

Synthesis of Dibenzosultams by “Transition-Metal Free” Photoinduced Intramolecular Arylation of *N*-aryl-2-halobenzenesulfonamides

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1 - Synthesis of 2-halo-*N*-phenylbenzenesulfonamides

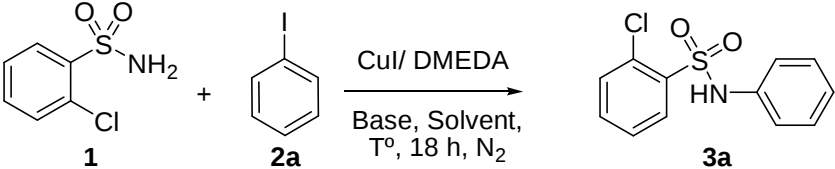
We investigated two methods to obtain *N*-aryl-2-halo-arylsulfonamides: One involves copper-catalyzed *N*-arylation of 2-chlorobenzenesulfonamide with aryl halides (Method A) and the other one involves a known reaction between a substituted benzenesulfonyl chlorides and different anilines (Method B).

1.1. *N*-arylation Copper-catalyzed

1.1.1. Screening of Optimal Conditions

The Copper-catalyzed *N*-arylation was initially optimized employing the cross-coupling reaction of 2-chlorobenzenesulfonamide (**1**) with iodobenzene (**2a**).

Table 1 Optimization of *N*-arylation Copper-catalyzed^a

						
Entry	Conditions					3a Yield (%) ^b
	2a (equiv)	CuI (mol %)	Solvent	Base	Temp (°C)	
1 ^c	1.2	5	CH ₃ CN	K ₂ CO ₃	70 °C	15
2	1.2	5	CH ₃ CN	K ₂ CO ₃	90 °C	25
3	1.2	10	CH ₃ CN	K ₂ CO ₃	90 °C	40
4 ^c	1.5	10	CH ₃ CN	K ₂ CO ₃	90 °C	45
5	2	10	CH ₃ CN	K ₂ CO ₃	90 °C	82
6 ^c						78
7	2	10	CH ₃ CN	KOH	90 °C	26
8	2	10	CH ₃ CN	<i>t</i> -BuOK	90 °C	<5
9	2	10	Toluene	K ₂ CO ₃	90 °C	<5
10	2	10	DMF	K ₂ CO ₃	90 °C	6

^aReactions carried out under N₂ with 0.6 mmol of **1** (2-chlorobenzenesulfonamide), 0.72 – 1.2 mmol of **2a** (iodobenzene), 0.03 – 0.06 mmol of CuI, 0.3 mmol of DMEDA (N,N'-Dimethylethylenediamine), 1.8 mmol of base in 4 mL of solvent. ^bYields were determined by GC (internal standard method). ^cIsolated yields after column chromatography.

The reaction was also tested with microwave irradiation making no progress even heating at 120 °C for 30 min obtaining only traces of the corresponding product **3a**.

Microwave-induced reactions were performed in a single mode instrument equipped with a noncontact infrared temperature sensor, direct pressure control system

for measuring the pressure of the reaction vessel contents and a cooling system by compressed air. In the method a temperature controlled reaction at 120°C, with a maximum power of 150 W.

2. Synthesis of Dibenzosultams

2.1 Uv-vis spectra for 3a and the corresponding anion

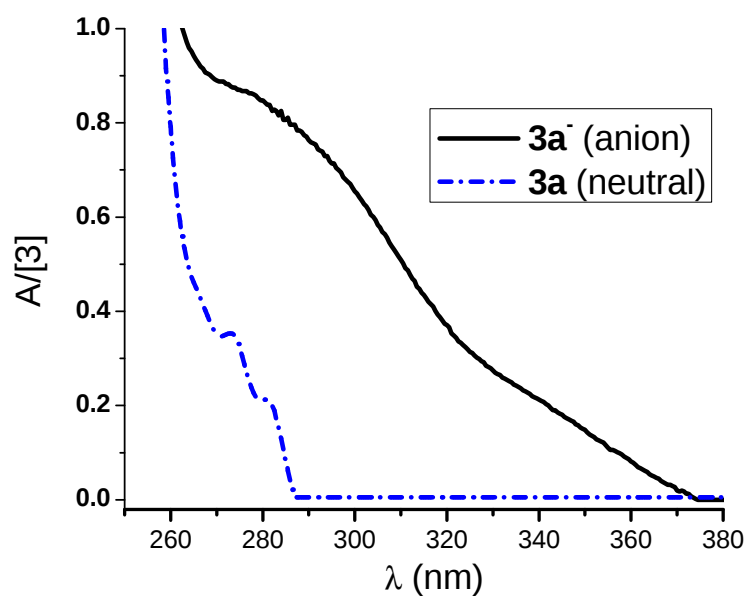
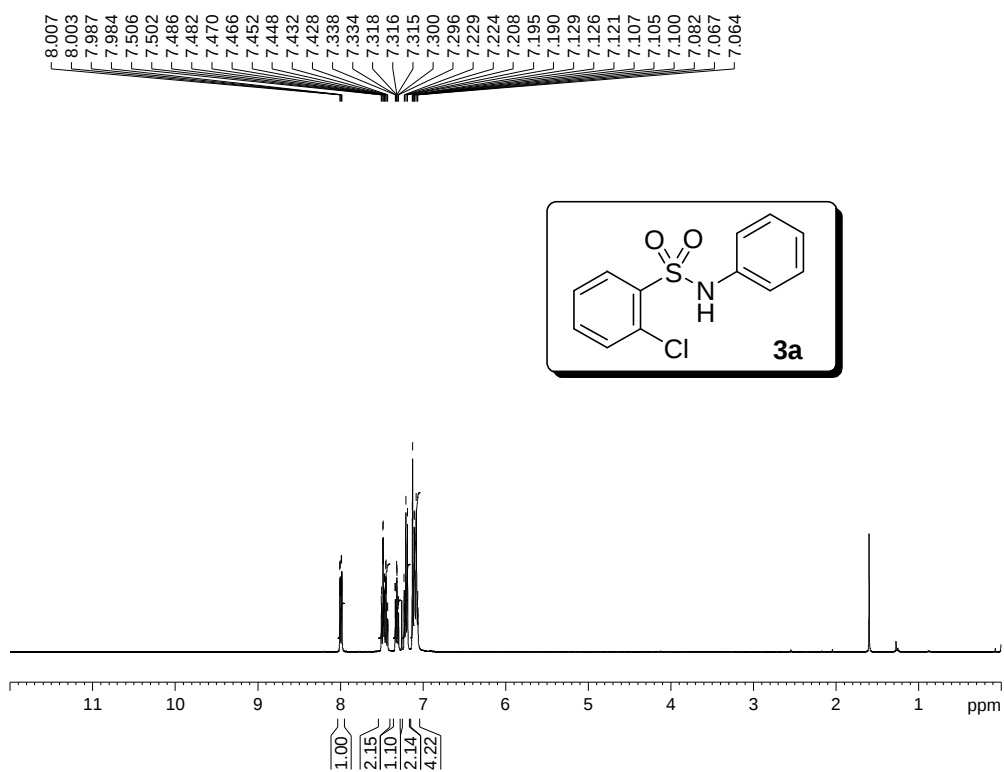
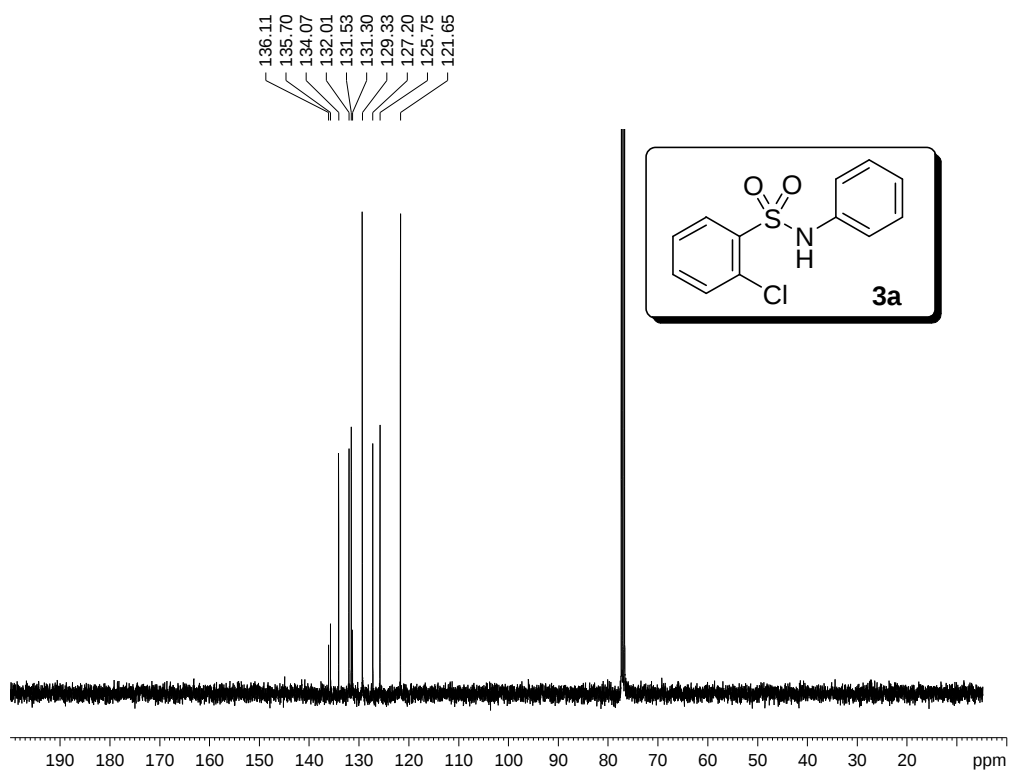
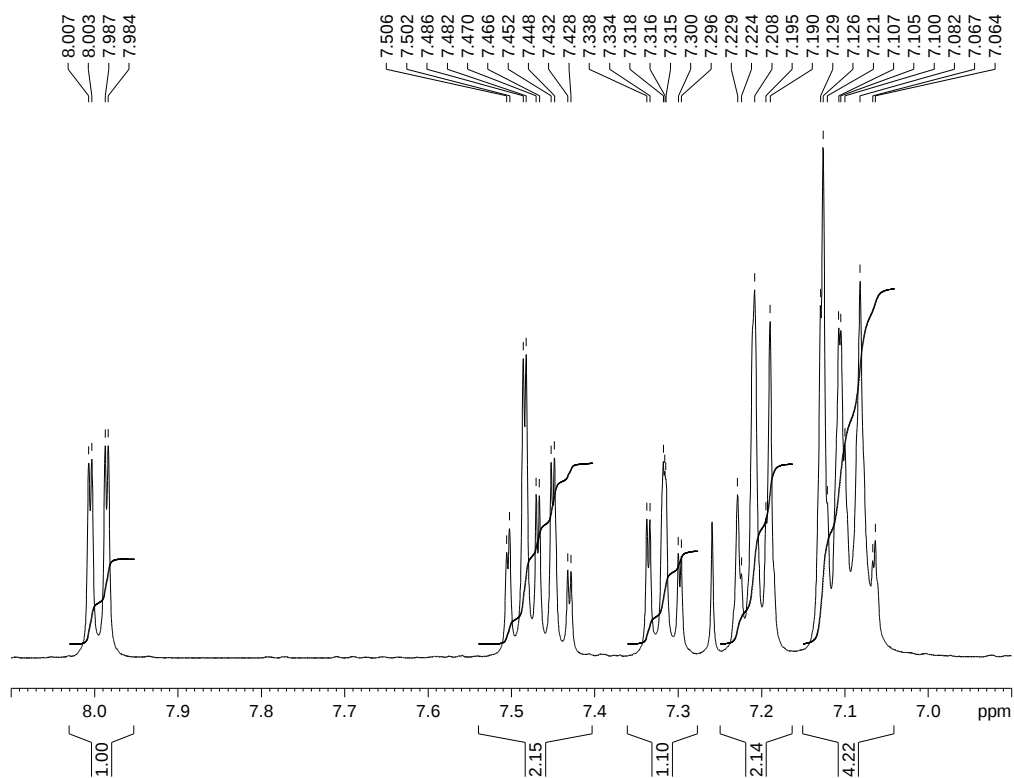


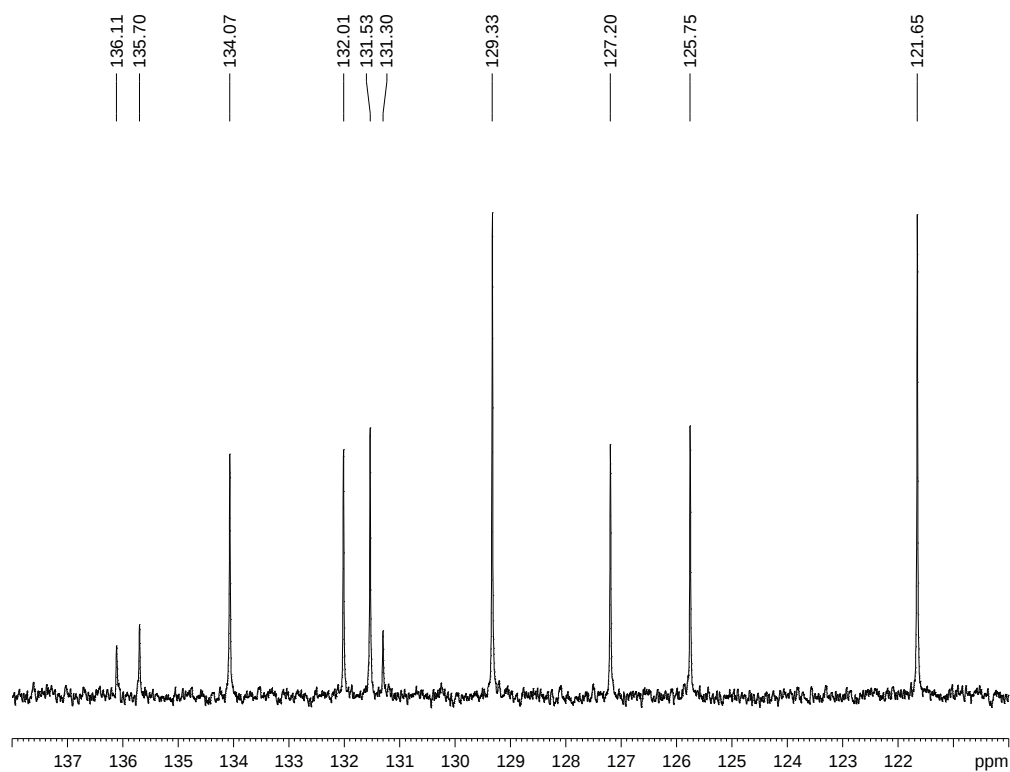
Figure 1: Uv-vis spectra for compounds **3a** (7.4×10^{-5} M) and anion derivative **3a⁻**. Spectra were performed under N₂ in DMSO. To form the anion, a solution of *t*-BuOK 1.35×10^{-3} M was added and the mixture was incubated for 5 min. Solid line: anion. Dashed line: neutral.

3. Copies of ^1H and ^{13}C NMR Spectra of substrates (3a-r)

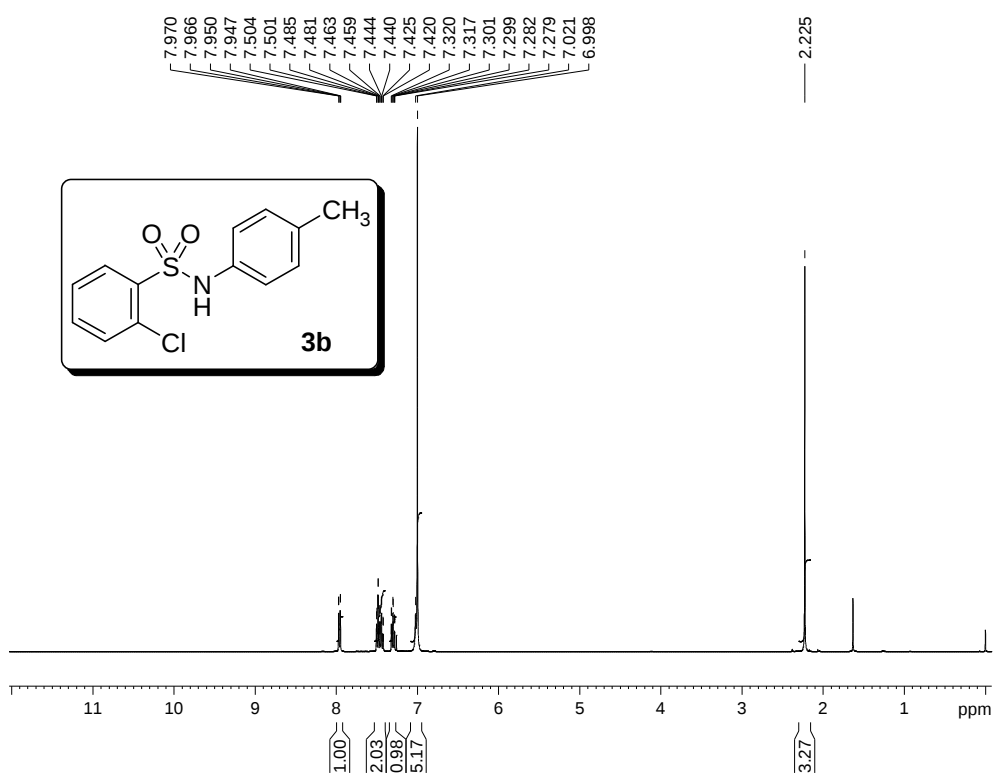
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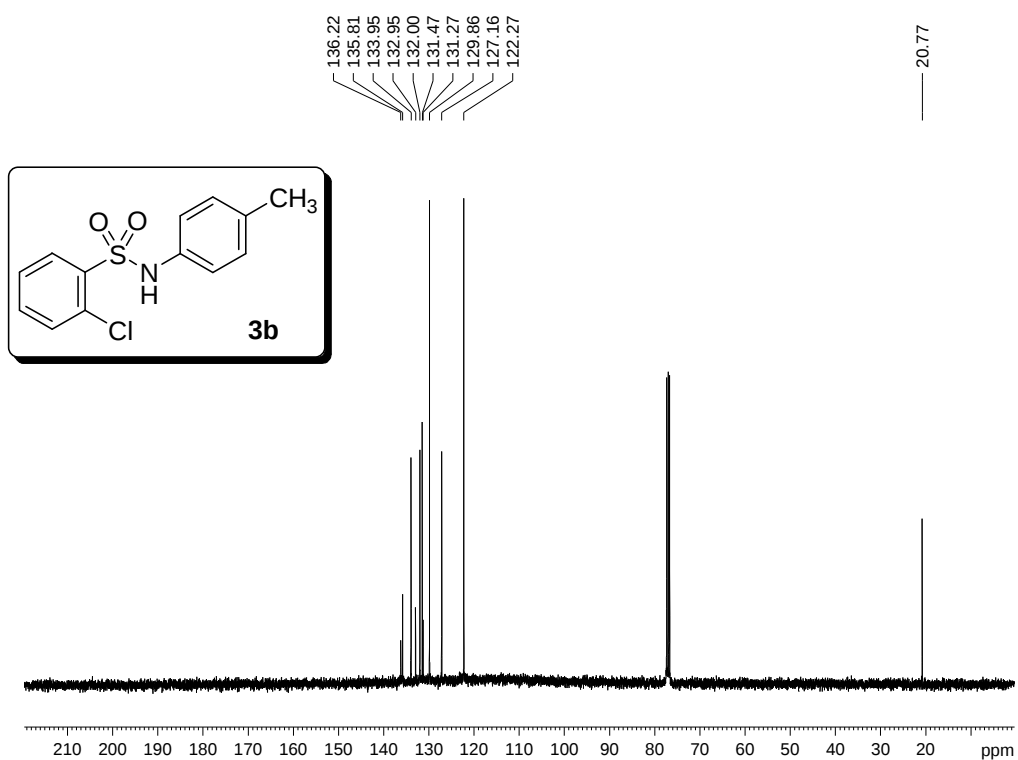
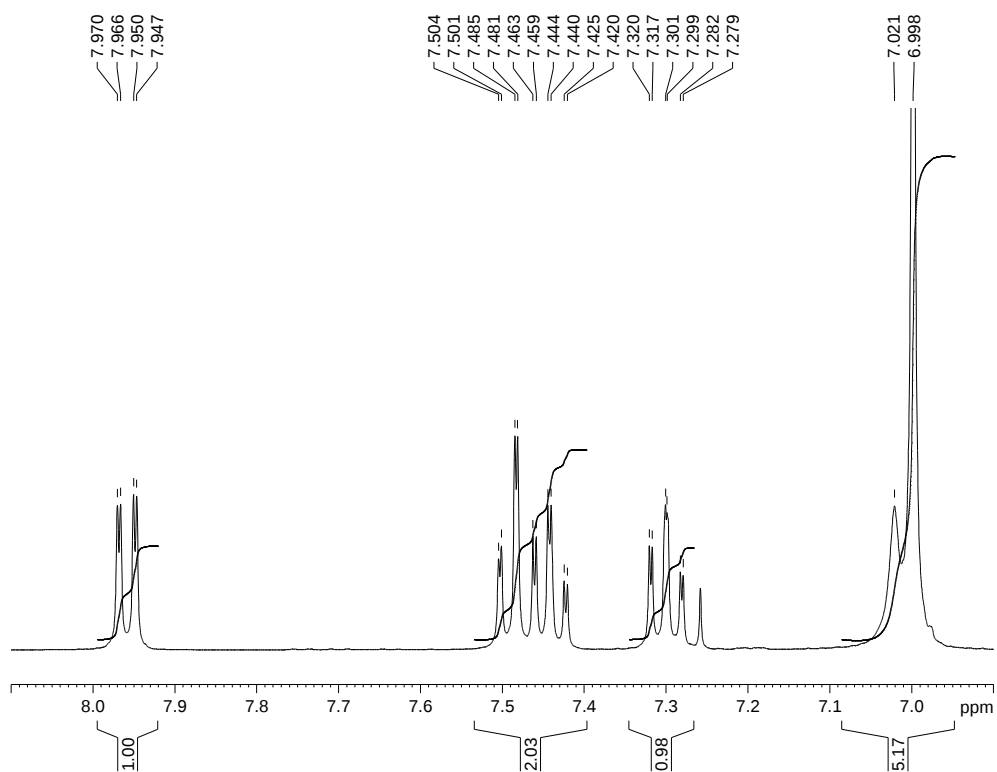


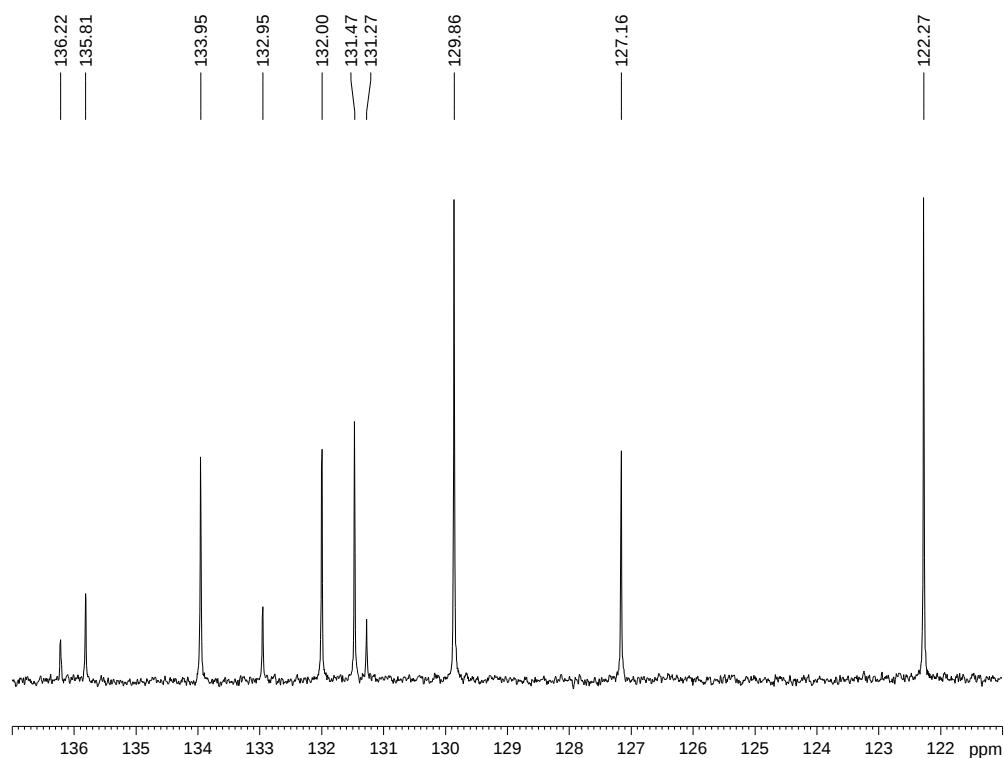




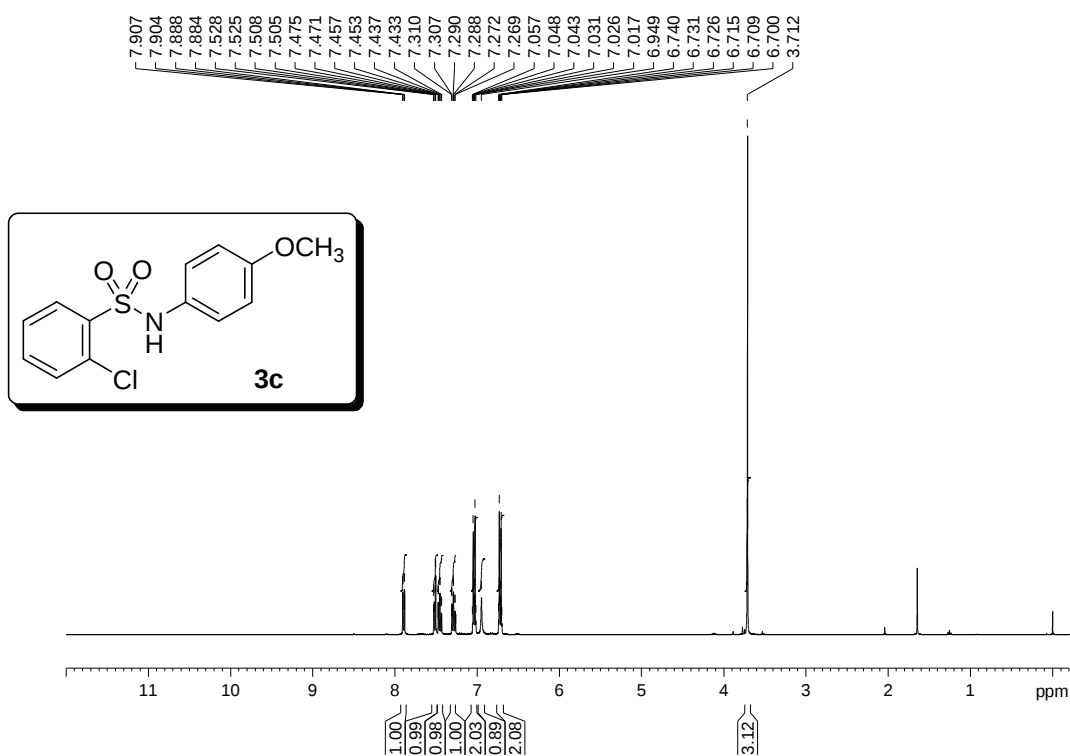
2-Chloro-N-(p-tolyl)benzenesulfonamide (**3b**)

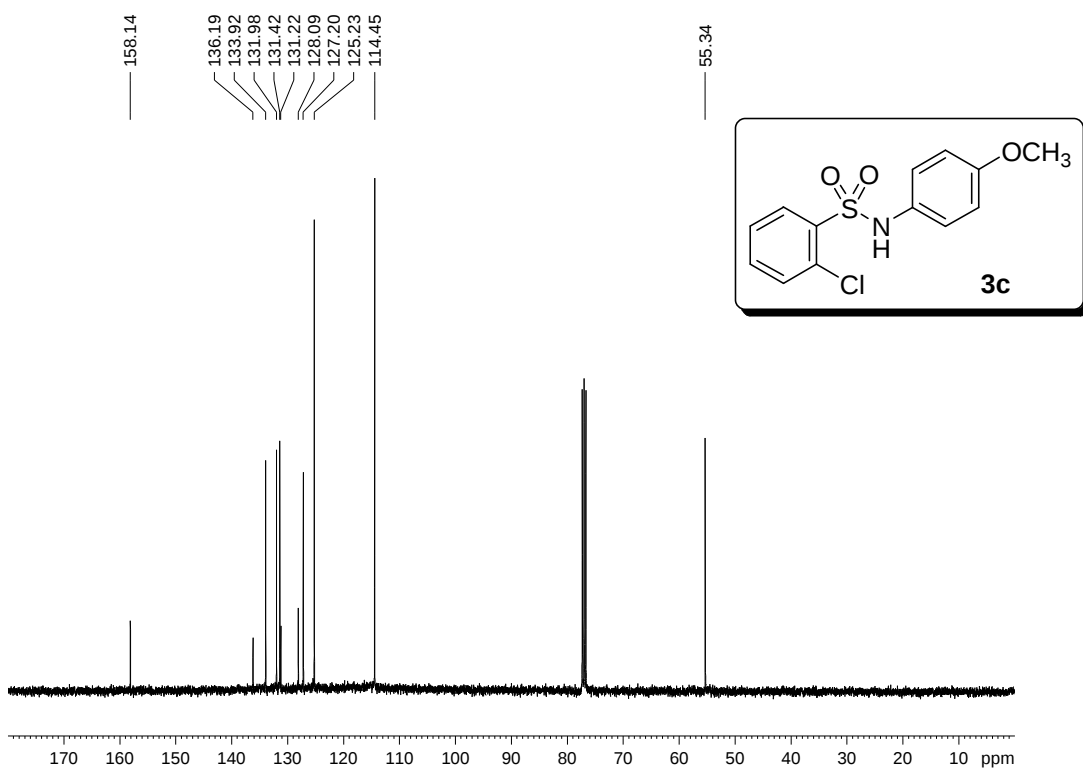
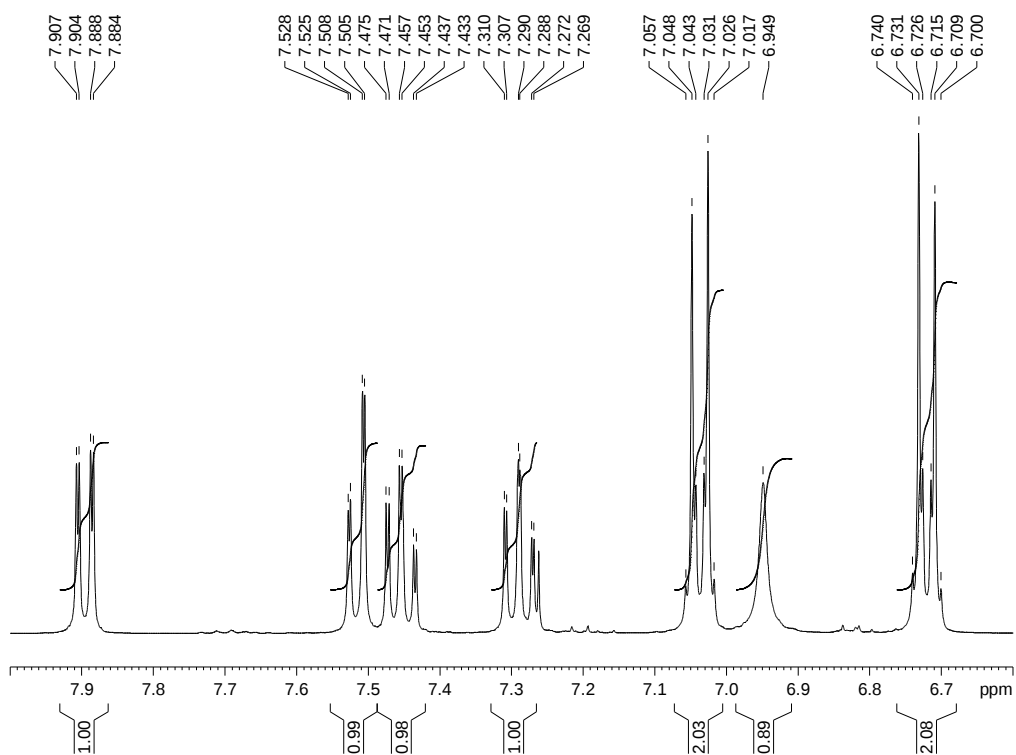


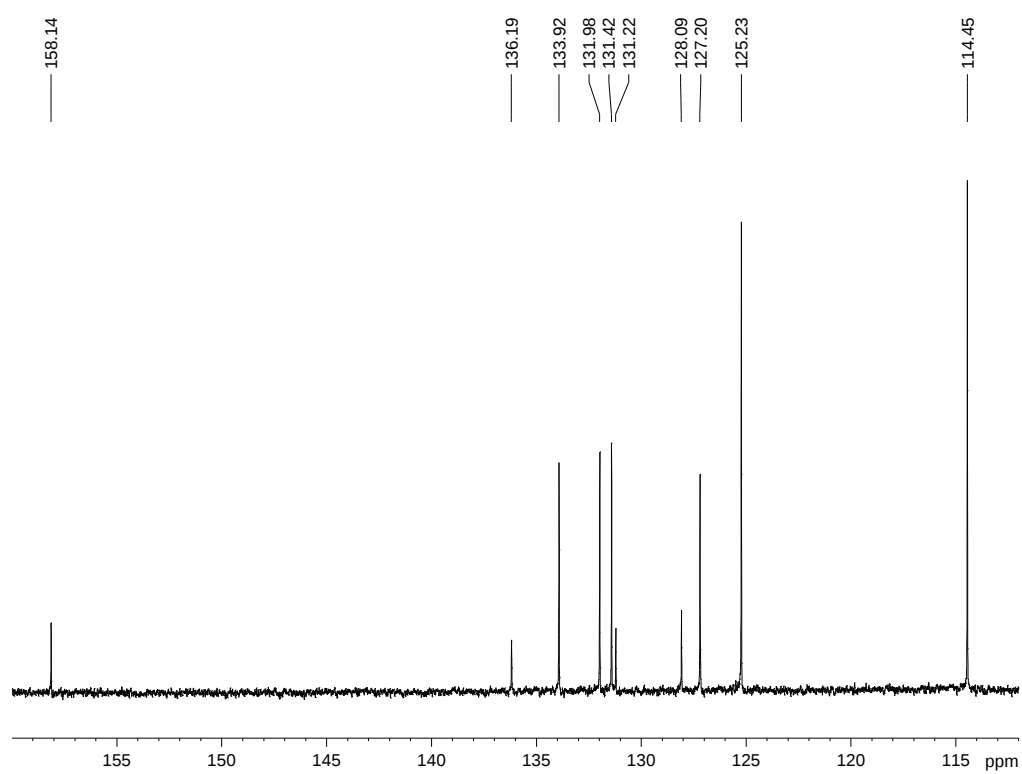




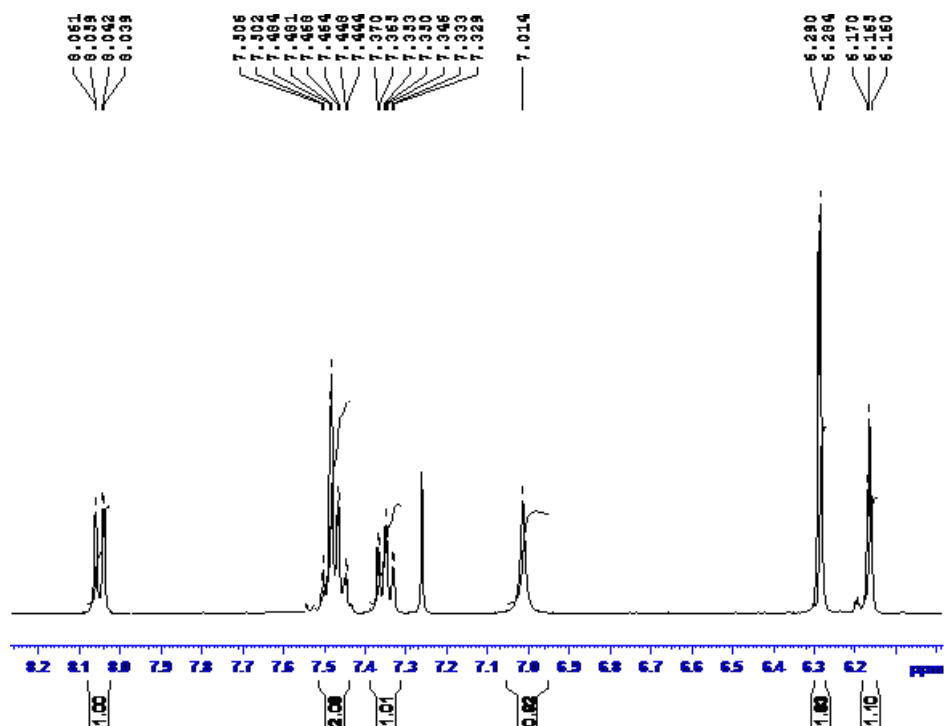
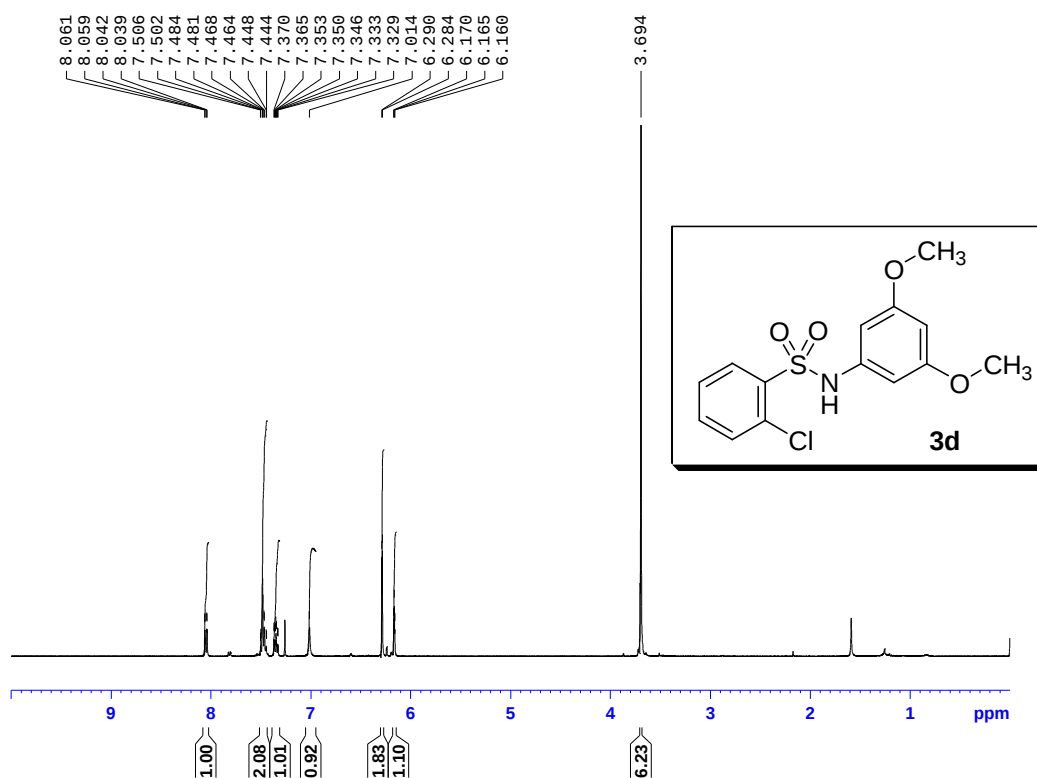
2-Chloro-N-(4-methoxyphenyl)benzenesulfonamide (3c)

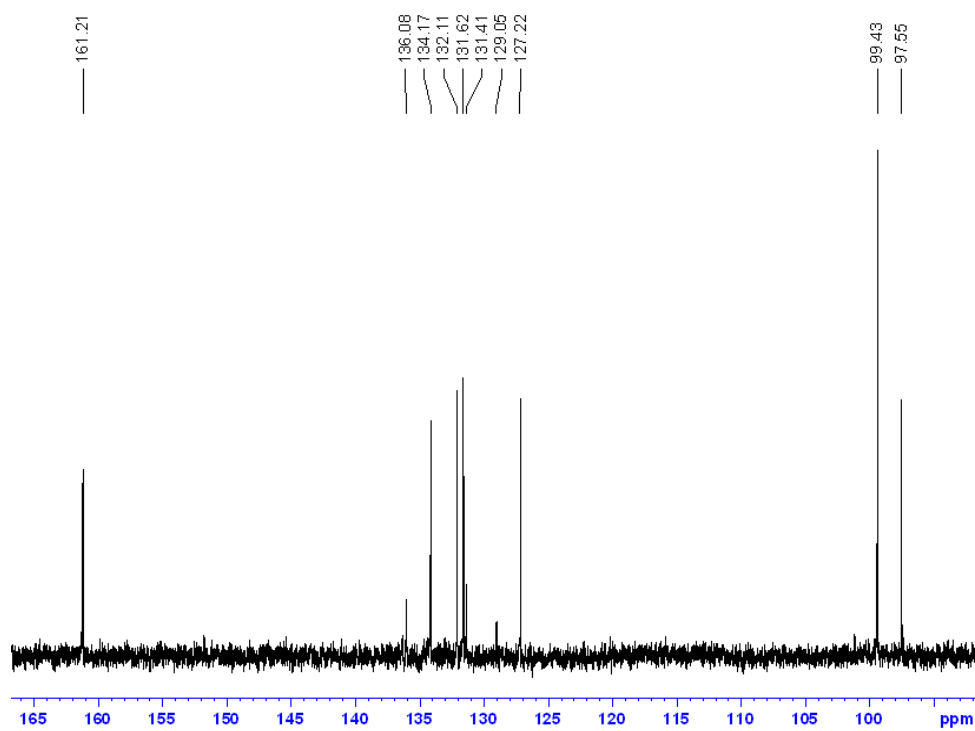
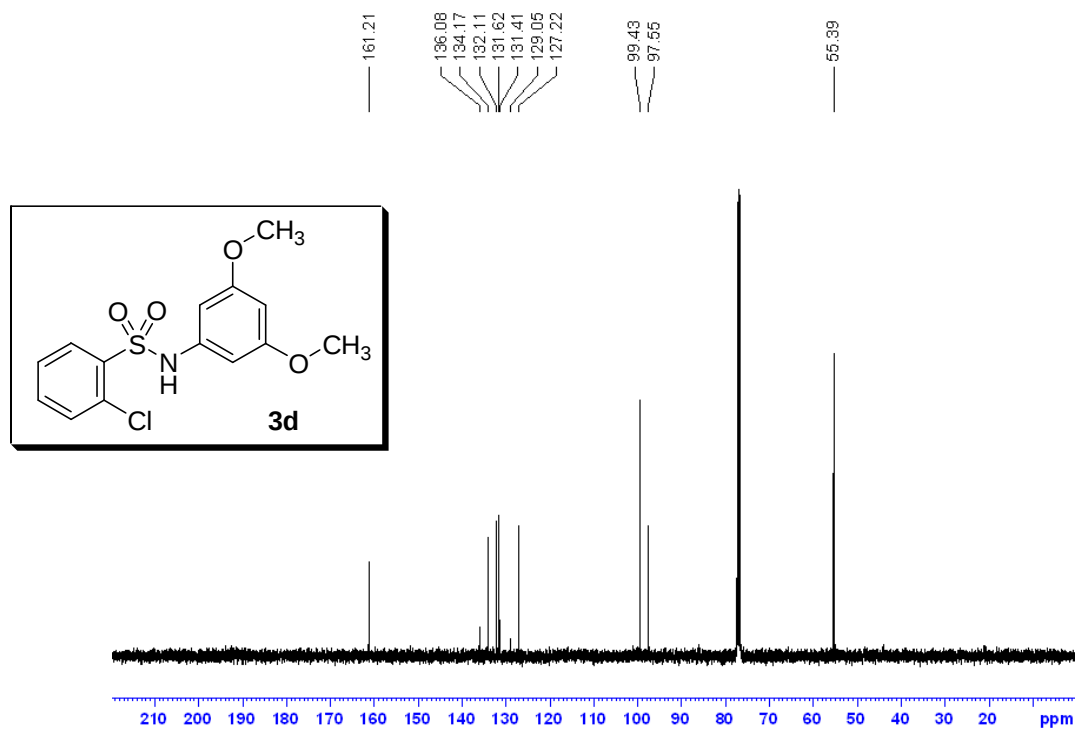




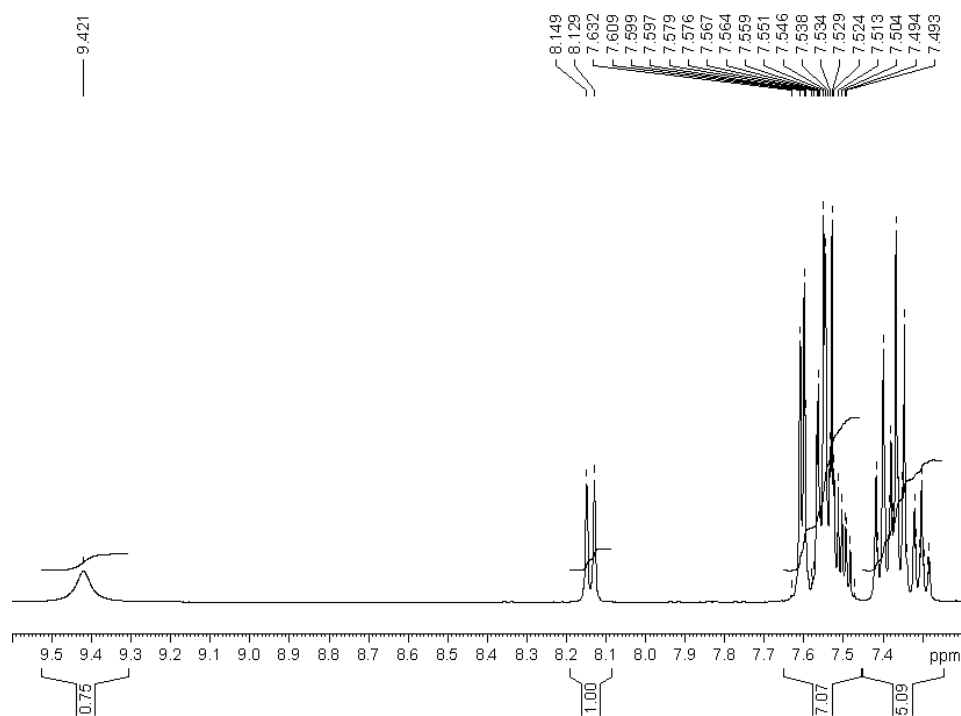
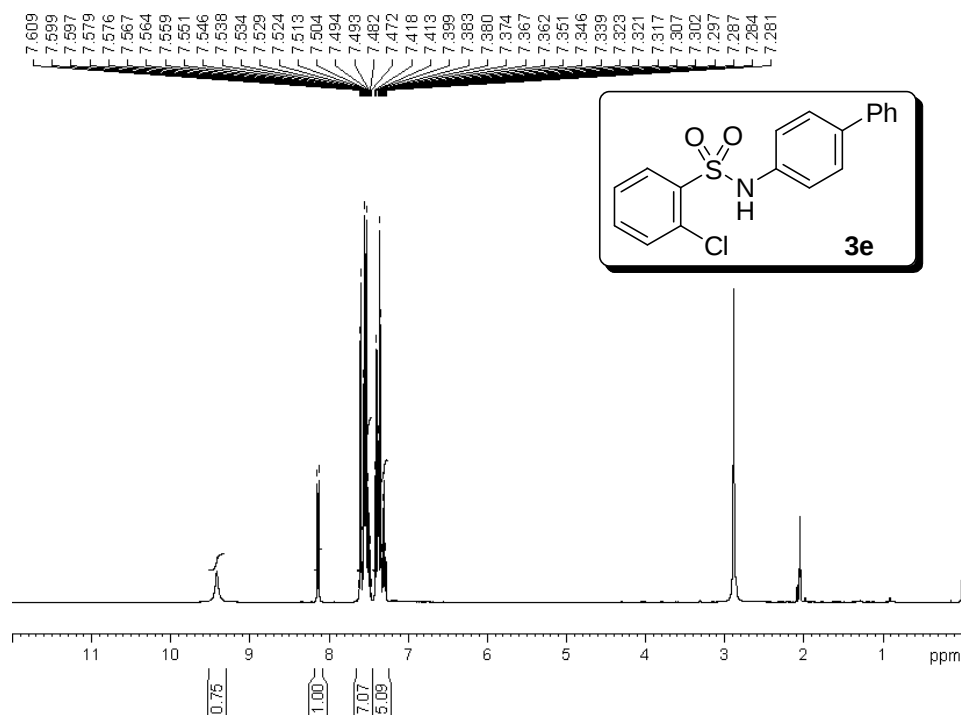


