Supporting Information

Pd/C-Catalyzed Deoxygenation of Phenol Derivatives Using Mg Metal in MeOH in the Presence of NH₄OAc

Hironao Sajiki,* Akinori Mori, Tomoteru Mizusaki, Takashi Ikawa, Tomohiro Maegawa, Kosaku Hirota

Laboratory of Medicinal Chemistry, Gifu Pharmaceutical University, Gifu 502-8585, Japan

Table of Contents:

Title and table of contents	S1
Experimental procedure and compound data	S2-S9
References	S10
NMR and Mass spectroscopy charts for compounds	S11-S31

General

10% Pd/C was purchased from Aldrich, N.E. ChemCat, or Merck.

General procedure for reductive cleavage of 3,4,5-trimethoxyphenyl triflate(1) using various metals and solvents (Table 1). After two vacuum/Ar cycles to remove air from the reaction tube, the mixture of 1 (158.2 mg, 0.5 mmol), 10% Pd/C (15.9 mg, 10 wt % of 1) and the metal (0.6 mmol) in solvent (2.0 mL) was stirred at ordinary pressure (balloon) and temperature (*ca.* 20°C) for 24 hours. The reaction mixture was filtrated using a membrane filter (Millipore, Millex[®]-LH, 0.45 µm) and the filtrate was partitioned between ether (10 mL) and water (10 mL). The aqueous layer was extracted with ether (10 mL × 3), and then combined organic layers were washed with brine (10 mL), dried with anhydrous MgSO₄, filtered and concentrated under reduced pressure. The ratio of the substrate (1) and the product (2) was confirmed by ¹H NMR of the crude mixture in CDCl₃.

General procedure for reductive cleavage of 3,4,5-trimethoxyphenyl triflate(1) in the presence of various additives (Table 2) After two vacuum/Ar cycles to remove air from the reaction tube, the mixture of 1 (158.2 mg, 0.5 mmol), 10% Pd/C (15.9 mg, 10 wt % of 1), magnesium metal (14.6 mg, 0.6 mmol) and additive (0.5 mmol) in MeOH (2.0 mL) was stirred at ordinary pressure (balloon) and temperature (*ca.* 20°C) for appropriate time (see Table 3). The reaction mixture was filtrated using a membrane filter (Millipore, Millex[®]-LH, 0.45 µm) and the filtrate was partitioned between ether (10 mL) and water (10 mL). The aqueous layer was extracted with ether (10 mL × 3), and then combined organic layers were washed with brine (10 mL), dried with anhydrous MgSO₄, filtered and concentrated under reduced pressure. The ratio of the substrate (1) and the product (2) was confirmed by ¹H NMR of the crude mixture in CDCl₃.

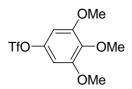
General procedure for reductive cleavage of various aryl triflates (Table 3). After two vacuum/Ar cycles to remove air from the reaction tube, the mixture of the aryl triflate (0.5 mmol), 10% Pd/C (10 wt % of the aryl triflate), magnesium metal (14.6 mg, 0.6 mmol) and ammonium acetate (38.6 mg, 0.5 mmol) in MeOH (1 mL) was stirred at ordinary pressure (balloon) and at temperature (*ca.* 20°C) for the appropriate time (see Table 3). The reaction mixture was filtrated using a membrane filter (Millipore, Millex[®]-LH, 0.45 µm) and the filtrate was partitioned between ether (10 mL) and water (10 mL). The aqueous layer was extracted with ether (10 mL × 3), and then combined organic layers were washed with brine (10 mL), dried with anhydrous MgSO₄, filtered and concentrated under reduced pressure.

General procedure for reductive cleavage of aryl mesylates or tosylate (Table 4). After two vacuum/Ar cycles to remove air from the reaction tube, the mixture of the aryl mesylate or tosylate (0.25 mmol), 10% Pd/C (10 wt % of the aryl mesylate or tosylate), magnesium metal (7.3 mg, 0.3 mmol) and ammonium acetate (0.25-7.5 mmol) in MeOH (1 mL) was stirred at ordinary pressure (balloon) and at temperature (*ca.* 20°C) for the appropriate time (see Table 4). The reaction mixture was filtrated using a membrane filter (Millipore, Millex[®]-LH, 0.45 µm) and the filtrate was partitioned between ether (10 mL) and water (10 mL). The aqueous layer was extracted with ether (10 mL × 3), and then combined organic layers were washed with brine (10 mL), dried with anhydrous MgSO₄, filtered and concentrated under reduced pressure.

General procedure for Table 5. After two vacuum/Ar cycles to remove air from the reaction tube, the mixture of **1** (79.1 mg, 0.25 mmol), 10% Pd/C (8.0 mg, 10 wt % of **1**) and the magnesium metal (7.3 mg, 0.3 mmol) in deuterated methanol (1.0 mL) was stirred at ordinary pressure (balloon) and temperature (*ca.* 20°C) for 24 hours. The reaction mixture was filtrated using a membrane filter (Millipore, Millex[®]-LH, 0.45 μ m) and the filtrate was partitioned between ether (10 mL) and water (10 mL). The aqueous layer was extracted with ether (10 mL × 3), and then combined organic layers were washed with brine (10 mL), dried with anhydrous MgSO₄, filtered and concentrated under reduced pressure. The hydrogen source of the reaction was confirmed by ¹H NMR of the crude mixture in CDCl₃.

