Supporting Information

Stereoselective Synthesis of P-Chiral Phosphorus Compounds From N-t-Butoxycarbonyl Amino Acids

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General. All manipulations of air-sensitive compounds were carried out in an inert atmosphere glovebox under recirculating nitrogen. ^{1}H , ^{13}C , and ^{31}P NMR spectra were recorded at 400, 100, and 160 MHz, respectively; chemical shifts are reported relative to TMS for ^{1}H and ^{13}C NMR, and to external 85% $H_{3}PO_{4}$ at 0 ppm (positive values downfield) for ^{31}P NMR. Infrared spectra were obtained in 0.1 mm NaCl solution cells on a computer-controlled FT-IR spectrometer. Optical rotations were measured at room temperature in a 1.0 dm cell.

All solvents were treated under nitrogen. Benzene and tetrahydrofuran were distilled from sodium benzophenone ketyl. Hexane was purified by washing successively with 5% nitric acid in sulfuric acid, water, sodium bicarbonate solution, and water, and then dried over calcium chloride and distilled from *n*-butyllithium in hexane. CDCl₃ was vacuum-transferred from phosphorus pentoxide. Silica gel (200-400 mesh) was dried for several hours under vacuum while heating with a heat gun and was transferred under vacuum into the glovebox. PhPCl₂ was degassed prior to use, NEt₃ was distilled from CaH₂, and *t*-Boc amino acids were used as received.

(2R, 4S)-3-tert-butyloxycarbonyl-5-oxo-2-phenyl-1,3,2-oxazaphospholidine (7). In the glovebox, t-Boc glycine (2.00g, 11.2 mmol) was dissolved in 3 mL of THF, and the solvent was removed under vacuum in order to remove any water that might be present. It was then redissolved in 40 mL of THF, NEt₃ (2.36 g, 24.4 mmol) was added in ~1 mL of THF, and then a solution of 2.18 g (12.2 mmol) of PhPCl₂ in 10 mL of THF was added dropwise over 5 min to the stirred solution, immediately giving a white precipitate. An aliquot taken for ³¹P NMR showed complete conversion to product, but stirring was still allowed to continue overnight. The mixture was filtered through a short column of silica gel (filling 30 mL of a 60 mL fritted funnel), washed with THF, and the solvent removed to give 2.12 g crude product (66% yield) that was close to spectroscopically pure. This was crystallized by dissolving in 2 mL of CH₂Cl₂, cooling at -30 °C, layering on an equal volume of hexanes, and cooling overnight to give 0.78 g of fine white crystals; a second crop of 0.55 g was obtained in the same way (41% total yield). IR (CHCl₃): 1797 (s), 1713 (s) cm⁻¹; ¹H NMR (CDCl₃): δ 7.50 (m, 5H), 4.23 (br dd, ³ J_{PH} = 4.9 Hz, ${}^{2}J_{HH} = 17.8$ Hz, 1H), 3.95 (dd, ${}^{3}J_{PH} = 1.4$ Hz, ${}^{2}J_{HH} = 17.8$ Hz, 1H) 1.45 (s, 9H). ${}^{31}P$ NMR (CDCl₃): 135.2 ppm. 13 C NMR (CDCl₃): 170.8 (br s), 152.4 (br s), 139.37 (d, J_{PC} = 43.3 Hz), 132.2, 129.2 (d, J_{PC} = 23.8 Hz), 129.0 (d, J_{PC} = 6.3 Hz), 83.1, 42.8 (br s), 28.2 ppm. Anal. Calcd for C₁₃H₁₆NO₄P: C, 55.52; H, 5.73; N, 4.98. Found: C, 55.79; H, 5.91; N, 4.78.

