

Unification of Reaction Metrics II: Evaluation of Named Organic Reactions and Application to Reaction Discovery

John Andraos

Department of Chemistry, York University, Toronto, ON M3J 1P3 CANADA

jandraos@yorku.ca

Supplementary Material

Table D10. Categorization of Named Substitution Reactions by Minimum Atom Economy and Maximum Environmental Impact Factor

(Reactions are listed alphabetically.)

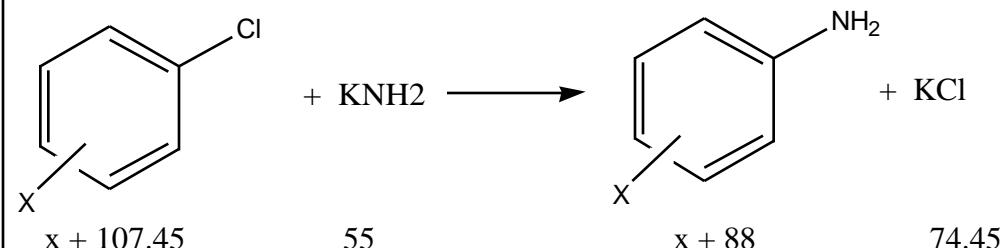
Aniline synthesis

Hrutfiord, B.F.; Bunnett, J.F. *J. Am. Chem. Soc.* **1958**, 80, 2021

Panar, M.; Roberts, J.D. *J. Am. Chem. Soc.* **1960**, 82, 3629

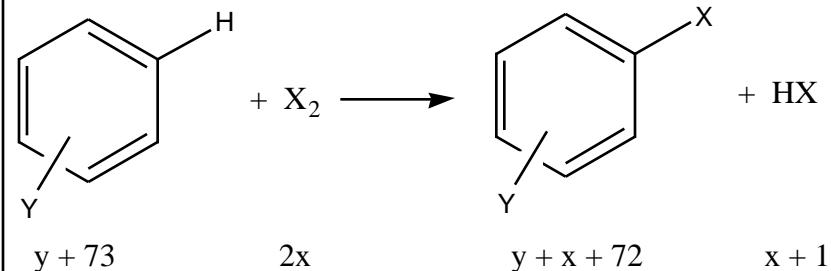
Bunnett, J.F.; Hrutfiord, B.F. *J. Org. Chem.* **1962**, 27, 4152

Bunnett, J.F.; Kim, J.K. *J. Am. Chem. Soc.* **1970**, 92, 7464



X = x (sum of substituents)

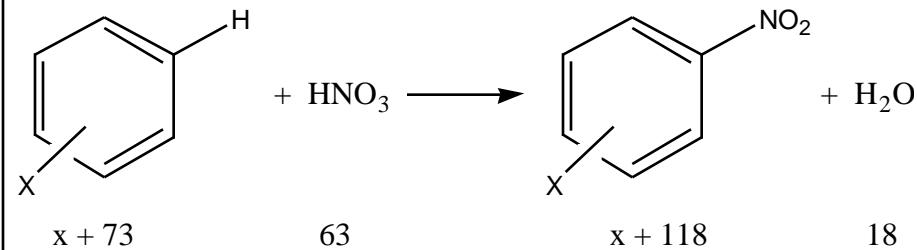
$$\text{AE} = [x + 88]/[x + 162.45]; \text{AE(min)} = \mathbf{0.56}; E = 74.45/[x + 88]; \text{E(max)} = \mathbf{0.80}$$

Aromatic HalogenationScheufelen, A. *Ann. Chem.* **1885**, 231, 152Schramm, J. *Chem. Ber.* **1885**, 18, 607Seelig, E. *Ann. Chem.* **1887**, 237, 129Rilliet, A.; Ador, E. *Chem. Ber.* **1875**, 8, 1286Mueller, H. *J. Chem. Soc.* **1862**, 15, 41

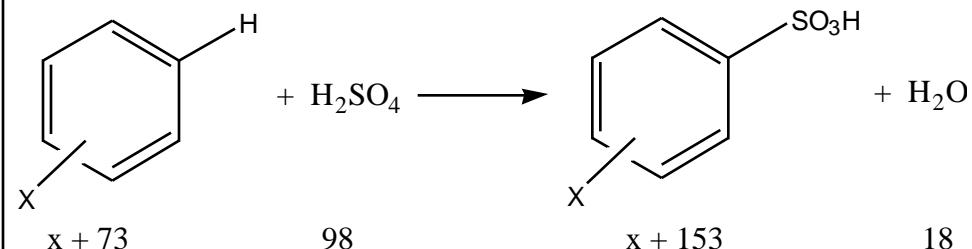
$Y = y$ (sum of substituents); $X = x$ (halogen)

$$AE = [x + y + 72]/[2x + y + 73]; E = [x + 1]/[x + y + 72]$$

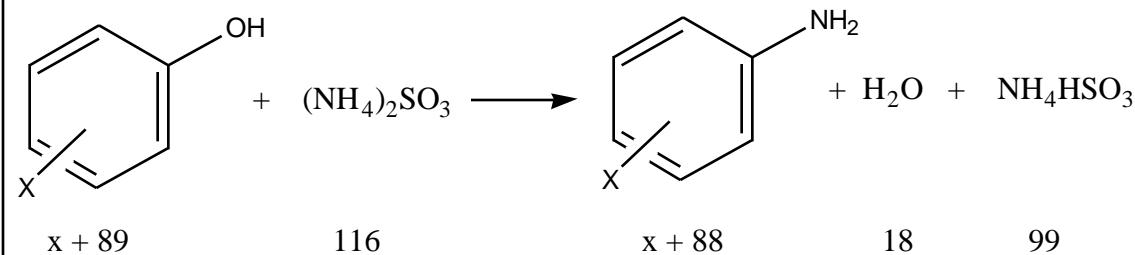
X	E(max)	AE(min)
Cl	0.32	0.76
Br	0.52	0.66
I	0.63	0.61

