SUPPORTING INFORMATION

PAni synthesis on HOPG and Modified HOPG:

An aqueous solution of 5.4 mM sodium dodecyl sulfate (SDS, 98% Aldrich Chemical Company, recrystallized once from 95% ethanol, cmc 8.1×10^{-3} M), and 5.4mM aniline (99% Aldrich, distilled under reduced pressure) was prepared at pH 2 using HCl (ACS plus grade, Fisher Scientific). 1-dodecanol (98% Avocado Research Chemicals) was used as received. All solutions were prepared with 18.2 M Ω -cm⁻¹ nanopure water in polypropylene (PP) vessels, stirred at 27 ± 1°C for 1 h and then used immediately. Although SDS was recrystallized once from 95% ethyl alcohol, almost certainly dodecanol remains as an impurity.¹ As low pH will also contribute to the hydrolysis of SDS, solutions were prepared and used in a timely manner to ensure reproducibility.

Advanced Ceramics, ZYB grade HOPG (12 x 12 mm) was cut into four pieces of approximately 5 x 5 mm on a band saw (ZYB grade HOPG was used to give a surface with fewer grain boundaries compared to ZYH HOPG. Thin sections of HOPG were cleaved with a razor blade exposing a fresh surface used for reaction. HOPG thin sections were not re-used for subsequent reactions because of contamination of underlying layers. Chemically modified HOPG was prepared by the treatment of a freshly cleaved graphite monochromator ZYB (5 mm x 5 mm Advanced Ceramics) with 5 mL of a 5.4 mM ammonium persulfate (APS, 98% Aldrich Chemical Company) at a pH of 2 adjusted with HCl for 3 hours at $22 \pm 1^{\circ}$ C in a polyethylene (PE) vial. The block was then briefly rinsed with nanopure water and dried for at least 12 hours in a dessicator prior to use.

Graphite substrates were contacted with 5 mL of the SDS/aniline solution for 5 h at 27 \pm 1°C in PE vials. A final monomer/oxidant molar ratio of 1:1 was obtained by the addition of 100 µL of an APS solution with stirring for approximately 5 minutes.ⁱⁱ Following polymerization, the substrate was rinsed thoroughly with pH 2 water (adjusted with HCl) and dried at room temperature in a dessicator for at least 12 h prior to imaging.

PPy synthesis on HOPG:

Sodium 1-octanesulfonate (99% Aldrich Chemical Company, cmc 1.6x10⁻¹M), sodium 1-decanesulfonate (98% Avocado Research Chemicals Ltd., cmc 4.3x10⁻²M) and sodium 1-dodecanesulfonate (99% Avocado Research Chemicals Ltd., cmc 1.2×10^{-2}) were all used without further purification as were 1-octanol and 1-decanol (99% Avocado Research Chemicals Ltd.). Pyrrole (98% Aldrich Chemical Company) was filtered through a basic alumina column prior to use. 5.4/2.7 mM surfactant/pyrrole solutions were used in all cases prepared with 18.2 M Ω -cm⁻¹nanopure water in PP vessels. A fresh surface of HOPG was rendered by cleaving a thin section of graphite monochromator ZYH (5 mm x 5 mm Advanced Ceramics) as described previously. HOPG substrates were then contacted with 5 mL of the surfactant/pyrrole solution for 2 h at $22 \pm 1^{\circ}$ C in PE vials. A final monomer/oxidant molar ratio in of 1:1 was obtained by addition of 100 µl of an APS solution with stirring for approximately 5 minutes," and oxidation carried out at $22 \pm 1^{\circ}$ C for 2 h. Following polymerization, the substrate was rinsed thoroughly with nanopure water and dried in a dessicator at room temperature for at least 12 h prior to imaging.

AFM Imaging

AFM images were collected by a Digital Instruments NanoScope III microscope, operated in tapping mode. MikroMasch Ultrasharp silicon nitride cantilevers were used with a backside aluminum coating and typical resonant frequencies of 325 kHz and force constants of 40 N/m. All scanning was performed under ambient conditions with typical relative humidity levels of around 30%. Images shown in Fig. 2 were captured at a scan angle of 0° with minimal engagement forces and relatively high scan rates (~15 µm/s). The wires appeared soft and easily distorted by lower setpoints. All other images were collected at a scan angle of 0° and tip velocities of ~3 µm/s. No filtering of images was performed. Calibration of the AFM was carried out using grids with a pitch of 10 µm and depth of 180nm.

XPS Experiments

XPS data were recorded on a Physical Electronics PHI 5800 ESCA System with a background pressure of approximately 2.0 X 10^{-9} Torr. The electron takeoff angle was 45° with respect to the sample surface. An 800 μ m spot size and 23 eV pass energy were typically used for the analysis. Binding energies were corrected by reference to the C_{1s} line at 284.8 eV for hydrocarbon. A nonlinear Shirley-type background was used for the area analysis of each peak. XPS data was fitted with Voigt Amplitude curves using PeakFit[®] software from Systat Software.

REFERENCES

i Turner, S. F.; Clarke, S. M.; Rennie, A.R.; Thirtle, P.N.; Cooke, D. J.; Li, Z. X.;Thomas, R. K. Langmuir, **1999**, 15, 1017.ii Gentle stirring to ensure that color changes occurring during synthesis were

homogeneous.