## Supplementary Material

# Non-Racemic $\alpha$-Allenyl Carbinols from Asymmetric Propargylboration with the 10-Trimethylsilyl-9-borabicyclo[3.3.2]decanes 

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General Information. All experiments were carried out in pre-dried glassware ( $1 \mathrm{~h}, 150$ ${ }^{\circ} \mathrm{C}$ ) under a nitrogen atmosphere. Standard handing techniques for air-sensitive compounds were employed for all the operations. Nuclear magnetic resonance (NMR) spectra were obtained using General Electric DPX-300 spectrometer. ${ }^{1} \mathrm{H}(300 \mathrm{MHz}),{ }^{13} \mathrm{C}$ $(75 \mathrm{MHz}),{ }^{31} \mathrm{P}(121.5 \mathrm{MHz})$ and ${ }^{11} \mathrm{~B}(96.5 \mathrm{MHz})$ NMR were recorded in $\mathrm{CDCl}_{3}$ or $\mathrm{C}_{6} \mathrm{D}_{6}$, unless otherwise used, and the chemical shift as were expressed in ppm relative to $\mathrm{CDCl}_{3}$ ( $\delta 7.26$ and 77.0 for ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, respectively) and of $\mathrm{C}_{6} \mathrm{D}_{6}$ ( $\delta 7.15$ and 128.0 ppm for ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, respectively) as the internal standard. Infrared spectra were recorded on a Perkin-Elmer 282 spectrophotometer. Mass spectral data were obtained with a Hewlett-Packard 5995A GC/MS spectrometer ( 70 eV ), Fisons VG Autospect or a Hewlett-Packard 5971A Mass Selective Ion Detector. High-resolution mass spectral data were obtained with a Micromass VG AutoSpec magnetic sector mass spectrometer (70 eV ). Optical rotations were measured employing a Perkin-Elmer 243B polarimeter. Ozonolyses were conducted with a Polymetrics Laboratory Ozonator Model T-408 operating at $70 \mathrm{~V}\left(\mathrm{O}_{2}\right.$ pressure $=8 \mathrm{psig}$, flow rate $=0.46$ (nominal)). Literature citations are provided for all known compounds together with the scanned spectra obtained in this study to consolidate more complete information herein.

## Experimental Procedures

3-Trimethylsilyl-2-propynylmagnesium bromide In a 100 mL three neck round bottom flask equipped with a stirring bar, a dry-ice condenser and a 50 mL addition funnel, To pulverized Mg powder $\left(0.177 \mathrm{~g}, 7.3 \mathrm{mmol}\right.$ ) was added and $\mathrm{HgCl}_{2} 1-2 \%(0.02 \mathrm{~g}, 0.073$ mmol ) as the initiator. Dry ether (from $\mathrm{Na} / \mathrm{Ph}_{2} \mathrm{CO}$ ) (ca. 1 mL ) was added and 3-bromo-1-(trimethylsilyl)-1-propyne ( $1.0 \mathrm{~mL}, 7.0 \mathrm{mmol}$ ) in ether ( 6.3 mL ) was added dropwise from the addition funnel with the contents of the flask being stirred vigorously. After the addition was complete the solution is refluxed for 2 h . A dark-green color indicates the formation of the Grignard reagent. The reaction is cooled at room temperature under positive pressure of nitrogen to produce the Grignard reagent $(6.0 \mathrm{~mL}, 0.67 \mathrm{M}, 4.0 \mathrm{mmol})$ (55\%).

( $\pm$ )- $B$-[ $\gamma$-(Trimethylsilyl)propargyl]-10-trimethylsilyl-9-borabicyclo[3.3.2]-decane (1). Method A. To a stirred solution of ( $\pm$ )-3 ${ }^{1}$ (Fig. 1) $(0.952 \mathrm{~g}, 4.0 \mathrm{mmol})$ in ether (110 mL ) at $-78 \quad{ }^{\circ} \mathrm{C}$, was added a solution of freshly prepared $\gamma$ -
(trimethylsilyl)propargylmagnesium bromide in ether ( $6.0 \mathrm{~mL}, 0.67 \mathrm{M}$ ) and the mixture was stirred for 1 h at $-78{ }^{\circ} \mathrm{C}$. The solution was allowed to slowly warm to room temperature and the solvents were removed in vacuo and hexane ( 10 mL ) was added to the residue. The slurry was filtered under a nitrogen atmosphere through a celite pad employing a double-ended needle to effect the transfer. This washing/filtration procedure was repeated two more times and the combined filtrates were concentrated at reduced pressure to obtain $1.25 \mathrm{~g}(98 \%)$ of $\mathbf{1} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 0.13(\mathrm{~s}, 3 \mathrm{H}), 0.14(\mathrm{~s}$, $3 \mathrm{H}), 1.4-1.8(\mathrm{~m}, 15 \mathrm{H}), 2.2(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}^{\mathrm{NMR}}\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta 0.06,1.5,20.0,21.7$, $24.8,25.1,27.8,29.2,30.9,31.4,33.6,34.9,40.1,84.0,127.4$ (Fig. 3); ${ }^{11} \mathrm{~B}$ NMR $\left(\mathrm{CDCl}_{3}, 96 \mathrm{MHz}\right) \delta 84.5$. IR (neat) $2913,2852,2172,1278,1247,835,758 \mathrm{~cm}^{-1}$. Method B. To a stirred solution of 1-(trimethylsilyl)propyne ( $0.74 \mathrm{~mL}, 5.0 \mathrm{mmol}$ ) in THF ( 5 mL ) at $0{ }^{\circ} \mathrm{C}$ was added tert-butyllithium in pentane ( $1.7 \mathrm{M}, 5.0 \mathrm{mmol}$ ) dropwise, and the reaction mixture was stirred for 1 h . The mixture was cooled to $-78^{\circ} \mathrm{C}$ and added dropwise via double ended needle to a solution of $( \pm)-3^{1}(1.2 \mathrm{~g}, 5.0 \mathrm{mmol})$ in ether (10 mL ) at $-78{ }^{\circ} \mathrm{C}$. After 30 min , trimethylsilyl triflate ( $1.0 \mathrm{~mL}, 5.3 \mathrm{mmol}$ ) was added dropwise, and the solution was stirred for 1 h at $-78{ }^{\circ} \mathrm{C}$, allowed to slowly warm to room temperature, and concentrated at reduced pressure. Hexane $(10 \mathrm{~mL})$ was added to the residue and the lithium salts were removed by filtration as above to give $1.1 \mathrm{~g}(85 \%)$ of $\mathbf{1}$. Note! The lithium triflate salt was difficult to remove and several filtrations were sometimes required which both lowered the yield and introduced the possibility of minor amounts of oxidation in $\mathbf{1}$. Note! Both methods were used to obtain (-)-1R from (-)-3R and $(+)-\mathbf{1} \boldsymbol{S}$ from $(+)-3 \boldsymbol{S} .{ }^{1}$
(-)-(10R)-B-[ $\gamma$-(Trimethylsilyl)propargyl]-10-TMS-9-borabicyclo[3.3.2]decane ((-)$\mathbf{1 R}$ ). To a stirred solution of (+)-2R(Fig. 2) ( $1.5 \mathrm{~g}, 4.0 \mathrm{mmol}$ ) in ether ( $16 \mathrm{~mL}, 0.25 \mathrm{M}$ ) at $-78 \quad{ }^{\circ} \mathrm{C}$ was added dropwise to a solution of freshly prepared trimethylsilylpropynemagnesium bromide $(6.0 \mathrm{~mL}, 0.67 \mathrm{M})$ in dry ether and the mixture was stirred for 1 h . The reaction mixture was allowed to slowly warm to room temperature. The solvent was removed in vacuo and hexane was added to the residue. The magnesium salts were filtered through a celite pad employing a double-ended needle to effect the transfer. The filtrate was dissolved in water and extracted with ether to give $0.602 \mathrm{~g}(92 \%)$ recovered $(1 S, 2 S)-(+)$-pseudoephedrine. The eluent was concentrated to obtain $1.25 \mathrm{~g}(98 \%)$ of $(-)-\mathbf{1 R} .[\alpha]^{20}{ }_{\mathrm{D}}=-10.5^{\circ}\left(c \quad 4.2, \mathrm{C}_{6} \mathrm{D}_{6}\right)$. (+)- $\mathbf{1} \boldsymbol{S}$ was similarly prepared from (-)-2S. $[\alpha]^{20}{ }_{D}=+11.3^{\circ}\left(c \quad 4.2, \mathrm{C}_{6} \mathrm{D}_{6}\right)$.