Aromatic NitrationKopp, H. *Ann. Chem.* **1856**, 98, 367 $X = x$ (sum of substituents)

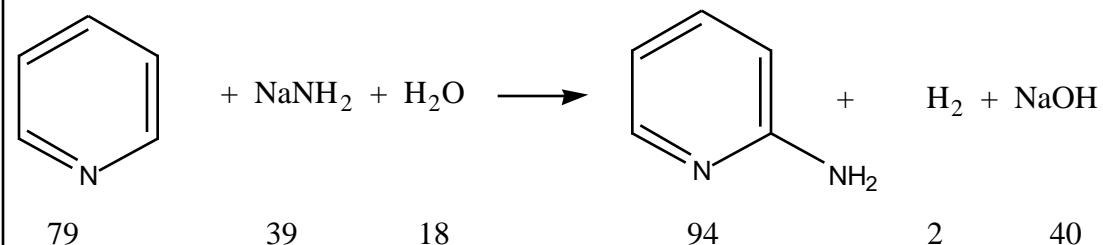
$$AE = [x + 118]/[x + 136]; AE(\min) = \mathbf{0.87}; E = 18/[x + 118]; E(\max) = \mathbf{0.15}$$

Aromatic SulphonationAdor, E.; Meyer, V. *Ann. Chem.* **1871**, 159, 1Meyer, V.; Michler, W. *Chem. Ber.* **1875**, 8, 672Barth, L.; Senhofer, C. *Chem. Ber.* **1875**, 8, 754Ascher, M.; Meyer, V. *Chem. Ber.* **1871**, 4, 323Noelting, E. *Chem. Ber.* **1875**, 8, 1091 $X = x$ (sum of substituents)

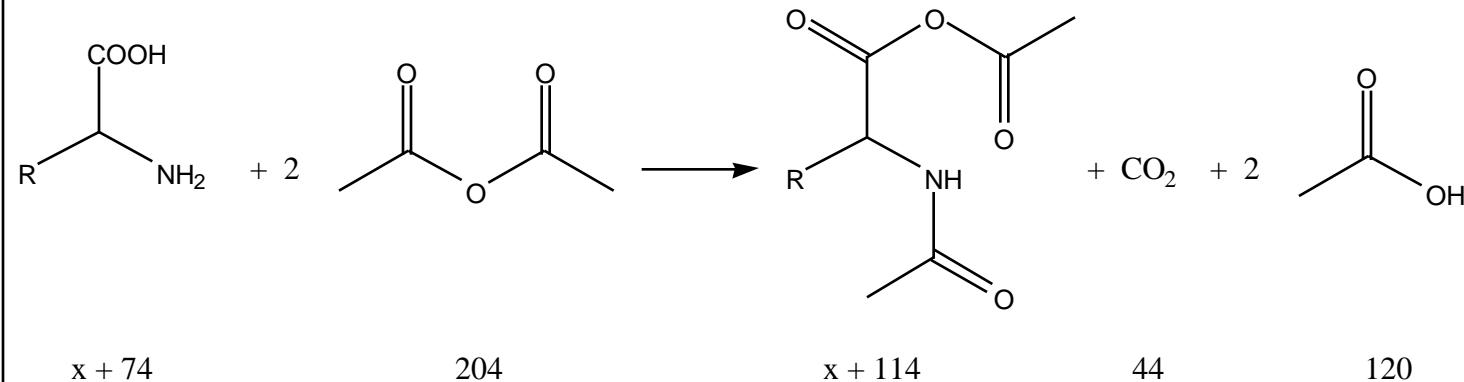
$$AE = [x + 153]/[x + 171]; AE(\min) = \mathbf{0.90}; E = 18/[x + 153]; E(\max) = \mathbf{0.11}$$

Bucherer reactionBucherer, T., *J. Prakt. Chem.* **1904**, 69, 49 $X = x$ (sum of substituents)

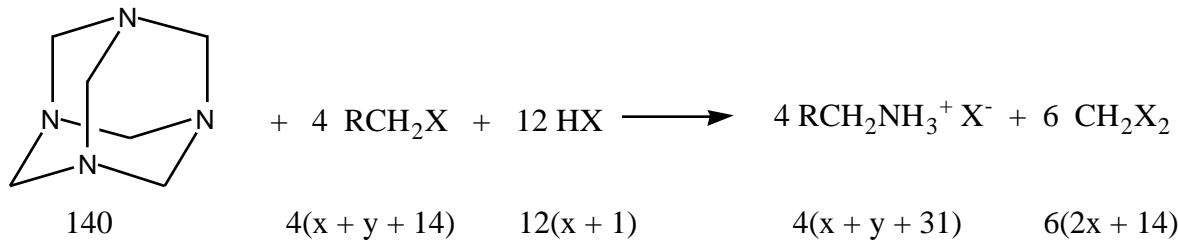
$$\Delta E = [x + 88]/[x + 205]; \Delta E(\min) = 0.43; E = 117/[x + 88]; E(\max) = 1.33$$

