## Supporting Information for

Synthesis of a Tripeptide Derivative Containing the Gln-Arg Hydroxyethylene Dipeptide Isostere

Matthias Brewer, Clint A. James and Daniel H. Rich<br>Department of Chemistry and School of Pharmacy, University of Wisconsin-Madison, Madison, WI 53706

## General:

All reactions were carried out under an atmosphere of argon using flame-dried glassware. Tetrahydrofuran (THF) was distilled from sodium metal-benzophenone ketyl. Dichloromethane $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$, DMPU, diisopropylamine, and toluene were distilled from calcium hydride. All other solvents and reagents were used without further purification. $\mathrm{N}, \mathrm{N}$ '-bis-Boc-1-guanylpyrazole was purchased from Advanced ChemTech. N-(9-Fluorenylmethoxycarbonyloxy)succinimide was purchased from Chem-Impex. All other reagents were purchased from Aldrich.

Flash column chromatography was performed using Merck grade 60 silica gel (230-400 mesh). ${ }^{1}$ H NMR spectra were taken on a Bruker AC 300 or a Bruker AC 250 spectrometer in $\mathrm{CDCl}_{3}$ at ambient temperature unless otherwise noted. Chemical shifts were reported in ppm ( $\delta$ units) downfield from tetramethylsilane. Mass spectra were taken on a Micromass AutoSpec magnetic sector mass spectrometer using 3-nitrobenzyl alcohol as the matrix.

## 2-tert-Butoxycarbonylamino-4-(trityl-carbamoyl)-butyric acid methyl ester (7).

 To a solution of BocGln(Trt)-OH (10.13g, 20.76 mmol$)$ in toluene:methanol (7:1, 300 mL ) was added a solution of $\mathrm{TMSCHN}_{2}(12.5 \mathrm{~mL}, 2.0 \mathrm{~mol} / \mathrm{L})$. The mixture was allowed to stir at rt until the evolution of $\mathrm{N}_{2}$ ceased (ca. 6 h ). The solvents were removed in vacuo to provide BocGln(Trt)-Ome (7) in quantitative yield. m.p. $153-154{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 7.12-7.40(\mathrm{~m}, 16 \mathrm{H}), 5.28(\mathrm{~d}, \mathrm{~J}=7.7,1 \mathrm{H}), 4.29$ (td, J = 8.1, 3.3, 1 H ), $3.70(\mathrm{~s}, 3 \mathrm{H}), 2.27-2.46(\mathrm{~m}, 2 \mathrm{H}), 2.07-2.22(\mathrm{~m}, 1 \mathrm{H}), 1.77-1.95(\mathrm{~m}, 1$ H), 1.43 ( $\mathrm{s}, 9 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR (75.4 MHz, $\mathrm{CDCl}_{3}$ ) $\delta: 172.74,170.88,155.84,144.60,128.67$, $127.83,126.88,80.06,70.54,52.97,52.35,33.55,29.02,28.22$; EI calculated for $\mathrm{C}_{30} \mathrm{H}_{34} \mathrm{~N}_{2} \mathrm{O}_{5}$ : 502.2468. Found 502.2468.

## [3-tert-Butoxycarbonylamino-2-oxo-5-(trityl-carbamoyl)-pentyl]-phosphonic acid dimethyl

 ester (4).

To a solution of dimethylmethylphosphonate ( $12.94 \mathrm{ml}, 119.4 \mathrm{mmol}$ ) in THF ( 120 mL ) at $-78^{\circ} \mathrm{C}$ was added nBuLi in hexane $(49.34 \mathrm{~mL}, 119.4$ stir at this temperature for 45 min at which point a solution of ester $7(2.481 \mathrm{~g}, 4.973 \mathrm{mmol})$ in THF ( 100 mL ) was added. The reaction was allowed to stir at $-78^{\circ} \mathrm{C}$ for 1 h , the cooling bath was warmed to $-30^{\circ} \mathrm{C}$, the mixture stirred for an additional hour and then quenched by dropwise addition of glacial acetic acid $(0.56 \mathrm{ml})$. The mixture was partitioned between ethyl acetate and saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}$, the layers were separated and the aqueous was extracted twice more with ethyl acetate. The organics were combined, washed with water, brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$,
and concentrated to ca. 5\% volume. Petroleum ether was added in a smooth stream with efficient stirring to provide a white solid which was stirred overnight at room temperature and then isolated by filtration to provide $10.26 \mathrm{~g}(87 \%)$ of phosphonate 4. m.p. $183-184{ }^{\circ} \mathrm{C} ;\left[\alpha_{\mathrm{D}}\right]-21.89$ (c 1.0, MeOH); $\mathrm{TLC} \mathrm{R}_{\mathrm{f}}=0.22(\mathrm{EtOAc}) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 7.18-7.32(\mathrm{~m}, 16 \mathrm{H})$, $5.56(\mathrm{~d}, \mathrm{~J}=7.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.31(\mathrm{~m}, 1 \mathrm{H}), 3.74(\mathrm{~d}, \mathrm{~J}=11.4,3 \mathrm{H}), 3.72(\mathrm{~d}, \mathrm{~J}=11.4,3 \mathrm{H}), 3.24(\mathrm{dd}$, $\mathrm{J}=22.3,14.5,1 \mathrm{H}), 3.05(\mathrm{dd}, \mathrm{J}=21.9,14.4,1 \mathrm{H}), 2.28-2.46(\mathrm{~m}, 2 \mathrm{H}), 2.16-2.26(\mathrm{~m}, 1 \mathrm{H}), 1.76$ - $1.86(\mathrm{~m}, 1 \mathrm{H}), 1.43(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}(75.4 \mathrm{MHz}) \delta: 201.13,171.24,155.96,144.84,128.89$, 128.04, 127.09, 80.36, 70.72, 60.10, 53.26, $37.86(\mathrm{~d}, \mathrm{~J}=128 \mathrm{~Hz}), 33.24,28.47$, 27.16; ESI calculated for $\left[\mathrm{C}_{32} \mathrm{H}_{39} \mathrm{~N}_{2} \mathrm{O}_{7} \mathrm{PNa}\right]^{+}$: 617.2393 . Found 617.2386.

## 5-(tert-Butyl-dimethyl-silanyloxy)-pentanoic acid methyl ester (9).



A solution of $\delta$-valerolactone ( $16.19 \mathrm{~g}, 161.7 \mathrm{mmol}$ ) containing concentrated $\mathrm{H}_{2} \mathrm{SO}_{4}$ (11 drops) was heated at reflux in anhydrous MeOH (350 mL) for $5 \mathrm{~h} .\{$ Huckstep, 1982 \#257\} The mixture cooled to rt and then in an ice-salt bath, $\mathrm{NaHCO}_{3}(1.85 \mathrm{~g})$ was added, the mixture was allowed to stir for 10 min and was then placed in the freezer for 2 h . The cold mixture was filtered and the solvent was removed in vacuo, without heating, and then under high vacuum. The residue was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 250 mL ), cooled to $0{ }^{\circ} \mathrm{C}$, and $\operatorname{TBSCl}(28.5 \mathrm{~g}, 189 \mathrm{mmol})$ and imidazole ( $29 \mathrm{~g}, 426 \mathrm{mmol}$ ) were added. The reaction mixture was allowed to stir while warming to rt over 14 h . The mixture was diluted with $\mathrm{H}_{2} \mathrm{O}$ and the layers were separated. The organic layer was washed with water, twice with saturated aqueous $\mathrm{CuSO}_{4}$, brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The residue was purified by flash column chromatography ( $10: 1$ petroleum ether : diethyl ether) to yield 30.41 g (76\%) of title compound 9. b.p. $87-90^{\circ} \mathrm{C} @ 1.0 \mathrm{mmHg} .{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 3.67$
$(\mathrm{s}, 3 \mathrm{H}), 3.62(\mathrm{t}, \mathrm{J}=6.2,2 \mathrm{H}), 2.34(\mathrm{t}, \mathrm{J}=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 1.63-1.72(\mathrm{~m}, 2 \mathrm{H}), 1.51-1.57(\mathrm{~m}, 2$
$\mathrm{H}), 0.89(\mathrm{~s}, \quad 9 \quad \mathrm{H}), \quad 0.05(\mathrm{~s}, \quad 6 \quad \mathrm{H}) .{ }^{13} \mathrm{C} \quad \mathrm{NMR} \quad\left(100 \quad \mathrm{MHz}, \quad \mathrm{CDCl}_{3}\right)$
$\delta: 174.07,62.61,51.38,33.77,32.13,25.90,21.42,18.28,-5.38 ;$ ESI calculated for $\left[\mathrm{C}_{12} \mathrm{H}_{26} \mathrm{O}_{3} \mathrm{SiNa}\right]^{+}: 269.1549$. Found 269.1556.

## 5-(tert-Butyl-dimethyl-silanyloxy)-2-hydroxy-pentanoic acid methyl ester (10).



To a solution of KHMDS ( $6.39 \mathrm{~g}, 20.3 \mathrm{mmol}$ ) in THF $(250 \mathrm{~mL})$ at -78
${ }^{\circ} \mathrm{C}$ was added a solution of ester $9(5.00 \mathrm{~g}, 20.3 \mathrm{mmol})$ in THF ( 50 mL ) in a dropwise manner via cannula. The solution was stirred for 20 min at this temperature and Davis oxaziridine ( $6.45 \mathrm{~g}, 24.7 \mathrm{mmol}$ ) in THF ( 50 ml ) was added in a dropwise manner via cannula. After 20 min the reaction was quenched with sat $\mathrm{NH}_{4} \mathrm{Cl}(50 \mathrm{~mL})$, the cooling bath was removed, and the mixture was allowed warm to room temperature. The solvent was removed in vacuo, the residue was dissolved in ethyl acetate, washed with water and brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The residue was suspended in chloroform ( 100 ml ) and a white precipitate was removed by filtration and discarded. To the mother liquor was added $\mathrm{tBuNH}_{2}(3$ $\mathrm{ml})$ and the mixture was allowed to stand 5 min . The solvent was removed in vacuo, the residue was suspended in a 5 to 1 mixture of hexanes and ethyl acetate $(30 \mathrm{ml})$, and the solution was refrigerated overnight. The resulting precipitate was removed by filtration and discarded, the mother liquor was concentrated, and the residue purified by flash column chromatography (80:20 hexane : ethyl acetate) to afford $3.79 \mathrm{~g}(71 \%)$ of the title compound as a light yellow liquid. TLC $\mathrm{R}_{\mathrm{f}}=0.43$ (80:20 Hexane:EtOAc); ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 4.24(\mathrm{dt}, \mathrm{J}=6.3,5.2,1 \mathrm{H})$, $3.78(\mathrm{~s}, 3 \mathrm{H}), 3.66(\mathrm{t}, \mathrm{J}=5.7,2 \mathrm{H}), 3.30(\mathrm{~d}, \mathrm{~J}=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.91(\mathrm{dddd}, \mathrm{J}=14.0,9.1,6.6,4.3,1$
H), $1.56-1.82(\mathrm{~m}, 3 \mathrm{H}), 0.90(\mathrm{~s}, 9 \mathrm{H}), 0.06(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 62.9 MHz$) \delta: 175.36,70.34$, $62.73,52.22,31.30,28.12,25.78,18.19,-5.49$.

## 5-(tert-Butyl-dimethyl-silanyloxy)-2-oxo-pentanoic acid methyl ester (11).



To a solution of Dess-Martin periodinane ( $8.67 \mathrm{~g}, 20.44 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(100 \mathrm{~mL})$ at rt was added $\mathrm{tBuOH}(1.54 \mathrm{~g}, 20.8 \mathrm{mmol})$ and the solution was stirred for 20 min . A solution of alcohol $\mathbf{1 0}(4.123 \mathrm{~g}, 15.71 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(17$ mL ) was added dropwise via cannula. The reaction mixture was allowed to stir for 15 min and was then poured into an efficiently stirring aqueous solution of $\mathrm{NaHCO}_{3}(200 \mathrm{~mL}, 1 \mathrm{~mol} / \mathrm{L})$ containing $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(36.3 \mathrm{~g}, 146.3 \mathrm{mmol})$. After stirring for 30 min the layers were separated and the aqueous phase was extracted twice more with dichloromethane. The combined organic layers were washed with saturated aqueous $\mathrm{NaHCO}_{3}$, water, brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The residue was purified by flash column chromatography (8:1 hexane : diethyl ether) to afforded $2.65 \mathrm{~g}(65 \%)$ of title compound $\mathbf{1 1}$ as a liquid. ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ : 3.87 (s, 3 H ), $3.65(\mathrm{t}, \mathrm{J}=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.92(\mathrm{t}, \mathrm{J}=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 1.88(\mathrm{tt}, \mathrm{J}=7.0,6.0,2 \mathrm{H}), 0.88$ (s, 9 H$), 0.03(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75.4 \mathrm{MHz}\right) \delta: 194.01,161.40,61.76,52.76,36.01$, 26.50, 25.87, 18.26, -5.41.