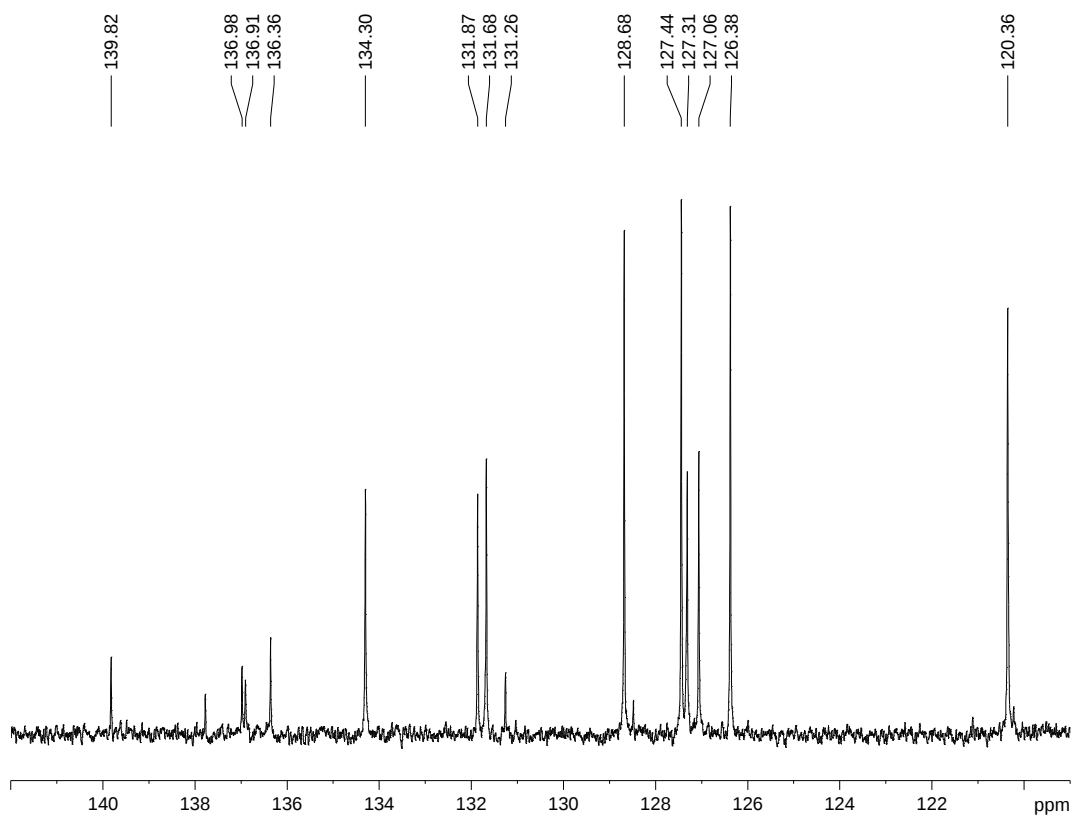
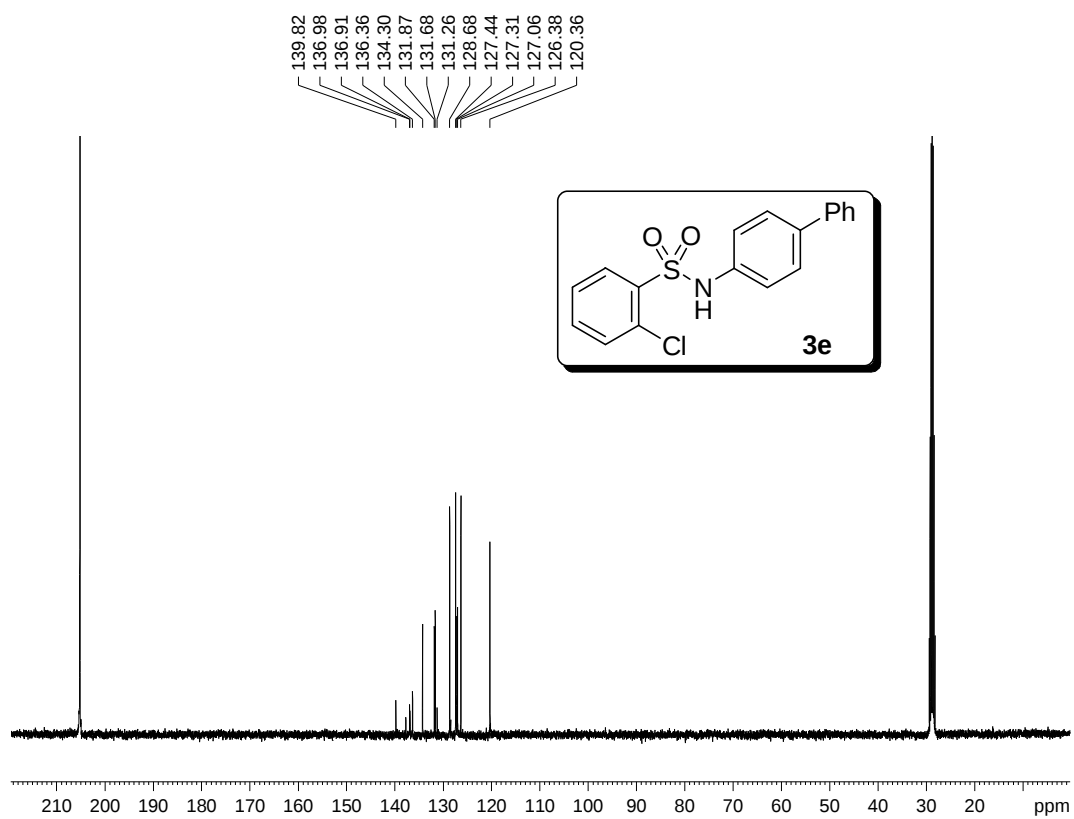
2-Chloro-N-(3,5-di-methoxyphenyl)benzenesulfonamide (3d)



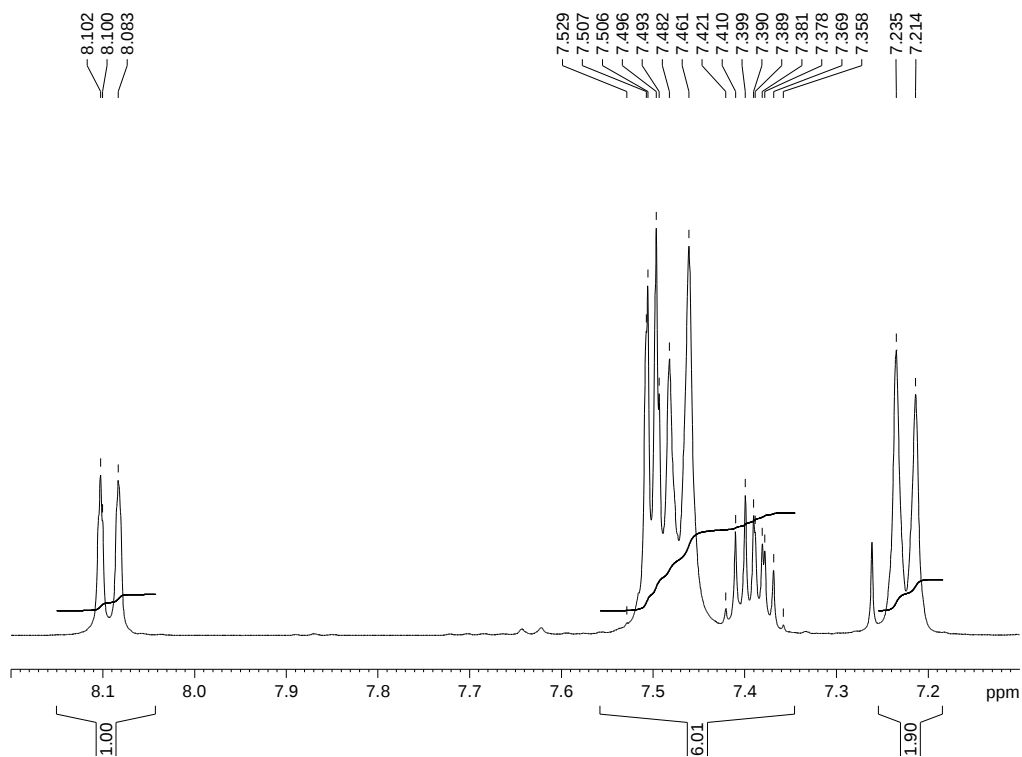
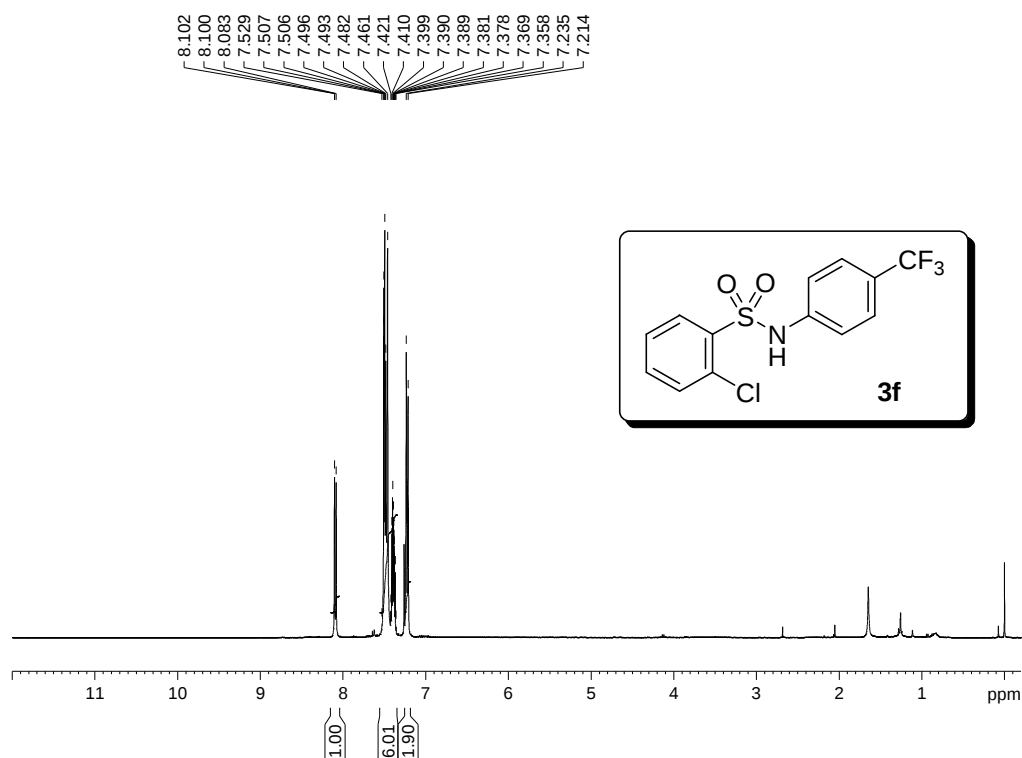


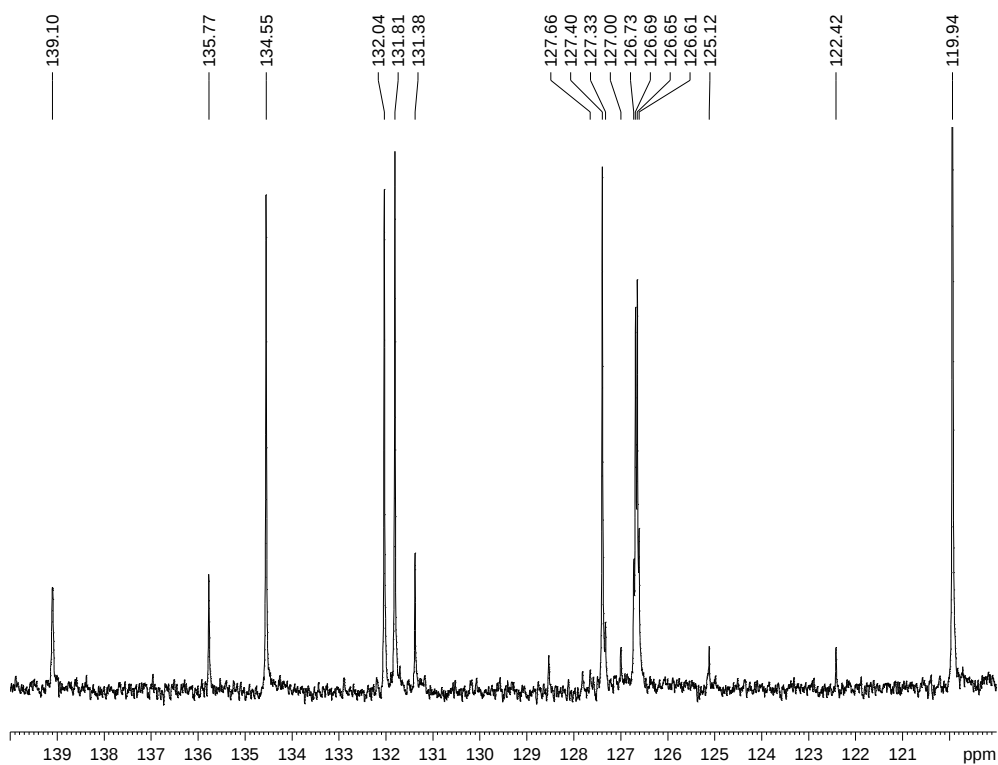
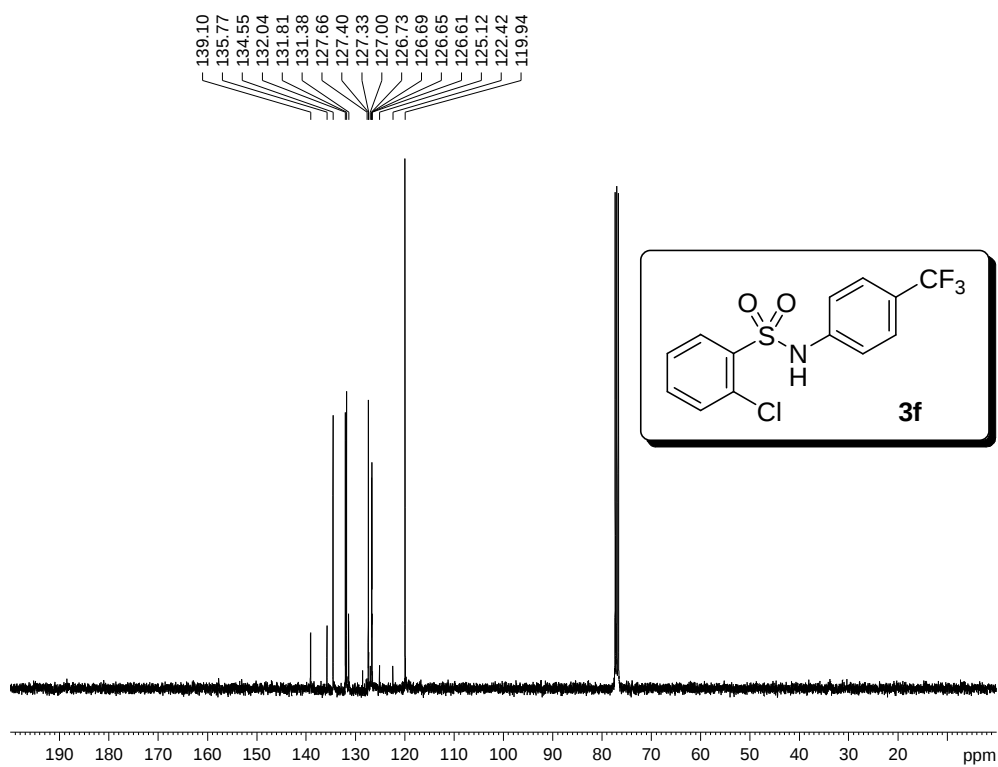
N-([1,1'-Biphenyl]-4-yl)-2-chlorobenzenesulfonamide (**3e**)



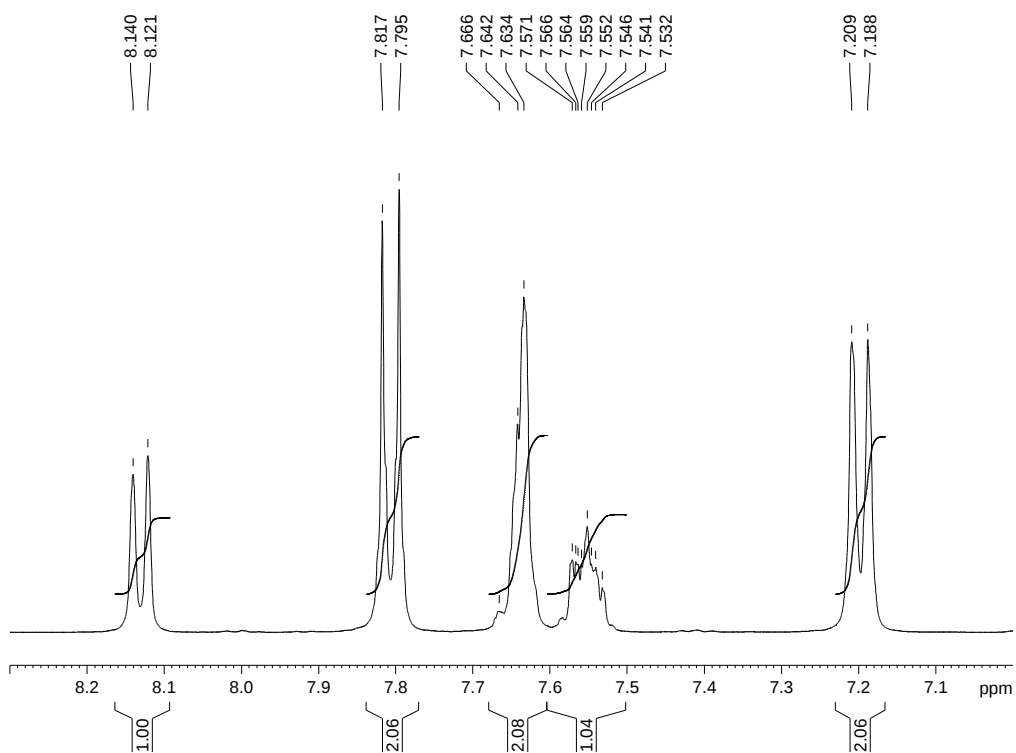
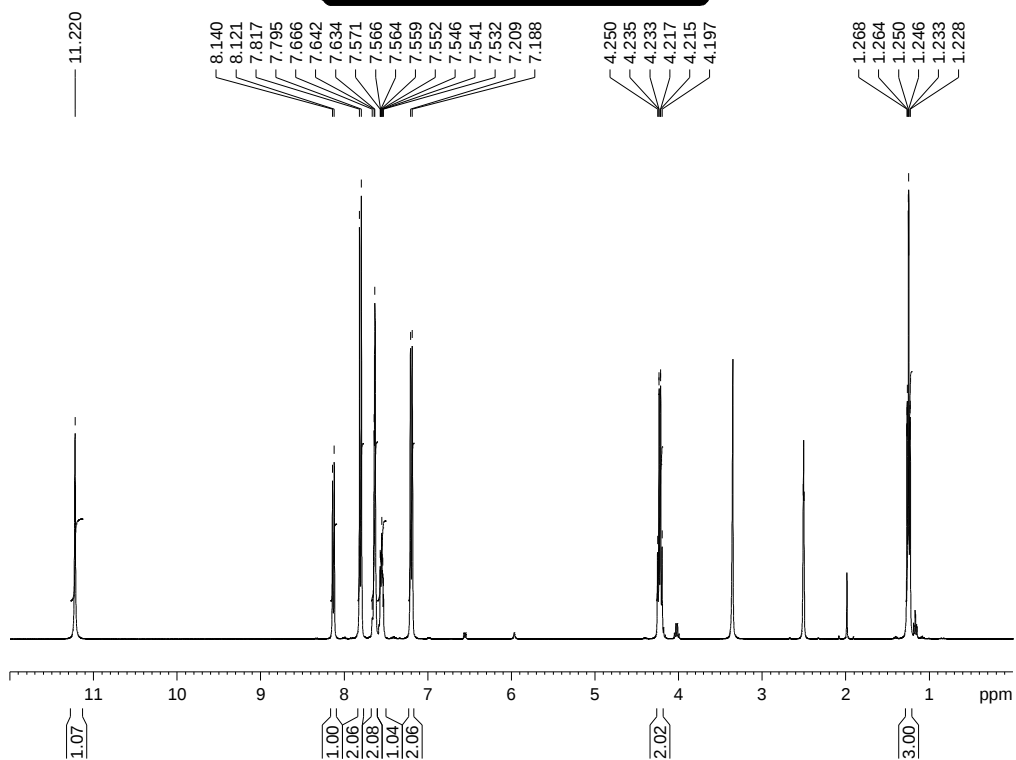
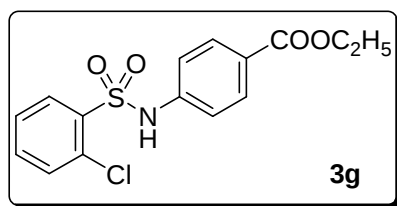


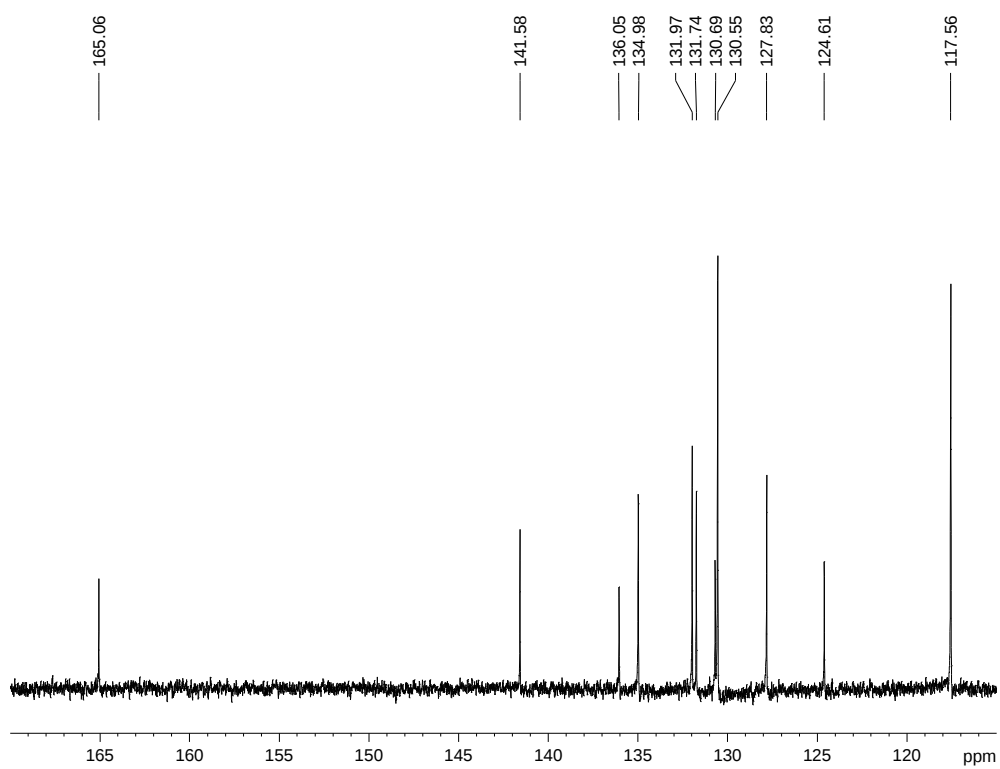
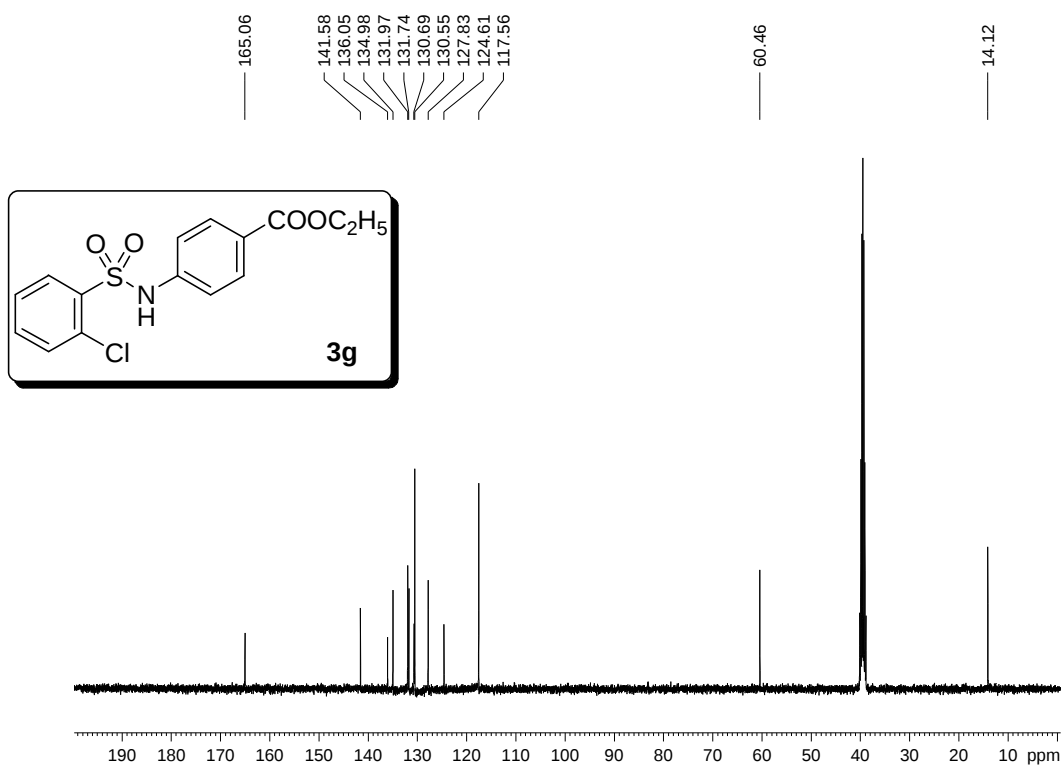
2-Chloro-N-(4-(trifluoromethyl)phenyl)benzenesulfonamide (**3f**)



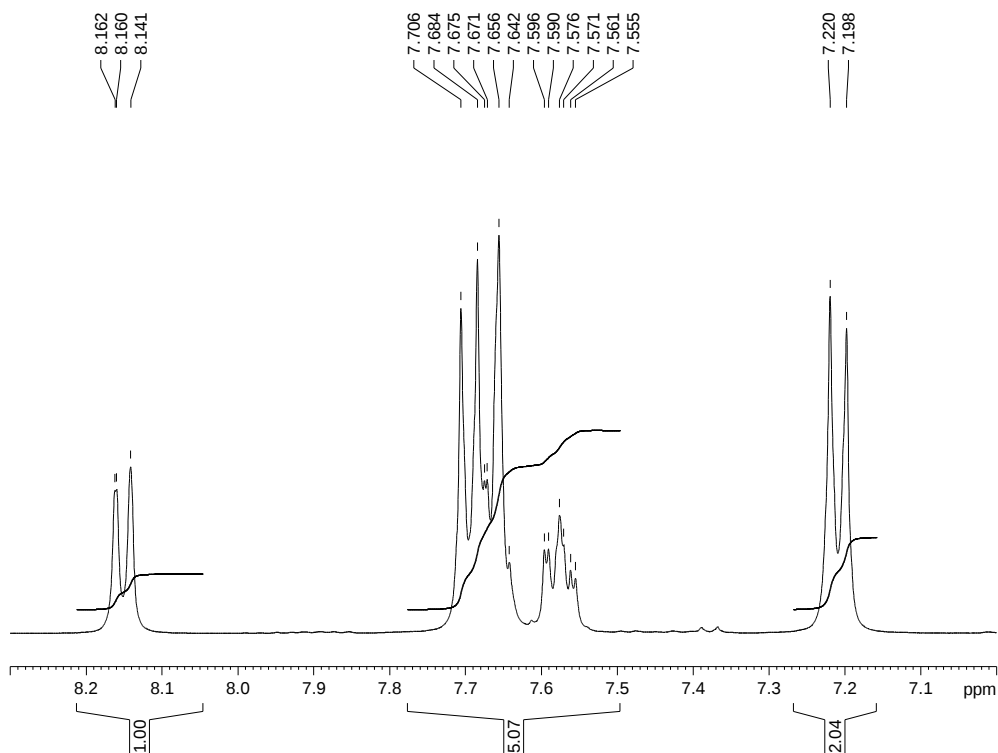
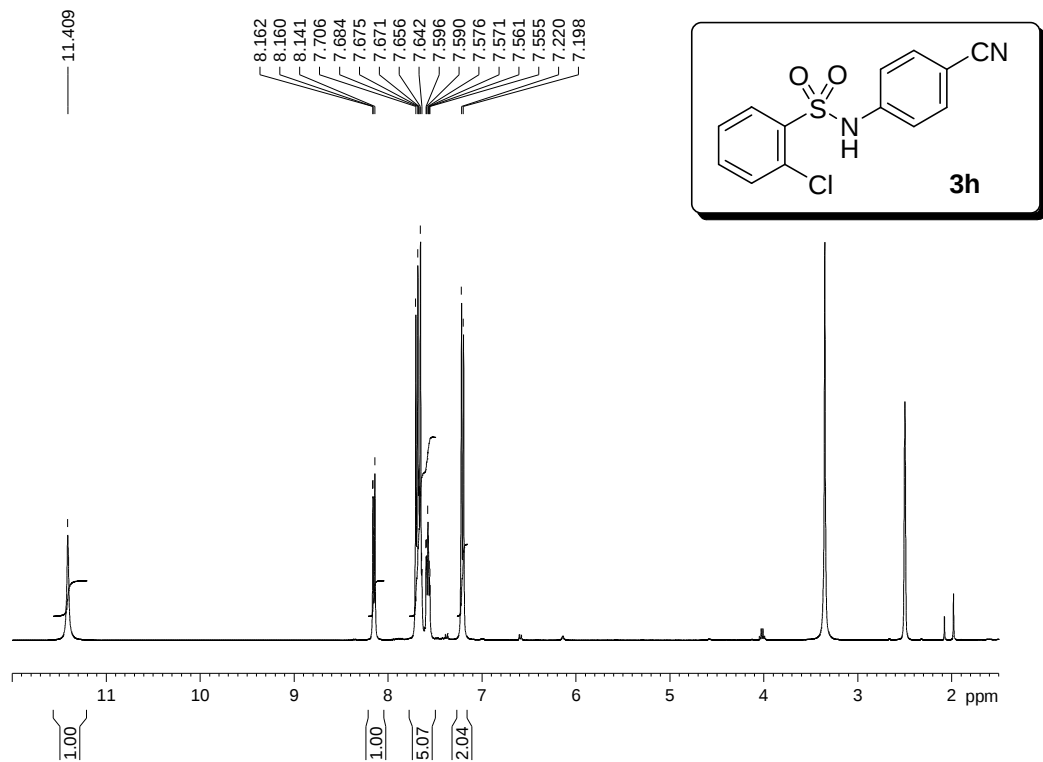


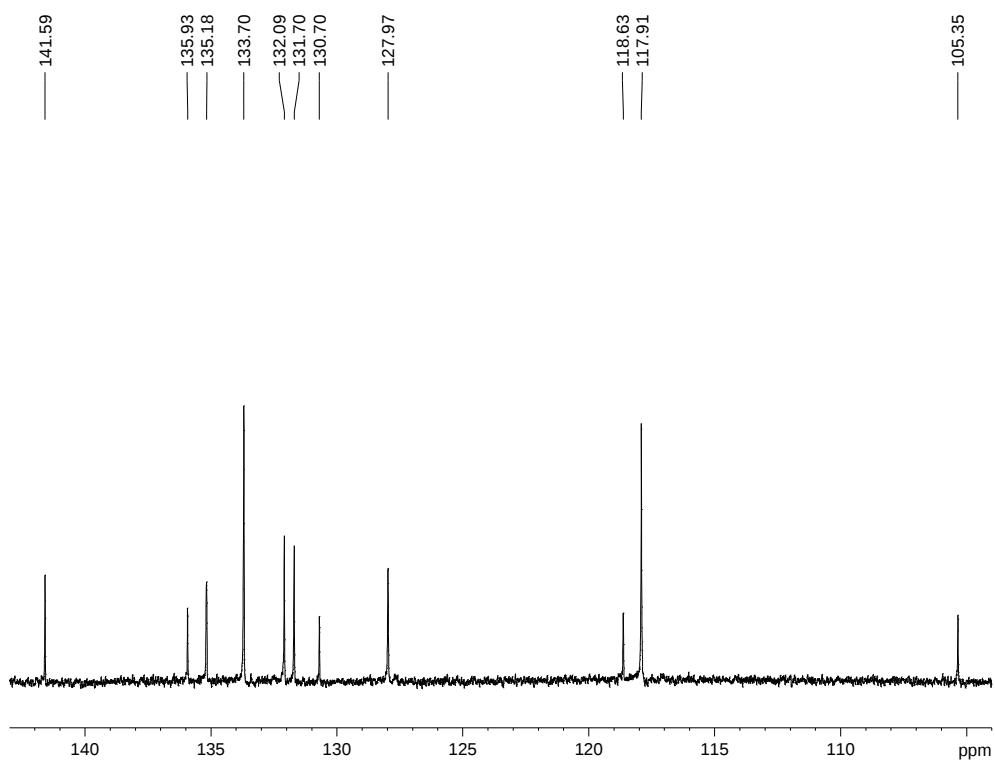
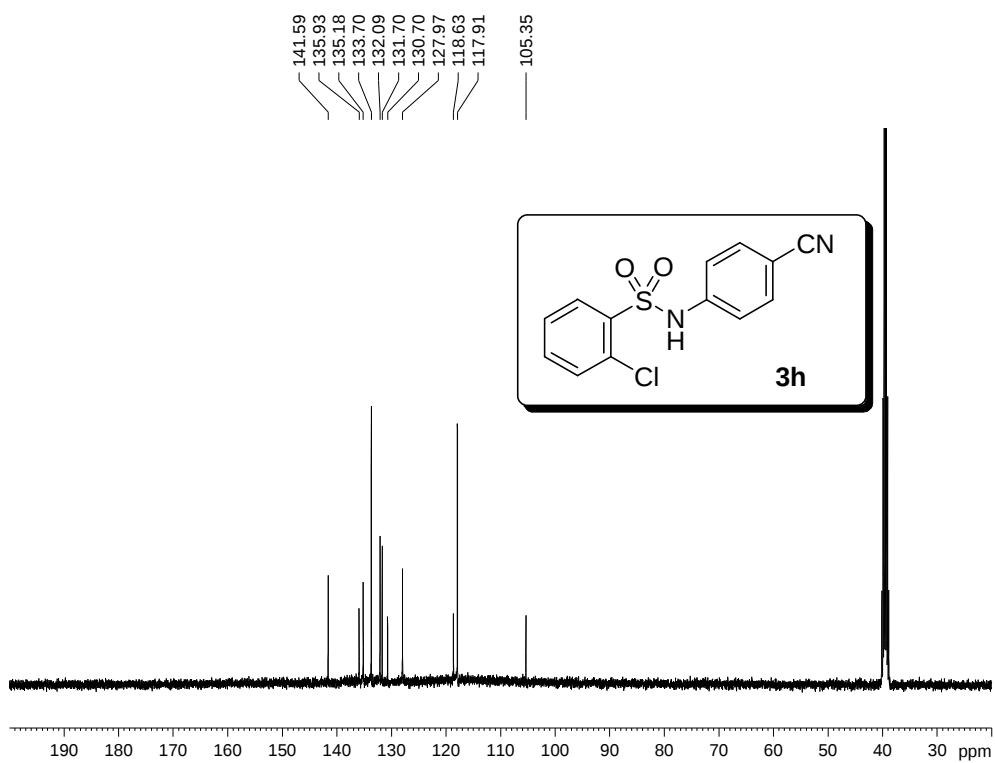
Ethyl 4-(2-chlorophenylsulfonamido)benzoate (**3g**)



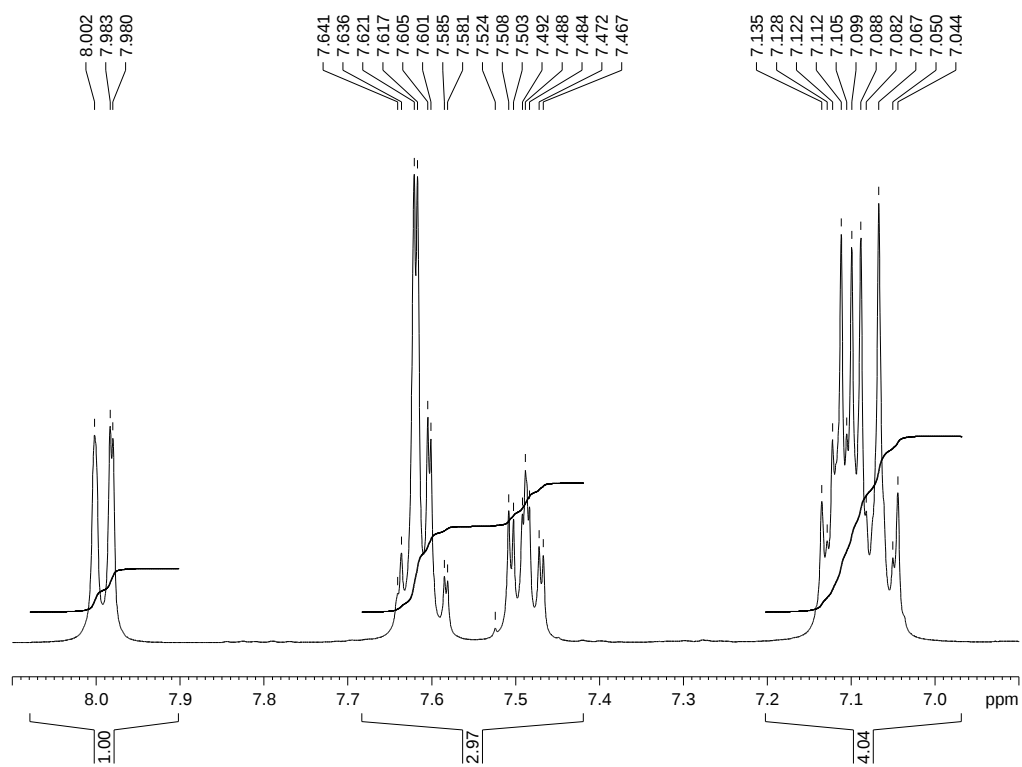
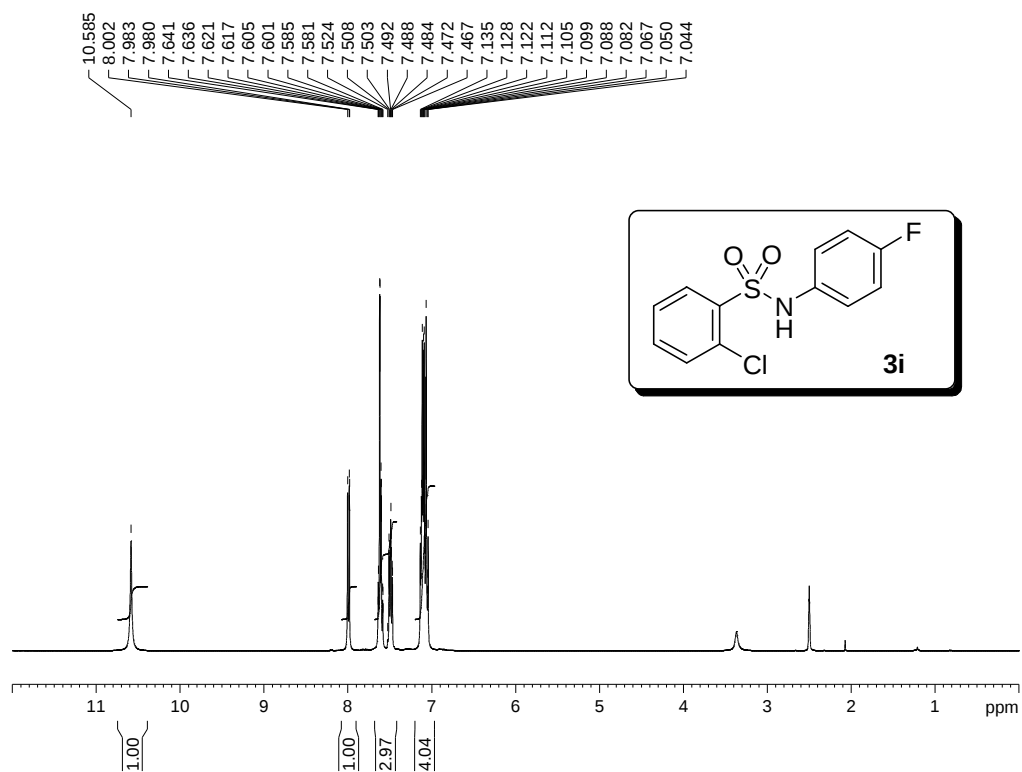


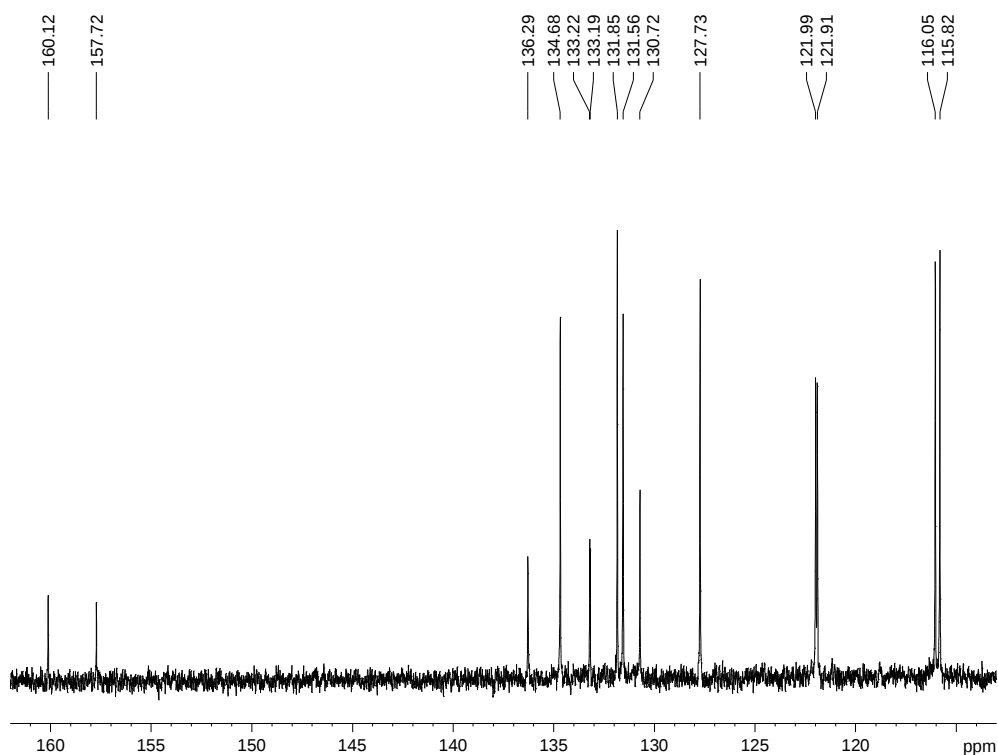
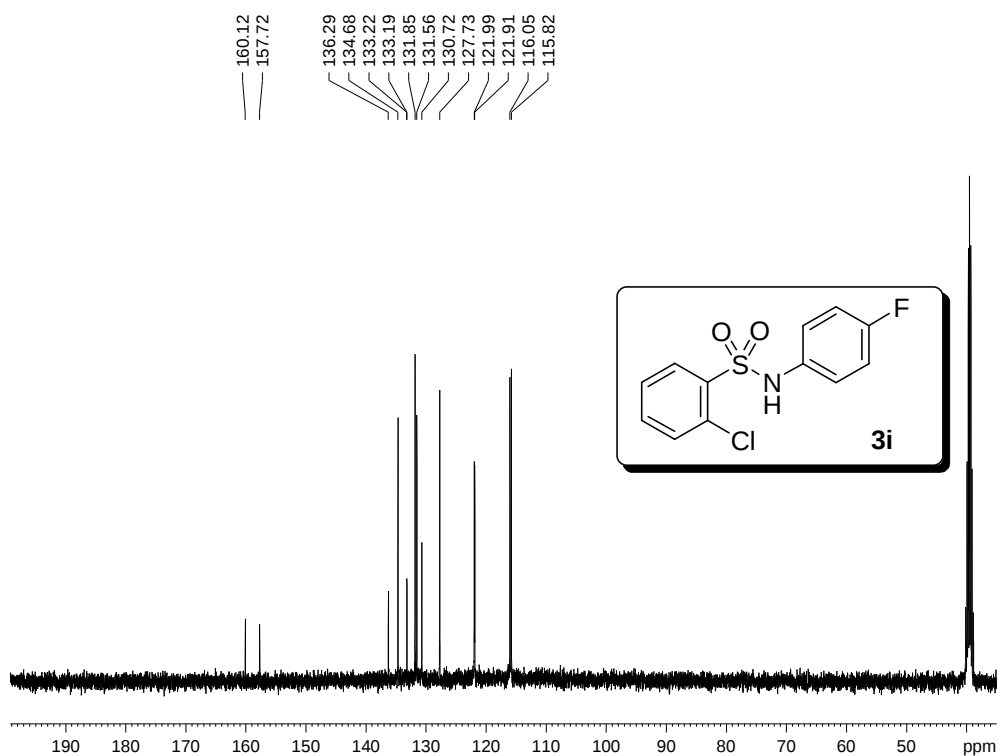
2-Chloro-N-(4-cyanophenyl)benzenesulfonamide (**3h**)



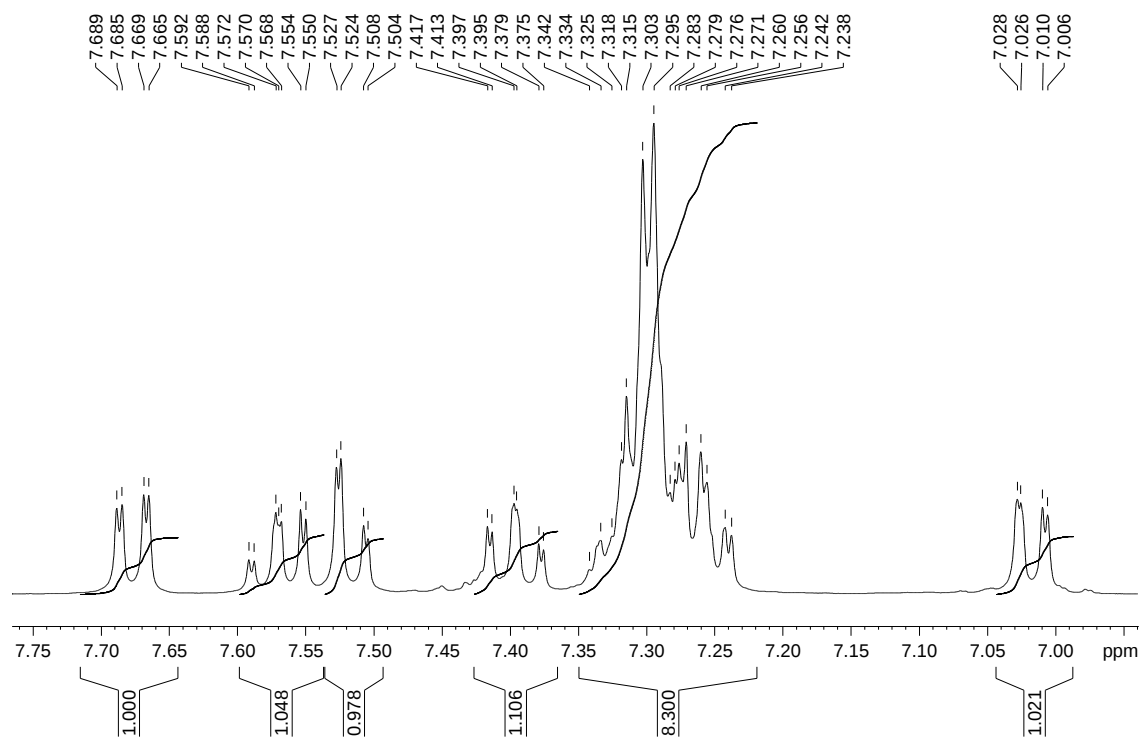
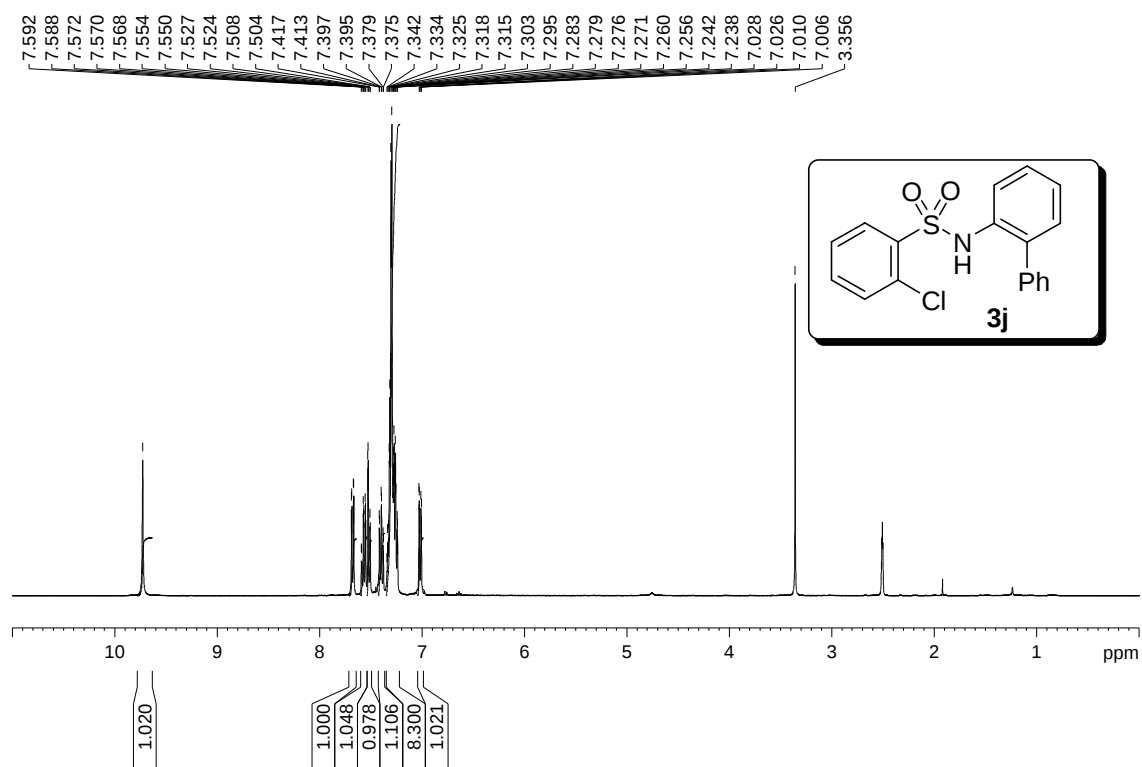