Chichibabin reactionChichibabin, A.E.; Zeide, O.A., *J. Russ. Phys. Chem. Soc.* **1914**, 46, 1216

$$\Delta E(\min) = 0.69; E(\max) = 0.45$$

Dakin-West reactionDakin, H.D.; West, R., *J. Biol. Chem.* **1928**, 78, 91; 745; 757 $R = x$

$$AE = [x + 114]/[x + 278]; AE(\text{min}) = \mathbf{0.41}; E = 164/[x + 114]; E(\text{max}) = \mathbf{1.43}$$

Delepine reactionDélépine, M., *Bull. Soc. Chim. Fr.* **1895**, 13, 352

X = x (halide); R = y

$$AE = [4(x + y + 31)]/[16x + 4y + 208]; E = [6(2x + 14)]/[4(x + y + 31)]$$

X	E(max)	AE(min)
F	1.56	0.39
Cl	1.92	0.34
Br	2.35	0.30
I	2.54	0.28

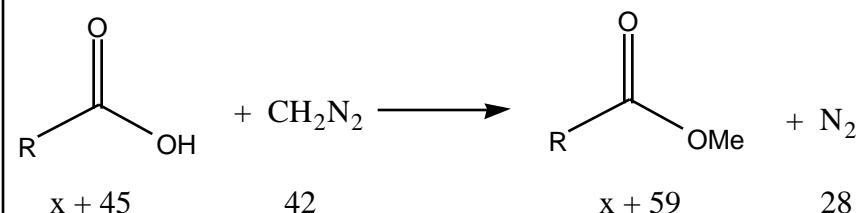
Esterification with Diazomethane

Herzig, J.; Pollak, J. *Monatsch. Chem.* **1908**, 29, 263

Wegschneidler, R.; Gehringer, H. *Monatsch. Chem.* **1909**, 29, 529

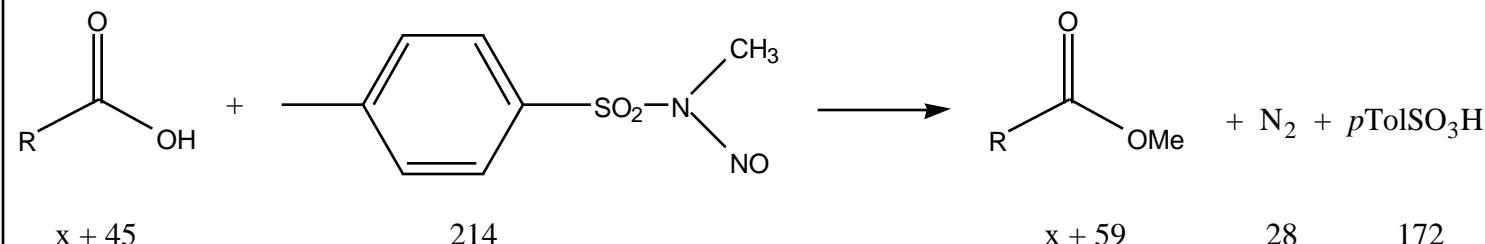
Bouveault, L.; Locquin, R. *Bull. Soc. Chim. Fr.* **1910**, 5[4], 1136

De Boer, T.J.; Backer, H.J. *Org. Synth. Coll. Vol. IV* **1963**, 250



R = x

$$\text{AE} = [\text{x} + 59]/[\text{x} + 87]; \text{AE(min)} = \mathbf{0.68}; \text{E} = 28/[\text{x} + 59]; \text{E(max)} = \mathbf{0.47}$$

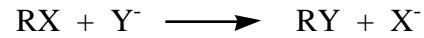


R = x

$$\text{AE} = [\text{x} + 59]/[\text{x} + 259]; \text{AE(min)} = \mathbf{0.23}; \text{E} = 200/[\text{x} + 59]; \text{E(max)} = \mathbf{3.33}$$

Finkelstein reaction

Perkin, W.H.; Duppa, B.F., *Ann. Chem.* **1859**, 112, 125
 Finkelstein, H., *Chem. Ber.* **1910**, 43, 1528

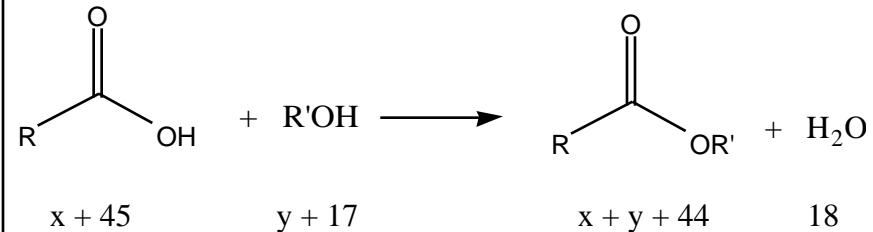


$$\Delta E = [y + r]/[x + y + r]; E = x/[y + r]$$

X	Y	E(max)	$\Delta E(\min)$	
Cl	Cl	0	1	(degenerate case)
Cl	Br	0.44	0.69	
Cl	I	0.28	0.78	
Br	Cl	2.25	0.31	
Br	Br	0	1	(degenerate case)
Br	I	0.63	0.61	
I	Cl	3.58	0.22	
I	Br	1.59	0.39	
I	I	0	1	(degenerate case)

