Regiospecific Functionalization of Methyl C-H Bonds of Alkyl Groups in Reagents with Heteroatom Functionality

Supporting Information

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General. Unless noted otherwise, all manipulations were carried out under an inert atmosphere using a nitrogen-filled glovebox or standard Schlenk techniques. All glassware was oven-dried for approximately 1 h prior to use. THF and diethyl ether were obtained as HPLC grade without inhibitors; benzene, toluene, dichloromethane, and pentane were obtained as ACS reagent grade. THF, diethyl ether, benzene, toluene, dichloromethane, and pentane were degassed by purging with nitrogen for 45 min and dried with a solvent purification system containing a 1 m column containing activated alumina. Cyclohexane was distilled from sodium benzophenone ketyl under nitrogen. All reagents were obtained from commercial sources and used without further purification, unless otherwise noted. N,N-diethyl-3-pentanamine,¹ 2-(4bromobutyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane $(Br(CH_2)_4Bpin)^2$, $Cp*Rh(C_6Me_6)^3$ $[(COD)IrCl]_{2}^{4}$ O,O'-(S)-(1,1'-dinaphthyl-2,2'-diyl)-N,N'-di-(S,S)-phenylethylphosphoramidite,⁵ and Pd(dba)₂⁶ were prepared using the reported procedures. Triethylamine, 1-ethylpiperidine, 1butylpiperidine, and N_{N} -diethyl-3-pentanamine were dried over CaH₂ and vacuum transferred prior to use. Tributylamine was dried over CaH₂ and distilled under vacuum prior to use. Perfluorooctylethane, butyl ethyl ether, tert-butyl ethyl ether, and p-bromo-tert-butylbenzene were dried over activated alumina and filtered prior to use. 1-Fluorooctane was dried over anhydrous CaSO₄ and vacuum transferred prior to use. C₆D₁₂ was dried over sodium benzophenone ketyl and vacuum transferred prior to use. ¹H NMR spectra were obtained at 400 or 500-MHz and recorded relative to residual protio-solvent. ¹³C NMR spectra were obtained at 100.6 or 125.0 MHz, and chemical shifts were recorded relative to the solvent resonance. Both ¹H NMR and ¹³C NMR chemical shifts are reported in parts per million downfield from tetramethylsilane. ¹³C NMR signals for carbon atoms α to boronate ester groups were either not observed or observed between $\delta = 20$ and 0 as broad peaks near the baseline. ¹⁹F NMR spectra were obtained at 376 MHz, and chemical shifts were recorded with a C_6F_6 external standard ($\delta =$ -164.9). ¹¹B NMR spectra were obtained at 160 MHz or 80 MHz, and chemical shifts were recorded relative to BF₃•OEt₂ as an external standard. GC analyses were performed using a DB-1301 narrow bore column. Response factors were calculated from the ratios of pure product to added dodecahydrotriphenylene in ¹H NMR spectra and GC traces.

Representative procedure for reactions with B_2pin_2 as the limiting reagent: *N*-[4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)butyl]-dibutylamine. Cp*Rh(η^4 -C₆Me₆) (2.8 mg, 7.0 µmol), B_2pin_2 (35.0 mg, 0.138 mmol), and Bu_3N (256 mg, 1.38 mmol) were placed into

an NMR tube. The tube was sealed under vacuum, and the tube was heated at 150 °C for 24 h. The NMR tube was opened, and the contents were dissolved in benzene. Dodecahydrotriphenylene (11.8 mg, 0.0491 mmol) was added as an internal standard, and an aliquot was removed and analyzed by GC. The yield of boronate ester product was 75% based on the reaction of tributylamine withi B₂pin₂ to generate two equiv of the functionalized product and H₂. The compound was independently synthesized as follows: Dibutylamine (12.8 mL, 76.2 mmol) and 2-(4-bromobutyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (2.01 g, 7.62 mmol) were heated in 20 mL of MeCN at 70 °C for 4 h. The reaction mixture was cooled and added to a 10:1 mixture of pentane and toluene. Dibutylammonium bromide precipitated as a white solid. Filtration and evaporation of the solvent gave the crude product, which was nearly pure, as indicated by ¹H NMR spectroscopy. The product was distilled at 75-80 °C and 50 mtorr to give 740 mg (31%) of the boronate ester product. ¹H NMR (C₆D₆, 500 MHz) δ = 2.42 (t, 2H), 2.38 (t, 4H), 1.66 (m, 2H), 1.65 (m, 2H), 1.42 (m, 4H), 1.33 (m, 4H), 1.06 (s, 12H), 1.03 (t, 2H), 0.91 (t, 6H); ¹¹B NMR (C₆D₆, 160 MHz) δ = 34.2; ¹³C{¹H} NMR (C₆D₆, 126 MHz) δ = 83.1, 54.9, 54.8, 30.9, 30.6, 25.3, 22.9, 21.4, 14.7. The carbon attached to boron was not observed. Anal. Calcd for C₁₈H₃₈NBO₂: C, 69.45; H, 12.30; N, 4.50. Found: C, 69.69; H, 12.54; N, 4.43.

