

Supporting Information

Fast and Stable Photochromic Oxazines

Massimiliano Tomasulo,[†] Salvatore Sortino,^{*,‡} Andrew J. P. White,[§]
and Francisco M. Raymo^{*,†}

Center for Supramolecular Science, Department of Chemistry, University of Miami, 1301 Memorial Drive, Florida, 33146-0431, Dipartimento di Scienze Chimiche, Universitá di Catania, viale Andrea Doria 8, Catania, I-95125, Italy, and Department of Chemistry, Imperial College London, South Kensington, London, SW7 2AY, UK

E-Mail: ssortino@unict.it; fraymo@miami.edu

• General Methods and Experimental Procedures for the Synthesis of 2 and 6	S2
• High-Performance Liquid Chromatograms of 3a and 4a	S3
• Partial ¹ H NMR Spectra of 4a at Various Temperatures	S4
• Partial ¹ H NMR Spectra of 3a Before and After the Addition of Bu ₄ NOH	S5
• Single-Crystal X-Ray Structures (ORTEP) of 3a and 4c	S6
• Ground-State Potential Energy Profile (AM1) of 3a	S7
• Computational Data (B3LYP) for 3a , 3b and the transition state.....	S8
• Steady-State Absorption Spectra of 4a Before and After the Addition of Bu ₄ NOH	S13
• Transient Absorption Spectra of 3a	S14
• Kinetic Trace for the Photoisomerization of 4a	S15

[†] University of Miami

[‡] Universitá di Catania

[§] Imperial College

General Methods and Experimental Procedures for the Synthesis of **2** and **6**

General Methods: Chemicals were purchased from commercial sources and used as received with the exception of MeCN and CH₂Cl₂, which were distilled over CaH₂. Compound **5** was prepared according to a literature protocol.¹ The experimental procedures for the preparation of **2** and **6** are reported in the Supporting Information (Figure S1). All reactions were monitored by thin-layer chromatography, using aluminum sheets coated with silica (60, F₂₅₄). High-performance liquid chromatography (HPLC) was performed with analytical (column dimensions = 4.6 × 250 mm, flow rate = 1.0 mL min⁻¹, injection volume = 10 μL, sample concentration = 0.1 mM) and semi-preparative (column dimensions = 21.4 × 250 mm, flow rate = 10 mL min⁻¹, injection volume = 10 mL, sample concentration = 0.1 mM) columns. The retention time (*RT*) and the peak asymmetry (*PA*) were determined at a wavelength of 254 nm. The average purity parameter (*APP*) was calculated for the peak heart in the wavelength range 215–700 nm. Fast atom bombardment mass spectra (FABMS) were recorded in a 3-nitrobenzyl alcohol matrix.

2-Phenyl-3,3'-dimethyl-3H-indole (2). A solution of *i*-propylphenylketone (1.6 mL, 11 mmol) and phenylhydrazine (1.1 mL, 11 mmol) in MeCO₂H (7 mL) was heated under reflux for 17 h. After cooling down to ambient temperature, the solution was diluted with H₂O (20 mL) and the pH was adjusted to *ca.* 8 with aqueous KOH (0.3 M). Then, the mixture was extracted with CH₂Cl₂ (4 × 15 mL). The organic phase was dried (MgSO₄), filtered and concentrated under reduced pressure to afford **2** (1.87 g, 76%) as an orange liquid. FABMS: *m/z* = 222 [M]⁺; ¹H NMR (500 MHz, CDCl₃): δ = 1.55 (6H, s), 7.30 (1H, d, 8 Hz), 7.34–7.40 (2H, m), 7.47–7.53 (3H, m), 7.72 (1H, d, 8 Hz), 8.14–8.20 (2H, m); ¹³C NMR (100 MHz, CDCl₃): δ = 24.7, 53.5, 120.7, 120.9, 126.0, 127.7, 128.3, 128.4, 128.5, 130.6, 132.9, 147.4, 183.3.

9,9-Dimethyl-9a-phenyl-2,3,9a-tetrahydro-oxazolo[3,2-*a*]indole (6). 2-Bromoethanol (1 mL, 28 mmol) was added to a solution of **2** (0.44 g, 2 mmol) in PhMe (25 mL) heated under reflux and Ar. The mixture was heated under reflux for a further 2 d. After cooling down to ambient temperature, the solvent was evaporated under reduced pressure. The residue was dissolved in CH₂Cl₂ (4 mL) and the solution was poured into a flask containing hexanes (35 mL) and sonicated for 30 min. The resulting precipitate was filtered off and the treatment was repeated three additional times. The solid was dissolved in a mixture of MeCN (5 mL), H₂O (10 mL) and aqueous KOH (0.3 M, 3 mL). After stirring for 20 min, the solution was concentrated under reduced pressure and extracted with CH₂Cl₂ (3 × 10 mL). The organic layer was dried (MgSO₄) and the solvent was evaporated under reduced pressure. The residue was purified column chromatography [SiO₂: hexanes/CH₂Cl₂ (1:1)] to afford **6** (13 mg, 3%) as a yellow solid. mp = 94°C; FABMS: *m/z* = 266 [M + H]⁺; ¹H NMR (400 MHz, CDCl₃): δ = 0.77 (3H, s), 1.54 (3H, s), 3.30–3.34 (1H, m), 3.50–3.54 (1H, m), 3.68–3.73 (2H, m), 6.84 (1H, d, 8 Hz), 6.95–6.99 (1H, m), 7.10 (1H, d, 8 Hz), 7.18–7.23 (1H, m), 7.33–7.44 (3H, m), 7.65–7.71 (2H, m); ¹³C NMR (100 MHz, CDCl₃): δ = 20.5, 29.3, 29.9, 47.8, 50.3, 63.3, 111.6, 112.1, 121.7, 122.7, 127.6, 127.8, 128.1, 138.8, 139.5, 151.2.