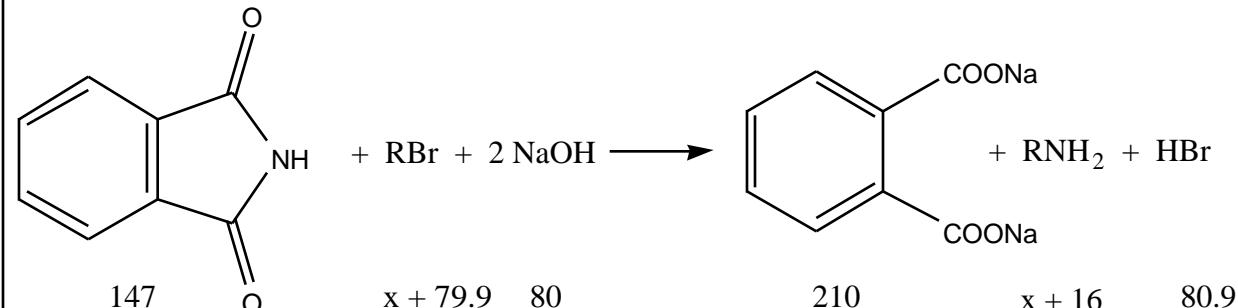
Fischer esterification

Fischer, E.; Speier, A., *Chem. Ber.* **1895**, 28, 3252

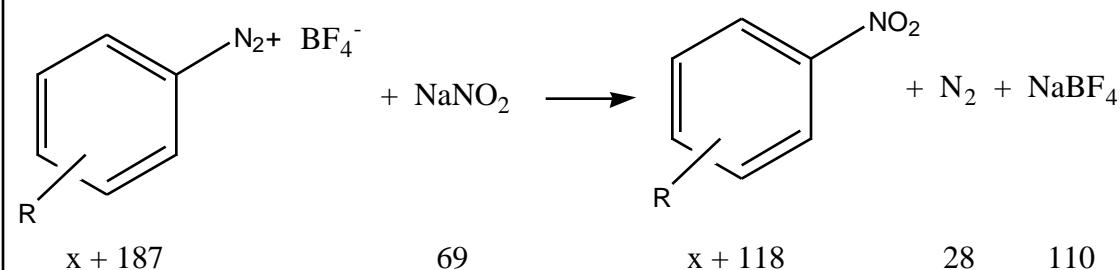


$$R = x; R' = y$$

$$\Delta E = [x + y + 44]/[x + y + 62]; \Delta E(\min) = \mathbf{0.72}; E = 18/[x + y + 44]; E(\max) = \mathbf{0.39}$$

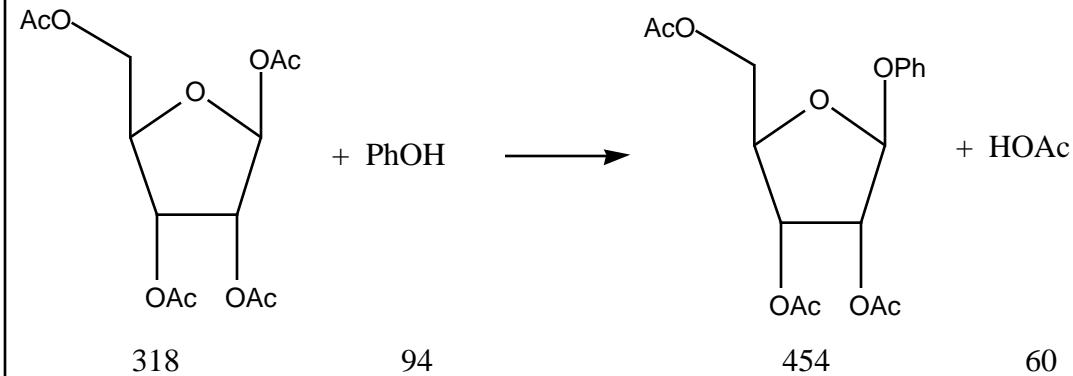
Gabriel synthesisGabriel, S., *Chem. Ber.* **1887**, 20, 2224

