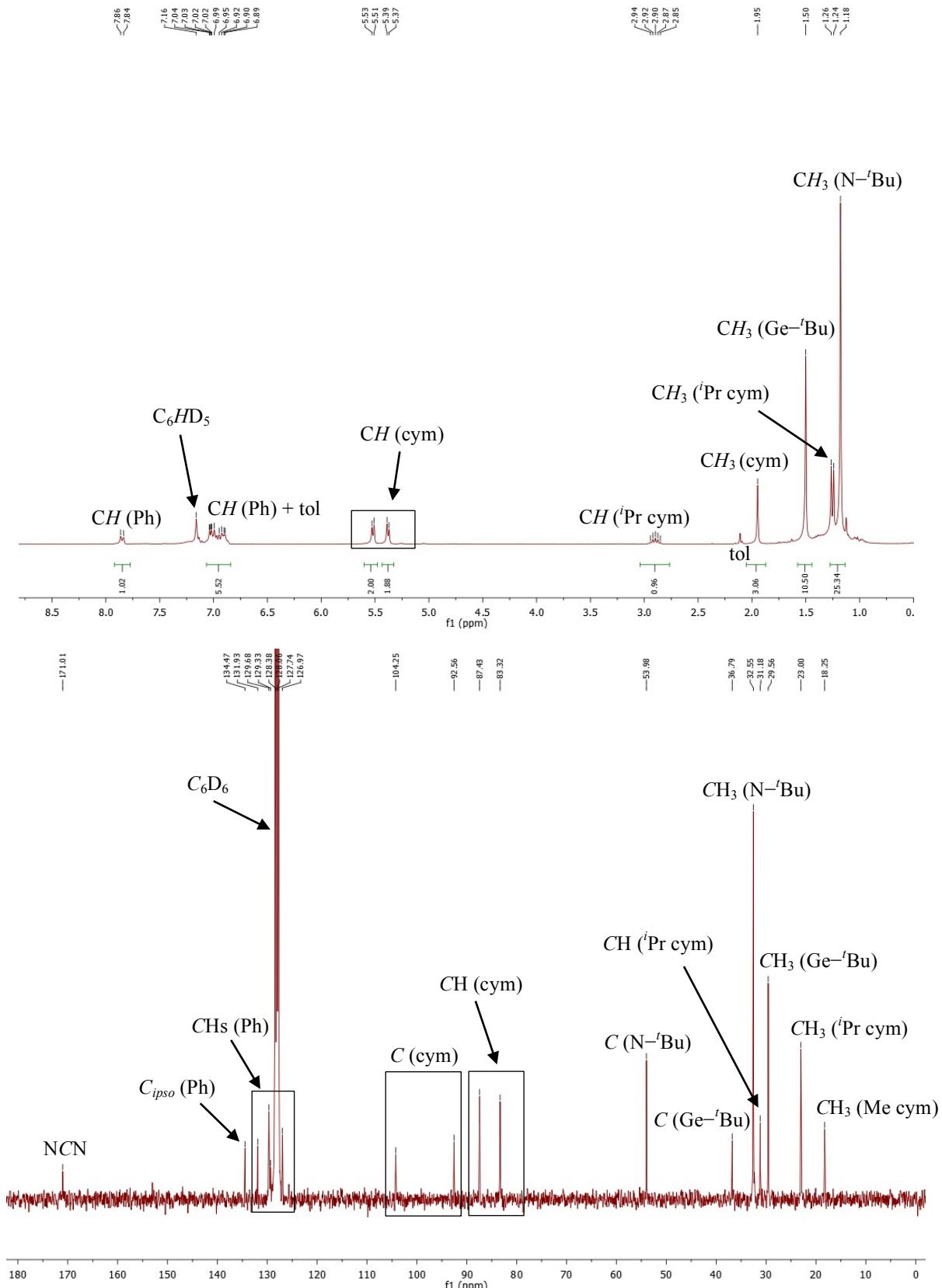
**Figure S4.**  $^{13}\text{C}\{^1\text{H}\}$ -DEPT 135 (75.5 MHz) NMR spectrum of **3** ( $\text{C}_6\text{D}_6$ , 25 °C).