## 5-tert-Butoxycarbonylamino-2-[3-(tert-butyl-dimethyl-silanyloxy)-propyl]-4-oxo-7-(trityl-

 carbamoyl)-hept-2-enoic acid methyl ester (12). To a suspension of $\mathrm{NaH}(1.65 \mathrm{~g}, 41.2 \mathrm{mmol}, 65 \%$ emulsion $)$ in THF (96 ml) at $0{ }^{\circ} \mathrm{C}$ was added a $0{ }^{\circ} \mathrm{C}$ solution of ketophosphonate 4 ( $7.91 \mathrm{~g}, 13.3 \mathrm{mmol}$ ) in THF ( 260 mL ). The
mixture was stirred at $0^{\circ} \mathrm{C}$ for 1 h , cooled to $-78^{\circ} \mathrm{C}$, and a solution of ketoester $\mathbf{1 1}(4.50 \mathrm{~g}, 17.3$ $\mathrm{mmol})$ in THF ( 26 mL ) was added. Stirring was continued at $-78^{\circ} \mathrm{C}$ for 0.5 h at which point the reaction was transferred to a $-30^{\circ} \mathrm{C}$ bath and kept at this temperature for 14 h . The reaction was quenched over an hour-long period by the slow addition of a solution of glacial acidic acid ( 12.53 ml ) in THF ( 25 ml ), and then warmed to room temperature and concentrated. The residue was partitioned between ethyl acetate and water and the layers were separated. The aqueous phase was extracted twice more with ethyl acetate and the organics were combined, washed with water, saturated aqueous $\mathrm{NaHCO}_{3}$, brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated to an oil that solidified upon standing. Purification by column chromatography (3:1 hexane : ethyl acetate) afforded title compound $\mathbf{1 2}$ as a colorless foam ( $7.79 \mathrm{~g}, 80 \%$ ). TLC $\mathrm{R}_{\mathrm{f}}=0.25$ (3:1 Hexane:EtOAc); ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 7.63$ (s, 1 H ), $7.19-7.32(\mathrm{~m}, 15 \mathrm{H}), 6.03$ ( $\mathrm{s}, 1$ H), $5.46(\mathrm{~d}, \mathrm{~J}=7.3,1 \mathrm{H}), 4.29(\mathrm{t}, \mathrm{J}=7.6,1 \mathrm{H}), 3.77(\mathrm{~s}, 3 \mathrm{H}), 3.62(\mathrm{t}, \mathrm{J}=6.1,2 \mathrm{H}), 2.36-2.52$ $(\mathrm{m}, 3 \mathrm{H}), 2.31(\mathrm{dt}, \mathrm{J}=14.2,5.4,1 \mathrm{H}), 2.14-2.23(\mathrm{~m}, 1 \mathrm{H}), 1.67($ pent, $\mathrm{J}=6.4,2 \mathrm{H}), 1.52-1.60$ $(\mathrm{m}, 1 \mathrm{H}), 1.41(\mathrm{~s}, 9 \mathrm{H}), 0.90(\mathrm{~s}, 9 \mathrm{H}), 0.05(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75.4 MHz) $\delta: 196.6,171.2$, $169.2,156.0,149.6,144.7,128.6,127.6,126.6,124.5,79.7,70.3,61.8,58.3,52.0,33.2,30.9$, 30.2, 28.5, 28.1, 25.8, 18.1, -5.5; ESI calculated for $\left[\mathrm{C}_{42} \mathrm{H}_{56} \mathrm{~N}_{2} \mathrm{O}_{7} \mathrm{SiNa}\right]^{+}$: 751.3755. Found 751.3779.
[1-\{4-[3-(tert-Butyl-dimethyl-silanyloxy)-propyl]-5-oxo-2,5-dihydro-furan-2-yl\}-3-(trityl-carbamoyl)-propyl]-carbamic acid tert-butyl ester (13a, 13b).

Procedure 1. To a solution of $\alpha, \beta$-unsaturated ester 12 ( 1.051 g ,
 1.415 mmol ) in $\mathrm{MeOH}(50 \mathrm{~mL})$ which was cooled to $-30^{\circ} \mathrm{C}$ by
means of a dry ice acetone bath, was added $\mathrm{NaBH}_{4}(54.3 \mathrm{mg}, 1.453 \mathrm{mmol})$. The solution was allowed to stir for 30 min during which time the temperature rose to $-20^{\circ} \mathrm{C}$. The solvent was removed on the rotary evaporator and the residue was purified by column chromatography ( $1 \%$ MeOH in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give the title compounds as a colorless foam ( $730.5 \mathrm{mg}, 74 \%$ ), as an inseparable mixture of diastereomers in a 2:3 ratio (by ${ }^{1} \mathrm{H} N \mathrm{NR}$ ) favoring the 4 S isomer $\mathbf{1 3 b}$. The mixture of diastereomers was carried on without further purification. $\mathrm{TLC}_{\mathrm{f}}=0.19$ (7:3 Hexane:EtOAc); EI calculated for $\left[\mathrm{C}_{41} \mathrm{H}_{54} \mathrm{~N}_{2} \mathrm{O}_{6} \mathrm{Si}^{+}\right]^{6}$ 69.3751. Found 698.3732.

Procedure 2. To a $-30^{\circ} \mathrm{C}$ solution of $\alpha, \beta$-unsaturated ester 12 ( $5.33 \mathrm{~g}, 7.32 \mathrm{mmol}$ ) in MeOH ( 275 mL ) was added $\mathrm{NaBH}_{4}(285 \mathrm{mg}, 7.54 \mathrm{mmol})$. After 30 min at this temperature the reaction was quenched with saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}$, and the solvent was removed in vacuo. The residue was partitioned between ethyl acetate and water, the layers were separated and the aqueous was extracted twice more with ethyl acetate. The combined organics were washed with water, brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated to give the title compound as an inseparable mixture of diastereomers ( $5.16 \mathrm{~g}, 100 \%$ ), in a $2: 1$ ratio (by ${ }^{1} \mathrm{H}$ NMR) favoring the 4 R isomer 13a. The colorless foam was carried on without further purification.

## [1-\{4-[3-(tert-Butyl-dimethyl-silanyloxy)-propyl]-5-oxo-tetrahydro-furan-2-yl\}-3-(trityl-

 carbamoyl)-propyl]-carbamic acid tert-butyl ester (15a, 15b). To a stirred solution of the diastereomeric mixture of olefins 13a and $\mathbf{1 3 b}(5.032 \mathrm{~g}, 7.20 \mathrm{mmol})$ formed by reduction procedure 2 , in ethanol ( 250 ml ) under a nitrogen atmosphere was added $\mathrm{Pt}(\mathrm{IV})$
oxide $(0.20 \mathrm{~g})$. The reaction vessel was evacuated and purged with nitrogen three times, and then
placed under an atmosphere of hydrogen using a balloon until TLC showed complete consumption of starting material (ca. 3 h ). At this time the hydrogen gas was evacuated, the catalyst was removed by filtration, and the solvent was removed in vacuo. Separation of the resulting diastereomers proved difficult but was achieved by several rounds of careful flash column chromatography (1:1:0.5 petroleum ether : diethyl ether : dichloromethane) to provide $3.14 \mathrm{~g}(62 \%)$ of [(1S)-1-\{(4R)-4-[3-(tert-Butyl-dimethyl-silanyloxy)-propyl]-5-oxo-tetrahydro-furan-(2R)-2-yl\}-3-(trityl-carbamoyl)-propyl]-carbamic acid tert-butyl ester 15a along with 1.52 g (30\%) of [(1S)-1-\{(4S)-4-[3-(tert-Butyl-dimethyl-silanyloxy)-propyl]-5-oxo-tetrahydro-furan-(2S)-2-yl\}-3-(trityl-carbamoyl)-propyl]-carbamic acid tert-butyl ester 15b.

2R-lactone 15a: TLC $\mathrm{R}_{\mathrm{f}}=0.20$ (1:1:0.5 Petroleum Ether: $\mathrm{Et}_{2} \mathrm{O}: \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) ${ }^{1} \mathrm{H}$ NMR ( 500 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta: 7.18-7.30(\mathrm{~m}, 15 \mathrm{H}), 7.07(\mathrm{~s}, 1 \mathrm{H}), 4.99(\mathrm{~d}, \mathrm{~J}=9.2,1 \mathrm{H}), 4.18-4.25(\mathrm{~m}, 1 \mathrm{H}), 3.69$ $-3.59(\mathrm{~m}, 1 \mathrm{H}), 3.62(\mathrm{t}, \mathrm{J}=6.2,2 \mathrm{H}), 2.49-2.60(\mathrm{~m}, 1 \mathrm{H}), 2.25(\mathrm{t}, \mathrm{J}=7.0,2 \mathrm{H}) 2.25-2.38(\mathrm{~m}$, $1 \mathrm{H}), 1.85-2.02(\mathrm{~m}, 2 \mathrm{H}), 1.52-1.69(\mathrm{~m}, 4 \mathrm{H}), 1.41-1.51(\mathrm{~m}, 1 \mathrm{H}), 1.43(\mathrm{~s}, 9 \mathrm{H}), 0.89(\mathrm{~s}, 9$ H), $0.05(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 178.35,171.64,156.20,144.78,128.82$, $128.00,127.0780 .24,79.91,70.67,62.83,53.04,40.49,33.70,31.86,30.52,28.47,26.96,26.36$, 26.09, 18.43, -5.19; ESI calculated for $\left[\mathrm{C}_{41} \mathrm{H}_{56} \mathrm{~N}_{2} \mathrm{O}_{6} \mathrm{SiNa}\right]^{+}$: 723.3805. Found 723.3818.

2S-lactone 15b: TLC $\mathrm{R}_{\mathrm{f}}=0.26$ (1:1:0.5 Petroleum Ether: $\mathrm{Et}_{2} \mathrm{O}: \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ); ${ }^{1} \mathrm{H}$ NMR for the S isomer $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 7.16-7.29(\mathrm{~m}, 16 \mathrm{H}), 4.79(\mathrm{~d}, \mathrm{~J}=9.8,1 \mathrm{H}), 4.25(\mathrm{brt} \mathrm{t}, \mathrm{J}=6.6,1$ H), $3.68(\mathrm{t}, \mathrm{J}=10.1,1 \mathrm{H}), 3.62(\mathrm{t}, \mathrm{J}=6.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.48-2.62(\mathrm{~m}, 1 \mathrm{H}), 2.30(\mathrm{t}, \mathrm{J}=6.5 \mathrm{~Hz}, 2 \mathrm{H})$, $2.19-2.34(\mathrm{~m}, 1 \mathrm{H}), 1.80-1.96(\mathrm{~m}, 2 \mathrm{H}), 1.66-1.79(\mathrm{~m}, 1 \mathrm{H}), 1.52-1.65(\mathrm{~m}, 3 \mathrm{H}), 1.42(\mathrm{~s}, 10$ H), $0.89(\mathrm{~s}, \quad 9 \quad \mathrm{H}), \quad 0.05 \quad(\mathrm{~s}, \quad 6 \quad \mathrm{H}) ; \quad{ }^{13} \mathrm{C} \quad \mathrm{NMR} \quad(125 \quad \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta: 178.46,171.26,156.48,144.75,128.74,127.83,126.87,79.91,79.82,70.48,62.74,53$ for $\left[\mathrm{C}_{41} \mathrm{H}_{56} \mathrm{~N}_{2} \mathrm{O}_{6} \mathrm{SiNa}\right]^{+}: 723.3805$. Found 723.3826.

