

SUPPORTING INFORMATION

A windmill-shaped hexacopper(II) molecule built up by template self-assembly of diaqua-tetra(μ_2 -N3,N9-adeninato)dicopper(II) and aqua(oxydiacetato)copper(II).

Josefa María González-Pérez^{*,†}, Carolina Alarcón-Payer[†], Alfonso Castiñeiras[‡], Tiziana Pivetta[§], Luis Lezama^{§§}, Duane Choquesillo-Lazarte[‡], Guido Crisponi[§] and Juan Nicolás-Gutiérrez^{*,†}.

Department of Inorganic Chemistry, Faculty of Pharmacy, University of Granada, E-18071 Granada, Spain. Department of Inorganic Chemistry, Faculty of Pharmacy, University of Santiago de Compostela, E-15706 Santiago de Compostela, Spain. Department of Chemical Sciences, University of Cagliari, I-09042 Monserrato-Cagliari, Italy and Department of Inorganic Chemistry, University of Basque country, E-4808 Bilbao, Spain.

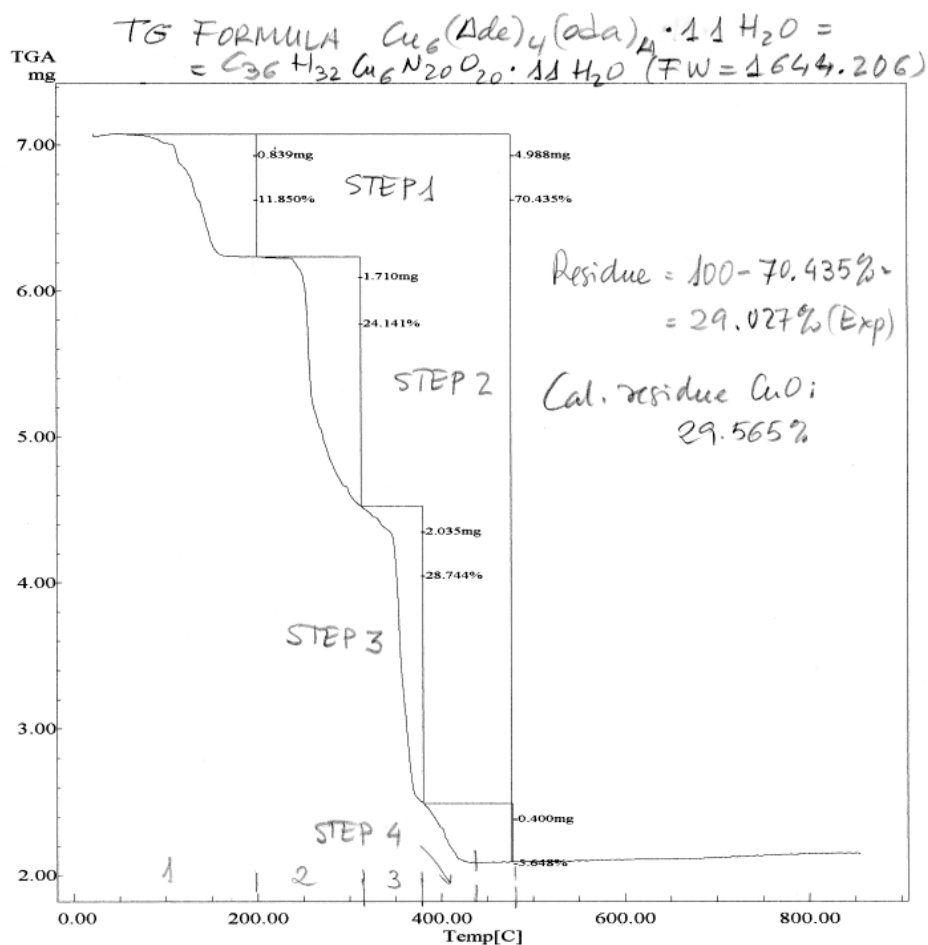
*To whom correspondence should be addressed: E-mail: jmgp@ugr.es (J.M.G.-P.), jniclos@ugr.es. Fax: +34-958246219

S1A. Thermogravimetric analysis, with FT-IR spectral identification of evolved gasses.