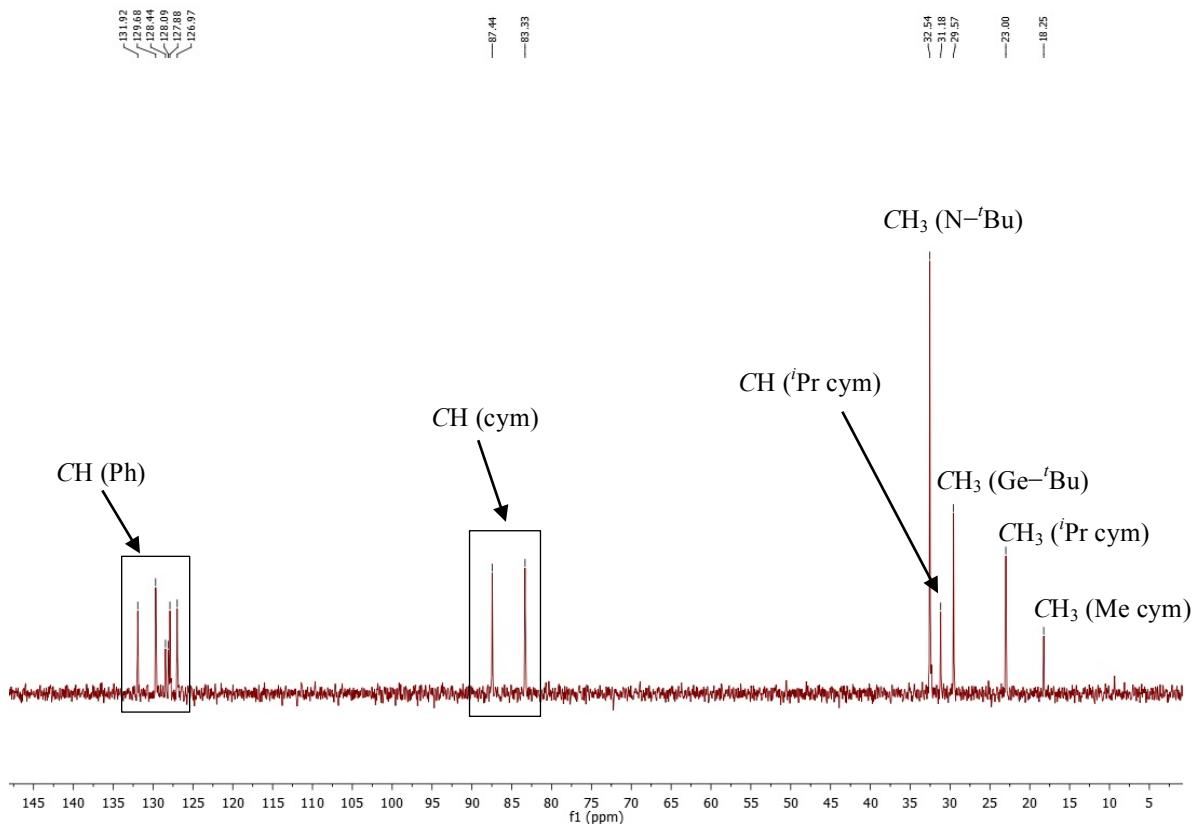
**Figure S5.**  $^1\text{H}$  (top; 300.1 MHz) and  $^{13}\text{C}\{^1\text{H}\}$  (bottom; 75.5 MHz) NMR spectra of **4** ( $\text{C}_6\text{D}_6$ , 25 °C).