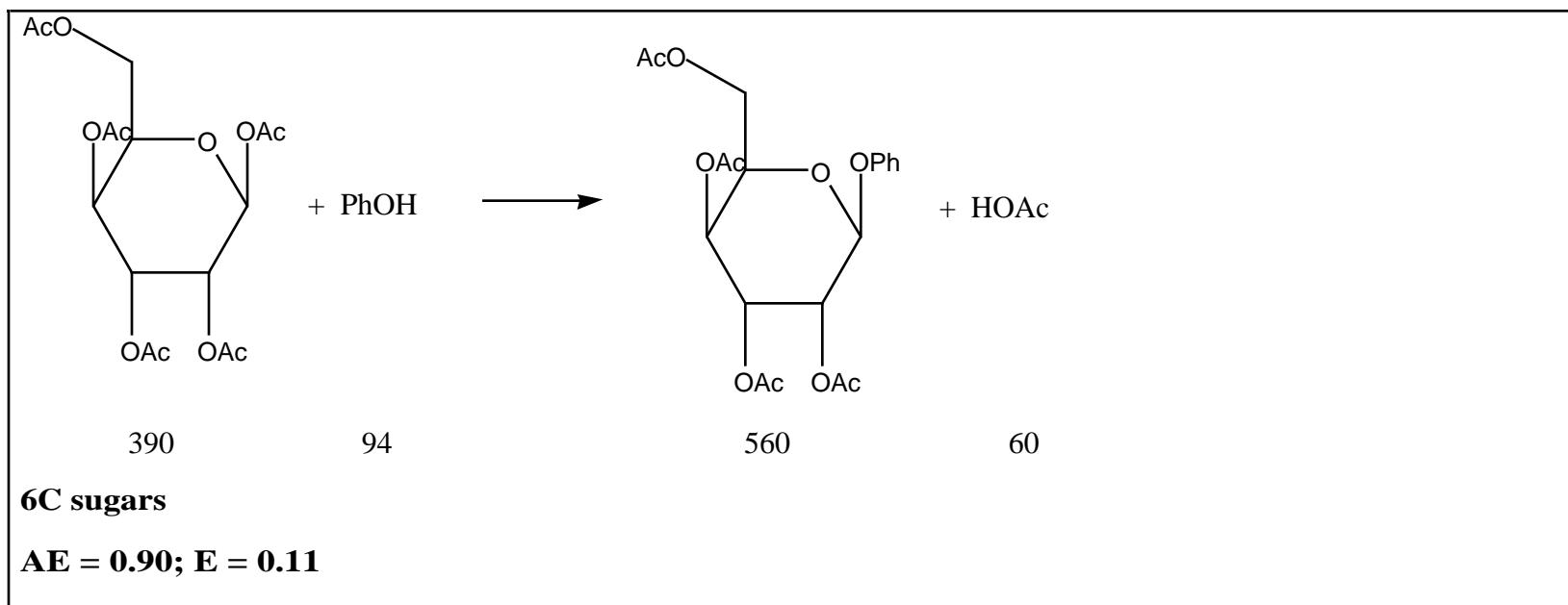
$$\text{AE} = [x + 16]/[x + 306.9]; \text{AE(min)} = \mathbf{0.055}; E = 290.9/[x + 16]; \text{E(max)} = \mathbf{17.11}$$

Gattermann reactionGattermann, L., *Chem. Ber.* **1898**, 31, 1149

$R = x$ (sum of substituents)

$$\text{AE} = [x + 118]/[x + 256]; \text{AE(min)} = \mathbf{0.47}; E = 138/[x + 118]; \text{E(max)} = \mathbf{1.12}$$

Helperich methodHelperich, B.; Schmitz-Hillebrecht, E. *Chem. Ber.* **1933**, 66, 378**5C sugars****AE = 0.88; E = 0.13**

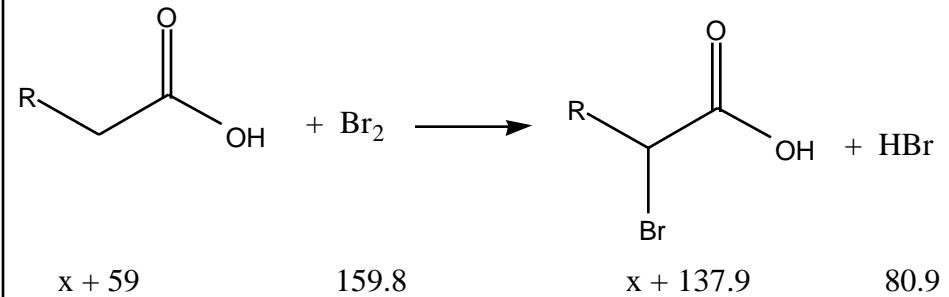


Hell-Volhard-Zelinsky reaction

Hell, C., *Chem. Ber.* **1881**, 14, 891

Volhard, J., *Ann. Chem.* **1887**, 242, 141

Zelinsky, N., *Chem. Ber.* **1887**, 20, 2026

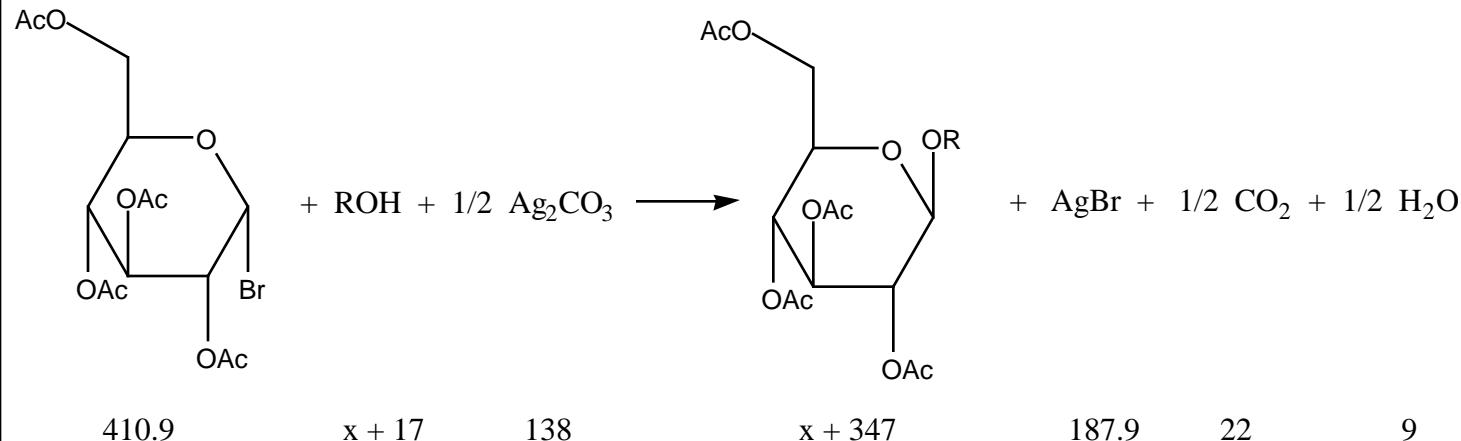


$R = x$

$$\text{AE} = [x + 137.9]/[x + 218.8]; \text{AE(min)} = 0.63; E = 80.9/[x + 137.9]; \text{E(max)} = 0.58$$