Representative procedure for reactions with the organic substrate as limiting reagent: 2-(3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,10-Heptadecafluoro-decyl)-4,4,5,5-tetramethyl-1,3,2dioxaborolane. Cp*Rh(η^4 -C₆Me₆) (2.8 mg, 7.0 µmol), B₂pin₂ (35.0 mg, 0.138 mmol), and CF₃(CF₂)₇CH₂CH₃ (61.8 mg, 0.138 mmol) were placed into an NMR tube and sealed under vacuum. The tube was heated at 150 °C for 24 h. The NMR tube was opened in a drybox, and additional Cp*Rh(η^4 -C₆Me₆) (2.8 mg, 7.0 µmol) and B₂pin₂ (35.0 mg, 0.138 mmol) were added. The tube was again sealed under vacuum and heated for an additional 24 h at 150 °C. The NMR tube was opened, and the contents were dissolved in benzene. Dodecahydrotriphenylene (16.2 mg, 0.0674 mmol) was added as an internal standard, and an aliquot was removed and analyzed by GC. The yield for the reaction of B₂pin₂ to form the functionalized product and HBpin was 84%. To obtain isolated material for spectral and analytical data, the borylation reaction was conducted with Cp*Rh(η^4 -C₆Me₆) (3.4 mg, 9 µmol), B₂pin₂ (43.7 mg, 0.172 mmol), and CF₃(CF₂)₇CH₂CH₃ (309 mg, 0.688 mmol). The product was purified by chromatography on silica gel, eluting with pentane, to give 31 mg (31 %) of the boronate ester product. ¹H NMR (C₆D₆, 500 MHz) δ = 2.12 (m, 2H), 1.00 (t, 2H), 0.99 (s, 12H); ¹¹B NMR (C₆D₆, 160 MHz) δ = 33.7; ¹³C{¹H} NMR (C₆D₆, 126 MHz) δ = 120 to 105 (m, *CF*'s), 83.9, 26.5, 25.1, 4.5, the carbon attached to boron was not observed; ¹⁹F NMR (C₆D₆, 376 MHz) δ = -82.1 (3F, t), -116.8 (2F, m), -122.7 (2F, m), -123.0 (4F, m), -123.9 (2F, m), -124.5 (2F, m), -127.3 (2F, m). Anal. Calcd for C₁₆H₁₆BF₁₇O₂: C, 33.47; H, 2.81. Found: C, 33.65; H, 2.64.

2-(2-tert-Butoxy-ethyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane. The yield for the borylation of *tert*-butyl ethyl ether with B₂pin₂ as the limiting reagent was obtained by the representative procedure. Procedure for borylation with *tert*-butyl ethyl ether as limiting reagent: $Cp*Rh(\eta^4-C_6Me_6)$ (5.6 mg, 14 µmol), B_2pin_2 (41.6 mg, 0.164 mmol), *tert*-butyl ethyl ether (28.2) mg, 0.276 mmol), and cyclohexane (90 µL) placed into an NMR tube and sealed under vacuum. The tube was heated at 150 °C for 24 h. After this time, the NMR tube was opened in a drybox, and additional Cp*Rh(η^4 -C₆Me₆) (5.6 mg, 14 µmol) and B₂pin₂ (28.4 mg, 0.112 mmol) were added. The tube was again sealed under vacuum and heated for an additional 24 h at 150 °C. The NMR tube was opened, and the contents were dissolved in benzene. Dodecahydrotriphenylene (10.0 mg, 0.0416 mmol) was added, and an aliquot was removed and analyzed by GC. The yield of the boronate ester product was 48%, based on the reaction of B₂pin₂ to generate the functionalized product and HBpin. To obtain isolated material for spectral and analytical data, the borylation reaction was conducted with Cp*Rh(η^4 -C₆Me₆) (80 mg, 0.20 mmol), B₂pin₂ (1.01 g, 4.00 mmol), and *tert*-butyl ethyl ether (4.11 g, 40.0 mmol). The product was isolated by chromatography on silica, eluting with pentane:ether (10:1) to give 464 mg (51%) of the boronate ester product. ¹H NMR (C₆D₆, 500 MHz) δ = 3.63 (t, 2H), 1.36 (t, 2H); 1.14 (s, 9H), 1.06 (s, 12H); ¹¹B NMR (C₆D₆, 160 MHz) δ = 33.7; ¹³C{¹H} NMR (C₆D₆, 126 MHz) δ = 83.2, 72.4, 58.9, 28.1, 25.3. The carbon attached to boron was not observed. Anal. Calcd for C₁₂H₂₅BO₃: C, 63.18; H, 11.05. Found: C, 63.30; H, 11.09, N < 0.02.

2-But-3-enyl-2,4,4,5,5-pentamethyl-[1,3]dioxolane. 5-Hexen-2-one (4.91 g, 50.0 mmol), trimethylorthoformate (12.7 g, 120 mmol), *p*-toluenesulfonic acid monohydrate (95 mg, 0.50 mmol), and C_6H_6 (40 mL) were heated at reflux under nitrogen for 5 h. The red-brown solution was cooled to room temperature and neutralized with NaHCO₃. The solution was filtered, and the solvent was evaporated to give a red-brown liquid. The product was heated at reflux with pinacol (3.55 g, 30.0 mmol), and *p*-toluenesulfonic acid monohydrate (57.1 mg, 0.300 mmol) in CHCl₃ (40 mL) under nitrogen for 6 h. The solution was cooled to room temperature, neutralized with NaHCO₃, filtered, and the solvent was evaporated under reduced pressure. The resulting oil

was purified by column chromatography on silica gel, eluting with ethyl acetate/hexane (1:6) to give 3.54 g (36%) of the the product. ¹H NMR (C₆D₆, 400 MHz) δ = 1.06 (s, 6H), 1.09 (s, 6H), 1.37 (s, 3H), 1.81 (m, 2H), 2.22 (m, 2H), 4.93 (ddt, 1H), 5.02 (dq, 1H), 5.81 (ddt, 1H); ¹³C{¹H} NMR (C₆D₆, 126 MHz) δ = 24.9, 25.0, 27.2, 29.9, 42.9, 82.3, 107.4, 114.2, 139.2. Anal. Calcd for C₁₂H₂₂O₂: C, 72.68; H, 11.18. Found: C, 72.77; H, 11.10.