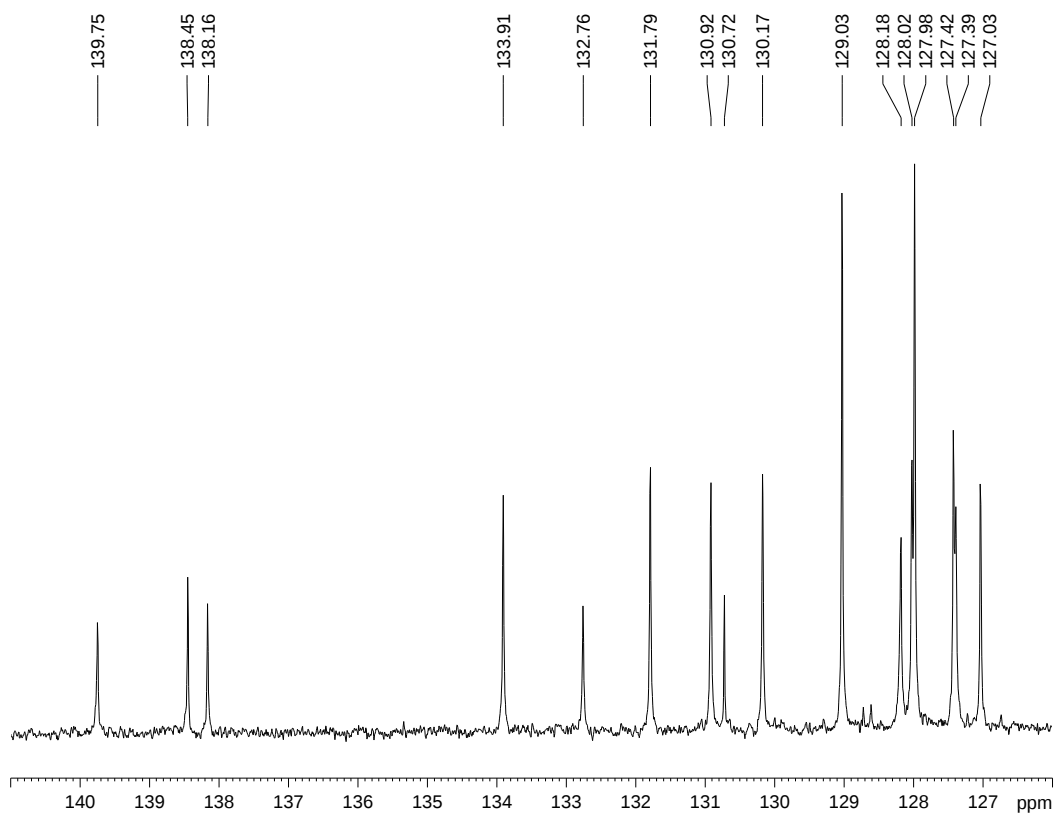
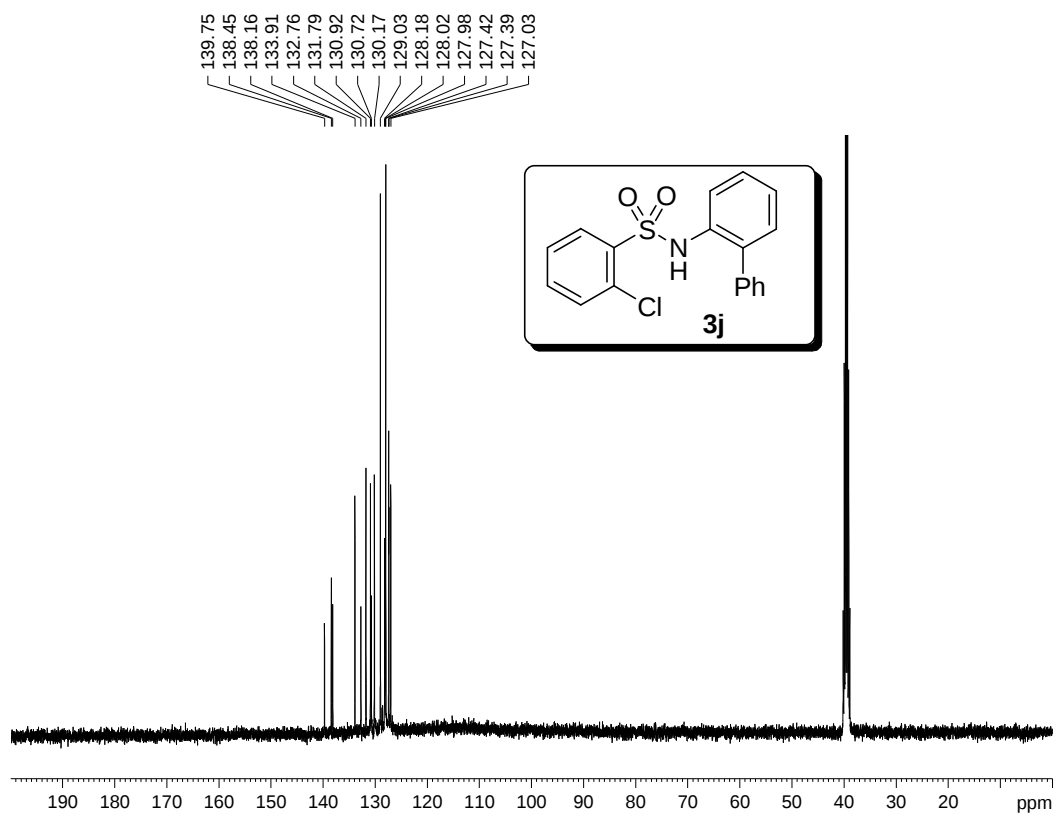
2-Chloro-N-(4-fluorophenyl)benzenesulfonamide (**3i**)



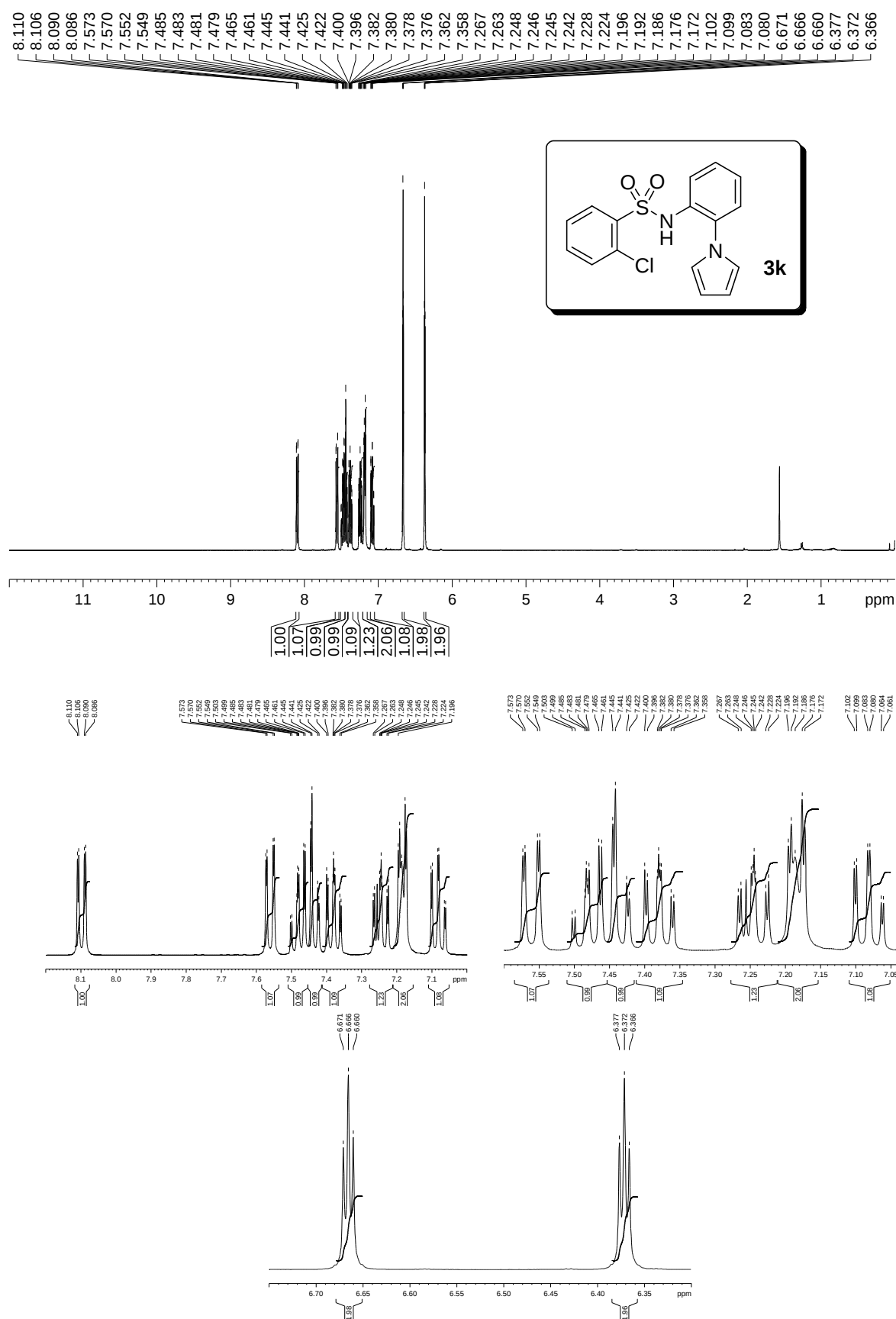


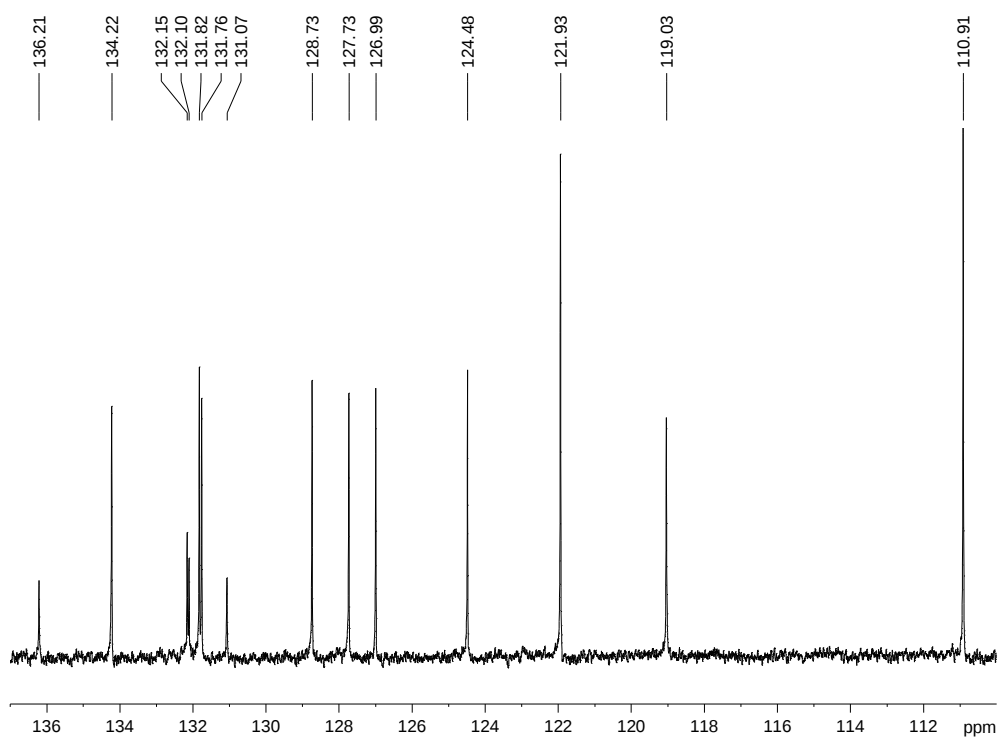
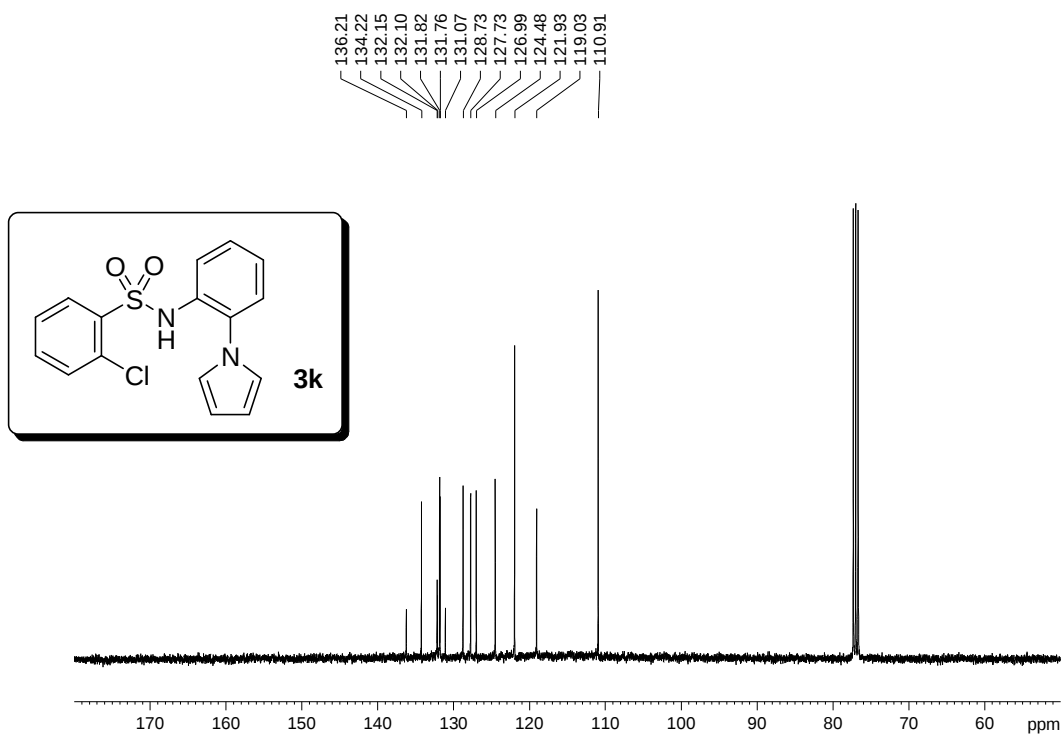
N-([1,1'-Biphenyl]-2-yl)-2-chlorobenzenesulfonamide (**3j**)



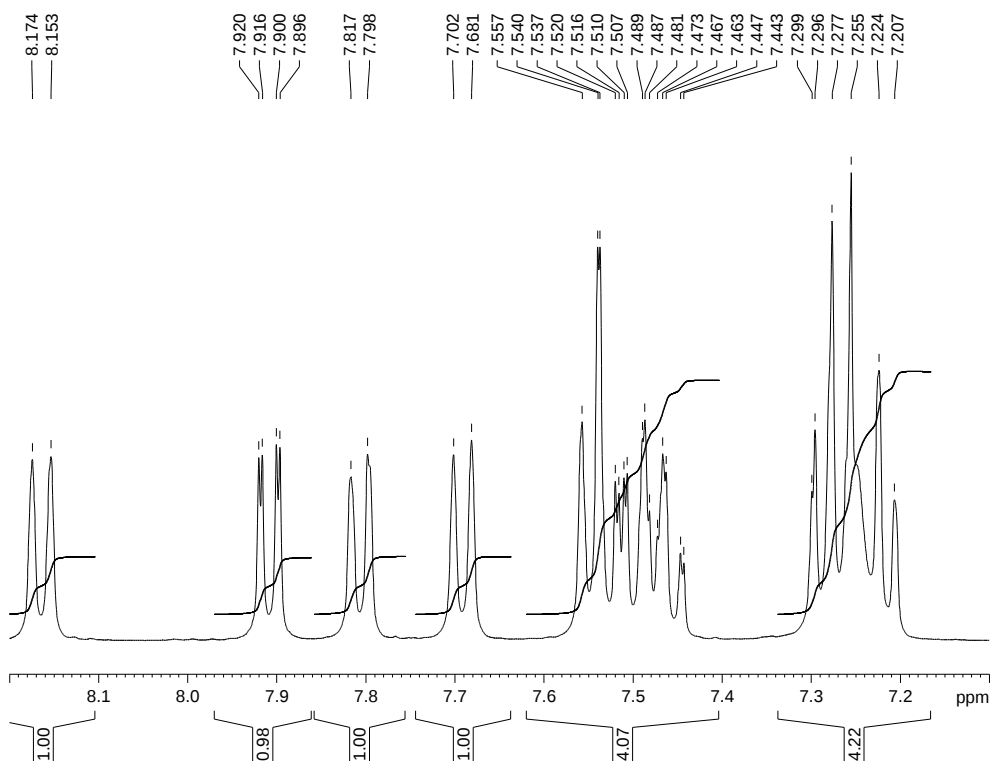
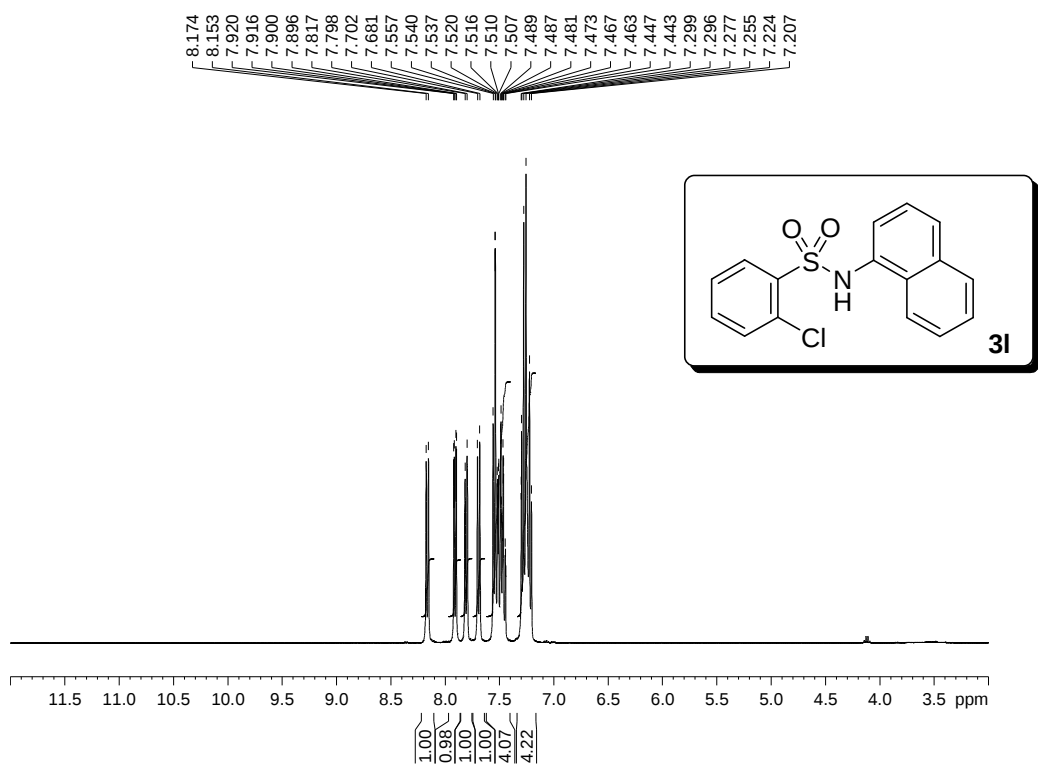


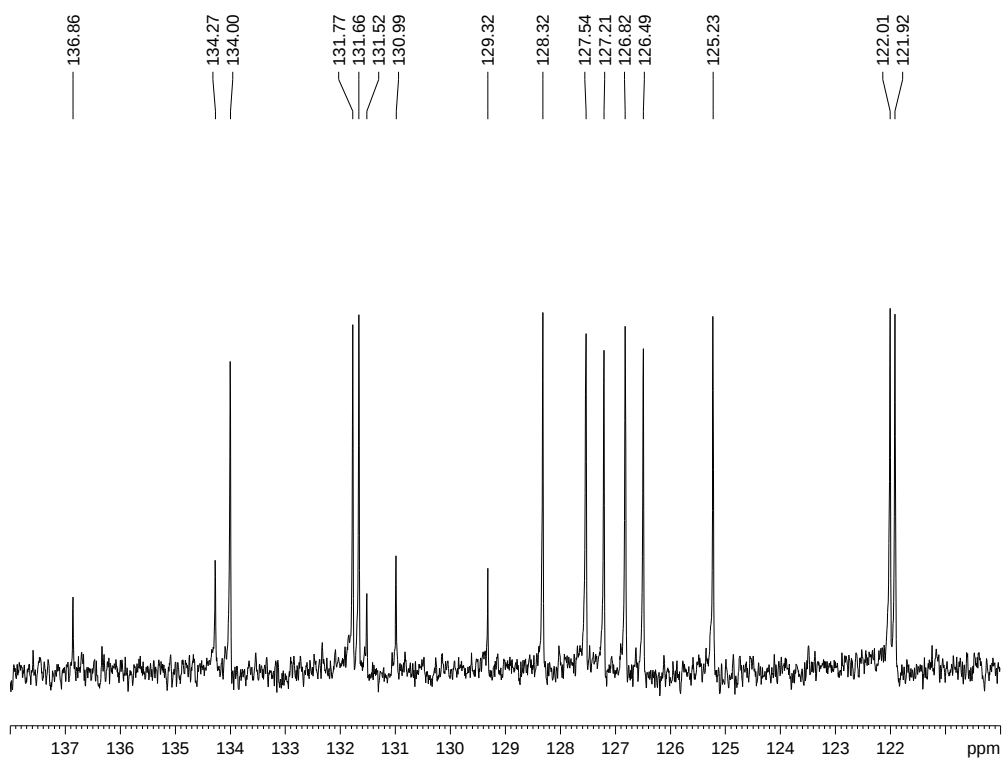
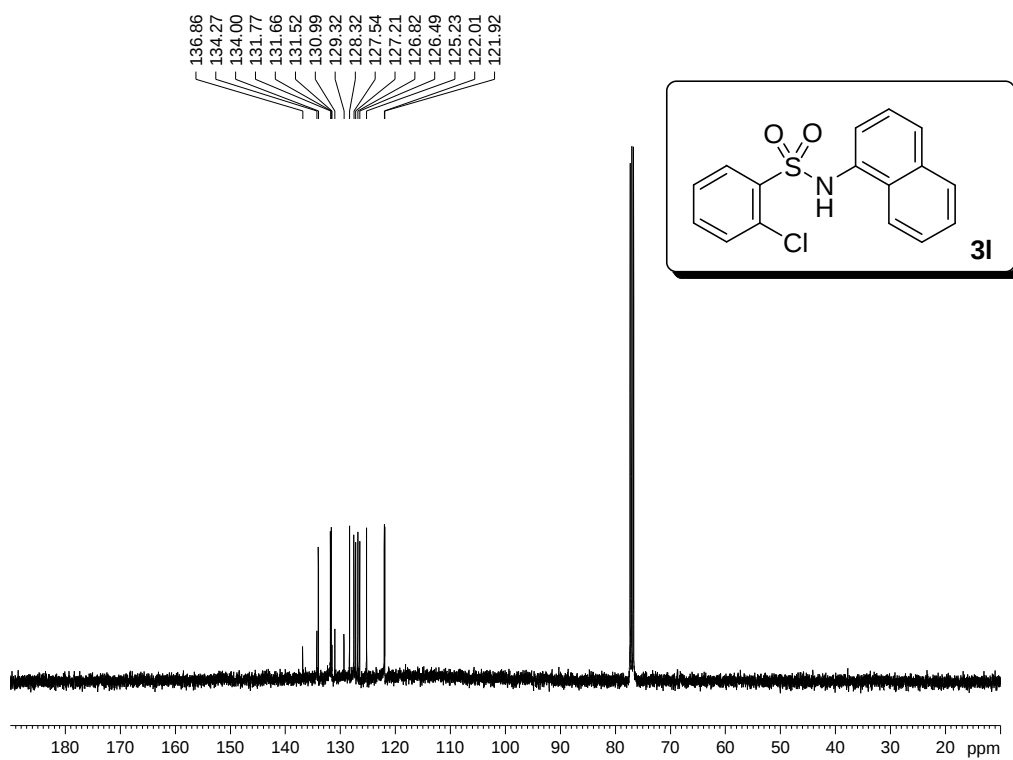
N-(2-(1*H*-Pyrrol-1-yl)phenyl)-2-chlorobenzenesulfonamide (**3k**)



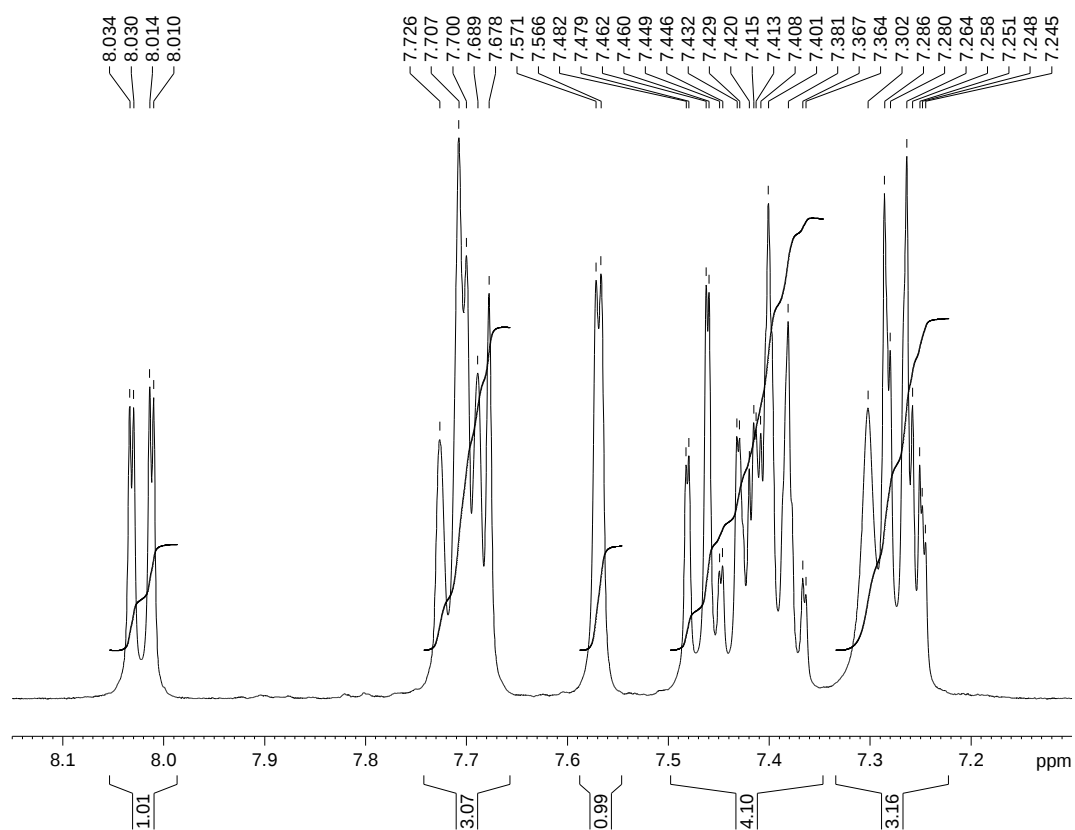
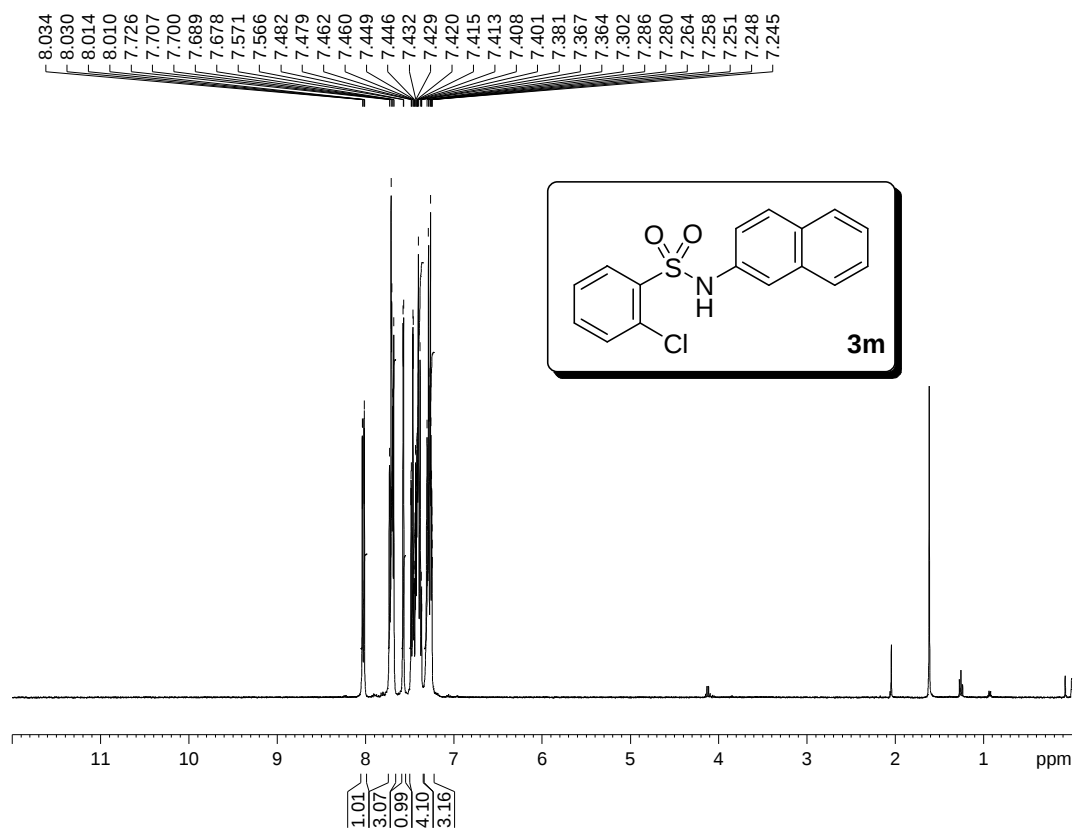


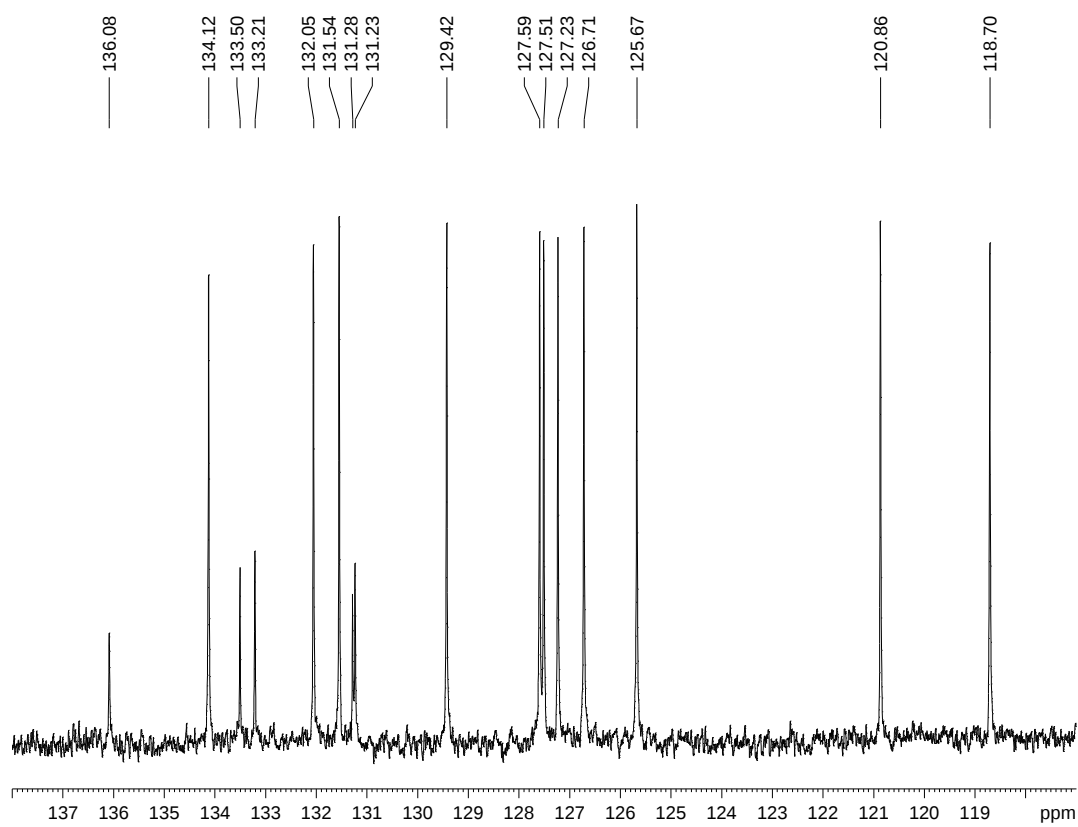
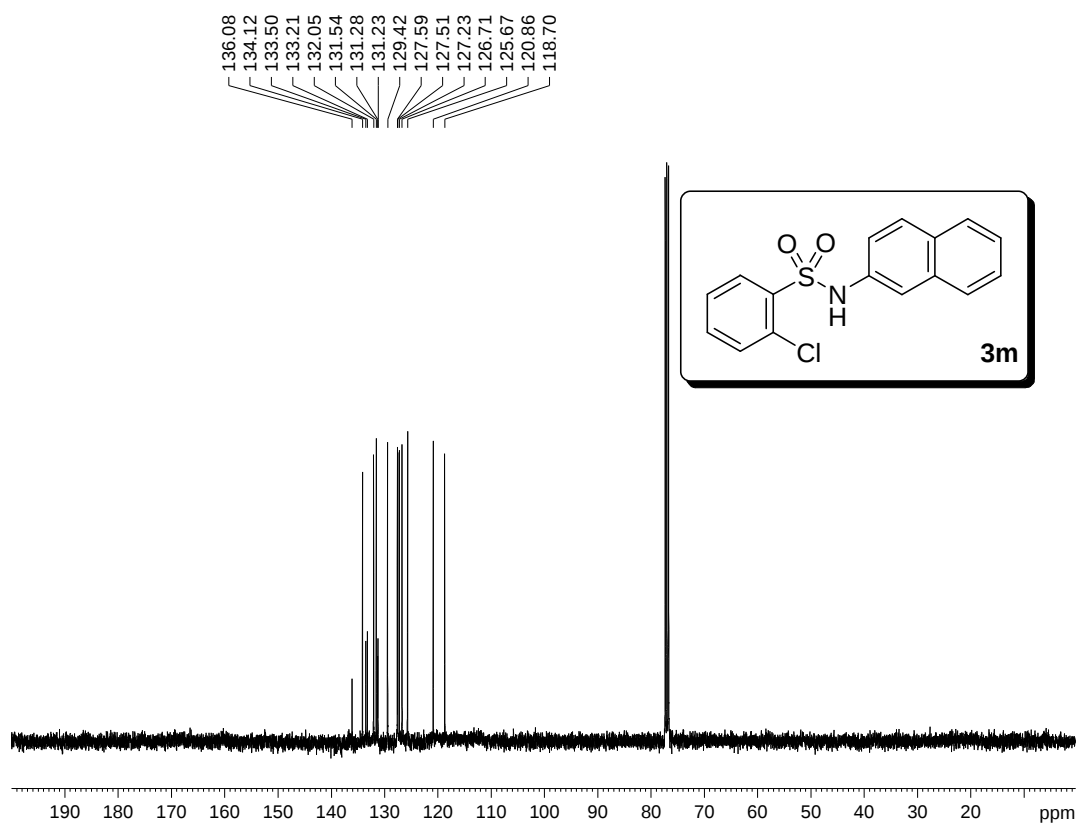
2-Chloro-N-(naphthalen-1-yl)benzenesulfonamide (**3I**)



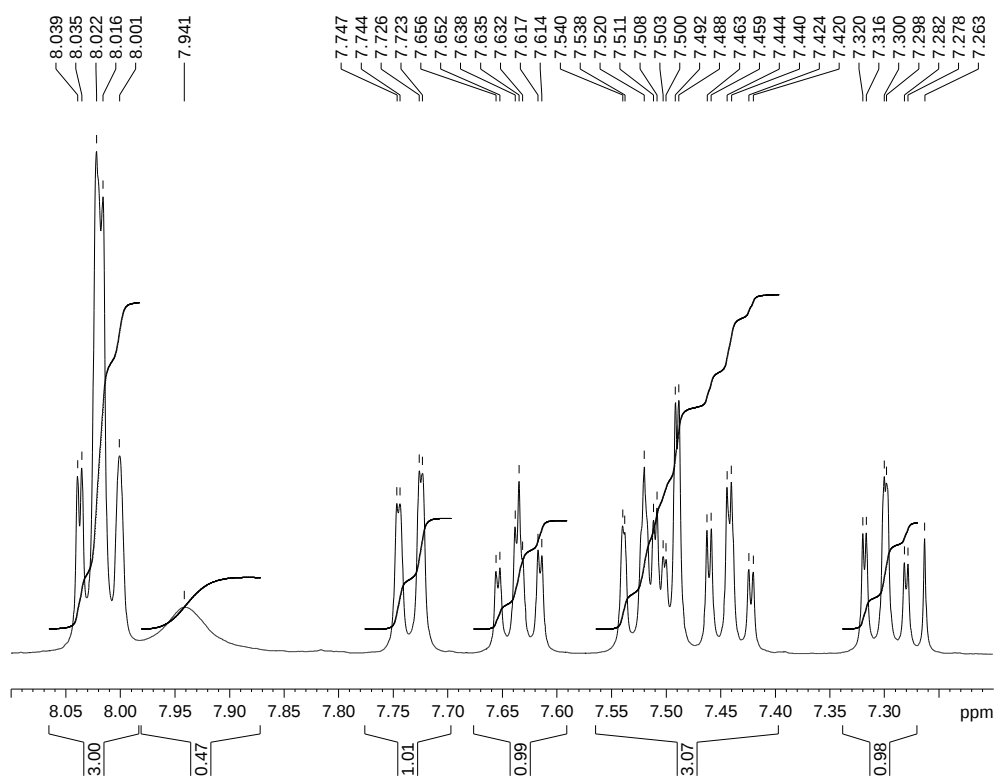
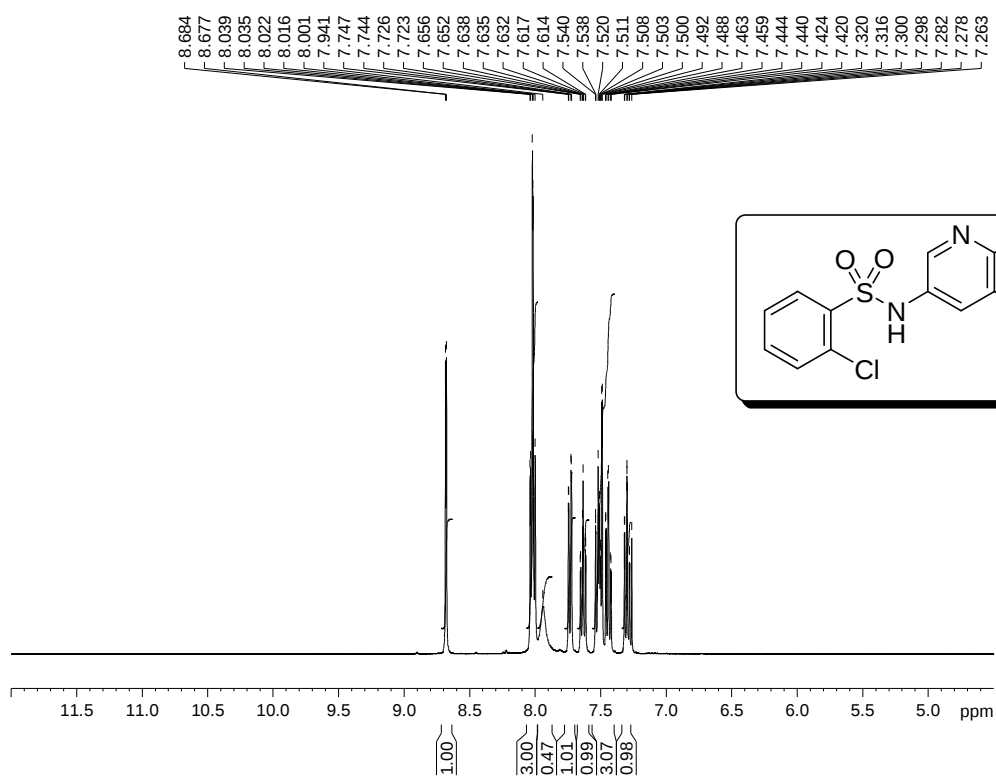


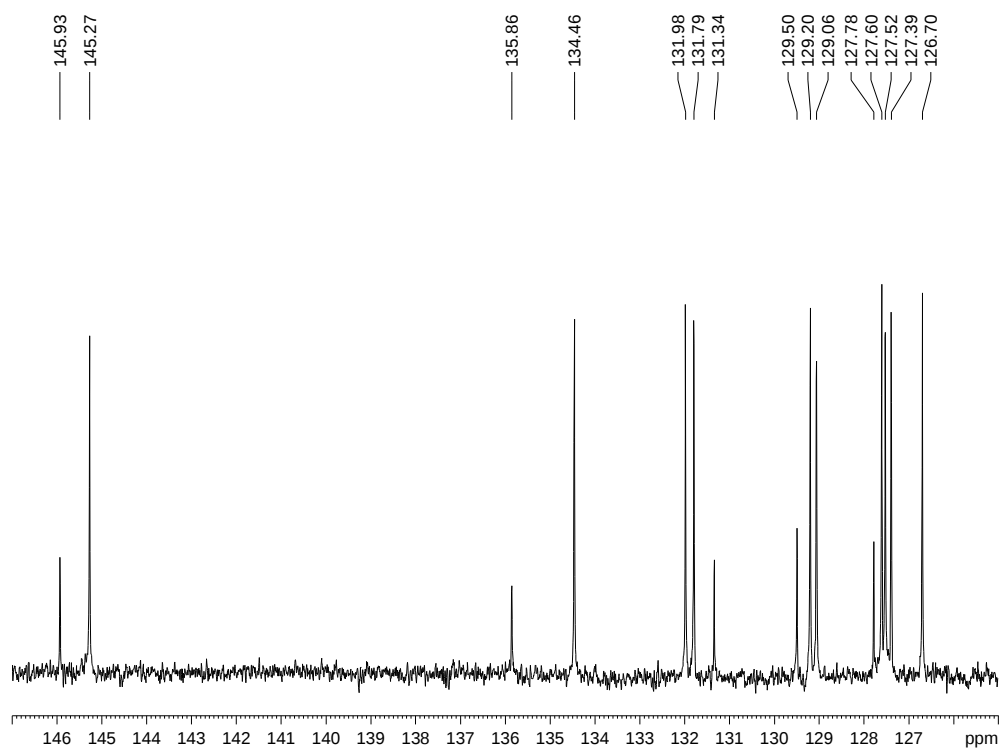
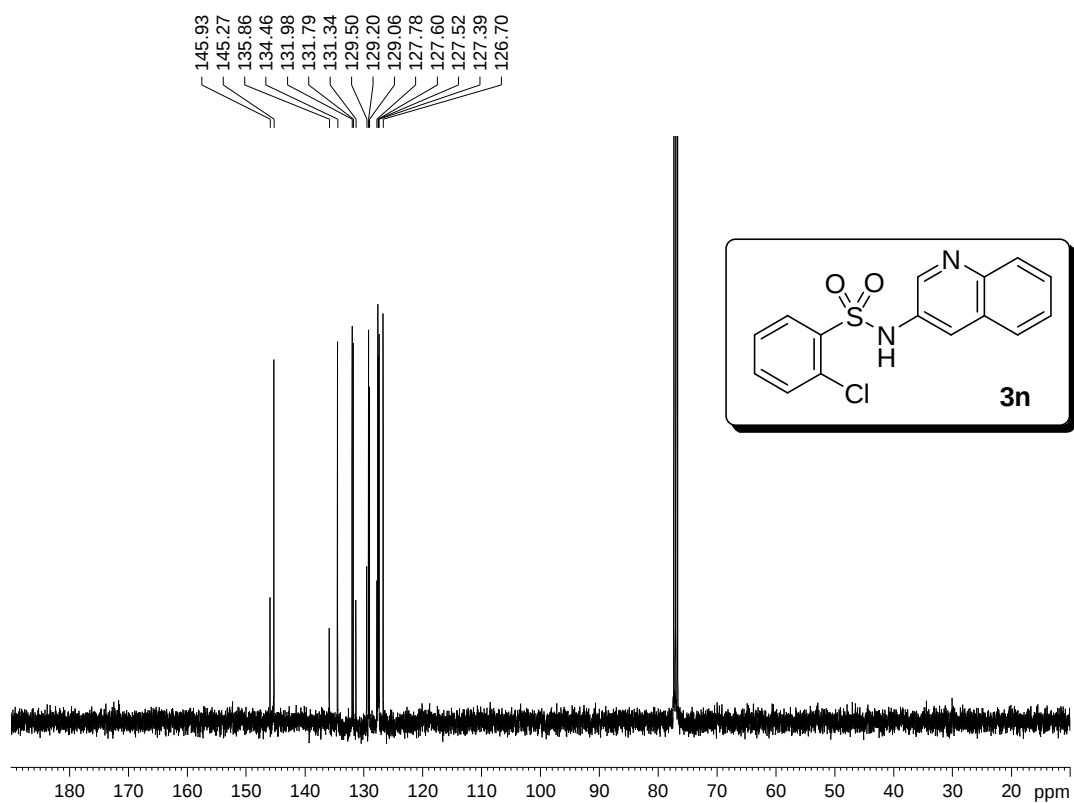
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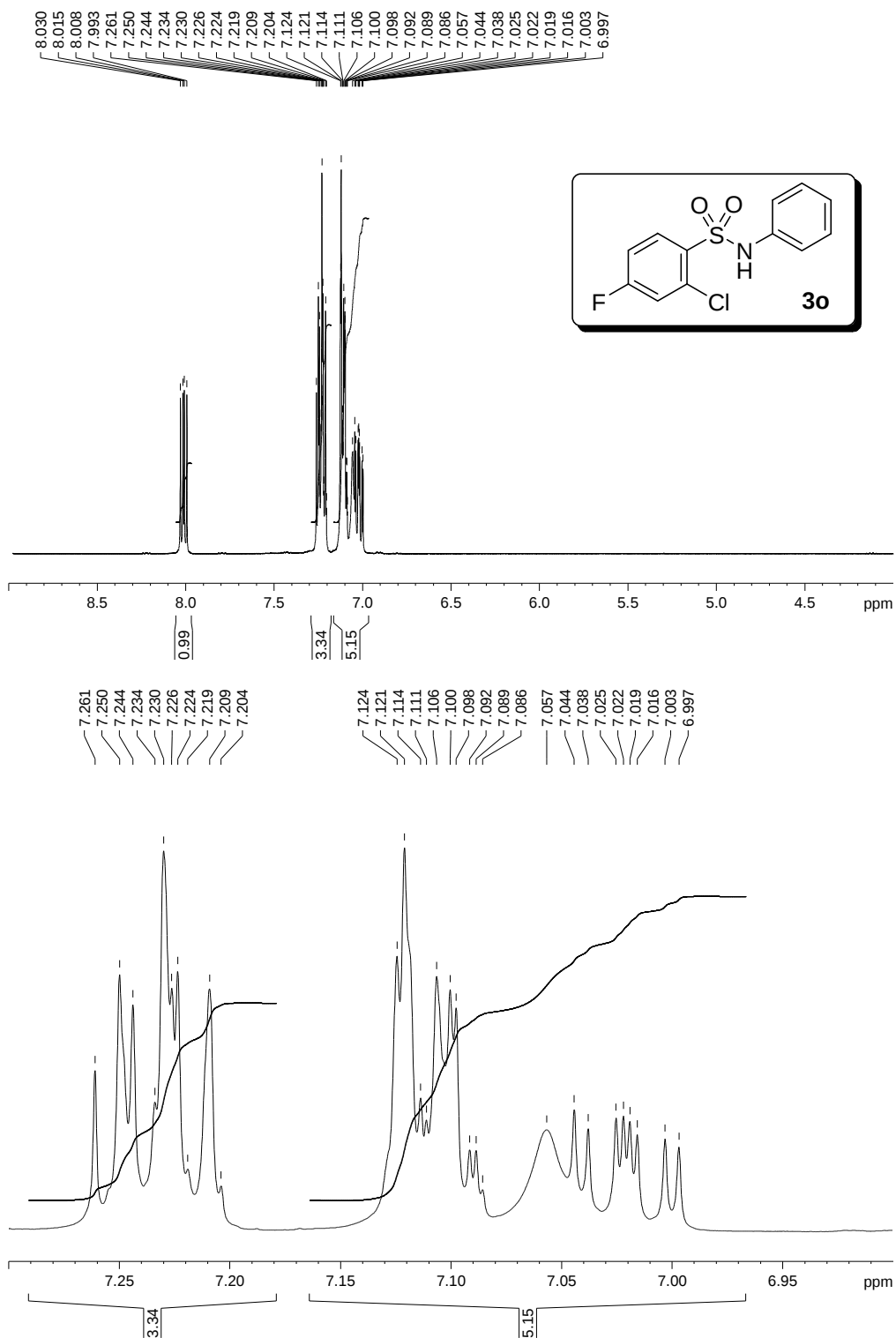