## Representative Procedure for the Propargylboration of Aldehydes with ( $\pm$ )-1.

1-Phenyl-2-(trimethylsilyl)-2,3-butadien-1-ol ( $\mathbf{\pm}$ )-(6e). Representative procedure: A solution of $\mathbf{1}(1.27 \mathrm{~g}, 4.0 \mathrm{mmol})$ in dry THF ( 5 mL ) was cooled to $-78{ }^{\circ} \mathrm{C}$ and PhCHO $(0.3 \mathrm{~mL}, 4.0 \mathrm{mmol})$, was added. After 3 h , the solvents were removed in vacuo, to give borinate $5 \mathbf{e}(1.68 \mathrm{~g}, 3.96 \mathrm{mmol})$ in $99 \%$ yield. 8 -Hydroxyquinoline ( $0.581 \mathrm{~g}, 4.0 \mathrm{mmol}$ ), was added followed by dry acetonitrile ( 6 mL ) and refluxed for 10 h . The mixture is cooled slowly and the supernatant is decanted into another flask. The precipitated brightyellow crystals were dried at reduced pressure ( $54 \%$ yield, $0.758 \mathrm{~g}, 2.16 \mathrm{mmol}$ ). The
solution was distilled to give $0.762 \mathrm{~g}(87 \%)$ of $( \pm)$-( $\mathbf{6 e})$ in $87 \%$ yield. bp $125^{\circ} \mathrm{C} 1.0 \mathrm{~mm}$ Hg . The spectral data is identical to that of $\mathbf{6 e}$.

## Propargylboration of Representative Aldehydes with (-)-1R or (+)-1S.


(-)-(2S)-3-(Trimethylsilyl)-3,4-pentadien-2-ol (6a). A solution of (-)-1R(1.27 g, 4.0 mmol ) in dry THF ( 16 mL ) was cooled to $-78^{\circ} \mathrm{C}$ and $\mathrm{MeCHO}(0.22 \mathrm{~mL}, 4.0 \mathrm{mmol})$ was added dropwise. After 3 h , the solvents were removed at reduced pressure to give borinate $5 \mathbf{a}(1.43 \mathrm{~g})$. The $(1 S, 2 S)-(+)$-pseudoephedrine $(0.66 \mathrm{~g}, 4.0 \mathrm{mmol})$ and freshly distilled acetonitrile ( 8 mL ) were added and the mixture was heated at reflux temperature for 4 h . The solution was slowly cooled to room temperature and decanted into another flask and the precipitated white crystals were washed with hexane $(3 \times 5 \mathrm{~mL})$ to yield 1.04 $\mathrm{g}(71 \%)$ of $(+) \mathbf{- 2 R}$. The residue was distilled to obtain $0.49 \mathrm{~g}(78 \%)$ of $\mathbf{6 a}, \mathrm{bp} 125^{\circ} \mathrm{C}, 60$ $\mathrm{mm} \mathrm{Hg} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 0.10(\mathrm{~s}, 9 \mathrm{H}), 1.30(\mathrm{~d}, 3 \mathrm{H}), 2.15(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.30$ $(\mathrm{m}, 1 \mathrm{H}), 4.50(\mathrm{~d}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta-0.95,24.2,66.6,71.9,101.8,206.7$ (Fig. 4). IR (neat) $3349,1925,1247,1076,835,810 \mathrm{~cm}^{-1} .[\alpha]^{26}{ }_{\mathrm{D}}=-8.9^{\circ}\left(c 1.31, \mathrm{CHCl}_{3}\right)$, $\operatorname{lit}^{3}(2 R):[\alpha]^{25}{ }_{\mathrm{D}}=+11.9^{\circ}\left(c\right.$ 1.27, $\left.\mathrm{CHCl}_{3}\right) ;$ LRGCMS $m / z[\mathrm{M}]^{+} 155,141,117,97,75,73$, 75, 66 .