(2R, 4S)-3-tert-butyloxycarbonyl-4-methyl-5-oxo-2-phenyl-1,3,2-oxazaphospholidine (8). In the glovebox, t-Boc alanine (2.00g, 10.6 mmol) was dissolved in 3 mL of THF, and the solvent was removed under vacuum in order to remove any water that might be present. It was then redissolved in 40 mL of THF, NEt₃ (2.11 g, 20.8 mmol) was added in ~1 mL of THF, and then a solution of 1.89 g (10.6 mmol) of PhPCl₂ in 10 mL of THF was added dropwise over 5 min to the stirred solution, immediately giving a white precipitate. The mixture was filtered through a short column of silica gel (filling 30 mL of a 60 mL fritted funnel), washed with THF, and the solvent removed to give 2.66 g of spectroscopically pure crude product (85% yield). This was dissolved in \sim 2 mL of ether, cooled to -30 °C, and then 2 ml of hexanes was layed on and the mixture allowed to stand at -30 °C overnight to give 0.68 g of fine white crystals; a second crop of 0.45 g was obtained in the same way (36% total yield). $[\alpha]_D^{26} = 372.8$ (c = 2.485, C_6H_6); IR (CHCl₃): 1787 (s), 1712 (s), 1678 (m) cm⁻¹; ¹H NMR (CDCl₃): δ 7.55 (m, 3H), 7.47 (m, 2H), 4.43 (br s, 1H), 1.70 (d, ${}^{3}J_{HH} = 6.8 \text{ Hz}$, 3H), 1.29 (s, 9H). ${}^{31}P$ NMR (CDCl₃): 134.6 ppm. 13 C NMR (CDCl₃): 173.5 (br s), 152.1 (br s), 139.8 (d, J_{PC} = 49.1 Hz), 132.6, 130.1 (d, J_{PC} = 26.1 Hz), 128.9 (d, J_{PC} = 7.1 Hz), 82.9, 50.4, 28.0, 18.3 ppm. Anal. Calcd for $C_{14}H_{18}NO_4P$: C_{14 56.95; H, 6.14; N, 4.74. Found: C, 56.84; H, 6.19; N, 4.79.

(2R, 4S)-3-tert-butyloxycarbonyl-4-isopropyl-5-oxo-2-phenyl-1,3,2-oxazaphospholidine (9). The synthetic details may be found in the published Note. Details of the structure determination may be found in the cif file, and comparison of selected bond lengths and angles to *cis*-3-toluenesulfonyl-5-oxo-2-phenyl-1,3,2-oxazaphospholidine (2) may be found in Table 1.

Variable temperature NMR spectra. Data collected as described in the published Note may be found in Table 2.

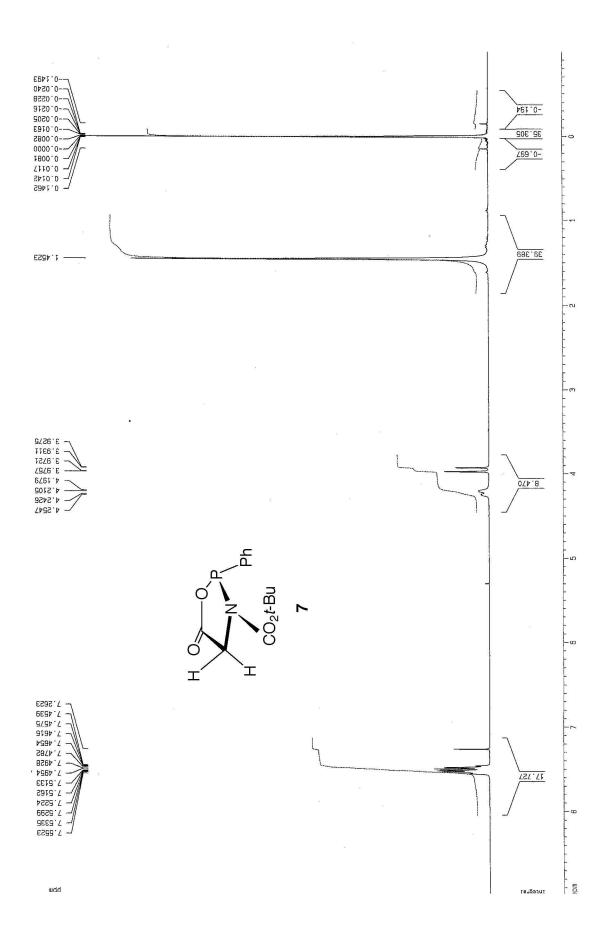
Table 1. Comparison of Bond Lengths (Å) and Angles (°) in **2** and **9**.