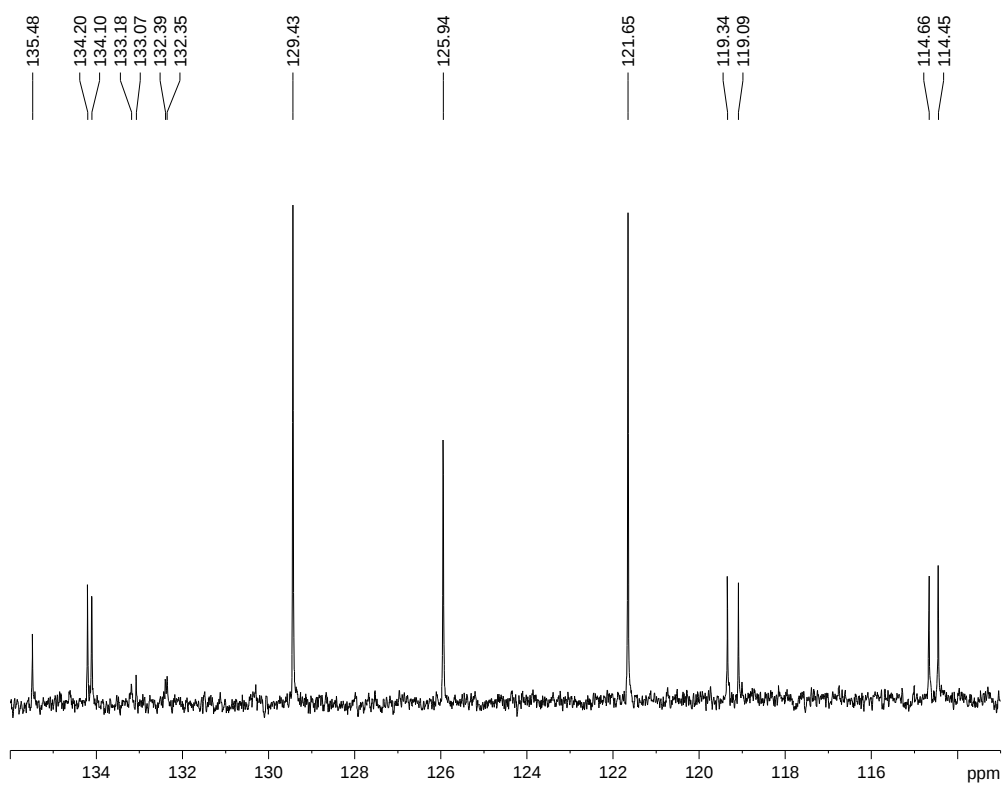
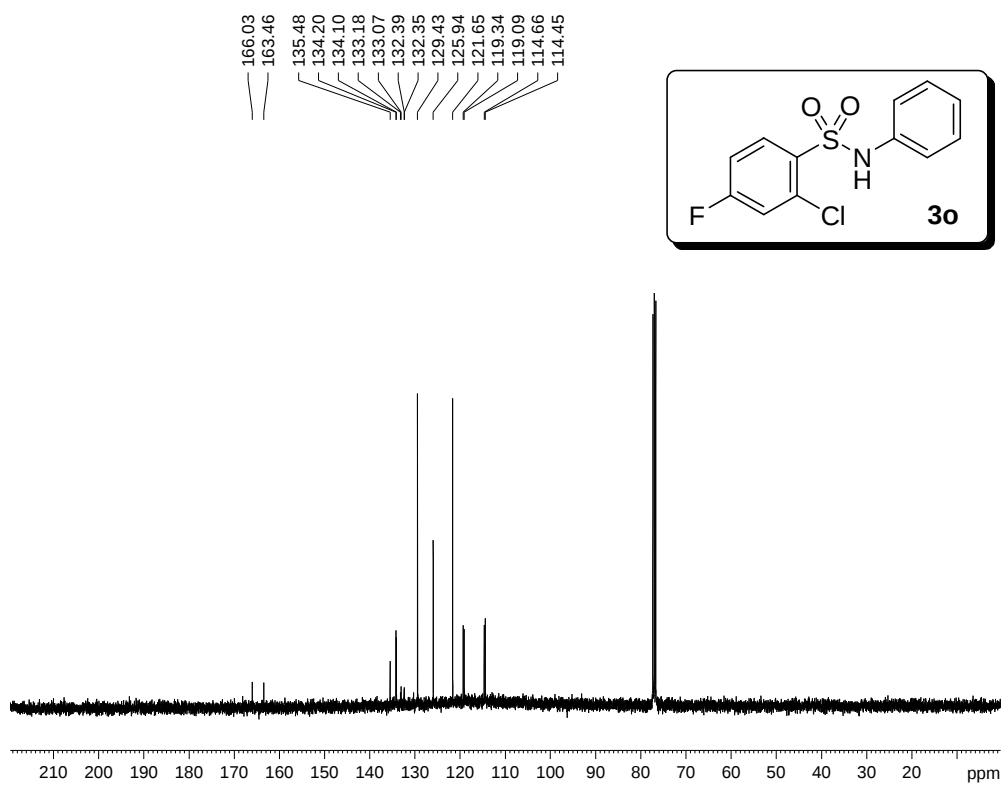
2-Chloro-N-(quinolin-3-yl)benzenesulfonamide (**3n**)



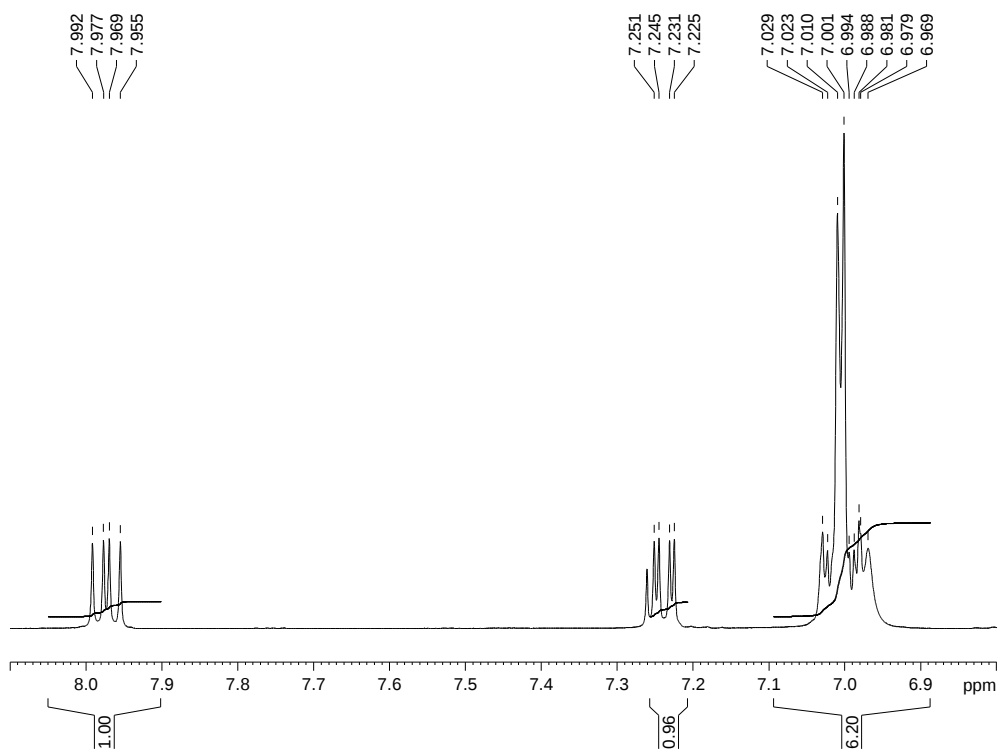
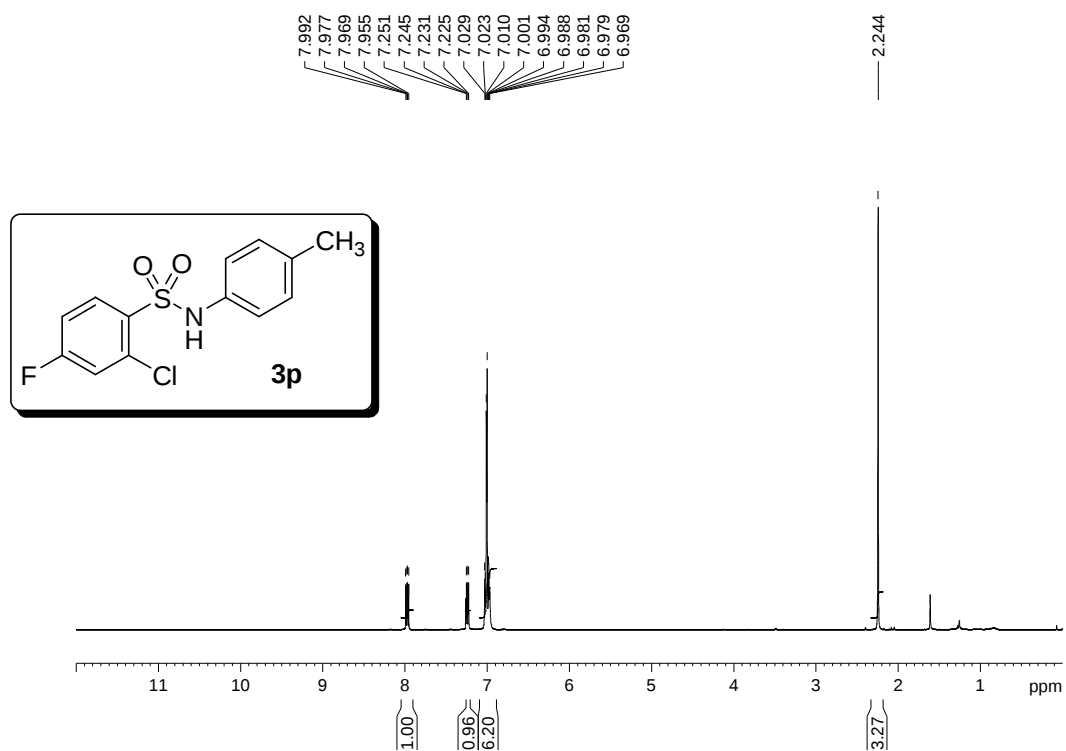


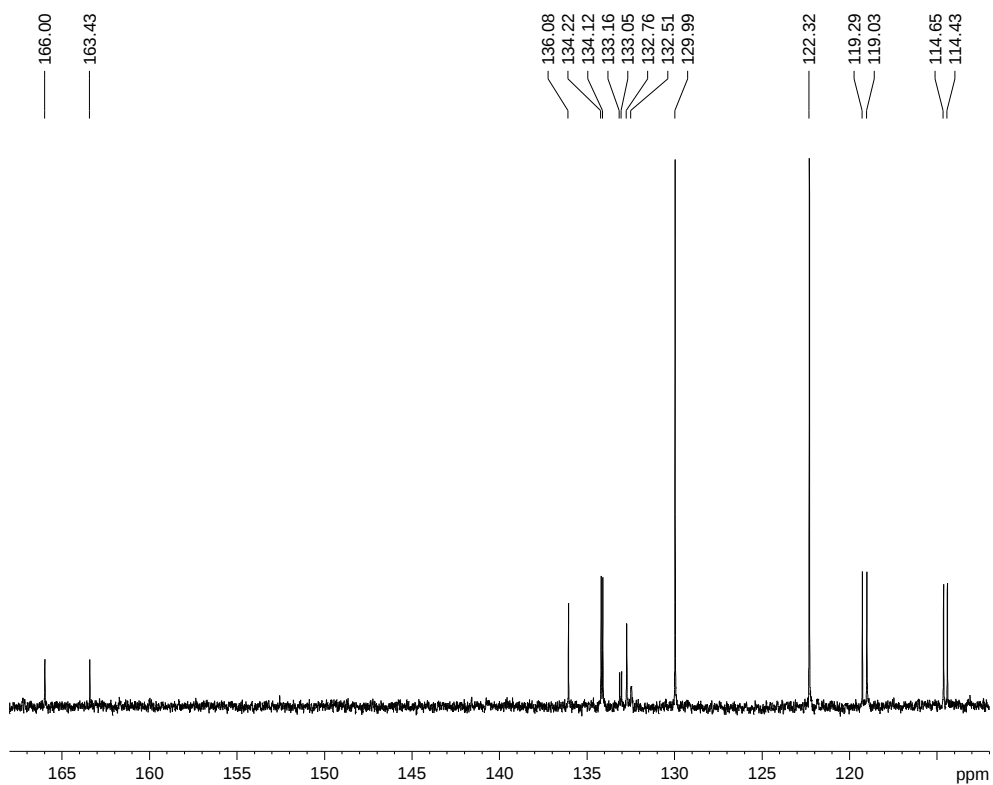
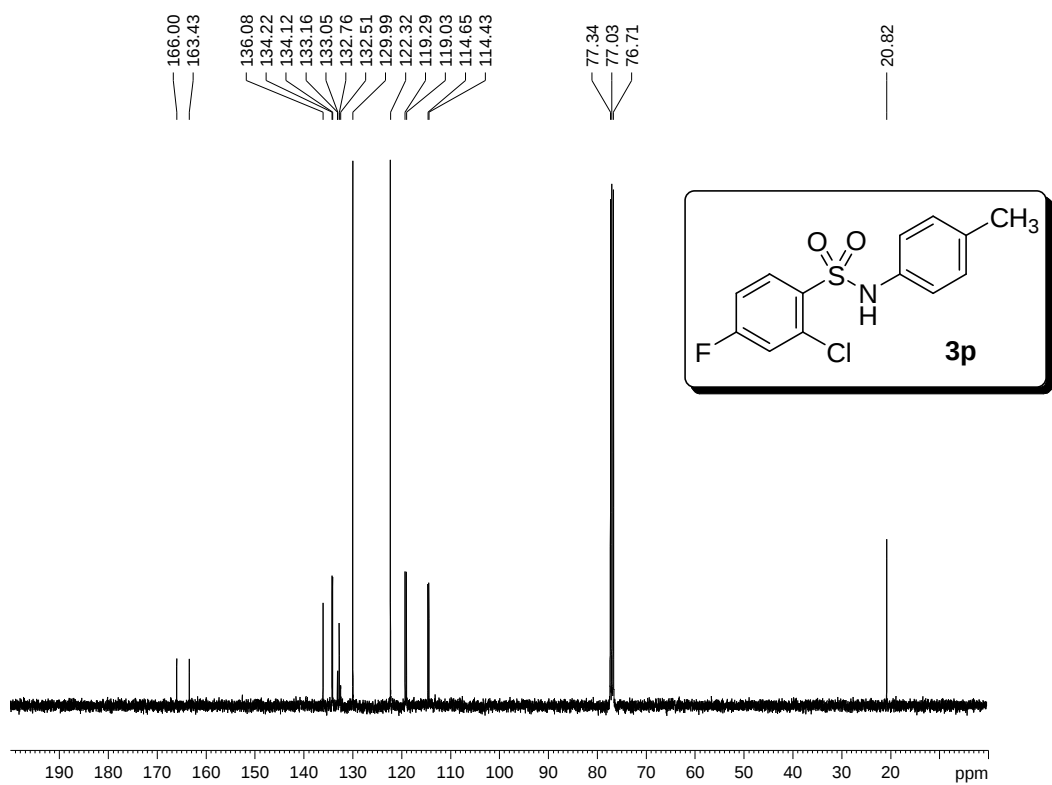
2-Chloro-4-fluoro-N-phenylbenzenesulfonamide (**3o**)

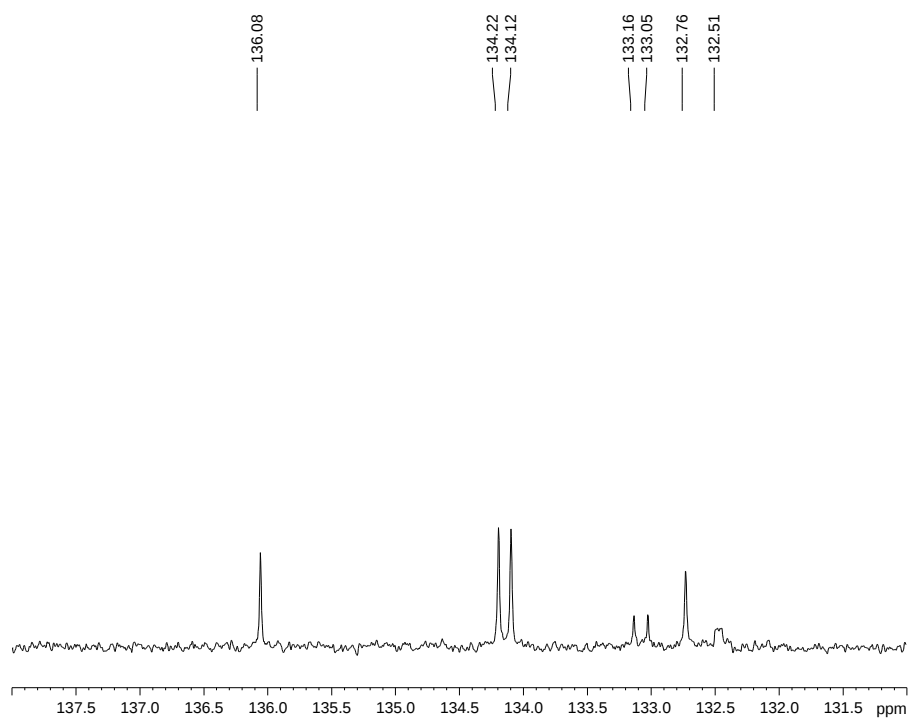




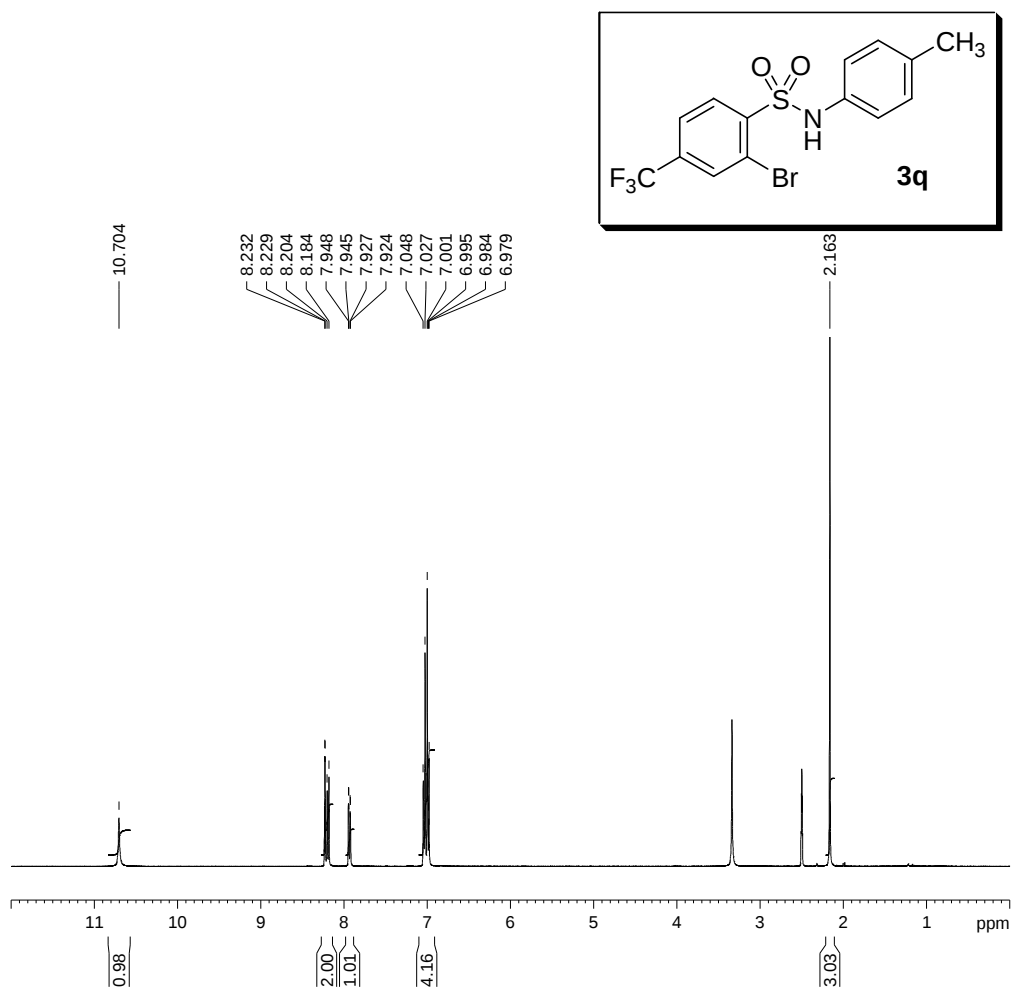
2-Chloro-4-fluoro-N-(p-tolyl)benzenesulfonamide (**3p**)

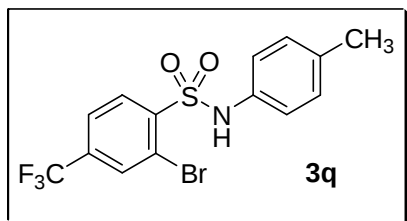
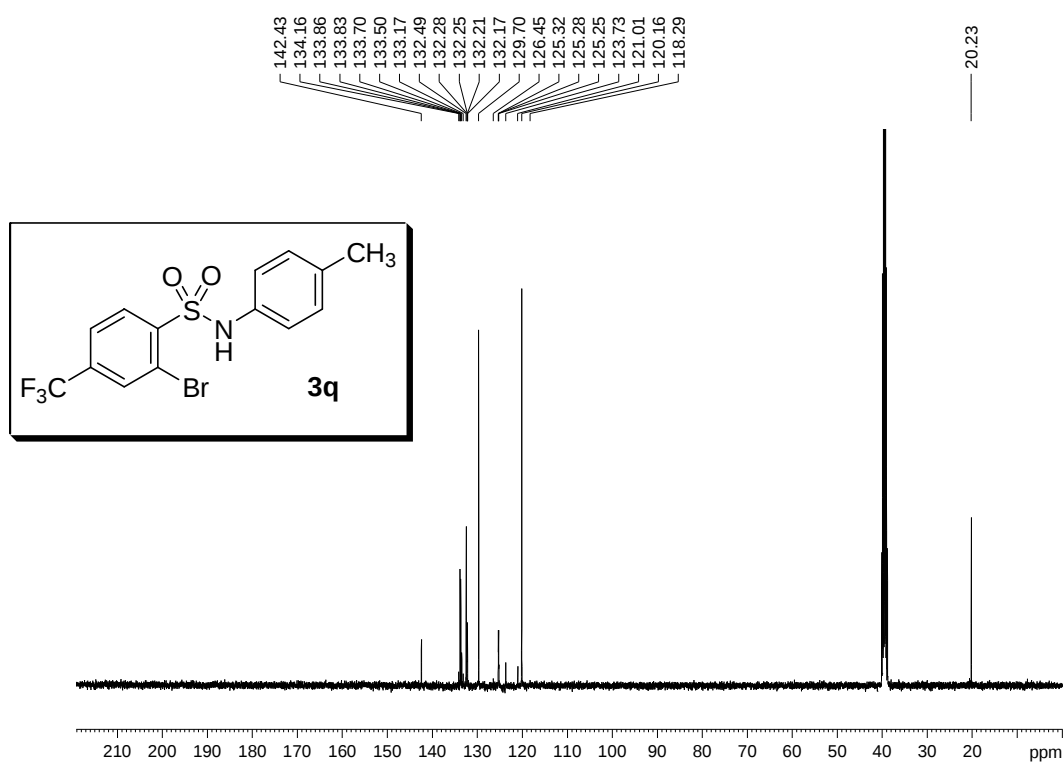
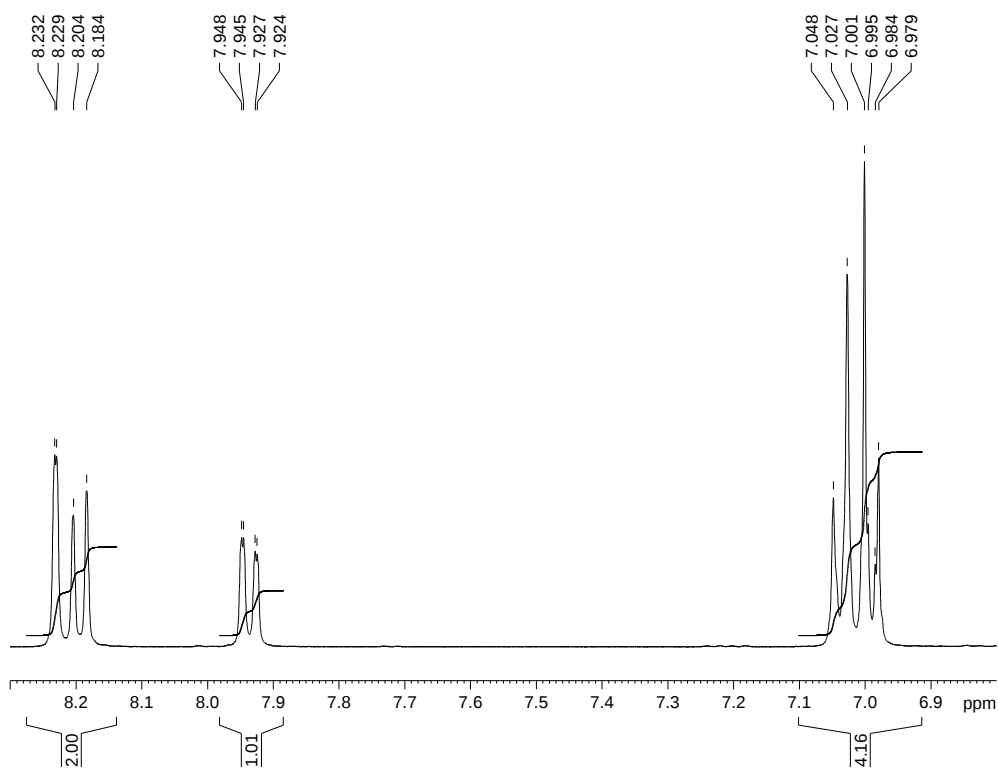


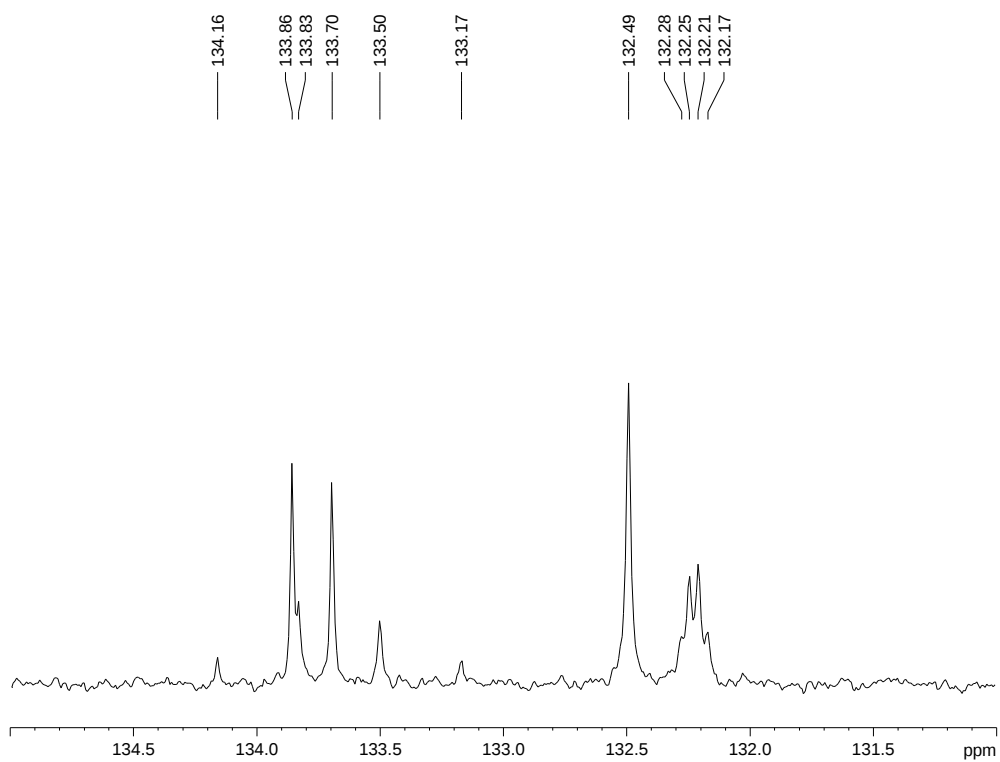
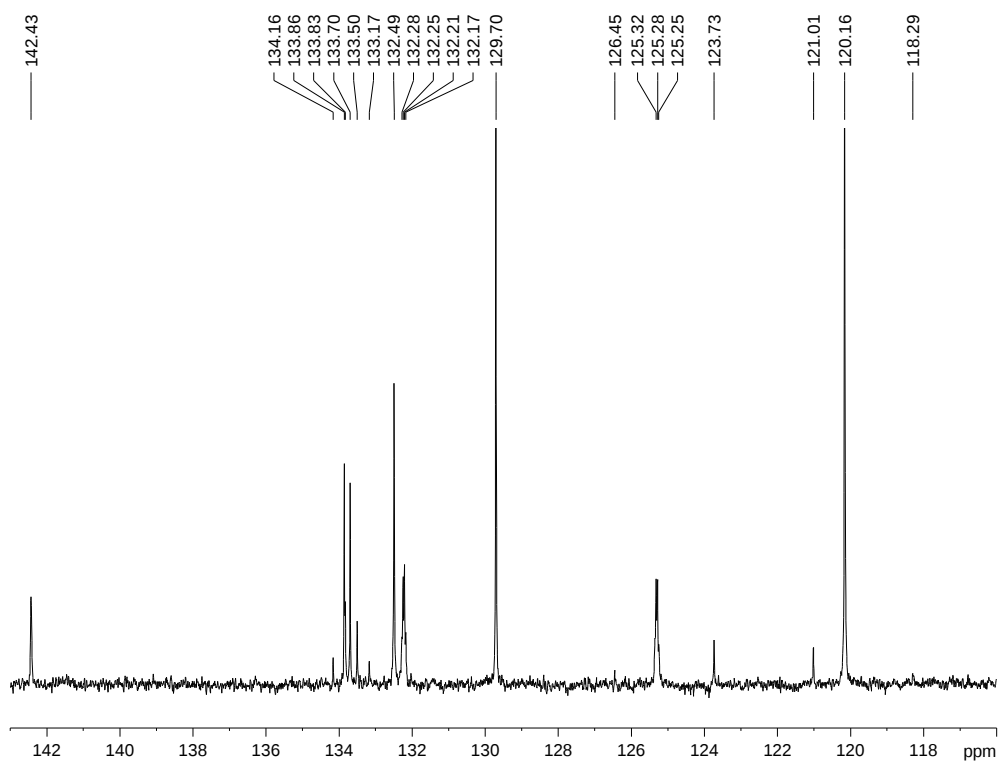


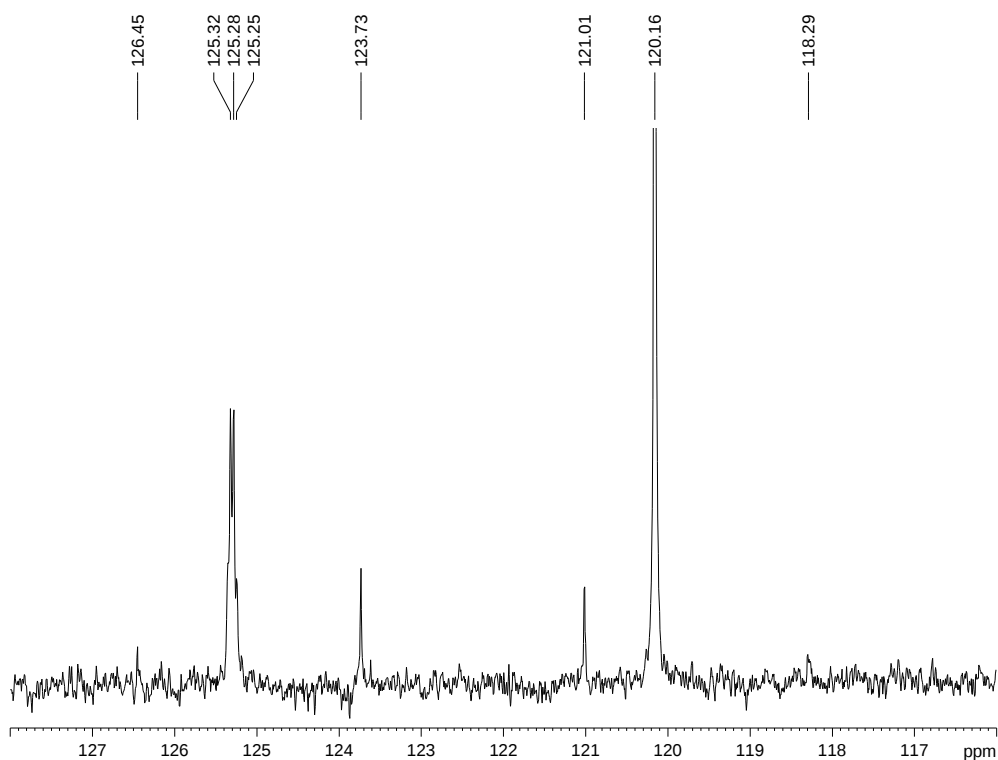


2-Bromo-N-(p-tolyl)-4-(trifluoromethyl)benzenesulfonamide (3q)

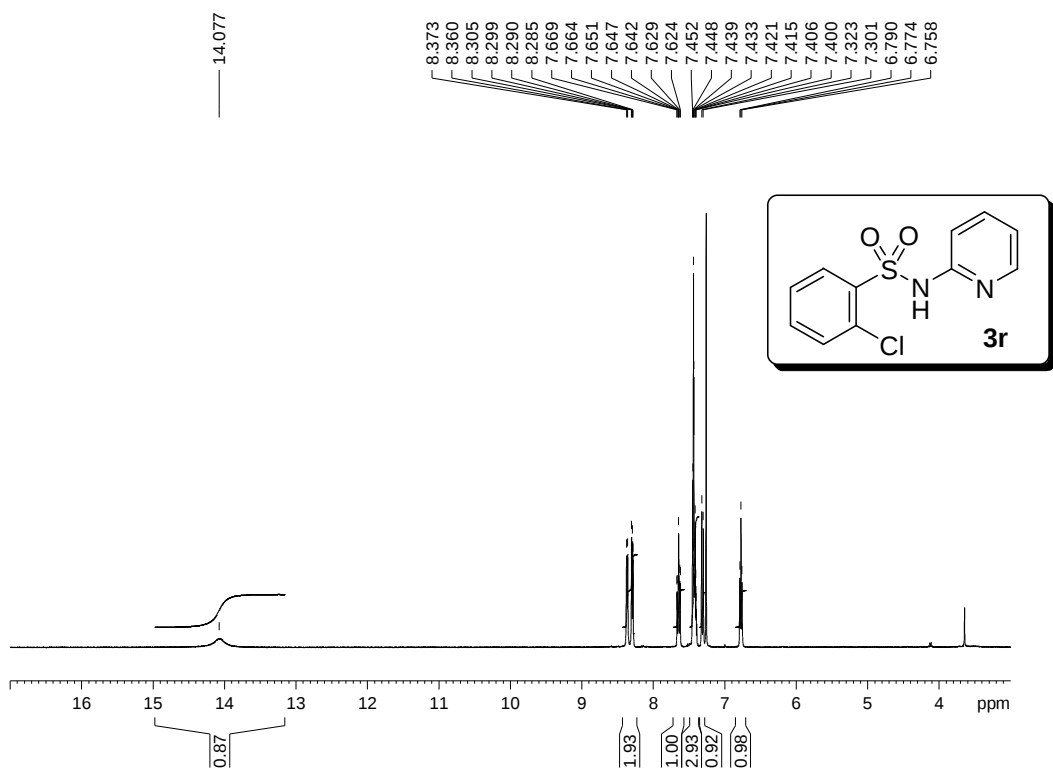


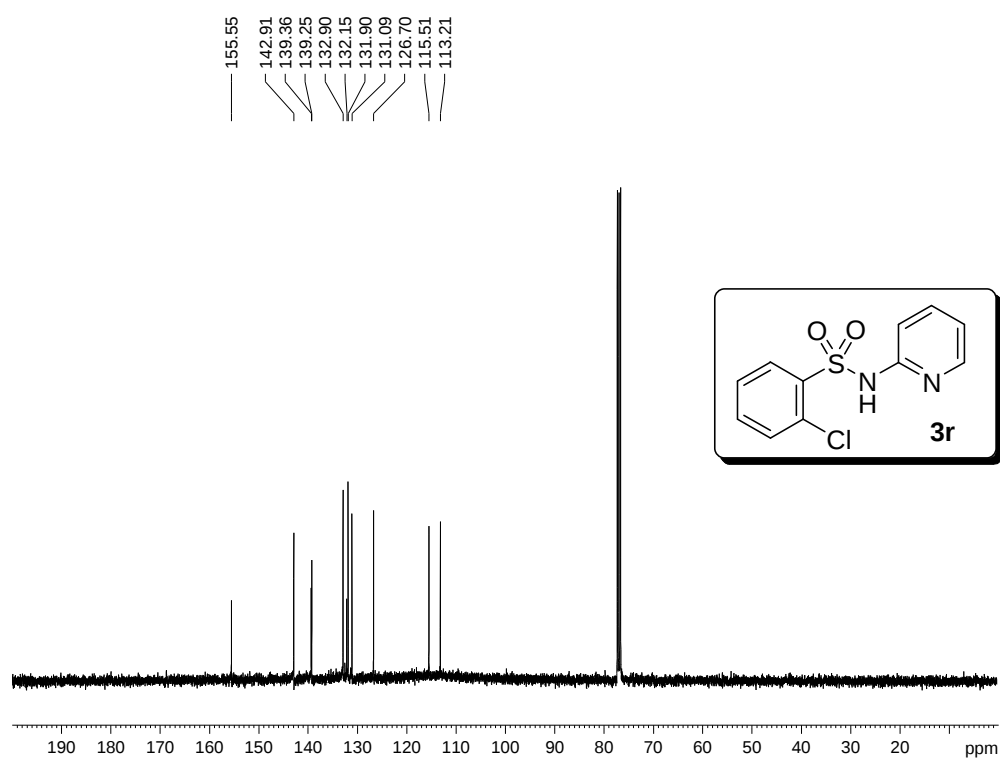
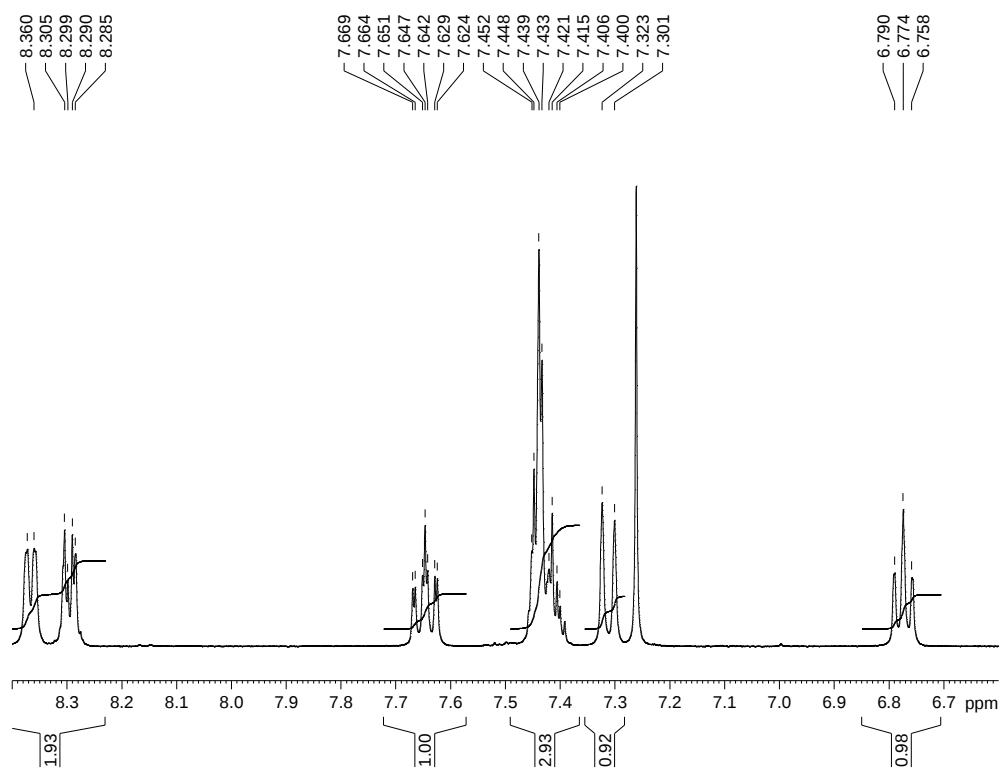


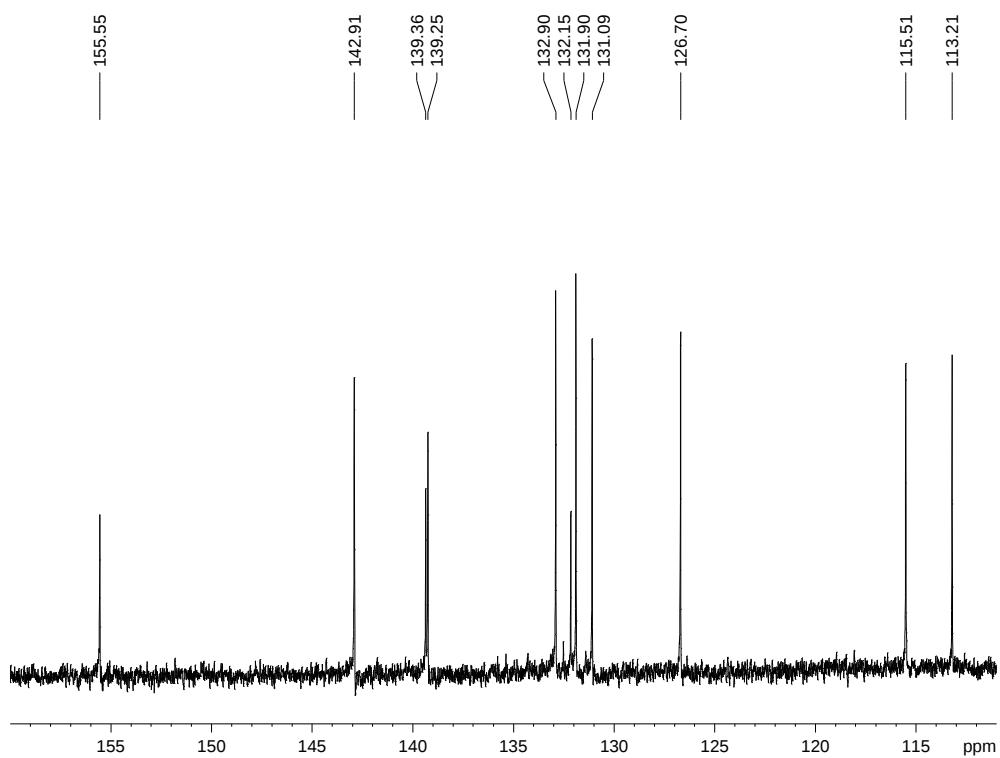




2-Chloro-N-(pyridin-2-yl)benzenesulfonamide (3r)

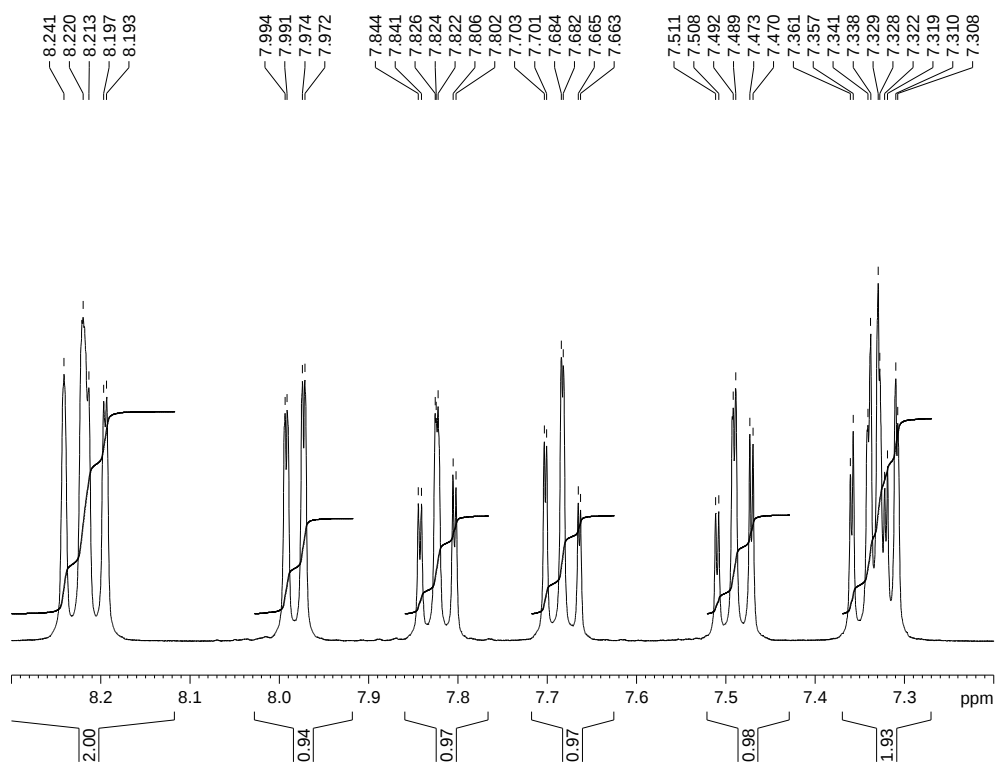
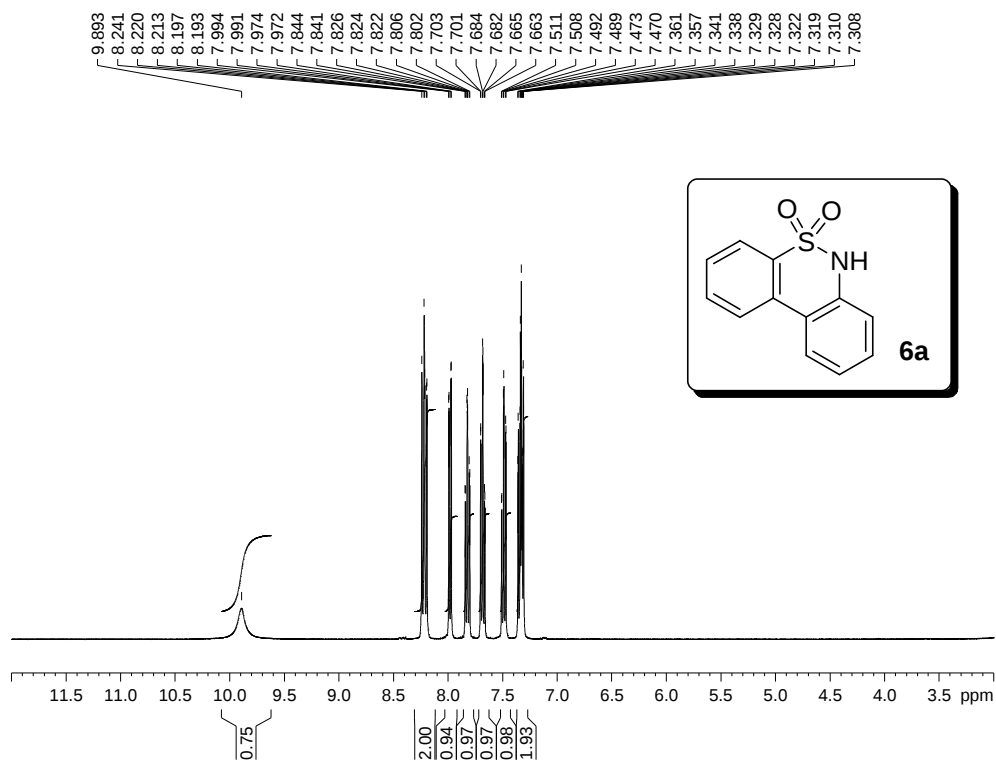