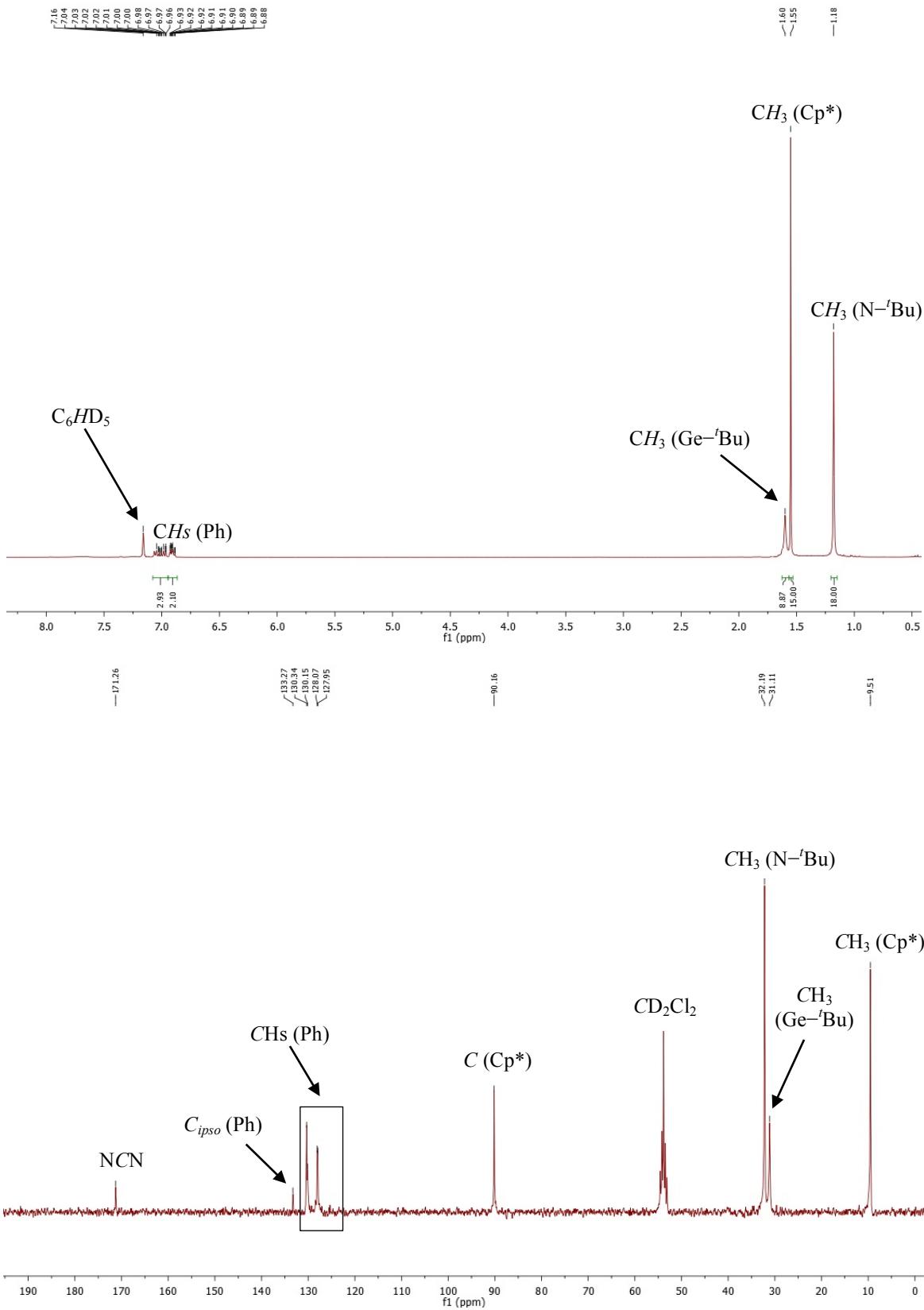
**Figure S6.**  $^{13}\text{C}\{^1\text{H}\}$ -DEPT 135 (75.5 MHz) NMR spectrum of **4** ( $\text{C}_6\text{D}_6$ , 25 °C).