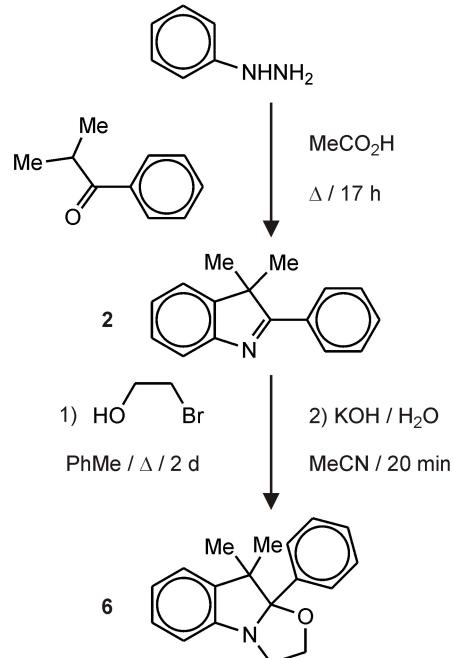
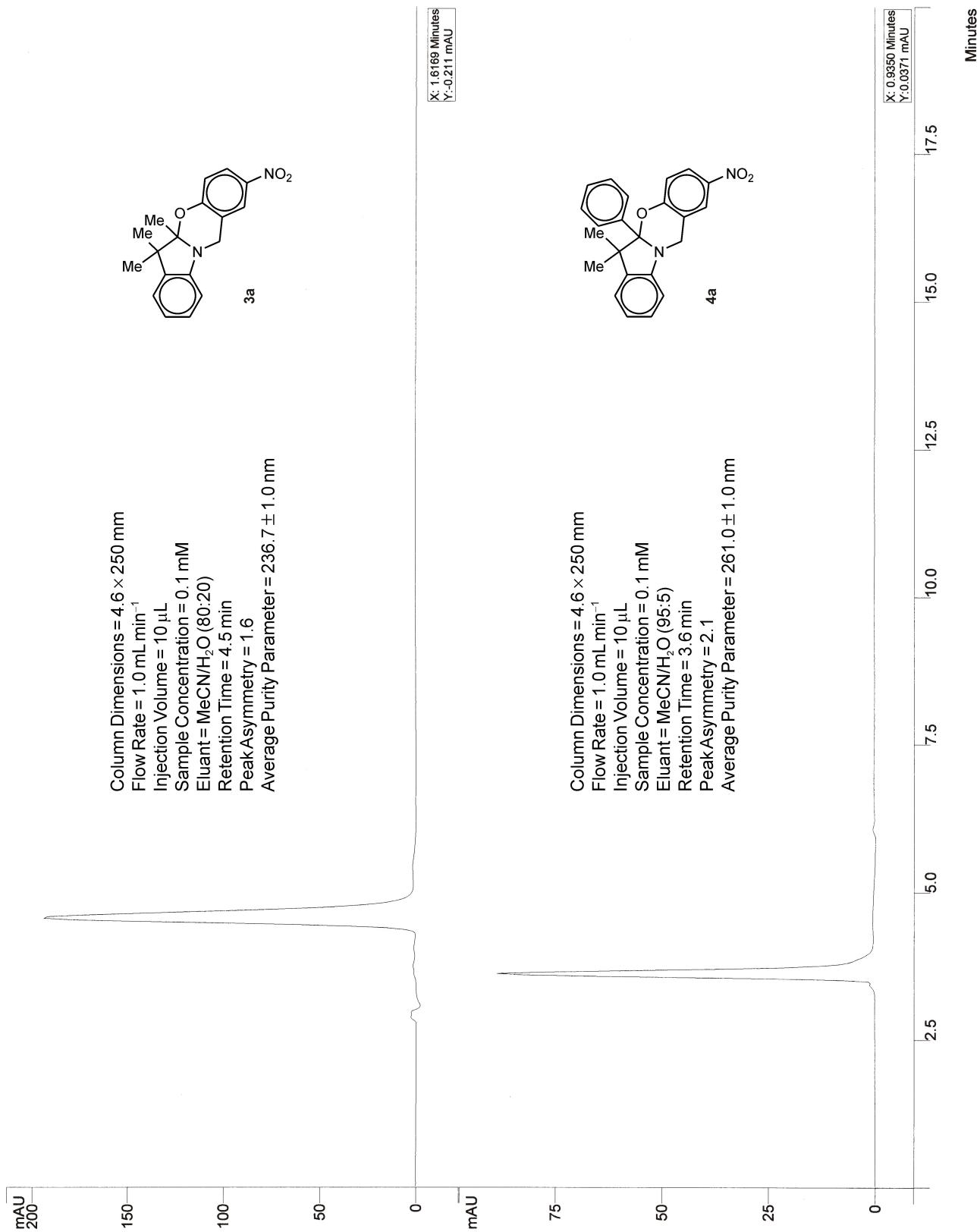


Figure S1. Synthesis of the indoline **6**.

