

Supporting Information for the manuscript entitled:

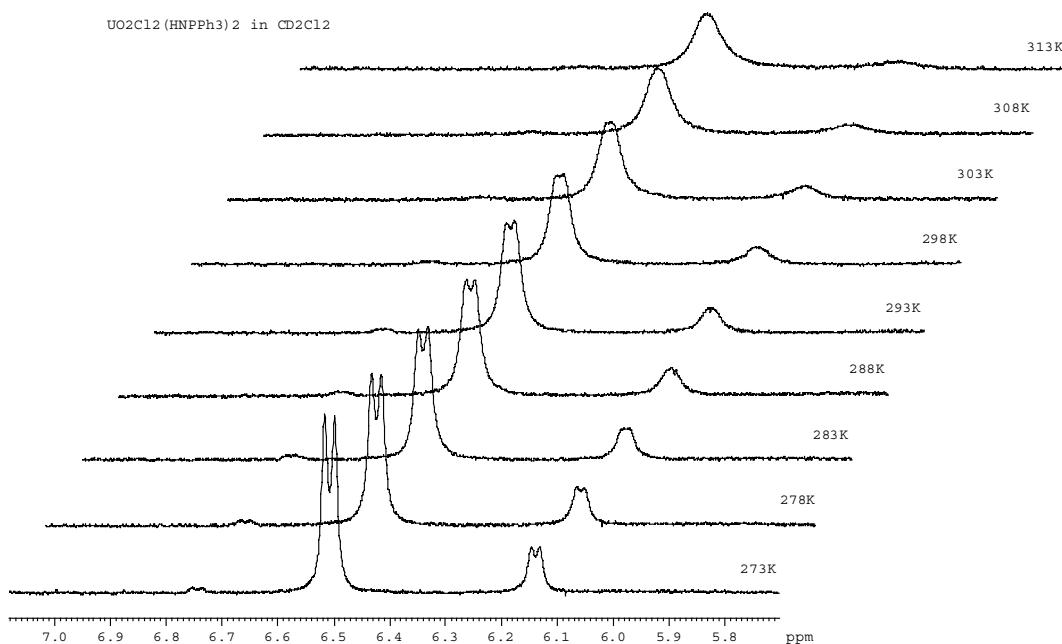
Preference for Nitrogen versus Oxygen Donor Coordination in Uranyl and Neptunyl (VI) Complexes.

