

## *Supporting Information*

# Reversible and Persistent Electrical Bistability in Single-Crystals of a Self-Assembled $\pi$ -Conjugated Tetraaryl System: A Sub-Micron Scale Electrical Characterization

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Technion City, 32000, Haifa, Israel.

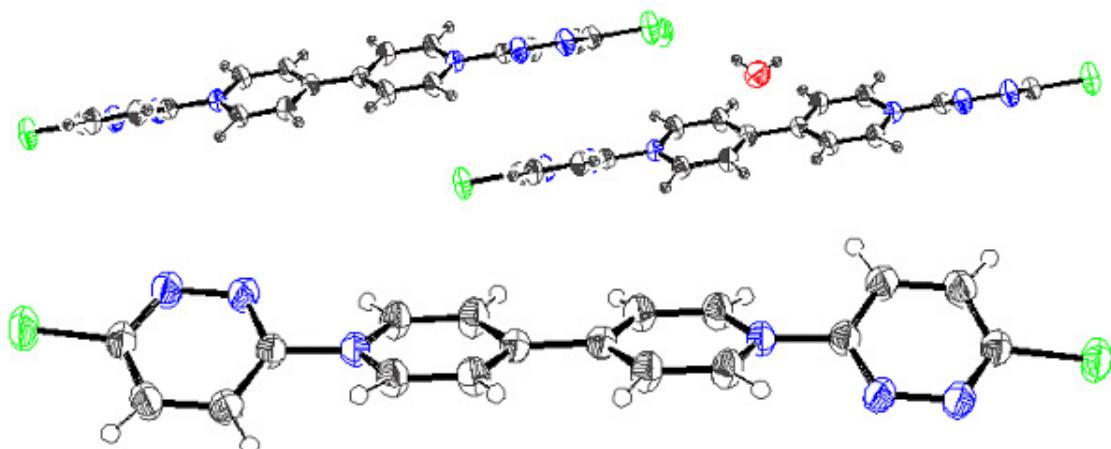
Fax: +927-4-8295307; Tel: +927-4-8293708; E-mail: chryoav@techunix.technion.ac.il

### **Materials $2^{2+} \cdot 2\text{PF}_6^-$ and $2^{2+} \cdot 2\text{Cl}^-$**

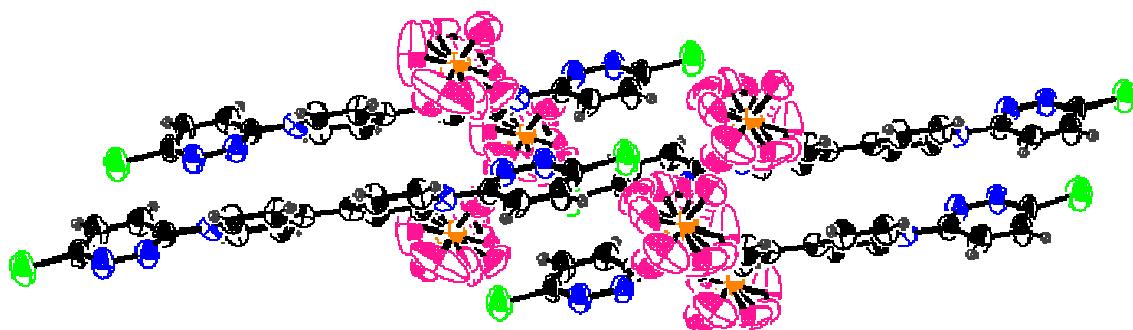
A 10:1 mixture of 3,6-dichloropyridazine and 4,4'-bipyridine was heated to 120°C under an inert atmosphere for two days. The crude was chromatographed (Silica, 5gL<sup>-1</sup> ammonium chloride in a

2:3 water:methanol solution). The fraction of the eluent that contained the product were combined and concentrated under reduced pressure. The product was precipitated from the concentrated solution by adding a concentrated solution of ammonium hexafluorophosphate. The resulting precipitate, N,N'-(3,3'-di-(6-chloropyridazine))-4,4'-bipyridinium bis hexafluorophosphate,  $\mathbf{2}^{2+}\cdot\mathbf{2PF}_6^-$ , was obtained in 80% yield. The dichloride salt  $\mathbf{2}^{2+}\cdot\mathbf{2Cl}^-$  was obtained by adding a saturated acetone solution of  $\mathbf{2}^{2+}\cdot\mathbf{2PF}_6^-$  to an acetone solution of tetrabutyl ammonium chloride.

$^1\text{H}$  NMR (DMSO-d<sub>6</sub>):  $\delta_{\text{ppm}}$  9.92 (4H, d), 9.12 (4H, d), 8.76 (2H, d), 8.56 (2H, d).  $^{13}\text{C}$  NMR (DMSO-d<sub>6</sub>):  $\delta_{\text{ppm}}$  158.69, 156.15, 151.4, 144.59, 132.55, 127.1, 126.97.