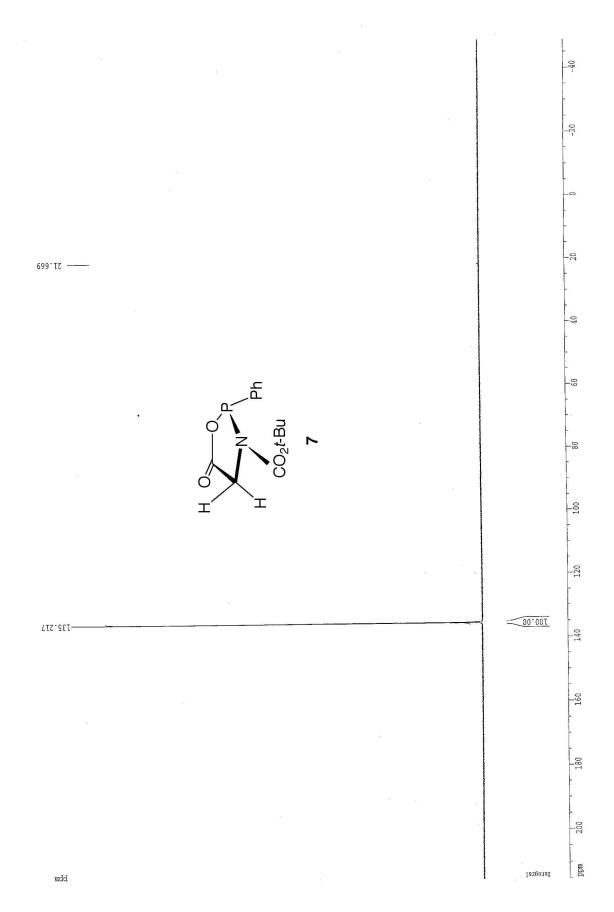
	2	9
P-O _{ring}	1.678 (3)	1.643 (6)
P-N	1.744 (4)	1.753 (6)
P-Ph	1.824 (5)	1.831 (8)
O_{ring} - C	1.367 (6)	1.372 (9)
N-C _{ring}	1.475 (6)	1.439 (9)
O_{ring} -P-N	90.9 (2)	88.7 (3)
O_{ring} -P-Ph	101.1 (2)	100.5 (3)
N-P-Ph	99.9 (2)	102.0 (3)
P-N-C _{ring}	112.5 (3)	116.1 4)
$P-N-R^a$	118.4 (2)	120.1 (5)
C_{ring} -N-R ^a	120.0 (3)	122.7 (6)
ΣN^b	350.9	358.9

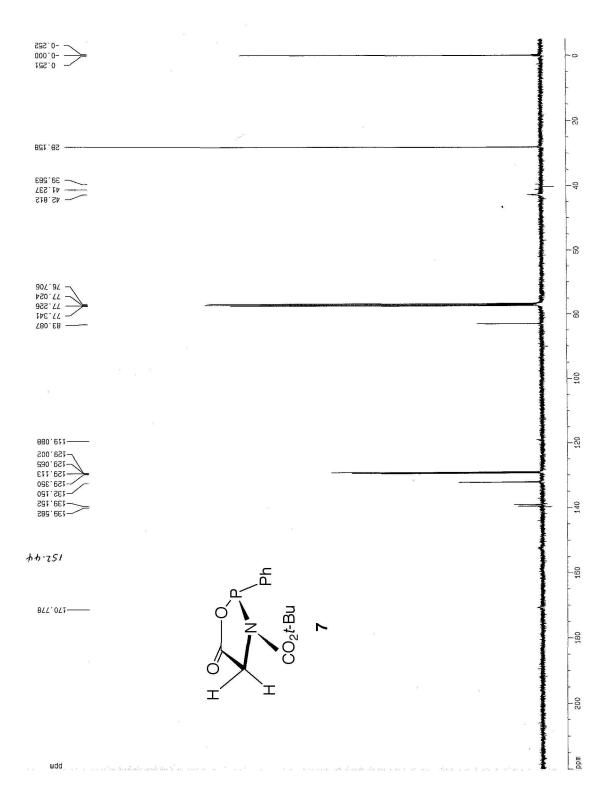
 $[\]overline{{}^{a}R'}$ = non-ring substituent on nitrogen. $\overline{{}^{b}}$ sum of the three angles at nitrogen.

