



We planned to choose an appropriate anhydride that possesses both sufficient reactivity and stability. To enhance the reactivity of the carbonyl position **A** ($\text{R}^1\text{C}=\text{O}$), the counter carbonyl position **B** ($\text{RC}=\text{O}$) should have an electron withdrawing group. In addition, the R group should be sterically bulky to block undesirable attack by *N*-methylimidazoles **1a-d**. Thus, screening was performed as shown in the following table. First, TsCl was examined, because this reagent is reported to be a good promoter of the esterification and amide formation (entry 1).¹¹ This trial, however, was disappointing. Next, we turned our attention to a few acyl-activating reagent, and determined that the use of $\text{Cl}_3\text{CC}(=\text{O})\text{Cl}$ and NaH was the most effective (entry 5).

$\begin{array}{c} \text{i) base} \\ \text{ii) } \text{R}^2\text{X} \\ \xrightarrow{\hspace{1cm}} \left[\text{R}^1\text{C}(=\text{O})\text{OR}^2 \xrightarrow{\text{1a}} \text{R}^1\text{C}(=\text{O})\text{N}^+ \text{imidazole ring-C(=O)-NMe-RCO}_2^- \right] \\ \left(\text{R}^1\text{CO}_2\text{H} = \text{cyclohexyl-CH}_2\text{-CH}_2\text{CO}_2\text{H} \right) \end{array}$

$\begin{array}{c} \text{Bu-CH}_2\text{-CO}_2\text{Me} \\ (1.0 \text{ eq.}) \\ \xrightarrow[\text{CH}_2\text{Cl}_2, -45^\circ\text{C}, 30 \text{ min}]{\text{TiCl}_4 - \text{Bu}_3\text{N}} \\ \text{R}^1\text{C}(=\text{O})\text{CH}_2\text{-Bu-CO}_2\text{Me} \text{ and } \text{Bu-CH}_2\text{-C(=O)-CH}_2\text{-Bu-CO}_2\text{Me} \\ \text{cross} \quad \text{self} \end{array}$

entry	R^2X	base	yield (%) ^a	cross / self^b
1	TsCl	Et_3N	trace ^c	ca. 1 / 1
2	$(\text{CF}_3\text{CO})_2\text{O}$	Et_3N	trace	ca. 1 / 1
3		Et_3N	0	—
4	CCl_3COCl	Et_3N	73	91 / 9
5	CCl_3COCl	NaH	76	95 / 5

^a Isolated. ^b Determined by ^1H NMR of the crude product. ^c $(\text{R}^1\text{CO})_2\text{O}$ was formed in 38% yield.

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