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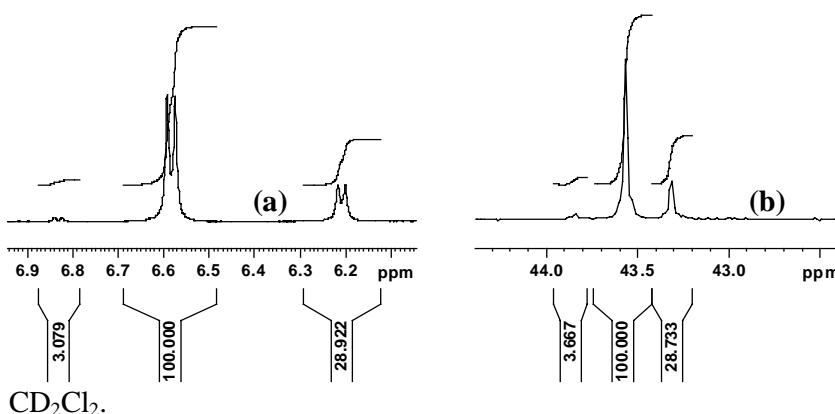
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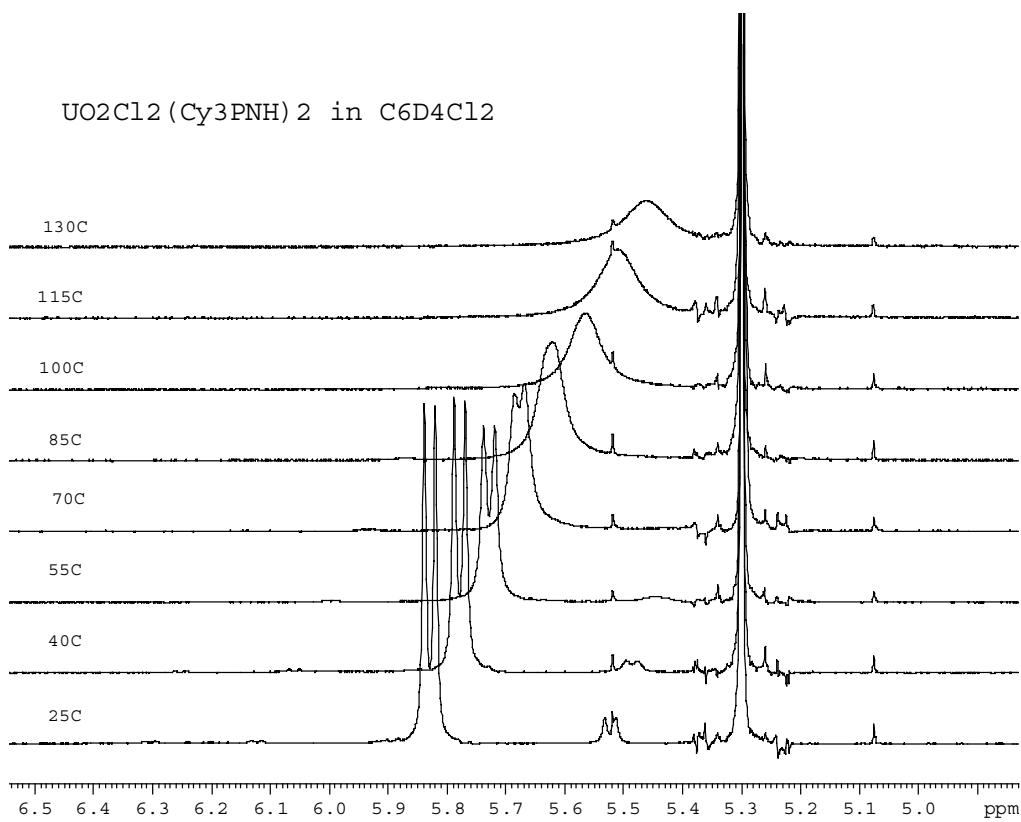
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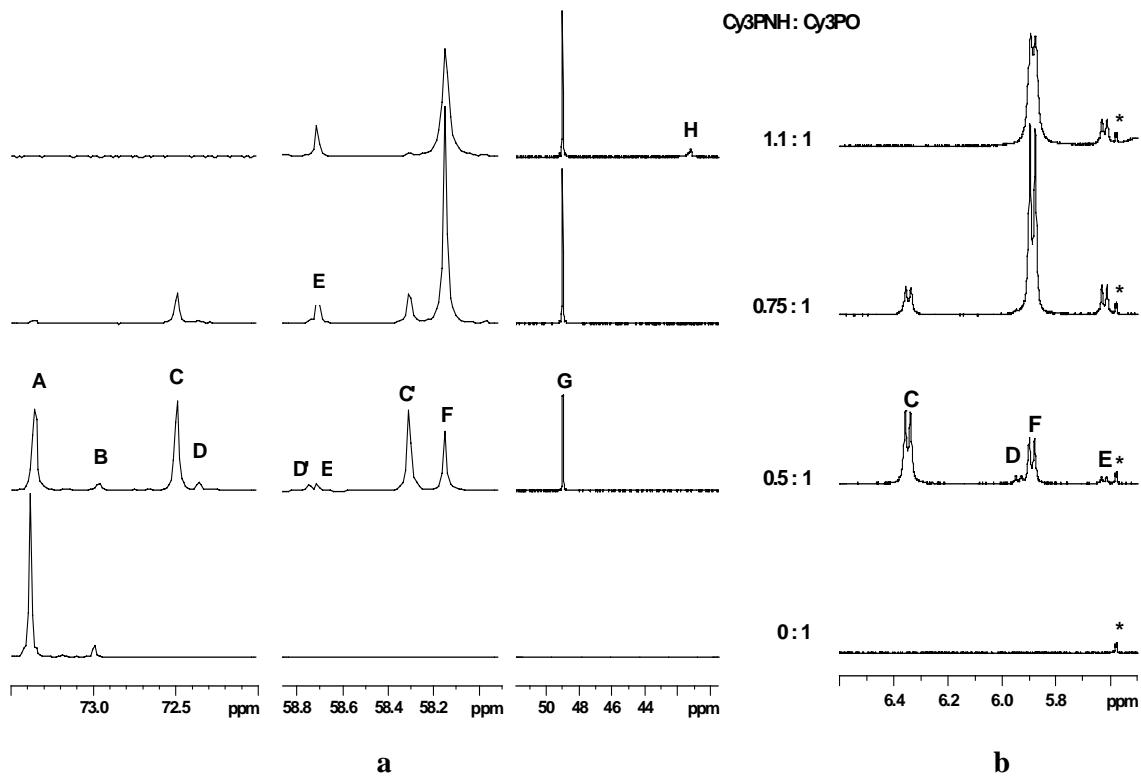
¹H Variable Temperature NMR spectra(P=NH region) of $\text{UO}_2\text{Cl}_2(\text{Ph}_3\text{PNH})_2$ (**1**) in