6-[2-(2,3-Dimethyl-2,3-butanedioxy)hexyl]-4,4,5,5-tetramethyl-1,3,2-dioxaborolane.

Yields for the catalytic borylation of 2-but-3-enyl-2,4,4,5,5-pentamethyl-[1,3]dioxolane were obtained by the representative procedures. This compound was independently synthesized as follows: 2-But-3-enyl-2,4,4,5,5-pentamethyl-[1,3]dioxolane (642 mg, 3.24 mmol), pinacolborane (498 mg, 3.89 mmol), Cp*Rh(C₂H₄)₂ (10 mg, 340 µmol), and CH₂Cl₂ (2 mL) were heated at reflux under nitrogen for 12 h. The brown solution was quenched with H₂O (10 mL), extracted with ether (3 x 30 mL), dried over MgSO₄, and reduced to an oil under vacuum. Vacuum distillation, followed by column chromatography on silica gel, eluting with ethyl acetate/hexane (1:6), gave 593 mg (80%) of the product. ¹H NMR (C₆D₆, 400 MHz) δ = 0.99 (t, 2H), 1.06 (s, 12H), 1.10 (s, 6H), 1.11 (s, 6H), 1.45 (s, 3H), 1.62 (m, 4H), 1.78 (t, 2H); ¹¹B NMR (C₆D₆, 80 MHz) δ = 34.0; ¹³C{¹H} NMR (C₆D₆, 100 MHz) δ = 24.9, 25.0, 25.1, 27.3, 28.2, 43.7, 82.1, 82.7, 107.9. The carbon attached to boron was not observed. Anal. Calcd for C₁₈H₃₅B₁O₄: C, 66.26; H, 10.81. Found: C, 66.33; H, 11.10.

2-(8-Fluoro-octyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane. Yields for the catalytic reactions of 1-fluorooctane were obtained by the representative procedure. Reaction of 1-fluorooctane with B₂pin₂ for extended times occasionally formed BF₃ and octene, as determined by ¹¹B NMR and GC-MS respectively. However, this decomposition was not observed after 12 h of reaction time. To obtain isolated material for spectral and analytical data, the borylation reaction was conducted with Cp*Rh(η^4 -C₆Me₆) (56 mg, 140 mmol), B₂pin₂ (700. mg, 2.76 mmol), and 1-fluorooctane (2.1 g, 16 mmol). After excess 1-fluorooctane and HBpin were evaporated, the product was isolated by chromatography on silica, eluting with hexane/ethylacetate (10:1) to give 562 mg (79%). ¹H NMR (C₆D₆, 400 MHz) δ = 4.10 (dt, 2H), 1.61 (m, 2H), 1.45-1.30 (m, 4H), 1.26-1.10 (m, 6H), 1.08 (s, 12H), 1.01 (t, 2H); ¹¹B NMR (C₆D₆, 80 MHz) δ = 34.0; ¹³C{¹H} NMR (C₆D₆, 126 MHz) δ = 84.0 (d, *J*=166 Hz), 83.1, 33.0, 31.1, 30.1, 29.8, 25.8, 25.3, 24.9; ¹⁹F{¹H} NMR (C₆D₆, 376 MHz) δ = -213. Anal. Calcd for C₁₄H₂₈BFO₂: C, 65.13; H, 10.93. Found: C, 65.50; H, 11.03.

1-[2-(4,4,5,5-Tetramethyl-1,3,2-dioxaborolan-2-yl)-ethyl]-piperidine. The yield for the catalytic borylation of ethylpiperidine was obtained by the representative procedure. To obtain isolated material for spectral and analytical data, the borylation reaction was conducted with Cp*Rh(η⁴-C₆Me₆). The product mixture was treated with portions of PL-DEAM scavenger in benzene until the B₂pin₃ and pinBOBpin side products were consumed, as determined by ¹¹B NMR spectroscopy (~2 equiv). Distillation at 55-60 °C and 60 mtorr gave 1.12 g (46%) of the boronate ester product. ¹H NMR (C₆D₆, 500 MHz) δ = 2.59 (t, 2H), 2.33 (broad, 4H), 1.50 (m, 4H), 1.30 (m, 2H), 1.19 (t, 2H), 1.08 (s, 12H); ¹¹B NMR (C₆D₆, 80 MHz) δ = 33.8; ¹³C {¹H} NMR (C₆D₆, 126 MHz) δ = 83.1, 55.4, 54.8, 26.9, 25.5, 25.4. The carbon attached to boron was not observed. Anal. Calcd for C₁₃H₂₆NBO₂: C, 65.29; H, 10.96; N, 5.86. Found: C, 65.50; H, 11.07; N, 5.91.