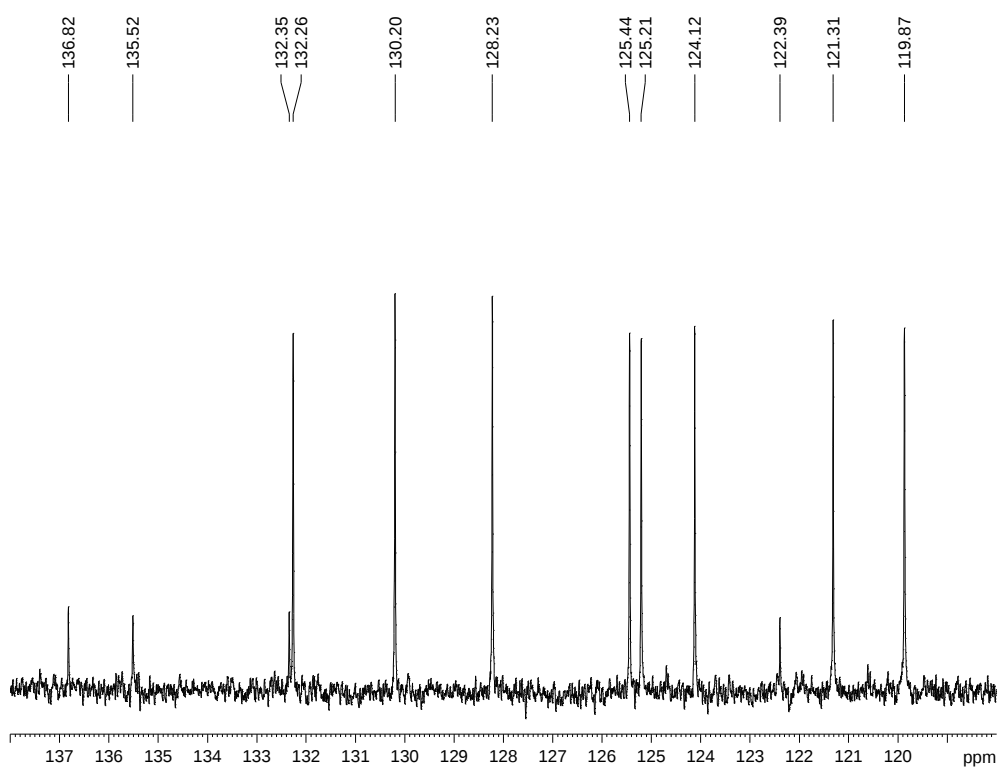
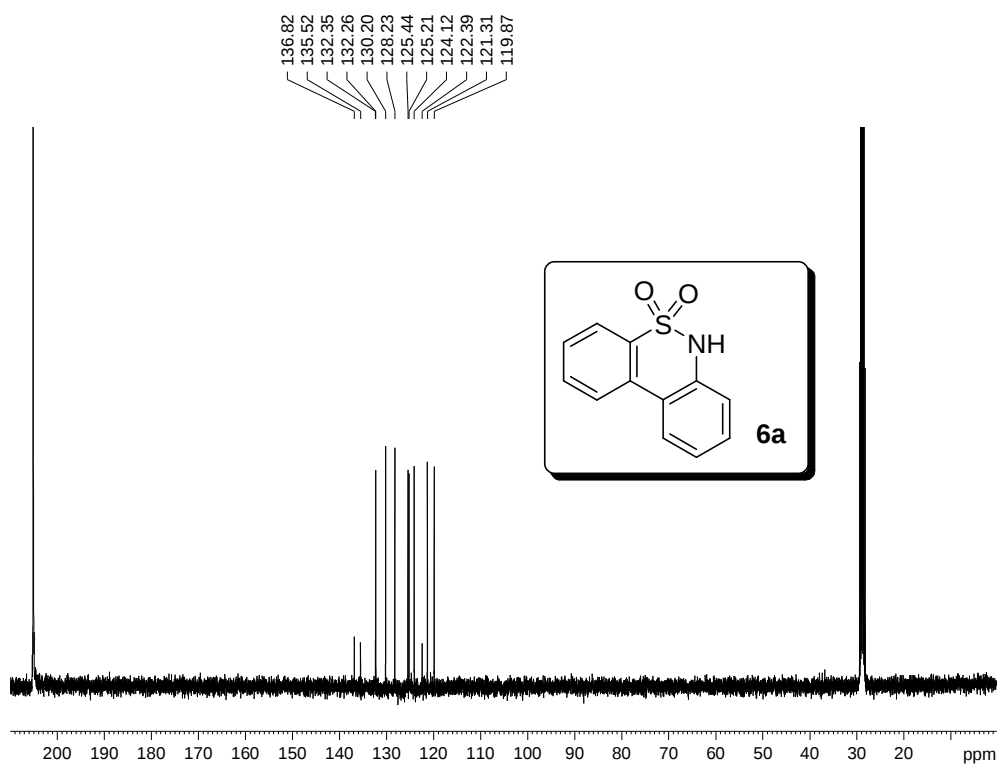




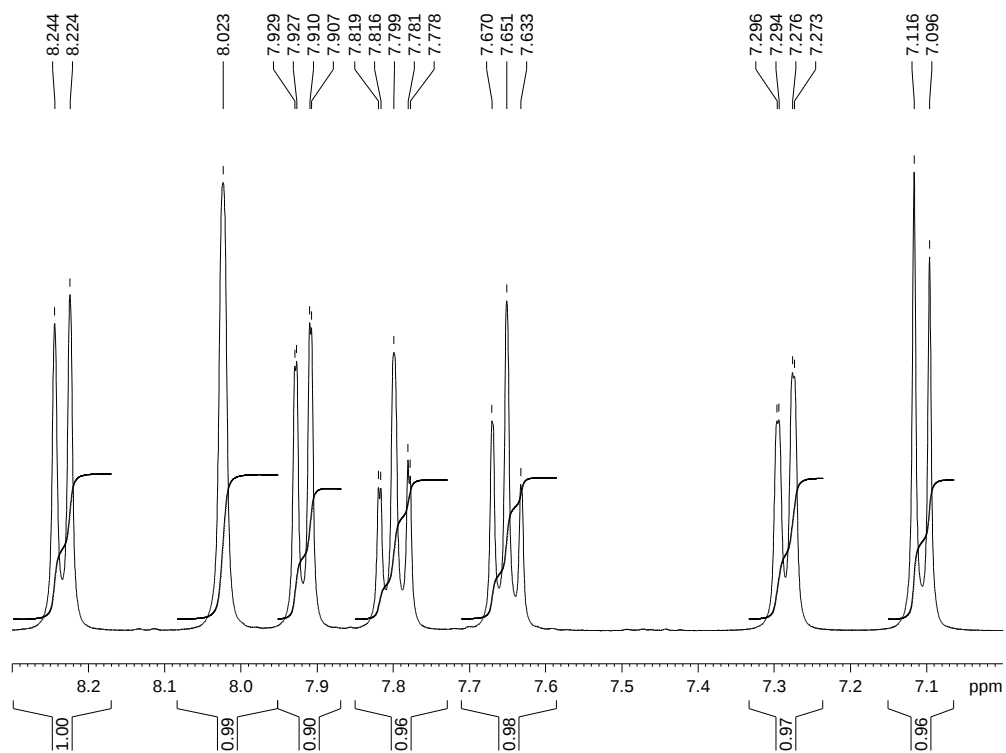
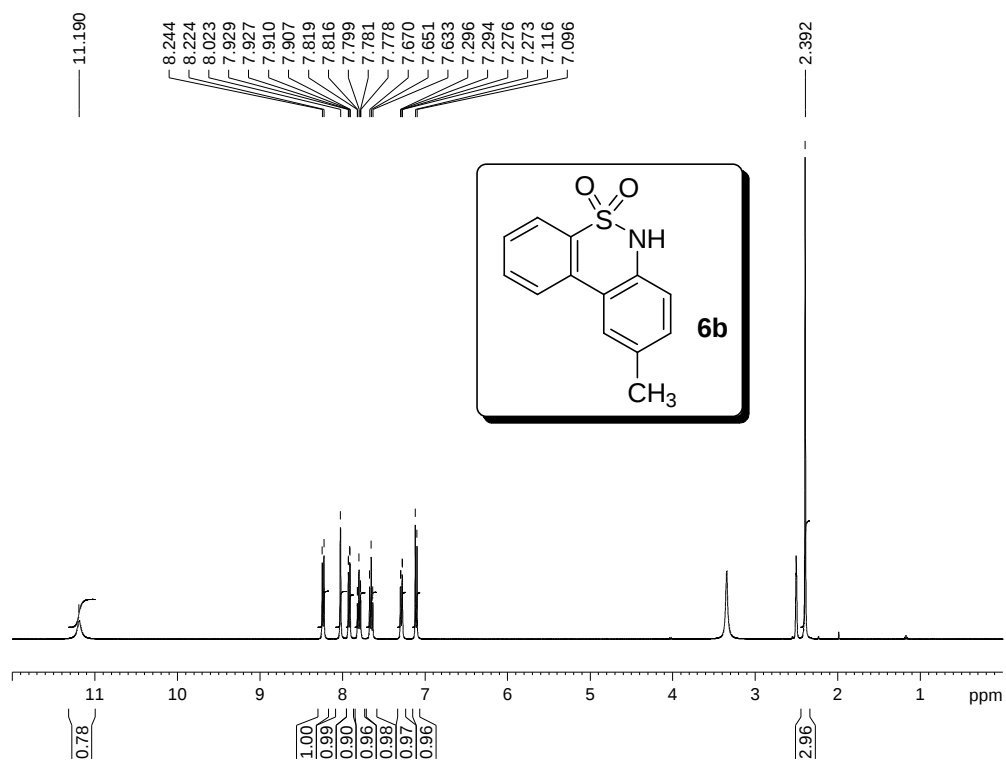
4. Copies of ^1H and ^{13}C NMR Spectra of products

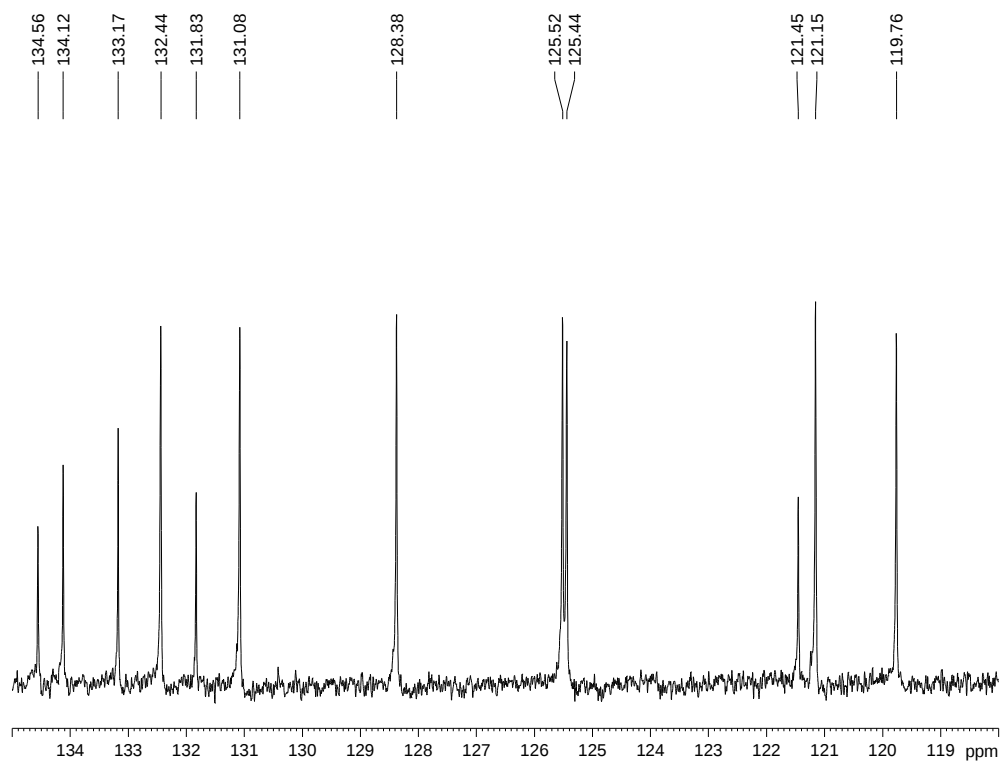
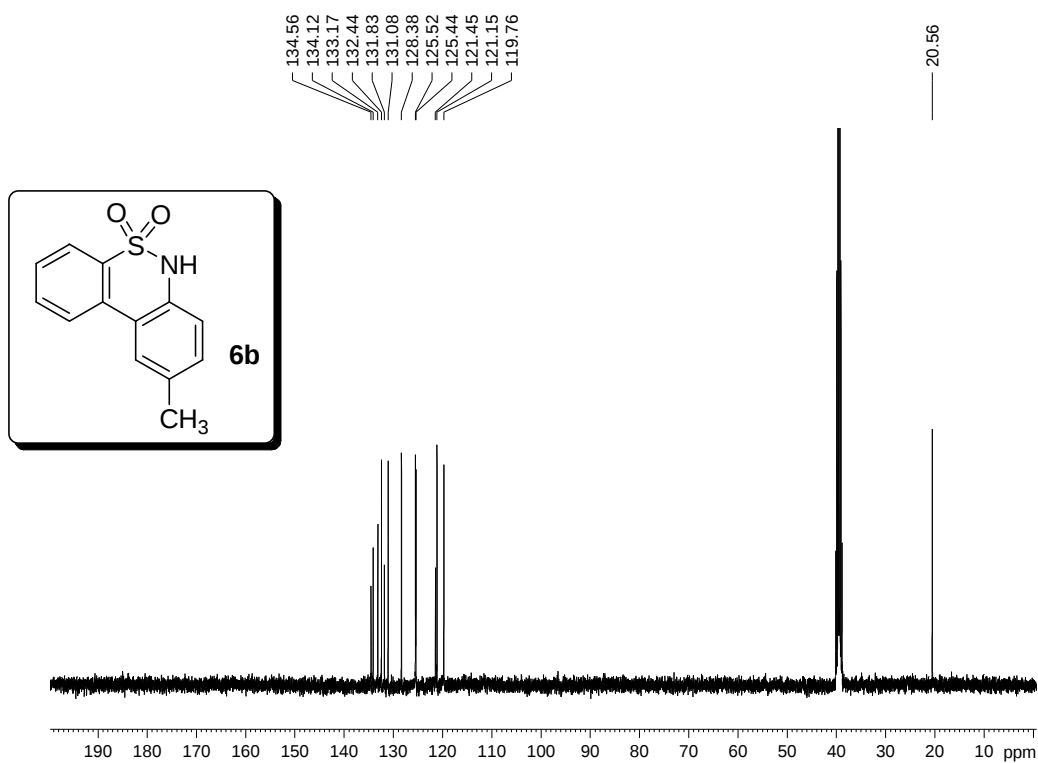
6H-Dibenzo[*c,e*][1,2]thiazine 5,5-dioxide (6a)



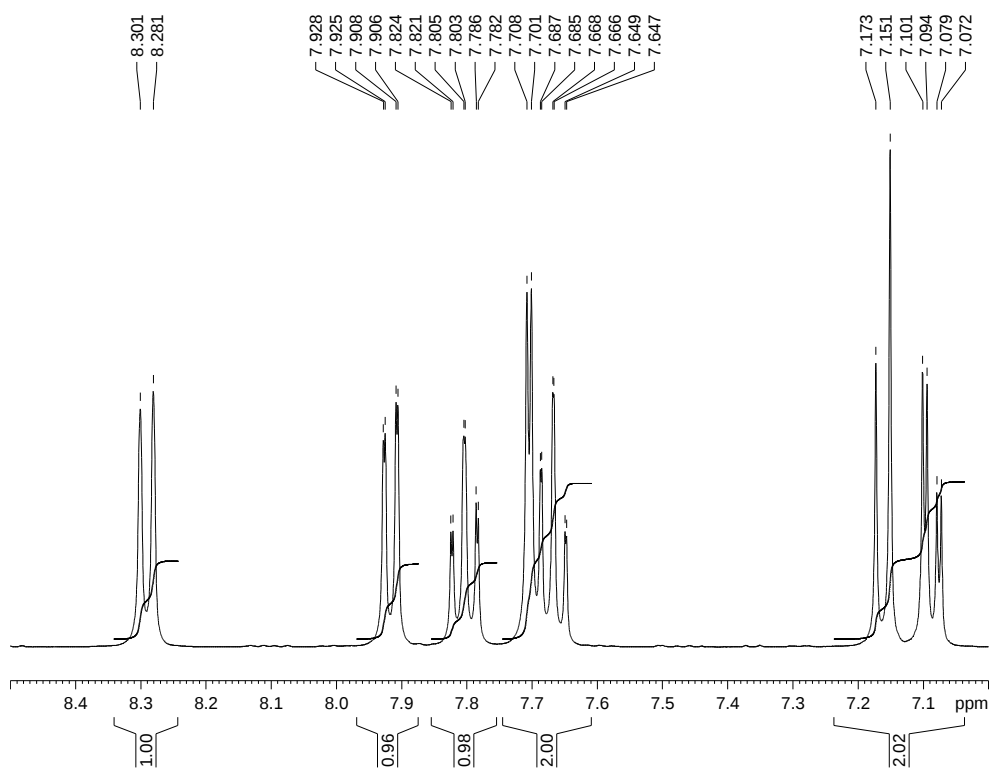
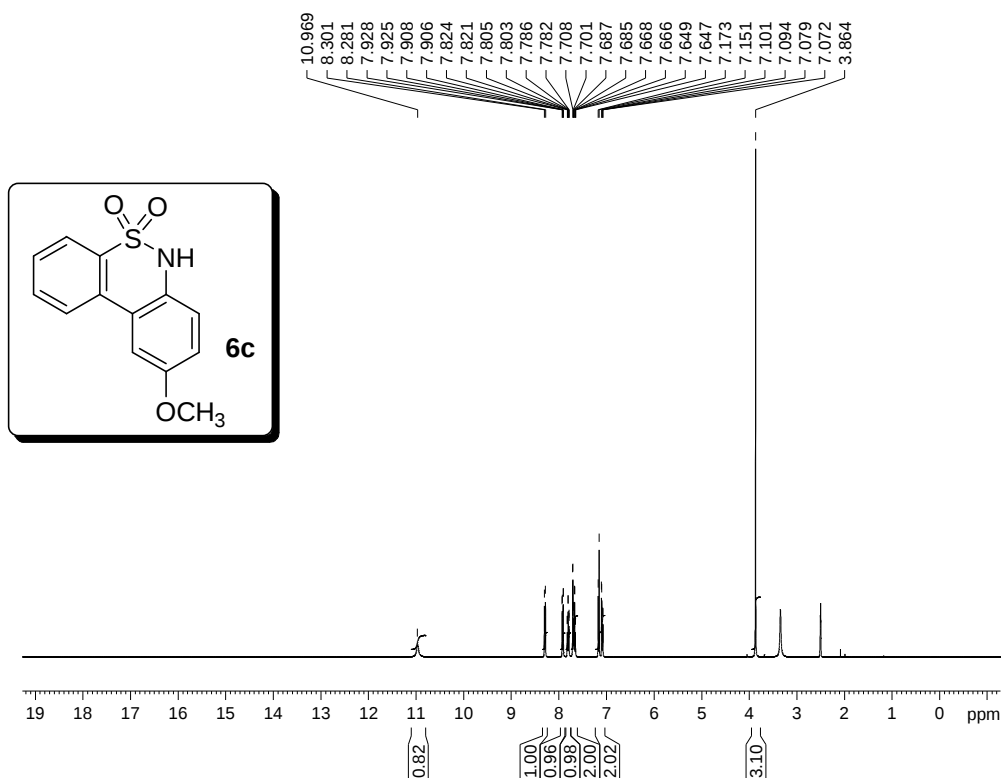


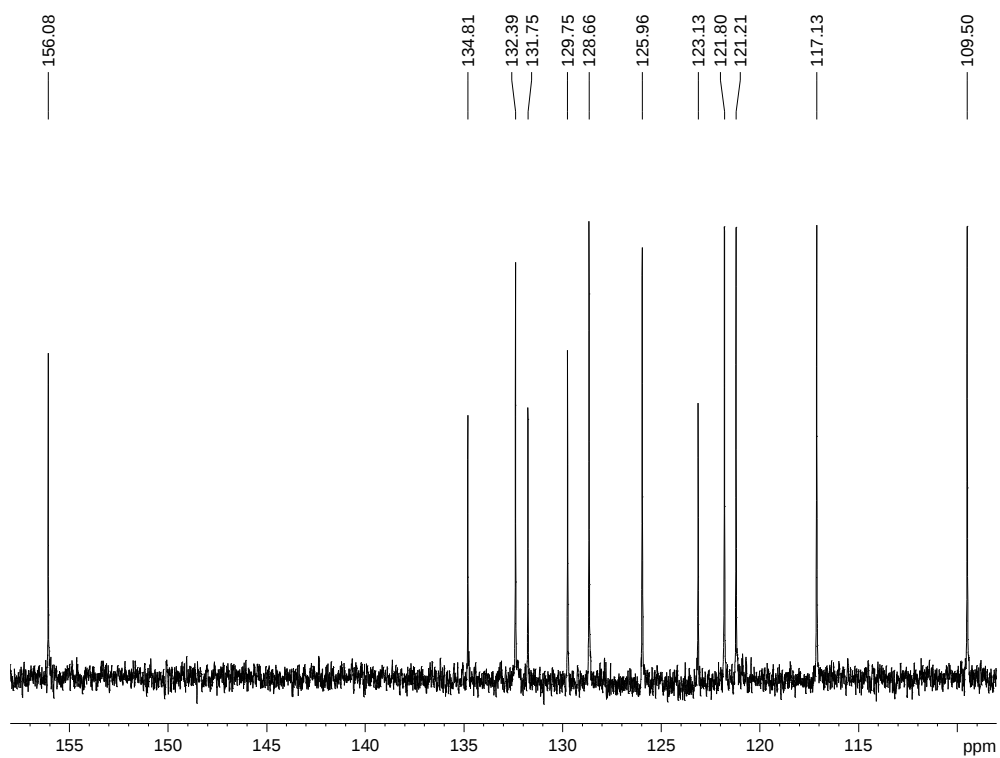
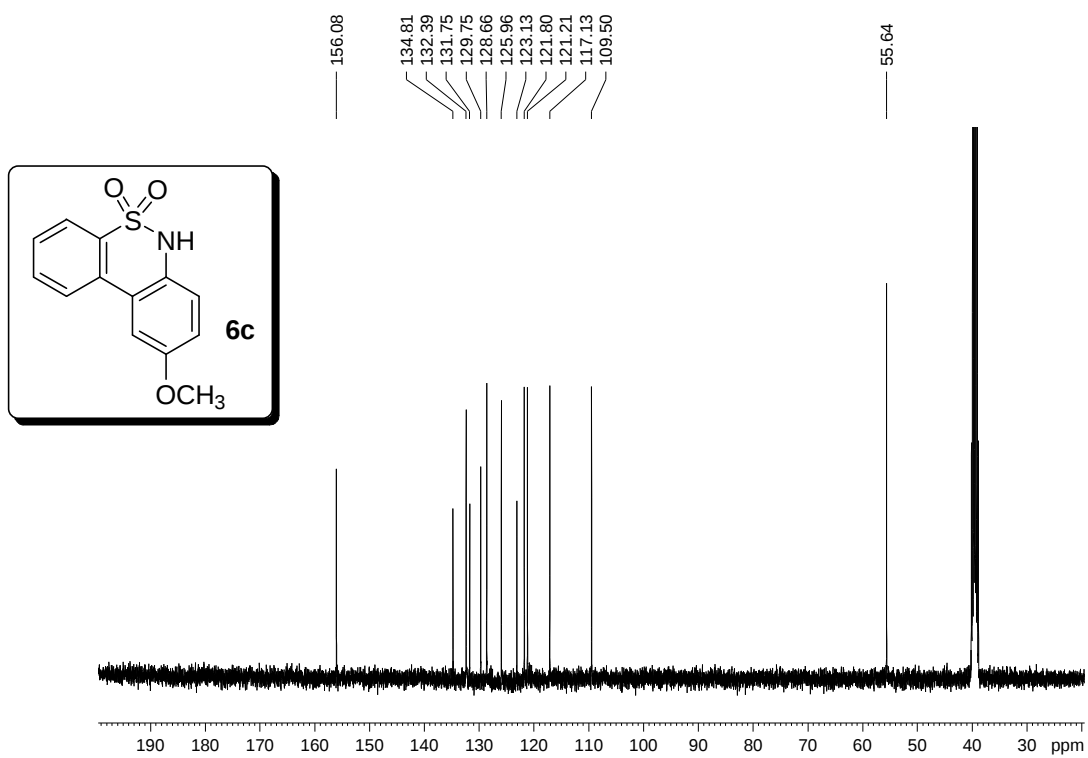
9-Methyl-6H-dibenzo[*c,e*][1,2]thiazine 5,5-dioxide (**6b**)



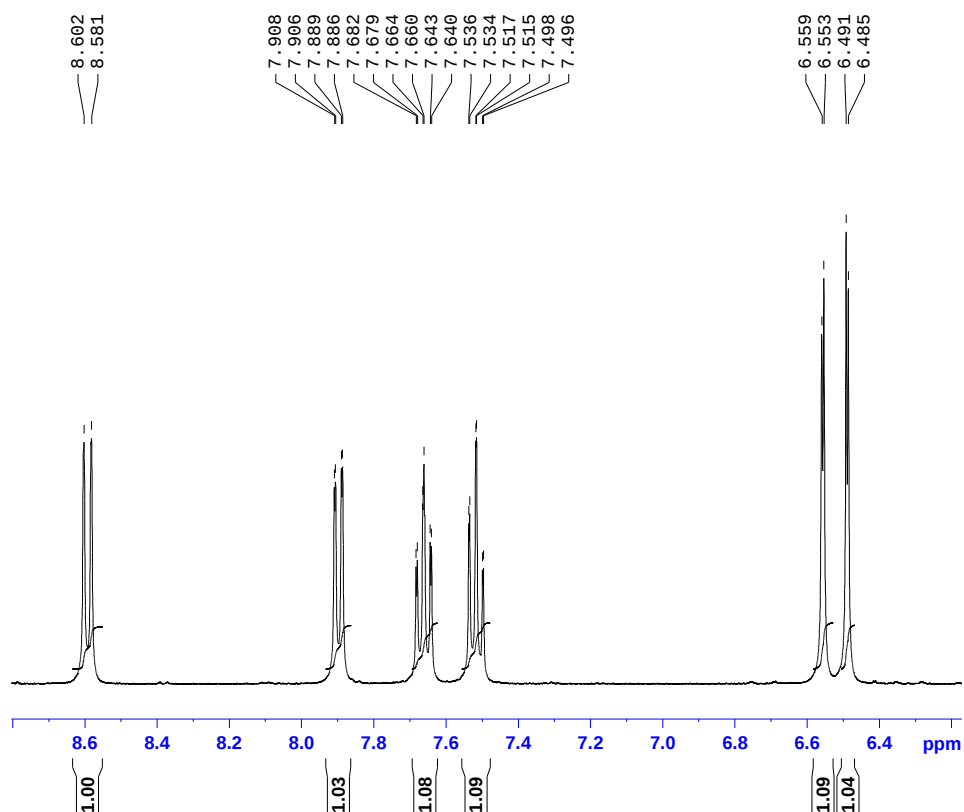
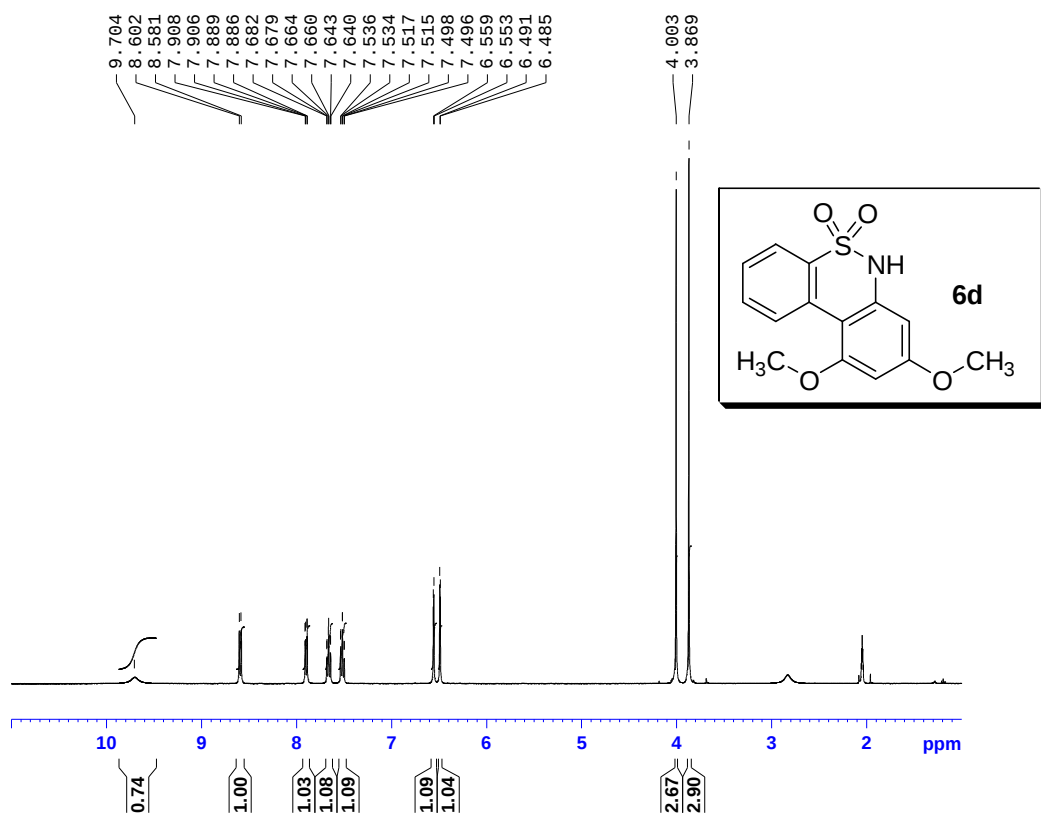


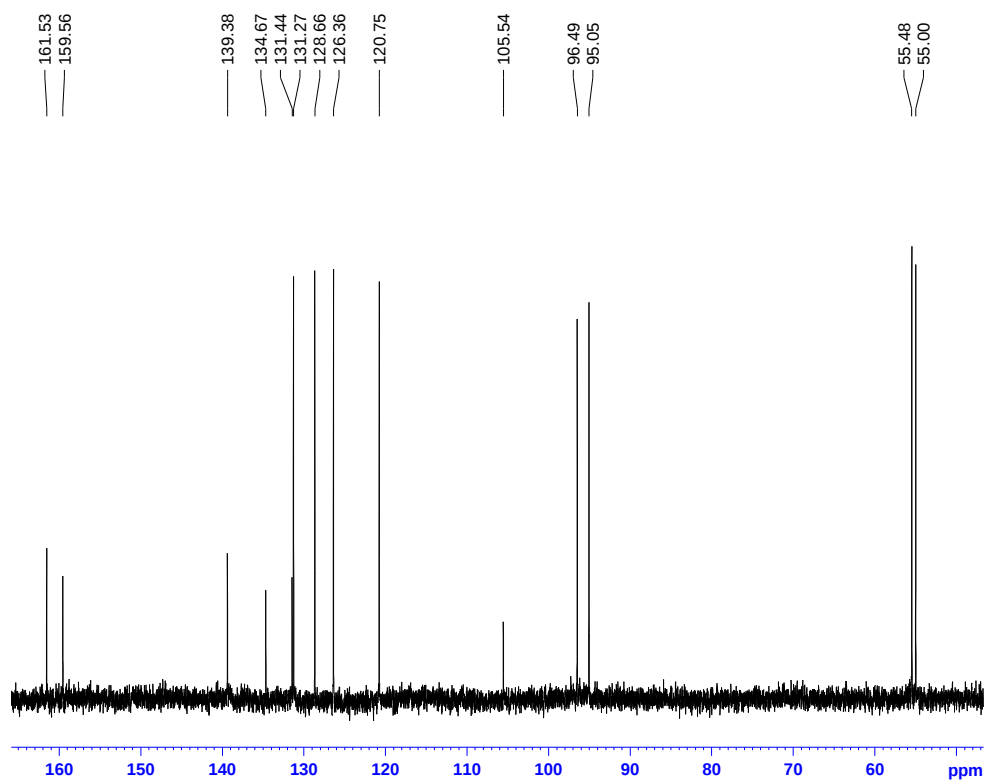
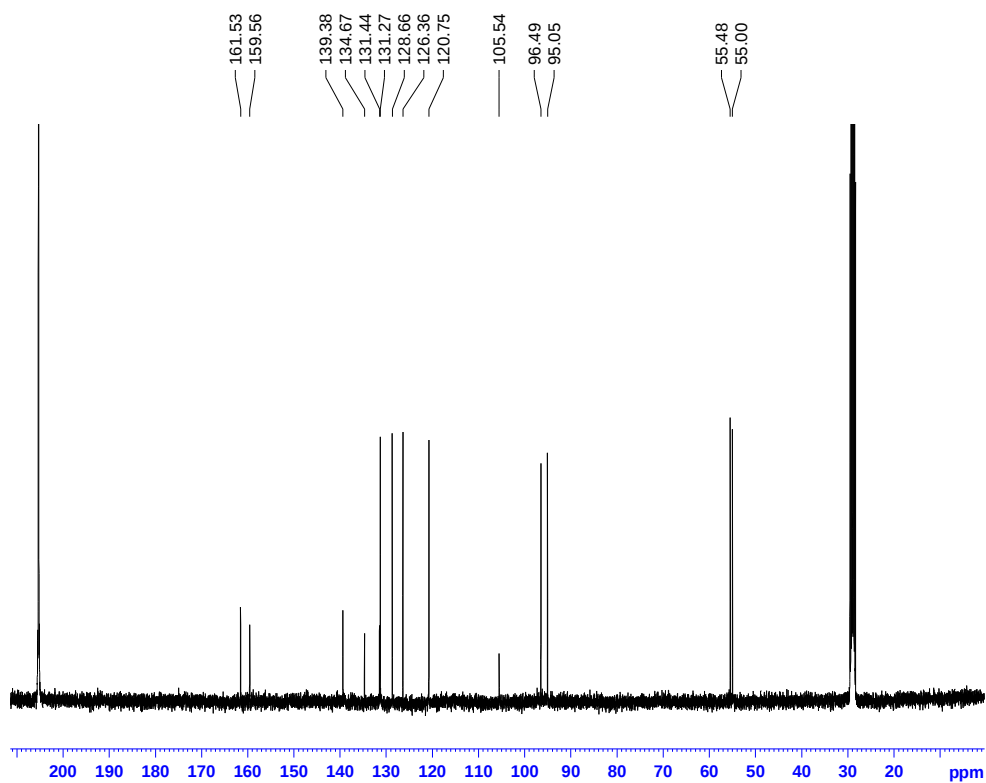
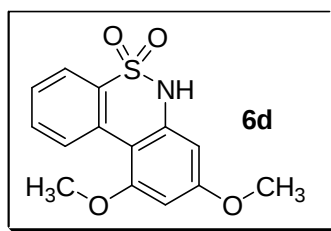
9-Methoxy-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (**6c**)



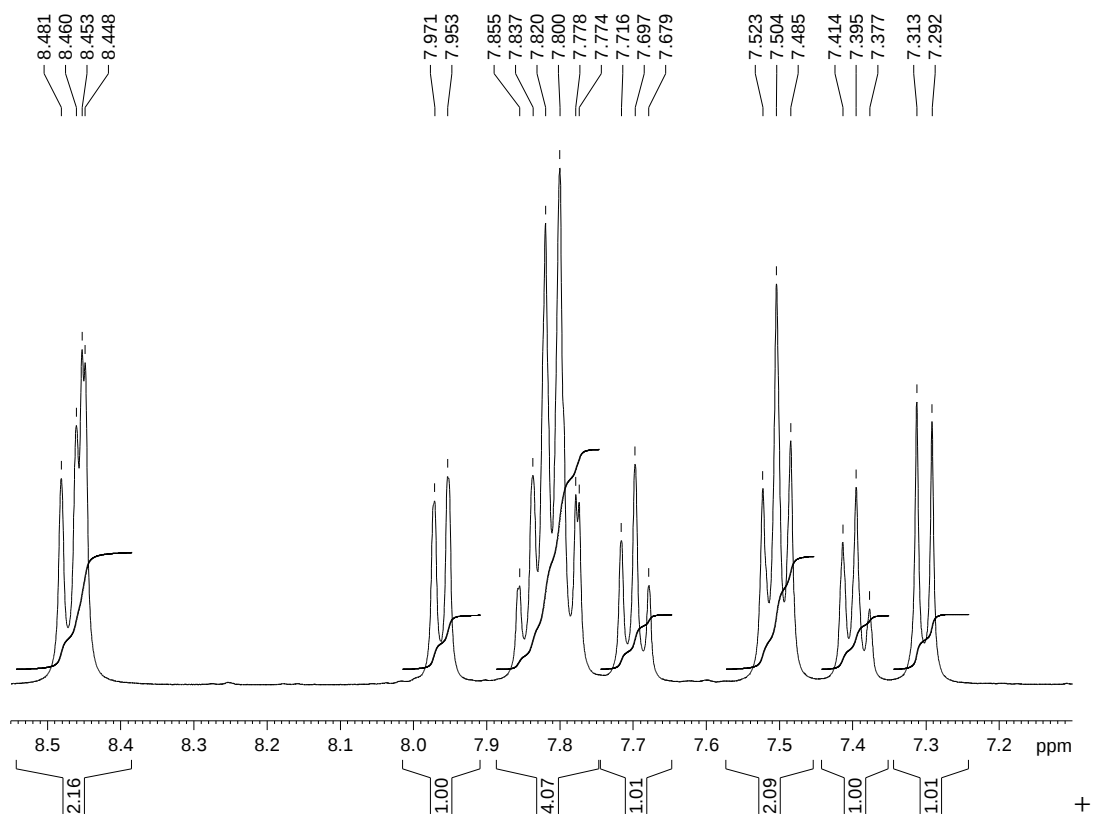
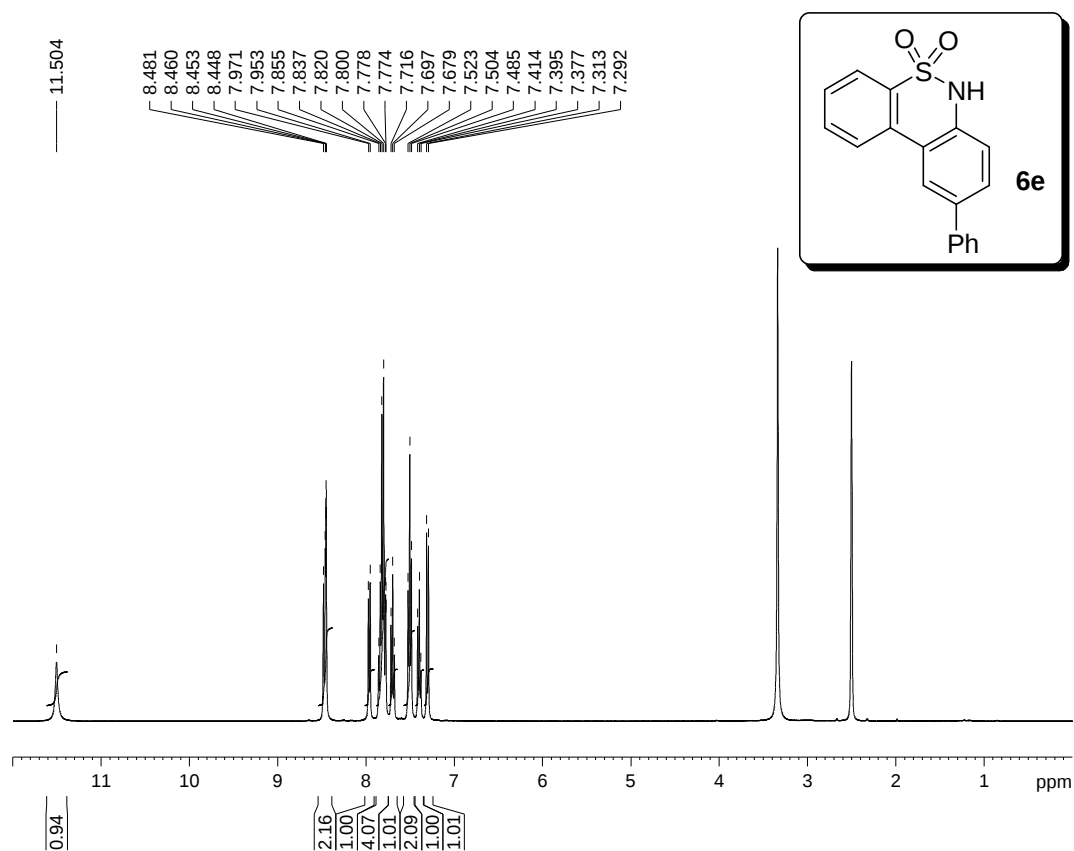


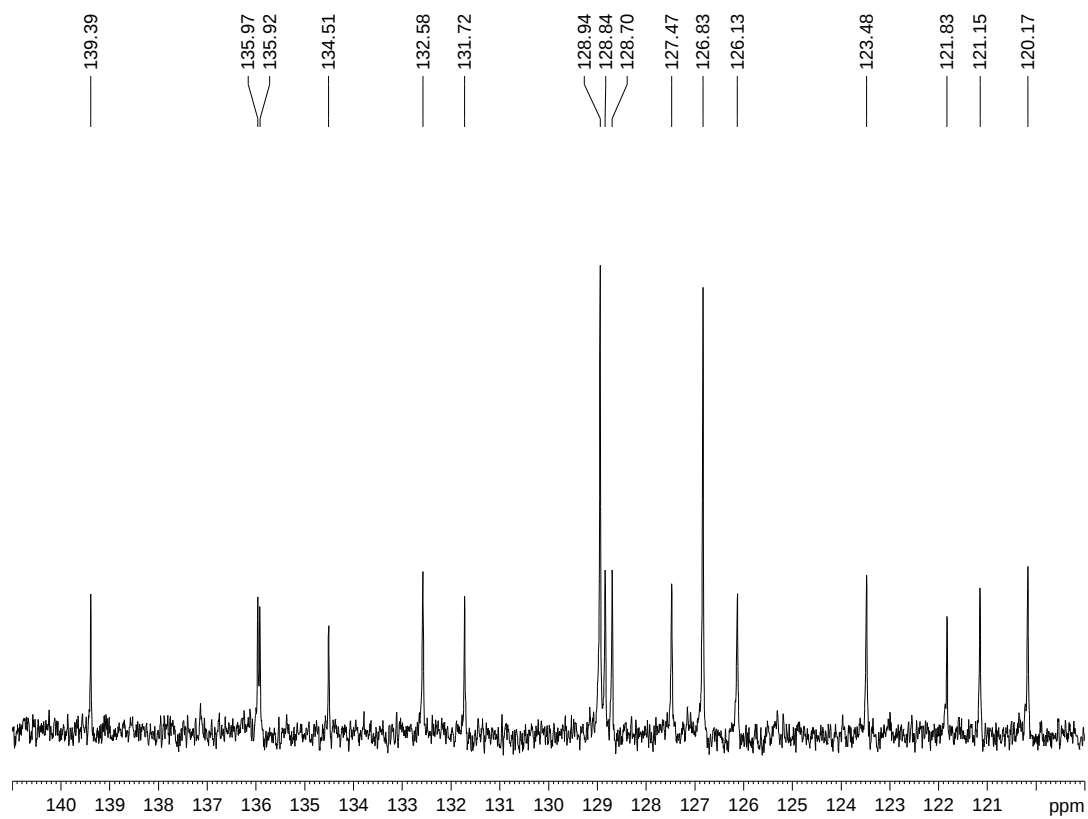
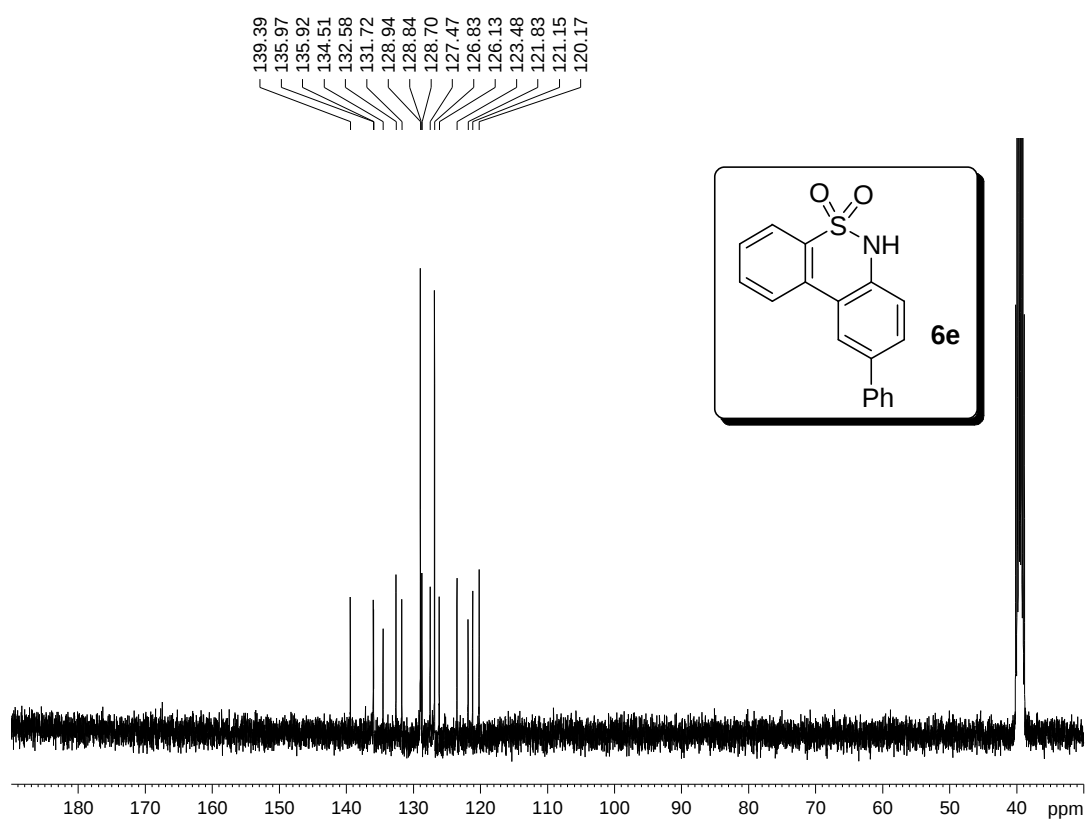
8,10-Di-methoxy-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (**6d**)