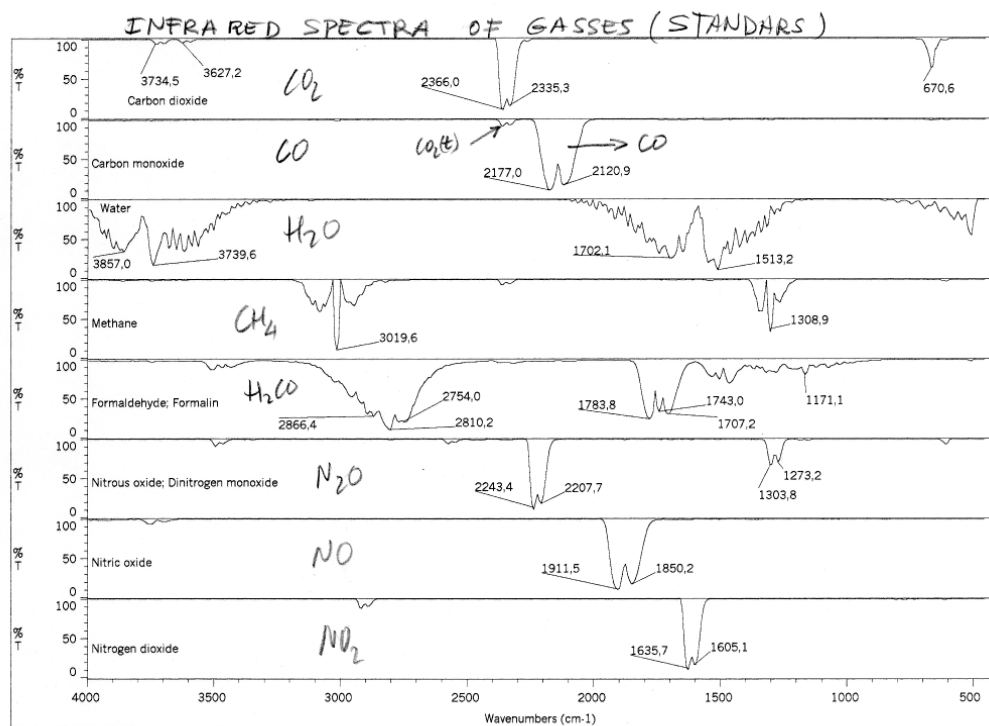
Unidad de TG, DSC, NIR.
Centro de Instrumentación Científica.
Universidad de Granada.

File Name: T2100156.D20 C-286
Detector Type: Shimadzu TGA-50H
Acquisition Date: 10/05/04
Acquisition Time: 09:51:44
Sample Name: C-286
Weight: 7.081[mg]
Cell: Alumina
Atmosphere: Air
Rate Flow: 100.0[ml/min]
Operator: MASP