Formation of a mixture of 2-(4-Butoxy-ethyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane and 2-(4-ethoxy-butyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane from n-butyl ethyl ether. The ratio of products from the borylation of n-butyl ethyl ether with B₂pin₂ as the limiting reagent was obtained by the representative procedure. The mixture of boronate ester products was purified by chromatography on silica gel, eluting with pentane:ether (10:1). After chromatography, the ratio of 2-(4-butoxy-ethyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane to 2-(4ethoxy-butyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane was greater than 6:1, and the NMR signals of the product from borylation for the ethyl group could be identified clearly. The product from borylation of the butyl group was prepared independently, as described in the following procedure. Signals due to CH₃(CH₂)₄O(CH₂)₂Bpin: ¹H NMR (C₆D₆, 400 MHz) δ = 3.67 (t, 2H), 3.32 (t, 2H), 1.53 (m, 2H), 1.34 (overlapping signals, m, 4H), 1.05 (s, 12H), 0.86 (t, 3H); ¹¹B NMR (C₆D₆, 160 MHz) δ = 34.0; ¹³C {¹H} NMR (C₆D₆, 126 MHz) δ = 83.3, 70.8, 68.2, 32.8, 25.3, 20.2, 14.5. The carbon attached to boron was not observed. Anal. Calcd for C₁₂H₂₅BO₃: C, 63.18; H, 11.05. Found: C, 63.21; H, 11.22, N < 0.02.

2-(4-Ethoxy-butyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane. This product was independently synthesized as follows: A solution of 2-(4-bromobutyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (2.50 g, 9.54 mmol) was treated with a solution of KOEt generated by the addition of KH (0.574 g, 14.3 mmol) to EtOH. The solution was stirred under nitrogen for 20 h. The product was isolated by chromatography on silica, eluting with pentane:ether (10:1) to give 267 mg (12%) of the boronate ester. ¹H NMR (C₆D₆, 500 MHz) δ = 3.32 (broad m, 2H), 3.27 (q, 2H),

1.71 (overlapping m, 4H), 1.10 (t, 3H), 1.04 (s, 12H), 0.98 (broad t, 2H); ¹¹B NMR (C₆D₆, 160 MHz) $\delta = 34.4$; ¹³C{¹H} NMR (C₆D₆, 126 MHz) $\delta = 83.1$, 71.1, 66.5, 33.3, 25.3, 21.7, 15.9. The carbon attached to boron was not observed. Anal. Calcd for C₁₂H₂₅BO₃: C, 63.18; H, 11.05. Found: C, 62.97; H, 10.97, N < 0.02.

Formation of a mixture of N, N-diethyl-[1-ethyl-3-(4,4,5,5-tetramethyl-1,3,2dioxaborolan-2-vl)-propyl]-amine and Ethyl-(1-ethyl-propyl)-[2-(4,4,5,5-tetramethyl-1,3,2dioxaborolan-2-vl)-ethyl]-amine. The ratio of products from the borylation of N,N-diethyl-3pentylamine with B₂pin₂ as the limiting reagent was obtained by the representative procedures. To obtain isolated material for spectral and analytical data, the borylation reaction was conducted with Cp*Rh(n⁴-C₆Me₆) (56 mg, 140 µmol), B₂pin₂ (0.700 g, 2.76 mmol), and N,N-diethyl-3pentylamine (0.700 g, 4.89 mmol). The product mixture was treated with portions of PL-DEAM scavenger in benzene until B₂pin₃ and pinBOBpin side products were consumed, as determined by ¹¹B NMR spectroscopy (~1 equiv). The mixture was distilled at 90 °C and 50 mtorr to give 243 mg (33%) of the boronate ester products. Signals due to ethyl-(1-ethyl-propyl)-[2-(4,4,5,5tetramethyl-1,3,2-dioxaborolan-2-yl)-ethyl]-amine: ¹H NMR (C₆D₆, 500 MHz) δ = 2.76 (t, 2H), 2.43 (m, 1H), 2.40 (m, 2H), 1.40 (m, 4H), 1.20 (m, 2H), 1.08 (s, 12H), 1.02 (t, 3H), 0.93 (t, 6H); the signals for the other isomer are listed in the following procedure; ${}^{11}B$ NMR (C₆D₆, 80 MHz) $\delta = 34.0$; ¹³C{¹H} NMR (C₆D₆, 126 MHz) $\delta = 83.1$, 63.6, 45.8, 43.6, 25.4, 23.4, 15.7, 12.8. Anal. Calcd for C₁₅H₃₂NBO₂: C, 66.92; H, 11.98, N, 5.20. Found for the product mixture: C, 66.65; H, 11.78; N, 4.92.

N,N-diethyl-[1-ethyl-3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-propyl]-amine.⁷