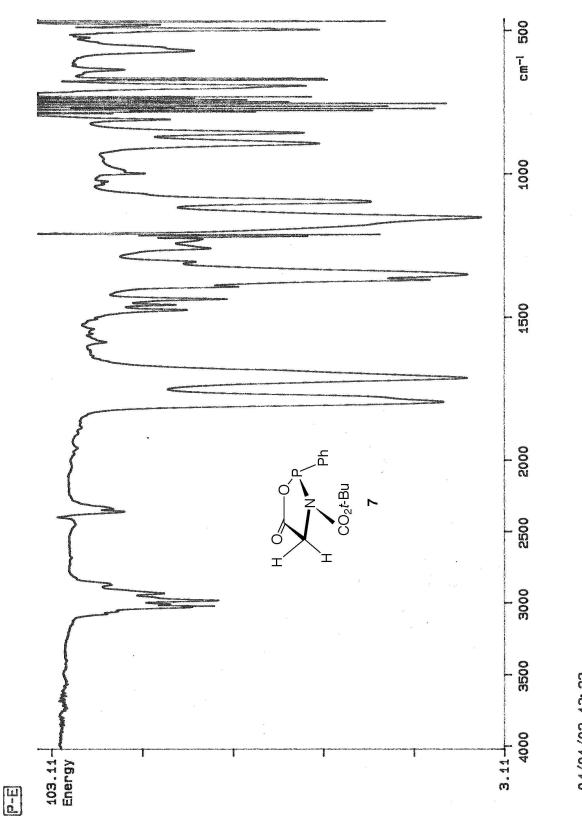
Table 2. Variable temperature NMR data

	A	В	С	D	F	F	G	н	T	1	K	- 1	М	N N	0	Р	0
1		ound 7		_	all chem, shi	ft differences in				-					-		`
2	T	31P integral	t-Bu integral	calcd 1/K		t-Bu	cis-CHN chem	cis-CHN	TMS	k(t-Bu) (s-1)	k(CHN) (s-1)						
3		K(31P)		for gNMR	shift diff.	extrapolated		extrapolated	v(1/2) (Hz)								
4	-30	4.3524	4.2424						1.624		0.05 ± 0.05						
5	-20	3.6333	3.9745	0.2561	0.0581	0.05705			1.867	0.7 ± 0.1	0.4 ± 0.2						
6	-10 0		3.7230	0.2763				0.06199	1.218 1.508	4 ± 1 7 ± 1	4 ± 1 7 ± 1						
8	10			0.2966		0.05531		0.06750	1.276	21 ± 1	22 ± 2						
9	20			0.3366		0,05357		0.07852	1.508	55 ± 5							
10	30			0.3564		0.0527		0.08403	1.16	00-0	150 ± 10						
11	40			0.3760		0.05183		0.08954	1.043		300 ± 20						
12	50			0.3954		0.05096		0.09505	0.986		600 ± 20						
13	60			0.4144		0.05009		0.10056	1.218		1100 ± 100						
14		K = [major]		For chem, shift	extrapolation					k for major	o minor						
15				slope=		-8.700E-05		5.510E-04									
16				int. =		0.05531		0.06750									
17	Comp	ound 8					all chem, shift	l lifferences in n	neo.								
19	Т		t-Bu integral	CHN integral	average K	calcd 1/K	t-Bu chem	t-Bu		CHN	TMS	k(t-Bu) (s-1)	L/CHN) (s-1)				
20	•	K(31P)		K(CHN)	a rorage K	for aNMR	shift diff.	extrapolated		extrapolated		V/(DO) (2-1)	1,0000				
21	-60	8.0588	7.5000	7.4706	7.6682	0.1341	0.2427		0.1143		2.2674						
22	-40		6.3158	6.0476	6.0940		0.2305	0.2311	0.137	0.1345		0.1 ± 0.1	0.5 ± 0.25				
23	-20		6.1257	5.7767	5.9468				0.1522	0.15345			1 ± 0.25				
24	-10					0.1832				0.162925	1.3338		5 ± 1				
25	22					0.2116		0.1953		0.193245	1.2758		60 ± 20				
26		K = [major]	[minor]				ft extrapolation:					k for major t	o minor				
27						slope=		-5.776E-04 0.2080		9.475E-04 0.1724							
28						int. =		0.2080		0.1/24							