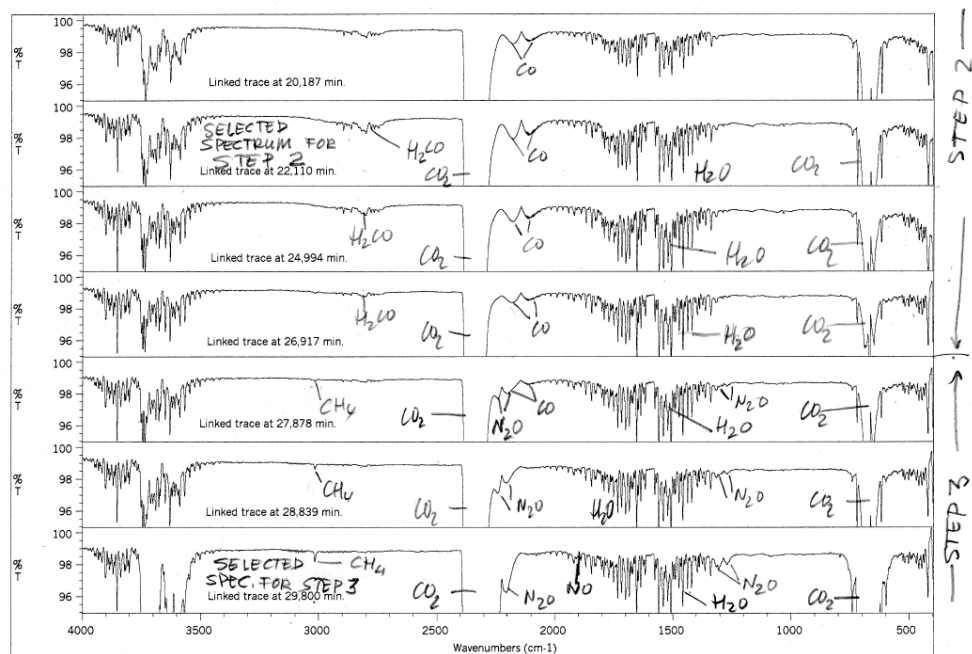
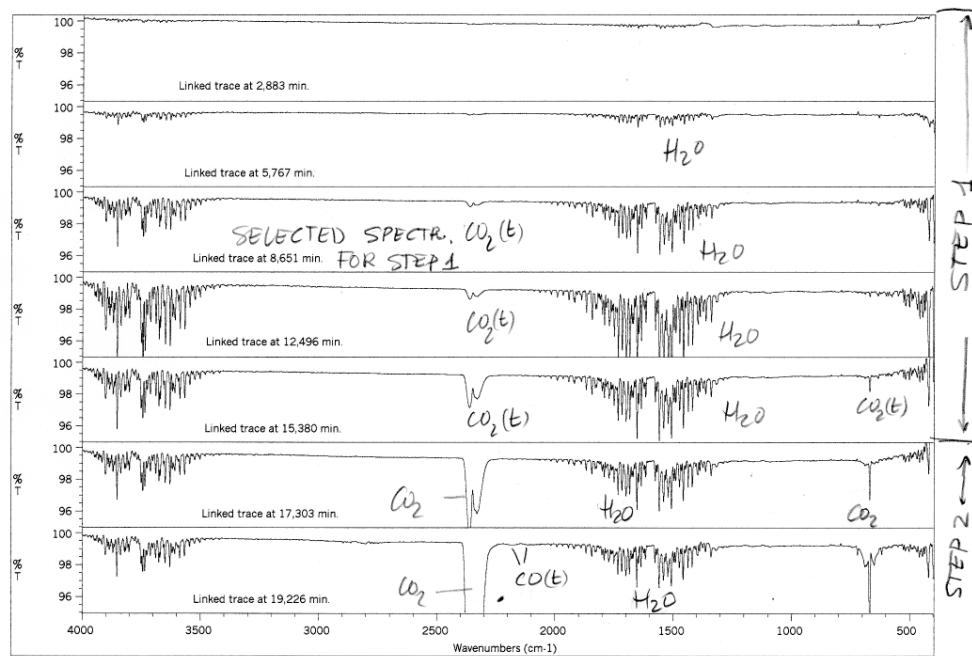
Temp Program			
Rate	Hold Temp	Hold Time	
[C/min]	[C]	[min]	
10.0	950.0	0.0	



