Supporting Information for:

Novel Polyphenylenes Containing Phenol Substituted Oxadiazole Moieties as Fluorescent Chemosensors for Fluoride Ion

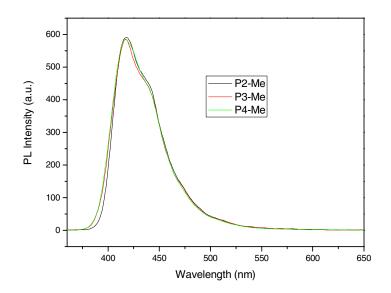
Gang Zhou, Yanxiang Cheng, Lixiang Wang, Xiabin Jing and Fosong Wang

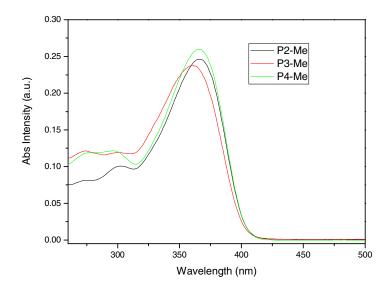
Contribution from the State Key Laboratory of Polymer Physics and Chemistry, Changchun Institute of Applied Chemistry, Chinese Academy of Sciences, Changchun 130022, P.R. China

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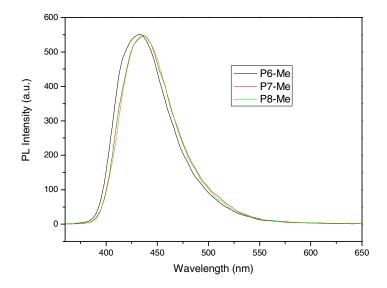
- 1. Fluorescence and absorption spectra of polymers **P2-P4** before their demethylation.
- 2. Fluorescence and absorption spectra of polymers **P6-P8** before their demethylation.
- 3. Fluorescence and absorption spectra of polymers **P2-P4** (5µM) in chloroform
- 4. Fluorescence and absorption spectra of polymers **P6-P8** (5μM) in chloroform
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- 6. Fluorescence titration-general procedure.
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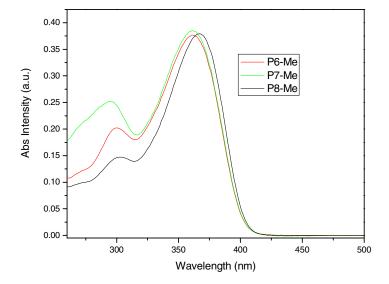
1. Fluorescence and absorption spectra of polymers **P2-P4** before their demethylation.



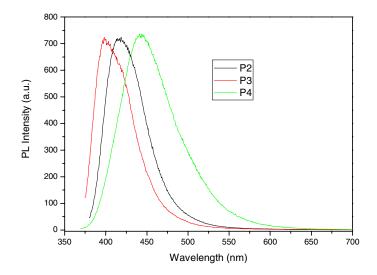


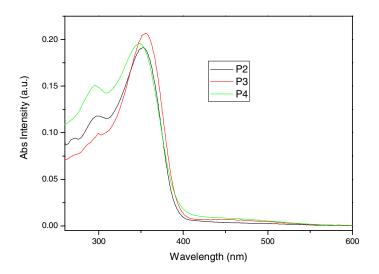
2. Fluorescence and absorption spectra of polymers **P6-P8** before their demethylation.



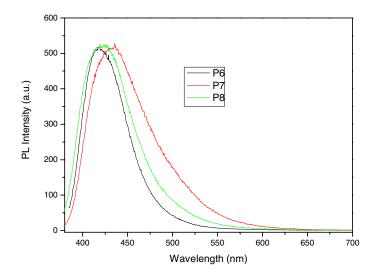


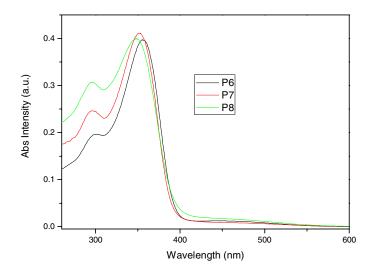
3. Fluorescence and absorption spectra of polymers **P2-P4** ($5\mu M$) in chloroform