(-)-(4R)-3-(Trimethylsilyl)-1,2-heptadien-4-ol (6b). ${ }^{\mathbf{2}}$ A solution of (+)- $\mathbf{S} \boldsymbol{S}(1.27 \mathrm{~g}, 4.0$ $\mathrm{mmol})$ in dry THF ( 16 mL ) was cooled to $-78^{\circ} \mathrm{C}$ and $n$ - $\mathrm{PrCHO}(0.36 \mathrm{~mL}, 4.0 \mathrm{mmol})$ was added dropwise. After 3 h , the solvents were removed at reduced pressure to give cleanly borinate $\mathbf{5 b}(1.39 \mathrm{~g})$. The $(1 R, 2 R)-(-)$-pseudoephedrine $(0.61 \mathrm{~g}, 3.7 \mathrm{mmol})$ and freshly distilled acetonitrile ( 8 mL ) were added and the mixture was heated at reflux temperature for 9 h . The solution was slowly cooled to room temperature and decanted into another flask and the precipitated white crystals were washed with hexane $(3 \times 5 \mathrm{~mL})$ to yield 1.16 $\mathrm{g}(85 \%)$ of (-)-2S. The residue was distilled to obtain $0.64 \mathrm{~g}(87 \%)$ of $\mathbf{6 b}$, bp $100^{\circ} \mathrm{C}, 1.0$ mm Hg . Anal. calcd for $\mathrm{C}_{10} \mathrm{H}_{20} \mathrm{OSi}$ : C 65.15, H 10.94; found: C 65.09 , H $10.90 .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 0.10(\mathrm{~s}, 9 \mathrm{H}), 0.85(\mathrm{t}, 3 \mathrm{H}), 1.3-1.65(\mathrm{~m}, 4 \mathrm{H}), 1.75(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.15(\mathrm{t}$, $1 \mathrm{H}), 4.5(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta-1.3,13.9,18.8,40.0,70.4,71.7,100.7$, 207.2 (Fig. 5). IR (neat) $3417,1928,1249,1100,839 \mathrm{~cm}^{-1} . \quad[\alpha]^{20}{ }_{\mathrm{D}}=-6.0^{\circ}$ (c 1.54, $\left.\mathrm{CHCl}_{3}\right)$; LRGCMS $m / z[\mathrm{M}]^{+}\left(\mathrm{C}_{10} \mathrm{H}_{20} \mathrm{OSi}\right) 183,169,145,79,75,73,55$. This experiment
was repeated with (-)-1R which gave $\mathbf{6 b}$ ' whose spectral properties were identical to those of $\mathbf{6 b}$, but exhibited the opposite specific rotation.

(-)-(3R)-2-Methyl-4-(trimethylsilyl)-4,5-hexadien-3-ol (6c). A solution of (+)- $\mathbf{1 S}$ (1.11 $\mathrm{g}, 3.5 \mathrm{mmol})$ in dry THF ( 16 mL ) was cooled to $-78{ }^{\circ} \mathrm{C}$ and $i$-PrCHO $(0.32 \mathrm{~mL}, 3.5$ mmol ) was added dropwise. After 3 h , the solvents were removed at reduced pressure to give borinate $5 \mathrm{c}(1.23 \mathrm{~g})$. The $(1 R, 2 R)-(-)$-pseudoephedrine $(0.54 \mathrm{~g}, 3.3 \mathrm{mmol})$ and freshly distilled acetonitrile ( 7 mL ) were added and the mixture was heated at reflux temperature for 10 h . The solution was slowly cooled to room temperature and decanted into another flask and the precipitated white crystals were washed with hexane ( $3 \times 5 \mathrm{~mL}$ ) to yield $0.94 \mathrm{~g}(78 \%)$ of ( $-\mathbf{)} \mathbf{- 2 S}$. The solution was distilled to obtain $0.49 \mathrm{~g}(77 \%)$ of $\mathbf{6 c}$, bp $85{ }^{\circ} \mathrm{C}, 1.0 \mathrm{~mm} \mathrm{Hg} .{ }^{1} \mathrm{H}\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 0.10(\mathrm{~s}, 9 \mathrm{H}), 0.85(\mathrm{~d}, 3 \mathrm{H}), 0.90(\mathrm{~d}, 3 \mathrm{H})$, $1.75(\mathrm{~m}, 1 \mathrm{H}), 1.80(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 3.85(\mathrm{~d}, 1 \mathrm{H}), 4.50(\mathrm{~d}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta-$ 1.3, 16.2, 20.0, 33.6, 71.6, 75.4, 99.8, 207.3 (Fig. 6). IR (neat) 3446, 1926, 1248, 1020, $837 \mathrm{~cm}^{-1} .[\alpha]^{23.5}{ }_{\mathrm{D}}=-5.4^{\circ}\left(c\right.$ 1.94, $\left.\mathrm{CHCl}_{3}\right), \mathrm{lit}^{3}[\alpha]^{26}{ }_{\mathrm{D}}=-5.3^{\circ}\left(c 1.96, \mathrm{CHCl}_{3}\right)$; LRGCMS $m / z[\mathrm{M}]^{+} 184,168,145,125,79,73,75$.