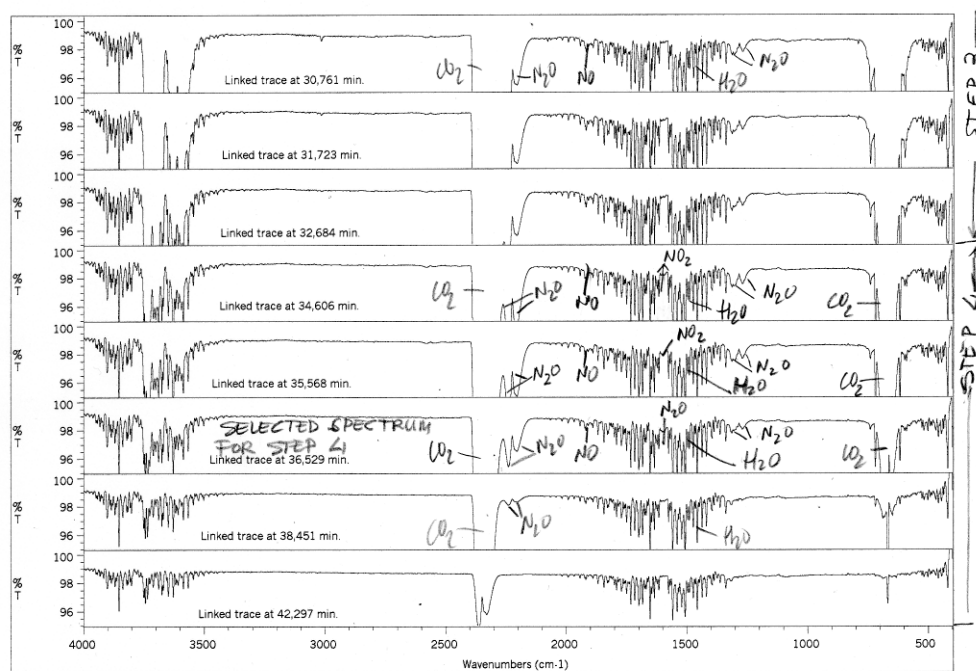
S1B. FT-IR spectra of the gasses (standards). They are used for the identification of gasses in the 22 IR spectra recorded during the experience (see S1C).



S1C. FT-IR spectra recorded during the TG analysis of compound 1-o. On the right-hand is indicated the 5 or 6 spectra corresponding to each step. Four of these spectra (one per step) were selected as is shown in S1E.



S1C (cont). FT-IR spectra recorded during the TG analysis of compound 1-o. On the right-hand is indicated the 5 or 6 spectra corresponding to each step. Four of these spectra (one per step) were selected as is shown in S1E.



S1D. Summary of the Thermogravimetric experience for compound **1-o**.

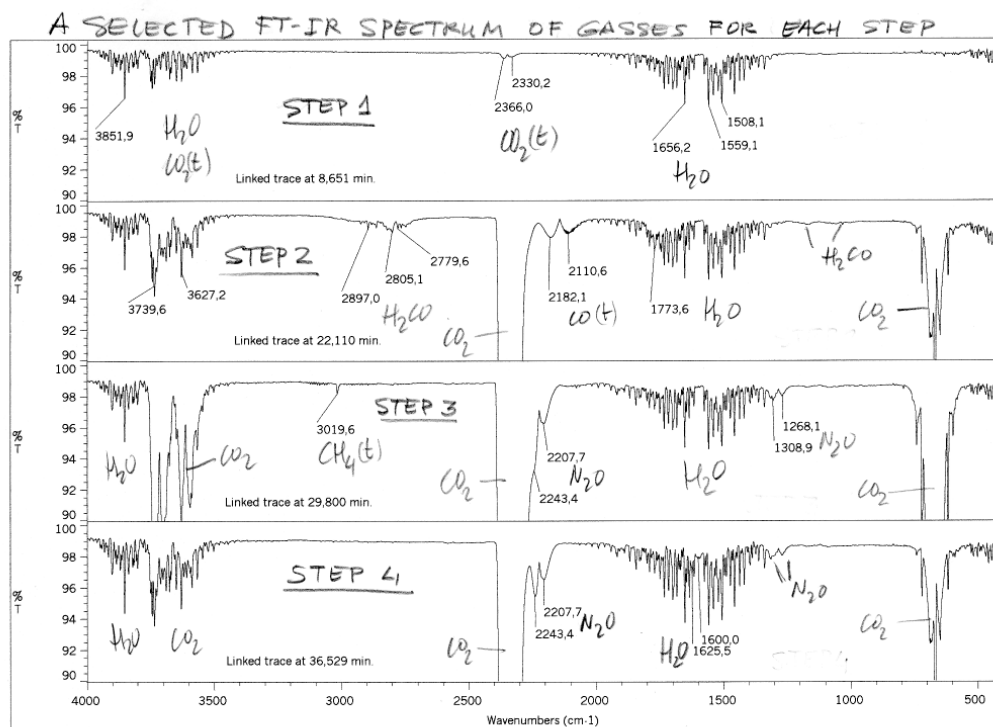
Step	temp (°C)	time (min)	IR N°	Weight loss %		Gasses or residue
				Exp.	Calcd*.	
1	r.t. – 200	2 - 15	5	11.850	12.053	(11)H ₂ O, CO ₂ (t) ***
2	200 - 315	15 - 27	6	24.141	-	H ₂ CO, CO ***
3	315 - 380	27 - 33	6	28.744	-	CO, N ₂ O, NO ***
4	380 - 435	33 - 44	5	5.648	-	N ₂ O, NO, NO ₂ (t) ***
Residue	480	-	-	29.565	29.027	CuO