## [(1S)-1-[(4R)-4-(3-Azido-propyl)-5-oxo-tetrahydro-furan-(2R)-2-yl]-3-(trityl-carbamoyl)-propyl]-carbamic acid tert-butyl ester (17).



Step 1: To a solution of 2R-lactone 15a (3.21 g, 4.57 mmol ) in methanol ( 25 ml ) was added 25 ml of a mixture composed of acetonitrile ( $70 \%$ ), water ( $30 \%$ ) and TFA ( $0.1 \%$ ). The reaction was stirred at room temperature for 1 h and concentrated to remove most of the organics. The remainder was partitioned between ethyl acetate and saturated aqueous $\mathrm{NaHCO}_{3}$, the layers were separated and the aqueous was extracted again with ethyl acetate. The organics were combined, washed with water and brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated to provide a foam.

Step 2: The foam from step 1 was dissolved in anhydrous toluene ( 30 ml ) and cooled to $0{ }^{\circ} \mathrm{C}$. Diisopropyl ethylamine ( $1.04 \mathrm{ml}, 5.95 \mathrm{mmol}$ ) was added and a waxy solid formed. Methane sulfonyl chloride ( $0.459 \mathrm{ml}, 5.95 \mathrm{mmol}$ ) was added and the hazy solution was warmed to room temperature. After a period of 0.5 h an additional 0.5 ml of methane sulfonyl chloride and 0.4 ml of diisopropyl ethylamine were added and the reaction was stirred for 10 min more. To this mixture was added an aqueous mixture ( 15 ml ) of sodium azide ( $2.38 \mathrm{~g}, 36.6 \mathrm{mmol}$ ) and tetrabutylammonium bromide $(0.147 \mathrm{~g}, 0.458 \mathrm{mmol})$ and the reaction was warmed to reflux for 3 h and then cooled to room temperature. The mixture was diluted with diethyl ether and the layers were separated. The aqueous was extracted twice more with diethyl ether and the organics were combined and washed with water, brine, dried over $\mathrm{MgSO}_{4}$, and concentrated. The residue was
purified by flash column chromatography ( $85: 15, \mathrm{CH}_{2} \mathrm{Cl}_{2}: \mathrm{Et}_{2} \mathrm{O}$ ) to provide 2.61 g ( $93 \%$ over 2 steps) of azide 17 as a colorless foam. $\mathrm{TLC}_{\mathrm{f}}=0.20\left(85: 15 \mathrm{CH}_{2} \mathrm{Cl}_{2}: \mathrm{Et}_{2} \mathrm{O}\right) ;{ }^{1} \mathrm{H}$ NMR ( 500 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta: 7.19-7.32(\mathrm{~m}, 15 \mathrm{H}), 6.99(\mathrm{~s}, 1 \mathrm{H}), 4.91(\mathrm{~d}, \mathrm{~J}=8.8,1 \mathrm{H}), 4.24-4.30(\mathrm{~m}, 1 \mathrm{H}), 3.64$ $-3.72(\mathrm{~m}, 1 \mathrm{H}), 3.32(\mathrm{t}, \mathrm{J}=6.8,2 \mathrm{H}), 2.52-2.61(\mathrm{~m}, 1 \mathrm{H}), 2.39(\mathrm{t}, \mathrm{J}=6.8,2 \mathrm{H}), 2.30-2.41(\mathrm{~m}$, $1 \mathrm{H}), 1.98-2.05(\mathrm{~m}, 1 \mathrm{H}), 1.88-1.96(\mathrm{~m}, 1 \mathrm{H}), 1.56-1.76(\mathrm{~m}, 4 \mathrm{H}), 1.45-1.55(\mathrm{~m}, 1 \mathrm{H}), 1.44$ (s, 9 H ); ${ }^{13} \mathrm{C}$ NMR (125 MHz, $\mathrm{CDCl}_{3}$ ) $\delta: 177.84,171.74,156.04,144.45,128.48,127.52$, $126.58,79.67,79.19,70.14,52.84,50.75,39.69,33.18,31.22,28.12,27.10,26.76,26.27,25.53$; ESI calculated for $\left[\mathrm{C}_{35} \mathrm{H}_{41} \mathrm{~N}_{5} \mathrm{O}_{5} \mathrm{Na}\right]^{+}$: 634.3005. Found 634.3002.

## [1-[4-(3-Azido-propyl)-5-oxo-tetrahydro-furan-2-yl]-3-(trityl-carbamoyl)-propyl]-carbamic acid benzyl ester (18).



Step 1: To a rt solution of lactone $17(0.50 \mathrm{~g}, 0.817 \mathrm{mmol})$ in anhydrous dichloromethane ( 10 mL ) containing 2,6-lutidine ( 0.190 $\mathrm{mL}, 1.63 \mathrm{mmol}$ ) was added TBSOTf ( $0.282 \mathrm{~mL}, 1.23 \mathrm{mmol}$ ). After 1 h the reaction mixture was quenched with saturated aqueous
$\mathrm{NH}_{4} \mathrm{Cl}$ and extracted three times with diethyl ether. The combined organic layers were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated.

Step 2: The residue from Step 1 was dissolved in THF ( 25 ml ), benzyl bromide ( $0.292 \mathrm{ml}, 2.45$ mmol ) was added and the mixture cooled to $0^{\circ} \mathrm{C}$. To this was added a 1 M solution of tetrabutyl ammonium fluoride in THF ( $0.98 \mathrm{ml}, 0.98 \mathrm{mmol}$ ) and after 2 h the reaction was quenched with saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ and extracted three times with diethyl ether. The organics were combined and washed with saturated aqueous $\mathrm{CuSO}_{4}$, twice with water, brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$
and concentrated. The residue was purified by column chromatography (60:40 hexane : ethyl acetate) to afford $0.379 \mathrm{~g}(72 \%)$ of the title compound $\mathbf{1 8}$ as a colorless foam. ${ }^{1} \mathrm{H}$ NMR ( 250 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 7.12-7.33(\mathrm{~m}, 20 \mathrm{H}), 6.79(\mathrm{~s}, 1 \mathrm{H}), 5.28(\mathrm{~d}, \mathrm{~J}=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.12(\mathrm{~d}, \mathrm{~J}=12.5$ $\mathrm{Hz}, 1 \mathrm{H}), 5.04(\mathrm{~d}, \mathrm{~J}=12.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.21(\mathrm{~m}, 1 \mathrm{H}), 3.70(\mathrm{~m}, 1 \mathrm{H}), 3.29(\mathrm{t}, \mathrm{J}=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.45-$ $2.55(\mathrm{~m}, 1 \mathrm{H}), 2.38(\mathrm{t}, \mathrm{J}=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.23-2.31(\mathrm{~m}, 1 \mathrm{H}), 1.73-2.03(\mathrm{~m}, 2 \mathrm{H}), 1.44-1.69(\mathrm{~m}$, 5H); ${ }^{13} \mathrm{C}$ NMR ( 62.9 MHz ) $\delta: 177.5,171.5,156.5,144.37,136.2,128.5,128.4,128.3,128.1$, $128.0,127.8,126.9,79.76,70.43,66.78,53.62,50.94,39.92,33.20,31.41,27.23,26.47,25.86$.

## [1-[1-[4-(3-Azido-propyl)-5-oxo-tetrahydro-furan-2-yl]-3-(trityl-carbamoyl)-

 propylcarbamoyl]-2-(trityl-carbamoyl)-ethyl]-carbamic acid benzyl ester (20).

Step 1: To a rt solution of Boc-lactone $17(1.33 \mathrm{~g}, 2.24$ mmol ) in anhydrous dichloromethane ( 30 mL ) containing 2,6-lutidine ( $0.522 \mathrm{~mL}, 4.48 \mathrm{mmol}$ ) was added TBSOTf $(0.772 \mathrm{~mL}, 3.36 \mathrm{mmol})$. After 1 h the reaction mixture was cooled to $0^{\circ} \mathrm{C}$ and a 1 M solution of tetrabutyl ammonium fluoride in THF ( $4.7 \mathrm{ml}, 4.7 \mathrm{mmol}$ ) was carefully added. The reaction was warmed to room temperature, and after 1 h quenched with saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ and extracted three times with ethyl acetate. The combined organic layers were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated to an oil.

Step 2: The residue from Step 1 was dissolved in DMF ( 25 ml ) and added to a $0^{\circ} \mathrm{C}$ mixture of Cbz-Asn(Trt)-OH (1.37 g, 2.69 mmol$)$, HOBt ( $0.535 \mathrm{~g}, 3.96 \mathrm{mmol}$ ), and EDCI ( $0.566 \mathrm{~g}, 2.96$ mmol ) in DMF ( 25 ml ). The reaction was allowed to warm to room temperature over 14 h , at which point the solvents were removed in vacuo. The residue was partitioned between ethyl acetate and aqueous $10 \%$ citric acid, the layers were separated and the aqueous extracted twice
more with ethyl acetate. The organics were combined and washed with water, saturated aqueous $\mathrm{NaHCO}_{3}$, and brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The residue was purified by flash column chromatography (80:18:2 dichloromethane: diethyl ether: methanol) to provide 2.11 g ( $94 \%$ ) of title compound 20. $\mathrm{TLC} \mathrm{R}_{\mathrm{f}}=0.34\left(80: 18: 2 \mathrm{CH}_{2} \mathrm{Cl}_{2}: \mathrm{Et}_{2} \mathrm{O}: \mathrm{MeOH}\right) ;{ }^{1} \mathrm{H}$ NMR ( 500 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta: 7.10-7.32(\mathrm{~m}, 35 \mathrm{H}), 7.00(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 6.95(\mathrm{~d}, \mathrm{~J}=8.3,1 \mathrm{H}), 6.76(\mathrm{~s}, 1 \mathrm{H}), 6.13(\mathrm{~s}, 1$ H), $5.04(\mathrm{~d}, \mathrm{~J}=12.6,1 \mathrm{H}), 4.96(\mathrm{~d}, \mathrm{~J}=12.3,1 \mathrm{H}), 4.38-4.44(\mathrm{br} \mathrm{m}, 1 \mathrm{H}), 3.98-4.06(\mathrm{br} \mathrm{m}, 1 \mathrm{H})$, 3.86-3.94 (br m, 1 H ), $3.213(\mathrm{t}, \mathrm{J}=6.8,2 \mathrm{H}), 2.97(\mathrm{~d}, \mathrm{~J}=13.7,1 \mathrm{H}), 2.59(\mathrm{dd}, \mathrm{J}=15.6,4.87,1$ H), 2.35-2.44 (br m, 1 H$), 1.99-2.20(\mathrm{~m}, 3 \mathrm{H}), 1.75-1.96(\mathrm{~m}, 3 \mathrm{H}), 1.35-1.66(\mathrm{~m}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 178.03,171.88,171.55,170.12,156.42,144.74,144.45,136.19,128.84$, 128.83, 128.66, 128.34, 128.18, 128.08, 127.97, 127.21, 127.02, 79.87, 70.90, 70.58, 67.28, $52.07,51.95,51.21,40.14,38.64,32.96,31.62,27.42,26.71,25.89$; ESI calculated for $\left[\mathrm{C}_{61} \mathrm{H}_{59} \mathrm{~N}_{7} \mathrm{O}_{7} \mathrm{Na}\right]^{+}: 1024.4374$. Found 1024.4370.

## (2R)-2-(3-Azido-propyl)-(5S)-5-[(2S)-2-benzyloxycarbonylamino-3-(trityl-carbamoyl)-propionylamino]-(4R)-4-(tert-butyl-dimethyl-silanyloxy)-7-(trityl-carbamoyl)-heptanoic acid (21).