This product was independently synthesized as follows: A mixture of $[(COD)IrCI]_2$ (129 mg, 0.096 mmol), *O*, *O*'-(*S*)-(1,1'-dinaphthyl-2,2'-diyl)-*N*,*N*'-di-(*S*,*S*)-phenylethylphosphoramidite (104 mg, 0.192 mmol) and Et₂NH (456 mg, 6.24 mmol) in THF (2 mL) was heated at 80 °C for 15 min. The mixture was cooled to 0 °C, and methyl *trans*-pent-2-enyl carbonate (692 mg, 4.80 mmol) was added. The reaction mixture was allowed to warm to room temperature over 3 h. After an additional 9 h, the solvent was evaporated, and the resulting liquid was purified by bulb-to-bulb distillation under vacuum, to give 260 mg (38%) of a 1:1 mixture of terminal and internal alkenes, as indicated by the ¹H NMR spectrum. HBpin (471 mg, 3.68 mmol) was added to the mixture of isomers, and the mixture was sealed under vacuum in an NMR tube and heated at 150 °C for 24 h. After this time, ¹¹B NMR indicated formation of the alkylboronate ester(s), along

with pinBOBpin, B₂pin₃, or both. The reaction mixture was concentrated to an oil under vacuum, dissolved in benzene, and treated with PL-DEAM scavenger until the ¹¹B NMR spectrum indicated that all the pinBOBpin and B₂pin₃ had been consumed (~2 equiv total, 48 h total). The product was distilled at 65 °C and 20 mtorr to give 20 mg (4%, 1.5% for two steps) of the boronate ester product. ¹H NMR (C₆D₆, 400 MHz) δ = 2.45 (m, 1H), 2.41 (m, 4H), 1.62 (m, 2H), 1.32 (m, 2H), 1.07 (s, 12H), 1.04 (t, 2H), 1.02 (t, 6H), 0.96 (t, 3H); ¹¹B NMR (C₆D₆, 80 MHz) δ = 33.4; ¹³C{¹H} NMR (C₆D₆, 126 MHz) δ = 83.0, 64.0, 44.0, 25.4, 25.1, 23.5, 15.8, 12.8. The carbon attached to boron was not observed. Anal. Calcd for C₁₅H₃₂NBO₂: C, 66.92; H, 11.98, N, 5.20. Found: C, 66.65; H, 12.21; N, 5.26.

Formation of a mixture of 2-(4-ethoxy-butyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane, 2-(4-Butoxy-ethyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane and 2-(4-butoxy-butyl)-4,4,5,5tetramethyl-1,3,2-dioxaborolane. Cp*Rh(η^4 -C₆Me₆) (2.8 mg, 7.0 µmol), B₂pin₂ (35.0 mg, 0.138 mmol), butyl ethyl ether (70.5 mg, 0.690 mmol), and dibutyl ether (89.9 mg, 0.690 mmol) were placed into an NMR tube and sealed under vacuum. The tube was heated at 150 °C for 24 h. The NMR tube was opened, and the contents were dissolved in benzene. Dodecahydrotriphenylene (23.5, 0.0978 mmol) was added, and an aliquot was removed and analyzed by GC. The ratio of 2-(4-butoxy-ethyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane to 2-(4ethoxy-butyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane to 2-(4butoxy-butyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane to 2-(4ethoxy-butyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane to 2-(4-butoxy-butyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane was determined to be 4:1:2, after correction for response factors

Formation of a mixture of 2-(3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,10-heptadecafluoro-decyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane to 2-octyl-4,4,5,5-tetramethyl-1,3,2-dioxaborolane. Cp*Rh(η^4 -C₆Me₆) (1.5 mg, 4.0 μ mol), B₂pin₂ (20.5 mg, 0.0807 mmol), (perfluoro-noctyl)ethane (77.0 mg, 1.72 mmol), and octane (9.8 mg, 0.0859 mmol) were placed into an NMR tube and sealed under vacuum. The tube was heated at 150 °C for 24 h. The NMR tube was opened, and the contents were dissolved in benzene. Dodecahydrotriphenylene (19.8 mg, 0.0824 mmol) was added, and an aliquot was removed and analyzed by GC. The ratio of 2-(3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,10-heptadecafluoro-decyl)-4,4,5,5-tetramethyl-1,3,2-

dioxaborolane to 2-octyl-4,4,5,5-tetramethyl-1,3,2-dioxaborolane was determined to be 94:6, after correction for response factors.

1-(2-*tert*-Butoxy-ethyl)-4-*tert*-butyl-benzene (Table 2, Entry 1). Cp*Rh(η^4 -C₆Me₆) (2.8 mg, 7.0 μ mol), B₂pin₂ (35.0 mg, 0.138 mmol), and *tert*-butyl ethyl ether (141 mg, 1.38 mmol)

were placed into an NMR tube and sealed under vacuum. The tube was heated at 150 °C for 24 h. The NMR tube was opened in a drybox, and the contents were dissolved in toluene (0.4 mL) and transferred to a vial containing *p*-bromo-*tert*-butylbenzene (70.6 mg, 0.331 mmol), CsOH·H₂O (139 mg, 0.820 mmol), Pd(dba)₂ (7.9 mg, 14 µmol), and 1,1'bis(diisopropylphosphino)ferrocene (5.8 mg, 14 µmol). The vial was sealed with a PTFE lined screw-cap, and was heated for 24 h at 100 °C. The vial was then opened, and the contents were diluted with benzene. Dodecahydrotriphenylene (12.9 mg, 0.0532 mmol) was added as an internal standard, and an aliquot was removed and analyzed by GC. The yield of the tertbutoxyalkylarene was 87%, based on the reaction of the ether and B₂pin₂ to form the functionalized product and HBpin. The alkylarene product was independently synthesized as follows: Isobutylene was bubbled through a solution of 2-(4-tert-butylphenyl)ethanol (1.10 g, 6.17 mmol) in 20 mL of hexane containing Amberlyst 15 (300 mg) with rapid stirring for 5 h. The solution was filtered through Celite, and the solvent was evaporated to give 788 mg (54%) of the alkylarene. ¹H NMR (C₆D₆, 400 MHz) δ = 7.52 (d, 2H), 7.12 (d, 2H), 3.37 (t, 2H), 2.78 (t, 2H), 1.12 (s, 9H), 0.98 (s, 9H); ${}^{13}C{}^{1}H$ NMR (C₆D₆, 126 MHz) δ = 149.2, 137.4, 129.5, 125.7, 72.7, 63.7, 37.8, 34.7, 31.9, 28.0. Anal. Calcd for C₁₇H₃₀O: C, 81.99; H, 11.18. Found: C, 81.70; H, 11.25, N < 0.02.

