SUPPORTING INFORMATION

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- **S5**: ¹H NMR of compound **5**c
- S6: ¹³C NMR of compound 5c
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- S11: ¹H NMR of compound 6bI
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- S22: Energy and Cartesian coordinates for structure 5a(H⁺)A
- S23: Energy and Cartesian coordinates for structure 5a(H⁺)B
- S24: Energy and Cartesian coordinates for structure 5d(H⁺)A
- S25: Energy and Cartesian coordinates for structure $5d(H^+)B$

General experimental methods

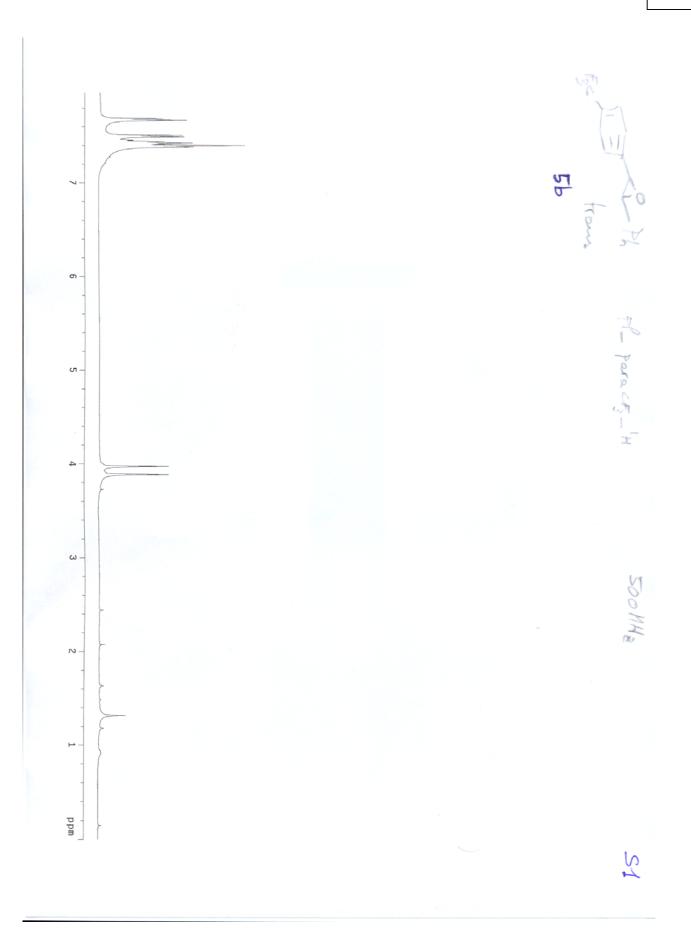
¹H and ¹³C NMR were recorded at 300 (or 500) and 75 (or 125) MHz respectively. MS spectra were recorded on a chromatograph equipped with a MS capillary column (30 m x 0.25 mm i.d. x 0.25 μ m film thickness; injection temperature: 250°C, column temperature program: 100°C for 3 min, then 10°C min⁻¹ until 250°C) and a mass selective detector (mass range: 15-800 a.m.u.; scan rate: 1.9 scans s⁻¹; EM voltage: 1435). Commercially available reagents were used without further purification. All reactions were monitored by TLC with silica gel-coated plates. Column chromatography was carried out using 60-240 mesh silica gel at atmospheric pressure.

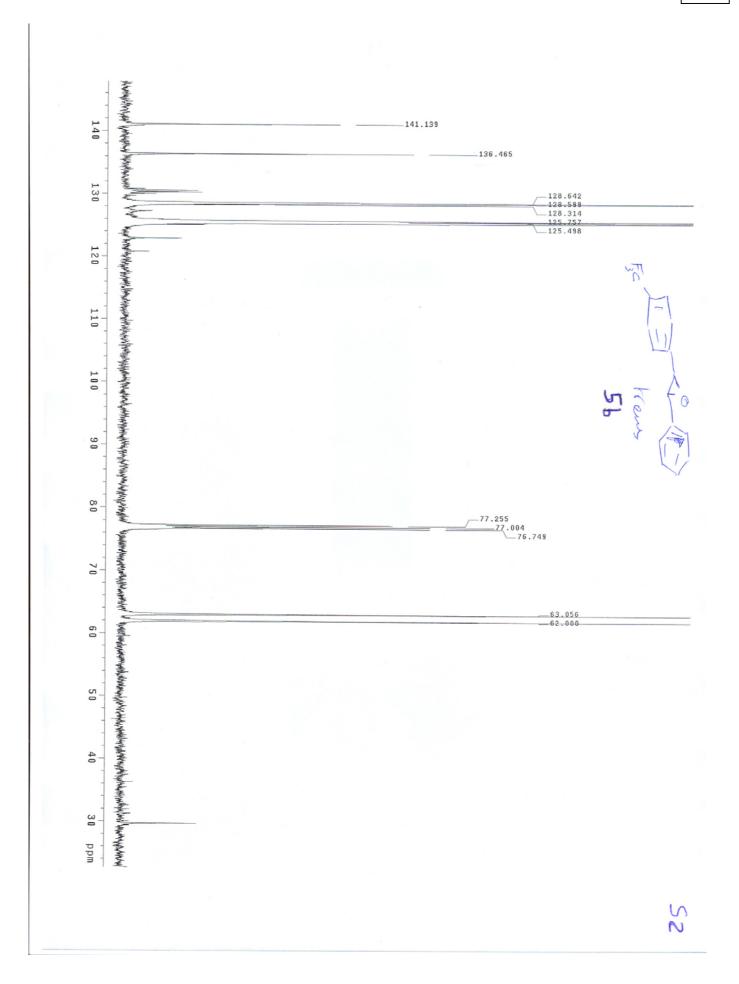
General procedure for the synthesis of epoxides 5(a-d), 9 and 10.