Preparation of 3,4,5- trimethoxyphenyl triflate¹

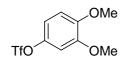
3,4,5-Trimethoxyphenol (1) (921 mg, 5.0 mmol) was dissolved in toluene(10 mL) and 30% K₃PO₄ solution (10 mL) was added. The reaction mixture was cooled to 0 °C and triflic anhydride (1.0 mL, 6.0 mmol) was added slowly dropwise with stirring to maintain a reaction temperature of <10 °C. The reaction mixture was allowed to warm to room temperature and stirred for a further 30 minutes, then extracted with ethylacetate (10 mL × 2). The combined organic layers were washed with water (30 mL), dried over MgSO₄ and evaporated to give 1 (1.54 g, 97%) as a white solid.



3,4,5-trimethoxyphenyl triflate² (Table 1, 2, 5 and Table 3 entry 2)



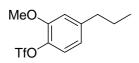
1,2,3-trimethoxybenzene (Table 1, 2, 5 and Table 3 entry 2, commercially available)



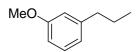
3,4-dimethoxyphenyl triflate (Table 3 entry 1) Colorless oil. ¹H NMR (CDCl₃): δ 6.85 (d, *J*=2.4 Hz, 2H), 6.78 (d, *J*=2.2 Hz, 1H), 3.90 (s, 3H), 3.89 (s, 3H). ¹³C NMR (CDCl₃): δ 149.8, 148.9, 142.9, 120.4, 117.2, 112.9, 111.1, 105.4, 56.2. MS (EI) *m/z* 286 (M⁺, 28%), 153 (100), 125 (30). HRMS (EI) Calcd for C₉H₉O₅F₃S (M⁺) 286.0123. Found 286.0116.



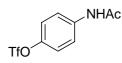
1,2-dimethoxybenzene (Table 3 entry 1, commercially available)



2-methoxy-4-propylphenyl triflate³ (Table 3 entry 3)



1-methoxy-3-propylbenzene³ (Table 3 entry 3)

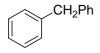


4-acetamidephenyl trflate⁴ (Table 3 entry 4)



N-phenylacetamide (Table 3 entry 4, commercially available)

4-benzylphenyl triflate⁵ (Table 3 entry 5)



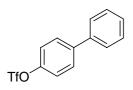
diphenylmethane (Table 3 entry 5, 6, Table 3 entries 7, 10, commercially available)



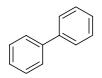
2-benzylphenyl triflate⁶ (Table 3 entry 6)

4-[(methoxycarbonyl) methyl]phenyl triflate⁸ (Table 3 entry 7)

methyl 2-phenylacetate (Table 3 entry 7, Table 3 entries 3,4,5 commercially available)



4-phenylphenyl triflate (Table 3 entry 8) White solid. ¹H NMR (CDCl₃): δ 7.64 (d, *J*=8.7 Hz, 2H), 7.55 (d, *J*=7.2 Hz, 2H), 7.46 (t, *J*=7.2 Hz, 2H), 7.39 (t, *J*=7.2 Hz, 1H), 7.34 (d, *J*=8.7 Hz, 2H). ¹³C NMR (CDCl₃): δ 148.9, 141.7, 139.3, 129.0, 128.9, 128.1, 127.2, 121.6. MS (EI) *m/z* 302 (M⁺, 40%), 169 (100), 141 (46), 115 (28). HRMS (EI) Calcd for C₁₃H₉O₃F₃S (M⁺) 302.0225. Found 302.0213. Anal. Calcd for C₁₃H₉O₃F₃S: C, 51.66; H, 3.00. Found C, 51.44; H, 3.15.



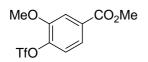
biphenyl (Table 3 entry 8, Table 3 entry 6, commercially available)



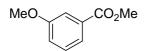
naphthalene 5-yl triflate (Table 3 entry 9, commercially available)



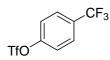
naphthalene (Table 3 entry 9, Table 3 entry 11, commercially available)



4-(methoxycarbonyl)-2-methoxyphenyl triflate⁹ (Table 3 entry 10)



methyl 3-methoxybenzene (Table 3 entry 10, commercially available)



4-(trifluoromethyl)phenyl triflate¹⁰ (Table 3 entry 11)



trifluoromethylbenzene (Table 3 entry 11, commercially available)

4-fluorophenyl triflate (Table 3 entry12) Colorless oil. ¹H NMR (CDCl₃): δ 7.24 (td, *J*=5.0, 2.6 Hz, 2H), 7.12 (td, *J*=7.2, 2.1 Hz, 2H). ¹³C NMR (CDCl₃): δ 123.2, 123.1, 123.1, 117.3, 117.2, 117.0, 117.0. MS (EI) *m*/*z* 244 (M⁺, 45%), 111 (100), 83 (78). HRMS (EI) Calcd for C₇H₄O₃F₄S (M⁺) 243.9817. Found 243.9812.



flurobenzene (Table 3 entry12, commercially available)

Procedure for the preparation of 4-[(methoxycarbonyl)methyl]phenyl tosylate. The mixture of methyl 2-(4-hydroxyphenyl) acetate (3.32 g, 20 mmol), *p*-toluenesulfonyl chloride (4.19 g, 22 mmol) and triethylamine (3.07 mL, 22mmol) in dry dichloromethane (50 mL) was stirred at room temperature for 10 hours. After evaporation of dichloromethane, the residue was partitioned between ethyl acetate (30 mL) and water (30 mL). The organic layer was washed with water (30 mL), sat. NaHCO₃ soln (30 mL), water (30 mL), dil. NaHSO₄ soln (30 mL), water (30 mL) and brine (30 mL) and dried over MgSO₄.After filtration, the solvent was removed *in vacuo* and the oilyresidue was chromatographed on silica gel (ethyl acetate : hexane = 1 : $10 \sim 1 : 5$) to give the pure product (5.77 g, 90%) as a clear oil.