1-tert-Butyl-4-(8-fluoro-octyl)-benzene (Table 2, Entry 2). Cp*Rh(n⁴-C₆Me₆) (2.8 mg, 7.0 µmol), B₂pin₂ (35.0 mg, 0.138 mmol), and 1-fluorooctane (182 mg, 1.38 mmol) were placed into an NMR tube and sealed under vacuum. The tube was heated at 150 °C for 12 h. The NMR tube was opened in a drybox, and the contents were dissolved in DMF (0.5 mL) and transferred to a vial containing p-bromo-tert-butylbenzene (58.8 mg, 0.276 mmol), CsF (83.8 mg, 0.552 mmol), Pd(dba)₂ (7.9 mg, 14 µmol), and 1,1'-bis(diisopropylphosphino)ferrocene (5.8 mg, 14 µmol). The vial was sealed with a PTFE-lined screw-cap and heated for 24 h at 100 °C. The vial was then opened, and the contents were diluted with benzene. Dodecahydrotriphenylene (19.6 mg, 0.0815 mmol) was added as an internal standard, and an aliquot was removed and analyzed by GC. The yield of the 8-fluorooctylarene product was 29%, based on the reaction of the fluorooctane with B₂pin₂ to form the functionalized product and HBpin. To obtain isolated material for spectral and analytical data, 2-(8-fluoro-octyl)-4,4,5,5-tetramethyl-1,3,2dioxaborolane (122 mg, 0.473 mmol), p-bromo-tert-butylbenzene (142 mg, 0.662 mmol), CsOH·H₂O (278 mg, 1.64 mmol), Pd(dba)₂ (15.8 mg, 28 umol), 1,1'-

bis(diisopropylphosphino)ferrocene (11.2 mg, 28 µmol) and toluene (0.8 mL) were placed in a vial. The vial was sealed with a PTFE lined screw-cap, and heated for 24 h at 100 °C. The product was purified by thin -layer chromatography on supported silica gel, eluting with hexanes followed by extraction with dichloromethane to give 30 mg (24%) of the fluoroalkylarene product. ¹H NMR (C₆D₆, 500 MHz) δ = 7.20 (d, 2H), 7.02 (d, 2H), 4.33 (dt, 2H), 2.47 (t, 2H), 1.53 (overlapping signals, m, 4H), 1.24 (overlapping signals, m, 8H), 1.21 (s, 9H); ¹³C{¹H} NMR (CDCl₃, 126 MHz) δ = 148.6, 140.0, 128.2, 125.3, 84.4 (d, *J*=164 Hz), 35.6, 34.5, 31.7, 30.7, 30.6, 29.6, 29.5, 25.4 (d, J=5.6 Hz) (d). Anal. Calcd for C₁₇H₂₇F: C, 81.54; H, 10.87. Found: C, 81.62; H, 10.92 N < 0.02.

1-tert-Butyl-4-(3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,10-heptadecafluoro-decyl)-benzene (Table 2, Entry 3). Cp*Rh(n⁴-C₆Me₆) (2.8 mg, 7.0 µmol), B₂pin₂ (35.0 mg, 0.138 mmol), and CF₃(CF₂)₇CH₂CH₃ (274 mg, 0.552 mmol) were placed into an NMR tube and sealed under vacuum. The tube was heated at 150 °C for 24 h. The NMR tube was opened in a drybox, and the contents were dissolved in toluene (0.4 mL) and transferred to a vial containing *p*-bromo-tertbutylbenzene (70.6 mg, 0.331 mmol), CsOH·H₂O (139 mg, 0.820 mmol), Pd(dba)₂ (7.9 mg, 14 μmol), and 1,1'-bis(diisopropylphosphino)ferrocene (5.8 mg, 14 μmol). The vial was sealed with a PTFE lined screw-cap and heated for 24 h at 100 °C. The vial was then opened, and the contents were diluted with benzene. Dodecahydrotriphenylene (22.8 mg, 0.0948 mmol) was added as an internal standard, and an aliquot was removed and analyzed by GC. The yield of fluoroalkylarene was 64%, based on the reaction of B_2pin_2 to form the functionalized product and HBpin. To obtain isolated material for spectral and analytical data, the borylation reaction was conducted with Cp*Rh(η^4 -C₆Me₆) (5.6 mg, 14 μ mol), B₂pin₂ (70.0 mg, 0.276 mmol), and $CF_3(CF_2)$ ₇CH₂CH₃ (589 mg, 1.10 mmol) and the coupling reaction was conducted with *p*-bromotert-butylbenzene (142 mg, 0.662 mmol), CsOH·H₂O (278 mg, 1.64 mmol), Pd(dba)₂ (15.8 mg, 28 µmol), 1,1'-bis(diisopropylphosphino)ferrocene (11.2 mg, 28 µmol). The product was isolated by chromatography on silica gel, eluting with hexane to give 28 mg (18 %) of the fluoroalkylarene product. ¹H NMR (C₆D₆, 500 MHz) δ = 7.21 (d, 2H), 6.81 (d, 2H), 2.63 (m, 2H), 2.07 (m, 2H), 1.22 (s, 9H); ${}^{13}C{}^{1}H$ NMR (CDCl₃, 126 MHz) δ = 149.9, 136.3, 128.2, 125.9, 120 to 105 (m, CF's), 34.7, 33.3, 31.5, 26.1. ¹⁹F NMR (C₆D₆, 376 MHz) δ = -82.1 (3F, m), -115.5 (2F, m), -122.7 (2F, m), -123.0 (4F, m), -123.8 (2F, m), -124.4 (2F, m), -127.3 (2F, m). Anal. Calcd for $C_{20}H_{17}F_{17}$: C, 41.39; H, 2.95. Found: C, 41.62; H, 2.77, N < 0.02.