	Comp	ound 9					all chem, shift	lifferences in n	nen								
	Т		t-Bu integral	CHN integral	average K	calcd 1/K	t-Bu chem	t-Bu		CHN	31P chem	31P	TMS	PPh3	k(t-Bu) (s-1)	k(CHN) (s-1)	k(P) (s-1)
32		K(31P)		K(CHN)		for gNMR	shift diff.	extrapolated		extrapolated		extrapolated		v(1/2) (Hz)		.,, , ,	
33	-40	10.5612	11.3750	11.6667	11.1826		0.2779	0.2780	0.1231	0.1243	0.9080	0.9045		2.78			
34	-20		9.5758	9.4904		0.1045	0.2633		0.1408	0.1392	0.8760	0.8800	1.07	3.05		0.2 ± 0.1	0.1 ± 0.1
35	- 0		8.4870	8.6235	8.0178	0.1172			0.1553	0.1540	0.8500	0.8554		2.46		0.7 ± 0.1	1 ± 0.5
36	10		9.6842	8.5030	8.6054				0.1598	0.1615	0.8490	0.8431				2 ± 0.5	1.5 ± 0.5
37	23 30					0.1312 0.1353		0.2318		0.1711 0.1763		0.8271 0.8185	0.88	3.16 1.65		7 ± 1 12 ± 1	5 ± 1 9 ± 1
39	40					0.1353		0.2267		0.1/63		0.8062	0.00			50 ± 10	35 ± 5
40	50					0.1411		0.2120		0.1912		0.7940	0.92			30 ± 10	50 ± 10
41		K = [major]	[minor]				ft extrapolation:	3.2120		0.1912		3 540			k for major t	o minor	35 - 10
42						slope=		-7.328E-04		7.427E-04		-1.229E-03		(line-width used t	or calculations		
43						int. =		0.2486		0.1540		0.8554		is shown above;	each is 1 Hz low	er .	
44														than actual PPh3			
		ound 10				lifferences in pp								at -40 °C, the t-E		was	
	Т	CH2 1/K		calcd 1/K	CH2 chem	extrapolated	TMS	k(CH2) (s-1)						1 Hz less than PP	h3)		
47	-20	1/K 0.11	calcd K(t-Bu) 9.0909		shift diff. 0.0845												
49	-20		8,3333	0.1100													
50	-10		7,6923	0.1326													
51	5	0.14	7.1429	0.1384													
52	15		6.6667	0.1500		0.0833											
53	25			0.1617		0.0829											
54	30			0.1676		0.0827	1.00										
	_	(= [major]/[minor] For chem. shift extrapolation:			1	k for major t	o minor				_						
55																	
				slope= int. =		-3.8983E-05 0.08388											