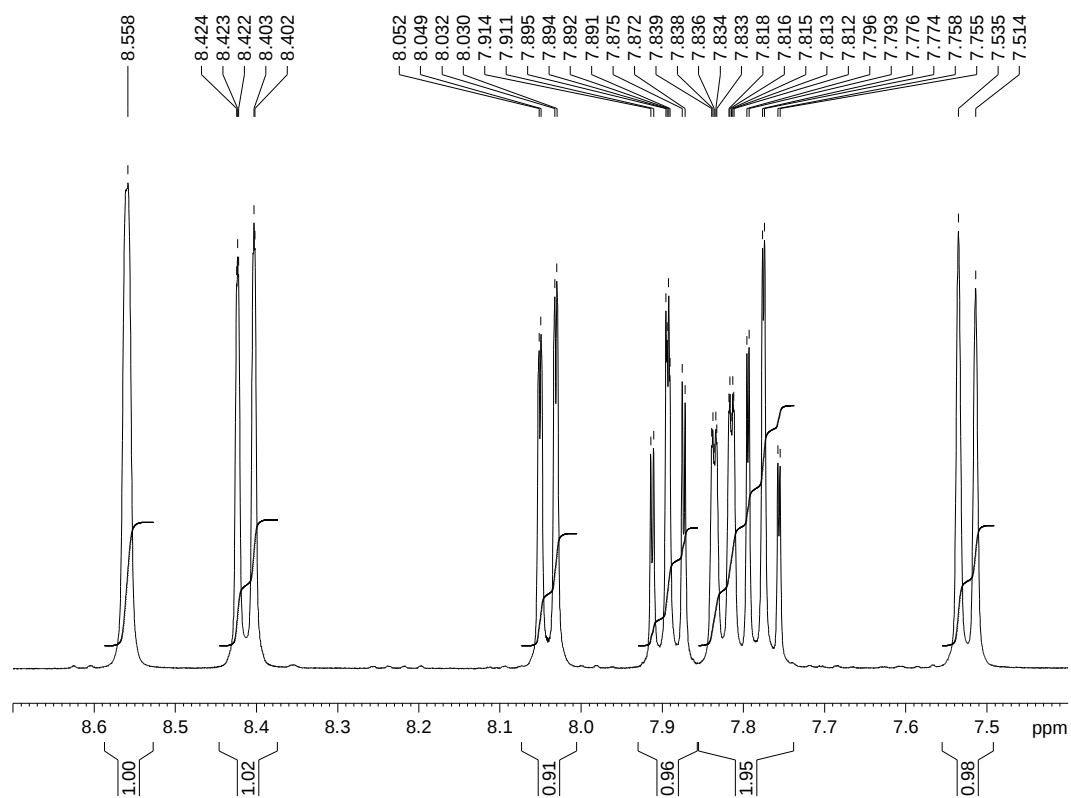
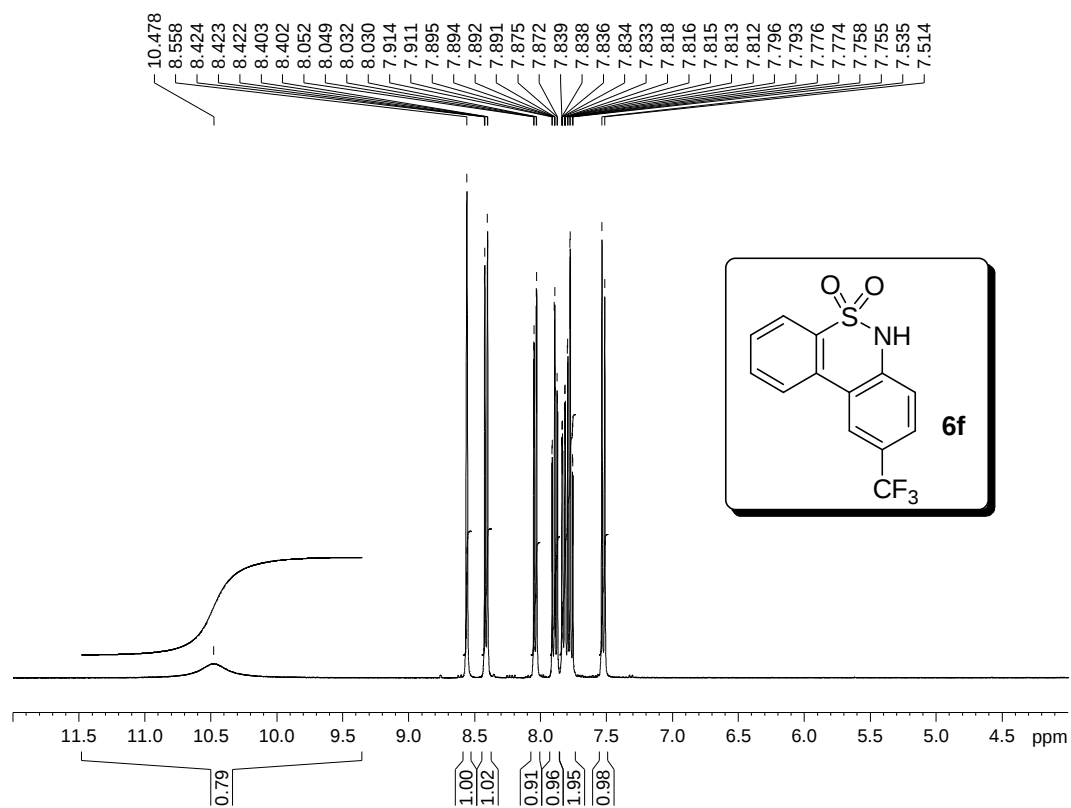


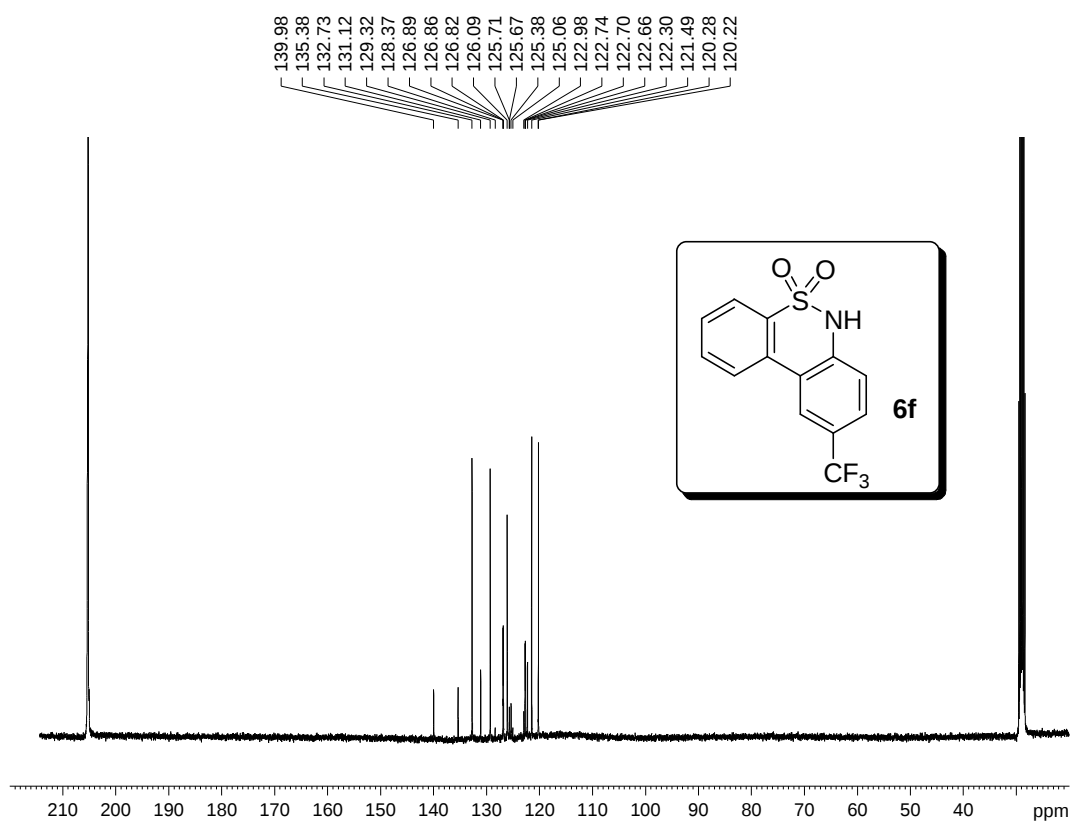
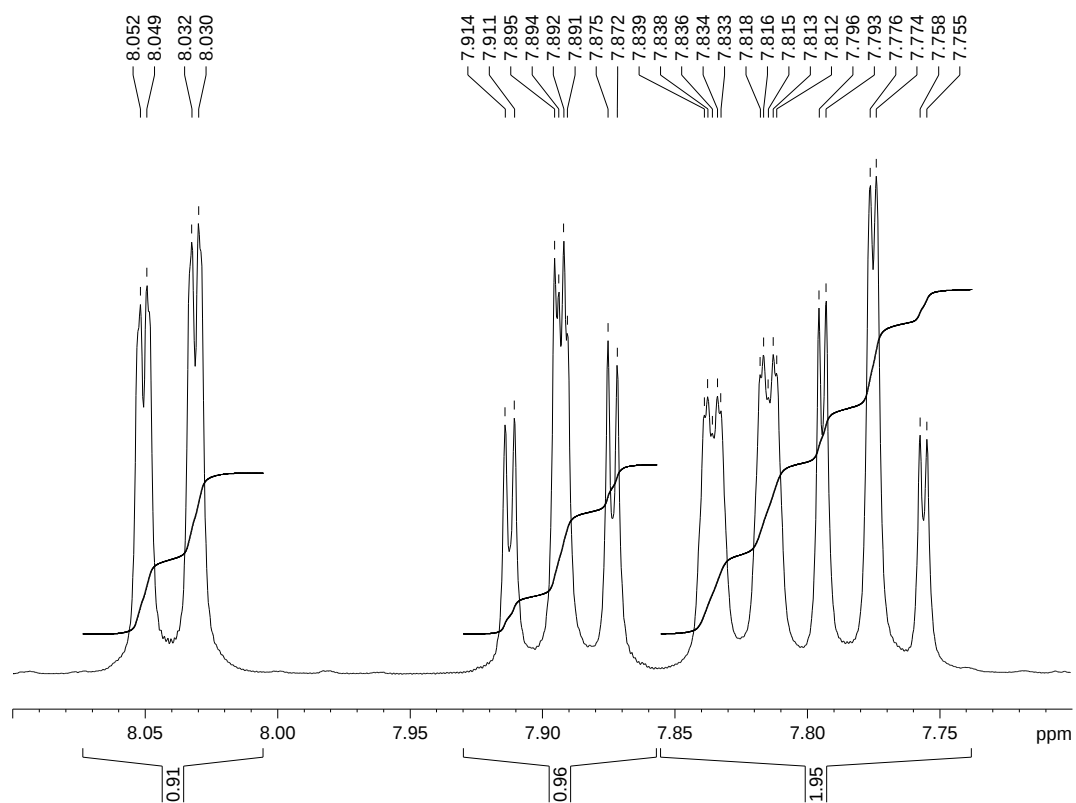
9-Phenyl-6H-dibenzo[*c,e*][1,2]thiazine 5,5-dioxide (**6e**)

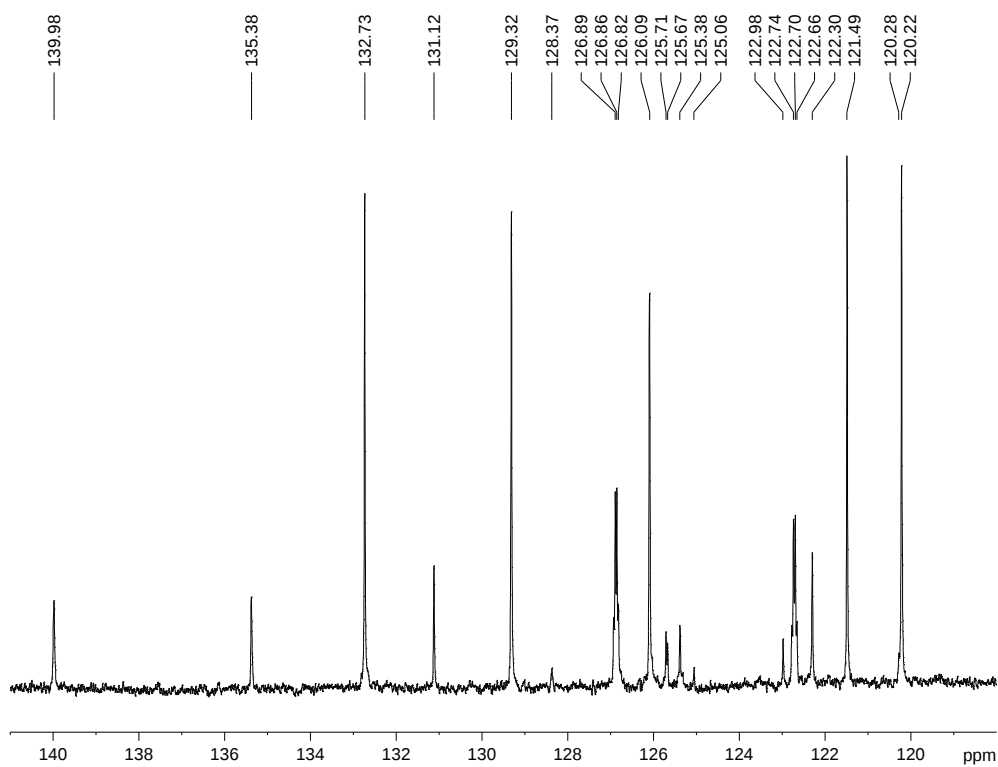




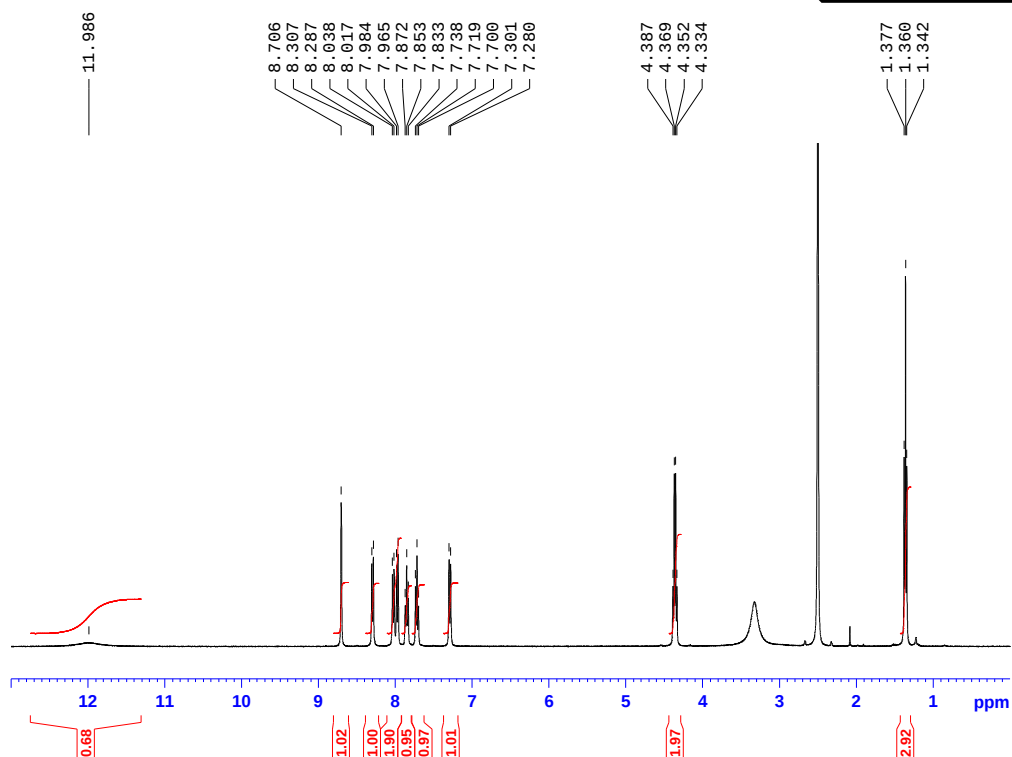
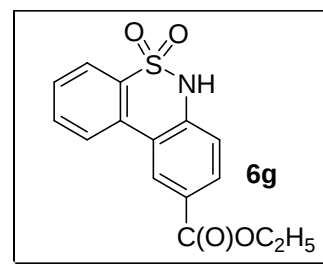
9-(Trifluoromethyl)-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (**6f**)

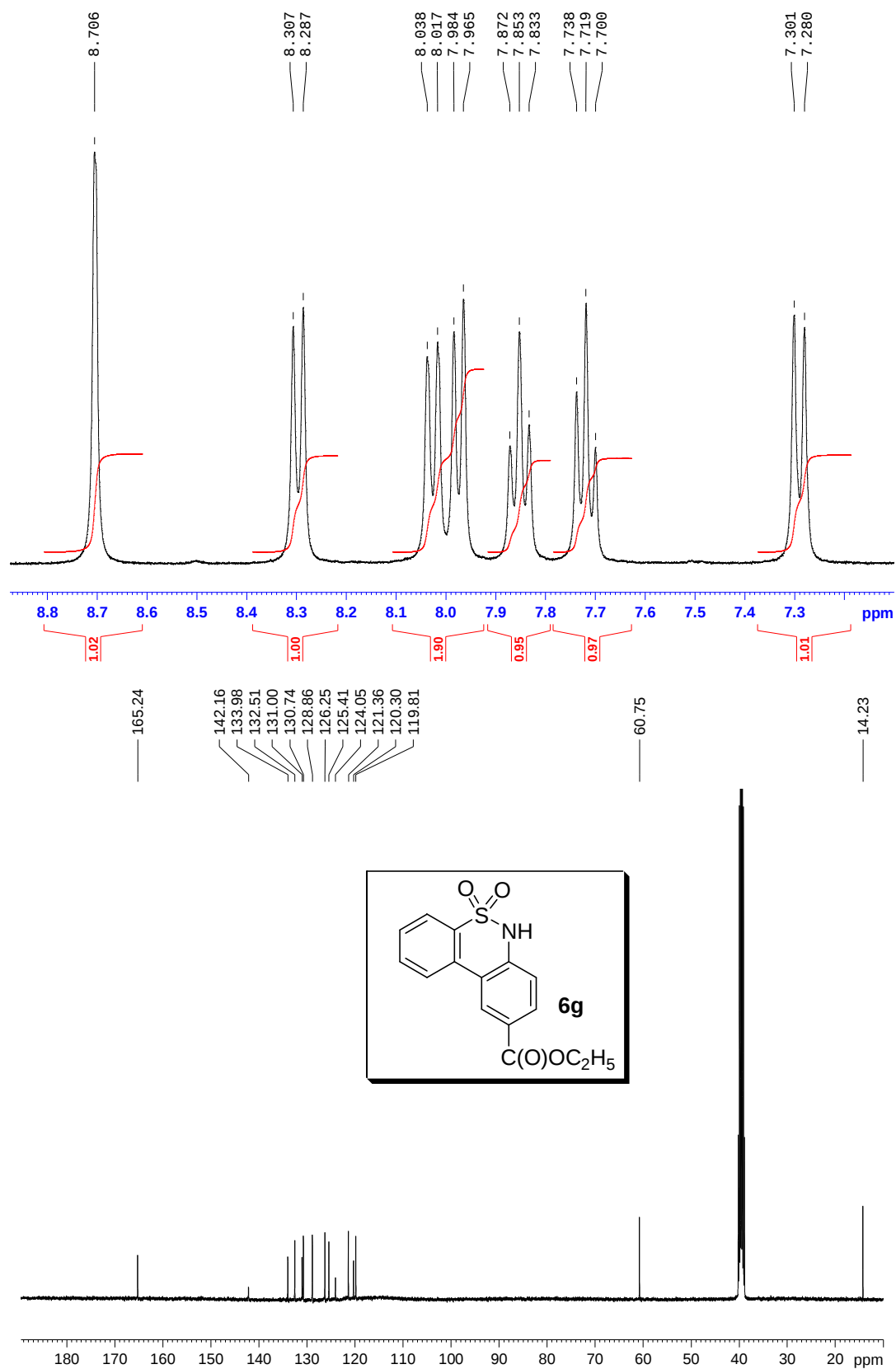


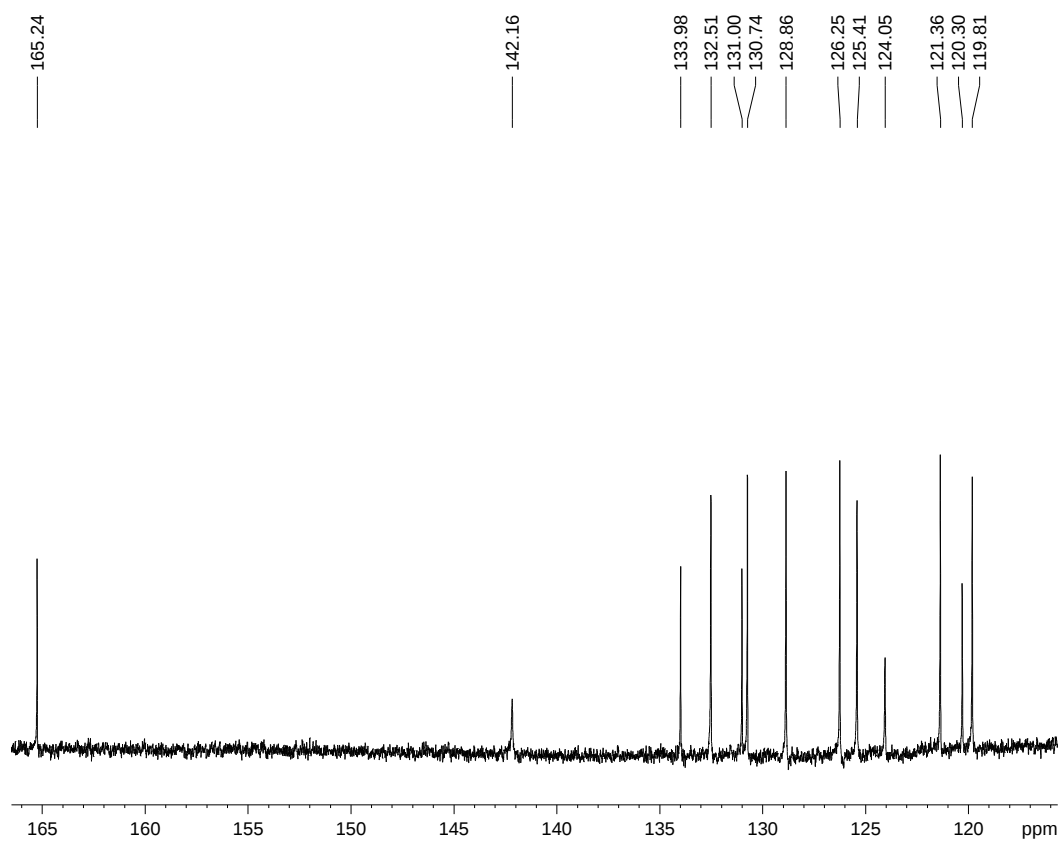




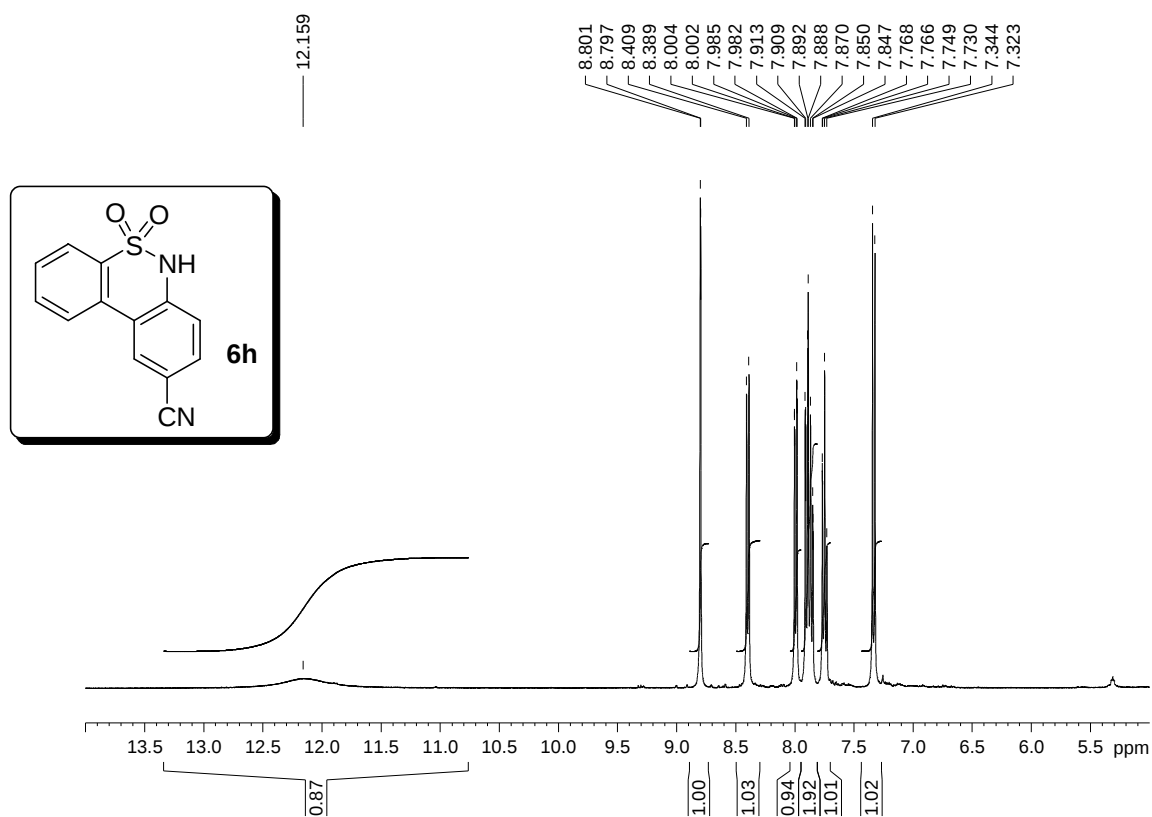
*Ethyl 6H-dibenzo[*c,e*][1,2]thiazine-9-carboxylate 5,5-dioxide (6g)*

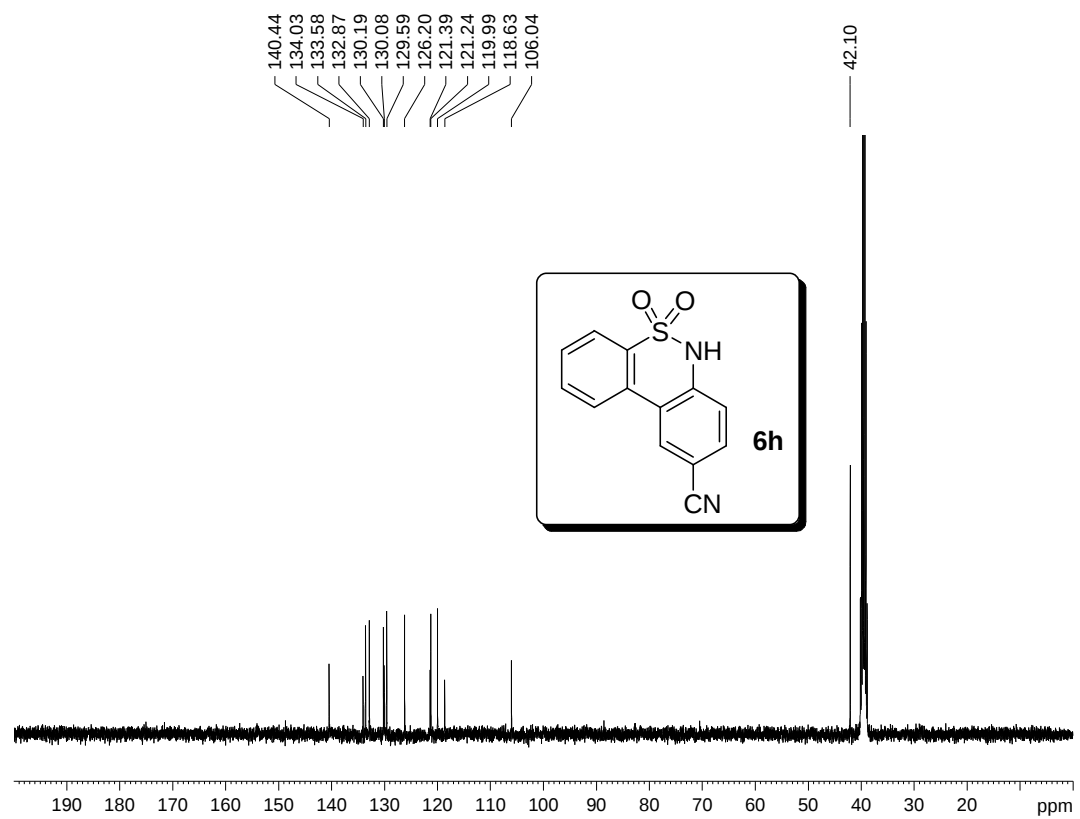
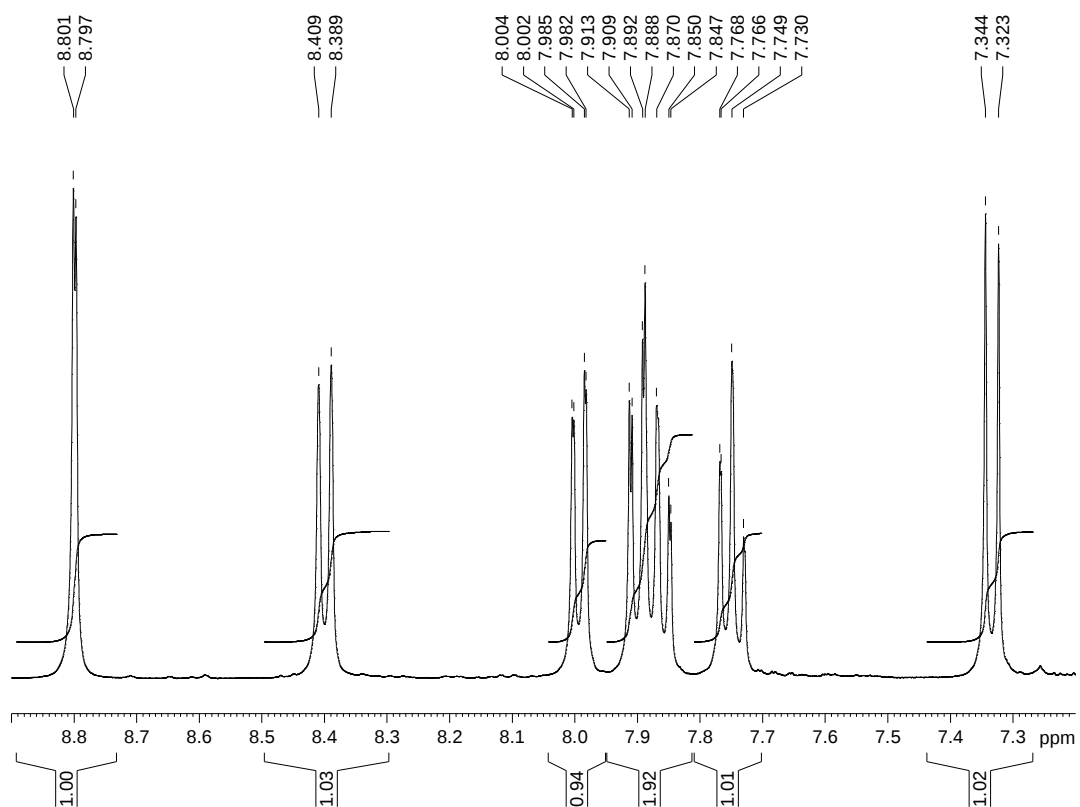


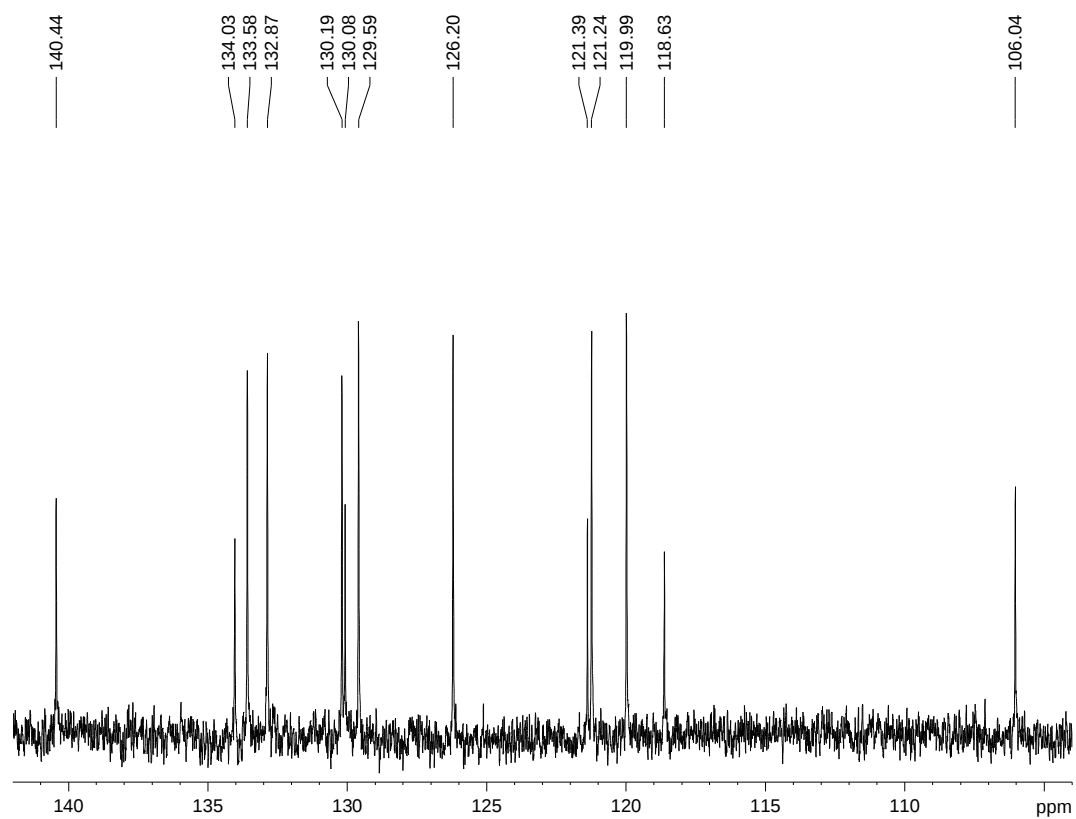




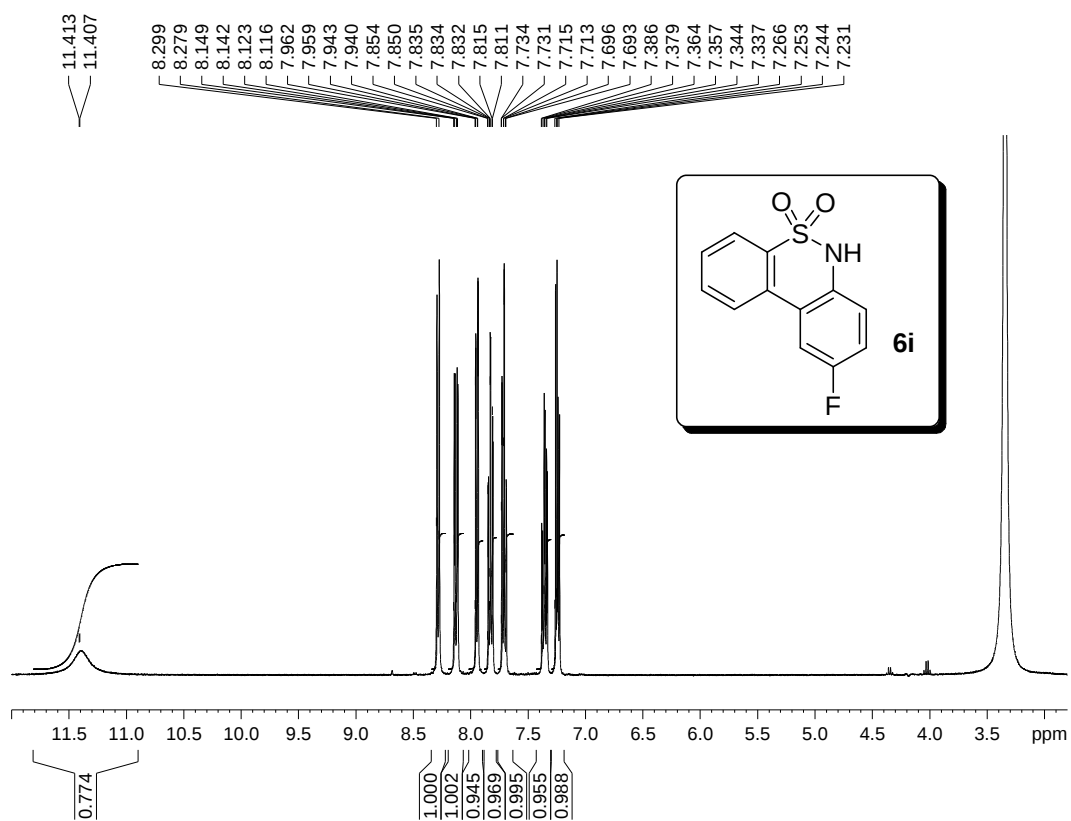
6H-Dibenzo[c,e][1,2]thiazine-9-carbonitrile 5,5-dioxide (6h)

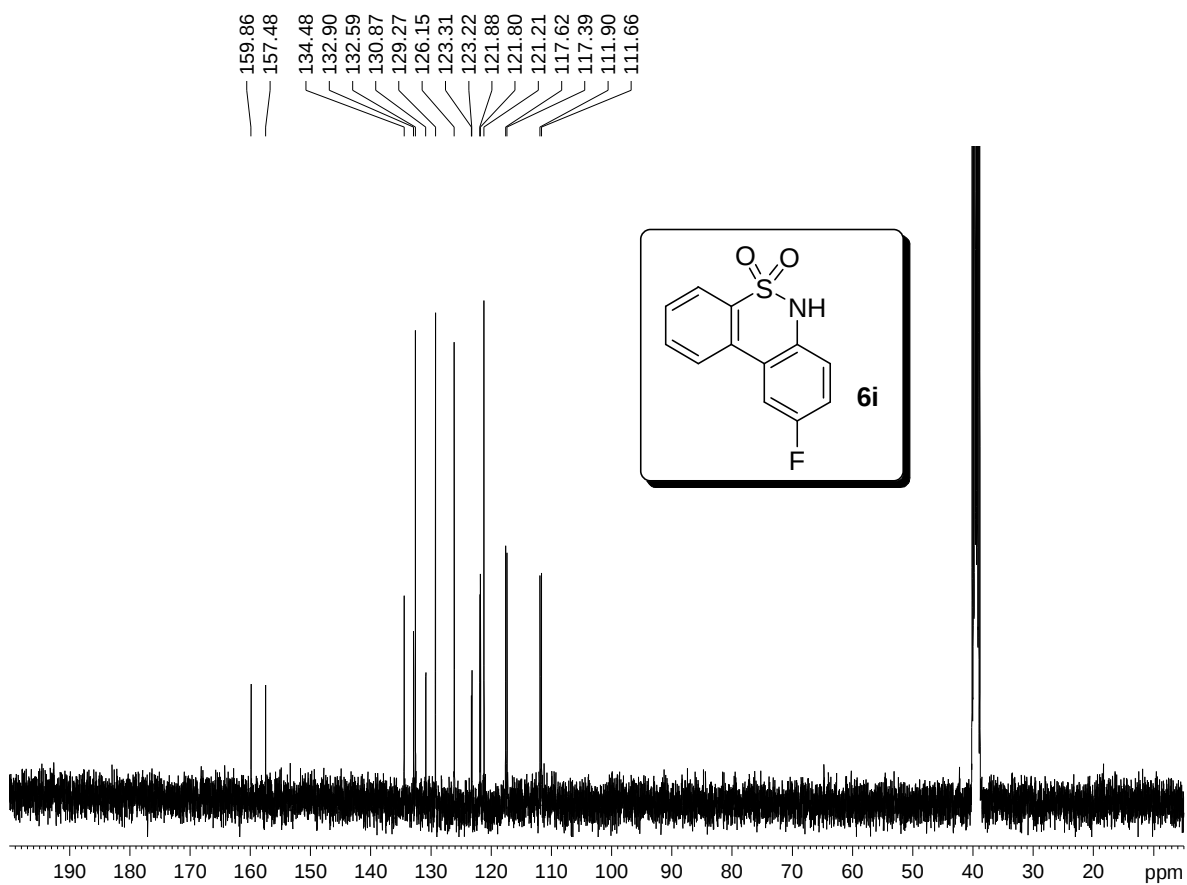
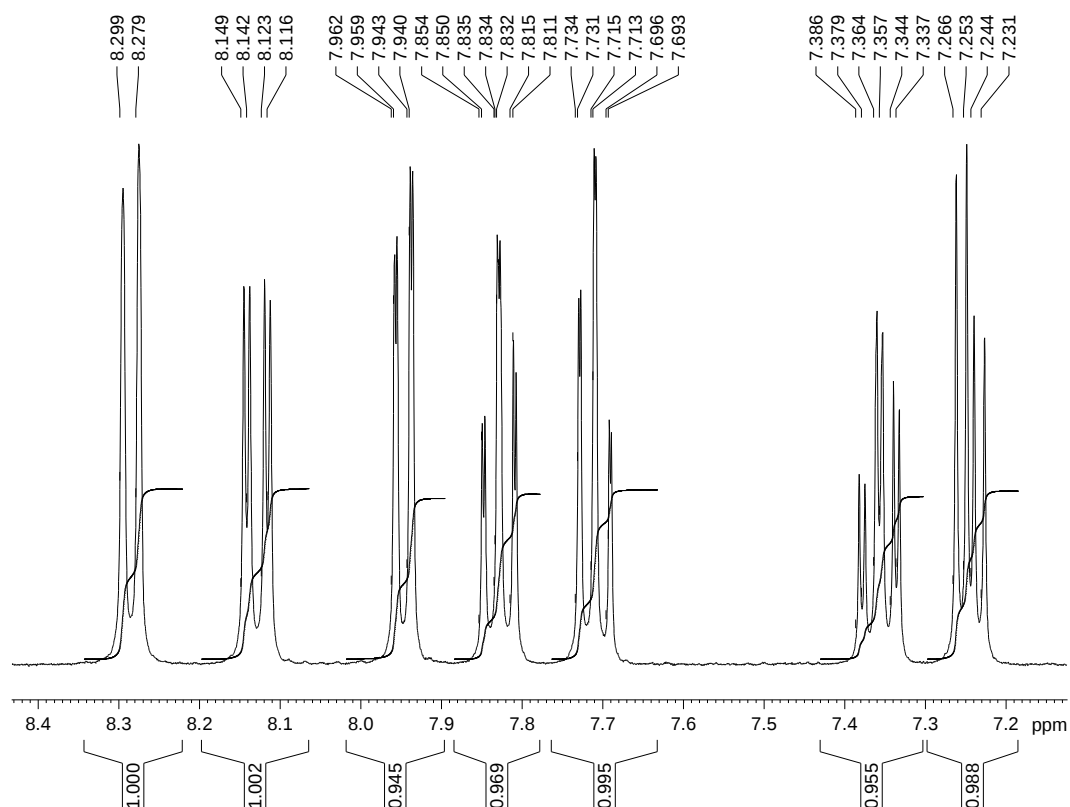


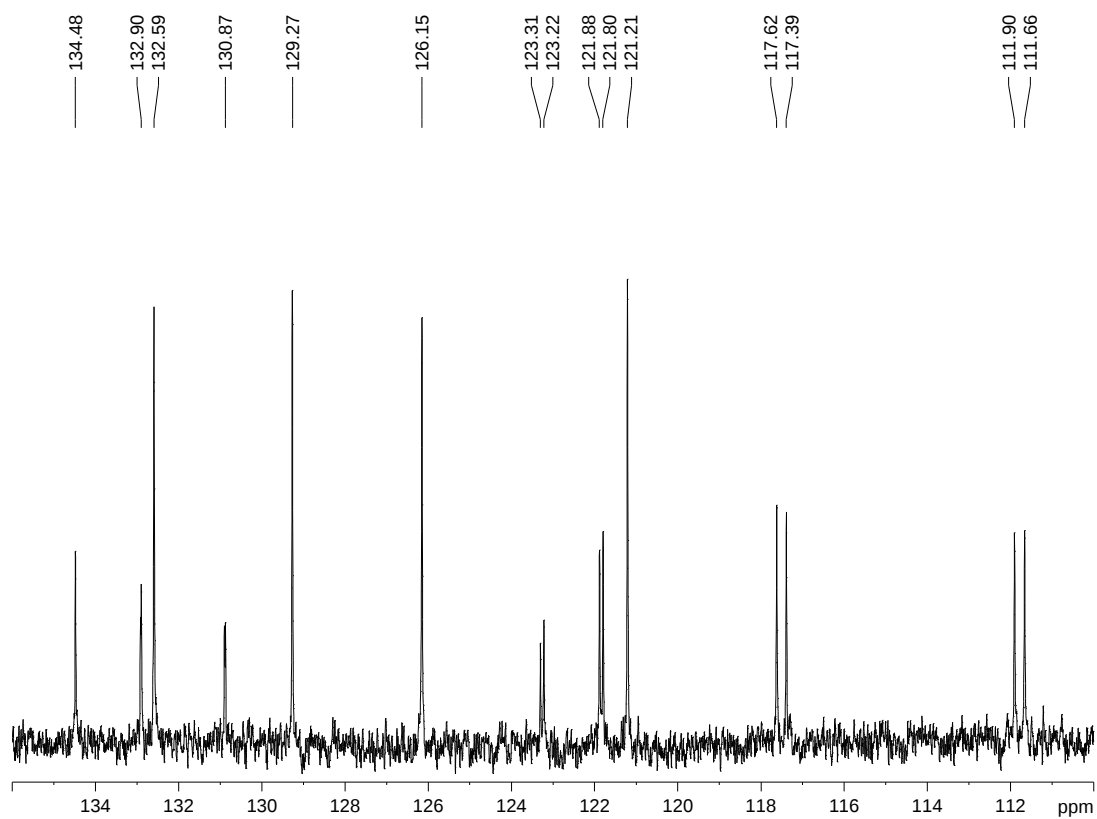




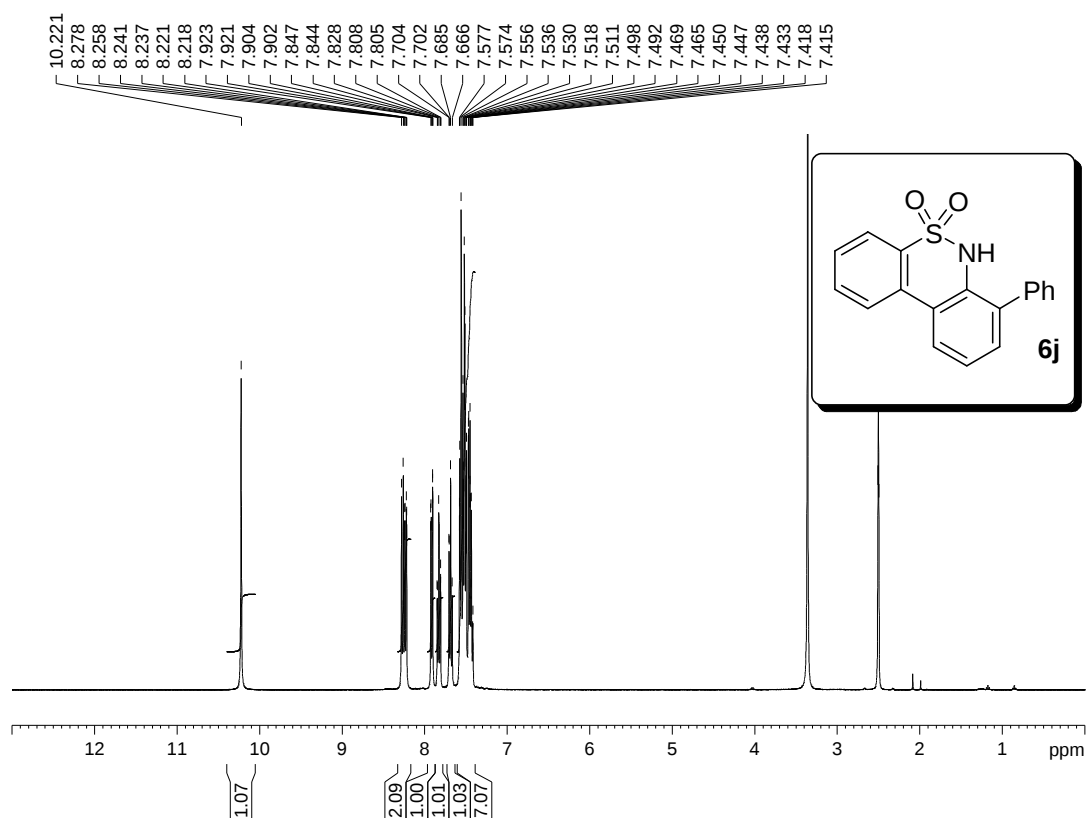
9-Fluoro-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (6i)