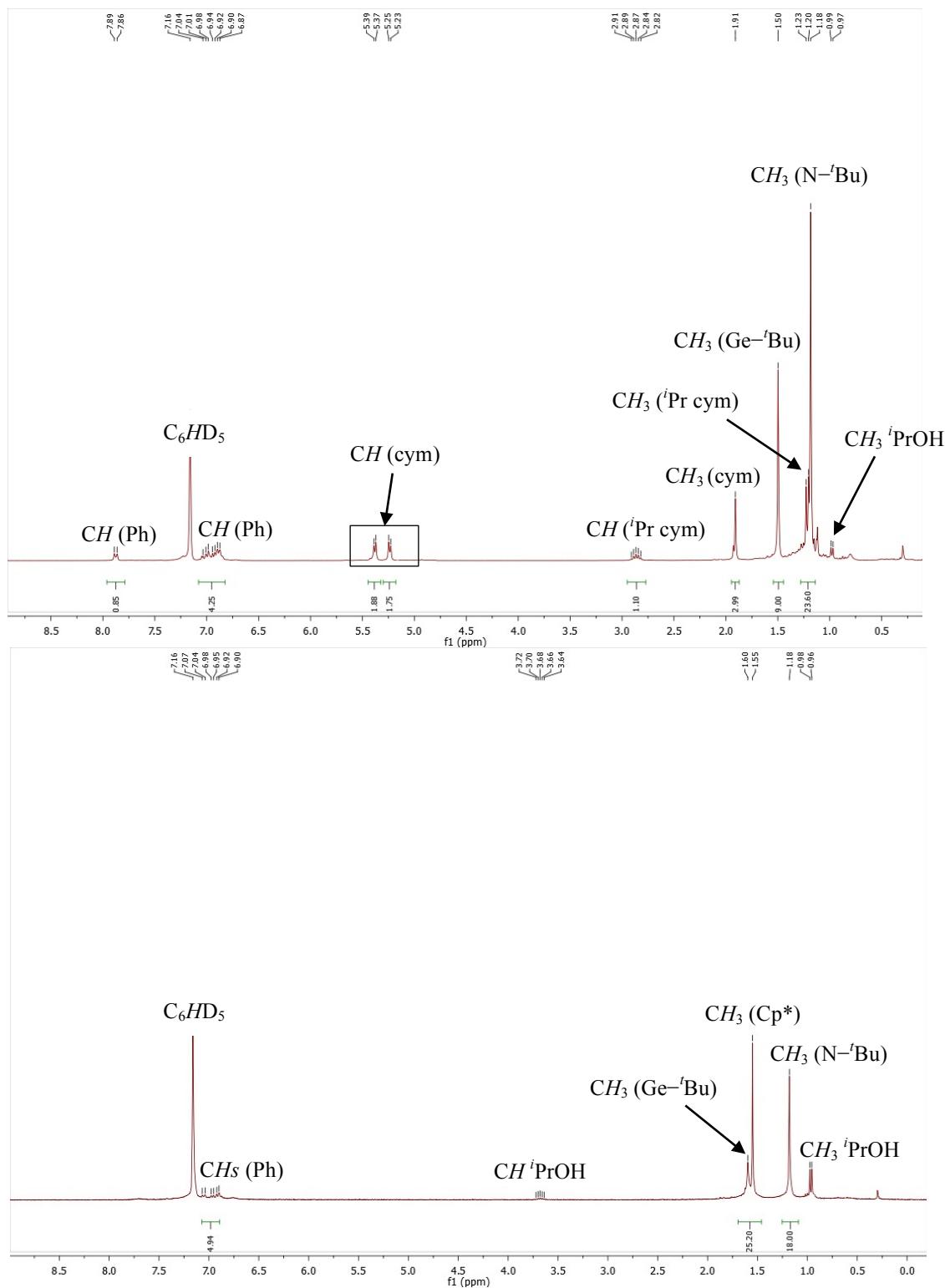
**Figure S7.**  $^1\text{H}$  (top; 300.1 MHz) and  $^{13}\text{C}\{^1\text{H}\}$  (bottom; 75.5 MHz) NMR spectra of **5** ( $\text{C}_6\text{D}_6$ , 25 °C).



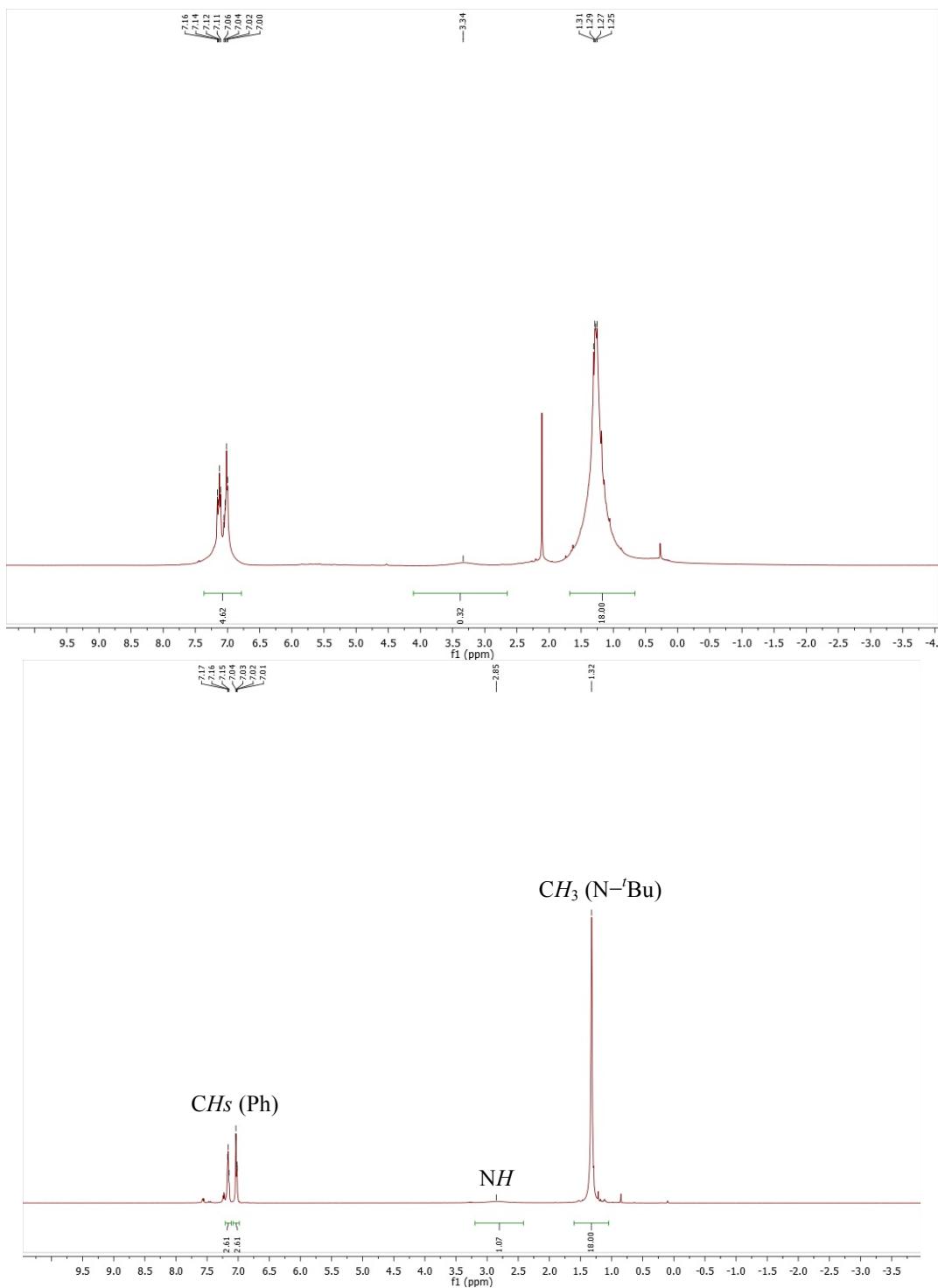
**Figure S8.**  $^{13}\text{C}\{^1\text{H}\}$ -DEPT 135 (75.5 MHz) NMR spectrum of **5** ( $\text{C}_6\text{D}_6$ ,  $25^\circ\text{C}$ ).