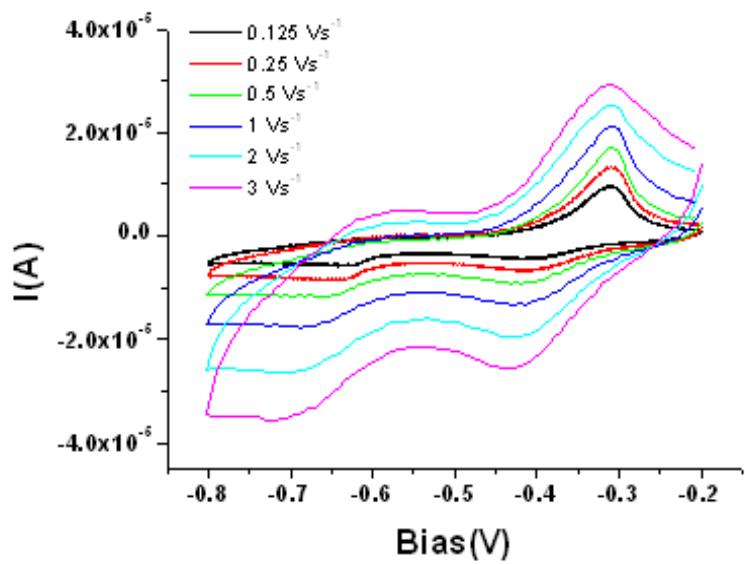


**Supplementary Figure 1.** Crystal structure of  $\mathbf{2}^{2+}\cdot\mathbf{2Cl}^-$ .



**Supplementary Figure 2.** Crystal structure of  $\mathbf{2}^{2+}\cdot\mathbf{2PF}_6^-$ .

## Material 1<sup>2+</sup>



**Supplementary Figure 3.** Cyclic voltammogram of **1** in acetonitrile in different scan rates. Upon decreasing the scan rate, the oxidation peak of the doubly reduced species disappears while the oxidation of the singly reduced species is increased, an indication of a disproportionation reaction.  
Electrolyte=1\*10<sup>-2</sup>M N(Bu)<sub>4</sub><sup>+</sup>PF<sub>6</sub><sup>-</sup>; scan rate=0.125 - 3V s<sup>-1</sup>; reference electrode=Ag/AgNO<sub>3</sub>; working and counter electrodes=Pt.

Cif file of  $\mathbf{1}^{2+}\cdot\mathbf{2I}_3^- \cdot \mathbf{A}$  :

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S.Mackay, C.J.Gilmore, C.Edwards, M. Tremayne, N. Stuart, K.Shankland
"maXus": a computer program for the solution and refinement of crystal
structures from diffraction data University of Glasgow, Scotland, UK,
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HKL Denzo and Scalepack (Otwinowski & Minor 1997)
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S.Mackay, C.J.Gilmore, C.Edwards, M. Tremayne, N. Stuart, K.Shankland
"maXus": a computer program for the solution and refinement of crystal
structures from diffraction data University of Glasgow, Scotland, UK,
Nonius BV, Delft, The Netherlands and MacScience Co. Ltd.,
Yokohama, Japan (1998)
;
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;
Molecular Structure Corporation (1999).
ORTEP. TEXRAY Structure Analysis Package.
MSC, 3200 Research Forest Drive, The
Woodlands, TX 77381, USA.
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; Refinement of F^2^ against ALL reflections. The weighted R-factor wR and
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; on F, with F set to zero for negative F^2^. The threshold expression of
; F^2^ > 2sigma(F^2^) is used only for calculating R-factors(gt) etc. and is
; not relevant to the choice of reflections for refinement. R-factors based
; on F^2^ are statistically about twice as large as those based on F, and R-
; factors based on ALL data will be even larger.
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 N3 N 0.4943(5) 0.59845(19) 0.1382(9) 0.0454(17) Uani 1 1 d . . .  
 C1 C 0.4846(7) 0.7171(3) 0.2781(15) 0.062(3) Uani 1 1 d . . .  
 C2 C 0.4874(8) 0.6890(3) 0.4138(13) 0.061(3) Uani 1 1 d . . .  
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 C5 C 0.4382(6) 0.5714(2) 0.2189(12) 0.051(2) Uani 1 1 d . . .  
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 C6 C 0.4393(6) 0.5336(2) 0.1673(12) 0.049(2) Uani 1 1 d . . .  
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 C7 C 0.4995(6) 0.5215(2) 0.0288(10) 0.0371(18) Uani 1 1 d . . .  
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 I3 0.0752(6) 0.0928(7) 0.0607(5) -0.0006(4) -0.0033(3) 0.0007(4)  
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 N2 0.083(6) 0.022(4) 0.055(4) 0.001(3) 0.002(4) 0.001(3)  
 N3 0.056(4) 0.032(4) 0.049(4) -0.007(3) 0.007(3) 0.000(3)  
 C1 0.063(6) 0.035(5) 0.088(8) -0.009(5) 0.006(5) 0.005(4)  
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All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic)

treatment of cell esds is used for estimating esds involving l.s. planes.