4-[(methoxycarbonyl)methyl] phenyl tosylate (Table 3 entries 1,2) Yellow oil. ¹H NMR (CDCl₃): δ 7.71 (d, *J*=8.3 Hz, 2H), 7.31 (d, *J*=8.0 Hz, 2H), 7.20 (d, *J*=8.8, 2H), 6.94 (d, *J*=8.5, 2H), 3.69 (s, 3H), 3.58 (s, 2H), 2.45 (s, 3H). ¹³C NMR (CDCl₃): δ 171.5, 148.7, 145.3, 132.9, 130.5, 129.7, 128.5, 122.4, 52.1, 40.4, 21.7. MS (EI) *m/z* 320 (M⁺, 90%), 155 (100), 91 (83). HRMS (EI) Calcd for C₁₆H₁₁₆O₅S (M⁺) 320.0718. Found 320.0724.

Typical procedure for the preparation of 4-[(methoxycarbonyl)methyl]phenyl mesylate.

Methyl 2-(4-hydroxyphenyl) acetate (1.66 g, 10 mmol) and triethylamine (1.67 mL, 12 mmol) were dissolved in dry dichloromethane (20 mL). The reaction mixture cooled to 0 °C and methanesulfonyl chloride (0.93 mL, 12 mmol) was added slowly dropwise with stirring to maintain a reaction temperature of <10 °C. The reaction mixture was allowed to warm to room temperature and stirred for a further 1 hour, then the reaction mixture was concentrated *in vacuo*. The residue was partitioned between ethyl acetate (20 mL) and water (20 mL). The organic layer was washed with aqueous saturated sodium hydrogen carbonate (10 mL), 10% aqueous sodium hydrogen sulfate (10 mL) and brine (20 mL), dried with anhydrous MgSO₄, filtered and concentrated under reduced pressure to give the corresponding mesylate (2.29 g, 94%) as a white solid.

CH₂CO₂Me

4-[(methoxycarbonyl)methyl]phenyl mesylate (Table 3 entries 3, 4, 5) White solid. mp 73-74°C. ¹H NMR (CDCl₃): δ 7.32 (d, *J*=8.6 Hz, 2H), 7.23 (d, *J*=8.6 Hz, 2H), 3.69 (s, 3H), 3.62 (s, 2H), 3.12 (s, 3H). ¹³C NMR (CDCl₃): δ 171.4, 148.33, 133.4, 131.0, 122.1, 52.2, 40.4, 37.3. MS (EI) *m/z* 244 (M⁺, 85%), 185 (75), 166 (42), 107 (100), 78 (34), 44 (30). HRMS (EI) Calcd for C₁₀H₁₂O₅S (M⁺) 244.0405. Found 244.0396. Anal. Calcd for C₁₀H₁₂O₅S: C, 49.17; H, 4.95. Found C, 49.08; H, 4.99.

4-phenylphenyl mesylate (Table 3 entry 6, commercially available)

2-benzylphenyl mesylate (Table 3 entry 7) White solid. mp 56-58°C. ¹H NMR (CDCl₃): δ 7.39 (d, *J*=7.8 Hz, 1H), 7.31-7.18 (m, 8H), 4.08 (s, 2H), 2.97 (s, 3H). ¹³C NMR (CDCl₃): δ 147.6, 139.5, 133.8, 131.6, 128.9, 128.5, 127.8, 127.2, 126.3, 121.7, 37.7, 36.2. MS (EI) *m*/*z* 262 (M⁺, 52%), 183 (100), 181 (85), 165 (50). HRMS (EI) Calcd for C₁₄H₁₄O₃S (M⁺) 262.0664. Found 262.0670. Anal. Calcd for C₁₄H₁₄O₃S: C, 64.10; H, 5.38. Found C, 64.00; H, 5.49.

4-(methoxycarbonyl) phenyl mesylate¹¹(Table 3 entry 8)

methylbenzoate (Table 3 entries 8, 9, commercially available)

2-(methoxycarbonyl) phenyl mesylate¹¹ (Table 3 entry 9)

