Selective Removal of Metallic Single-Walled Carbon Nanotubes with Small-Diameters by Using Nitric and Sulfuric Acids

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Supporting Information-Figure S1

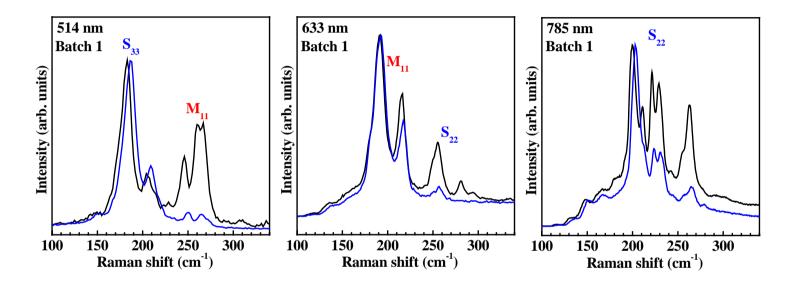


Figure S1. RBM Raman spectra for batch 1 HiPco SWCNTs at 514, 633, and 785 nm excitation wavelengths. The black and blue lines are from the pristine and air-oxidized HiPco SWCNTs (at 350 $^{\circ}$ C in air atmosphere). M_{ii} and S_{ii} correspond to metallic and semiconducting interband transitions. Gas-phase oxidation treatment led to diameter-selective removal of SWCNTs, regardless of metallicity.

Fig. S1. Yang et al.

Supporting Information-Figure S2

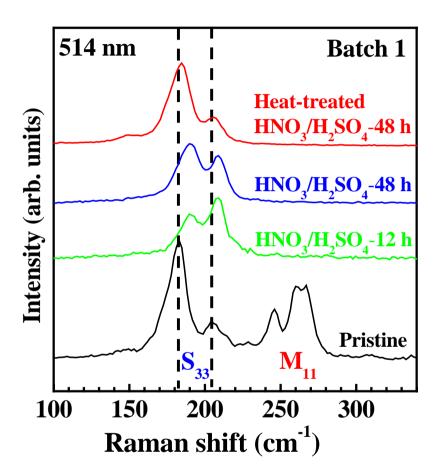


Figure S2. RBM Raman spectra from the leftover on the filter for batch 1 HiPco SWCNTs at 514 nm excitation wavelength. The black, green, blue, and red lines are from the pristine, HNO₃/H₂SO₄ (1:9) treatment for 12 h, 48 h, and HNO₃/H₂SO₄ (1:9) treatment for 48 h followed by heat treatment at 900 °C in Ar atmosphere, respectively. The peak shift was recovered after the heat treatment.

Fig. S2. Yang et al.

Supporting Information-Figure S3

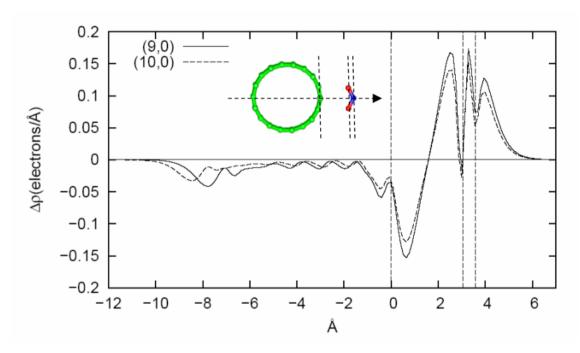


Figure S3. The redistribution of electronic density was calculated by $\Delta \rho = \rho(CNT+NO_2) - \rho(CNT) - \rho(NO_2)$. The figure was drawn along the arrowed direction shown in the inset figure and $\Delta \rho$ is averaged over the plane perpendicular to the plotting direction. Three vertical lines in the inset and figure indicate the positions of the outermost C, O, and N atoms, respectively. The enhanced charge transfer of metallic (9,0) nanotube was clearly visible.

Fig. S3. Yang et al.