(a) ¹H (in the PNH region) and (b) ³¹P NMR spectroscopy of $\text{UO}_2\text{Cl}_2(\text{Ph}_3\text{PNH})_2$ (**1**) (in CD_2Cl_2 at 0°C).



¹H Variable Temperature NMR spectra (P=NH region) of UO₂Cl₂(Cy₃PNH)₂ (**2**) in C₆D₄Cl₂. The peak at 5.31 ppm is due to the residual C₆H₄Cl₂ solvent with ¹³C satellites at 5.07 and 5.52 ppm.

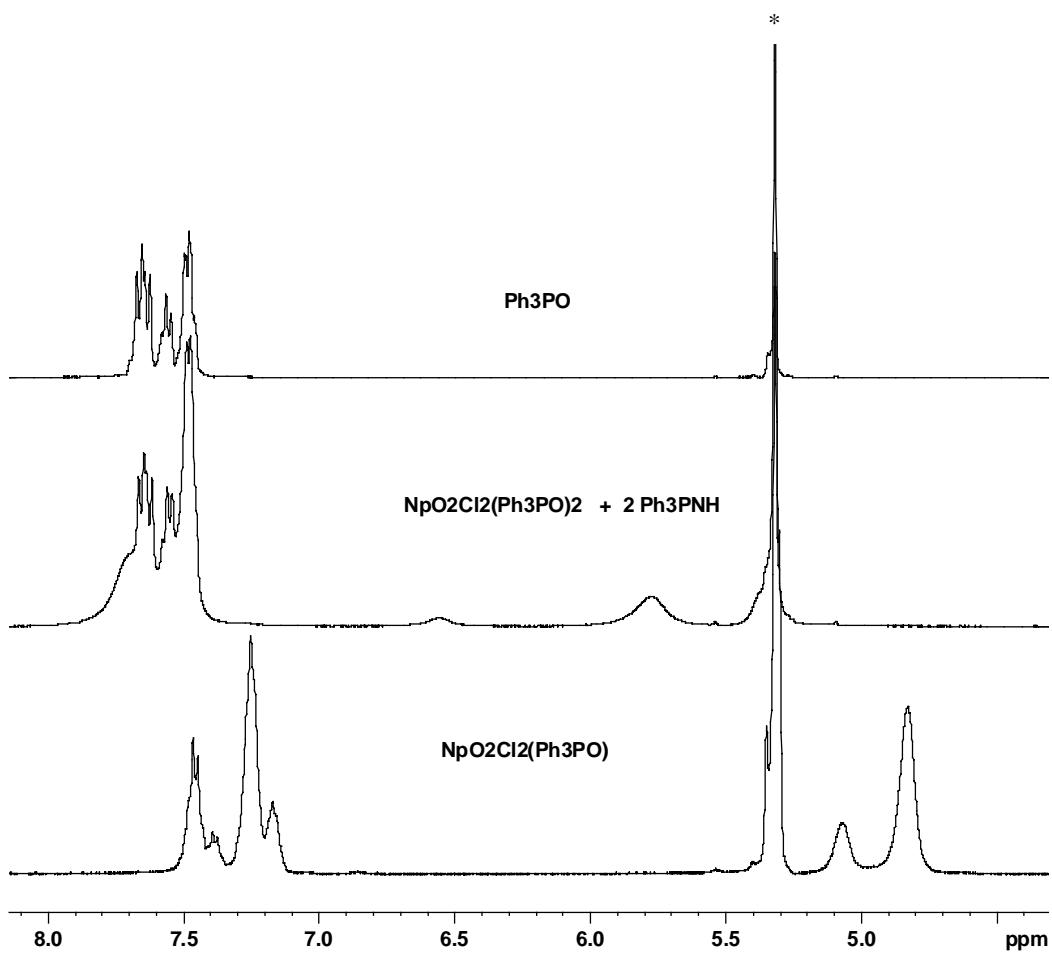


(a) ^{31}P NMR and (b) ^1H NMR ($\text{P}=\text{NH}$ region) of the reaction between $\text{UO}_2\text{Cl}_2(\text{Cy}_3\text{PO})_2$ (**3**) and Cy_3PNH at varying ratios of $\text{Cy}_3\text{PO} : \text{Cy}_3\text{PNH}$ in the system. (*) denotes a ^{13}C satellite of CD_2Cl_2 solvent) (see Table 1 for peak assignments).

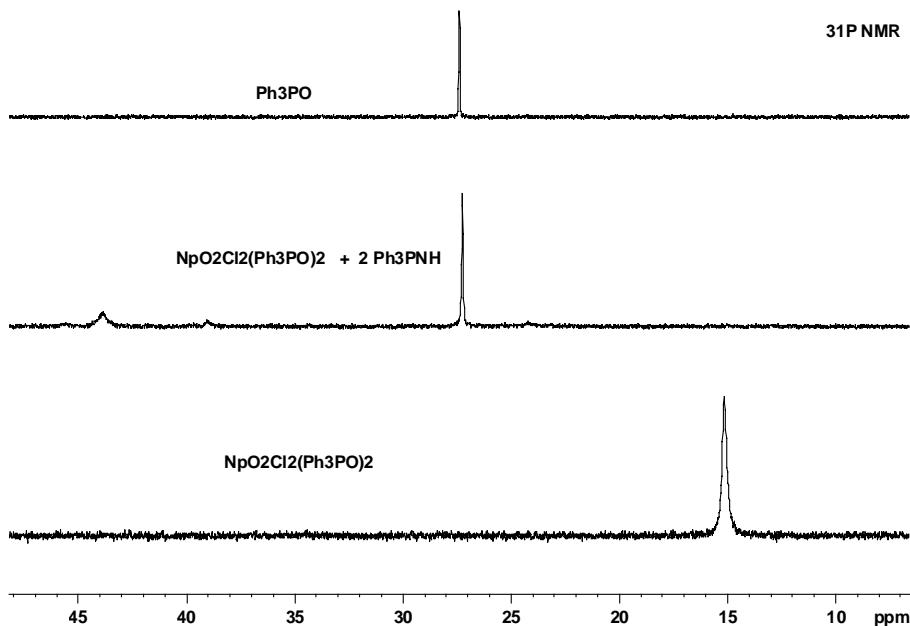