* Calculated values for C₃₆H₃₂Cu₆N₂₀O₂₀·11H₂O (FW 1644.206)

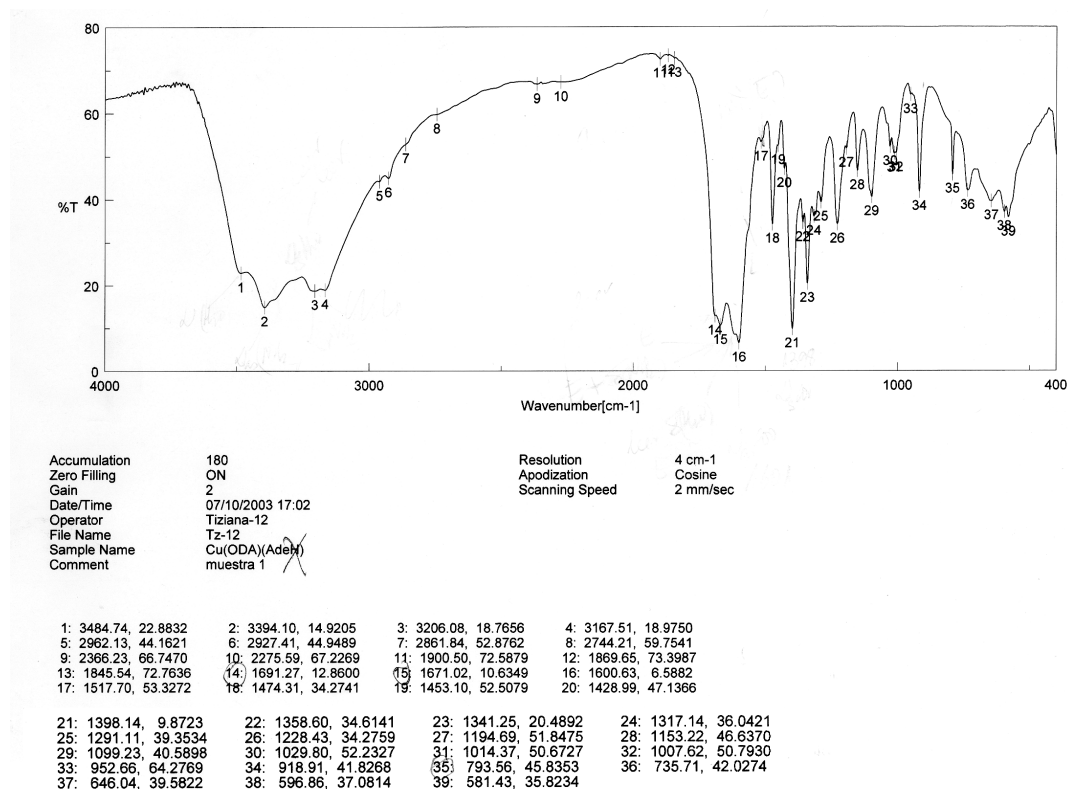
** t = trace amounts

*** In addition to H₂O and CO₂

S1E. Selected FT-IR spectra of gasses for each step.