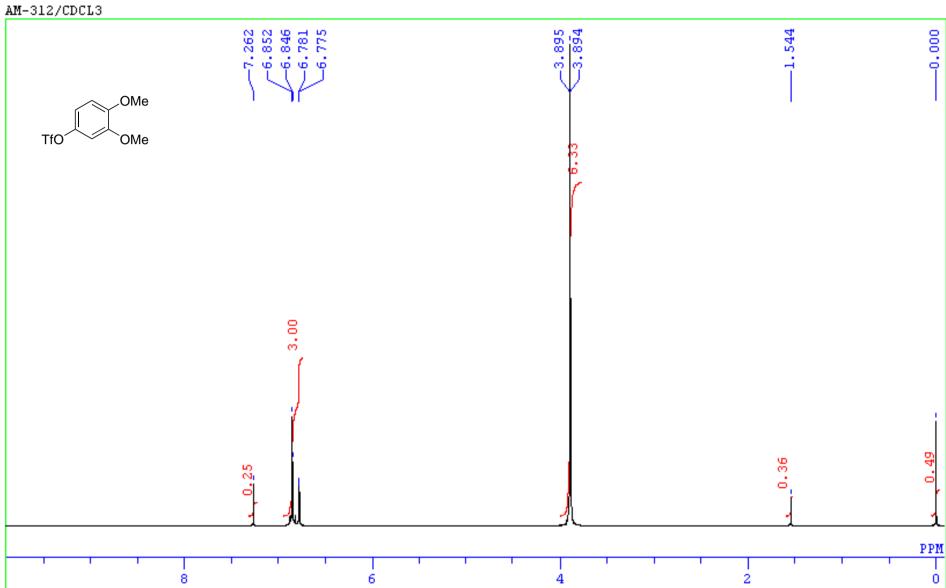
4-benzylphenyl mesylate (Table 3 entry 10) White solid ¹H NMR (CDCl₃): δ 7.32-7.16 (m, 9H), 3.98 (s, 2H), 3.11 (s, 3H). ¹³C NMR (CDCl₃): δ 147.6, 140.6, 140.2, 130.3, 128.9, 128.6, 126.4, 121.9, 41.3, 37.3. MS (EI) *m/z* 262 (M⁺, 100%), 183 (80), 155 (67). HRMS (EI) Calcd for C₁₄H₁₄O₃S (M⁺) 262.0664. Found 262.0660. Anal. Calcd for C₁₄H₁₄O₃S: C, 64.10; H, 5.38. Found C, 64.10; H, 5.40.



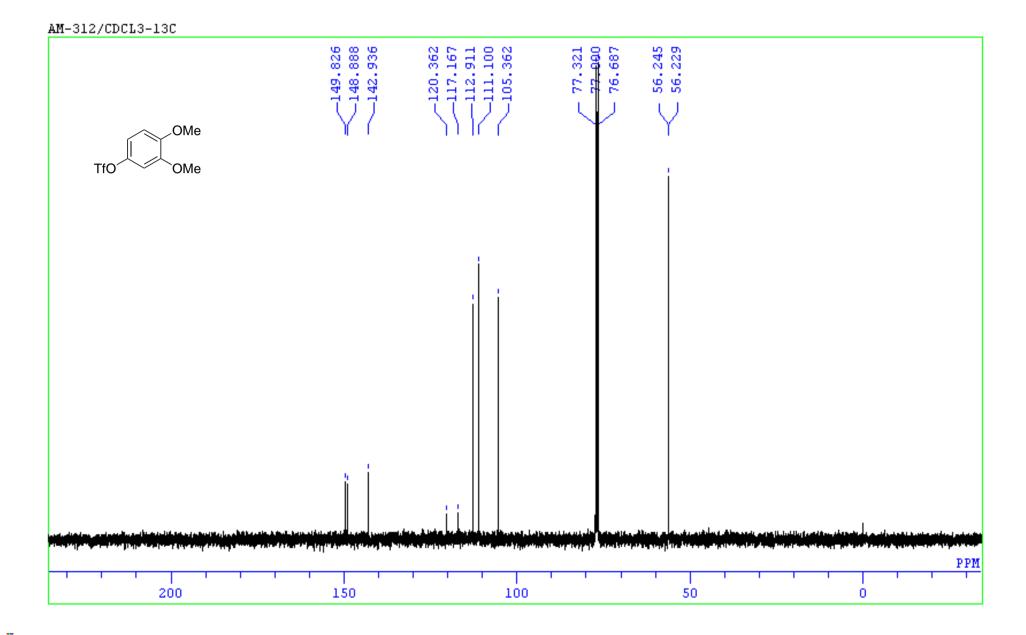
naphthalene 5-yl mesylate (Table 3 entry 11, commercially available)