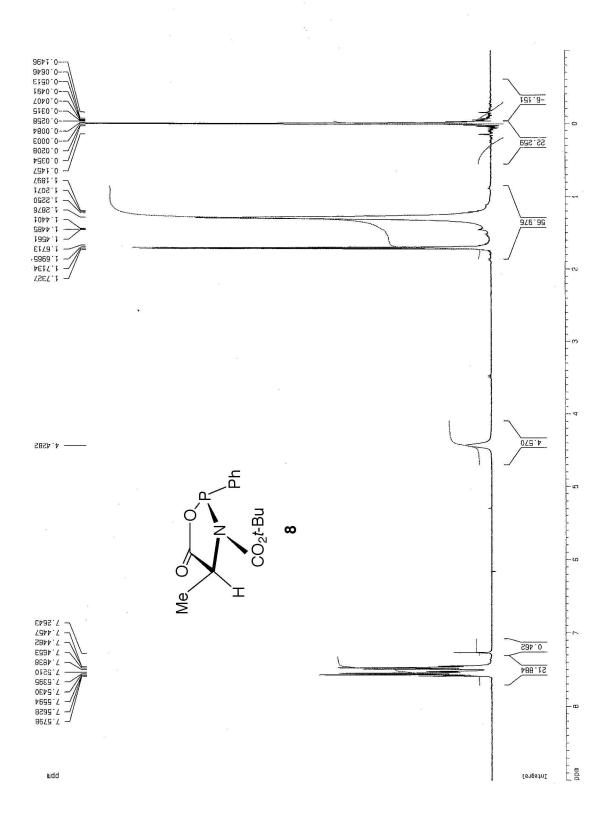


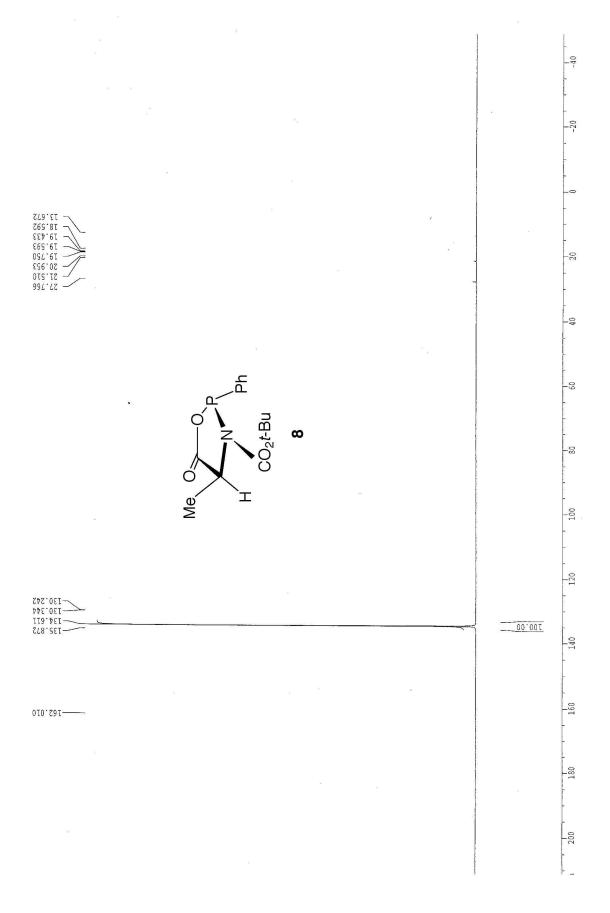


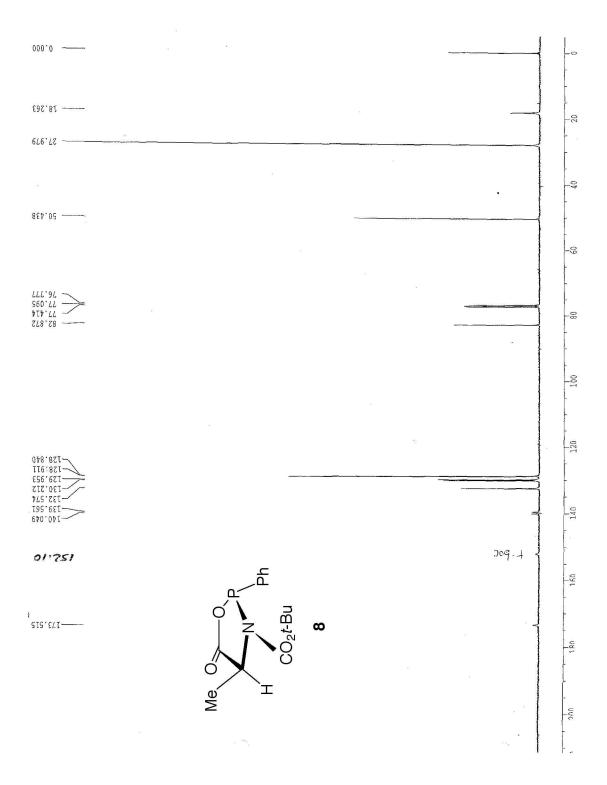


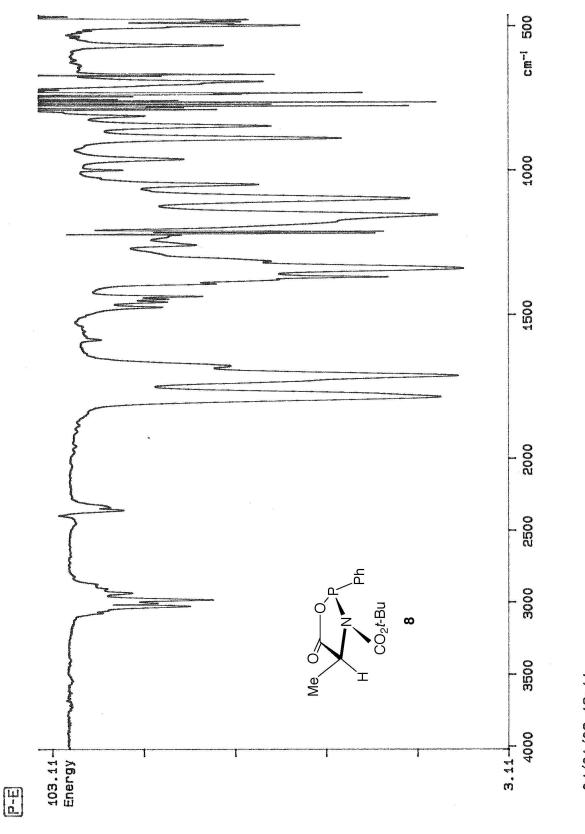


04/01/02 13:23 Y: 16 scans, 4.0cm-1, single, diff









04/01/02 12:11 Y: 16 scans, 4.0cm-1, single, diff