4-(2,4,4,5,5-Pentamethyl-[1,3]dioxolan-2-yl)-butan-1-ol (Table 2, Entry 4).⁸ Cp*Rh(n⁴-C₆Me₆) (2.8 mg, 7.0 µmol), B₂pin₂ (35.0 mg, 0.138 mmol), and 2-butyl-2,4,4,5,5-pentamethyl-[1,3]dioxolane (140 mg, 1.38 mmol) were placed into an NMR tube and sealed under vacuum. The tube was heated at 150 °C for 24 h. The NMR tube was opened, and the contents were dissolved in THF (3 mL). This solution was treated dropwise with aqueous KOH (2 mL, 2 M) and H₂O₂ (2 mL, 30%) at 0 °C and then allowed to warm to room temperature. After 2 h, the aqueous solution was extracted with diethyl ether. The ether solution was dried over MgSO₄ and concentrated to an oil under vacuum. A solution of 1.3,5-trimethoxybenzene (10.1 mg, 60, umol) in CD₃OD was added to the oil and comparison of the integrations of the ¹H NMR resonances of the product versus the integrations of the methyl resonance of 1,3,5-trimethoxybenzene indicated that the yield of alcohol was 68%, based on the reaction of the dioxolane and B_2pin_2 to the boronate ester and HBpin. To obtain isolated material for spectral and analytical data, the borylation reaction was conducted with Cp*Rh(η^4 -C₆Me₆) (83 mg, 0.21 mmol), B₂pin₂ (754 mg, 2.97 mmol), and 2-butyl-2,4,4,5,5-pentamethyl-[1,3]dioxolane (2.0 g, 11.9 mmol) and the crude product was converted to the alcohol with aqueous KOH (2 mL, 2 M) and H₂O₂ (2 mL, 30%). Analytically pure product (43 mg, 8%) was obtained by column chromatography on silica gel, eluting with ethyl acetate/hexane (1:10) and then with ethyl acetate/hexane (2:1), followed by distillation at 60 °C and 20 mtorr. ¹H NMR (C₆D₆, 400 MHz) δ = 3.38 (t, 2H), 1.71 (t, 2H), 1.54 (m, 4H), 1.43 (s, 3H), 1.11 (s, 6H), 1.10 (s, 6H), 1.05 (broad, 1H); ¹³C{¹H} NMR (C₆D₆, 126 MHz) $\delta = 108.2, 82.7, 62.9, 43.8, 33.9, 27.5, 25.3, 25.2, 22.2$. The carbon attached to boron was not observed. Anal. Calcd for $C_{12}H_{24}O_5$: C, 66.63; H, 11.18. Found: C, 66.46; H, 10.93, N < 0.02.

Potassium 2-(*tert***-butoxy)ethyltrifluoroborate (Table 2, Entry 5).**⁹ Cp*Rh(η^4 -C₆Me₆) (2.8 mg, 7.0 µmol), B₂pin₂ (35.0 mg, 0.138 mmol), and *tert*-butyl ethyl ether (141 mg, 1.38 mmol) were placed into an NMR tube and sealed under vacuum. The tube was heated at 150 °C for 24 h. The NMR tube was opened, the contents were dissolved in methanol, and the resulting solution was rapidly stirred with solid KHF₂ (100 mg, 1.28 mmol) for 2 h. Filtration and evaporation of the solvent gave an oily solid. A solution of 1,3,5-trimethoxybenzene (19.8 mg, 0.118 mmol) in CD₃OD was added to this material, and comparison of the integrations of the ¹H NMR resonances of the product versus the integrations of the methyl resonance of 1,3,5-trimethoxybenzene indicated that the yield of the *tert*-butoxyalkyl trifluoroborate was 86% yield

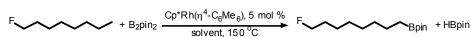
of the alkyltrifluoroborate based on the reaction of the ether and B_2pin_2 to form the functionalized product and HBpin. To obtain isolated material for spectral and analytical data, the borylation reaction was conducted with Cp*Rh(η^4 -C₆Me₆) (40 mg, 0.10 mmol), B₂pin₂ (520 mg, 2.05 mmol), and *tert*-butyl ethyl ether (2.10 g, 20 mmol). The borylated *tert*-butyl ethyl ether was purified by passing the mixture through a plug of silica gel, eluting with a 1:1 mixture of pentane and diethyl ether and was converted to the alkyltrifluoroborate by treatment of the oil resulting from evaporation of solvent with KHF₂ in methanol. Analytically pure alkoxyalkylfluoroborate (120 mg, 28%) was obtained by precipitating the product three times from MeCN with Et₂O. ¹H NMR (CD₃OD, 500 MHz) δ = 3.46 (m, 2H), 1.19 (s, 9H), 0.55 (broad m, 2H); ¹³C{¹H} NMR (CD₃OD, 126 MHz) δ = 73.9, 62.8, 28.2, the carbon attached to boron was not observed; ¹⁹F{¹H} NMR (CD₃OD, 376 MHz) δ = -143. Anal. Calcd for C₆H₁₃BF₃KO: C, 34.63; H, 6.30. Found: C, 34.29; H, 6.22, N < 0.02.