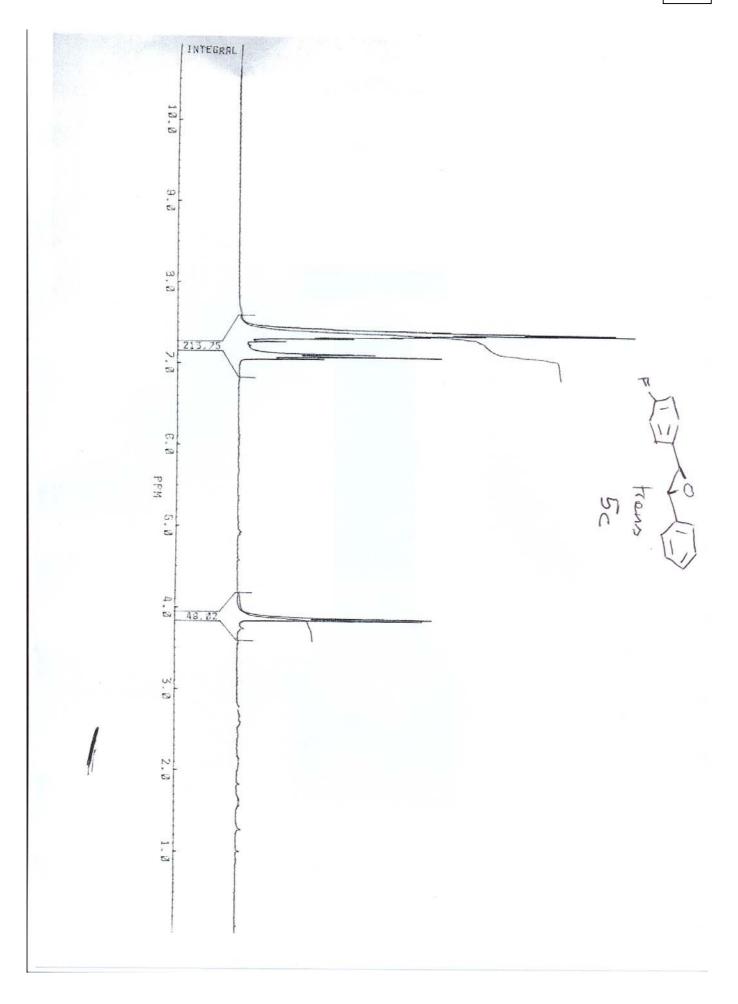
NaOH (7.1 mL of 50% aqueous solution) was added dropwise at 0°C to a stirred solution of benzyldimethylsulfonium chloride (1.3 equiv., 10.5 mmol), the appropriate aldehyde (1 equiv.) and tetrabutylammonium hydrogensulfate (0.05 equiv.) in CH_2Cl_2 (15 mL). After 5h the reaction is quenched and worked up as usual.¹³

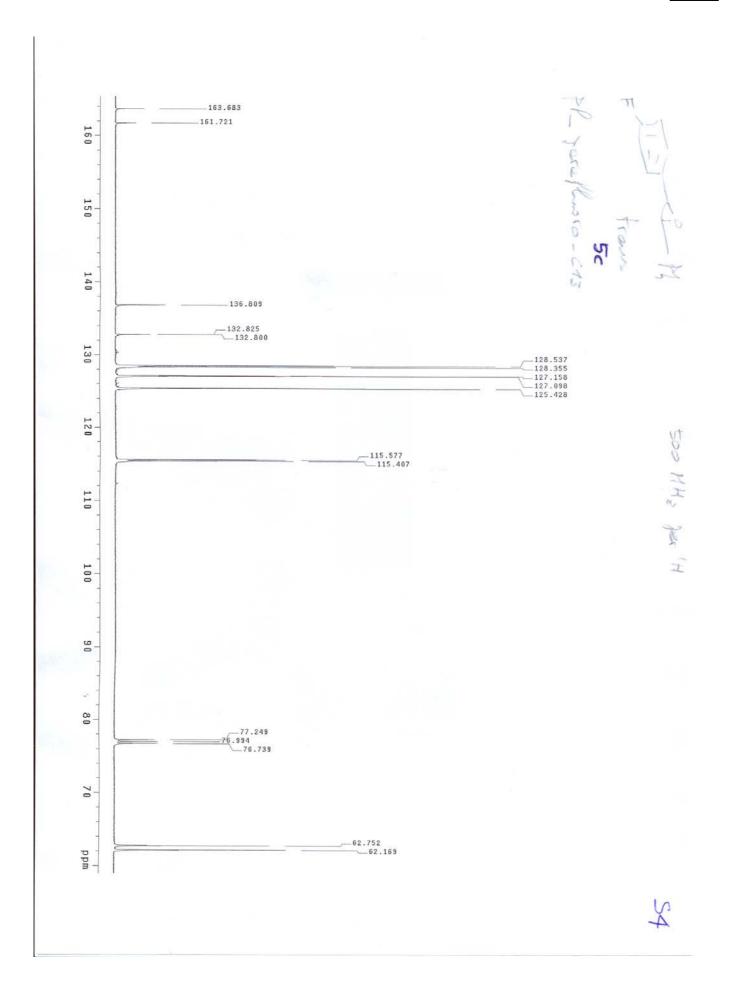
General procedure for the LiBr opening of (*E*)-2,3-diaryloxiranes. To a solution of the epoxide (0.50 mmol) in 10 mL of acetonitrile, kept at the desired temperature, were added solid LiBr (2-4 equiv.) and Amberlyst 15 (220-440 mg/mmol) consecutively in single portions. The mixture was stirred at the desired temperature and monitored by TLC until complete. The mixture was filtered and the filtrate was evaporated under vacuum. The residue was dissolved in Et₂O and the resulting organic layer was dried with Na₂SO₄. The solvent was evaporated under vacuum and the crude product was analyzed by NMR.

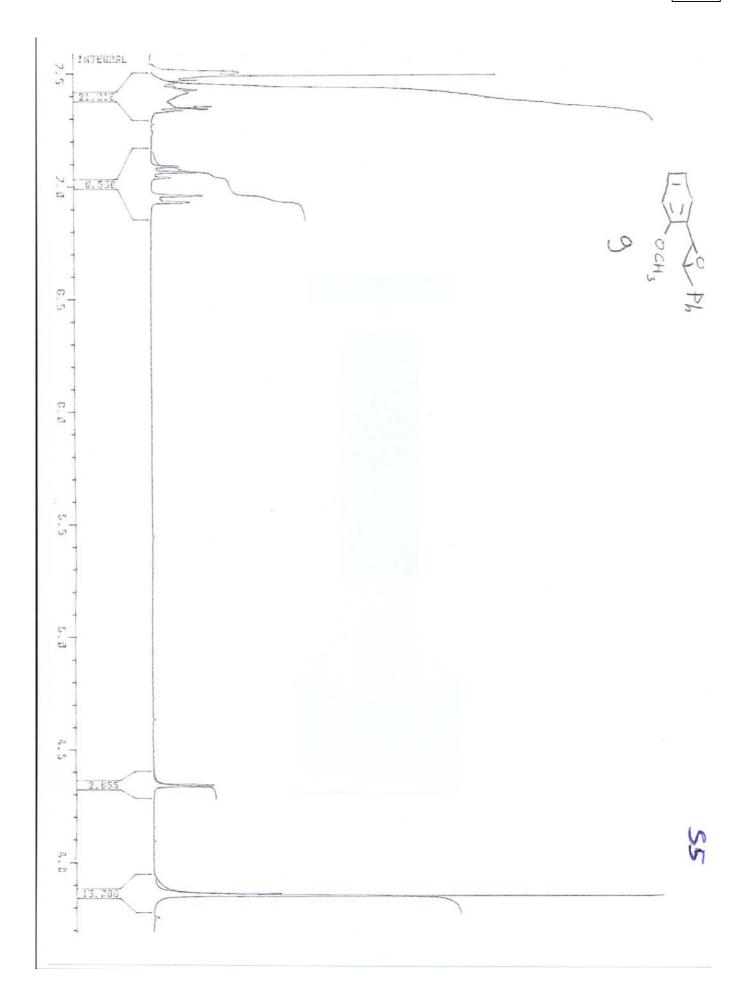
General procedure for the alkali ring closure of bromohydrins mixtures. To a solution of the bromohydrins (0.10 mmol) in 2 mL of anhydrous THF, kept at 0°C, was added NaH (1.5 equiv.) in a single portion. The mixture was stirred and monitored by TLC until complete. The mixture was quenched by cold water and the organic layer extracted by Et_2O , dried with Na₂SO₄. The solvent was evaporated under vacuum and the crude product was analyzed by ¹H NMR.