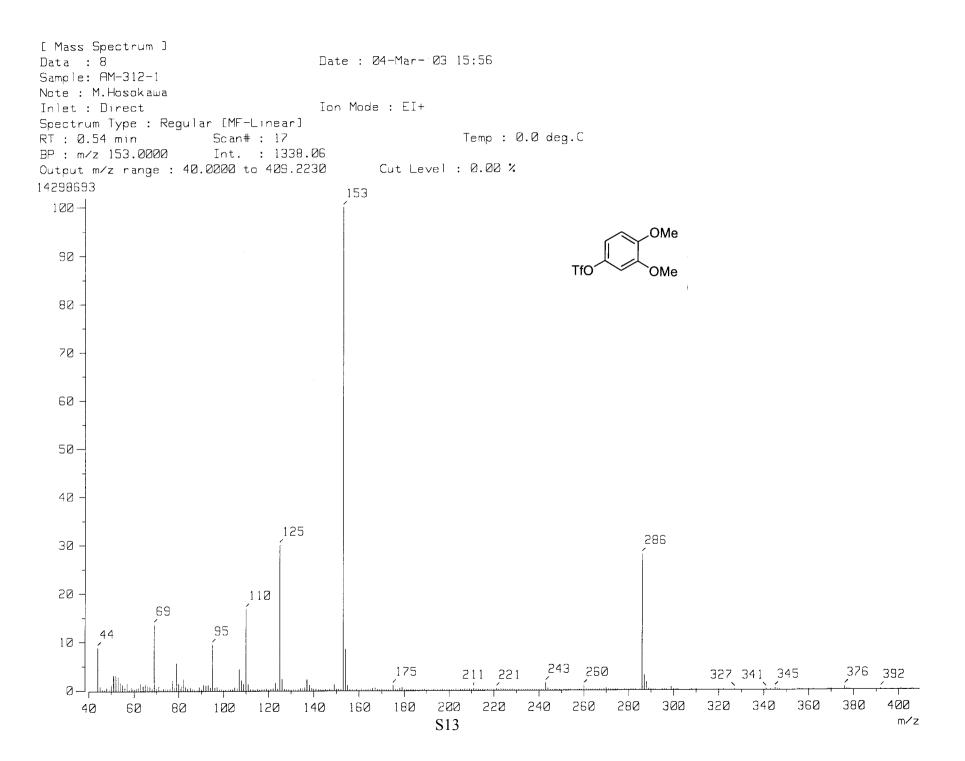
References

- (1) Frantz, D. E.; Weaver, D. G.; Carey, J. P.; Kress, M. H.; Dolling, U. H. Org. Lett. **2002**, *4*, 4717.
- (2) Macmillan, D.; Anderson, D. W. Org. Lett. 2004, 6, 4659.
- (3) Hu, T. Q.; Cairns, G. R.; James, B. R. Holzforschung 2000, 54, 127.
- (4) Niederpruem, H.; Voss, P.; Beyl, V. Liebigs Ann. Chem. 1973, 20.
- (5) Molander, G. A.; Ito, T. Org. Lett. 2001, 3, 393.
- (6) Wang, J. –Q.; Harvey, R. G. Tetrahedron 2002, 58, 5927.
- (7) Ichimura, M.; Ueno, K.; Shibanuma, T.; Takada, K. *Jpn. Kokai Tokkyo Koho* **2002**, JP 2002-322100.
- (8) Graffner-Nordeberg, M.; Sjodin, K.; Tunek, A.; Hallberg, A. Chem. Pharm. Bull. 1998, 46, 591.
- (9) Eicher, T.; Fey, S.; Puhl, W.; Buchel, E.; Speicher, A. Eur. J.Org. Chem. 1998, 877.
- (10) Qing, F. –L.; Fan, J.; Sun, H. –B.; Yue, X. –J. J. Chem. Soc., Perkin Trans. 1, 1997, 3053.
- (11) Percec, V.; Bae, J. -Y.; Zhao, M.; Hill, D. H. J. Org. Chem. 1995, 60, 176.