Step 1: To a rt solution of lactone $20(2.06 \mathrm{~g}, 2.05 \mathrm{mmol})$ in dioxane ( 60 ml ) was added a 1 M aqueous solution of LiOH ( $12.31 \mathrm{ml}, 12.3 \mathrm{mmol}$ ). After 20 min the reaction was partitioned between a $0{ }^{\circ} \mathrm{C}$ mixture of ethyl acetate and ethanol (8:2) and saturated aqueous $\mathrm{NaHCO}_{3}$ and brine (1:1). The layers were separated and the aqueous was extracted twice more with fresh $0{ }^{\circ} \mathrm{C}$ organic. The organic layers were combined,
washed with $0{ }^{\circ} \mathrm{C}$ brine, dried over a $1: 1$ mixture of $\mathrm{MgSO}_{4}$ and $\mathrm{K}_{2} \mathrm{CO}_{3}$, and concentrated in vacuo without heating, and then thoroughly dried under high vacuum.

Step 2: To a $0{ }^{\circ} \mathrm{C}$ mixture of the residue from Step 1 and 2,6-lutidine ( $1.91 \mathrm{ml}, 16.4 \mathrm{mmol}$ ) in anhydrous dichloromethane ( 50 ml ) was added TBSOTf ( $1.88 \mathrm{ml}, 8.2 \mathrm{mmol}$ ). After 1 h methanol ( 20 ml ) was added, the reaction was warmed to room temperature and the solvents were removed in vacuo. The residue was dissolved in methanol $(50 \mathrm{ml}), \mathrm{K}_{2} \mathrm{CO}_{3}(0.34 \mathrm{~g})$ was added and the mixture was stirred overnight. The reaction was concentrated, the residue partitioned between ethyl acetate and aqueous $10 \%$ citric acid, the layers separated, and the aqueous extracted twice more with ethyl acetate. The organics were combined, washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The residue was purified by flash column chromatography (dichloromethane followed by $88: 18: 2: 0.1$ dichloromethane : diethyl ether : methanol : acetic acid) to provide 1.85 g ( $80 \%$ over two steps) of the title compound 21 as a foam. TLC $\mathrm{R}_{\mathrm{f}}=0.22$ (95:5 $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}: \mathrm{MeOH}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 7.09-7.36(\mathrm{~m}, 36 \mathrm{H}), 6.74(\mathrm{~d}, \mathrm{~J}=10.2,1 \mathrm{H})$, $6.71(\mathrm{~s}, 1 \mathrm{H}), 6.17(\mathrm{~d}, \mathrm{~J}=9.0,1 \mathrm{H}), 5.06(\mathrm{~d}, \mathrm{~J}=12.2,1 \mathrm{H}), 4.98(\mathrm{~d}, \mathrm{~J}=12.2,1 \mathrm{H}), 4.57(\mathrm{dt}, 9.0$, $4.4,1 \mathrm{H}), 4.09(\mathrm{brt}, \mathrm{J}=10.5,1 \mathrm{H}), 3.64-3.75(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 3.35(\mathrm{dd}, \mathrm{J}=16.1,4.4,1 \mathrm{H}), 2.95-3.06$ $(\mathrm{m}, 2 \mathrm{H}), 2.64(\mathrm{dd}, \mathrm{J}=16.1,4.2,1 \mathrm{H}), 2.31-2.39(\mathrm{br} \mathrm{m}, 1 \mathrm{H}), 2.01-2.14(\mathrm{~m}, 2 \mathrm{H}), 1.84-1.94(\mathrm{~m}, 1$ H), 1.70-1.78 (m, 1 H$), 1.25-1.63(\mathrm{~m}, 6 \mathrm{H}), 0.82(\mathrm{~s}, 9 \mathrm{H}), 0.02(\mathrm{~s}, 3 \mathrm{H}),-0.04(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 178.16,172.10,171.38,170.57,156.33,144.72,144.30,135.82,128.70$, $128.63,128.45,128.18,128.09,127.81,127.70,126.92,126.62,72.21,70.65,70.22,67.13$, $52.14,51.77,50.94,49.94,40.50,37.87,36.84,33.06,30.02,26.26,25.73,23.67,17.80,-4.34,-$ 4.88; ESI calculated for $\left[\mathrm{C}_{67} \mathrm{H}_{74} \mathrm{~N}_{7} \mathrm{O}_{8} \mathrm{Si}^{-}\right.$: 1132.5368 . Found 1132.5374.
(5S)-5-[(2S)-2-benzyloxycarbonylamino-3-(trityl-carbamoyl)-propionylamino]-(4R)-4-(tert-butyl-dimethyl-silanyloxy)-(2R)-2-(3-N,N'-bis-t-Boc-guanidino-propyl)-7-(tritylcarbamoyl)heptanoic acid (22).


To a solution of azide 21 ( $0.826 \mathrm{~g}, 0.728 \mathrm{mmol}$ ), $\mathrm{N}, \mathrm{N}$ '-bis-Boc-1-guanylpyrazole ( $0.452 \mathrm{~g}, 1.68 \mathrm{mmol}$ ), and aqueous $1 \mathrm{M} \mathrm{LiOH}(1.46 \mathrm{ml})$ in a mixture of THF (16.5 $\mathrm{ml})$ and water ( 1.5 ml ) was added triphenylphosphine ( $0.573 \mathrm{~g}, 2.18 \mathrm{mmol})$. Gas evolution commenced, and the reaction was allowed to stir at room temperature for 48 h . The THF was removed in vacuo, the residue was partitioned between aqueous $10 \%$ citric acid and ethyl acetate, and the layers were separated. The aqueous layer was extracted twice with ethyl acetate, the organics were combined, washed with water and brine, dried over $\mathrm{MgSO}_{4}$, and concentrated in vacuo. In order to separate the triphenylphosphine oxide, the residue was dissolved in 20 ml of $9: 1$ mixture of diethyl ether and dichloromethane and filtered through alumina ( 35 g ) in a sintered glass funnel. The flask was rinsed twice with 10 ml portions of the organic and these were also passed through the alumina. The alumina was rinsed with 300 ml of a 93:7 mixture of diethyl ether and methanol. The mother liquor was discarded and the desired product was flushed from the alumina with methanol (ca. 600 ml ). The methanol was removed in vacuo and the residue was purified by flash column chromatography ( $95: 5$ dichloromethane : methanol) to provide 0.548 g $(56 \%)$ of the title compound 22 as a foam. TLC $\mathrm{R}_{\mathrm{f}}=0.20\left(95: 5 \mathrm{CH}_{2} \mathrm{Cl}_{2}: \mathrm{MeOH}\right) ;{ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 11.49(\mathrm{~s}, 1 \mathrm{H}), 8.20(\mathrm{t}, \mathrm{J}=6.5,1 \mathrm{H}), 7.10-7.35(\mathrm{~m}, 36 \mathrm{H}), 6.84(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 6.77$ $(\mathrm{s}, 1 \mathrm{H}), 6.25(\mathrm{~s}, 1 \mathrm{H}), 5.04(\mathrm{~d}, \mathrm{~J}=15.0,1 \mathrm{H}), 4.97(\mathrm{~d}, \mathrm{~J}=15.0,1 \mathrm{H}), 4.54(\mathrm{dt}, \mathrm{J}=11.0,6.0,1 \mathrm{H})$, $4.06(\mathrm{br} \mathrm{t}, \mathrm{J}=11.5,1 \mathrm{H}), 3.62-3.75(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 3.12-3.38(\mathrm{~m}, 3 \mathrm{H}), 2.63(\mathrm{~d}, \mathrm{~J}=18.0,1 \mathrm{H}), 2.34$
(br s, 1 H$), 1.99-2.16(\mathrm{~m}, 2 \mathrm{H}), 1.81-1.94(\mathrm{~m}, 1 \mathrm{H}), 1.68-1.79(\mathrm{~m}, 1 \mathrm{H}), 1.30-1.60(\mathrm{~m}, 24 \mathrm{H}), 0.82$ $(\mathrm{s}, 9 \mathrm{H}), 0.02(\mathrm{~s}, 3 \mathrm{H}),-0.05(\mathrm{~s}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( 100 MHz ) $\delta: 178.84,172.29,171.50,170.79$, $163.67,156.48,156.13,153.22,144.90,144.54,136.02,128.87,128.62,128.29,128.00,127.86$, $127.09,126.80,83.01,79.19,72.40,70.82,70.38,67.31,52.28,52.00,41.40,40.77,38.00$, $37.04,33.33,30.62,28.44,28.21,26.80,25.90,23.81,17.96,-4.12,-4.77$, ESI calculated for $\left[\mathrm{C}_{78} \mathrm{H}_{94} \mathrm{~N}_{7} \mathrm{O}_{12} \mathrm{Si}\right]^{-}: 1348.6730$. Found 1348.6714.

## (4R)-4-(tert-butyl-dimethyl-silanyloxy)-(5S)-5-[(2S)-2-(9,9a-dihydro-4aH-flouren-9-

 ylmethoxycarbonylamino)-3-(trityl-carbamoyl)-propionylamino]-(2R)-2-(3-N,N'-bis-t-Boc-guanidino-propyl)-7-(tritylcarbamoyl)-heptanoic acid (1).

Step 1: To a stirred solution of guanidine $22(0.548 \mathrm{~g}$, 0.405 mmol ) in $\mathrm{MeOH}(20 \mathrm{ml})$ under a nitrogen atmosphere was added $20 \% \mathrm{Pd}(\mathrm{OH})_{2}$ on carbon (0.06 g). The reaction vessel was evacuated and purged with nitrogen three times, and then placed under an atmosphere of hydrogen using a balloon until TLC showed complete consumption of starting material (ca. 24 h ). At this time the hydrogen gas was evacuated, the catalyst was removed by filtration, and the solvent was removed in vacuo.

Step 2: The residue from Step 1 was dissolved in dioxane ( 5 ml ), aqueous $10 \%$ sodium carbonate $(3.2 \mathrm{ml})$ was added, and the mixture was cooled to $0{ }^{\circ} \mathrm{C}$. To this was added N - $(9-$ Fluorenylmethoxycarbonyloxy)succinimide ( $0.21 \mathrm{~g}, 0.061 \mathrm{mmol}$ ) and the reaction was allowed to warm to room temperature overnight. The reaction mixture was partitioned between aqueous $10 \%$ citric acid and ethyl acetate and the layers were separated. The aqueous layer was extracted
twice with ethyl acetate, the organics were combined, washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated in vacuo. The product was purified by flash column chromatography (98/2 dichloromethane / methanol) to yield 0.181 g ( $31 \%$ over two steps) of title compound $\mathbf{1}$ as a foam: $\mathrm{TLC}_{\mathrm{f}}=0.18\left(95: 5 \mathrm{CH}_{2} \mathrm{Cl}_{2}: \mathrm{MeOH}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 11.48(\mathrm{~s}, 1 \mathrm{H}), 8.18$ (br t, J=6.5, 1 H), $7.74(\mathrm{~d}, \mathrm{~J}=7.3,1 \mathrm{H}), 7.70(\mathrm{~d}, \mathrm{~J}=7.3,1 \mathrm{H}), 7.53(\mathrm{~d}, \mathrm{~J}=7.3,1 \mathrm{H}), 7.48(\mathrm{~d}, \mathrm{~J}=$ 7.3, 1 H$), 7.35-7.42(\mathrm{~m}, \mathrm{~J}=2 \mathrm{H}), 7.23-7.29(\mathrm{~m}, 2 \mathrm{H}), 7.11-7.22(\mathrm{~m}, 31 \mathrm{H}), 6.80(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 6.22$ (br s, 1 H), $4.56(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.40(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.04-4.22(\mathrm{br} \mathrm{m}, 3 \mathrm{H}), 3.70(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 3.33(\mathrm{~d}, \mathrm{~J}=$ $11.2,1 \mathrm{H}), 3.23(\mathrm{br} \mathrm{d}, \mathrm{J}=4.9,2 \mathrm{H}), 5.21(\mathrm{br} \mathrm{d}, \mathrm{J}=14.2,1 \mathrm{H}), 2.12(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 1.90(\mathrm{dt}, \mathrm{J}=12.7$, $7.8,1 \mathrm{H}), 1.76(\mathrm{td}, \mathrm{J}=10.2,3.9,1 \mathrm{H}), 1.20-1.63(\mathrm{~m}, 24 \mathrm{H}), 0.80(\mathrm{~s}, 9 \mathrm{H}), 0.02(\mathrm{~s}, 3 \mathrm{H}),-0.05(\mathrm{~s}$, $3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 62.9 MHz ) $\delta: 178.24,172.14,171.32,170.89,163.69,156.55,156.12,153.21$, $144.93,144.52,143.89,143.68,141.38,141.32,128.89,128.81,128.04,127.87,127.29,127.22$, $127.16,126.78,125.41,125.22,120.12,120.02,82.99,79.18,72.48,70.88,70.40,67.58,52.34$, $51.87,47.11,41.01,40.72,38.10,37.10,33.41,30.57,29.80,28.45,28.21,26.74,25.91,23.88$, 17.96, -4.12, -4.70; ESI calculated for $\left[\mathrm{C}_{85} \mathrm{H}_{98} \mathrm{~N}_{7} \mathrm{O}_{12} \mathrm{Si}^{+}\right.$: 1436.7043 . Found 1436.7062.