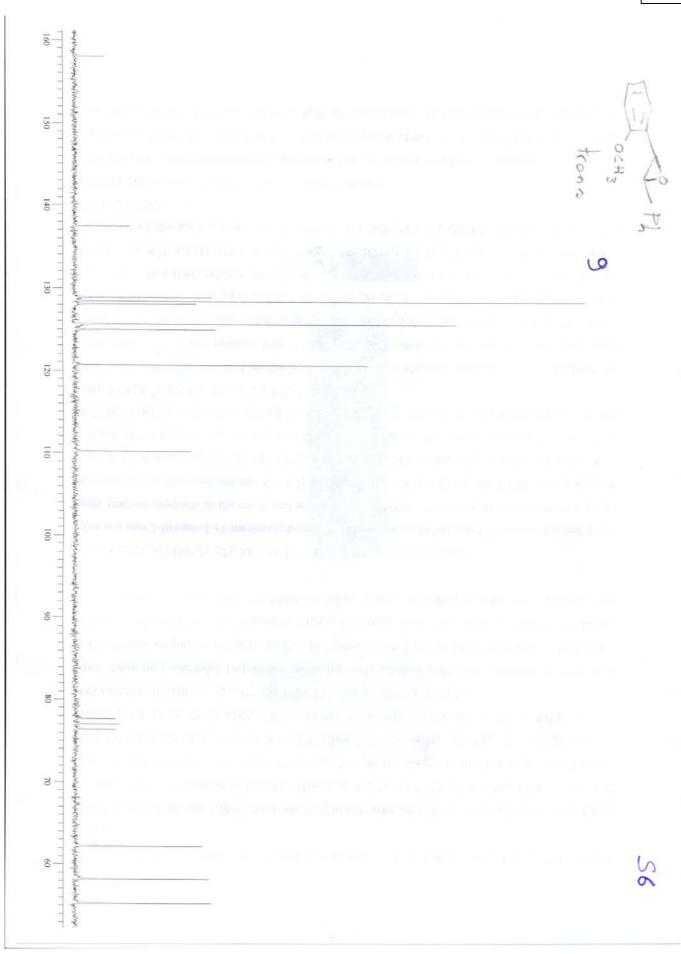


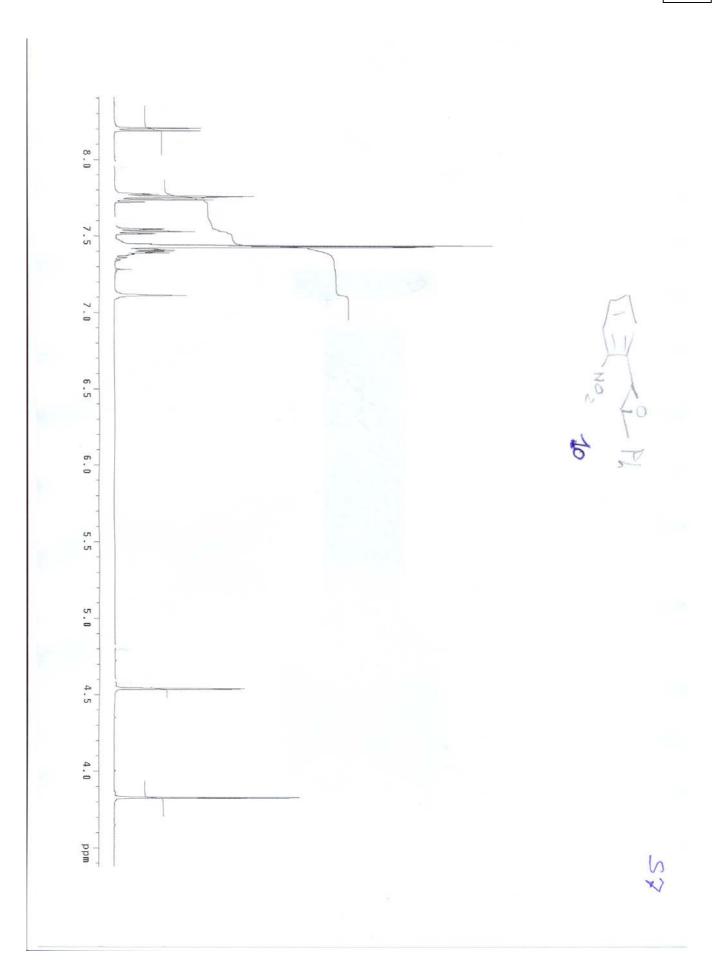


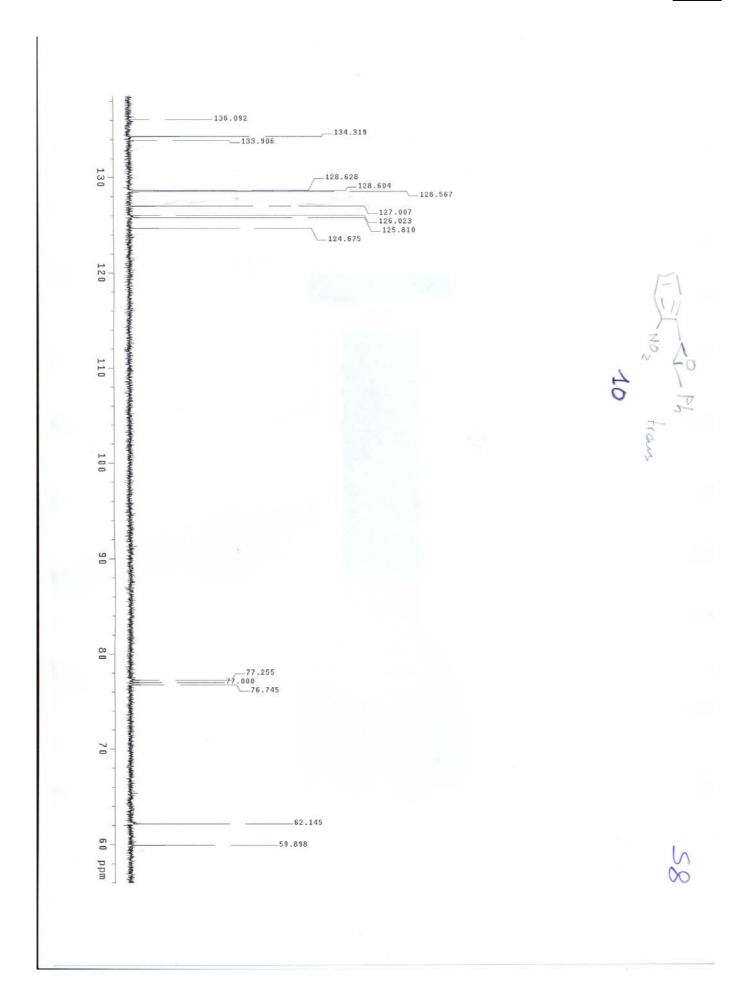


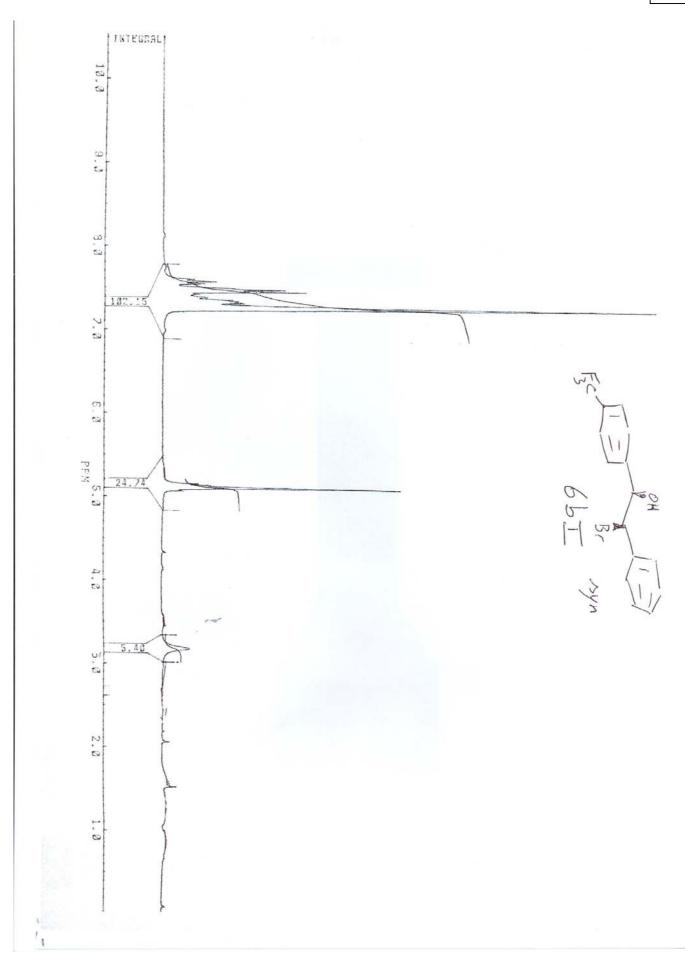