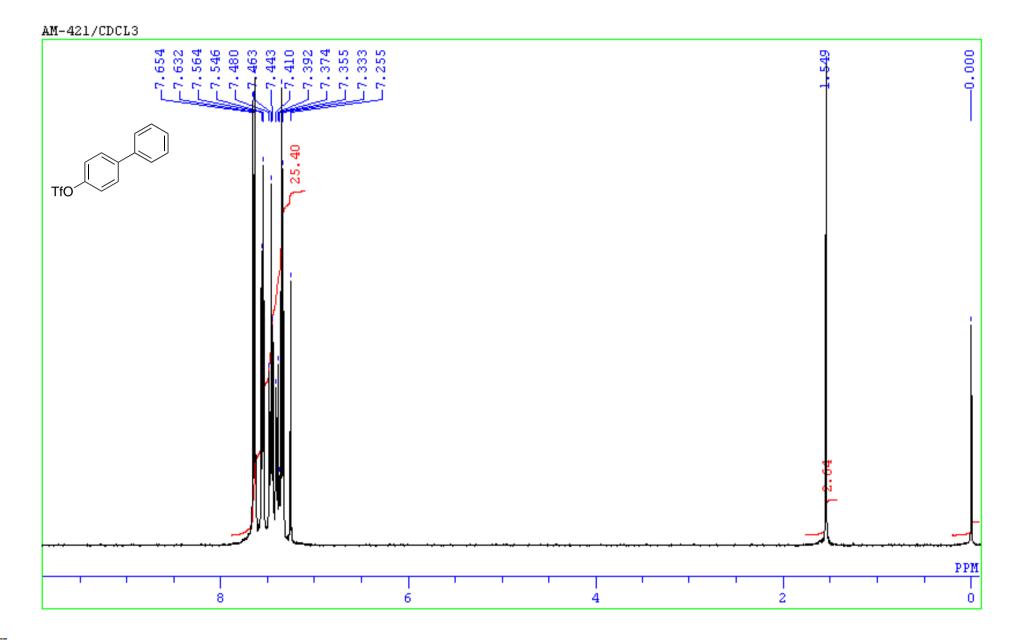
-

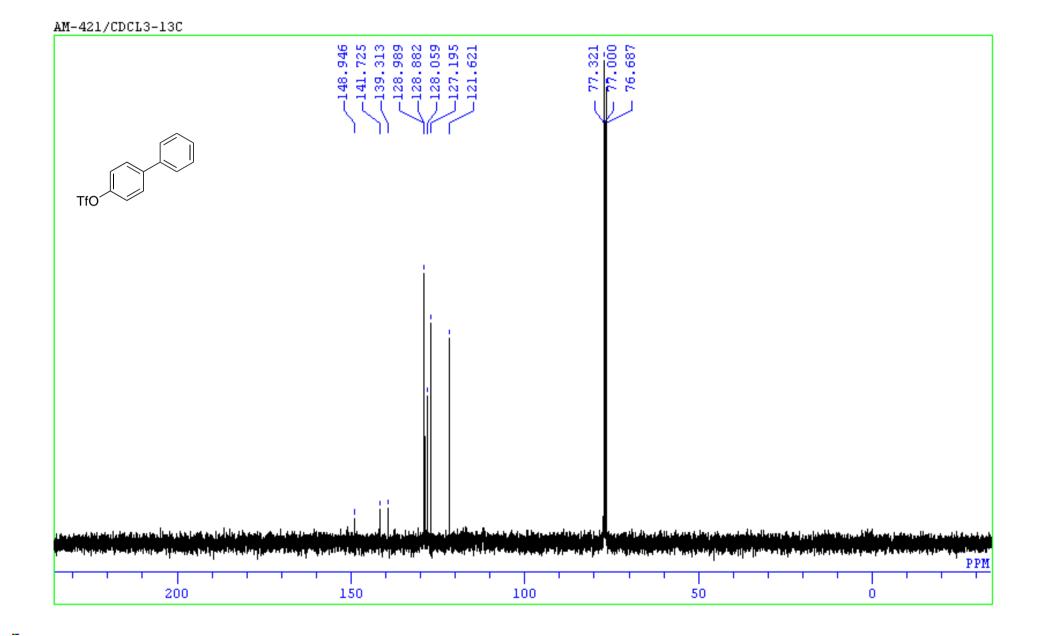




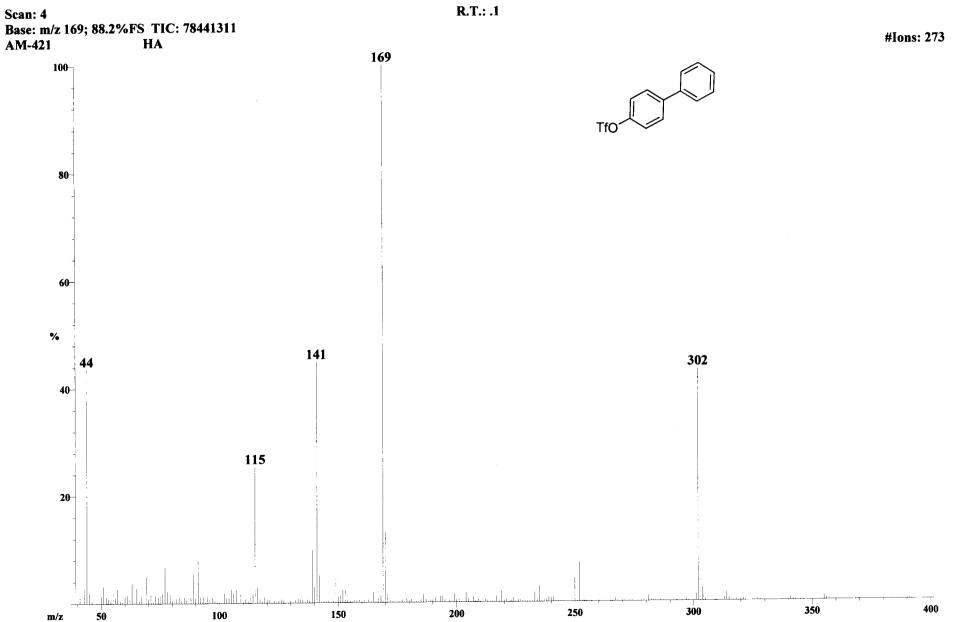
.

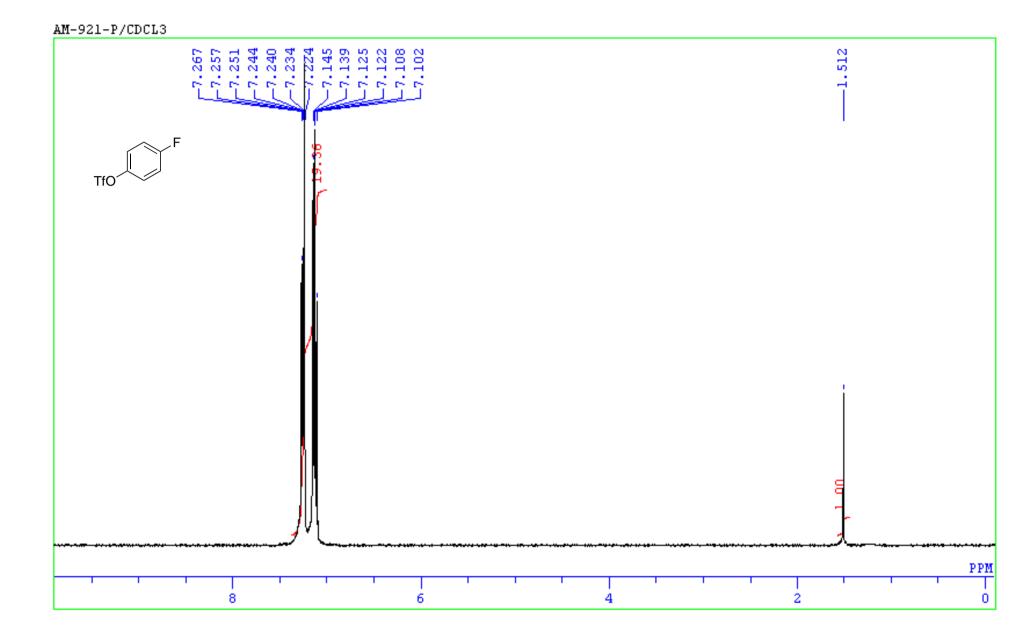
.