Koenigs-Knorr synthesis

Koenigs, W.; Knorr, E. *Chem. Ber.* **1901**, 34, 957

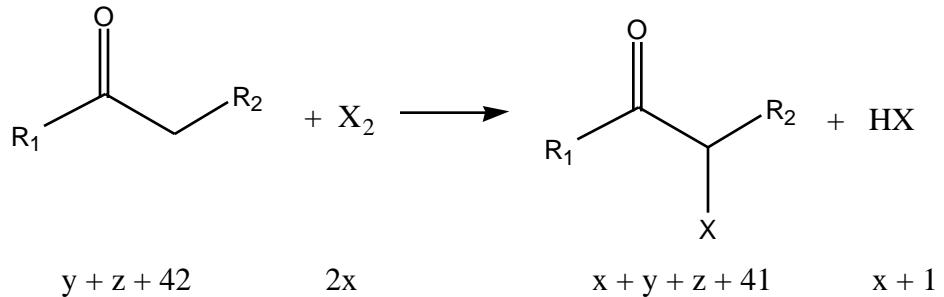


$R = x$

$$\text{AE} = [x + 347]/[x + 565.9]; \text{AE(min)} = \mathbf{0.61}; E = 218.9/[x + 347]; \text{E(max)} = \mathbf{0.62}$$

Lapworth reaction

Lapworth, A., *J. Chem. Soc.* **1903**, 83, 995
 Lapworth, A., *J. Chem. Soc.* **1904**, 85, 1206



$\text{R}_1 = \text{y}$; $\text{R}_2 = \text{z}$; $\text{X} = \text{x}$ (halide)

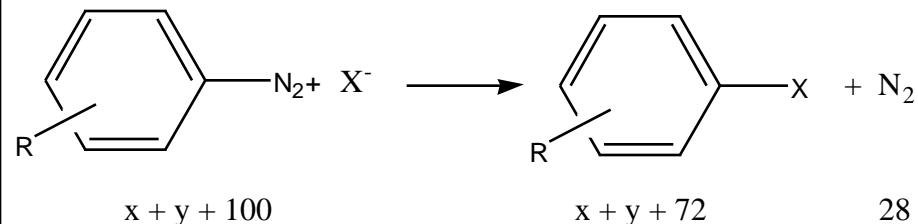
$$\text{AE} = [\text{x} + \text{y} + \text{z} + 41]/[2\text{x} + \text{y} + \text{z} + 42]; \text{E} = [\text{x} + 1]/[\text{x} + \text{y} + \text{z} + 41]$$

$$\text{AE(min)} = [\text{x} + 43]/[2\text{x} + 42]; \text{E(max)} = [\text{x} + 1]/[\text{x} + 43]$$

X	E(max)	AE(min)
Cl	0.46	0.68
Br	0.66	0.60
I	0.75	0.57

Sandmeyer reaction

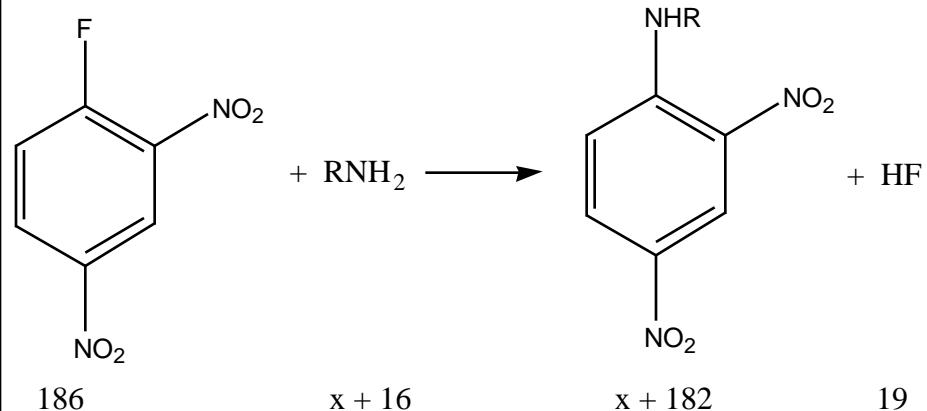
Sandmeyer, T., *Chem. Ber.* **1884**, 17, 1633
 Sandmeyer, T., *Chem. Ber.* **1884**, 17, 2650



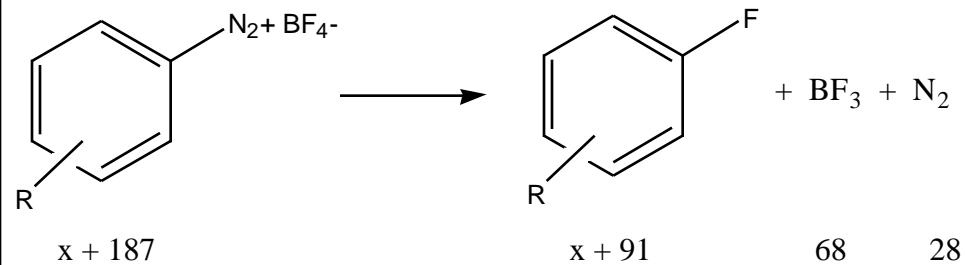
$\text{R} = \text{x}$ (sum of substituents); $\text{X} = \text{x}$ (halide)

$$\text{AE} = [\text{x} + \text{y} + 72]/[\text{x} + \text{y} + 100]; \text{E} = 28/[\text{x} + \text{y} + 72]$$