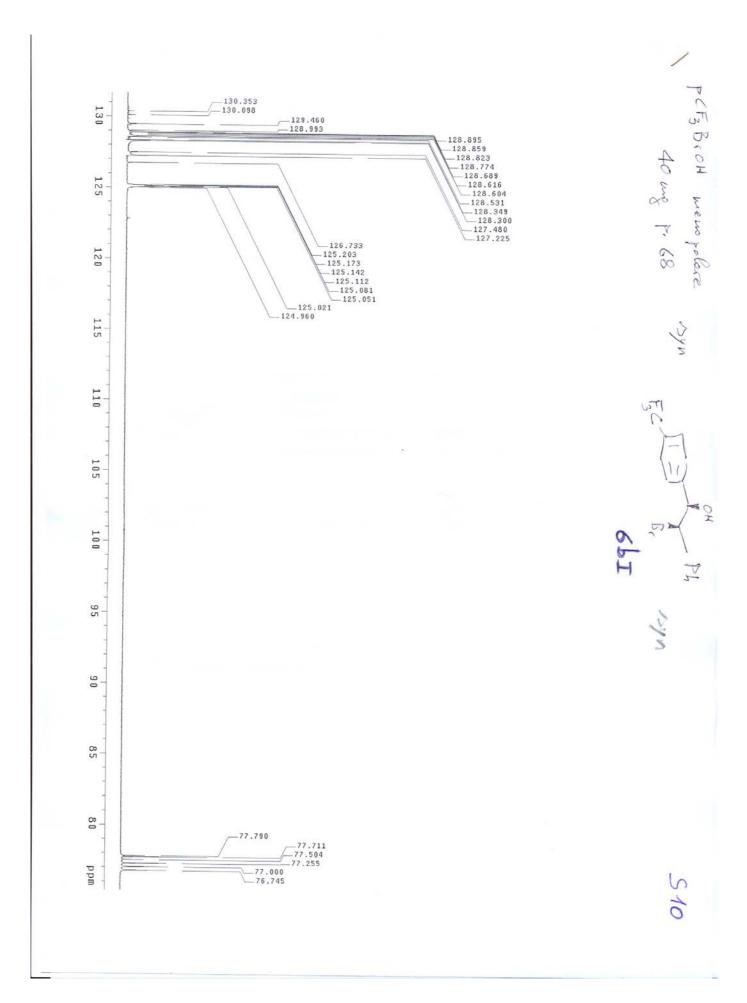




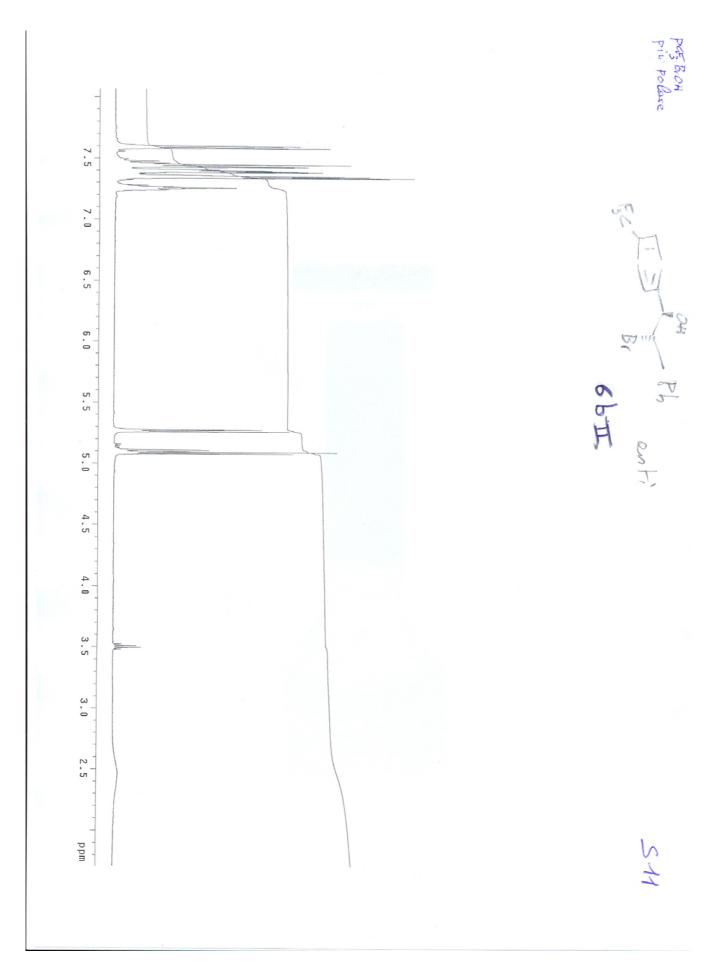


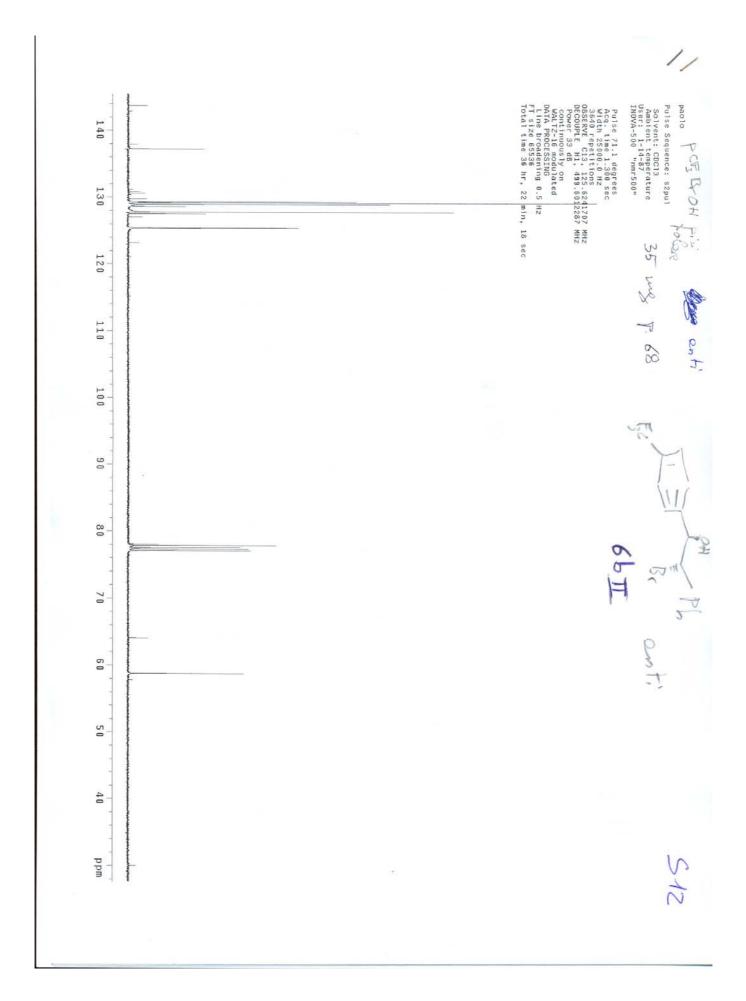




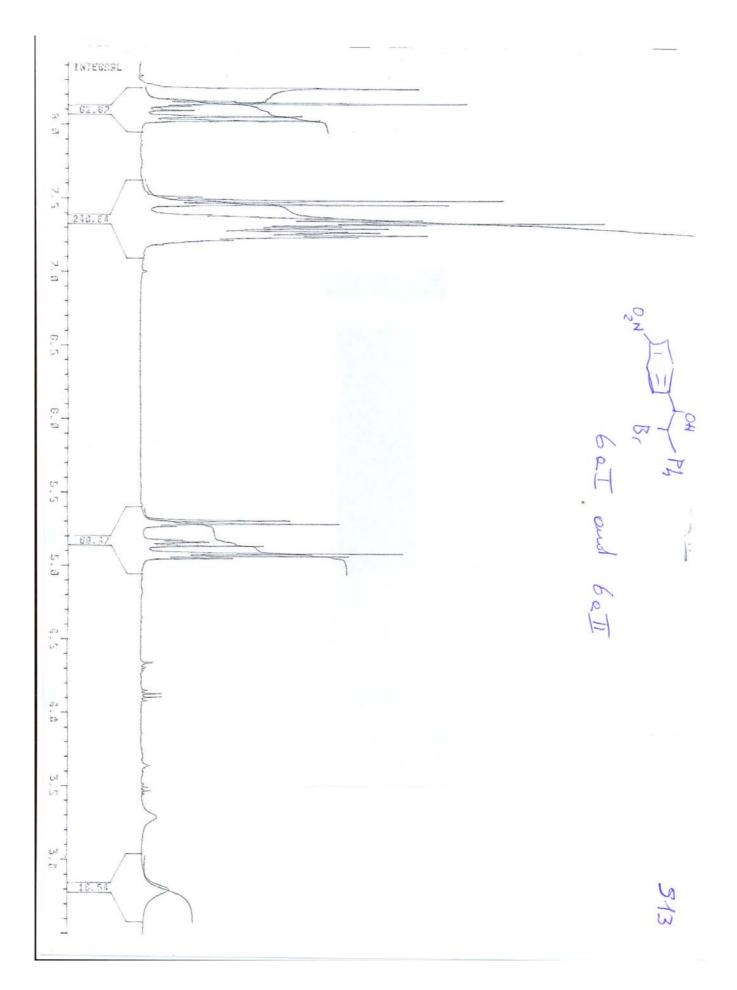