(+)-(3R)-2,2-Dimethyl-4-(trimethylsilyl)-4,5-hexadien-3-ol (6d). A solution of (+)-1S $(1.25 \mathrm{~g}, 3.92 \mathrm{mmol})$ in dry THF $(16 \mathrm{~mL})$ was cooled to $-78{ }^{\circ} \mathrm{C}$ and $t$-BuCHO $(0.38 \mathrm{~mL}$, 3.5 mmol ) was added dropwise. After 3 h , the solvents were removed at reduced pressure to give borinate $5 \mathbf{d}(1.58 \mathrm{~g})$. The $(1 R, 2 R)-(-)$-pseudoephedrine $(0.64 \mathrm{~g}, 3.88$ mmol ) and freshly distilled acetonitrile ( 8 mL ) were added and the mixture was heated at reflux temperature for 10 h . The solution was slowly cooled to room temperature and decanted into another flask and the precipitated crystals were washed with hexane ( $3 \times 5$ $\mathrm{mL})$ to yield $1.00 \mathrm{~g}(70 \%)$ of $(-)-\mathbf{2 S}$. The residue was distilled to obtain $0.56 \mathrm{~g}(80 \%)$ of 6d, bp $105{ }^{\circ} \mathrm{C}, 1.0 \mathrm{~mm} \mathrm{Hg} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 0.11(\mathrm{~s}, 9 \mathrm{H}), 0.92(\mathrm{~s}, 9 \mathrm{H}), 1.8$ (br s, 1 H ), $3.75(\mathrm{~s}, 1 \mathrm{H}), 4.5(\mathrm{dd}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta-0.6,26.1,37.4,71$. 4, 77. 6, 98.4, 208.6 (Fig. 7). IR (neat) 3479, 1921, 1248, 1050, 1007, 836, $803 \mathrm{~cm}^{-1}$. $[\alpha]^{25}{ }_{\mathrm{D}}=+5.0^{\circ}\left(c 1.13, \mathrm{CHCl}_{3}\right), \mathrm{lit}^{3}[\alpha]^{25}=+5.4^{0}\left(c 2.60, \mathrm{CHCl}_{3}\right) ;$ LRGCMS $m / z \quad[\mathrm{M}]^{+}$ $197,141,125,93,75,73,57$.

(-)-(1S)-1-Phenyl-2-(trimethylsilyl)-2,3-butadien-1-ol (6e). A solution of (-)-1R(0.96 $\mathrm{g}, 3.0 \mathrm{mmol})$ in dry THF $(16 \mathrm{~mL})$ was cooled to $-78^{\circ} \mathrm{C}$ and PhCHO $(0.30 \mathrm{~mL}, 3.0 \mathrm{mmol})$ was added dropwise. After 3 h , the solvents were removed at reduced pressure to give borinate $5 \mathbf{e}(1.23 \mathrm{~g})$. The ( $1 S, 2 S$ )-(+)-pseudoephedrine ( $0.48 \mathrm{~g}, 2.91 \mathrm{mmol}$ ) and freshly distilled acetonitrile ( 8 mL ) were added and the mixture was heated at reflux temperature for 15 h . The solution was slowly cooled to room temperature and decanted into another flask and the precipitated white crystals were washed with hexane $(3 \times 5 \mathrm{~mL})$ to yield 0.76 $\mathrm{g}(70 \%)$ of $(+) \mathbf{- 2 R}$. The solution was distilled to obtain $0.40 \mathrm{~g}(60 \%)$ of $\mathbf{6 e}, \mathrm{bp} 125^{\circ} \mathrm{C}$, $1.0 \mathrm{~mm} \mathrm{Hg} .{ }^{1} \mathrm{H}\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 0.0(\mathrm{~s}, 9 \mathrm{H}), 2.61(\mathrm{~d}, 1 \mathrm{H}), 4.65(\mathrm{dd}, 2 \mathrm{H}), 5.25(\mathrm{~s}$, $1 \mathrm{H}), 7.35(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta-1.2,72.5,72.8,99.5,126.8,127.6$, 128.1, 143.0, 207.0 (Fig. 8). IR (neat) 3393, 3029, 1955, 1267, 1024, 905, 843, 749, 698 $\mathrm{cm}^{-1} .[\alpha]^{20}{ }_{\mathrm{D}}=-122.8^{\circ}\left(c 1.40, \mathrm{CHCl}_{3}\right), \mathrm{lit}^{3}[\alpha]^{25}{ }_{\mathrm{D}}=-139.4^{\circ}\left(c 1.40, \mathrm{CHCl}_{3}\right) ;$ LRGCMS $m / z[\mathrm{M}]^{+} 218,179,128,107,79,77,75,73,51$. This experiment was repeated with (+)$\mathbf{1 S}$ which gave $\mathbf{6 e}$ ' whose spectral properties were identical to those of $\mathbf{6 e}$, but exhibited the opposite specific rotation.

$(+)-(4 S)-(E)$-3-(Trimethylsilyl)-1,2,5-heptatrien-4-ol (6f). A solution of (-)-1R(1.11 g, $3.5 \mathrm{mmol})$ in dry THF ( 16 mL ) was cooled to $-78^{\circ} \mathrm{C}$ and $(E)$-crotonaldehyde $(0.3 \mathrm{~mL}, 3.5$ mmol ) was added dropwise. After 3 h , the solvents were removed at reduced pressure to give borinate $5 \mathrm{f}(1.35 \mathrm{~g})$. The $(1 R, 2 R)-(-)-$ pseudoephedrine $(0.58 \mathrm{~g}, 3.5 \mathrm{mmol})$ and freshly distilled acetonitrile ( 8 mL ) were added and the mixture was heated at reflux temperature for 10 h . The solution was slowly cooled to room temperature and decanted into another flask and the precipitated white crystals were washed with hexane ( $3 \times 5 \mathrm{~mL}$ ) to yield $1.10 \mathrm{~g}(85 \%)$ of (-)-2S. The solution was distilled to obtain $0.56 \mathrm{~g}(87 \%)$ of $\mathbf{6 f}$, bp $80^{\circ} \mathrm{C}, 1.0 \mathrm{~mm} \mathrm{Hg} .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 0.15(\mathrm{~s}, 9 \mathrm{H}), 1.70(\mathrm{dd}, J=6.4 \mathrm{~Hz}, J=1.4 \mathrm{~Hz}$, $3 \mathrm{H}), 1.89(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.60(\mathrm{~m}, 3 \mathrm{H}), 5.50(\mathrm{ddq}, J=15.3 \mathrm{~Hz}, J=7.4 \mathrm{~Hz}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H})$, $5.70(\mathrm{dq}, J=15.4 \mathrm{~Hz}, J=6.4 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta-1.0,17.4,71.4$, 72.3, 100.3, 126.9, 133.2, 206.8 (Fig. 9). IR (neat) 3308, 3044, 2084, 1639, 1247, 1025, $977,851 \mathrm{~cm}^{-1} .[\alpha]^{25}{ }_{\mathrm{D}}=+57.7^{\circ}\left(c 1.60, \mathrm{CHCl}_{3}\right), \mathrm{lit}^{3}(4 R)[\alpha]^{25}{ }_{\mathrm{D}}=-87.8^{\circ}\left(c 1.96, \mathrm{CHCl}_{3}\right)$; LRGCMS $m / z[\mathrm{M}]^{+} 182,167,143,112,75,73,71,53$.