Table 1. Assignments of the chemical shifts for the above spectra. Note the assignment of *cis* and *trans* isomers are estimated on the basis that the *trans* isomer is always the higher chemical shift of the two isomers in the ^{31}P NMR spectra.

Complex	^{31}P NMR δ (ppm)	^1H NMR δ (ppm) [J_{PH}]
	(ppm)	(Hz)]
<i>trans</i> - $\text{UO}_2\text{Cl}_2(\text{Cy}_3\text{PO})_2$ (A)	73.4	n/a
<i>cis</i> - $\text{UO}_2\text{Cl}_2(\text{Cy}_3\text{PO})_2$ (B)	73.0	n/a
<i>trans</i> - $\text{UO}_2\text{Cl}_2(\text{Cy}_3\text{PO})(\text{Cy}_3\text{PNH})$ (C,C')	72.5, 58.3	6.35 [7.6]
<i>cis</i> - $\text{UO}_2\text{Cl}_2(\text{Cy}_3\text{PO})(\text{Cy}_3\text{PNH})$ (D,D')	72.4, 58.8	5.94 [8.1]
<i>trans</i> - $\text{UO}_2\text{Cl}_2(\text{Cy}_3\text{PNH})_2$ (E)	58.7	5.62 [7.6]
<i>cis</i> - $\text{UO}_2\text{Cl}_2(\text{Cy}_3\text{PNH})_2$ (F)	58.2	5.89 [7.8]
Cy_3PO (G)	49.0	n/a
Cy_3PNH (H)	41.0	0.40