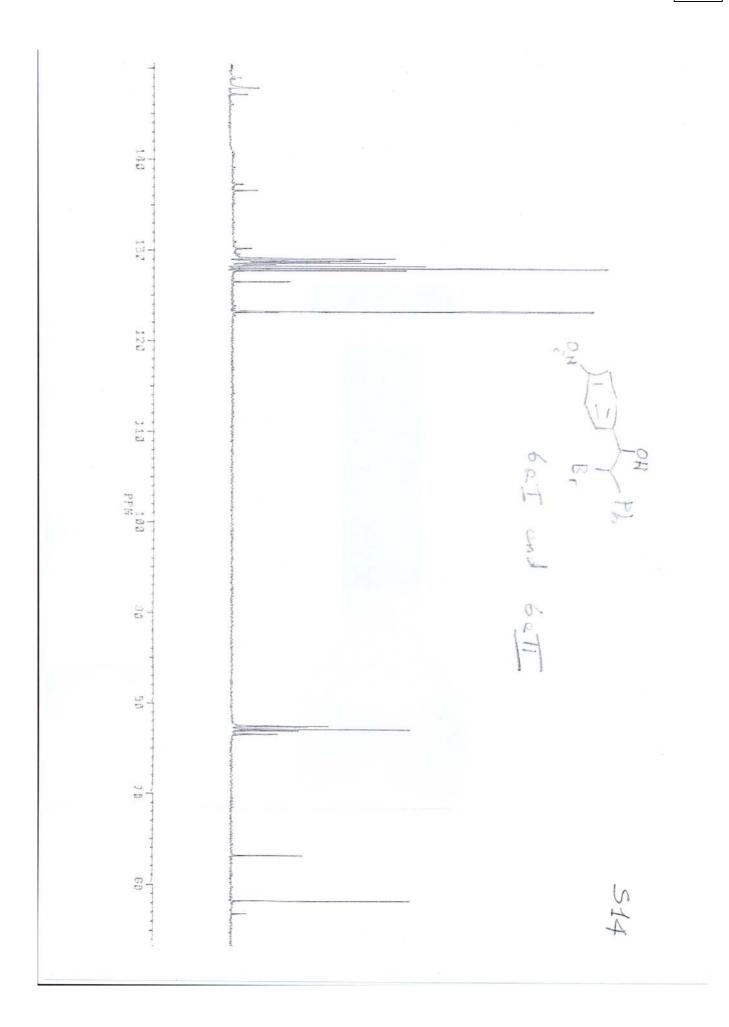
<u>S1</u>2

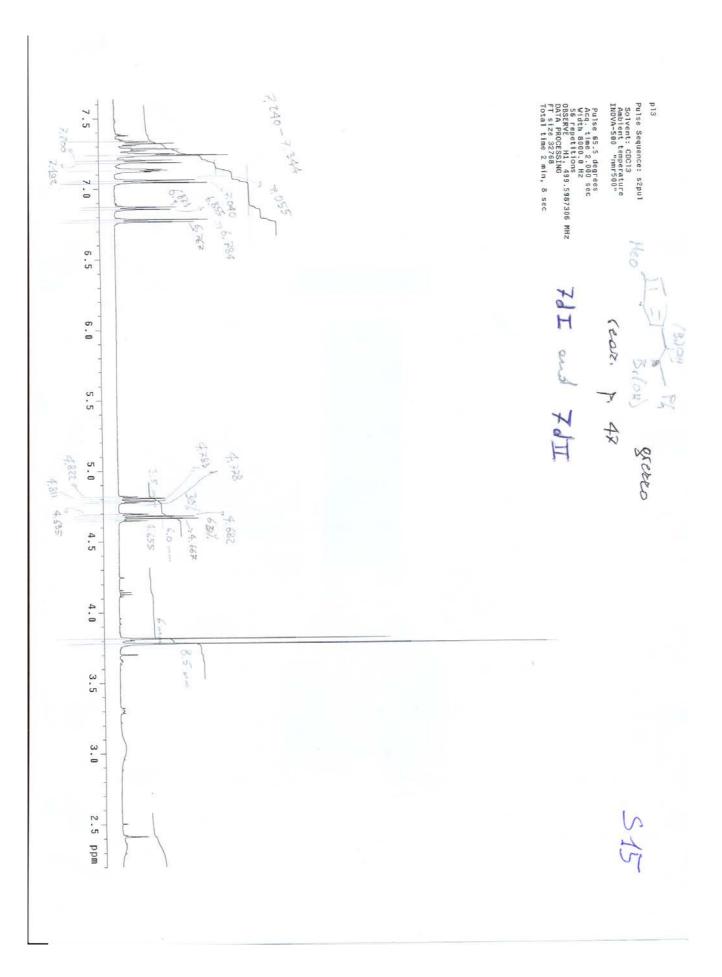


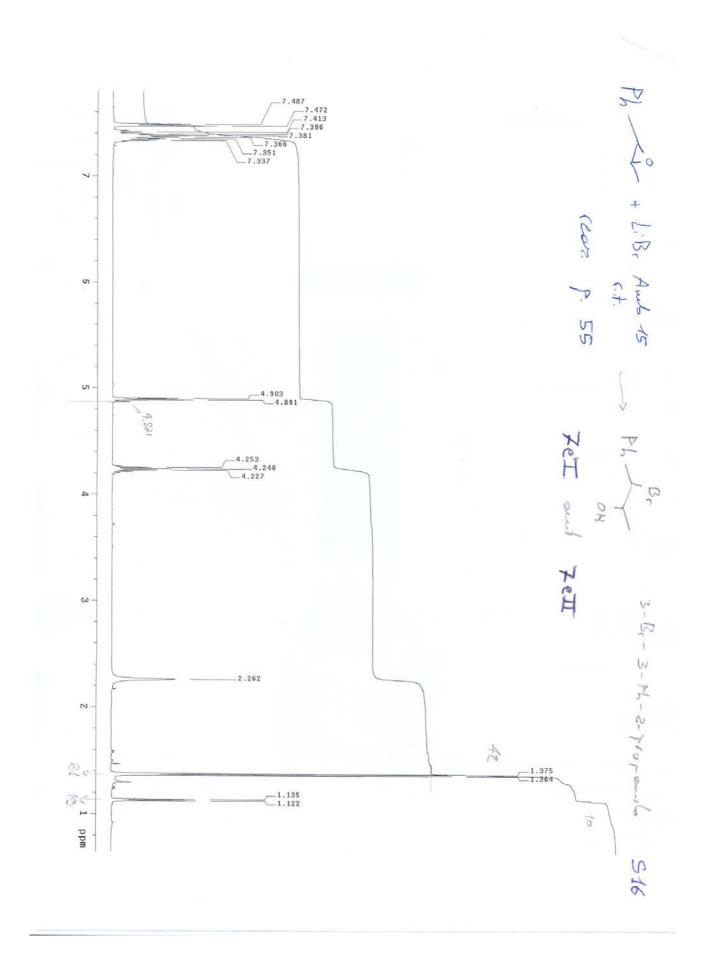