X	E(max)	AE(min)
Cl	0.25	0.80
Br	0.18	0.85
I	0.14	0.88

Sanger reactionSanger, F., *Biochem. J.* **1945**, 39, 507 $R = x$

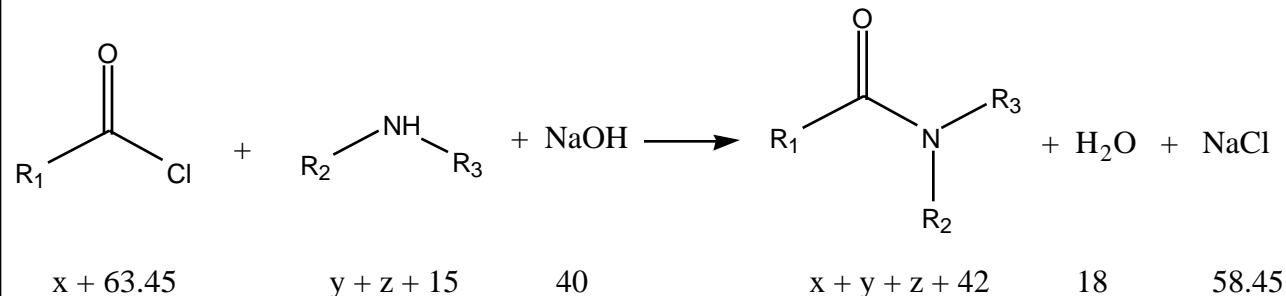
$$\text{AE} = [x + 182]/[x + 202]; \text{AE(min)} = \mathbf{0.90}; E = 20/[x + 182]; \text{E(max)} = \mathbf{0.11}$$

Schiemann reactionBalz, G.; Schiemann, G., *Chem. Ber.* **1927**, 60, 1186 $R = x$ (sum of substituents)

$$\text{AE} = [x + 91]/[x + 187]; \text{AE(min)} = \mathbf{0.50}; E = 96/[x + 91]; \text{E(max)} = \mathbf{1}$$

Schotten-Baumann reaction

Schotten, C., *Chem. Ber.* **1884**, 17, 2544
Baumann, E., *Chem. Ber.* **1886**, 19, 3218



R1 = x; R2 = y; R3 = z

$$\text{AE} = [x + y + z + 42]/[x + y + z + 118.45]; \textbf{AE(min)} = 0.37; E = 76.45/[x + y + z + 42]; \textbf{E(max)} = 1.70$$

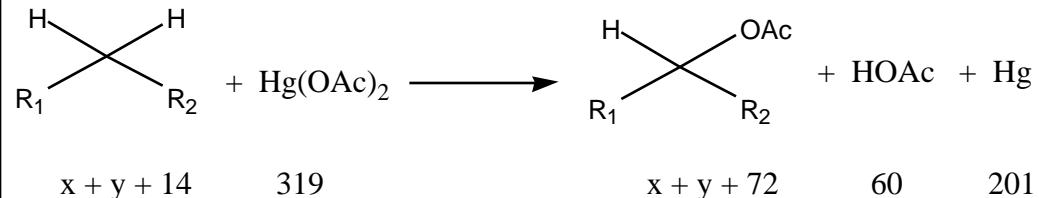
Synthesis of acetates via oxymercuration

Brook, A.G.; Wright, G.F. *Can. J. Res.* **1950**, 28B, 623

Wright, G.F. *Chemistry in Canada* **1950**, 2(9), 29

Wright, G.F. *Ann. N.Y. Acad. Sci.* **1957**, 65, 436

Abercrombie, M.J.; Rodgman, A.; Bharucha, K.R.; Wright, G.F. *Can. J. Chem.* **1959**, *37*, 1328



R1 = x; R2 = y

$$AE = [x + y + 72]/[x + y + 333]; AE(\min) = 0.22; E = 261/[x + y + 72]; E(\max) = 3.53$$

Synthesis of Carbonate diesters, Carbamates (Urethanes), and Ureas

Kym, P. J. *Prakt. Chem.* **1907**, 75, 323

Friedrich Bayer & Co., US Pat. 935,017 (1909/09/28)

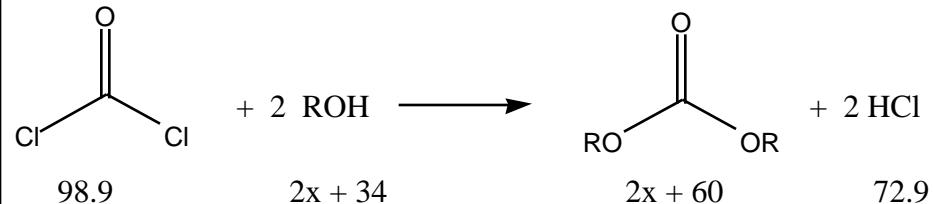
Friedrich Bayer & Co., US Pat. 935,018 (1909/09/28)

Friedrich Bayer & Co., DE Pat. 234,637 (1910/05/21)

Friedrich Bayer & Co., GB Pat. 1,012,433 (1910/05/21)