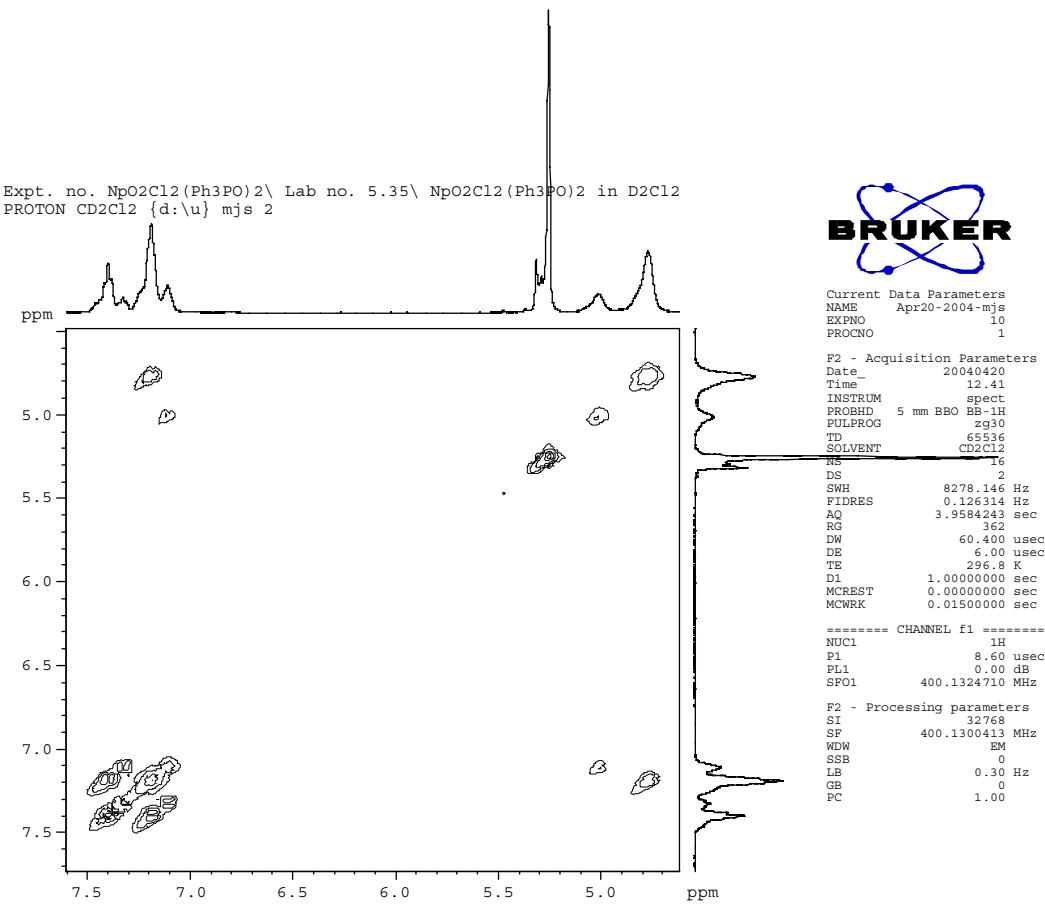
n/a = not applicable



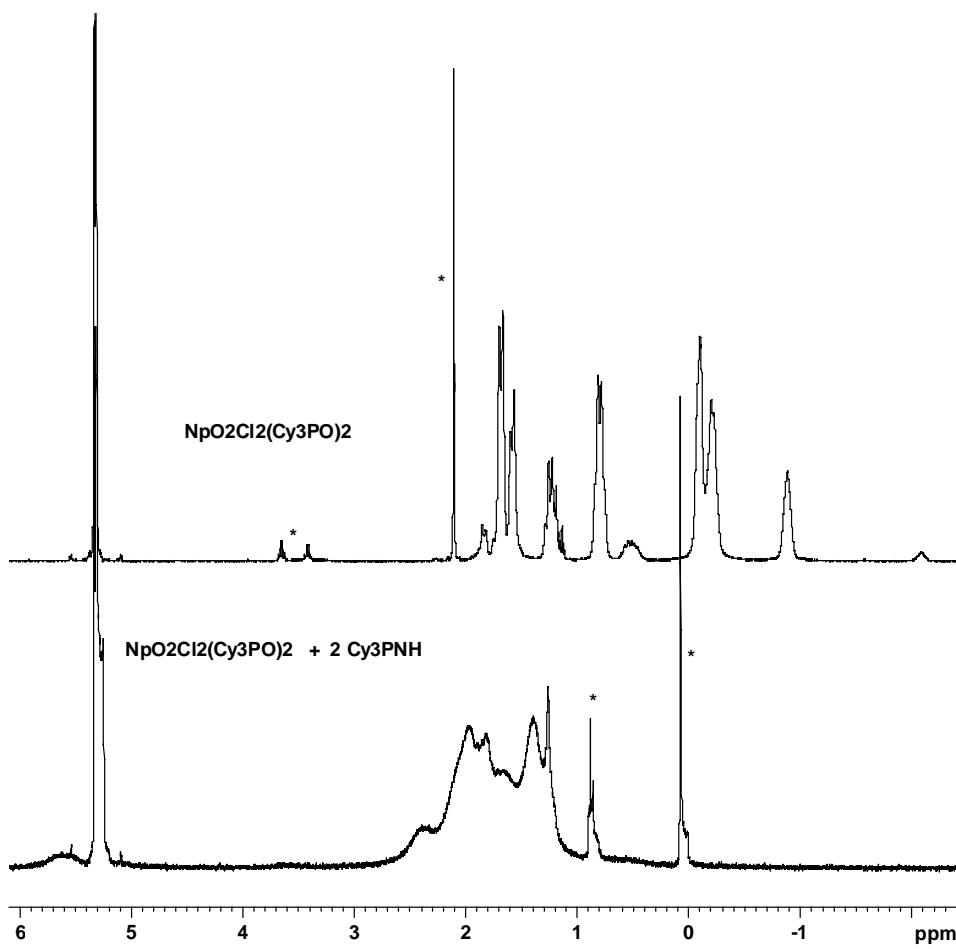
¹H NMR of the reaction of NpO₂Cl₂(Ph₃PO)₂ (**5**) with Ph₃PNH (* = residual solvent peak)



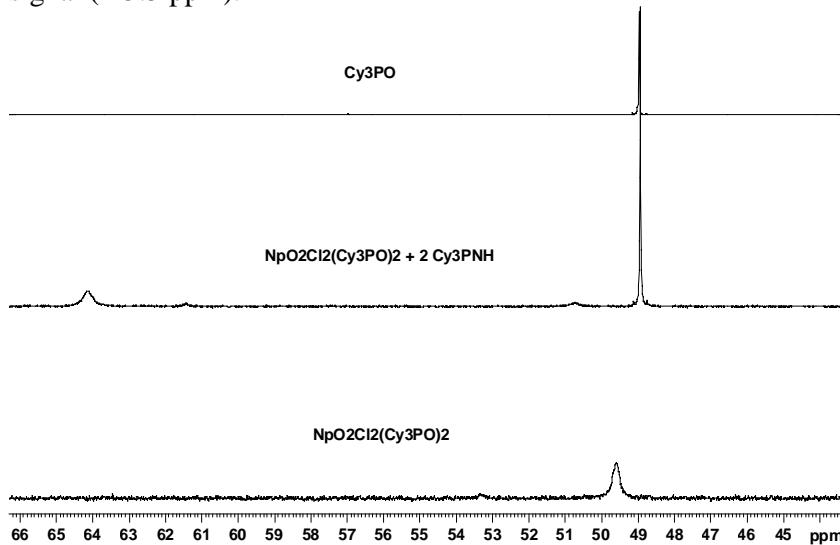
³¹P NMR of the reaction of NpO₂Cl₂(Ph₃PO)₂ (**5**) with Ph₃PNH