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N2 C4 1.309(11) . ?
N3 C5 1.349(10) . ?
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  _geom_angle_atom_site_label_1
  _geom_angle_atom_site_label_2
  _geom_angle_atom_site_label_3
  _geom_angle
  _geom_angle_site_symmetry_1
  _geom_angle_site_symmetry_3
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I3 I2 I1 179.18(3) . ?
N2 N1 C1 127.1(8) . ?
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N2 N1 H1 116.4 . . ?  
C1 N1 H1 116.4 . . ?  
C4 N2 N1 113.8(8) . . ?  
C5 N3 C9 120.0(7) . . ?  
C5 N3 C4 120.0(7) . . ?  
C9 N3 C4 119.9(7) . . ?  
O1 C1 N1 118.9(10) . . ?  
O1 C1 C2 124.9(10) . . ?  
N1 C1 C2 116.2(8) . . ?  
C3 C2 C1 119.4(10) . . ?  
C3 C2 H2 120.3 . . ?  
C1 C2 H2 120.3 . . ?  
C2 C3 C4 118.6(9) . . ?  
C2 C3 H3 120.7 . . ?  
C4 C3 H3 120.7 . . ?  
N2 C4 C3 124.6(8) . . ?  
N2 C4 N3 112.8(7) . . ?  
C3 C4 N3 122.5(8) . . ?  
C6 C5 N3 120.6(8) . . ?  
C6 C5 H5 119.7 . . ?  
N3 C5 H5 119.7 . . ?  
C5 C6 C7 120.4(8) . . ?  
C5 C6 H6 119.8 . . ?  
C7 C6 H6 119.8 . . ?  
C8 C7 C6 117.7(7) . . ?  
C8 C7 C7 122.2(9) . 5\_665 ?  
C6 C7 C7 120.1(9) . 5\_665 ?  
C9 C8 C7 121.0(8) . . ?  
C9 C8 H8 119.5 . . ?  
C7 C8 H8 119.5 . . ?  
C8 C9 N3 120.3(9) . . ?  
C8 C9 H9 119.8 . . ?  
N3 C9 H9 119.8 . . ?

loop\_  
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  \_geom\_torsion\_atom\_site\_label\_3  
  \_geom\_torsion\_atom\_site\_label\_4  
  \_geom\_torsion  
    \_geom\_torsion\_site\_symmetry\_1  
    \_geom\_torsion\_site\_symmetry\_2  
    \_geom\_torsion\_site\_symmetry\_3  
    \_geom\_torsion\_site\_symmetry\_4

\_geom\_torsion\_publ\_flag  
C1 N1 N2 C4 -1.1(13) .... ?  
N2 N1 C1 O1 -177.1(9) .... ?  
N2 N1 C1 C2 4.8(14) .... ?  
O1 C1 C2 C3 176.3(9) .... ?  
N1 C1 C2 C3 -5.7(15) .... ?  
C1 C2 C3 C4 3.3(15) .... ?  
N1 N2 C4 C3 -1.8(13) .... ?  
N1 N2 C4 N3 179.5(7) .... ?  
C2 C3 C4 N2 0.6(15) .... ?  
C2 C3 C4 N3 179.2(8) .... ?  
C5 N3 C4 N2 -139.7(8) .... ?  
C9 N3 C4 N2 41.3(11) .... ?  
C5 N3 C4 C3 41.6(12) .... ?  
C9 N3 C4 C3 -137.4(9) .... ?  
C9 N3 C5 C6 -1.7(13) .... ?  
C4 N3 C5 C6 179.3(8) .... ?  
N3 C5 C6 C7 0.6(14) .... ?  
C5 C6 C7 C8 -0.5(12) .... ?  
C5 C6 C7 C7 179.8(9) ... 5\_665 ?  
C6 C7 C8 C9 1.7(13) .... ?  
C7 C7 C8 C9 -178.7(10) 5\_665 ... ?  
C7 C8 C9 N3 -2.9(15) .... ?  
C5 N3 C9 C8 2.9(14) .... ?  
C4 N3 C9 C8 -178.1(9) .... ?

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\_diffrn\_reflns\_theta\_full 25.05  
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\_refine\_diff\_density\_rms 0.251