S2. FT-IR spectrum of compound **1-o** in solid state (KBr disc).



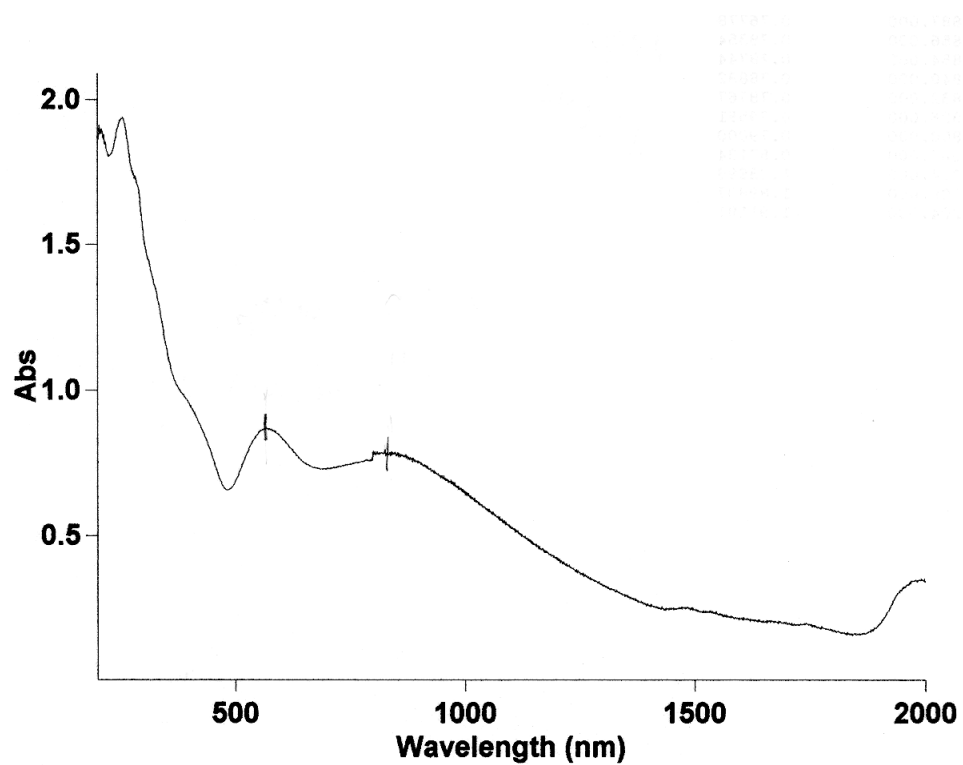
Wavenumbers in cm^{-1} (the number of the band is indicated in brackets).

H_2O ν_{as} 3485 [1] $\nu_{\text{s}} \sim 3280$ δ 1625

Ade $\nu_{\text{as}}(\text{NH}_2)$ 3390 [2], 3350 $\nu_{\text{s}}(\text{NH}_2)$ 3206 [3], 3168 [4] $\delta(\text{NH}_2)$ 1671 [15]
 $\pi(\text{C-H})$ 794 [35]

oda $\nu_{\text{as}}(\text{COO})$ 1601 [16] $\nu_{\text{s}}(\text{COO})$ 1398 [21]

S3. Electronic (reflectance) spectrum of compound **1-o**.



Maxima	840 nm	11900 cm ⁻¹
	567 nm	17640 cm ⁻¹

S4. X-band ESR spectrum of compound **1-o** at room temperature. Apparent $g_{iso} = 2.137$.