¹ Raymo, F. M.; Giordani, S.; White, A. J. P.; Williams, D. J. *J. Org. Chem.* **2003**, *68*, 4158–4169.

High-Performance Liquid Chromatograms of 3a and 4a



Partial ^1H NMR Spectra of **4a at Various Temperatures**

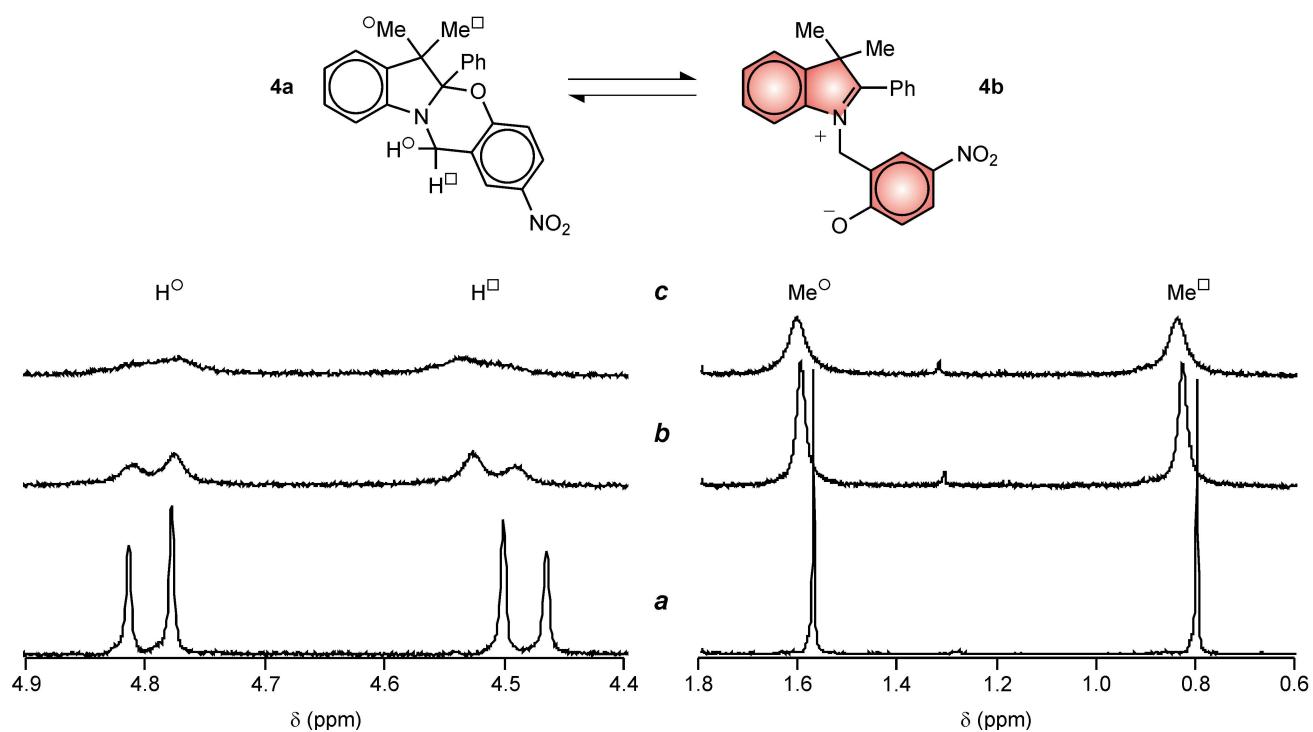


Figure S2. Partial ^1H NMR spectra (500 MHz, acetonitrile-d3, 5 mM) of **4a** at 303 (**a**), 333 (**b**) and 343 K (**c**).