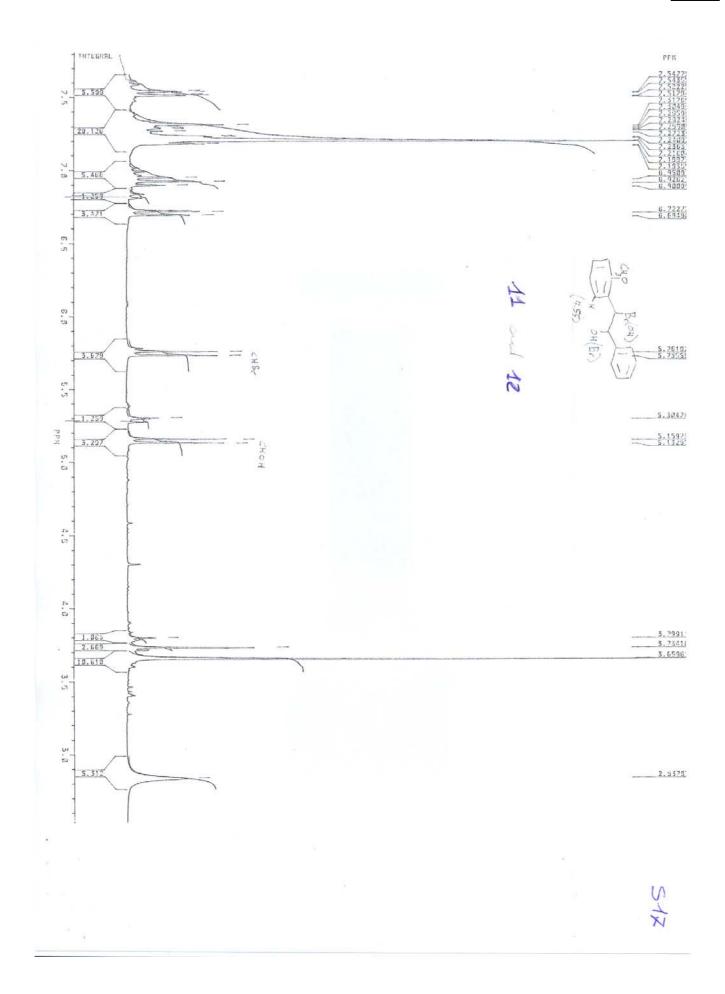
<u>S1</u>4

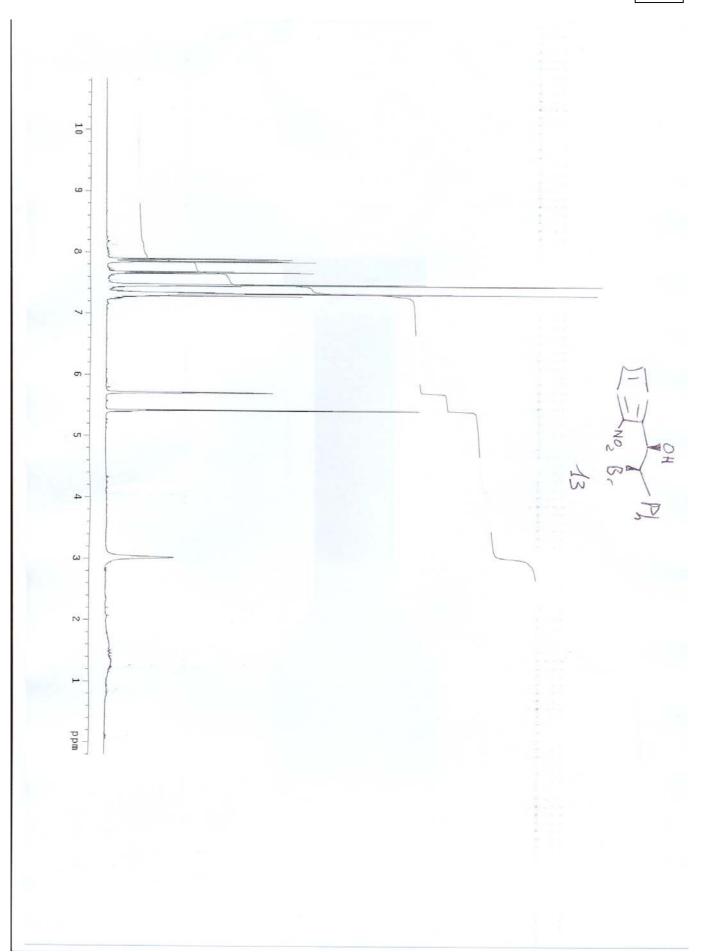


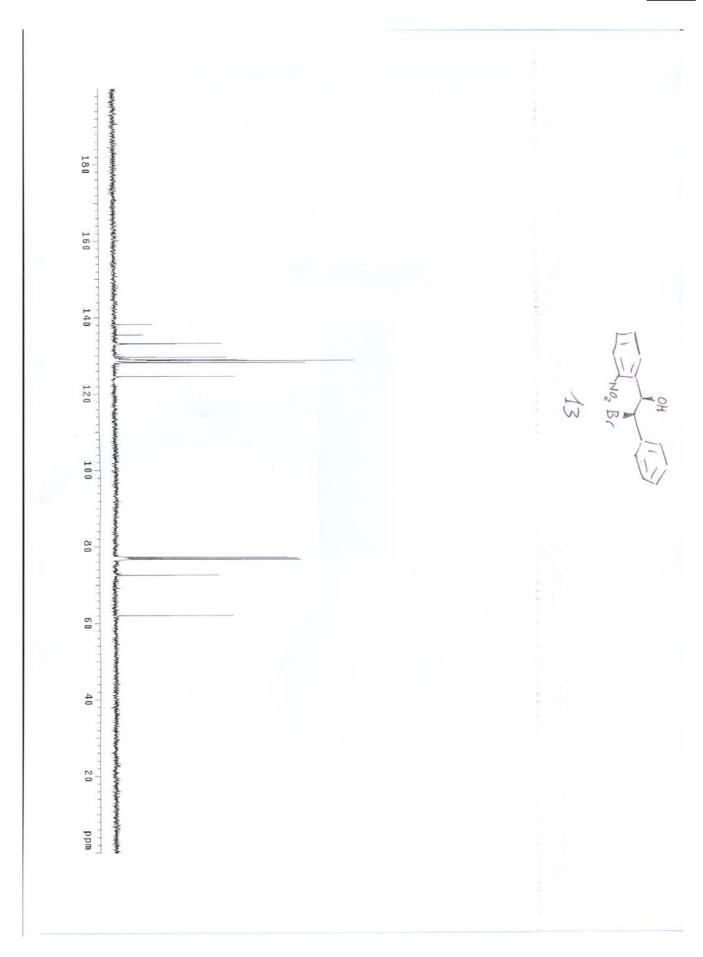












Geometry of structure 7a(H+)A

b3lyp/6-31g(d)

Η

С

С

С

C

С

С

Η

Η

Н

С

Η

N

O C H C

С С С С С Н Н Н Н Н О Н

Sum of electronic and thermal Free Energies E=-820.560887(Hartr ee/Particle)