## Crystallographic Experimental Section for ent-15a.

## Data Collection

A colorless crystal with approximate dimensions $0.5 \times 0.4 \times 0.4 \mathrm{~mm}^{3}$ was selected under oil under ambient conditions and attached to the tip of a glass capillary. The crystal was mounted in a stream of cold nitrogen at $100(2) \mathrm{K}$ and centered in the X-ray beam by using a video camera.

The crystal evaluation and data collection were performed on a Bruker CCD-1000 diffractometer with Mo K $\quad(=0.71073 \AA)$ radiation and the diffractometer to crystal distance of 4.9 cm .

The initial cell constants were obtained from three series of $\omega$ scans at different starting angles. Each series consisted of 20 frames collected at intervals of $0.3^{\circ}$ in a $6^{\circ}$ range about $\omega$ with the exposure time of 10 seconds per frame. A total of 184 reflections was obtained. The reflections were successfully indexed by an automated indexing routine built in the SMART program. The final cell constants were calculated from a set of 16953 strong reflections from the actual data collection.

The data were collected by using the hemisphere data collection routine. The reciprocal space was surveyed to the extent of a full sphere to a resolution of $0.80 \AA$. A total of 30516 data were harvested by collecting three sets of frames with $0.3^{\circ}$ scans in $\omega$ with an exposure time 30 sec per frame. These highly redundant datasets were corrected for Lorentz and polarization effects. The absorption correction was based on fitting a function to the empirical transmission surface as sampled by multiple equivalent measurements. [1]

## Structure Solution and Refinement

The systematic absences in the diffraction data were consistent for the space groups Pna2 ${ }_{l}$ and Pnma. The E-statistics strongly suggested the non-centrosymmetric space group $\operatorname{Pna} 2_{1}$ that yielded chemically reasonable and computationally stable results of refinement [2]. A successful solution by the direct methods provided most non-hydrogen atoms from the $E$-map. The remaining non-hydrogen atoms were located in an alternating series of least-squares cycles and difference Fourier maps. All non-hydrogen atoms were refined with anisotropic displacement coefficients. All hydrogen atoms were included in the structure factor calculation at idealized positions and were allowed to ride on the neighboring atoms with relative isotropic displacement coefficients.

The final least-squares refinement of 572 parameters against 7950 data resulted in residuals $R$ (based on $F^{2}$ for $I \geq 2$ ) and $w R$ (based on $F^{2}$ for all data) of 0.0412 and 0.1083 , respectively. The final difference Fourier map was featureless.

The ORTEP diagram is drawn with $30 \%$ probability ellipsoids.

## References

[1] Blessing, R.H. Acta Cryst. 1995, A51, 33-38.
[2] All software and sources of the scattering factors are contained in the SHELXTL (version 5.1) program library (G. Sheldrick, Bruker Analytical X-Ray Systems, Madison, WI).


Table 1. Crystal data and structure refinement for rich08.

Identification code
Empirical formula
Formula weight
Temperature
Wavelength
Crystal system
Space group
Unit cell dimensions

Volume
Z
Density (calculated)
Absorption coefficient
F(000)
Crystal size
Theta range for data collection Index ranges
Reflections collected
Independent reflections
Completeness to theta $=26.40^{\circ}$
Absorption correction
Max. and min. transmission Refinement method Data / restraints / parameters
Goodness-of-fit on $\mathrm{F}^{2}$
Final R indices [ $\mathrm{I}>2 \operatorname{sigma}(\mathrm{I})$ ]
R indices (all data)
Absolute structure parameter
Largest diff. peak and hole
rich08
$\mathrm{C}_{41} \mathrm{H}_{56} \mathrm{~N}_{2} \mathrm{O}_{6} \mathrm{Si}$
700.97

100(2) K
$0.71073 \AA$
Orthorhombic
Pna2 ${ }_{1}$
$\begin{array}{ll}\mathrm{a}=14.3530(13) \AA & \alpha=90^{\circ} . \\ \mathrm{b}=15.3106(13) \AA & \beta=90^{\circ} . \\ \mathrm{c}=17.6827(15) \AA & \gamma=90^{\circ} .\end{array}$
3885.8(6) $\AA^{3}$

4
$1.198 \mathrm{Mg} / \mathrm{m}^{3}$
$0.108 \mathrm{~mm}^{-1}$
1512
$0.50 \times 0.40 \times 0.40 \mathrm{~mm}^{3}$
1.76 to $26.40^{\circ}$.
$-17<=\mathrm{h}<=17,-19<=\mathrm{k}<=19,-22<=\mathrm{l}<=22$
30516
$7950[\mathrm{R}($ int $)=0.0441]$
99.9 \%

Empirical with SADABS
0.9580 and 0.9479

Full-matrix least-squares on $\mathrm{F}^{2}$
7950 / $1 / 459$
1.043
$\mathrm{R} 1=0.0412, \mathrm{wR} 2=0.1071$
$\mathrm{R} 1=0.0428, \mathrm{wR} 2=0.1083$
0.05(10)
0.428 and -0.217 e. $\AA^{-3}$

Table 2. Atomic coordinates ( $\mathrm{x} 10^{4}$ ) and equivalent isotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for rich08. $\mathrm{U}(\mathrm{eq})$ is defined as one third of the trace of the orthogonalized $\mathrm{U}^{\mathrm{ij}}$ tensor.

|  | x | y | z | U(eq) |
| :---: | :---: | :---: | :---: | :---: |
| Si | 1885(1) | 10648(1) | 4915(1) | 23(1) |
| N(1) | 5370(1) | 9332(1) | 2854(1) | 16(1) |
| $\mathrm{N}(2)$ | 2880(1) | 7990(1) | 2752(1) | 19(1) |
| $\mathrm{O}(1)$ | 6383(1) | 8207(1) | 3030(1) | 20(1) |
| $\mathrm{O}(2)$ | 3368(1) | 9352(1) | 2413(1) | 22(1) |
| $\mathrm{O}(3)$ | 1946(1) | 8816(1) | 2079(1) | 20(1) |
| $\mathrm{O}(4)$ | 4032(1) | 7013(1) | 4407(1) | 25(1) |
| $\mathrm{O}(5)$ | 4176(1) | 6785(1) | 5647(1) | 36(1) |
| $\mathrm{O}(6)$ | 1843(1) | 10013(1) | 5669(1) | 32(1) |
| C(1) | 7299(1) | 9878(1) | 2575(1) | 20(1) |
| C(2) | 8264(1) | 9923(1) | 2502(1) | 23(1) |
| C(3) | 8816(1) | 10065(1) | 3136(1) | 24(1) |
| C(4) | 8403(1) | 10156(1) | 3839(1) | 24(1) |
| C(5) | 7437(1) | 10102(1) | 3911(1) | 20(1) |
| C(6) | 6875(1) | 9966(1) | 3281(1) | 17(1) |
| C(7) | 6089(2) | 11585(1) | 2894(1) | 31(1) |
| C(8) | 5761(2) | 12385(2) | 2631(2) | 44(1) |
| C(9) | 4826(2) | 12515(1) | 2504(2) | 41(1) |
| $\mathrm{C}(10)$ | 4204(2) | 11842(1) | 2648(1) | 35(1) |
| $\mathrm{C}(11)$ | 4527(1) | 11046(1) | 2912(1) | 25(1) |
| $\mathrm{C}(12)$ | 5475(1) | 10901(1) | 3031(1) | 19(1) |
| C(13) | 5403(1) | 10618(1) | 4645(1) | 21(1) |
| C(14) | 5187(1) | 10516(1) | 5408(1) | 26(1) |
| C(15) | 5044(1) | 9690(2) | 5710(1) | 28(1) |
| C(16) | 5132(1) | 8962(1) | 5244(1) | 25(1) |
| C(17) | 5360(1) | 9065(1) | 4481(1) | 21(1) |
| C(18) | 5491(1) | 9894(1) | 4174(1) | 17(1) |
| C(19) | 5810(1) | 10008(1) | 3343(1) | 16(1) |
| C(20) | 5665(1) | 8503(1) | 2757(1) | 16(1) |
| C(21) | 5018(1) | 7921(1) | 2291(1) | 17(1) |
| C(22) | 4460(1) | 7315(1) | 2816(1) | 19(1) |
| C(23) | 3670(1) | 7779(1) | 3247(1) | 17(1) |
| C(24) | 2784(1) | 8768(1) | 2415(1) | 17(1) |
| C(25) | 1668(1) | 9601(1) | 1656(1) | 18(1) |
| C(26) | 2314(2) | 9755(1) | 988(1) | 28(1) |
| C(27) | 1619(2) | 10390(1) | 2175(1) | 30(1) |
| C(28) | 693(1) | 9364(1) | 1380(1) | 30(1) |
| C(29) | 3277(1) | 7208(1) | 3878(1) | 19(1) |
| C(30) | 3688(2) | 6969(1) | 5124(1) | 27(1) |
| C(31) | 2647(2) | 7155(1) | 5121(1) | 29(1) |
| C(32) | 2533(1) | 7640(1) | 4365(1) | 23(1) |
| C(33) | 2279(2) | 7579(2) | 5842(1) | 36(1) |
| C(34) | 2458(2) | 8562(2) | 5915(1) | 36(1) |
| C(35) | 1635(2) | 9102(1) | 5625(1) | 32(1) |
| C(36) | 738(2) | 10634(2) | 4423(1) | 37(1) |
| C(37) | 2785(2) | 10248(2) | 4239(1) | 32(1) |
| C(38) | 2164(1) | 11781(1) | 5265(1) | 25(1) |
| C(39) | 3023(2) | 11773(2) | 5772(2) | 53(1) |
| C(40) | 2332(2) | 12386(2) | 4582(2) | 47(1) |
| C(41) | 1358(2) | 12146(2) | 5726(2) | 45(1) |

Table 3. Bond lengths $\left[\AA \AA\right.$ ] and angles [ ${ }^{\circ}$ ] for rich08.