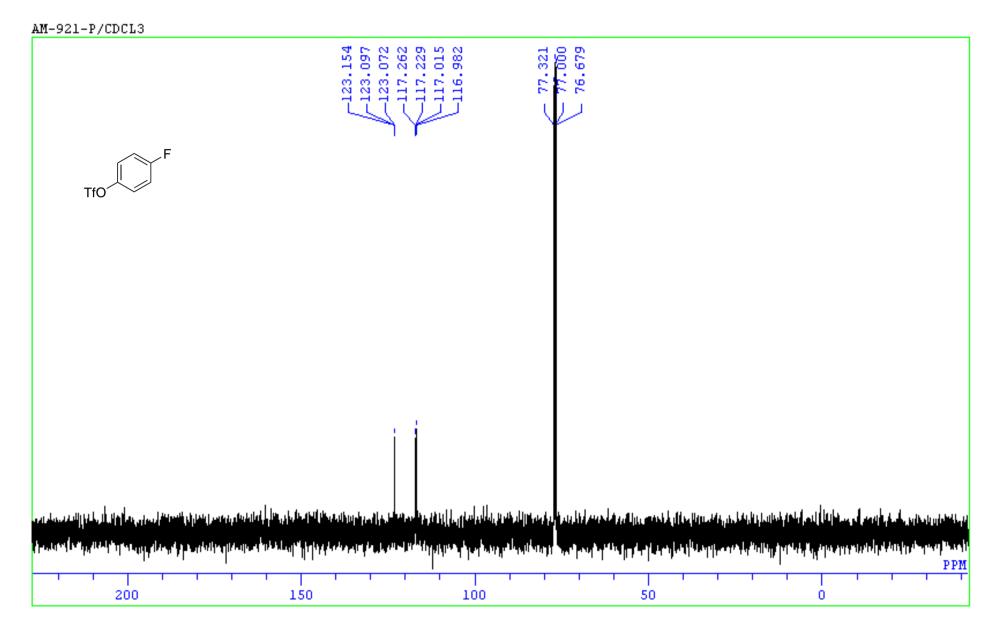




Date Run: 10-18-2005 (Time Run: 10:48:12) Ionization mode: EI+



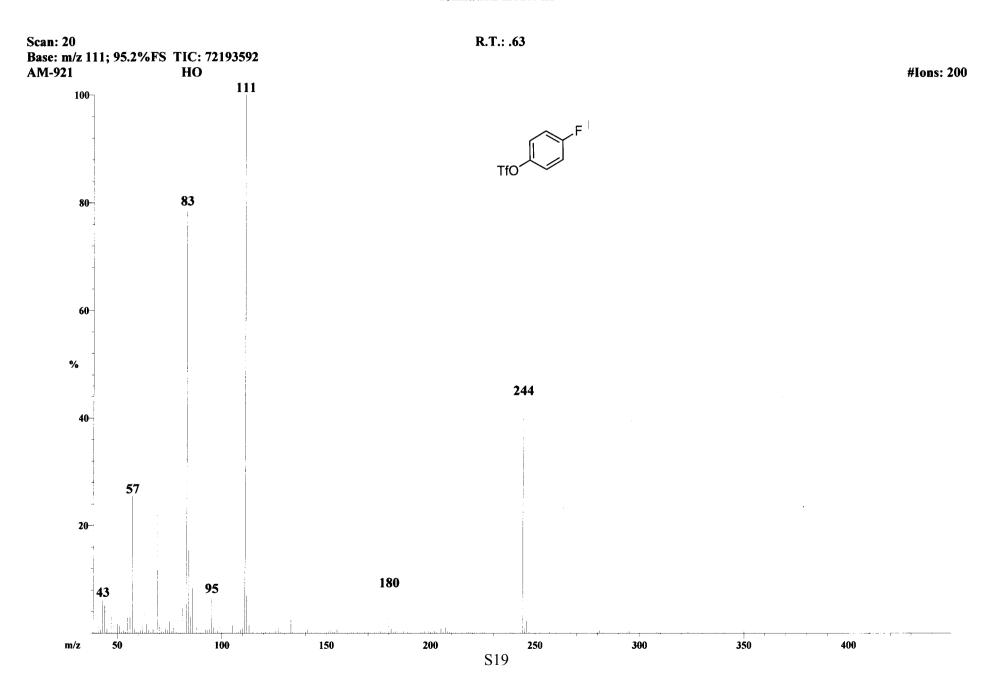


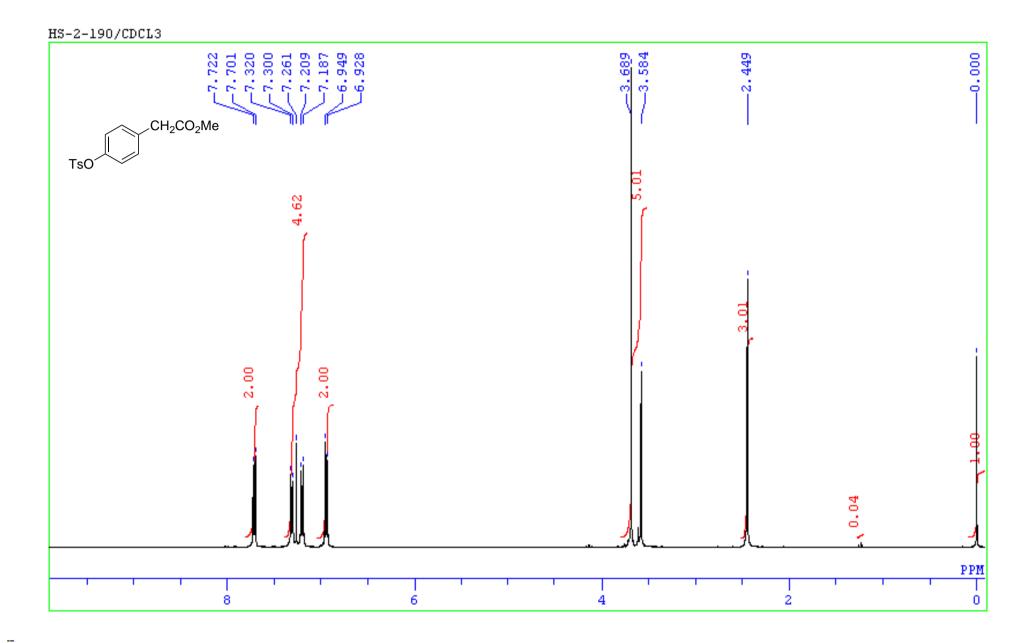