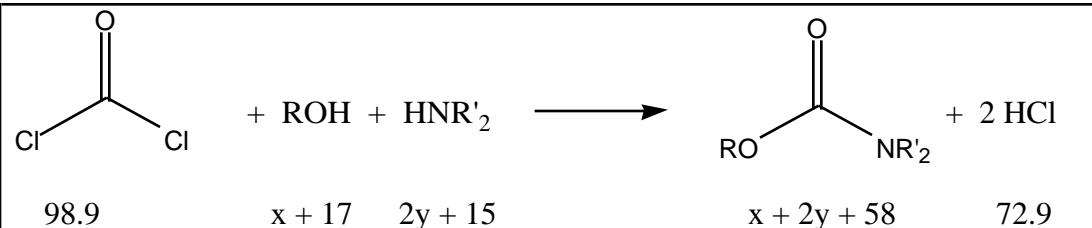
Abbott Labs, GB 255,971 (1925/05/06)

Miescher, K.; Kaeg, H., US Pat. 2,173,423 (1939/09/19)



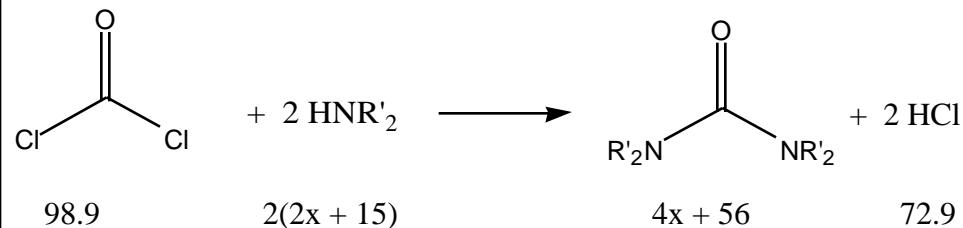
$\text{R} = \text{x}$

$$\text{AE} = [2x + 60]/[2x + 132.9]; \text{AE(min)} = \mathbf{0.46}; \text{E} = 72.9/[2x + 60]; \text{E(max)} = \mathbf{1.18}$$



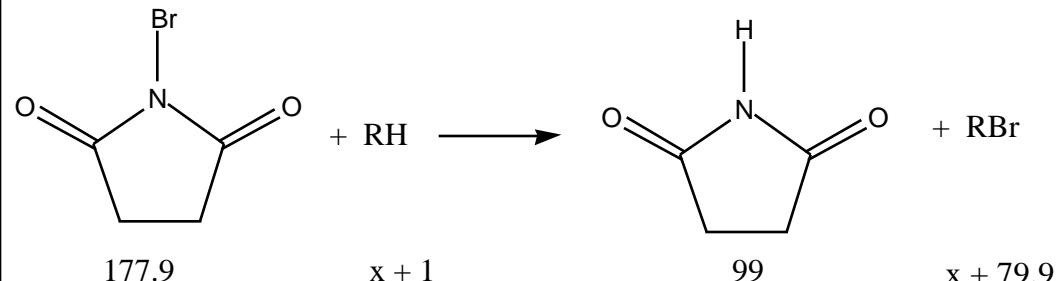
$\text{R} = x; \text{R}' = y$

$$\text{AE} = [x + 2y + 58]/[2x + 130.9]; \text{AE(min)} = \mathbf{0.46}; \text{E} = 72.9/[x + 2y + 58]; \text{E(max)} = \mathbf{1.20}$$

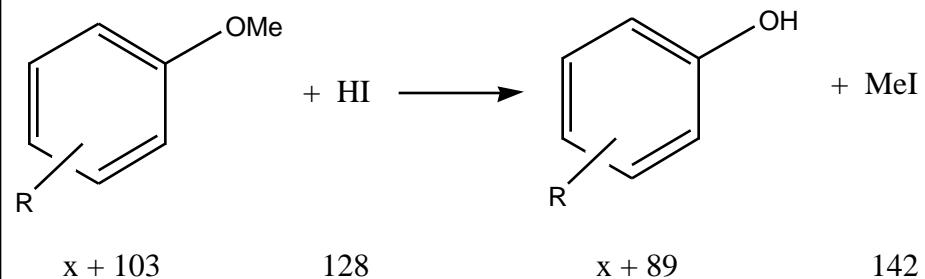


$\text{R}' = x$

$$\text{AE} = [4x + 56]/[4x + 128.9]; \text{AE(min)} = \mathbf{0.45}; \text{E} = 72.9/[4x + 56]; \text{E(max)} = \mathbf{1.22}$$

Wohl-Ziegler brominationWohl, A. *Chem. Ber.* **1919**, 52, 51Ziegler, K.; Spath, E.; Schaaf, E.; Schumann, W.; Winkelmann, E. *Ann. Chem.* **1942**, 551, 80 $R = x$

$$AE = [x + 79.9]/[x + 177.9]; \mathbf{AE(\min)} = \mathbf{0.45}; E = 99/[x + 79.9]; \mathbf{E(\max)} = \mathbf{1.22}$$

Zeisel determinationZeisel, S. *Monatsch. Chem.* **1885**, 6, 989Zeisel, S. *Monatsch. Chem.* **1886**, 7, 406Zeisel, S.; Fanto, R. *Z. Anal. Chem.* **1903**, 42, 549 $R = x$ (sum of substituents)

$$AE = [x + 89]/[x + 231]; \mathbf{AE(\min)} = \mathbf{0.40}; E = 142/[x + 89]; \mathbf{E(\max)} = \mathbf{1.53}$$