| Si-O(6) | $1.6518(16)$ | $\mathrm{C}(10)-\mathrm{C}(11)$ | 1.384(3) |
| :---: | :---: | :---: | :---: |
| Si-C(36) | 1.861(2) | $\mathrm{C}(11)-\mathrm{C}(12)$ | 1.394 (3) |
| Si-C(37) | 1.863(2) | $\mathrm{C}(12)-\mathrm{C}(19)$ | 1.551(2) |
| Si-C(38) | $1.886(2)$ | $\mathrm{C}(13)-\mathrm{C}(18)$ | 1.393(3) |
| $\mathrm{N}(1)-\mathrm{C}(20)$ | 1.349 (2) | $\mathrm{C}(13)-\mathrm{C}(14)$ | 1.393 (3) |
| $\mathrm{N}(1)-\mathrm{C}(19)$ | 1.489 (2) | $\mathrm{C}(14)-\mathrm{C}(15)$ | $1.388(3)$ |
| $\mathrm{N}(2)-\mathrm{C}(24)$ | $1.339(2)$ | $\mathrm{C}(15)-\mathrm{C}(16)$ | 1.392(3) |
| $\mathrm{N}(2)-\mathrm{C}(23)$ | $1.468(2)$ | $\mathrm{C}(16)-\mathrm{C}(17)$ | $1.398(3)$ |
| $\mathrm{O}(1)-\mathrm{C}(20)$ | $1.225(2)$ | $\mathrm{C}(17)-\mathrm{C}(18)$ | 1.392(3) |
| $\mathrm{O}(2)-\mathrm{C}(24)$ | 1.227(2) | $\mathrm{C}(18)-\mathrm{C}(19)$ | 1.549 (2) |
| $\mathrm{O}(3)-\mathrm{C}(24)$ | $1.344(2)$ | $\mathrm{C}(20)-\mathrm{C}(21)$ | 1.528(2) |
| $\mathrm{O}(3)-\mathrm{C}(25)$ | 1.471(2) | $\mathrm{C}(21)-\mathrm{C}(22)$ | $1.538(2)$ |
| $\mathrm{O}(4)-\mathrm{C}(30)$ | $1.362(2)$ | $\mathrm{C}(22)-\mathrm{C}(23)$ | $1.539(3)$ |
| $\mathrm{O}(4)-\mathrm{C}(29)$ | $1.462(2)$ | $\mathrm{C}(23)-\mathrm{C}(29)$ | 1.526 (2) |
| $\mathrm{O}(5)-\mathrm{C}(30)$ | 1.194(2) | $\mathrm{C}(25)-\mathrm{C}(27)$ | 1.518(3) |
| $\mathrm{O}(6)-\mathrm{C}(35)$ | 1.428 (3) | $\mathrm{C}(25)-\mathrm{C}(26)$ | 1.520 (3) |
| $\mathrm{C}(1)-\mathrm{C}(2)$ | 1.393 (3) | C(25)-C(28) | 1.526 (3) |
| $\mathrm{C}(1)-\mathrm{C}(6)$ | 1.397 (3) | $\mathrm{C}(29)$-C(32) | 1.523(3) |
| $\mathrm{C}(2)-\mathrm{C}(3)$ | 1.390 (3) | $\mathrm{C}(30)-\mathrm{C}(31)$ | 1.521 (3) |
| $\mathrm{C}(3)-\mathrm{C}(4)$ | $1.385(3)$ | $\mathrm{C}(31)-\mathrm{C}(33)$ | 1.526 (3) |
| $\mathrm{C}(4)-\mathrm{C}(5)$ | $1.394(3)$ | $\mathrm{C}(31)-\mathrm{C}(32)$ | 1.537(3) |
| $\mathrm{C}(5)-\mathrm{C}(6)$ | 1.390 (3) | $\mathrm{C}(33)-\mathrm{C}(34)$ | 1.532(3) |
| $\mathrm{C}(6)-\mathrm{C}(19)$ | $1.533(2)$ | $\mathrm{C}(34)-\mathrm{C}(35)$ | 1.530 (3) |
| $\mathrm{C}(7)-\mathrm{C}(12)$ | 1.390 (3) | $\mathrm{C}(38)-\mathrm{C}(41)$ | 1.521 (3) |
| $\mathrm{C}(7)-\mathrm{C}(8)$ | $1.391(3)$ | $\mathrm{C}(38)-\mathrm{C}(39)$ | $1.524(3)$ |
| $\mathrm{C}(8)-\mathrm{C}(9)$ | $1.375(4)$ | $\mathrm{C}(38)-\mathrm{C}(40)$ | 1.540(3) |
| C(9)-C(10) | 1.387(4) |  |  |
| $\mathrm{O}(6)-\mathrm{Si}-\mathrm{C}(36)$ | 109.77(10) | $\mathrm{C}(15)-\mathrm{C}(14)-\mathrm{C}(13)$ | 120.41(18) |
| $\mathrm{O}(6)-\mathrm{Si}-\mathrm{C}(37)$ | 110.47(10) | $\mathrm{C}(14)-\mathrm{C}(15)-\mathrm{C}(16)$ | 119.27(18) |
| $\mathrm{C}(36)-\mathrm{Si}-\mathrm{C}(37)$ | 108.10(11) | $\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{C}(17)$ | 120.16(19) |
| $\mathrm{O}(6)-\mathrm{Si}-\mathrm{C}(38)$ | 106.49(9) | $\mathrm{C}(18)-\mathrm{C}(17)-\mathrm{C}(16)$ | 120.73(18) |
| $\mathrm{C}(36)-\mathrm{Si}-\mathrm{C}(38)$ | 110.56(10) | $\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(13)$ | 118.64(17) |
| $\mathrm{C}(37)-\mathrm{Si}-\mathrm{C}(38)$ | 111.46(10) | $\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(19)$ | 120.84(15) |
| $\mathrm{C}(20)-\mathrm{N}(1)-\mathrm{C}(19)$ | 126.49(14) | $\mathrm{C}(13)-\mathrm{C}(18)-\mathrm{C}(19)$ | 120.33(16) |
| $\mathrm{C}(24)-\mathrm{N}(2)-\mathrm{C}(23)$ | 122.72(14) | $\mathrm{N}(1)-\mathrm{C}(19)-\mathrm{C}(6)$ | 110.68(14) |
| $\mathrm{C}(24)-\mathrm{O}(3)-\mathrm{C}(25)$ | 120.87(13) | $\mathrm{N}(1)-\mathrm{C}(19)-\mathrm{C}(18)$ | 110.25(14) |
| $\mathrm{C}(30)-\mathrm{O}(4)-\mathrm{C}(29)$ | 109.69(15) | $\mathrm{C}(6)-\mathrm{C}(19)-\mathrm{C}(18)$ | 110.92(14) |
| $\mathrm{C}(35)-\mathrm{O}(6)-\mathrm{Si}$ | 122.52(14) | $\mathrm{N}(1)-\mathrm{C}(19)-\mathrm{C}(12)$ | 105.96(13) |
| $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(6)$ | 120.75(17) | $\mathrm{C}(6)-\mathrm{C}(19)-\mathrm{C}(12)$ | 108.72(14) |
| $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{C}(1)$ | 120.08(18) | $\mathrm{C}(18)-\mathrm{C}(19)-\mathrm{C}(12)$ | 110.19(14) |
| $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}(2)$ | 119.66(18) | $\mathrm{O}(1)-\mathrm{C}(20)-\mathrm{N}(1)$ | 124.22(16) |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | 120.10(18) | $\mathrm{O}(1)-\mathrm{C}(20)-\mathrm{C}(21)$ | 120.54(15) |
| $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(4)$ | 120.90(17) | $\mathrm{N}(1)-\mathrm{C}(20)-\mathrm{C}(21)$ | 115.22(15) |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(1)$ | 118.50(16) | $\mathrm{C}(20)-\mathrm{C}(21)-\mathrm{C}(22)$ | 110.01(14) |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(19)$ | 121.06(16) | $\mathrm{C}(21)-\mathrm{C}(22)-\mathrm{C}(23)$ | 113.76(14) |
| $\mathrm{C}(1)-\mathrm{C}(6)-\mathrm{C}(19)$ | 120.12(16) | $\mathrm{N}(2)-\mathrm{C}(23)-\mathrm{C}(29)$ | 106.02(14) |
| $\mathrm{C}(12)-\mathrm{C}(7)-\mathrm{C}(8)$ | 120.5(2) | $\mathrm{N}(2)-\mathrm{C}(23)-\mathrm{C}(22)$ | 112.06(14) |
| $\mathrm{C}(9)-\mathrm{C}(8)-\mathrm{C}(7)$ | 120.8(2) | $\mathrm{C}(29)-\mathrm{C}(23)-\mathrm{C}(22)$ | 111.70(14) |
| C(8)-C(9)-C(10) | 119.4(2) | $\mathrm{O}(2)-\mathrm{C}(24)-\mathrm{N}(2)$ | 125.39(16) |
| $\mathrm{C}(11)-\mathrm{C}(10)-\mathrm{C}(9)$ | 120.0(2) | $\mathrm{O}(2)-\mathrm{C}(24)-\mathrm{O}(3)$ | 124.79(16) |
| $\mathrm{C}(10)-\mathrm{C}(11)-\mathrm{C}(12)$ | 121.17(19) | $\mathrm{N}(2)-\mathrm{C}(24)-\mathrm{O}(3)$ | 109.83(14) |
| $\mathrm{C}(7)-\mathrm{C}(12)-\mathrm{C}(11)$ | 118.18(17) | $\mathrm{O}(3)-\mathrm{C}(25)-\mathrm{C}(27)$ | 110.82(15) |
| $\mathrm{C}(7)-\mathrm{C}(12)-\mathrm{C}(19)$ | 121.96(17) | $\mathrm{O}(3)-\mathrm{C}(25)-\mathrm{C}(26)$ | 110.86(14) |
| $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(19)$ | 119.80(16) | $\mathrm{C}(27)-\mathrm{C}(25)-\mathrm{C}(26)$ | 112.05(16) |
| C(18)-C(13)-C(14) | 120.78(18) | $\mathrm{O}(3)-\mathrm{C}(25)-\mathrm{C}(28)$ | 102.51(14) |


| $\mathrm{C}(27)-\mathrm{C}(25)-\mathrm{C}(28)$ | $109.87(16)$ | $\mathrm{C}(29)-\mathrm{C}(32)-\mathrm{C}(31)$ | $101.95(15)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{C}(26)-\mathrm{C}(25)-\mathrm{C}(28)$ | $110.34(16)$ | $\mathrm{C}(31)-\mathrm{C}(33)-\mathrm{C}(34)$ | $115.50(19)$ |
| $\mathrm{O}(4)-\mathrm{C}(29)-\mathrm{C}(32)$ | $104.29(14)$ | $\mathrm{C}(35)-\mathrm{C}(34)-\mathrm{C}(33)$ | $111.9(2)$ |
| $\mathrm{O}(4)-\mathrm{C}(29)-\mathrm{C}(23)$ | $108.11(14)$ | $\mathrm{O}(6)-\mathrm{C}(35)-\mathrm{C}(34)$ | $110.34(18)$ |
| $\mathrm{C}(32)-\mathrm{C}(29)-\mathrm{C}(23)$ | $115.11(15)$ | $\mathrm{C}(41)-\mathrm{C}(38)-\mathrm{C}(39)$ | $107.6(2)$ |
| $\mathrm{O}(5)-\mathrm{C}(30)-\mathrm{O}(4)$ | $121.3(2)$ | $\mathrm{C}(41)-\mathrm{C}(38)-\mathrm{C}(40)$ | $108.6(2)$ |
| $\mathrm{O}(5)-\mathrm{C}(30)-\mathrm{C}(31)$ | $128.61(19)$ | $\mathrm{C}(39)-\mathrm{C}(38)-\mathrm{C}(40)$ | $109.8(2)$ |
| $\mathrm{O}(4)-\mathrm{C}(30)-\mathrm{C}(31)$ | $110.06(16)$ | $\mathrm{C}(41)-\mathrm{C}(38)-\mathrm{Si}$ | $110.69(15)$ |
| $\mathrm{C}(30)-\mathrm{C}(31)-\mathrm{C}(33)$ | $114.60(18)$ | $\mathrm{C}(39)-\mathrm{C}(38)-\mathrm{Si}$ | $110.90(15)$ |
| $\mathrm{C}(30)-\mathrm{C}(31)-\mathrm{C}(32)$ | $101.43(16)$ | $\mathrm{C}(40)-\mathrm{C}(38)-\mathrm{Si}$ | $109.20(15)$ |
| $\mathrm{C}(33)-\mathrm{C}(31)-\mathrm{C}(32)$ | $118.97(17)$ |  |  |