(S)-(+)- $\alpha$-Hydroxyphenylacetic acid (Mandelic acid, 9). 6e ( $0.060 \mathrm{~g}, 0.27 \mathrm{mmol}$ ) was dissolved in dichloromethane ( 30 mL ), and the solution was cooled to $-78^{\circ} \mathrm{C}$. Ozone was bubbled through the solution until a blue color persisted ( 11 min ). The solvents were removed to give the trimethylsilyl ester intermediate $8\left({ }^{13} \mathrm{C}\right.$ NMR $\delta 176(\mathrm{SiOC}=\mathrm{O}), 1.8$ (TMS)). THF ( 3 mL ) was added followed by 1 equiv of water ( $0.005 \mathrm{~g}, 0.27 \mathrm{mmol}$ ). The mixture was stirred for 3 h at room temperature and the solvents were removed in vacuo to give $9(0.041 \mathrm{~g}, 100 \%) .[\alpha]^{20}{ }_{\mathrm{D}}=+149^{\circ}\left(c 2.5 \mathrm{H}_{2} \mathrm{O}\right)$, lit. ${ }^{4}[\alpha]^{25}{ }_{\mathrm{D}}=+155^{\circ}\left(c 5 \mathrm{H}_{2} \mathrm{O}\right)$; ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{CD}_{3} \mathrm{OD}\right) ~ \oint 4.90(\mathrm{~s}, 1 \mathrm{H}), 7.10-7.35(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 75 \mathrm{MHz}\right) \delta 72.6$, 126.6, 127.6, 127.7, 138.2, 175.4 (Fig. 10). In a separate experiment, the ozonolysis was interrupted after 2.5 min , the mixture was concentrated and its ${ }^{13} \mathrm{C}$ NMR spectrum was recorded to clearly reveal the acylsilane intermediate 7 (i.e. $\delta 240.6$ (TMSC=O); -2.8 (TMS) (Fig. 17).

( $\boldsymbol{R}$ )-(-)- $\boldsymbol{\alpha}$-(Acetoxy)phenylacetic acid ( $\boldsymbol{O}$-Acetylmandelic acid, 9') ( $0.978 \mathrm{~g}, 0.45$ mmol ) was dissolved in 3 mL of dry THF. Pyridine ( $1.0 \mathrm{~mL}, 12.4 \mathrm{mmol}$ ) was added followed by acetic anhydride ( $1.0 \mathrm{~mL}, 10.6 \mathrm{mmol}$ ) at room temperature. The solution was allowed to react overnight. The mixture was extracted with water ( 4 X 3 mL ) to remove the precipitated salts. The organic layer was dried with $\mathrm{MgSO}_{4}$, filtrated and concentrated to give the acetylated product ( $0.117 \mathrm{~g}, 0.45 \mathrm{mmol}$ ) quantitatively. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CHCl}_{3}, 300 \mathrm{MHz}\right) \delta 0.02(\mathrm{~s}, 9 \mathrm{H}), 2.09(\mathrm{~s}, 3 \mathrm{H}), 4.53(\mathrm{~d}, 2 \mathrm{H}), 6.29(\mathrm{t}, 1 \mathrm{H}), 7.27-7.37(\mathrm{~m}$, $5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CHCl}_{3}, 75 \mathrm{MHz}\right) \delta-1.14,21.2,71.64,71.68,97.6,127.5,128.1,128.2$ $139.5,169.9,209.0$. This material was dissolved in dichloromethane ( 25 mL ) and cooled to $-78{ }^{\circ} \mathrm{C}$. Ozone was bubbled through the solution until a blue solution persisted ( 10 min .). The mixture was concentrated and dissolved in THF ( 5 mL ). Hydrogen peroxide ( 1 mL of $30 \%$ ) was added and the mixture was stirred at room temperature for 3 h . Water ( 3 mL ) was added and the layers were separated and the organic layer was dried $\left(\mathrm{MgSO}_{4}\right)$, filtered and the solvents were removed in vacuo to afford $0.087 \mathrm{~g}(100 \%)$ of $\mathbf{9}^{\prime}$. $[\alpha]^{21}{ }_{\mathrm{D}}=-145.5\left(c\right.$ 1.78, $\left.\mathrm{CH}_{3} \mathrm{COCH}_{3}\right)$, lit. ${ }^{5}[\alpha]^{25}{ }_{\mathrm{D}}=-152.4\left(c \quad 2, \mathrm{CH}_{3} \mathrm{COCH}_{3}\right) .{ }^{1} \mathrm{H}$ NMR
$\left(\mathrm{CHCl}_{3}, 300 \mathrm{MHz}\right) \delta 2.22(\mathrm{~s}, 3 \mathrm{H}), 5.90(\mathrm{~s}, 1 \mathrm{H}), 7.31-7.50(\mathrm{~m}, 5 \mathrm{H}), 8.80(\mathrm{bs}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CHCl}_{3}, 75 \mathrm{MHz}\right) \delta 20.5,74.1,127.55,128.50,128.74,133.11$, 170.6, 173.4 (Fig. 10).