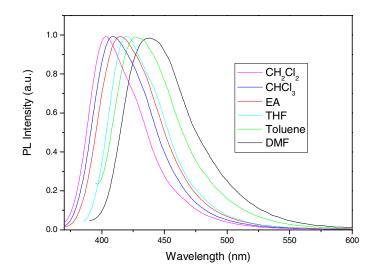


4. Fluorescence and absorption spectra of polymers P6-P8 $(5\mu M)$ in chloroform





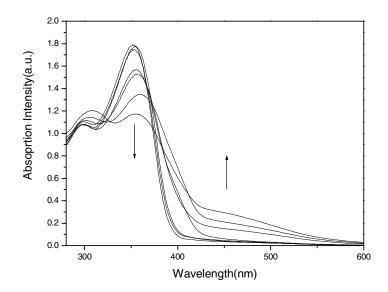
5. Fluorescence spectra of polymer $\mbox{\bf P3}\,(5\mu\mbox{\bf M})$ in different solvents



6. Fluorescence titration-general procedure:

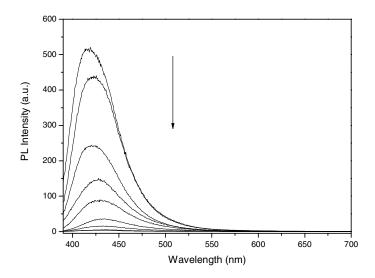
Fluorescence spectra were obtained on a Perkin-Elmer LS 50B luminescence with discharge lamp excitation. Measurements spectrometer Xenon Ultraviolet-Visible absorption spectra were carried out on Perkin-Elmer Lambda 35 UV/Vis spectrometer. Solution of polymers P1-P10 were dilute at a concentration of 5×10⁻⁶ M (the concentration of polymers is based on the moles of the repeated unit) in CHCl₃. Solutions of tetrabutylammonium anionic salts with the concentration of typically 1×10⁻³ M were prepared by successive dilution. Fluorescence titration was carried out by sequentially adding 0.1mL aliquots of anions solutions in CHCl₃ to 5.00 ml of the polymer solutions, and CHCl₃ was added to acquire 10.00ml solutions. The solutions were stirred for 5 min prior to obtaining fluorescence spectra.

7. Absorption spectra for the titration of ${\bf P2}$ with ${\bf Bu_4NF}$.

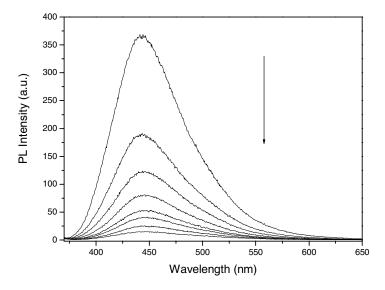


8. Fluorescence spectra for the titration of **P2**, **P4**, **P6-P8** with Bu₄NF.

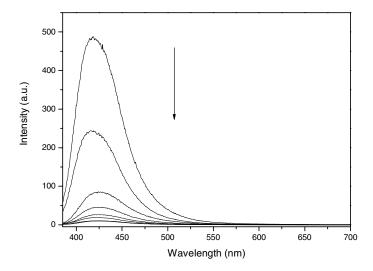
Fluorescence emission for titration of P2 with 0-100 μM Bu₄NF in CHCl₃



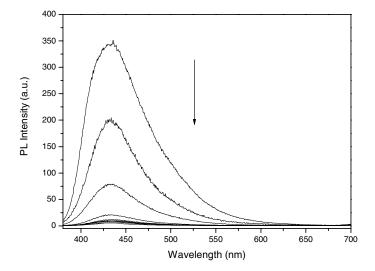
Fluorescence emission for titration of P4 with 0-100 μM Bu₄NF in CHCl₃

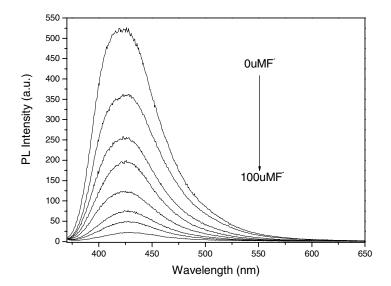


Fluorescence emission for titration of **P6** with 0-100 µM Bu₄NF in CHCl₃



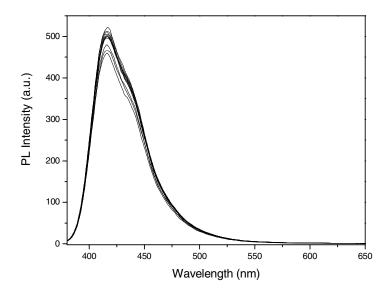
Fluorescence emission for titration of $\boldsymbol{P7}$ with 0-100 μM Bu_4NF in $CHCl_3$





9. Absorption and fluorescence spectra for the titration of P1+P5 with Bu₄NF.

Fluorescence emission for titration of P1+P5 with 0-100 μM Bu₄NF in CHCl₃



Absorption spectra for titration of P1+P5 with 0-100 μM Bu₄NF in CHCl₃