Partial ^1H NMR Spectra of **3a Before and After the Addition of Bu_4NOH**

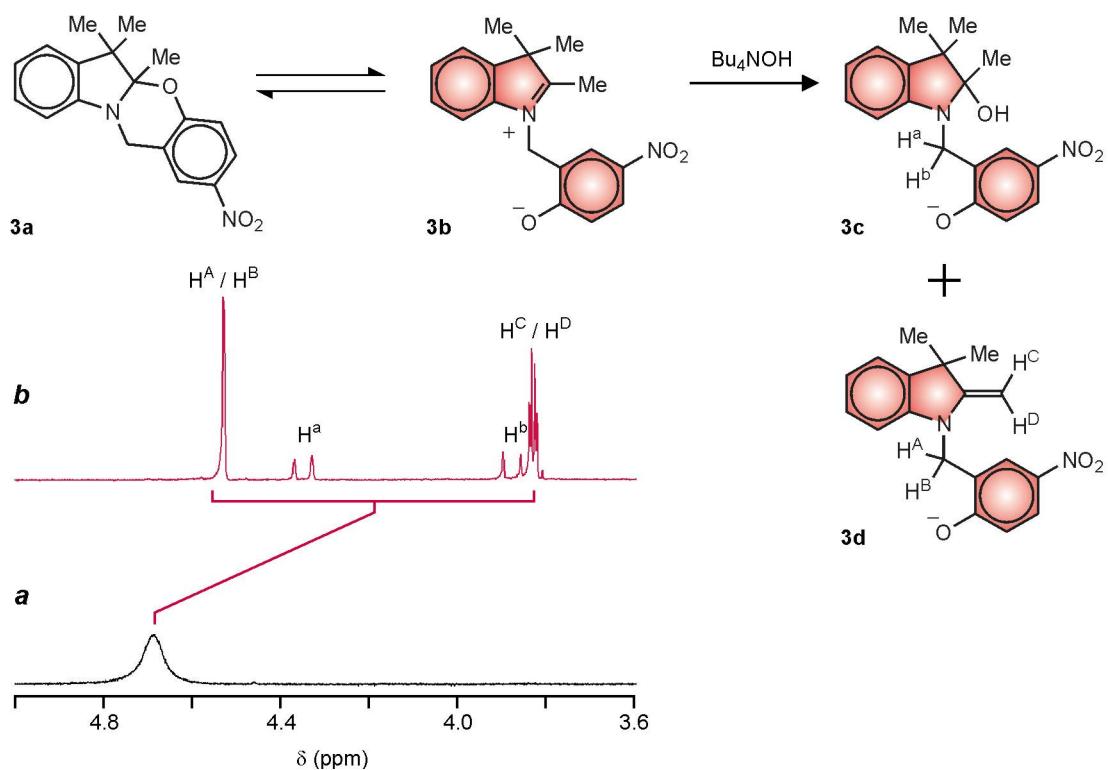


Figure S3. Partial ^1H NMR spectra (400 MHz , acetonitrile- d_3 , 10 mM) of **3a** before (a) and after (b) the addition of Bu_4NOH (2 eq.).

Single-Crystal X-Ray Structures (ORTEP) of **3a** and **4c**

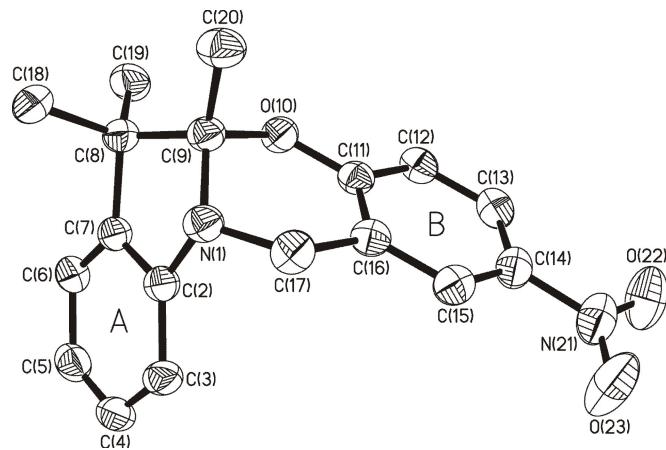


Figure S4. The single-crystal X-ray structure of **3a** (50% probability ellipsoids).

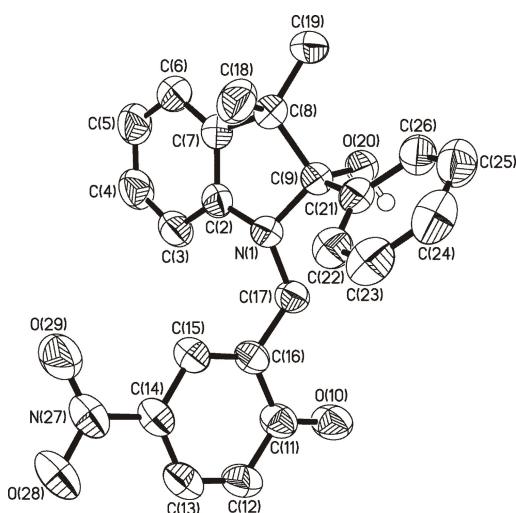


Figure S5. The single-crystal X-ray structure of **4c** (50% probability ellipsoids).