General procedure for the preparation of Mosher esters.
3-(Trimethylsilyl)-3,4-pentadien-2-yl (R)- $\alpha$-methoxy- $\alpha$-(trifluoromethyl)phenylacetate ( $\mathbf{1 0 a}{ }^{\mathbf{S}}$ ). In a 25 mL round bottom flask dried at $150{ }^{\circ} \mathrm{C}$ and cooled under $\mathrm{N}_{2}, \mathbf{6 a}$ $(0.062 \mathrm{~g}, 0.40 \mathrm{mmol})$ is added with 4-(dimethylamino)pyridine (DMAP) ( $0.10 \mathrm{~g}, 0.80$ mmol ) in 4.0 mL of THF at room temperature. With a constant stirring ( $R$ )-(-)- $\alpha-$ methoxy- $\alpha$-(trifluoromethyl)phenylacetyl chloride (Mosher's acid chloride) ${ }^{6}$ ( 0.126 g , 0.50 mmol ) was added via syringe dropwise at the same temperature for 10 h . Water (ca. 5 mL ) is added and the layers are separated. The aqueous layer is washed with dichloromethane ( $3 \mathrm{X} \mathrm{ca}$.5 mL ) and the organic layers combined and concentrated. Hexane is added and the solution was eluted through an alumina gel chromatographic column to afford $\mathbf{1 0 a}{ }^{\boldsymbol{S}}(0.132 \mathrm{~g}, 0.35 \mathrm{mmol}, 70 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 0.13$ $(\mathrm{s}, 9 \mathrm{H}), 1.5(\mathrm{~d}, 3 \mathrm{H}), 3.61(\mathrm{q}, 3 \mathrm{H}), 4.35(\mathrm{~d}, 1 \mathrm{H}), 4.5(\mathrm{~d}, 1 \mathrm{H}), 5.6(\mathrm{q}, J=6 \mathrm{~Hz}, 1 \mathrm{H}), 7.3-7.6$ $(\mathrm{m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta-0.9,20.9,55.5,72.4,84.2(\mathrm{q}, J=27.4 \mathrm{~Hz}), 97.1$, 126.8 ( $\mathrm{q}, ~ J=288.8 \mathrm{~Hz}$ ), 127.2, 128.4, 130.5, 132.4, 165.8, 209.5. This was determined from the analysis, by ${ }^{1} \mathrm{H}$ NMR of the $\mathrm{OCH}_{3}$ with $\delta 3.61$ and 3.55 in $50: 50$ ratio for $\mathbf{1 0 a}$ from ( $\pm$ )-( $\mathbf{6 a}$ ) and 97:3 from 6a (Fig. 11). This was confirmed by analysis of the ${ }^{13} \mathrm{C}$ NMR spectrum of $\mathbf{1 0 a}$ for $\mathrm{C}-2$ in $\mathrm{CH}_{3} \mathbf{C H}$ (OMosher) $\mathrm{C}(\mathrm{TMS})=\mathrm{C}=\mathrm{CH}_{2}$ with signals at $\delta$ 72.0 and 71.6 ppm in a $50: 50$ ratio and 3:97 for $\mathbf{1 0 a}{ }^{S}$.

3-(Trimethylsilyl)-1,2-heptadien-4-yl (R)- $\alpha$-methoxy- $\alpha$-(trifluoromethyl)phenylacetate ( $\mathbf{1 0 b}^{R}$ ). $80 \%$ yield, ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 0.08(\mathrm{~s}, 9 \mathrm{H}), 0.75(\mathrm{t}$, $3 \mathrm{H}), 1.25(\mathrm{~m}, 2 \mathrm{H}), 1.70(\mathrm{~m}, 2 \mathrm{H}), 3.56(\mathrm{q}, 3 \mathrm{H}), 4.5(\mathrm{~d}, 1 \mathrm{H}), 4.6(\mathrm{~d}, 1 \mathrm{H}), 5.4(\mathrm{dd}, J=5.1$, $8.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.3-7.4(\mathrm{~m}, 3 \mathrm{H}), 7.6-7.7(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta-1.2,13.5$, $18.5,37.2,55.4,71.7,75.7,84.4(\mathrm{q}, J=27.5 \mathrm{~Hz}), 96.2,123.4(\mathrm{q}, J=288.6 \mathrm{~Hz}), 127.5$, $128.9,129.4,165.9,210.0$. The $\%$ ee was determined from the analysis by ${ }^{13} \mathrm{C}$ NMR, methylene carbons at C-6 $\left(\mathrm{CH}_{3} \mathbf{C H}_{2} \mathrm{CH}_{2} \mathrm{CH}(\mathrm{OMosher}) \mathrm{C}(\mathrm{TMS})=\mathrm{C}=\mathrm{CH}_{2}\right)$ with $\delta 18.8$ and 18.5 a peak area ratio of $50: 50$ for $\mathbf{1 0 b}$ from $( \pm)-(\mathbf{6 b})$, and 1:99 from $\mathbf{6 b}$, and 99:1 from $\mathbf{6 b}$ ' (Fig. 12). This was confirmed by analysis of C-1 $\left(\mathrm{PrCH}(\mathrm{OMosher}) \mathrm{C}(\mathrm{TMS})=\mathrm{C}=\mathbf{C H}_{\mathbf{2}}\right.$ ) with $\delta 71.7$ and 71.5 ppm in a peak area ratio of 50:50 for $\mathbf{1 0 b}$ from $( \pm)-(\mathbf{6 b})$, and $99: 1$ for $\mathbf{1 0 b}^{R}$.