1.45495100	2.12756900	1.77615400
1.62786200	1.37471900	1.01132000
2.09555900	-0.53398400	-0.99318200
0.64299900	1.09363200	0.05411200
2.85114300	0.71207200	0.96669700
3.06072300	-0.23238800	-0.03598300
0.87234100	0.12842400	-0.93790400
3.63940800	0.91996800	1.68010800
0.10822100	-0.08381500	-1.67892800
2.31012000	-1.27454900	-1.75407300
-0.71434700	1.83625000	0.09020200
-0.59340500	2.72972100	0.72616500
4.36056600	-0.94260700	-0.08595300
4.50949900	-1.76990200	-0.97940800
5.18778900	-0.65236000	
-1.57428100	0.88299000	0.77182500
-1.24779800	0.75871500	0.83191300
-2.65622100	0.09952500	1.86354500
-4.85700500	-1.51694700	0.43475700
-3.19871800	-0.80356800	-0.17242000
-3.26668800		1.41308400
-4.35162500	0.16398500	-0.86448200
-4.28299200	-0.64077500	-1.14956500
-2.74086900	-1.59876000	1.10873700
	-0.85300000	2.39703000
-2.87243800	0.85439300	-1.59756300
-4.82120000	-0.59482300	-2.12658000
-4.69222500	-2.28051700	1.84667000
-5.71308600	-2.14217200	-0.41128000
-1.22812200	2.15926100	-1.17247500
-0.60145000	2.74971800	-1.62283000

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pNO2B

Geometry of structure 7a(H+)B

b3lyp/6-31g(d)

Η

С

С

С

С

С

С

Η

Η

Η

С

Η

С

C

C

C

С

C

C

Η

H

Η

Η

Η

Ν

0

0

Η

0

H

Sum of electronic and thermal Free Energies E=-820.554404 (Hart ree/Particle)

-3.93514200 0.26594900 -1.89432500 -3.72772500 -0.18558900 -0.92703100 -3.22109100 -1.31281000 1.59015600 -2.74104400 0.36284000 -0.08476800 -4.46938500 -1.27895500 -0.49606500 -4.21631600 -1.84321900 0.76102900 -2.46775000 -0.22405900 1.16490600 -5.24943100 -1.68771000 -1.13049500 -1.69647100 0.19540200 1.80279000 -3.03293900 -1.75219900 2.56468600 -1.95867200 1.63567100 -0.52867100 -2.54432900 2.16833400 -1.28907800 -0.89888500 0.83186000 -1.14983300 0.40645600 0.50363300 -0.70756300 3.01399100 -0.16283500 -0.05320000 1.13668000 -0.43594300 -1.49427900 1.03819300 1.10335200 0.42205400 2.34218900 0.76668300 0.74414000 2.43925700 -0.77150600 -1.17021600 0.66318700 -0.89298700 -2.35830400 0.49913300 1.83302100 1.01087500 2.85311400 1.20826600 1.59112500 3.01491400 -1.48144500 -1.75142100 -4.79642300 -2.69855800 1.09397600 4.41560300 -0.52037300 0.30384100 4.88632900 0.02152300 1.29625900 4.97964800 -1.32834400 -0.42367000 -1.20241100 0.38879600 -2.09532800 -1.52190800 2.46324000 0.50341000 -2.29037100 2.90965000 0.89665700

Suppl3

Geometry of structure 7d(H+)A

Η

C

C

C

C

C

C

Η

Η

Η

0

С

Η

Η

Η

C

Η

0

Η

C

H

C

С

С

С

C

С

Η

Η

Η

Η

Η

b3lyp/6-31g(d) geom=connectivity

Sum of electronic and thermal Free Energies $E{=}{-}730.571423~({\rm Hart\,ree/Particle})$

2.19302900

1.41233800

-0.56856400

1.20111700

0.66415400

-0.33602100

0.18243500

0.83167500

0.00216800

-1.32775100

-0.99943400

-2.02327300

-1.61225000

-2.38866600

-2.84099700

2.07404100

3.00509300

2.29755500

2.86945000

1.11764800

1.15683500

0.15039700

-1.74325200

0.04226100

-0.72020800

-1.65370600

-0.89714400

0.71213700

-0.64066100

-2.31034900

-0.97435700

-2.47365800

1.83095600

1.09149300

-0.84839700

0.06536200

1.13794900

0.16584400

1.89989800

-1.66023600

-1.60272000

0.29699000

-0.64488500

-1.65747000

-0.32597300

-0.61823700

-0.01924700

0.54072000

-1.30627800

-1.78341500

0.77508200

1.83538700

0.39637300

-0.14644600

-0.91456300

1.42032200

1.14953800

-1.16991900

-1.69159800

2.42139900

1.93344900

-2.16400200

-0.35988800

-0.87916800

57441300
.73683700
2.17983100
.78707100
.89356500
.13198900
.01493700
.64725100
.28293300
.34694000
.28488500
.63317900
.70484000
.60854800
.90506800
.48334700
.32373900
.98101600
.35726000
.27333800
.03516000
.23162400
.22393100
.78561300
.71200300
.69610100
.76959000
.44262900
.29709400
.05892000
.19813800
.99917200

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Suppl4

Geometry of structure 7d(H+)B

b3lyp/6-31g(d)

Η С С C С С С Η Н Η 0 C Н Η Η С Н С Η 0 Η С С C С С С Η Η Η Η Η

Sum of electronic and thermal Free Energies E=-730.583531 (Hart ree/Particle)

1.27297500	-0.92952400	2.22905900
1.67247600	-0.46060300	1.33411600
2.65965800	0.77261800	-0.98745300
0.82308700	0.42629700	0.58054600
2.96063200	-0.72792300	0.95400100
3.47357100	-0.11169800	-0.22175100
1.37652400	1.04013500	-0.60376000
3.57598300	-1.39834000	1.54120800
0.75896200	1.72144200	-1.17302700
3.09702900	1.22523900	-1.87081500
4.69386900	-0.29308700	-0.68334100
5.63226200	-1.15721700	-0.00465300
5.83196000	-0.78216900	1.00284200
6.53836700	-1.11512000	-0.60626400
5.24967400	-2.18099200	0.02823800
-0.45793400	0.63874400	1.05784200
-0.71896100	0.12404400	1.98132000
-1.60588200	1.41071100	0.51197400
-2.01854300	2.00077200	1.34870200
-1.22433100	2.24124900	-0.55890300
-2.02936900	2.61675100	-0.95161100
-2.64434900	0.33771600	0.13670200
-4.52076000	-1.60742100	-0.56240200
-3.72028900	0.06449300	0.99049300
-2.49918400	-0.37774000	-1.06081200
-3.44421700	-1.34008100	-1.41219500
-4.65810500	-0.90514900	0.63815200
-3.83871400	0.62283100	1.91635400
-1.66449200	-0.16265000	-1.72176300
-3.33990200	-1.88293400	-2.34685000
-5.50140300	-1.10295300	1.29286500
-5.25480100	-2.35898900	-0.83711200

S25

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