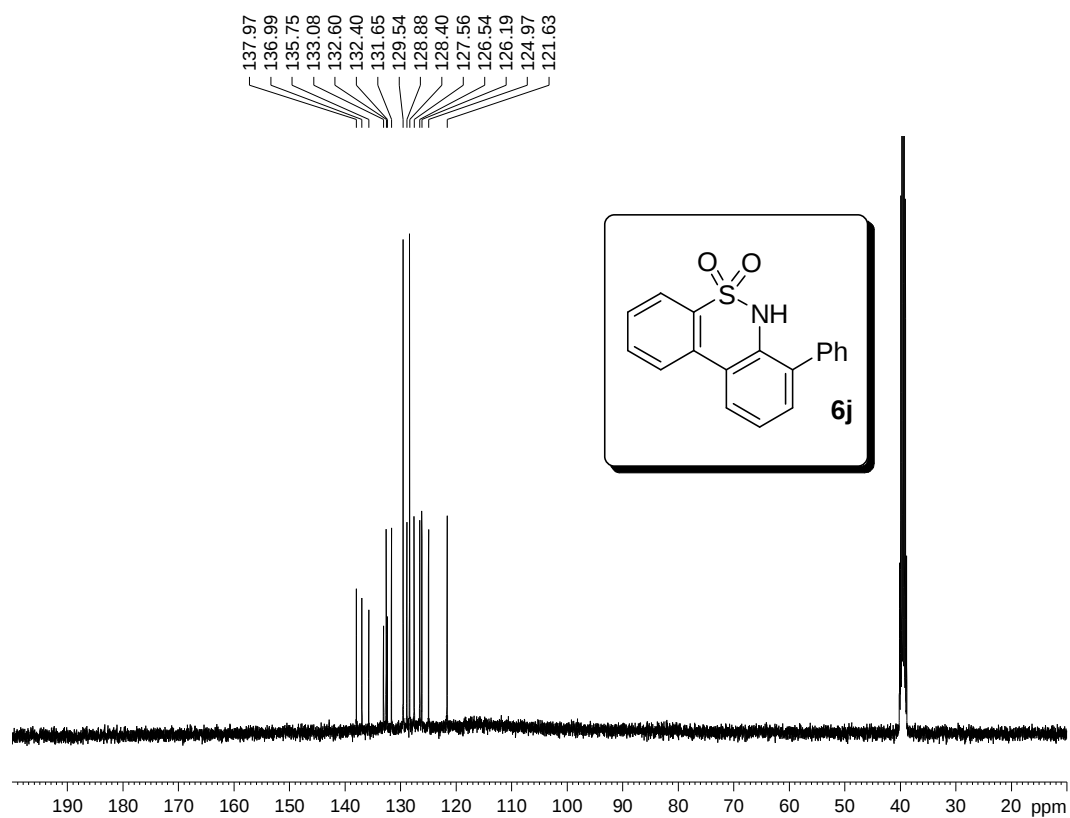
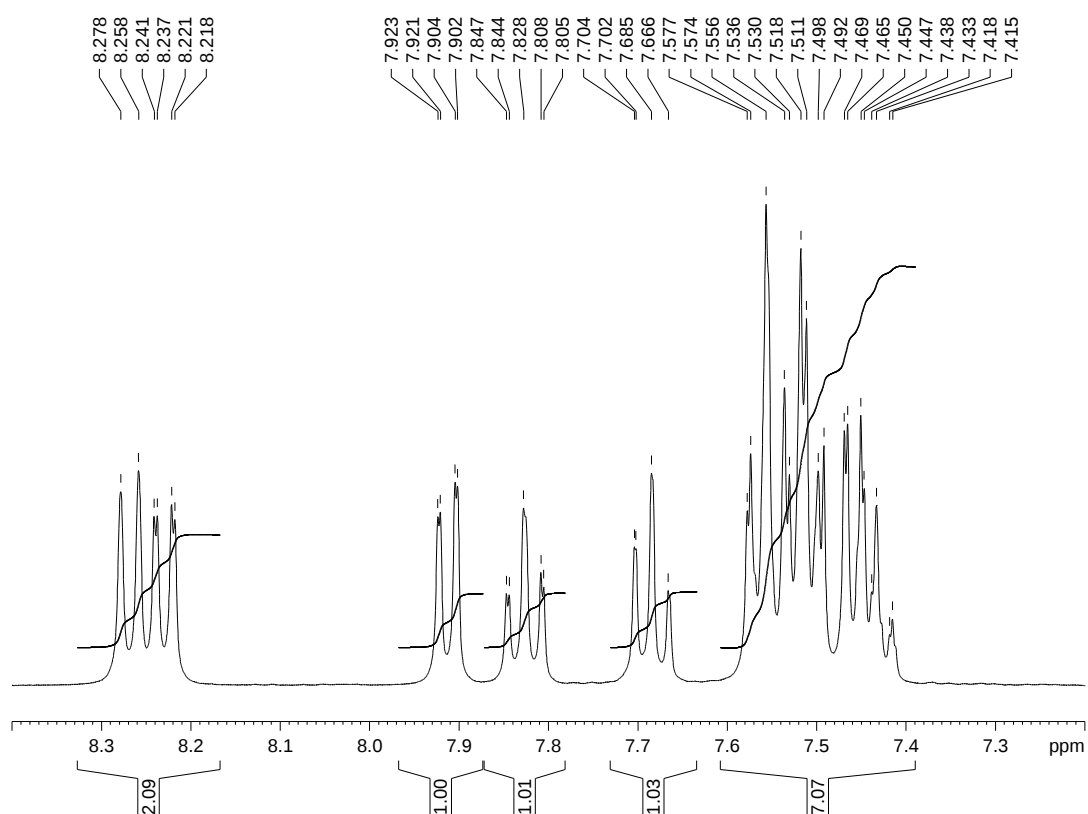


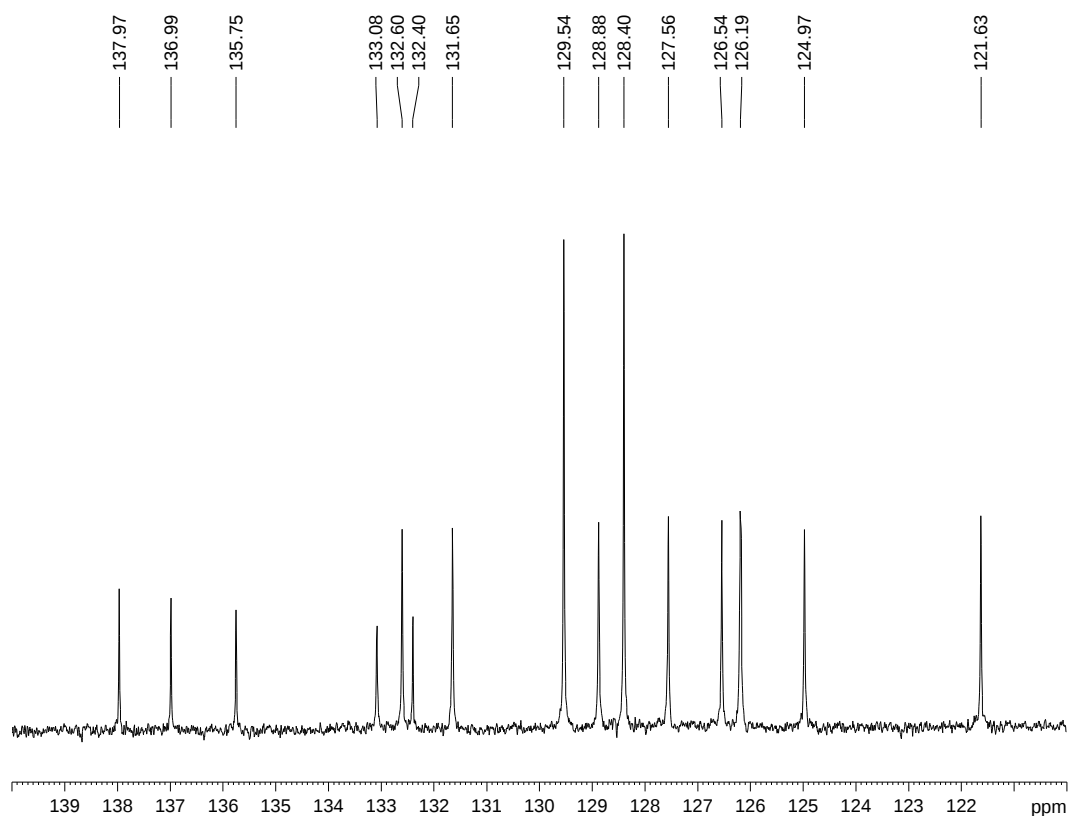




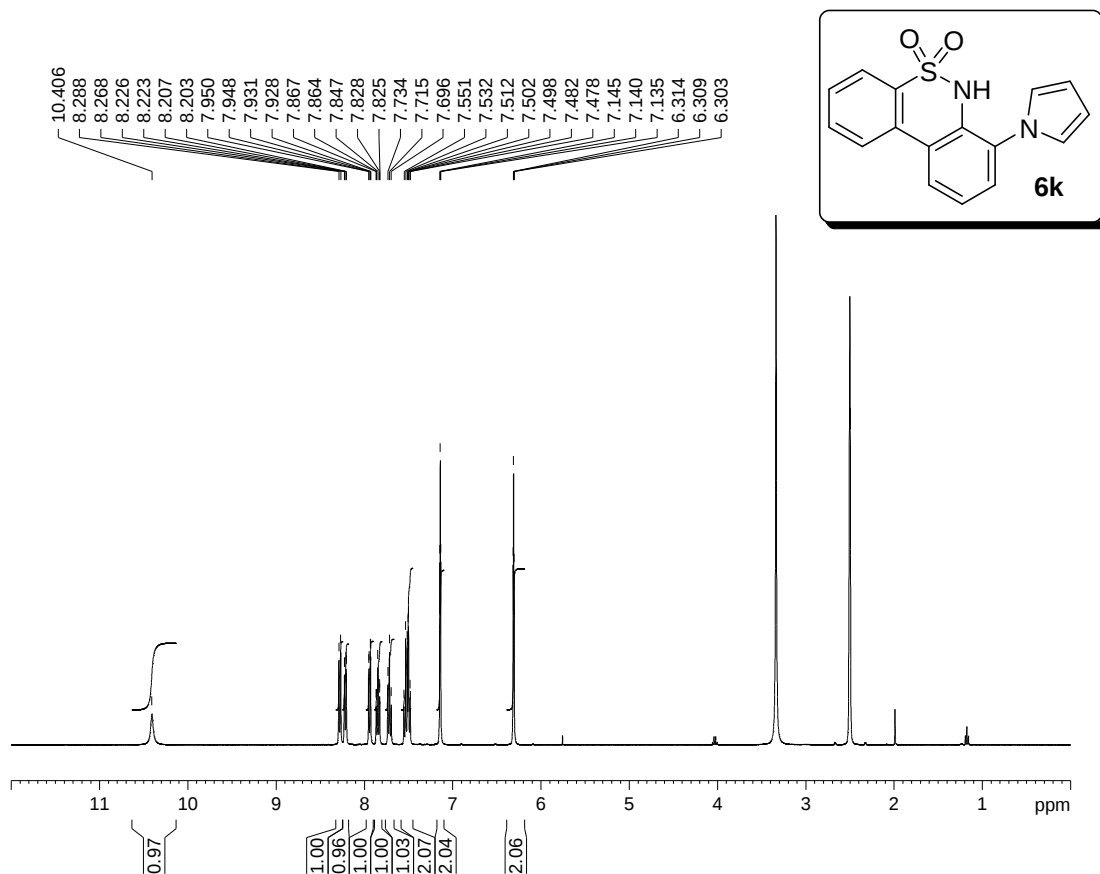
7-Phenyl-6H-dibenzo[*c,e*][1,2]thiazine 5,5-dioxide (**6j**)

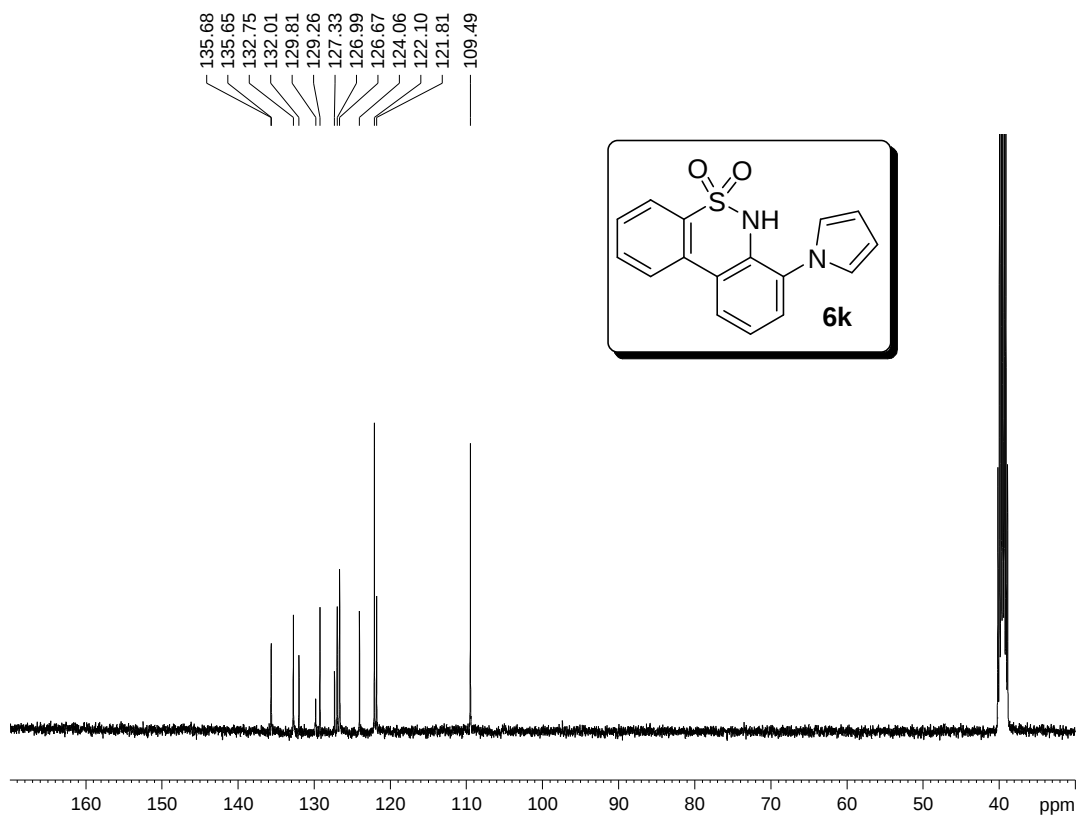
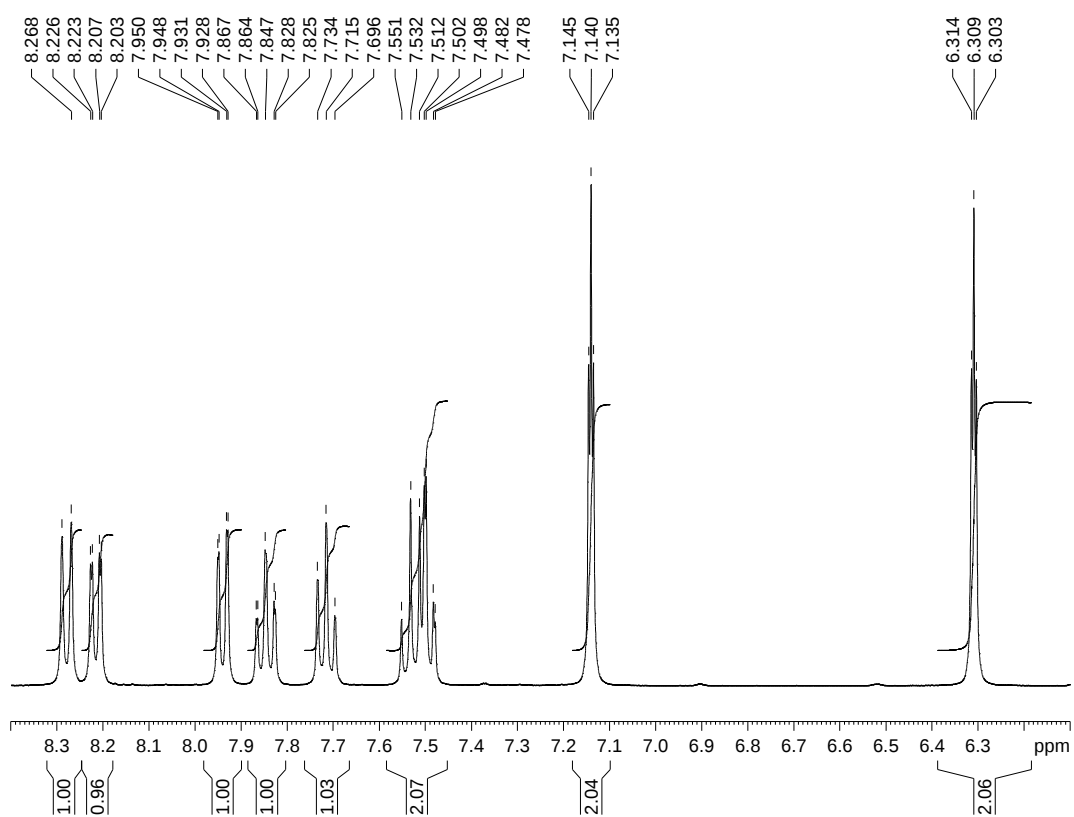


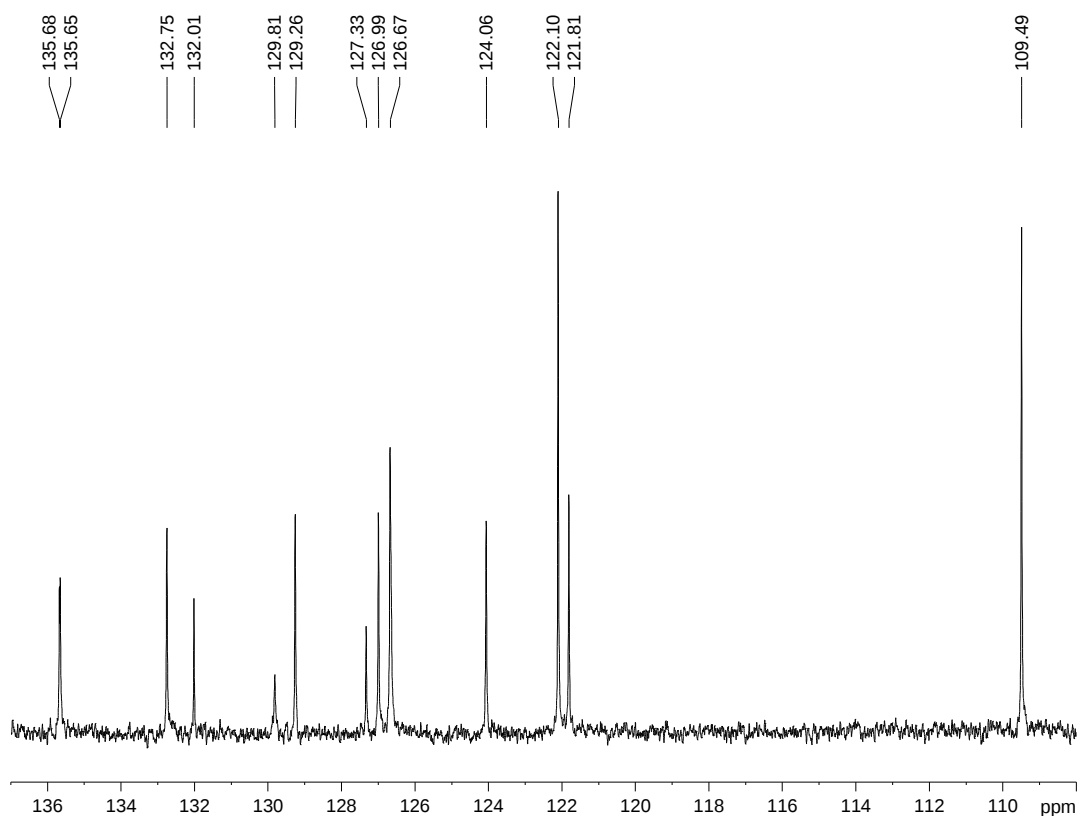




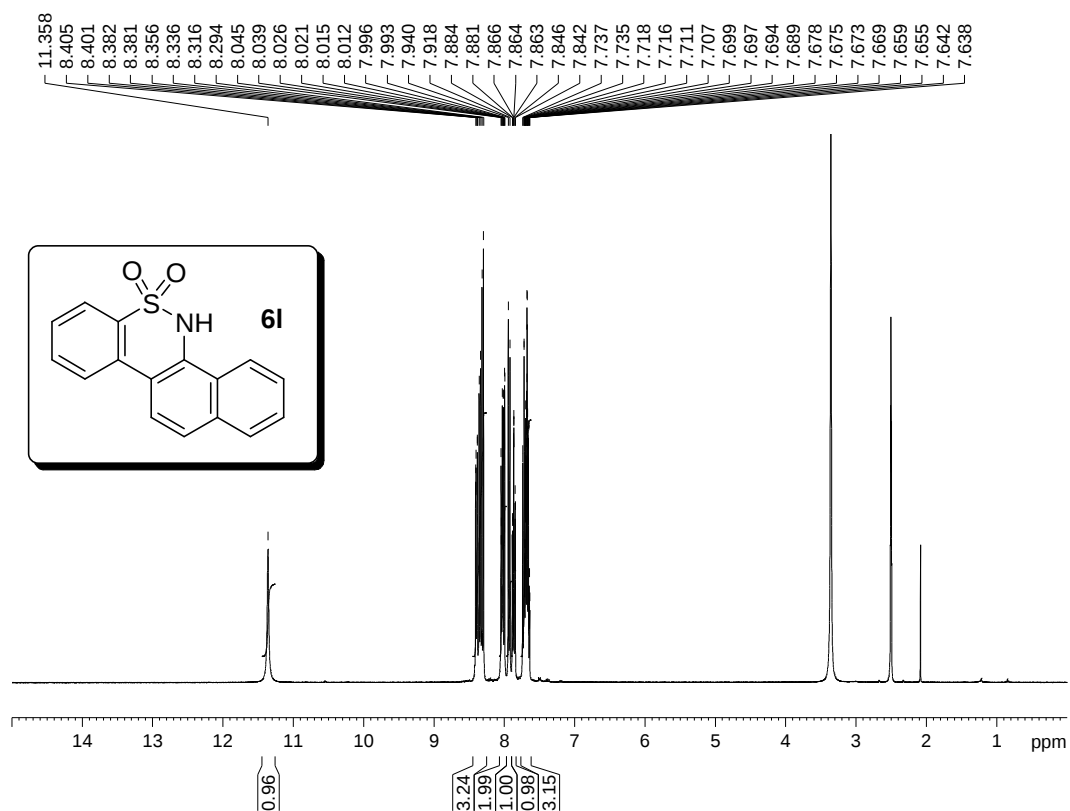
7-(1H-Pyrrol-1-yl)-6H-dibenzo[*c,e*][1,2]thiazine 5,5-dioxide (**6k**)

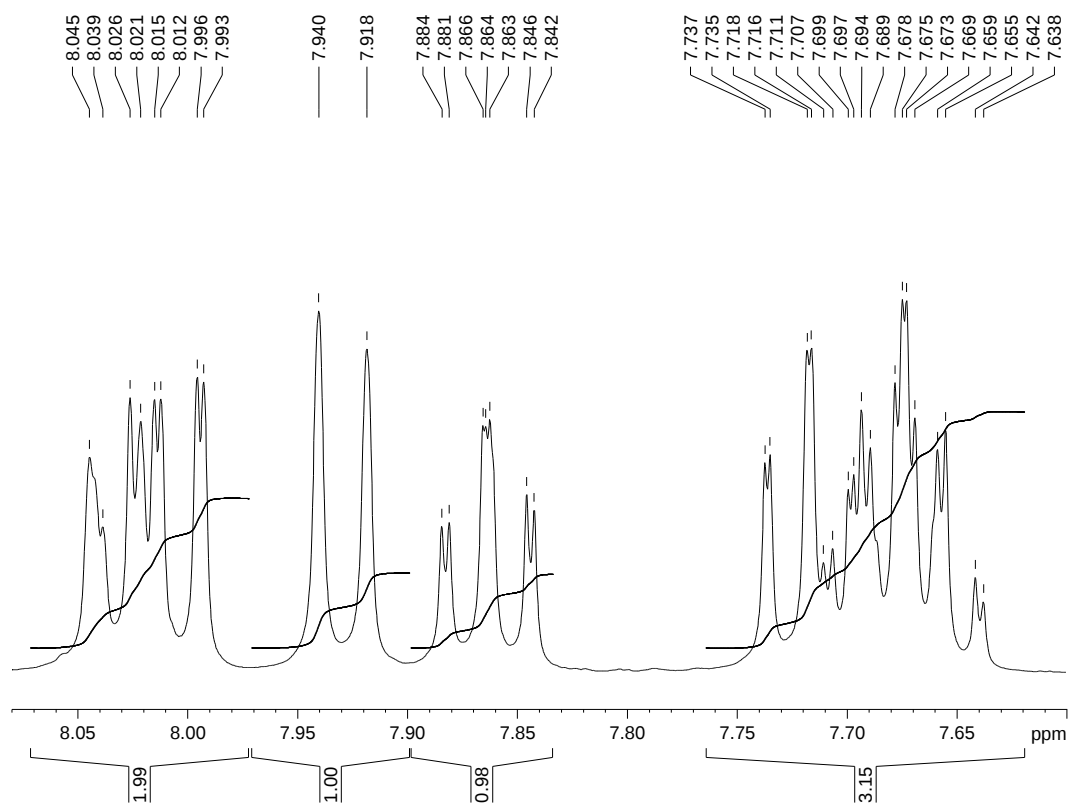
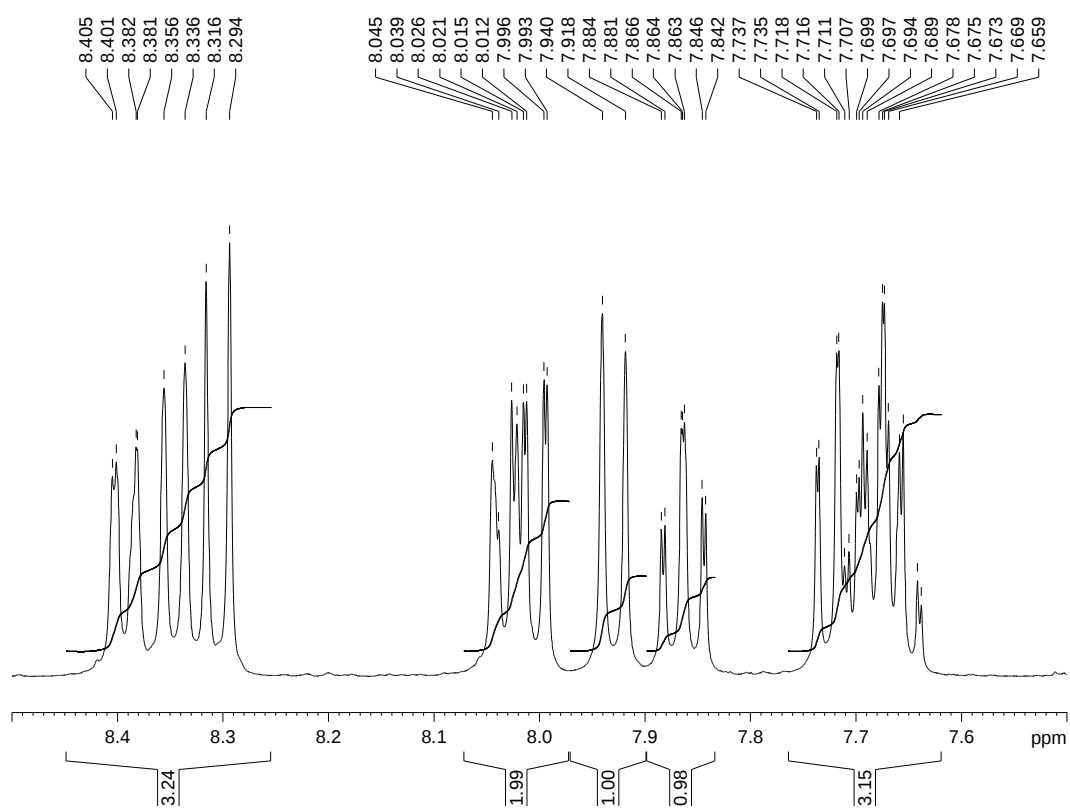


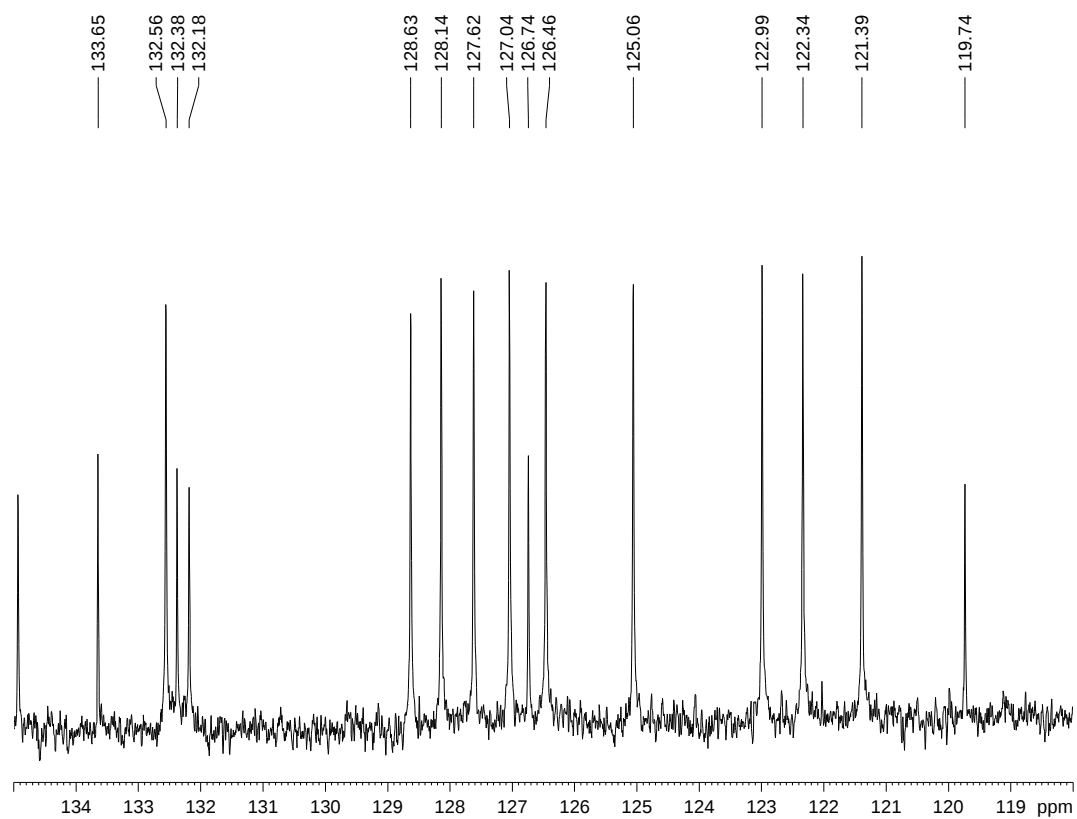
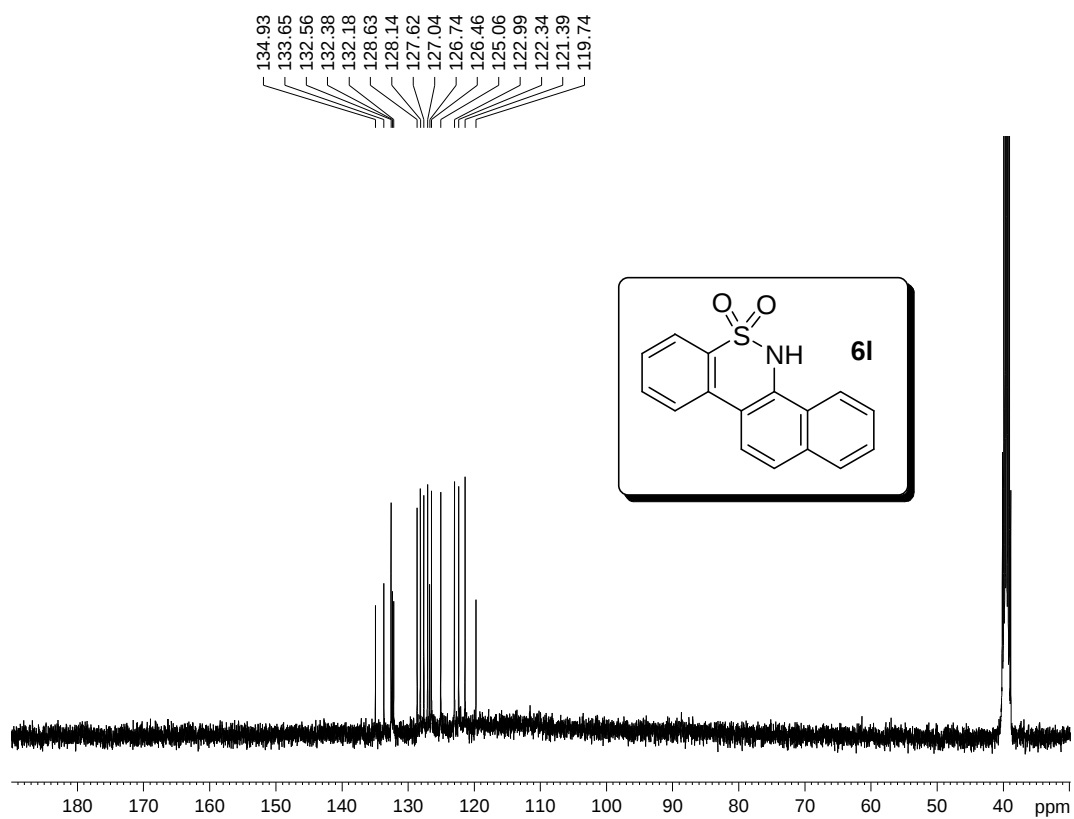




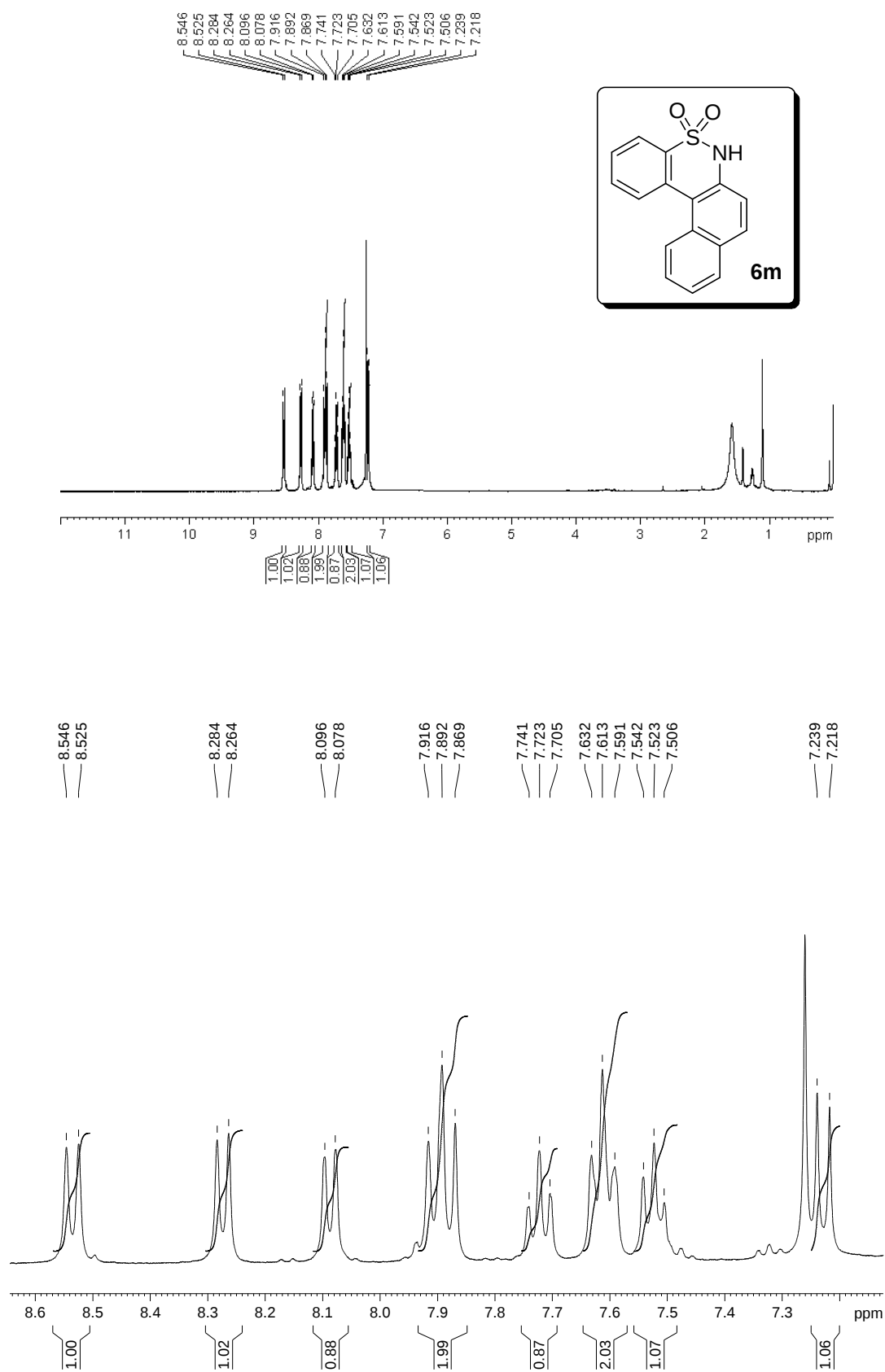
5H-Benzo[e]naphtho[1,2-c][1,2]thiazine 6,6-dioxide (6I)



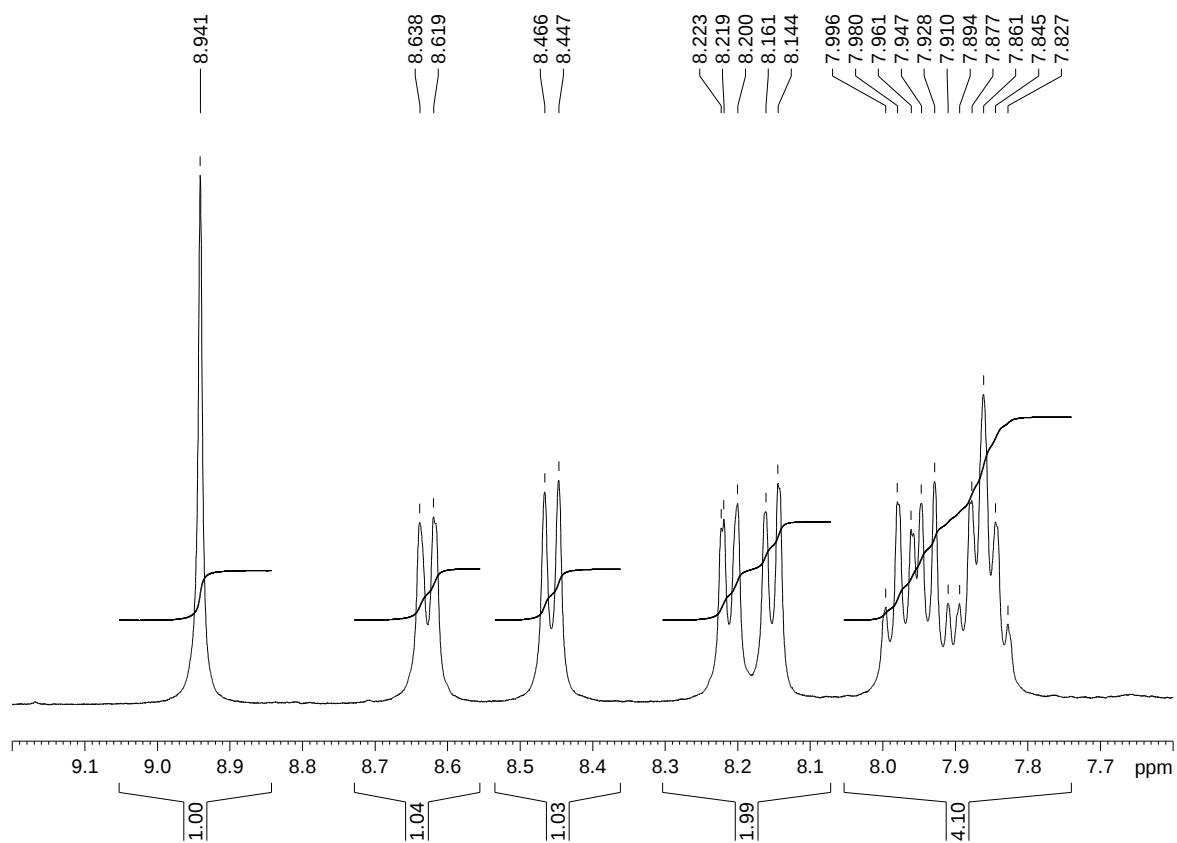
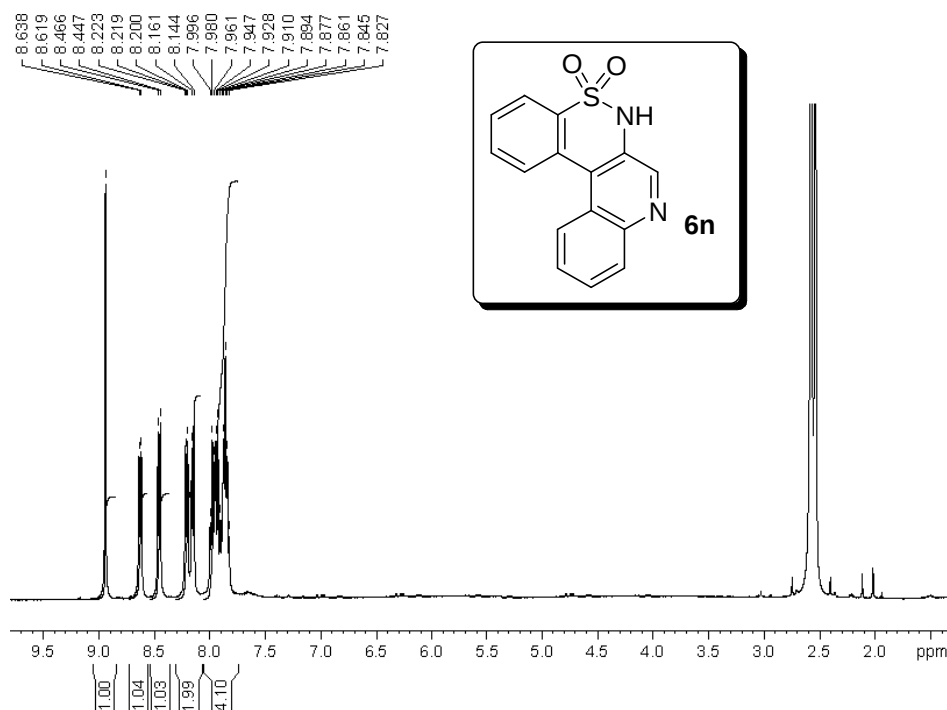


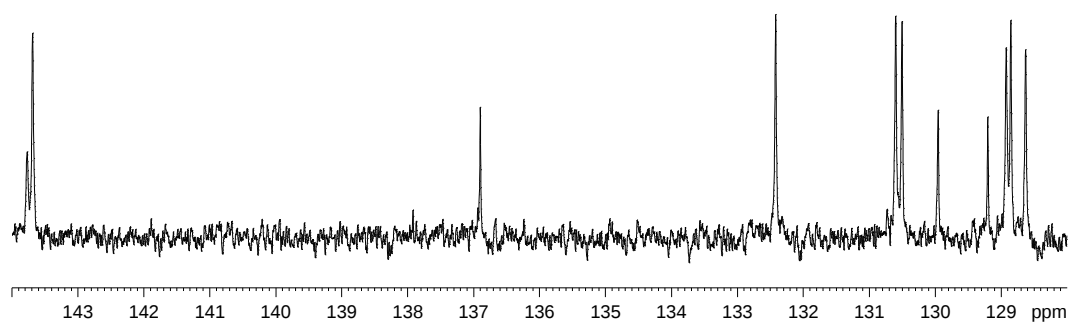
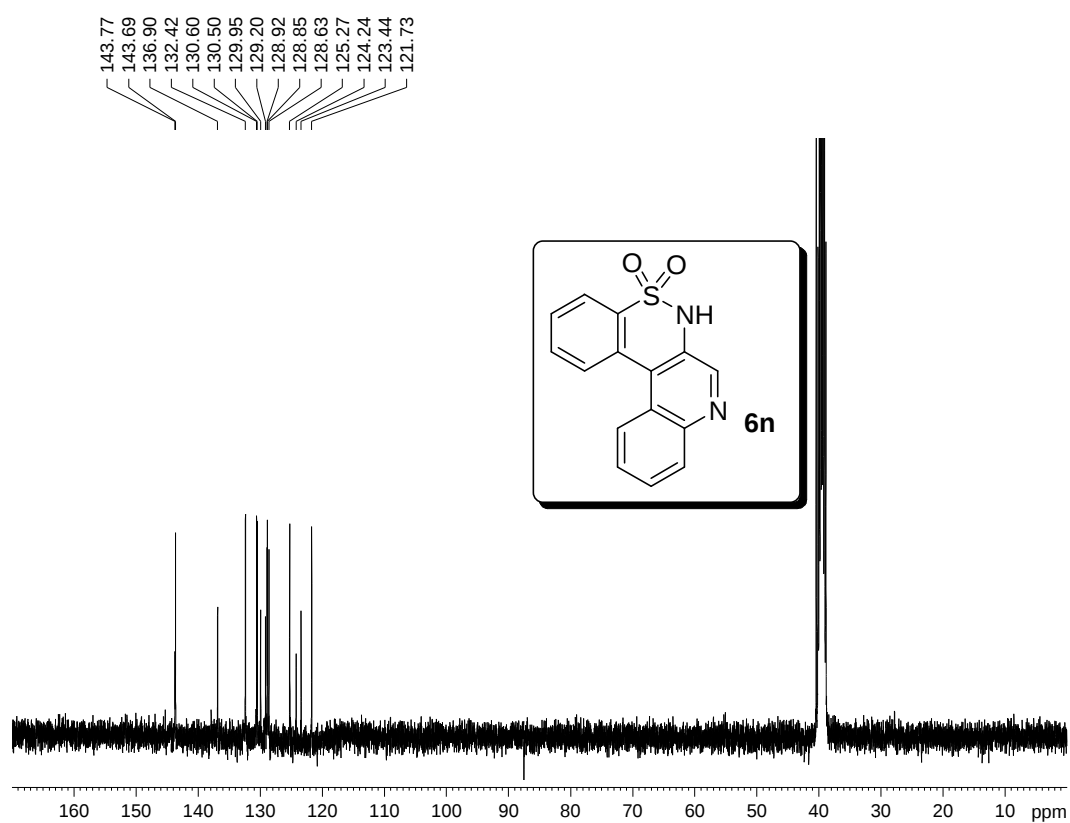


6*H*-Benzo[*e*]naphtho[2,1-*c*][1,2]thiazine 5,5-dioxide (**6m**)