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for rich08. The anisotropic displacement factor exponent takes the form: $-2 \pi^{2}\left[h^{2} a^{* 2} U^{11}+\ldots+2 h k a^{*} b^{*} U^{12}\right]$

|  | $\mathrm{U}^{11}$ | $\mathrm{U}^{22}$ | $\mathrm{U}^{33}$ | $\mathrm{U}^{23}$ | $\mathrm{U}^{13}$ | $\mathrm{U}^{12}$ |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| Si | $22(1)$ | $27(1)$ | $19(1)$ | $-2(1)$ | $0(1)$ | $-1(1)$ |
| $\mathrm{N}(1)$ | $14(1)$ | $17(1)$ | $16(1)$ | $0(1)$ | $-3(1)$ | $-1(1)$ |
| $\mathrm{N}(2)$ | $15(1)$ | $20(1)$ | $21(1)$ | $4(1)$ | $-4(1)$ | $-4(1)$ |
| $\mathrm{O}(1)$ | $16(1)$ | $20(1)$ | $23(1)$ | $-3(1)$ | $-3(1)$ | $3(1)$ |
| $\mathrm{O}(2)$ | $19(1)$ | $19(1)$ | $28(1)$ | $2(1)$ | $-5(1)$ | $-2(1)$ |
| $\mathrm{O}(3)$ | $18(1)$ | $20(1)$ | $22(1)$ | $5(1)$ | $-4(1)$ | $-2(1)$ |
| $\mathrm{O}(4)$ | $26(1)$ | $27(1)$ | $22(1)$ | $6(1)$ | $-7(1)$ | $-1(1)$ |
| $\mathrm{O}(5)$ | $43(1)$ | $39(1)$ | $27(1)$ | $13(1)$ | $-13(1)$ | $-5(1)$ |
| $\mathrm{O}(6)$ | $42(1)$ | $31(1)$ | $24(1)$ | $1(1)$ | $1(1)$ | $-4(1)$ |
| $\mathrm{C}(1)$ | $22(1)$ | $20(1)$ | $17(1)$ | $-1(1)$ | $-2(1)$ | $-2(1)$ |
| $\mathrm{C}(2)$ | $22(1)$ | $21(1)$ | $26(1)$ | $-1(1)$ | $4(1)$ | $-2(1)$ |
| $\mathrm{C}(3)$ | $17(1)$ | $26(1)$ | $31(1)$ | $0(1)$ | $3(1)$ | $-3(1)$ |
| $\mathrm{C}(4)$ | $20(1)$ | $26(1)$ | $26(1)$ | $-5(1)$ | $-4(1)$ | $-5(1)$ |
| $\mathrm{C}(5)$ | $21(1)$ | $20(1)$ | $19(1)$ | $-3(1)$ | $2(1)$ | $-1(1)$ |
| $\mathrm{C}(6)$ | $17(1)$ | $13(1)$ | $22(1)$ | $0(1)$ | $0(1)$ | $-2(1)$ |
| $\mathrm{C}(7)$ | $31(1)$ | $23(1)$ | $40(1)$ | $4(1)$ | $7(1)$ | $-1(1)$ |
| $\mathrm{C}(8)$ | $48(1)$ | $24(1)$ | $60(2)$ | $13(1)$ | $10(1)$ | $-2(1)$ |
| $\mathrm{C}(9)$ | $54(2)$ | $22(1)$ | $46(1)$ | $13(1)$ | $-4(1)$ | $8(1)$ |
| $\mathrm{C}(10)$ | $37(1)$ | $30(1)$ | $37(1)$ | $-1(1)$ | $-11(1)$ | $9(1)$ |
| $\mathrm{C}(11)$ | $26(1)$ | $20(1)$ | $28(1)$ | $-3(1)$ | $-6(1)$ | $1(1)$ |
| $\mathrm{C}(12)$ | $25(1)$ | $17(1)$ | $15(1)$ | $-1(1)$ | $0(1)$ | $2(1)$ |
| $\mathrm{C}(13)$ | $18(1)$ | $25(1)$ | $22(1)$ | $-3(1)$ | $-2(1)$ | $0(1)$ |
| $\mathrm{C}(14)$ | $23(1)$ | $36(1)$ | $19(1)$ | $-10(1)$ | $0(1)$ | $2(1)$ |
| $\mathrm{C}(15)$ | $20(1)$ | $48(1)$ | $15(1)$ | $1(1)$ | $1(1)$ | $-2(1)$ |
| $\mathrm{C}(16)$ | $21(1)$ | $34(1)$ | $21(1)$ | $6(1)$ | $0(1)$ | $-3(1)$ |
| $\mathrm{C}(17)$ | $20(1)$ | $24(1)$ | $19(1)$ | $0(1)$ | $-1(1)$ | $-1(1)$ |
| $\mathrm{C}(18)$ | $12(1)$ | $23(1)$ | $14(1)$ | $-2(1)$ | $-1)$ | $0(1)$ |
| $\mathrm{C}(19)$ | $18(1)$ | $16(1)$ | $14(1)$ | $-2(1)$ | $-1)$ | $-2(1)$ |
| $\mathrm{C}(20)$ | $16(1)$ | $17(1)$ | $14(1)$ | $-1(1)$ | $3(1)$ | $-1(1)$ |
| $\mathrm{C}(21)$ | $19(1)$ | $17(1)$ | $17(1)$ | $-2(1)$ | $1(1)$ | $-1(1)$ |
| $\mathrm{C}(22)$ | $18(1)$ | $16(1)$ | $23(1)$ | $1(1)$ | $-3(1)$ | $-2(1)$ |
| $\mathrm{C}(23)$ | $19(1)$ | $15(1)$ | $18(1)$ | $1(1)$ | $-3(1)$ | $-2(1)$ |
| $\mathrm{C}(24)$ | $15(1)$ | $22(1)$ | $15(1)$ | $0(1)$ | $2(1)$ | $1(1)$ |
| $\mathrm{C}(25)$ | $19(1)$ | $17(1)$ | $18(1)$ | $2(1)$ | $-2(1)$ | $2(1)$ |
| $\mathrm{C}(26)$ | $31(1)$ | $28(1)$ | $23(1)$ | $8(1)$ | $5(1)$ | $6(1)$ |
| $\mathrm{C}(27)$ | $33(1)$ | $28(1)$ | $27(1)$ | $-5(1)$ | $-3(1)$ | $8(1)$ |
| $\mathrm{C}(28)$ | $25(1)$ | $29(1)$ | $36(1)$ | $8(1)$ | $-9(1)$ | $2(1)$ |
| $\mathrm{C}(29)$ | $22(1)$ | $19(1)$ | $18(1)$ | $3(1)$ | $-4(1)$ | $-1(1)$ |
|  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |


| C(30) | $38(1)$ | $21(1)$ | $23(1)$ | $6(1)$ | $-5(1)$ | $-6(1)$ |
| :--- | :--- | :--- | :--- | :---: | :---: | :---: |
| $\mathrm{C}(31)$ | $39(1)$ | $27(1)$ | $21(1)$ | $5(1)$ | $1(1)$ | $1(1)$ |
| $\mathrm{C}(32)$ | $31(1)$ | $24(1)$ | $16(1)$ | $0(1)$ | $0(1)$ | $3(1)$ |
| $\mathrm{C}(33)$ | $47(1)$ | $33(1)$ | $26(1)$ | $2(1)$ | $5(1)$ | $-2(1)$ |
| $\mathrm{C}(34)$ | $43(1)$ | $39(1)$ | $24(1)$ | $-5(1)$ | $-2(1)$ | $-2(1)$ |
| $\mathrm{C}(35)$ | $39(1)$ | $30(1)$ | $28(1)$ | $2(1)$ | $9(1)$ | $-2(1)$ |
| $\mathrm{C}(36)$ | $28(1)$ | $50(1)$ | $35(1)$ | $-8(1)$ | $-10(1)$ | $-3(1)$ |
| $\mathrm{C}(37)$ | $35(1)$ | $32(1)$ | $29(1)$ | $-3(1)$ | $8(1)$ | $1(1)$ |
| $\mathrm{C}(38)$ | $25(1)$ | $25(1)$ | $25(1)$ | $-2(1)$ | $1(1)$ | $1(1)$ |
| $\mathrm{C}(39)$ | $49(2)$ | $35(1)$ | $76(2)$ | $-11(1)$ | $-32(1)$ | $4(1)$ |
| $\mathrm{C}(40)$ | $76(2)$ | $29(1)$ | $36(1)$ | $-1(1)$ | $7(1)$ | $-5(1)$ |
| $\mathrm{C}(41)$ | $44(1)$ | $43(1)$ | $47(1)$ | $-21(1)$ | $18(1)$ | $-6(1)$ |

Table 5. Hydrogen coordinates ( $\times 10^{4}$ ) and isotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for rich08.

|  | x | y | Z | U(eq) |
| :---: | :---: | :---: | :---: | :---: |
| H(1) | 4866 | 9488 | 2604 | 19 |
| $\mathrm{H}(2)$ | 2452 | 7588 | 2674 | 22 |
| H(1A) | 6925 | 9787 | 2139 | 24 |
| $\mathrm{H}(2 \mathrm{~A})$ | 8545 | 9857 | 2019 | 27 |
| H(3) | 9474 | 10100 | 3087 | 29 |
| H(4) | 8778 | 10255 | 4273 | 29 |
| H(5) | 7159 | 10159 | 4396 | 24 |
| H(7) | 6736 | 11506 | 2980 | 38 |
| H(8) | 6189 | 12846 | 2539 | 53 |
| H(9) | 4609 | 13061 | 2320 | 49 |
| H(10) | 3556 | 11927 | 2566 | 42 |
| $\mathrm{H}(11)$ | 4095 | 10591 | 3014 | 30 |
| H(13) | 5491 | 11187 | 4444 | 26 |
| H(14) | 5137 | 11016 | 5724 | 31 |
| $\mathrm{H}(15)$ | 4888 | 9622 | 6229 | 33 |
| $\mathrm{H}(16)$ | 5038 | 8394 | 5445 | 30 |
| H(17) | 5425 | 8565 | 4168 | 25 |
| H(21A) | 5390 | 7567 | 1933 | 21 |
| H(21B) | 4584 | 8289 | 1994 | 21 |
| H(22A) | 4892 | 7049 | 3188 | 23 |
| H(22B) | 4189 | 6836 | 2511 | 23 |
| H(23) | 3918 | 8332 | 3471 | 21 |
| H(26A) | 2928 | 9935 | 1172 | 41 |
| H(26B) | 2055 | 10214 | 664 | 41 |
| H(26C) | 2374 | 9214 | 696 | 41 |
| H(27A) | 1250 | 10244 | 2624 | 44 |
| H(27B) | 1326 | 10878 | 1907 | 44 |
| H(27C) | 2250 | 10556 | 2330 | 44 |
| H(28A) | 728 | 8850 | 1052 | 44 |
| H(28B) | 431 | 9856 | 1096 | 44 |
| H(28C) | 294 | 9235 | 1816 | 44 |
| H(29) | 3029 | 6652 | 3660 | 23 |
| H(31) | 2323 | 6580 | 5067 | 35 |
| H(32A) | 1903 | 7552 | 4150 | 28 |
| H(32B) | 2651 | 8273 | 4423 | 28 |
| H(33A) | 2566 | 7282 | 6281 | 43 |
| H(33B) | 1598 | 7477 | 5870 | 43 |


| H(34A) | 2572 | 8707 | 6453 | 43 |
| :--- | ---: | ---: | ---: | ---: |
| H(34B) | 3024 | 8717 | 5625 | 43 |
| H(35A) | 1501 | 8941 | 5093 | 39 |
| H(35B) | 1075 | 8972 | 5931 | 39 |
| H(36A) | 590 | 10035 | 4268 | 56 |
| H(36B) | 767 | 11011 | 3976 | 56 |
| H(36C) | 254 | 10848 | 4767 | 56 |
| H(37A) | 3375 | 10159 | 4508 | 48 |
| H(37B) | 2874 | 10681 | 3837 | 48 |
| H(37C) | 2581 | 9694 | 4017 | 48 |
| H(39A) | 3124 | 12358 | 5980 | 80 |
| H(39B) | 3568 | 11597 | 5475 | 80 |
| H(39C) | 2928 | 11357 | 6186 | 80 |
| H(40A) | 1783 | 12378 | 4252 | 70 |
| H(40B) | 2877 | 12181 | 4299 | 70 |
| H(40C) | 2442 | 12983 | 4761 | 70 |
| H(41A) | 1225 | 11754 | 6150 | 67 |
| H(41B) | 805 | 12195 | 5404 | 67 |
| H(41C) | 1526 | 12725 | 5920 | 67 |

Table 6. Torsion angles [ ${ }^{\circ}$ ] for rich08.