2-Methyl-4-(trimethylsilyl)-4,5-hexadien-3-yl (R)- $\alpha$-methoxy- $\alpha$-(trifluoromethyl)phenylacetate ( $10 \mathrm{c}^{\boldsymbol{R}}$ ). $73 \%$ yield, ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 0.15(\mathrm{~s}, 9 \mathrm{H})$, $0.9(\mathrm{dd}, 6 \mathrm{H}), 2.0(\mathrm{~m}, 1 \mathrm{H}), 3.57(\mathrm{q}, 3 \mathrm{H}), 4.2(\mathrm{dd}, 1 \mathrm{H}), 4.4(\mathrm{dd}, 1 \mathrm{H}), 5.1(\mathrm{~d}, 1 \mathrm{H}), 7.4(\mathrm{~m}$, $3 \mathrm{H}), 7.5-7.6(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta-1.1,19.5,32.7,38.7,71.3,75.3$, $84.5(\mathrm{q}, J=27.7 \mathrm{~Hz}), 95.5,124.2(\mathrm{q}, J=288.1 \mathrm{~Hz}), 127.6,128.8,129.5,165.6,210.2$. The $\%$ ee was determined from the analysis by ${ }^{13} \mathrm{C}$ NMR, of C-6 $\left(\mathrm{CH}_{3} \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}(\mathrm{OMosher}) \mathbf{C}(\mathrm{TMS})=\mathrm{C}=\mathbf{C H}_{2}\right)$ with $\delta 71.5$ and 71.3 in a peak ratio of

50:50 for $\mathbf{1 0 c}$ from ( $\pm$ )-( $\mathbf{6 c}$ ), and 1:99 from $\mathbf{6 c}$ (Fig. 13). This was confirmed by analysis of C-4 $\left(\mathrm{CH}_{3} \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}(\mathrm{OMosher}) \mathrm{C}(\mathrm{TMS})=\mathrm{C}=\mathrm{CH}_{2}\right)$ with 95.6 and 95.5 in a peak area ratio of 50:50 for $\mathbf{1 0 c}$ from $( \pm)-(\mathbf{6 c})$, and 1:99 from $\mathbf{6 c}$ for $\mathbf{1 0} \mathbf{c}^{R}$.

2,2-Dimethyl-4-(trimethylsilyl)-4,5-hexadien-3-yl ( $R$ )- $\alpha$-methoxy- $\alpha$-(trifluoromethyl)phenylacetate ( $\mathbf{1 0 d}{ }^{R}$ ). $82 \%$ yield, ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta .0 .13(\mathrm{~s}, 9 \mathrm{H})$, $0.9(\mathrm{~s}, 9 \mathrm{H}), 3.58(\mathrm{q}, 3 \mathrm{H}), 4.46(\mathrm{~d}, 1 \mathrm{H}), 4.55(\mathrm{~d}, 1 \mathrm{H}), 4.95(\mathrm{~s}, 1 \mathrm{H}), 7.4-7.6(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta-0.4,25.9,36.5,55.3,67.8,71.4,81.1,84.1$ (q, $J=27.8 \mathrm{~Hz}$ ), $94.5,123.4(\mathrm{q}, ~ J=288.4 \mathrm{~Hz}), 127.5,128.1,129.4,132.5,165.7,211.0$. The $\%$ ee was determined from the analysis by ${ }^{1} \mathrm{H}$ NMR, of the $\mathrm{C}-3$ methine proton $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CCH}(\mathrm{OMosher}) \mathrm{C}(\mathrm{TMS})=\mathrm{C}=\mathrm{CH}_{2}\right)$ with $\delta 4.95$ and 4.90 in a $50: 50$ peak area ratio for $\mathbf{1 0 d}$ from $( \pm)-(\mathbf{6 d})$ and $99: 1$ from $\mathbf{6 d}$ (Fig. 14). This was confirmed by analysis of ${ }^{13} \mathrm{C}$ NMR, of C-6 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CCH}(\mathrm{OMosher}) \mathrm{C}(\mathrm{TMS})=\mathbf{C}=\mathrm{CH}_{2}\right)$ with $\delta 211.0$ and 210.8 ppm in a 50:50 peak area ratio for $\mathbf{1 0 d}$ from $( \pm)-(\mathbf{6 d})$ and $99: 1$ for $\mathbf{1 0 d}{ }^{R}$.

1-Phenyl-2-(trimethylsilyl)-2,3-butadien-1-yl (R)- $\alpha$-methoxy- $\alpha$-(trifluoromethyl)phenylacetate $\left(10 \mathbf{e}^{\boldsymbol{S}}\right)$. $78 \%$ yield, ${ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 0.01(\mathrm{~s}, 9 \mathrm{H})$, $3.55(\mathrm{bs}, 3 \mathrm{H}), 4.6(\mathrm{~d}, 2 \mathrm{H}), 6.5(\mathrm{bs}, 1 \mathrm{H}), 7.2-7.4(\mathrm{~m}, 6 \mathrm{H}), 7.45-7.55(\mathrm{~m}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta-1.3,55.5,72.8,77.2,84.6(\mathrm{q}, J=27.7 \mathrm{~Hz}), 96.6,123.3(\mathrm{q}, J=288.4$ $\mathrm{Hz}), 127.5,127.5,128.1,128.2,128.5,128.8,130.5,132.9,138.3,165.6,209.5$. The $\%$ ee was determined from the analysis by ${ }^{13} \mathrm{C}$ NMR, of $\mathrm{C}-1$ $\left((\mathrm{Ph}) \mathbf{C H}(\mathrm{OMosher}) \mathrm{C}(\mathrm{TMS})=\mathrm{C}=\mathrm{CH}_{2}\right)$ with $\delta 72.8$ and 72.4 ppm in a peak area ratio of 50:50 for $10 \mathbf{e}$ from $( \pm)-(6 e)$, and 99:1 from $\mathbf{6 e}$ (Fig. 15). This was confirmed by analysis of C-3 $\left(\mathrm{PhCH}(\mathrm{OMosher}) \mathrm{C}(\mathrm{TMS})=\mathbf{C}=\mathrm{CH}_{2}\right)$ with $\delta 209.5$ and 209.0 ppm in a peak area ratio of 50:50 for $\mathbf{1 0 e}$ from $( \pm)-(\mathbf{6 e})$, and 99:1 for $\mathbf{1 0} \mathbf{e}^{S}$.