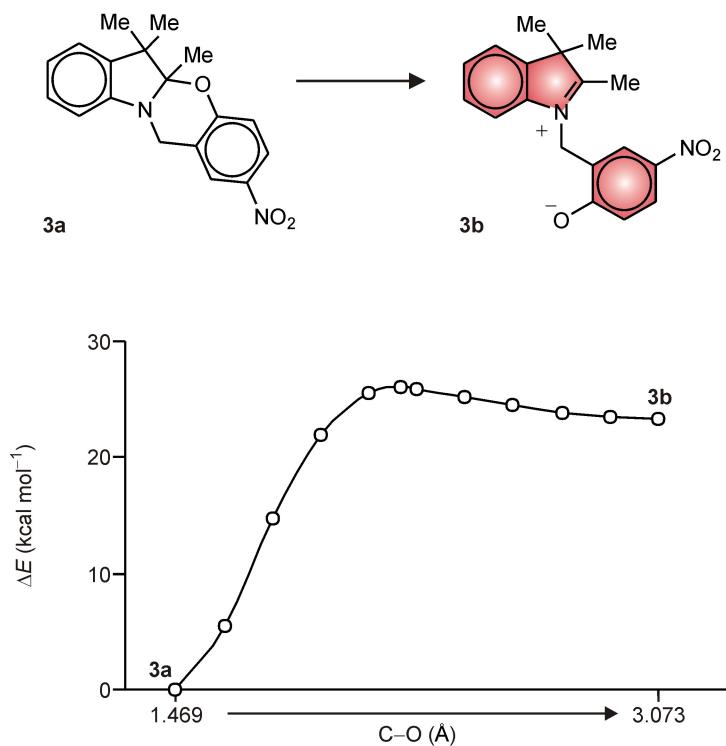
Ground-State Potential Energy Profile (AM1) of 3a

Figure S6. Energy (AM1) dependence on the C–O distance associated with the bond cleaved in the transformation of **3a** into **3b**.

Computational Data (B3LYP) for 3a, 3b and the transition state

3a

Zero-Point Corrected Energy = -1043.790847 hartrees

Number of Imaginary Frequencies = 0

Optimized Geometry:

C						
C	1	1.402515				
C	2	1.386390	1	120.297267		
C	3	1.402747	2	119.270999	1	-0.468500
C	4	1.394716	3	119.974546	2	0.437358
C	1	1.392327	2	121.153022	3	0.262029
N	1	1.415170	2	109.770008	3	-177.761053
C	7	1.451784	1	107.440126	2	-19.610471
C	2	1.523249	1	108.986085	6	177.387111
H	3	1.087342	2	120.576356	1	-179.936367
H	4	1.086056	3	119.965279	2	-179.360723
H	5	1.086721	4	119.768685	3	-179.798943
H	6	1.085800	1	121.870869	2	179.676289
C	9	1.547941	2	108.811538	1	-97.357889
H	14	1.096021	9	111.543223	2	178.831339
H	14	1.094709	9	109.253007	2	-62.085392
H	14	1.094484	9	111.600365	2	57.062283
C	9	1.535252	2	114.201893	1	139.959323
H	18	1.095669	9	109.593665	2	67.126370
H	18	1.096405	9	111.160682	2	-174.393152
H	18	1.091014	9	111.739644	2	-53.185795
C	7	1.458109	1	121.200860	2	-155.231136
H	22	1.092968	7	108.923830	1	-36.619293
C	8	1.524262	7	113.527985	1	158.570858
H	24	1.092413	8	111.021549	7	-54.288822
C	22	1.514414	7	111.090960	1	85.823545
C	26	1.410406	22	118.773581	7	17.544147
C	27	1.405838	26	120.428306	22	-179.051565
C	28	1.384804	27	120.295971	26	-0.716365
C	29	1.397825	28	118.932282	27	0.122882
C	26	1.391283	22	122.367914	7	-162.255015
H	28	1.084914	27	118.482489	26	179.506298
H	29	1.082894	28	121.536911	27	-179.767568
H	31	1.084681	26	120.951233	22	0.007168
O	27	1.353651	26	123.053501	22	0.040107
N	30	1.460826	29	119.406119	28	-179.970862
O	36	1.232966	30	117.822136	29	0.284823
O	36	1.233657	30	117.883852	29	-179.785382
H	22	1.098084	7	108.805293	1	-151.838842
H	24	1.094253	8	110.693621	7	66.261717
H	24	1.093797	8	109.951292	7	-175.125137

1 2 1.5 6 1.5 7 1.0
 2 3 1.5 9 1.0
 3 4 1.5 10 1.0

```

4 5 1.5 11 1.0
5 6 1.5 12 1.0
6 13 1.0
7 8 1.0 22 1.0
8 9 1.0 24 1.0 35 1.0
9 14 1.0 18 1.0
10
11
12
13
14 15 1.0 16 1.0 17 1.0
15
16
17
18 19 1.0 20 1.0 21 1.0
19
20
21
22 23 1.0 26 1.0 39 1.0
23
24 25 1.0 40 1.0 41 1.0
25
26 27 1.5 31 1.5
27 28 1.5 35 1.0
28 29 2.0 32 1.0
29 30 1.5 33 1.0
30 31 1.5 36 1.0
31 34 1.0
32
33
34
35
36 37 1.5 38 1.5
37
38
39
40
41

```

3b

Zero-Point Corrected Energy = -1043.755388 hartrees

Number of Imaginary Frequencies = 0

Optimized Geometry:

C						
C	1	1.394790				
C	2	1.389174	1	119.647364		
C	3	1.399653	2	118.543350	1	0.097222
C	4	1.398981	3	120.834638	2	-0.107278
C	1	1.389205	2	123.042332	3	-0.000267
N	1	1.432625	2	108.293904	3	-178.718136
C	7	1.311877	1	110.955950	2	0.107663
C	2	1.514903	1	108.756186	6	179.747861
H	3	1.086591	2	121.064602	1	-179.808999

H	4	1.085912	3	119.594783	2	-179.837079
H	5	1.085628	4	119.747001	3	-179.487009
H	6	1.082971	1	122.229518	2	178.445019
C	9	1.546005	2	112.307439	1	-120.824785
H	14	1.094637	9	111.204735	2	175.652022
H	14	1.094578	9	109.345618	2	-65.450660
H	14	1.094676	9	111.639411	2	54.045614
C	9	1.554316	2	112.088088	1	113.690447
H	18	1.094834	9	109.170205	2	60.393658
H	18	1.094994	9	110.525269	2	179.340939
H	18	1.092652	9	110.852760	2	-60.606443
C	7	1.492599	1	122.528307	2	-167.324585
H	22	1.093949	7	107.297255	1	-59.023213
C	8	1.483743	7	124.465321	1	179.815483
H	24	1.095150	8	111.916399	7	-51.745702
C	22	1.501651	7	110.296038	1	62.986505
C	26	1.454422	22	117.765287	7	58.280975
C	27	1.443519	26	115.476661	22	179.751297
C	28	1.373842	27	122.282227	26	-0.250216
C	29	1.411649	28	120.037098	27	0.440696
C	26	1.385418	22	120.563086	7	-121.587134
H	28	1.086129	27	116.781279	26	179.954862
H	29	1.084089	28	121.493204	27	-179.748556
H	31	1.086030	26	121.582997	22	-0.385713
O	27	1.269518	26	121.725346	22	-0.946782
N	30	1.437587	29	120.242156	28	179.757412
O	36	1.239516	30	118.356946	29	0.312636
O	36	1.242678	30	118.396686	29	-179.344970
H	22	1.091605	7	106.084142	1	-174.763330
H	24	1.100279	8	106.417290	7	67.398109
H	24	1.091382	8	110.773422	7	-173.130624

1 2 1.5 6 1.5 7 1.0
 2 3 1.5 9 1.0
 3 4 1.5 10 1.0
 4 5 1.5 11 1.0
 5 6 1.5 12 1.0
 6 13 1.0
 7 8 2.0 22 1.0
 8 9 1.0 24 1.0
 9 14 1.0 18 1.0
 10
 11
 12
 13
 14 15 1.0 16 1.0 17 1.0
 15
 16
 17
 18 19 1.0 20 1.0 21 1.0
 19
 20
 21
 22 23 1.0 26 1.0 39 1.0
 23
 24 25 1.0 40 1.0 41 1.0
 25
 26 27 1.0 31 2.0
 27 28 1.5 35 2.0

```

28 29 2.0 32 1.0
29 30 1.5 33 1.0
30 31 1.5 36 1.0
31 34 1.0
32
33
34
35
36 37 1.5 38 1.5
37
38
39
40
41

```

Transition State

Zero-Point Corrected Energy = -1043.762001 hartrees

Number of Imaginary Frequencies = 1

Optimized Geometry:

C						
C	1	1.394021				
C	2	1.389352	1	119.657527		
C	3	1.399529	2	118.611087	1	0.041402
C	4	1.398908	3	120.781672	2	0.013410
C	1	1.389503	2	122.950133	3	-0.035576
N	1	1.431079	2	108.571318	3	-179.014525
C	7	1.317727	1	110.845762	2	-1.721386
C	2	1.515242	1	108.833263	6	178.972271
H	3	1.086670	2	121.030071	1	-179.746258
H	4	1.085923	3	119.622846	2	-179.696222
H	5	1.085695	4	119.743723	3	-179.507649
H	6	1.083483	1	122.191203	2	178.513135
C	9	1.549972	2	110.901740	1	-116.373580
H	14	1.094620	9	111.278592	2	177.868966
H	14	1.094386	9	109.135080	2	-63.377432
H	14	1.094919	9	111.752493	2	55.950303
C	9	1.549306	2	113.060553	1	119.073658
H	18	1.095100	9	109.293427	2	60.189344
H	18	1.095359	9	110.379035	2	178.966690
H	18	1.091505	9	110.711787	2	-61.175463
C	7	1.490965	1	121.502340	2	-161.612381
H	22	1.093409	7	107.924605	1	-58.486682
C	8	1.487454	7	125.977559	1	175.386592
H	24	1.094584	8	109.843775	7	-134.994530
C	22	1.501723	7	108.906102	1	63.399564
C	26	1.452089	22	118.035862	7	53.162675
C	27	1.440437	26	115.833969	22	-178.948663
C	28	1.374767	27	122.145152	26	-0.851987
C	29	1.410182	28	119.946875	27	1.223026
C	26	1.385831	22	120.525809	7	-125.263830
H	28	1.086106	27	116.920064	26	179.875263
H	29	1.084005	28	121.507622	27	-179.276134
H	31	1.085863	26	121.455547	22	-1.278748
O	27	1.274357	26	121.560558	22	0.305328
N	30	1.439651	29	120.137833	28	179.430764