| $\mathrm{C}(36)-\mathrm{Si}-\mathrm{O}(6)-\mathrm{C}(35)$ | 59.45(19) | $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(19)-\mathrm{C}(12)$ | 101.89(18) |
| :---: | :---: | :---: | :---: |
| $\mathrm{C}(37)-\mathrm{Si}-\mathrm{O}(6)-\mathrm{C}(35)$ | -59.66(18) | $\mathrm{C}(1)-\mathrm{C}(6)-\mathrm{C}(19)-\mathrm{C}(12)$ | -71.58(19) |
| $\mathrm{C}(38)-\mathrm{Si}-\mathrm{O}(6)-\mathrm{C}(35)$ | 179.16(15) | $\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(19)-\mathrm{N}(1)$ | 37.1(2) |
| $\mathrm{C}(6)-\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | -0.6(3) | $\mathrm{C}(13)-\mathrm{C}(18)-\mathrm{C}(19)-\mathrm{N}(1)$ | -148.07(15) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | 0.4(3) | $\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(19)-\mathrm{C}(6)$ | -85.81(19) |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | 0.2(3) | $\mathrm{C}(13)-\mathrm{C}(18)-\mathrm{C}(19)-\mathrm{C}(6)$ | 88.98(19) |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | -0.7(3) | $\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(19)-\mathrm{C}(12)$ | 153.75(16) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(1)$ | 0.4(3) | $\mathrm{C}(13)-\mathrm{C}(18)-\mathrm{C}(19)-\mathrm{C}(12)$ | -31.5(2) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(19)$ | -173.13(17) | $\mathrm{C}(7)-\mathrm{C}(12)-\mathrm{C}(19)-\mathrm{N}(1)$ | -132.07(18) |
| $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(6)-\mathrm{C}(5)$ | 0.2(3) | $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(19)-\mathrm{N}(1)$ | 50.8(2) |
| $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(6)-\mathrm{C}(19)$ | 173.82(16) | $\mathrm{C}(7)-\mathrm{C}(12)-\mathrm{C}(19)-\mathrm{C}(6)$ | -13.1(2) |
| $\mathrm{C}(12)-\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9)$ | 0.2(4) | $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(19)-\mathrm{C}(6)$ | 169.81(16) |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(10)$ | 0.7(4) | $\mathrm{C}(7)-\mathrm{C}(12)-\mathrm{C}(19)-\mathrm{C}(18)$ | 108.7(2) |
| $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)$ | -0.5(4) | $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(19)-\mathrm{C}(18)$ | -68.4(2) |
| $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)-\mathrm{C}(12)$ | -0.7(3) | $\mathrm{C}(19)-\mathrm{N}(1)-\mathrm{C}(20)-\mathrm{O}(1)$ | -4.6(3) |
| $\mathrm{C}(8)-\mathrm{C}(7)-\mathrm{C}(12)-\mathrm{C}(11)$ | -1.3(3) | $\mathrm{C}(19)-\mathrm{N}(1)-\mathrm{C}(20)-\mathrm{C}(21)$ | 173.52(15) |
| $\mathrm{C}(8)-\mathrm{C}(7)-\mathrm{C}(12)-\mathrm{C}(19)$ | -178.4(2) | $\mathrm{O}(1)-\mathrm{C}(20)-\mathrm{C}(21)-\mathrm{C}(22)$ | 75.8(2) |
| $\mathrm{C}(10)-\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(7)$ | 1.5(3) | $\mathrm{N}(1)-\mathrm{C}(20)-\mathrm{C}(21)-\mathrm{C}(22)$ | -102.36(17) |
| $\mathrm{C}(10)-\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(19)$ | 178.74(18) | $\mathrm{C}(20)-\mathrm{C}(21)-\mathrm{C}(22)-\mathrm{C}(23)$ | 74.73(18) |
| $\mathrm{C}(18)-\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(15)$ | -0.8(3) | $\mathrm{C}(24)-\mathrm{N}(2)-\mathrm{C}(23)-\mathrm{C}(29)$ | 143.20(16) |
| $\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(15)-\mathrm{C}(16)$ | 0.9(3) | $\mathrm{C}(24)-\mathrm{N}(2)-\mathrm{C}(23)-\mathrm{C}(22)$ | -94.69(19) |
| $\mathrm{C}(14)-\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{C}(17)$ | -0.2(3) | $\mathrm{C}(21)-\mathrm{C}(22)-\mathrm{C}(23)-\mathrm{N}(2)$ | 73.44(18) |
| $\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{C}(17)-\mathrm{C}(18)$ | -0.7(3) | $\mathrm{C}(21)-\mathrm{C}(22)-\mathrm{C}(23)-\mathrm{C}(29)$ | -167.76(14) |
| $\mathrm{C}(16)-\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(13)$ | 0.8(3) | $\mathrm{C}(23)-\mathrm{N}(2)-\mathrm{C}(24)-\mathrm{O}(2)$ | 7.7(3) |
| $\mathrm{C}(16)-\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(19)$ | 175.68(16) | $\mathrm{C}(23)-\mathrm{N}(2)-\mathrm{C}(24)-\mathrm{O}(3)$ | -172.17(15) |
| $\mathrm{C}(14)-\mathrm{C}(13)-\mathrm{C}(18)-\mathrm{C}(17)$ | -0.1(3) | $\mathrm{C}(25)-\mathrm{O}(3)-\mathrm{C}(24)-\mathrm{O}(2)$ | 1.1(3) |
| $\mathrm{C}(14)-\mathrm{C}(13)-\mathrm{C}(18)-\mathrm{C}(19)$ | -174.98(16) | $\mathrm{C}(25)-\mathrm{O}(3)-\mathrm{C}(24)-\mathrm{N}(2)$ | -179.07(14) |
| $\mathrm{C}(20)-\mathrm{N}(1)-\mathrm{C}(19)-\mathrm{C}(6)$ | 40.6(2) | $\mathrm{C}(24)-\mathrm{O}(3)-\mathrm{C}(25)-\mathrm{C}(27)$ | -62.9(2) |
| $\mathrm{C}(20)-\mathrm{N}(1)-\mathrm{C}(19)-\mathrm{C}(18)$ | -82.5(2) | $\mathrm{C}(24)-\mathrm{O}(3)-\mathrm{C}(25)-\mathrm{C}(26)$ | 62.1(2) |
| $\mathrm{C}(20)-\mathrm{N}(1)-\mathrm{C}(19)-\mathrm{C}(12)$ | 158.28(16) | $\mathrm{C}(24)-\mathrm{O}(3)-\mathrm{C}(25)-\mathrm{C}(28)$ | 179.88(16) |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(19)-\mathrm{N}(1)$ | -142.12(16) | $\mathrm{C}(30)-\mathrm{O}(4)-\mathrm{C}(29)-\mathrm{C}(32)$ | 22.69(18) |
| $\mathrm{C}(1)-\mathrm{C}(6)-\mathrm{C}(19)-\mathrm{N}(1)$ | 44.4(2) | $\mathrm{C}(30)-\mathrm{O}(4)-\mathrm{C}(29)-\mathrm{C}(23)$ | 145.65(15) |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(19)-\mathrm{C}(18)$ | -19.4(2) | $\mathrm{N}(2)-\mathrm{C}(23)-\mathrm{C}(29)-\mathrm{O}(4)$ | -177.42(13) |
| $\mathrm{C}(1)-\mathrm{C}(6)-\mathrm{C}(19)-\mathrm{C}(18)$ | 167.12(15) | $\mathrm{C}(22)-\mathrm{C}(23)-\mathrm{C}(29)-\mathrm{O}(4)$ | 60.24(18) |


| $\mathrm{N}(2)-\mathrm{C}(23)-\mathrm{C}(29)-\mathrm{C}(32)$ | $-61.32(19)$ | $\mathrm{C}(31)-\mathrm{C}(33)-\mathrm{C}(34)-\mathrm{C}(35)$ | $93.8(2)$ |
| :--- | :---: | :--- | :---: |
| $\mathrm{C}(22)-\mathrm{C}(23)-\mathrm{C}(29)-\mathrm{C}(32)$ | $176.35(15)$ | $\mathrm{Si}-\mathrm{O}(6)-\mathrm{C}(35)-\mathrm{C}(34)$ | $118.25(17)$ |
| $\mathrm{C}(29)-\mathrm{O}(4)-\mathrm{C}(30)-\mathrm{O}(5)$ | $177.23(18)$ | $\mathrm{C}(33)-\mathrm{C}(34)-\mathrm{C}(35)-\mathrm{O}(6)$ | $-177.34(17)$ |
| $\mathrm{C}(29)-\mathrm{O}(4)-\mathrm{C}(30)-\mathrm{C}(31)$ | $-1.1(2)$ | $\mathrm{O}(6)-\mathrm{Si}-\mathrm{C}(38)-\mathrm{C}(41)$ | $-67.41(18)$ |
| $\mathrm{O}(5)-\mathrm{C}(30)-\mathrm{C}(31)-\mathrm{C}(33)$ | $31.8(3)$ | $\mathrm{C}(36)-\mathrm{Si}-\mathrm{C}(38)-\mathrm{C}(41)$ | $51.78(19)$ |
| $\mathrm{O}(4)-\mathrm{C}(30)-\mathrm{C}(31)-\mathrm{C}(33)$ | $-150.00(17)$ | $\mathrm{C}(37)-\mathrm{Si}-\mathrm{C}(38)-\mathrm{C}(41)$ | $172.04(17)$ |
| $\mathrm{O}(5)-\mathrm{C}(30)-\mathrm{C}(31)-\mathrm{C}(32)$ | $161.3(2)$ | $\mathrm{O}(6)-\mathrm{Si}-\mathrm{C}(38)-\mathrm{C}(39)$ | $52.0(2)$ |
| $\mathrm{O}(4)-\mathrm{C}(30)-\mathrm{C}(31)-\mathrm{C}(32)$ | $-20.5(2)$ | $\mathrm{C}(36)-\mathrm{Si}-\mathrm{C}(38)-\mathrm{C}(39)$ | $171.17(19)$ |
| $\mathrm{O}(4)-\mathrm{C}(29)-\mathrm{C}(32)-\mathrm{C}(31)$ | $-34.18(18)$ | $\mathrm{C}(37)-\mathrm{Si}-\mathrm{C}(38)-\mathrm{C}(39)$ | $-68.6(2)$ |
| $\mathrm{C}(23)-\mathrm{C}(29)-\mathrm{C}(32)-\mathrm{C}(31)$ | $-152.45(16)$ | $\mathrm{O}(6)-\mathrm{Si}-\mathrm{C}(38)-\mathrm{C}(40)$ | $173.14(16)$ |
| $\mathrm{C}(30)-\mathrm{C}(31)-\mathrm{C}(32)-\mathrm{C}(29)$ | $32.47(19)$ | $\mathrm{C}(36)-\mathrm{Si}-\mathrm{C}(38)-\mathrm{C}(40)$ | $-67.67(19)$ |
| $\mathrm{C}(33)-\mathrm{C}(31)-\mathrm{C}(32)-\mathrm{C}(29)$ | $159.16(19)$ | $\mathrm{C}(37)-\mathrm{Si}-\mathrm{C}(38)-\mathrm{C}(40)$ | $52.59(19)$ |
| $\mathrm{C}(30)-\mathrm{C}(31)-\mathrm{C}(33)-\mathrm{C}(34)$ | $78.7(2)$ |  |  |
| $\mathrm{C}(32)-\mathrm{C}(31)-\mathrm{C}(33)-\mathrm{C}(34)$ | $-41.5(3)$ |  |  |

Table 7. Hydrogen bonds for rich08 [ $\AA$ and ${ }^{\circ}$ ].

| D-H...A | $\mathrm{d}(\mathrm{D}-\mathrm{H})$ | $\mathrm{d}(\mathrm{H} \ldots \mathrm{A})$ | $\mathrm{d}(\mathrm{D} \ldots \mathrm{A})$ | $<$ (DHA) |
| :--- | :---: | :---: | :---: | :---: |
| $\mathrm{N}(1)-\mathrm{H}(1) \ldots \mathrm{O}(2)$ | 0.88 | 2.19 | $2.9767(19)$ | $149(1)$ |
| $\mathrm{N}(2)-\mathrm{H}(2) \ldots \mathrm{O}(1) \# 1$ | 0.88 | 2.06 | $2.8665(19)$ | $152(1)$ |

Symmetry transformations used to generate equivalent atoms:
\#1 x-1/2,-y+3/2,z

Brewer/James/Rich Supporting Information