Potassium 2-(1-piperidino)ethyltrifluoroborate (Table 2, Entry 6). Cp*Rh(n⁴-C₆Me₆) (5.6 mg, 14 µmol), B₂pin₂ (70.0 mg, 0.276 mmol), and 1-ethylpiperidine (312 mg, 2.76 mmol) were placed into an NMR tube. The tube was placed into an NMR tube and sealed under vacuum. The tube was heated at 150 °C for 24 h. The NMR tube was opened, the contents were dissolved in methanol, and the resulting solution was rapidly stirred with solid KHF₂ (216 mg, 2.76 mmol) for 2 h. Filtration and evaporation of the solvent gave an oily solid. A solution of 1,3,5-trimethoxybenzene (38.9 mg, 0.231 mmol) in CD₃OD was added to this material, and comparison of the integrations of the ¹H NMR resonances of the product versus the integrations of the methyl resonance of 1,3,5-trimethoxybenzene indicated that the yield of the aminoalkyl trifluoroborate was 69% yield, based on the reaction of amine and B₂pin₂ to form two equiv of the functionalized product and H₂. To obtain isolated material for spectral and analytical data, the borylation reaction was conducted with Cp*Rh(η^4 -C₆Me₆) (78.8 mg, 0.197 mmol), B₂pin₂ (1.00 g, 3.94 mmol), and 1-ethylpiperidine (4.5 g, 39 mmol). The borylated product was isolated by treatment with PL-DEAM scavenger in benzene and distilled. The distillate was diluted with MeOH and converted to the fluoroborate by addition of solid KHF₂ (2.3 g, 29 mmol) for 2 h. An analytically pure sample (252 mg, 15%) of the product was obtained by precipitating the product three times from MeCN with Et₂O. ¹H NMR (CD₃OD, 500 MHz) δ = 3.51 (m, 2H), 3.04 (m, 2H), 2.80 (m, 2H), 1.95 (m, 2H), 1.83 (m, 1H), 1.71 (m, 2H), 1.51 (m, 1H), 0.61 (broad m, 2H); ¹¹B NMR (C₆D₆, 160 MHz) δ = 5.2; ¹³C{¹H} NMR (CD₃OD, 126 MHz) δ = 58.9, 53.5, 24.6,

23.1, the carbon attached to boron was not observed; ${}^{19}F{}^{1}H$ NMR (CD₃OD, 376 MHz) $\delta = -$ 142.8. Anal. Calcd for C₇H₁₄NBO₂: C, 38.37; H, 6.44; N, 6.39. Found: C, 38.61; H, 6.56; N, 6.20.

Table S1. Yields by GC of the borylation of fluorooctane as a function of the ratio of 1-fluorooctane:B₂pin₂:solvent.

solvent, 150 °C			
Solvent = cyclohexane	Yield(%)	Solvent = cyclooctane	Yield(%)
1-fluorooctane:B2pin2:cyclohexane		1-fluorooctane:B2pin2:cyclooctane	
10:1:10	55	10:1:10	61
1:1:3	39	1:1:3	32
1:2:3	40	1:2:3	18
10:1:0	83		
1:2:0	46		



References

(1) Maxim, N. Bull. Soc. Chim. Fr. **1927**, 41, 809; Maxim, N.; Mavrodineanu, R. Bull. Soc. Chim. Fr. **1935**, 2, 591.

- (2) Feng, Z.; Hellberg, M. *Tetrahedron Lett.* **2000**, *41*, 5813.
- (3) Bowyer, W. J.; Merkert, J. W.; Geiger, W. E. Organometallics 1989, 8, 191.
- (4) Herde, J. L.; Lambert, J. C.; Senoff, C. V. Inorg. Synth. 1974, 15, 18.

(5) Alexakis, A.; Rosset, S.; Allamand, J.; March, S.; Guillen, F.; Benhaim, C. *Synlett* **2001**, 1375.

- (6) Ukai, T.; Kawazura, H.; Ishii, Y. J. Organomet. Chem. 1974, 65, 253.
- (7) (a) Leitner, A.; Shu, C.; Hartwig, J. F. Proc. Natl. Acad. Sci. U. S. A. 2004, 101, 5830. (b)
- Kiener, C. A.; Shu, C.; Incarvito, C.; Hartwig, J. F. J. Am. Chem. Soc. 2003, 125, 14272. (c)

Ohmura, T.; Hartwig, J. F. J. Am. Chem. Soc. 2002, 124, 15164.

- (8) Wallace, R. H.; Zong, K. K. *Tetrahedron Lett.* **1992**, *33*, 6941.
- (9) Fang, G.-H.; Yan, Z.-J.; Deng, M.-Z. Org. Lett. 2004, 6, 357.