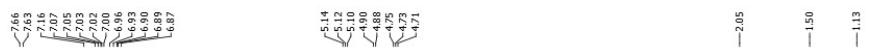
**Figure S9.**  $^1\text{H}$  (top; 300.1 MHz,  $\text{C}_6\text{D}_6$ ) and  $^{13}\text{C}\{^1\text{H}\}$  (bottom; 75.5 MHz,  $\text{CD}_2\text{Cl}_2$ ) NMR spectra of **6** (25 °C).



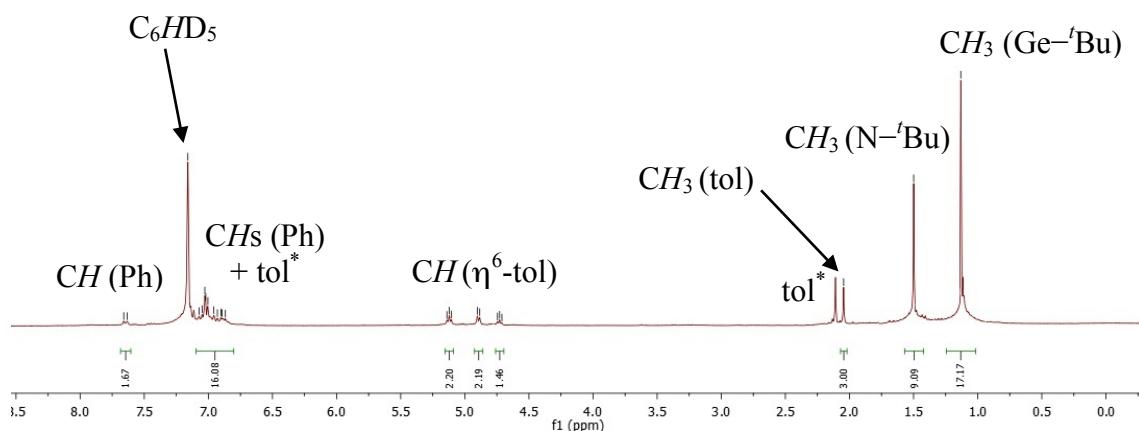
**Figure S10.**  $^1\text{H}$  NMR spectra (300.1 MHz,  $\text{C}_6\text{D}_6$ , 25 °C) of **5** (top) and **6** (bottom) after one day in isopropanol under air.



**Figure S11.**  $^1\text{H}$  NMR spectra (300.1 MHz,  $\text{C}_6\text{D}_6$ , 25 °C) of **3** after one day in toluene under air (top) and of pure  $^t\text{Bu}_2\text{bzamH}$  (bottom).



$\text{tol}^*$  = reaction solvent



**Figure S12.**  ${}^1\text{H}$  NMR spectra (300.1 MHz,  $\text{C}_6\text{D}_6$ , 25 °C) of the solid obtained by solvent evaporation after heating complex **5** in toluene at reflux temperature for 24 h.