O	36	1.238951	30	118.310767	29	0.366831
O	36	1.241690	30	118.357184	29	-179.333373
H	22	1.091210	7	106.623787	1	-174.454736
H	24	1.090588	8	112.827610	7	-13.854669
H	24	1.094184	8	108.317531	7	105.487943
1	2 1.5	6 1.5	7 1.0			
2	3 1.5	9 1.0				
3	4 1.5	10 1.0				
4	5 1.5	11 1.0				
5	6 1.5	12 1.0				
6	13 1.0					
7	8 2.0	22 1.0				
8	9 1.0	24 1.0				
9	14 1.0	18 1.0				
10						
11						
12						
13						
14	15 1.0	16 1.0	17 1.0			
15						
16						
17						
18	19 1.0	20 1.0	21 1.0			
19						
20						
21						
22	23 1.0	26 1.0	39 1.0			
23						
24	25 1.0	40 1.0	41 1.0			
25						
26	27 1.0	31 2.0				
27	28 1.5	35 2.0				
28	29 2.0	32 1.0				
29	30 1.5	33 1.0				
30	31 1.5	36 1.0				
31	34 1.0					
32						
33						
34						
35						
36	37 1.5	38 1.5				
37						
38						
39						
40						
41						

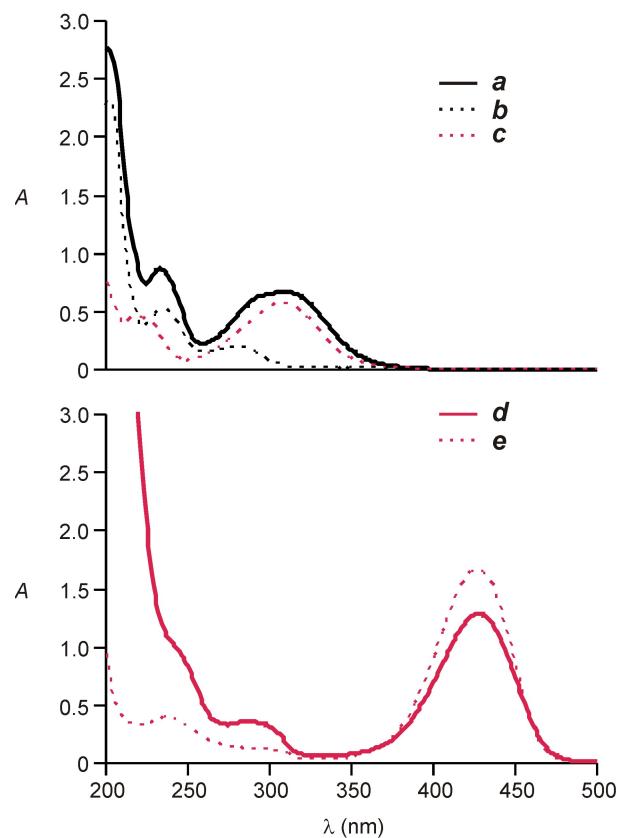
Steady-State Absorption Spectra of **4a Before and After the Addition of Bu_4NOH** 

Figure S7. Steady-state absorption spectra (0.1 mM, MeCN, 298 K) of **4a** (**a**), **6** (**b**) and **7** (**c**) as well as of **4a** after the addition of Bu_4NOH (100 eq.) (**d**) and **8** (**e**).