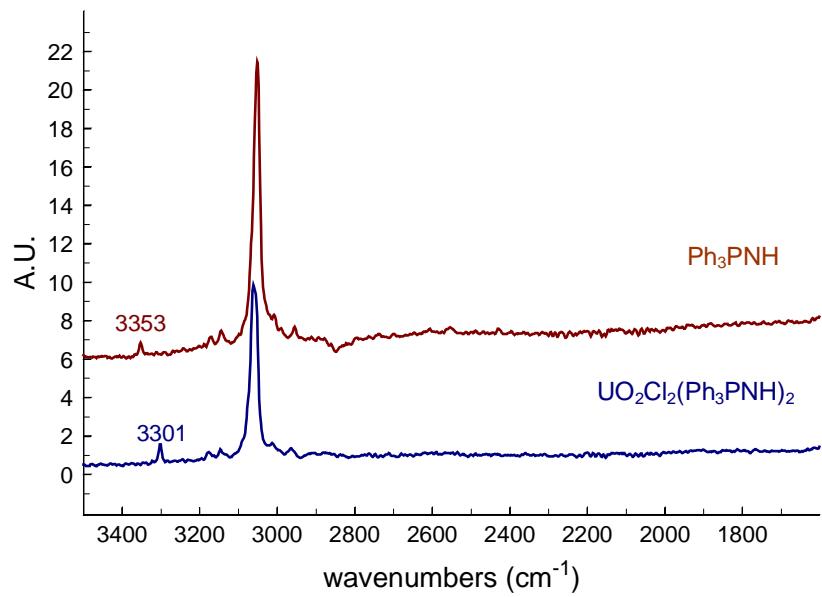
COSY spectrum of NpO₂Cl₂(Ph₃PO)₂ (**5**) confirming the assignment of the ortho-protons at ~4.8 ppm.



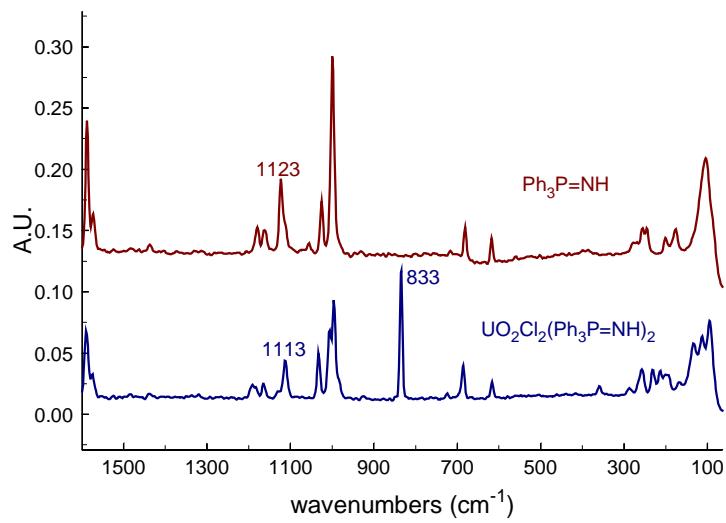
¹H NMR of the reaction of $\text{NpO}_2\text{Cl}_2(\text{Cy}_3\text{PO})_2$ (**6**) with Cy_3PNH . (*) denotes impurity in solvent). Note broad peak just to the left of the solvent peak is assigned the NH signal (~ 5.5 ppm).



³¹P NMR of the reaction of $\text{NpO}_2\text{Cl}_2(\text{Cy}_3\text{PO})_2$ (**6**) with Cy_3PNH . (Signal at 51 ppm in the middle spectrum indicates hydrolysis of some of the ligand and formation of $[\text{Cy}_3\text{PNH}_2]^+$).



Raman spectrum of $\text{UO}_2\text{Cl}_2(\text{Ph}_3\text{PNH})_2$ (**1**) showing the change in NH stretch on going from free to complexed ligand



Raman spectrum of $\text{UO}_2\text{Cl}_2(\text{Ph}_3\text{PNH})_2$ (**1**) showing the O=U=O_{sym} stretch