## (E)-3-(Trimethylsilyl)-1,2,5-heptatrien-4-yl (R)- $\alpha$-methoxy- $\alpha$-(trifluoro-

 methyl)phenylacetate ( $\mathbf{1 0 f}{ }^{S}$ ). $75 \%$ yield, ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 0.2(\mathrm{~s}, 9 \mathrm{H})$, $1.81(\mathrm{dd}, J=6.4,1.5 \mathrm{~Hz}, 3 \mathrm{H}), 3.6(\mathrm{bs}, 3 \mathrm{H}), 4.7(\mathrm{~d}, 2 \mathrm{H}), 5.6(\mathrm{ddq}, J=15.2,7.4,1.4 \mathrm{~Hz}$, $1 \mathrm{H}), 5.8(\mathrm{dq}, J=15.3,6.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.9,(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.2-7.4(\mathrm{~m}, 3 \mathrm{H}), 7.5-7.6(\mathrm{~m}$, $2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta-1.1,20.0,55.4,69.8,74.1,85.5$, (q, $J=28.2 \mathrm{~Hz}$ ), $95.6,123.3$ (q, $J=288.6 \mathrm{~Hz}$ ), 127.3, 128.5, 128.6, 129.5, 130.6, 130.8, 165.7, 213.2. The $\%$ ee was determined from the analysis by ${ }^{13} \mathrm{C}$ NMR, of the ester $\mathrm{C}=\mathrm{O}$ signal with $\delta$ 165.69 and 165.61 ppm in a peak area ratio of $50: 50$ for $\mathbf{1 0 f}$ from $( \pm)-(\mathbf{6 f})$, and $<1.5: 98.5$ from $6 \mathbf{f}$ (Fig. 16). This was confirmed by analysis of $\mathrm{C}-5$ (MeHC $=\mathbf{C H C H}(O M o s h e r)-$ $\mathrm{C}(\mathrm{TMS})=\mathrm{C}=\mathrm{CH}_{2}$ ) with $\delta 130.8$ and 130.5 ppm in a peak area ratio of $50: 50$ for $\mathbf{1 0 f}$ from $( \pm)-(\mathbf{6 f})$, and $99: 1$ for $\mathbf{1 0 f} \mathbf{f}^{S}$. Unfortunately, this clearer spectral comparison was lost in a hurricane-related laboratory fire (September 18, 2004).
$(-)-3 R$


Figure 1. ${ }^{\mathbf{1}} \mathbf{H}$ and ${ }^{13} \mathrm{C}$ NMR Spectra of $\boldsymbol{B}$-Methoxy-10-trimethylsilyl-9-borabicyclo[3.3.2]decane (3)

$(+)-2 R$



Figure 2. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR Spectra of $(+)-(10 R)-B-(1 S, 2 S)-$ Pseudoephedrinyl-10-trimethylsilyl-9-borabicyclo[3.3.2]decane ((+)-2R)


Figure 3. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR Spectra of $(-)-(10 R)-B-[\gamma-$
(Trimethylsilyl)propargyl]-10-TMS-9-borabicyclo[3.3.2]decane ((-)-1R)






Figure 4. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR Spectra of (-)-(2S)-3-(Trimethylsilyl)-3,4-pentadien-2-ol (6a).



Figure 5. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR Spectra of (-)-(4R)-3-(Trimethylsilyl)-1,2-
heptadien-4-ol (6b).


Figure 6. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR Spectra of (-)-(3R)-2-Methyl-4-(trimethylsilyl)-4,5-hexadien-3-ol (6c).


Figure 7. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR Spectra of $(+)-(3 R)$-2,2-Dimethyl-4-(trimethylsilyl)-4,5-hexadien-3-ol (6d).


Figure 8. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR Spectra of (-)-(1S)-1-Phenyl-2-(trimethylsilyl)-2,3-butadien-1-ol (6e).







Figure 9. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR Spectra of $(+)-(4 S)-(E)$-3-(Trimethylsilyl)-1,2,5-heptatrien-4-ol (6f).






Figure 10. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR Spectra of $S$-(+)-Mandelic acid (9) ( $\left.\mathrm{CD}_{3} \mathrm{OD}\right)$ (above). ${ }^{1} H$ NMR Spectrum of $\boldsymbol{R}$-(-)- $O$-Acetylmandelic acid ( ${ }^{\prime}$ ) ( $\mathrm{CDCl}_{3}$ ) (below).


Figure 11. ${ }^{1} \mathrm{H}$ NMR of OMe Region for 3-(Trimethylsilyl)-3,4-pentadien-2yl (R)- $\alpha$-methoxy- $\alpha$-(trifluoromethyl)phenylacetate ( $10 \mathrm{a}^{S}$ ).







Figure 12. ${ }^{13} \mathrm{C}$ NMR of $\mathrm{C}-1$ Region for 3-(Trimethylsilyl)-1,2-heptadien-4-yl (R)- $\alpha$ -methoxy- $\alpha$-(trifluoromethyl)phenylacetate ( $\mathbf{1 0 b}{ }^{R}$ ).



Figure 13. ${ }^{13}$ C NMR of C-6 Region for 2-Methyl-4-(trimethylsilyl)-4,5-hexadien-3-yl ( $R$ )- $\alpha$-methoxy- $\alpha$-(trifluoro-methyl)phenylacetate ( $10 \mathrm{c}^{R}$ ).


Figure 14. ${ }^{1} \mathbf{H}$ NMR of C-3 Methine Region for 2,2-Dimethyl-4-(trimethylsilyl)-4,5-hexadien-3-yl (R)- $\alpha$-methoxy- $\alpha$-(trifluoromethyl)phenylacetate (10d ${ }^{R}$ ).


Figure 15. ${ }^{13}$ C NMR of C-1 Region for 1-Phenyl-2-(trimethylsilyl)-2,3-butadien-1-yl (R)- $\alpha$-methoxy- $\alpha$-(trifluoro-methyl)phenylacetate (10e ${ }^{S}$ ).


Figure 16. ${ }^{13} \mathrm{C}$ NMR of $\mathrm{C}=\mathrm{O}$ Region for ( E )-3-(Trimethylsilyl)-1,2,5-heptatrien-4-yl (R)- $\alpha$-methoxy- $\alpha$-(trifluoromethyl)phenylacetate $\left(10 f^{S}\right)$.



Figure 17. (Above): ${ }^{13} \mathrm{C}$ NMR of TMSC $=\mathrm{O}$ (left) and TMS (right) for 7. (Below): ${ }^{13} \mathrm{C}$ NMR of TMSOC=O (left) and TMS (right) for 8.

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