Transient Absorption Spectra of 3a

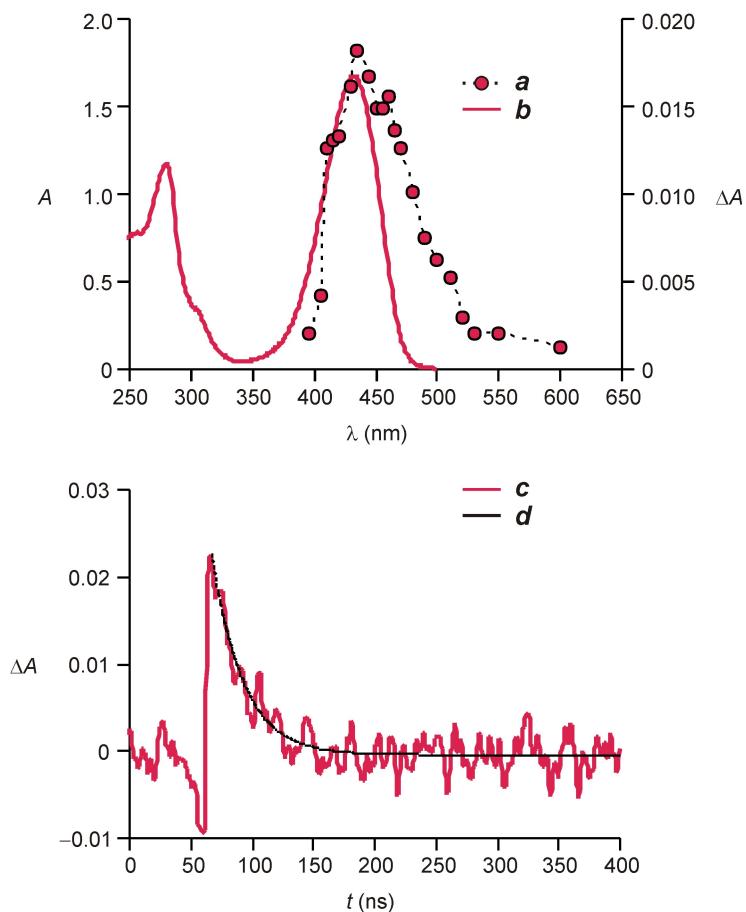


Figure S8. Transient absorption spectrum (*a*) of **3a** recorded 30 ns after the laser pulse (355 nm, 6 ns, 12 mJ, 0.1 mM, MeCN, 295 K) and steady-state absorption spectrum (*b*) of **3a** and Bu₄NOH (0.1 mM, MeCN, 298 K). Evolution of the absorbance at 440 nm (*c*) upon laser excitation of **3a** and the corresponding mono-exponential curve fitting (*d*).

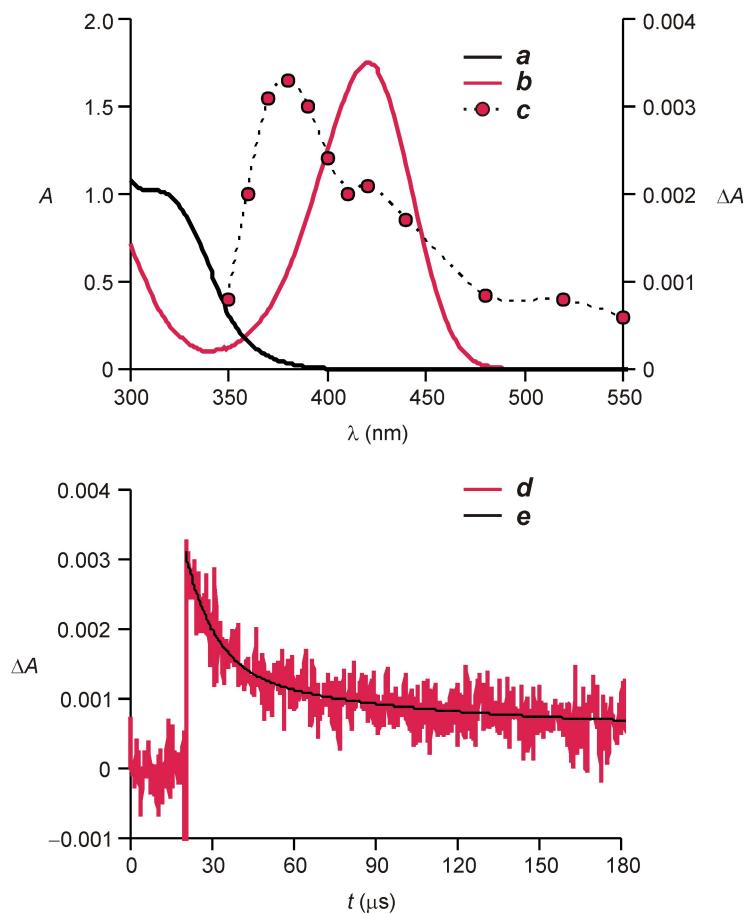


Figure S9. Steady-state absorption spectra (5%, PMMA, 298 K) of **3a** without (**a**) and with (**b**) Bu₄NOH (7 eq.) and transient absorption spectrum (**c**) of **3a** recorded 1 μ s after the laser pulse (355 nm, 6 ns, 12 mJ, 5%, PMMA, 295 K). Evolution of the absorbance at 380 nm (**d**) upon laser excitation of **3a** (355 nm, 6 ns, 12 mJ, 5%, PMMA, 295 K) and the corresponding mono-exponential curve fitting (**e**).

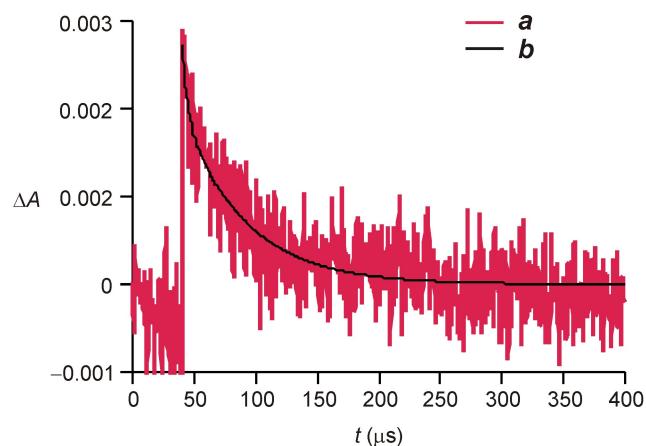
Kinetic Trace for the Photoisomerization of 4a

Figure S10. Evolution of the absorbance at 430 nm (*a*) upon laser excitation of **4a** (355 nm, 6 ns, 12 mJ, 5%, PMMA, 295 K) and the corresponding bi-exponential curve fitting (*b*).