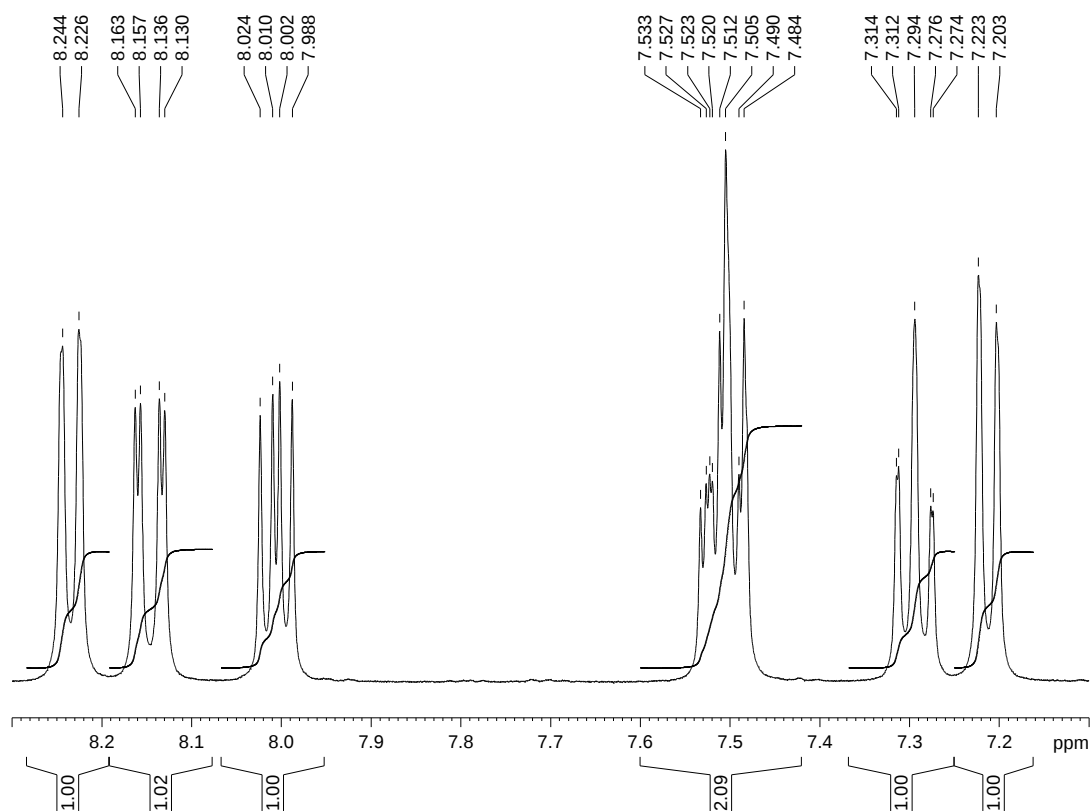
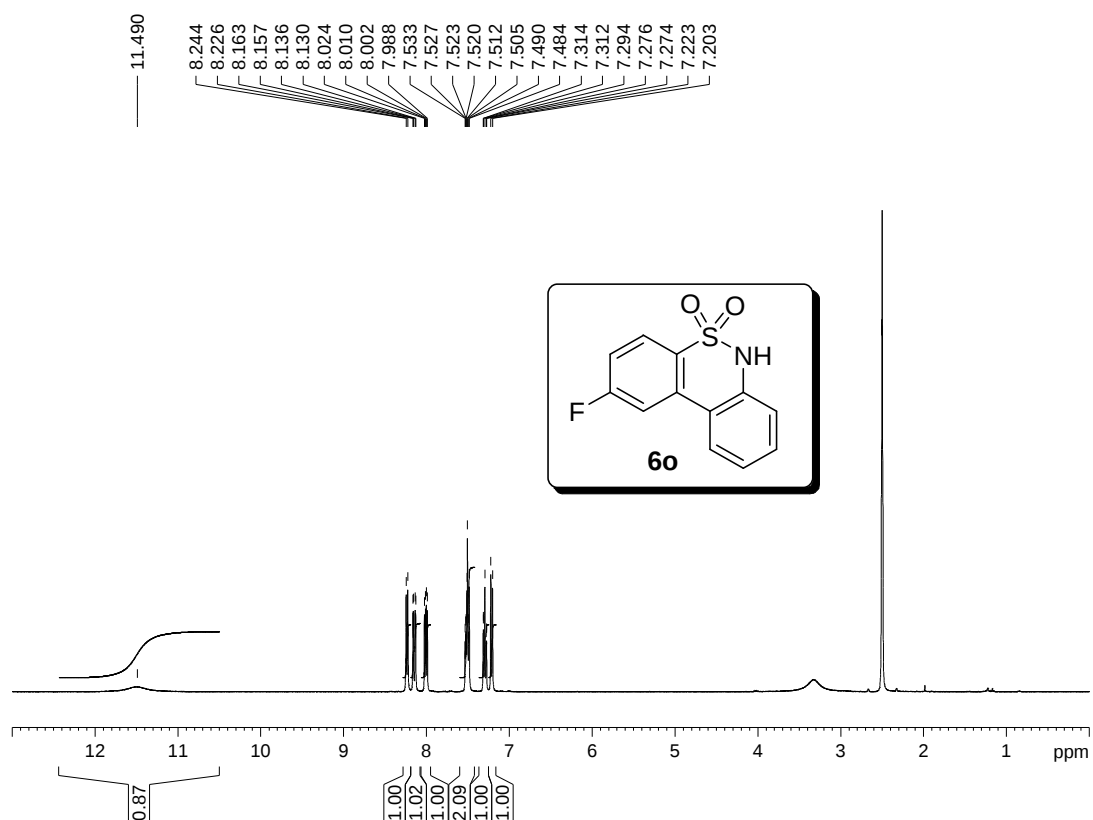


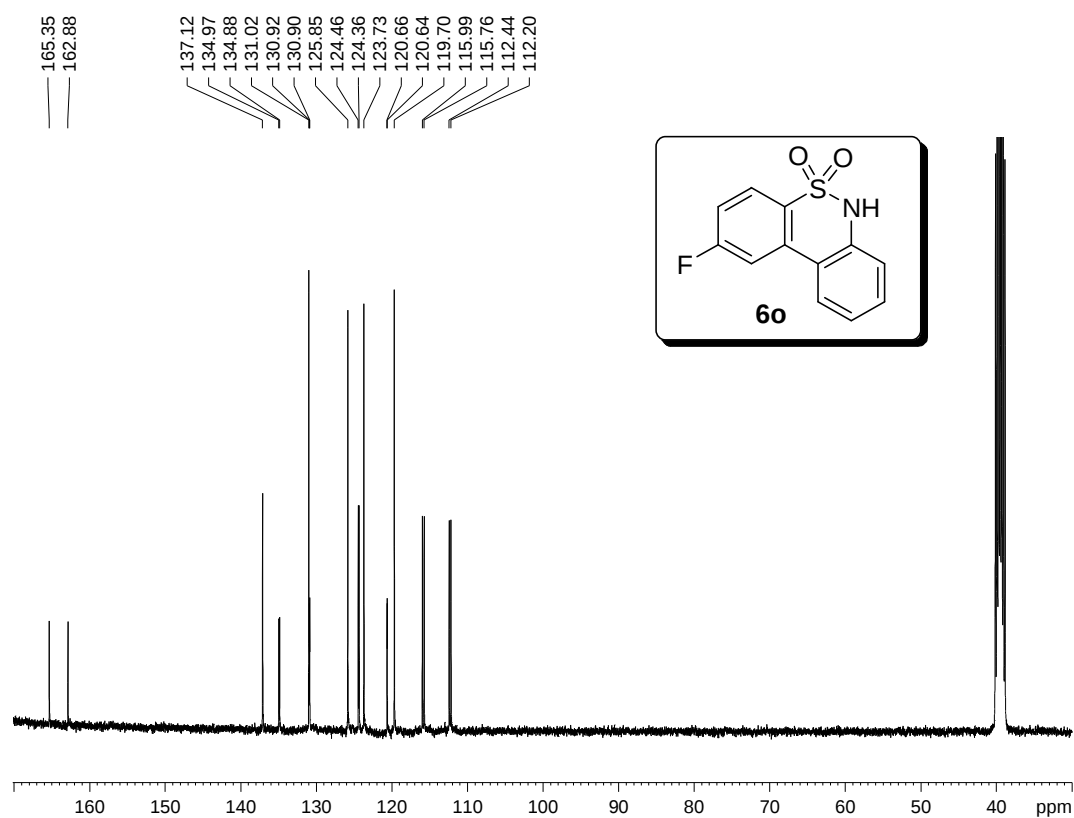
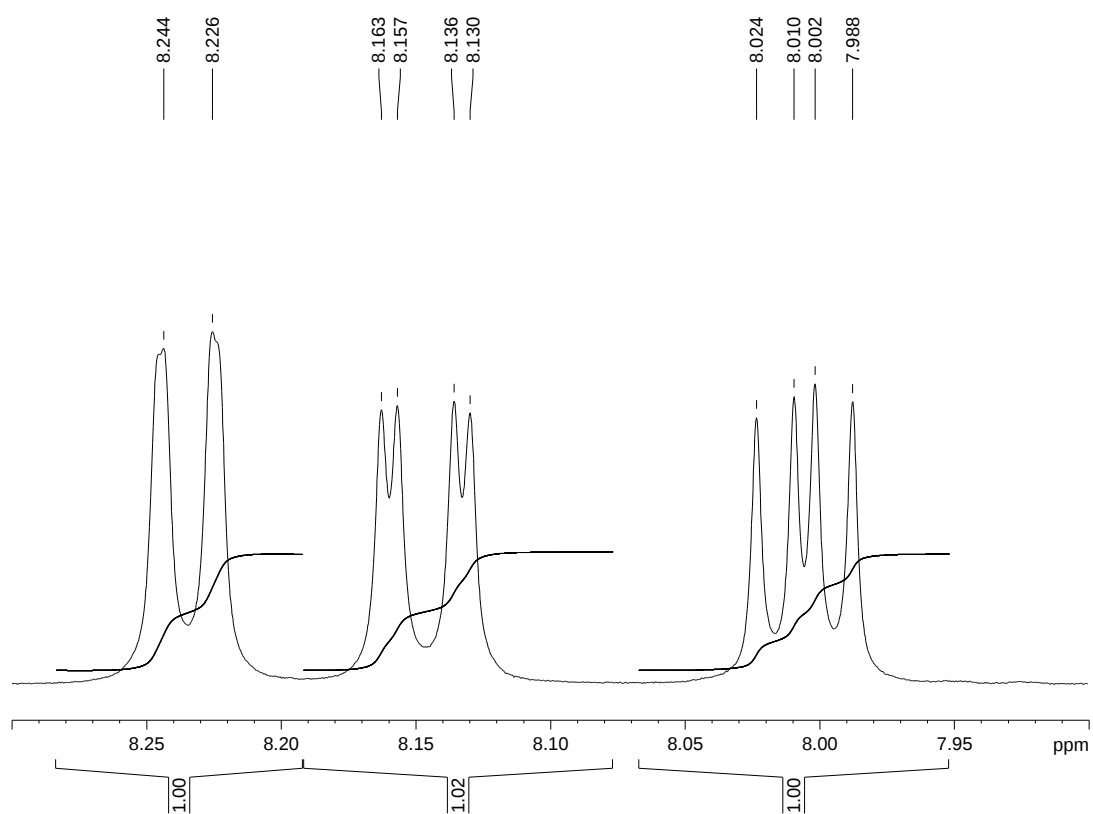
6H-Benzo[5,6][1,2]thiazino[3,4-c]quinoline 5,5-dioxide (**6n**)

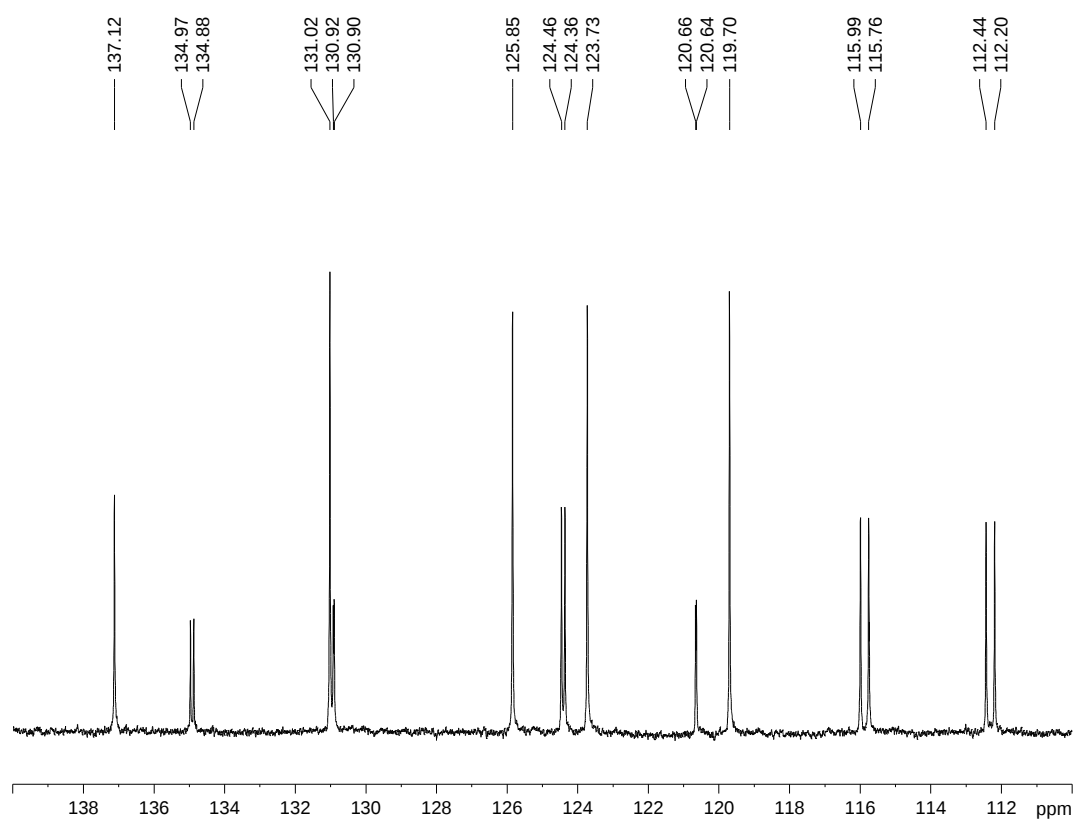




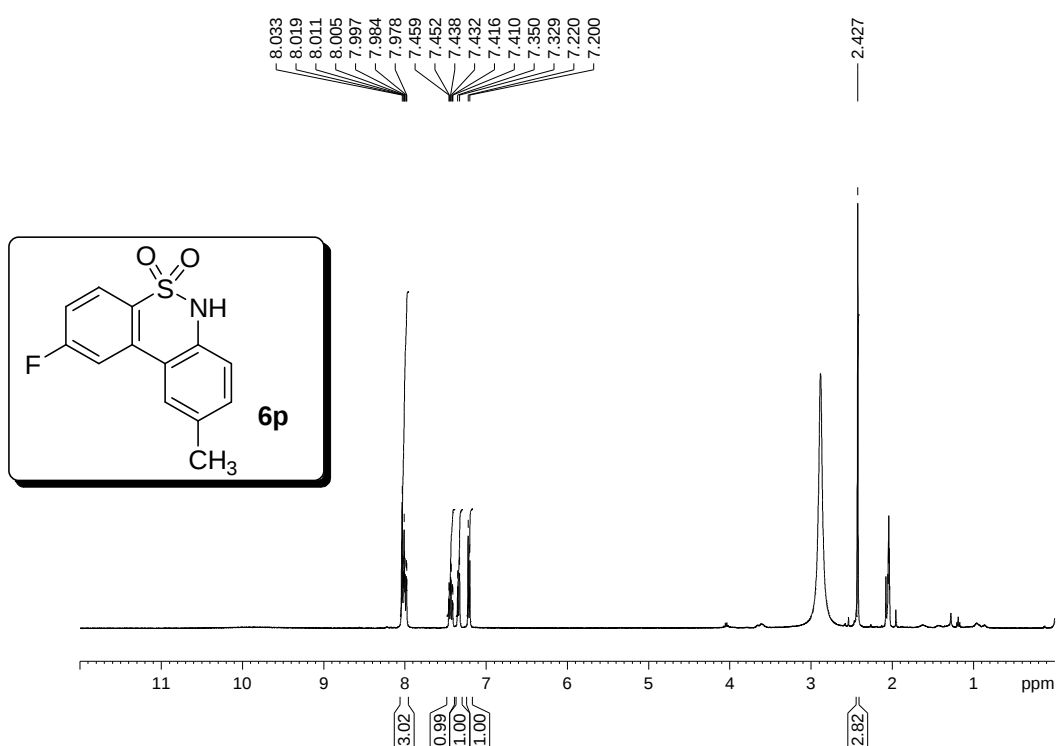
2-Fluoro-6H-dibenzo[*c,e*][1,2]thiazine 5,5-dioxide (**6o**)

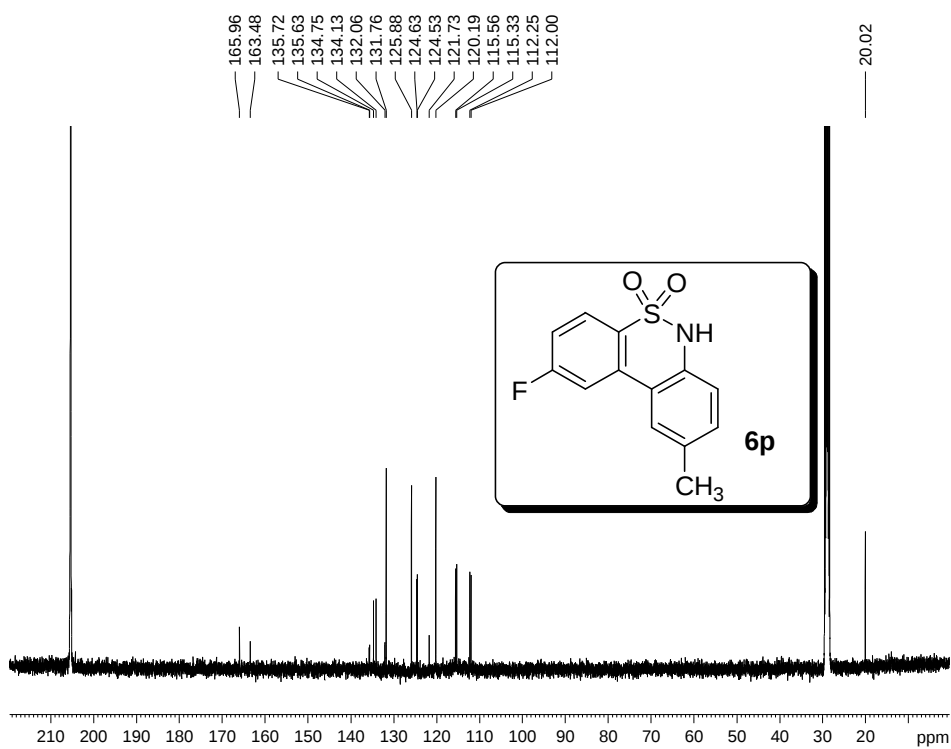
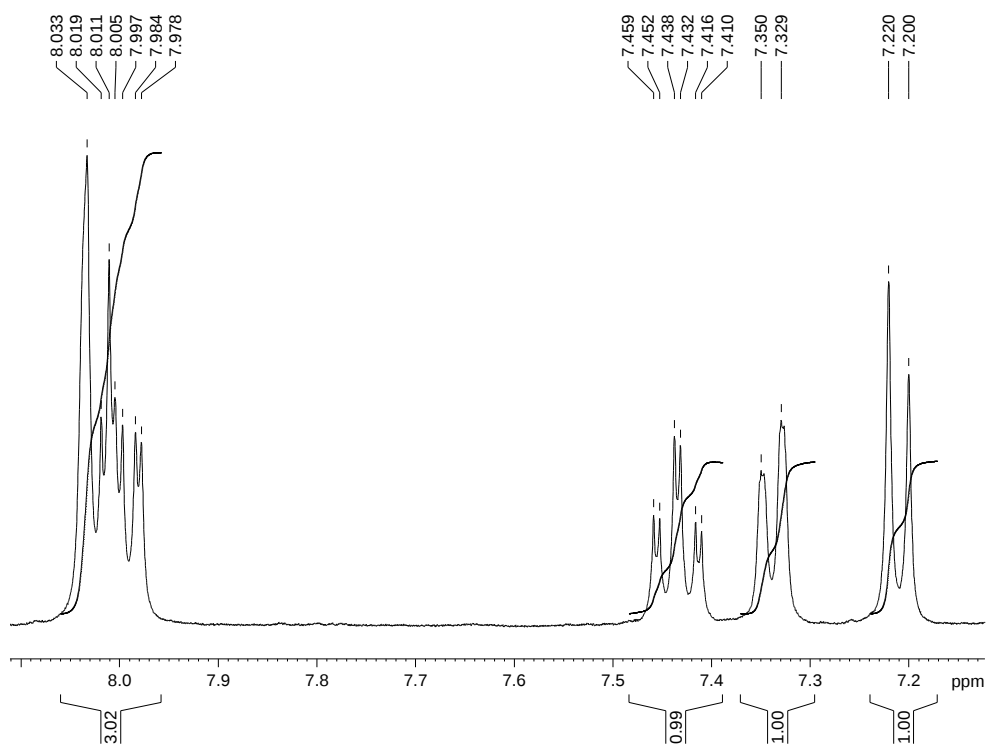


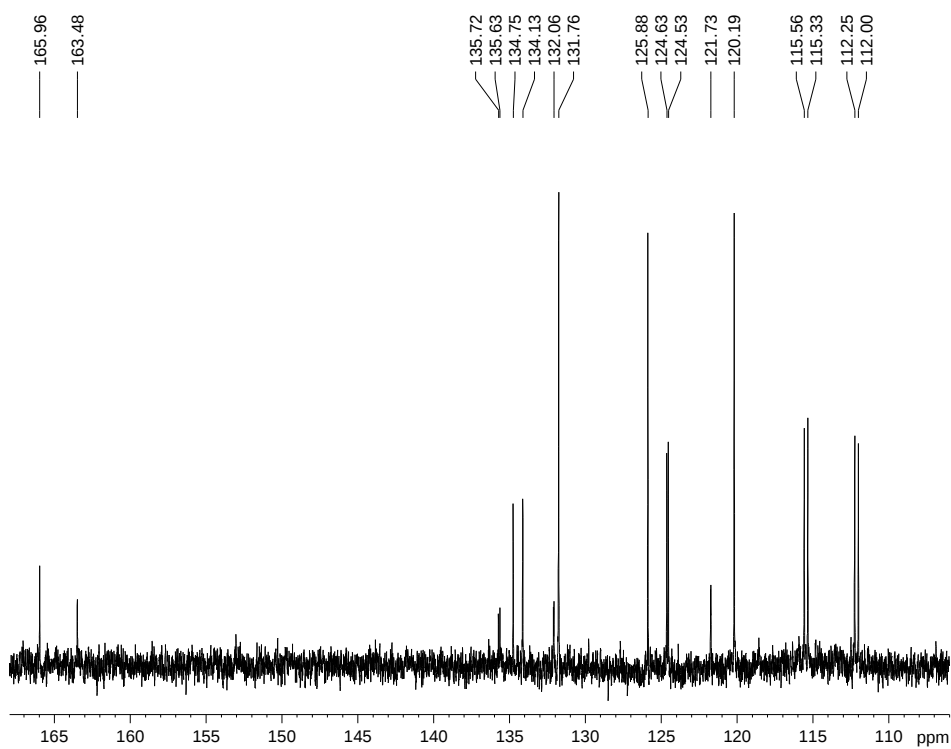




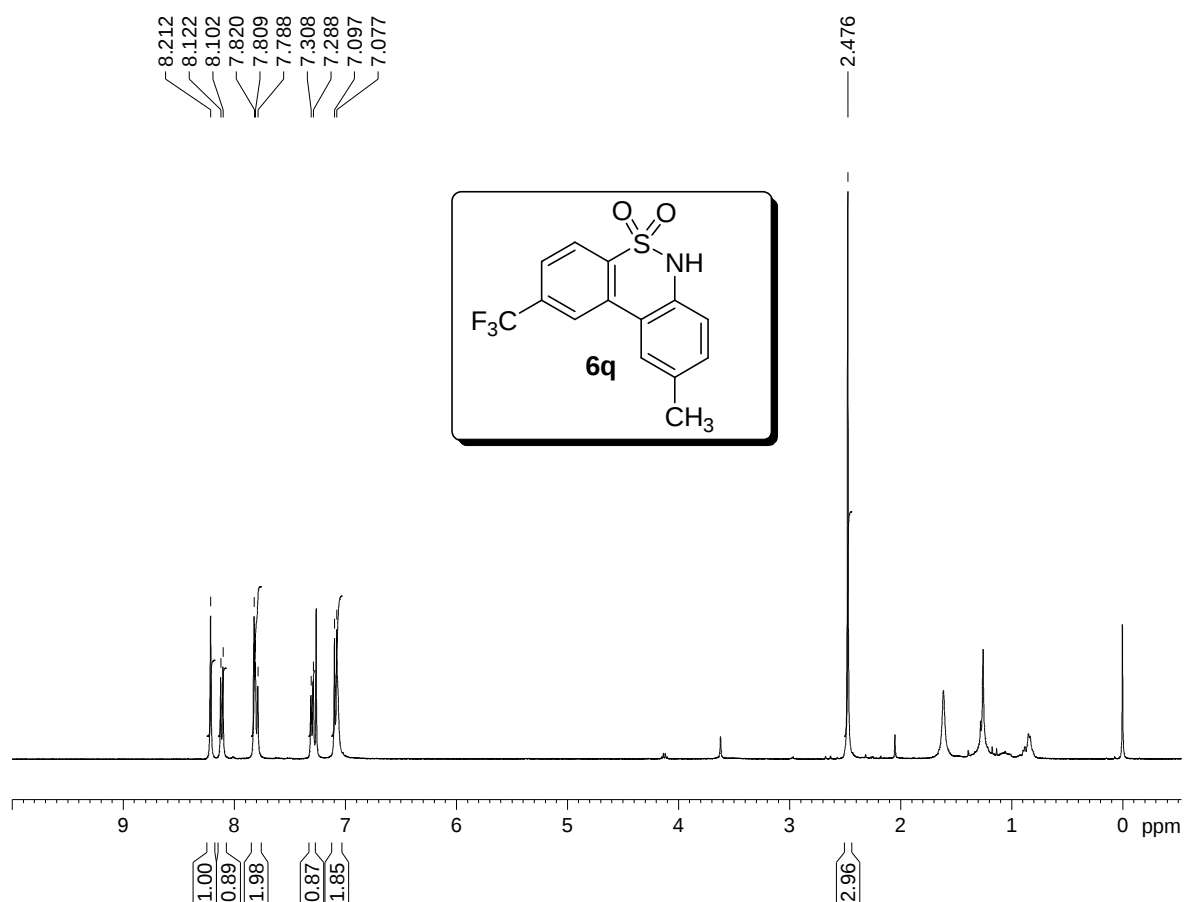
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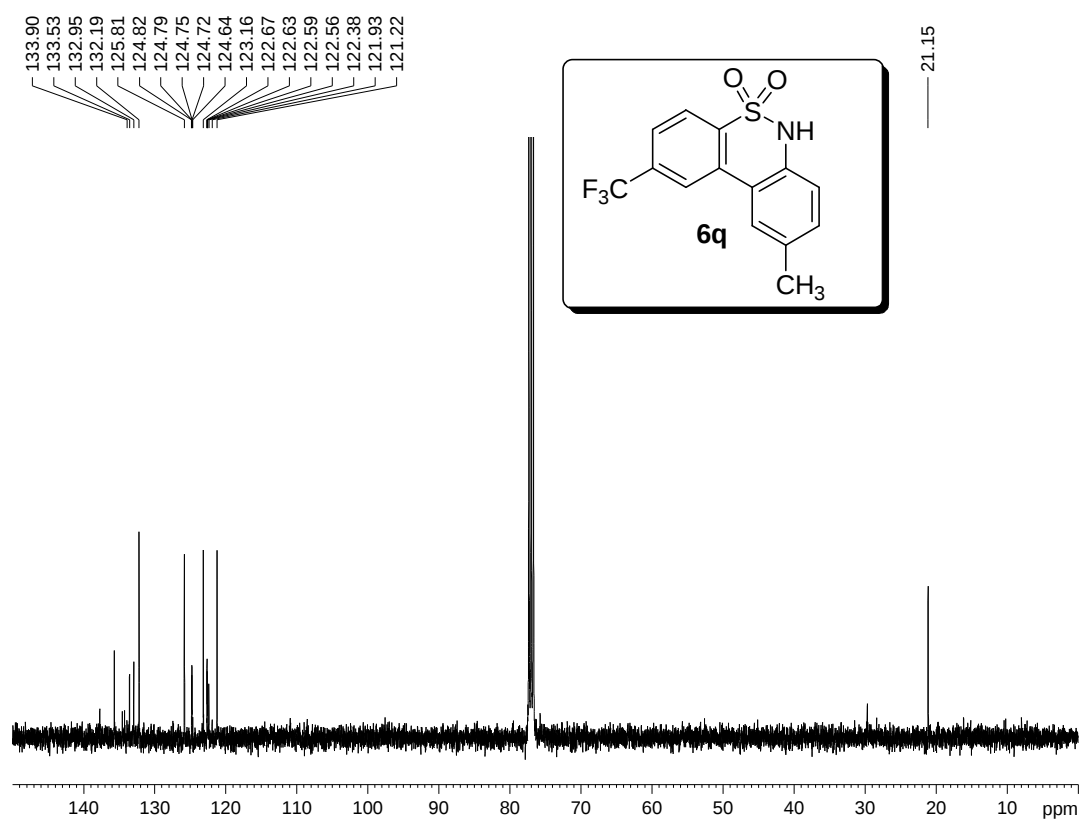
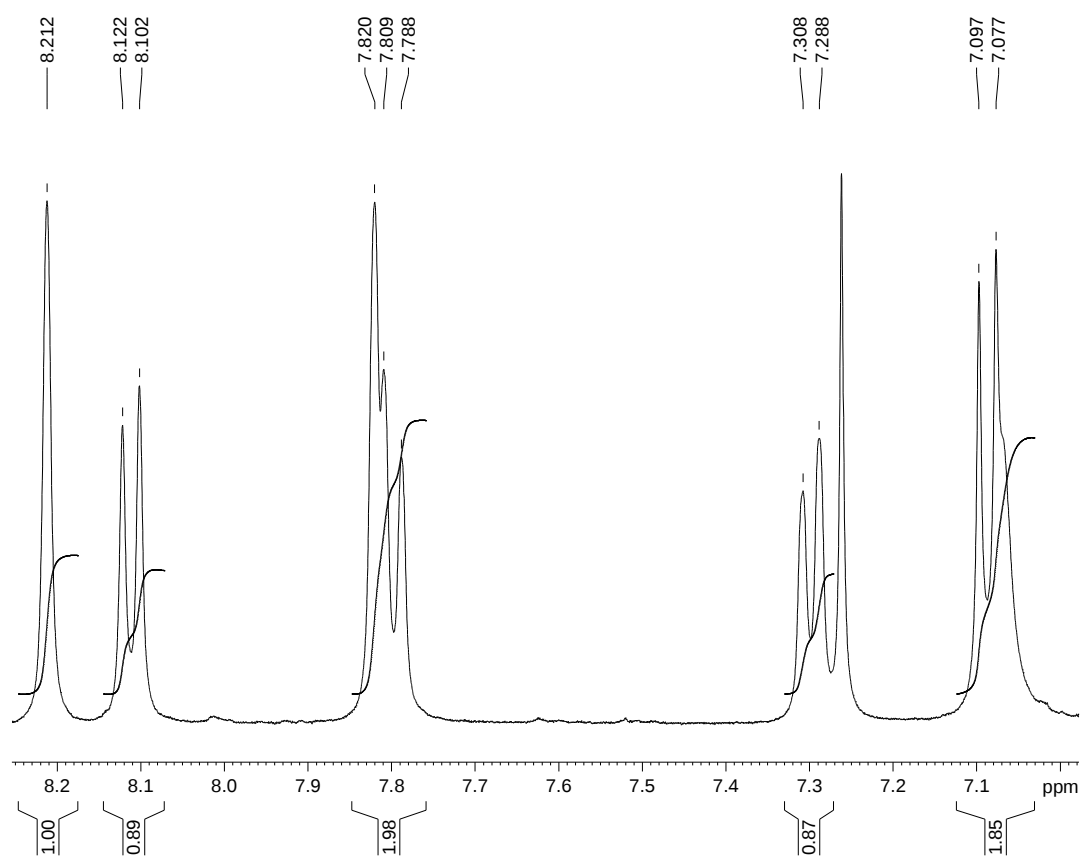


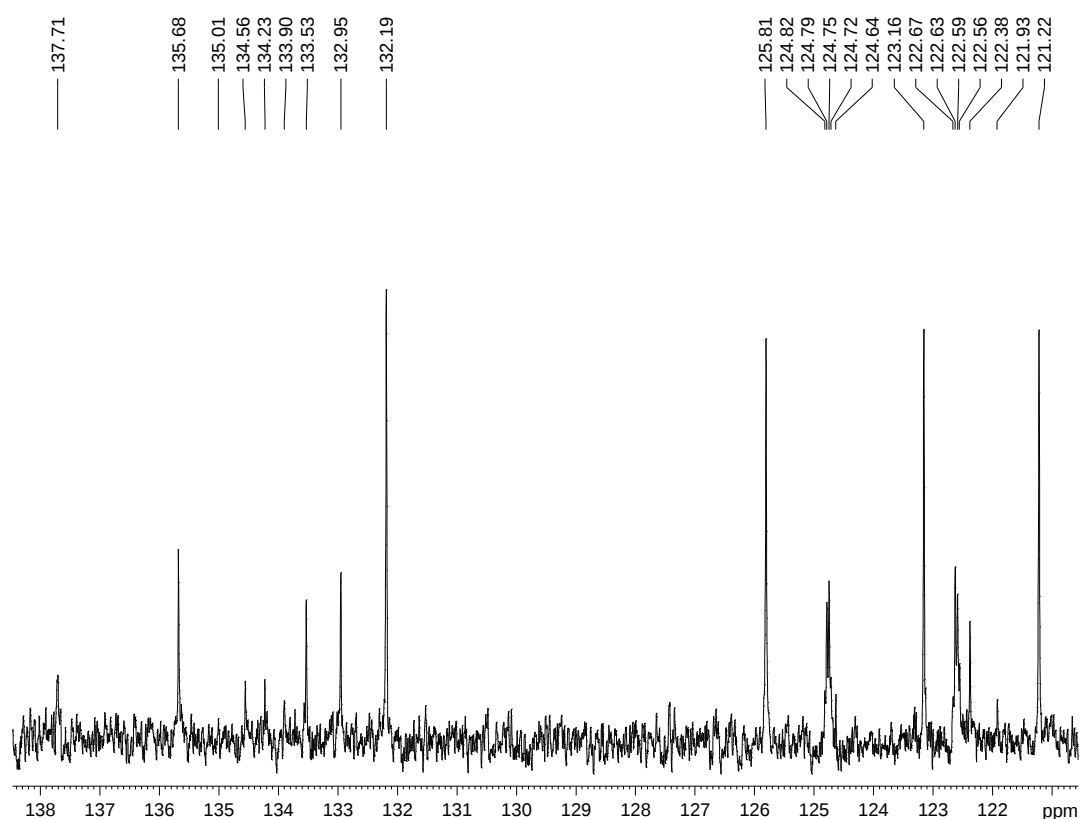




9-Methyl-2-(trifluoromethyl)-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (6q)



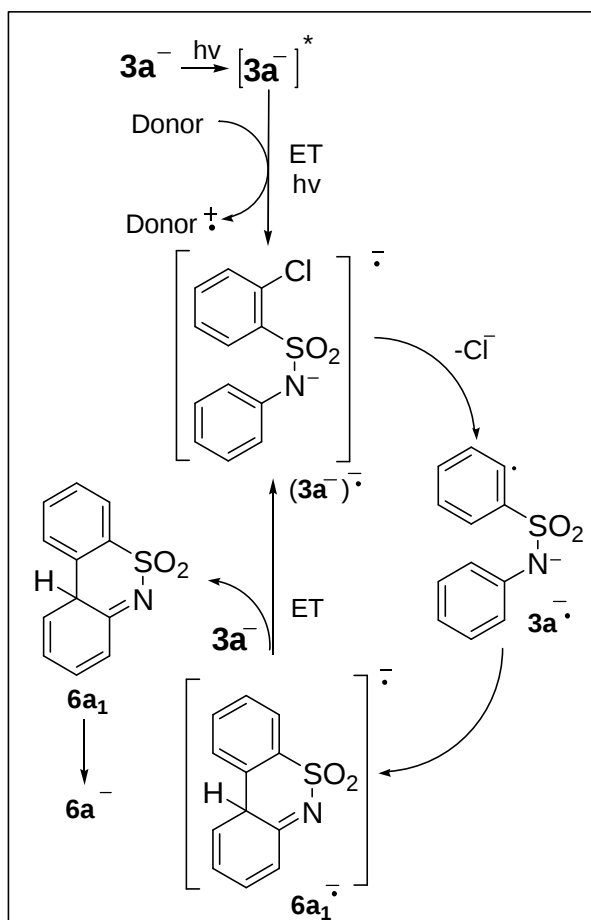




5. Theoretical Section

Full *Gaussian09* Reference

Gaussian 09, Revision B.01, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, T. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, O. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, and D. J. Fox, Gaussian, Inc., Wallingford CT, 2010.



$3a^-$ (anion 3a)

m06-2x/6-311+g(d) scrf=(solvent=dms0,pcm)

Sum of electronic and zero-point Energies=

-1526.065603 a.u.

Charge = -1 Multiplicity = 1

C	0.13270	0.53200	0.66039
C	1.16273	0.59909	1.59665
C	2.43800	0.15018	1.26237
C	2.68103	-0.36793	-0.00705
C	1.64620	-0.43174	-0.93677
C	0.36006	0.01443	-0.62025
H	0.96114	1.00080	2.58253
H	3.23360	0.20717	1.99717
H	3.67006	-0.72178	-0.27639
H	1.82002	-0.83536	-1.92585
Cl	-1.44611	1.11611	1.18056
S	-0.93377	-0.11655	-1.92400
O	-0.21921	-0.75646	-3.06597
O	-2.00845	-0.95622	-1.34035
N	-1.53620	1.31682	-2.21281
C	-0.80156	2.36423	-2.76767
C	0.52104	2.31646	-3.26510
C	-1.46734	3.61089	-2.83682
C	1.13393	3.45781	-3.78376
H	1.06700	1.38208	-3.26289

C	-0.85115	4.73933	-3.36347
H	-2.48449	3.66718	-2.46170
C	0.46207	4.67817	-3.84111
H	2.15276	3.38346	-4.15417
H	-1.39925	5.67671	-3.39812
H	0.94580	5.55958	-4.24906

(**3a⁻**)*(anion 3a S₁)

TD DFT m06-2x/6-311+g(d) scrf=(solvent=dmsol,pcm)

The total energy including the PCM contribution is -1526.090199236 a.u.

Charge = -1 Multiplicity = 1

C	0.13270	0.53200	0.66039
C	1.16273	0.59909	1.59665
C	2.43800	0.15018	1.26237
C	2.68103	-0.36793	-0.00705
C	1.64620	-0.43174	-0.93677
C	0.36006	0.01443	-0.62025
H	0.96114	1.00080	2.58253
H	3.23360	0.20717	1.99717
H	3.67006	-0.72178	-0.27639
H	1.82002	-0.83536	-1.92585
Cl	-1.44611	1.11611	1.18056
S	-0.93377	-0.11655	-1.92400
O	-0.21921	-0.75646	-3.06597
O	-2.00845	-0.95622	-1.34035
N	-1.53620	1.31682	-2.21281
C	-0.80156	2.36423	-2.76767
C	0.52104	2.31646	-3.26510
C	-1.46734	3.61089	-2.83682
C	1.13393	3.45781	-3.78376
H	1.06700	1.38208	-3.26289
C	-0.85115	4.73933	-3.36347
H	-2.48449	3.66718	-2.46170
C	0.46207	4.67817	-3.84111
H	2.15276	3.38346	-4.15417
H	-1.39925	5.67671	-3.39812
H	0.94580	5.55958	-4.24906

(**3a⁻**)^{•-} (dianion radical 3a)

m06-2x/6-311+g(d) scrf=(solvent=dmsol,pcm)

Sum of electronic and zero-point Energies= -1526.131042 a.u.

Charge = -2 Multiplicity = 2

C	0.05330	0.47402	0.58983
C	1.11829	0.81895	1.38929
C	2.45188	0.55895	0.98635
C	2.64764	-0.12197	-0.25365
C	1.58577	-0.47408	-1.05194
C	0.21985	-0.19083	-0.68301
H	0.91097	1.27615	2.36032
H	3.29056	0.85532	1.61589

H	3.66071	-0.38055	-0.57501
H	1.74988	-0.99296	-1.99801
Cl	-1.57448	0.72568	1.23416
S	-1.02839	-0.19458	-1.91238
O	-0.38919	-0.77315	-3.12791
O	-2.18122	-0.98426	-1.42809
N	-1.62172	1.28491	-2.16854
C	-0.82723	2.28405	-2.64792
C	0.54241	2.18871	-3.03261
C	-1.41627	3.57477	-2.78502
C	1.24470	3.30084	-3.49382
H	1.04495	1.22460	-2.97732
C	-0.70581	4.67158	-3.25234
H	-2.46668	3.67400	-2.50186
C	0.64290	4.55505	-3.61233
H	2.29476	3.17758	-3.77360
H	-1.21103	5.63743	-3.33654
H	1.20404	5.41627	-3.97759

TS C-Cl (Transition State for C-Cl fragmentation)

m06-2x/6-311+g(d) scrf=(solvent=dmsol,pcm)

Sum of electronic and zero-point Energies= -1526.131649 a.u.

Charge = -2 Multiplicity = 2

C	0.01172	0.79536	0.31345
C	1.12292	1.21356	1.05771
C	2.39947	0.78722	0.74248
C	2.57969	-0.18349	-0.29219
C	1.49143	-0.62757	-1.00805
C	0.18059	-0.16102	-0.74630
H	0.96735	1.89297	1.89229
H	3.25423	1.19002	1.27528
H	3.56800	-0.57479	-0.50837
H	1.63114	-1.34156	-1.81348
Cl	-1.58170	0.85044	1.23399
S	-1.03529	-0.30755	-2.02544
O	-0.34690	-0.95090	-3.18343
O	-2.17647	-1.13043	-1.54859
N	-1.64765	1.13493	-2.34617
C	-0.80868	2.18611	-2.66530
C	0.56242	2.10899	-3.00901
C	-1.38697	3.47736	-2.67118
C	1.29451	3.25536	-3.30621
H	1.05362	1.14508	-3.04889
C	-0.65076	4.61121	-2.97988
H	-2.43869	3.56137	-2.41563
C	0.70631	4.51740	-3.29688
H	2.34571	3.15288	-3.55980
H	-1.13977	5.58068	-2.96809
H	1.28435	5.40286	-3.53598

3a⁻ (radical anion 3a)

m06-2x/6-311+g(d) scrf=(solvent=dms0,pcm)

Sum of electronic and zero-point Energies= -1065.787758 a.u.

Charge = -1 Multiplicity = 2

C	2.07937	0.74349	-1.36146
C	2.82856	1.03541	-0.24673
C	2.50976	0.34090	0.92503
C	1.47580	-0.59553	0.93015
C	0.74544	-0.85549	-0.22629
C	1.05823	-0.16934	-1.40302
H	3.62938	1.76602	-0.26410
H	3.07156	0.53447	1.83215
H	1.23975	-1.13104	1.84227
H	-0.05401	-1.58801	-0.23337
S	0.08103	-0.42870	-2.89884
O	1.05652	-0.40875	-4.00876
O	-0.59615	-1.72001	-2.70097
N	-1.02085	0.69599	-2.93692
C	-0.69249	2.04060	-3.08705
C	0.55704	2.57578	-3.46532
C	-1.73581	2.95882	-2.84064
C	0.74174	3.95376	-3.56621
H	1.37974	1.91459	-3.70487
C	-1.54335	4.32675	-2.95308
H	-2.70353	2.56172	-2.55281
C	-0.29671	4.84290	-3.31323
H	1.71718	4.33043	-3.85751
H	-2.37225	4.99834	-2.75391
H	-0.14372	5.91253	-3.39841

TS C-C (Transition State for C-C coupling)

m06-2x/6-311+g(d) scrf=(solvent=dms0,pcm)

Sum of electronic and zero-point Energies= -1065.773184 a.u.

Charge = -1 Multiplicity = 2

C	0.46176	1.32920	-0.97103
C	1.03560	1.79248	0.19743
C	1.73252	0.88435	0.99749
C	1.83532	-0.45458	0.61619
C	1.25399	-0.89646	-0.56825
C	0.56289	0.01629	-1.36938
H	0.94586	2.83471	0.49116
H	2.19312	1.22023	1.92087
H	2.37200	-1.15583	1.24492
H	1.33412	-1.93582	-0.87196
S	-0.13486	-0.52346	-2.95022
O	1.04467	-0.69260	-3.83040
O	-0.86226	-1.78279	-2.70743
N	-1.14990	0.59906	-3.37267
C	-0.59926	1.89212	-3.33871

C	0.23746	2.40986	-4.32647
C	-0.83864	2.67311	-2.17159
C	0.79744	3.68150	-4.19047
H	0.45731	1.79964	-5.19557
C	-0.33185	3.98527	-2.08227
H	-1.63653	2.35679	-1.50872
C	0.51173	4.46893	-3.07112
H	1.44850	4.06610	-4.96807
H	-0.58797	4.60082	-1.22668
H	0.92918	5.46670	-2.99095

6a₁⁻ (radical anion 6a₁)

m06-2x/6-311+g(d) scrf=(solvent=dms0,pcm)

Sum of electronic and zero-point Energies= -1065.825920 a.u.

Charge = -1 Multiplicity = 2

C	0.46526	1.32619	-1.10793
C	1.13041	1.84937	-0.00271
C	1.79720	1.00468	0.88463
C	1.83032	-0.37025	0.67082
C	1.19178	-0.91075	-0.44228
C	0.51173	-0.05746	-1.29848
H	1.14045	2.91988	0.16556
H	2.30332	1.42911	1.74454
H	2.35930	-1.01791	1.36021
H	1.22204	-1.97433	-0.64994
S	-0.28605	-0.65203	-2.77583
O	0.26960	-1.97010	-3.11011
O	-1.74166	-0.65768	-2.51386
N	0.15872	0.40298	-3.88726
C	0.12733	1.73065	-3.54832
C	0.49510	2.68509	-4.48350
C	-0.29318	2.15913	-2.14137
C	0.51659	4.05630	-4.19539
H	0.77438	2.33109	-5.47130
C	-0.21441	3.64279	-1.92011
H	-1.34709	1.84287	-2.02720
C	0.15222	4.51039	-2.89890
H	0.81174	4.76667	-4.95816
H	-0.52477	4.01337	-0.94853
H	0.16187	5.57561	-2.68848

6a₁ (6a₁ neutral)

m06-2x/6-311+g(d) scrf=(solvent=dms0,pcm)

Sum of electronic and zero-point Energies= -1065.698372 a.u.

Charge = 0 Multiplicity = 1

C	0.48002	1.32431	-1.09786
C	1.12577	1.84518	0.01850
C	1.77834	0.99315	0.90799
C	1.81352	-0.38074	0.69108

C	1.19509	-0.91786	-0.43278
C	0.53155	-0.05523	-1.29205
H	1.14148	2.91307	0.19682
H	2.27650	1.41477	1.77312
H	2.33081	-1.03092	1.38588
H	1.22512	-1.98044	-0.64309
S	-0.27360	-0.66672	-2.74479
O	0.31212	-1.93938	-3.12783
O	-1.72231	-0.61073	-2.55565
N	0.15169	0.45886	-3.89978
C	0.12825	1.70433	-3.54205
C	0.46655	2.70208	-4.53411
C	-0.24766	2.16489	-2.13948
C	0.48190	4.01174	-4.21756
H	0.72106	2.34721	-5.52524
C	-0.17671	3.64174	-1.91380
H	-1.31592	1.89576	-2.05017
C	0.14595	4.49742	-2.88761
H	0.74774	4.73856	-4.97740
H	-0.45852	3.99965	-0.93004
H	0.15344	5.56494	-2.70600

6a⁻ (anion 6a)

m06-2x/6-311+g(d) scrf=(solvent=dms0,pcm)

Sum of electronic and zero-point Energies=

-1065.299912 a.u.

Charge = -1 Multiplicity = 1

C	3.83017	-0.13449	-0.04985
C	2.82625	-1.05561	-0.27749
C	1.46348	-0.70734	-0.16826
C	1.14051	0.64892	0.11594
C	2.18135	1.55668	0.37123
C	3.51153	1.18330	0.29910
C	-0.26213	1.09390	0.05351
C	-1.30743	0.15957	0.06670
C	-2.64470	0.52142	-0.06653
H	-3.40870	-0.24817	-0.06105
C	-2.97483	1.86086	-0.20360
C	-1.95687	2.81743	-0.22079
C	-0.62740	2.44220	-0.10663
H	4.86792	-0.44224	-0.12428
H	3.06209	-2.08509	-0.52504
H	1.94059	2.57925	0.64044
H	4.29339	1.90403	0.50774
H	-4.01118	2.16093	-0.30434
H	-2.20366	3.86598	-0.34402
H	0.13800	3.20594	-0.17272
S	-0.86255	-1.53760	0.31947
O	-1.87031	-2.39268	-0.32775
O	-0.77991	-1.72592	1.78450
N	0.53109	-1.70274	-0.41007

6a (6a neutral)

m06-2x/6-311+g(d) scrf=(solvent=dms,pcm)

Sum of electronic and zero-point Energies=

-1065.743453 a.u.

Charge = 0 Multiplicity = 1

C	3.83143	-0.15555	0.01495
C	2.82845	-1.06679	-0.28025
C	1.49580	-0.66131	-0.26170
C	1.14777	0.66688	0.04173
C	2.17858	1.55593	0.36808
C	3.50613	1.15705	0.35284
C	-0.26451	1.09754	-0.00794
C	-1.31199	0.16906	0.06567
C	-2.65366	0.51966	-0.01094
H	-3.41959	-0.24569	0.04222
C	-2.98403	1.85905	-0.16091
C	-1.96907	2.81077	-0.24410
C	-0.63344	2.43756	-0.17892
H	4.86632	-0.47687	0.00079
H	3.07122	-2.09611	-0.52069
H	1.93607	2.57373	0.64946
H	4.28542	1.86439	0.60901
H	-4.02337	2.15705	-0.22368
H	-2.22042	3.85639	-0.37766
H	0.13150	3.19646	-0.28812
S	-0.90699	-1.52738	0.33655
O	-1.87884	-2.41247	-0.28181
O	-0.60805	-1.73119	1.74681
N	0.47197	-1.57962	-0.59518
H	0.76452	-2.52967	-0.80756