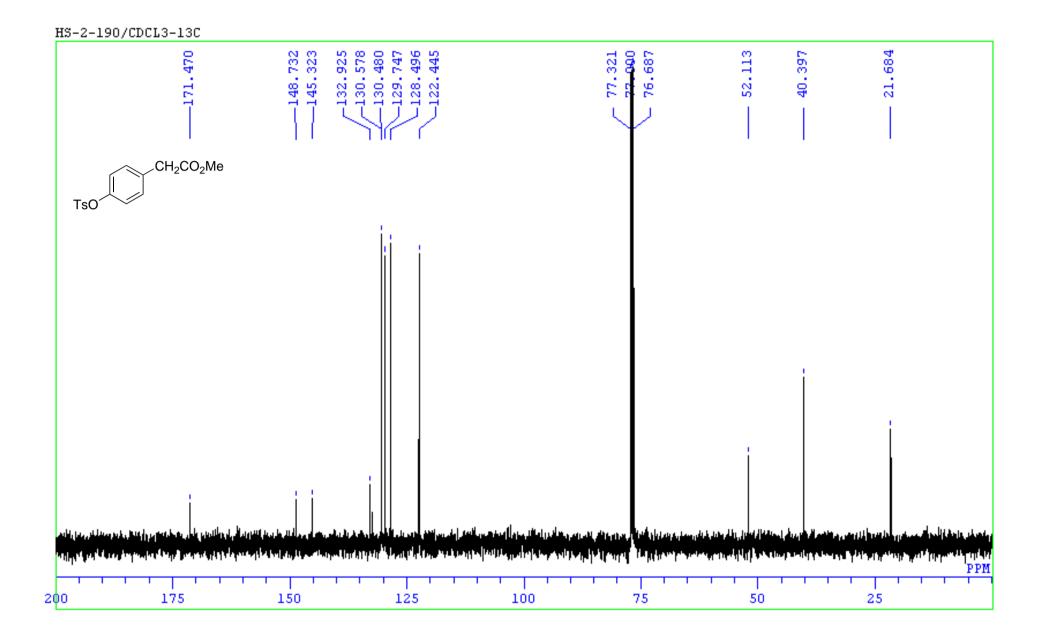
File: LR-OCT184 Instrument: SX102A

Date Run: 10-14-2004 (Time Run: 08:57:17) Ionization mode: EI+

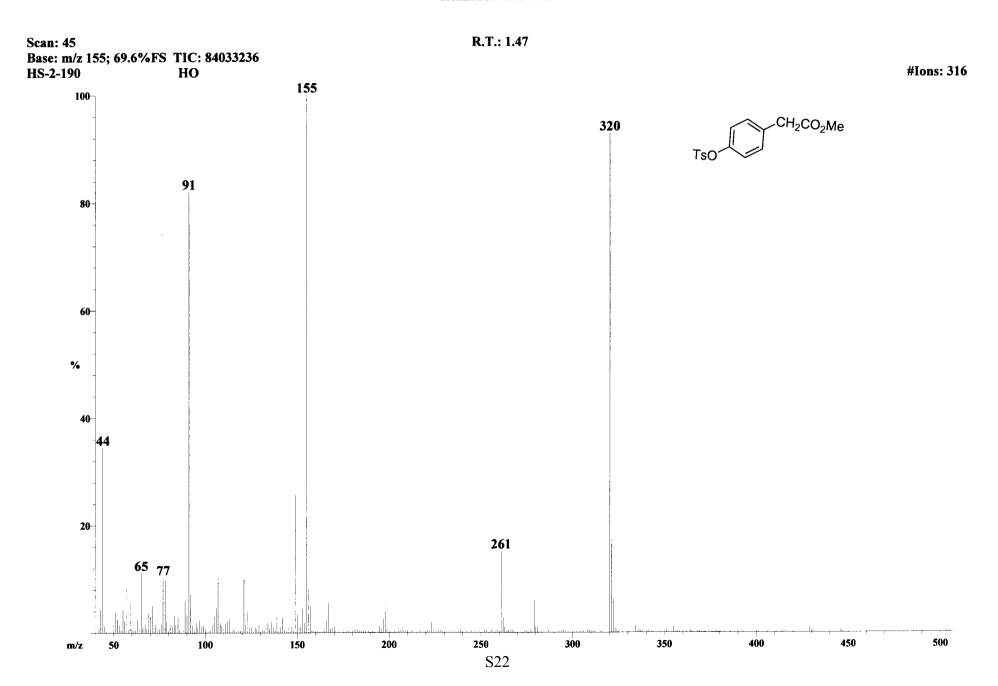
ł

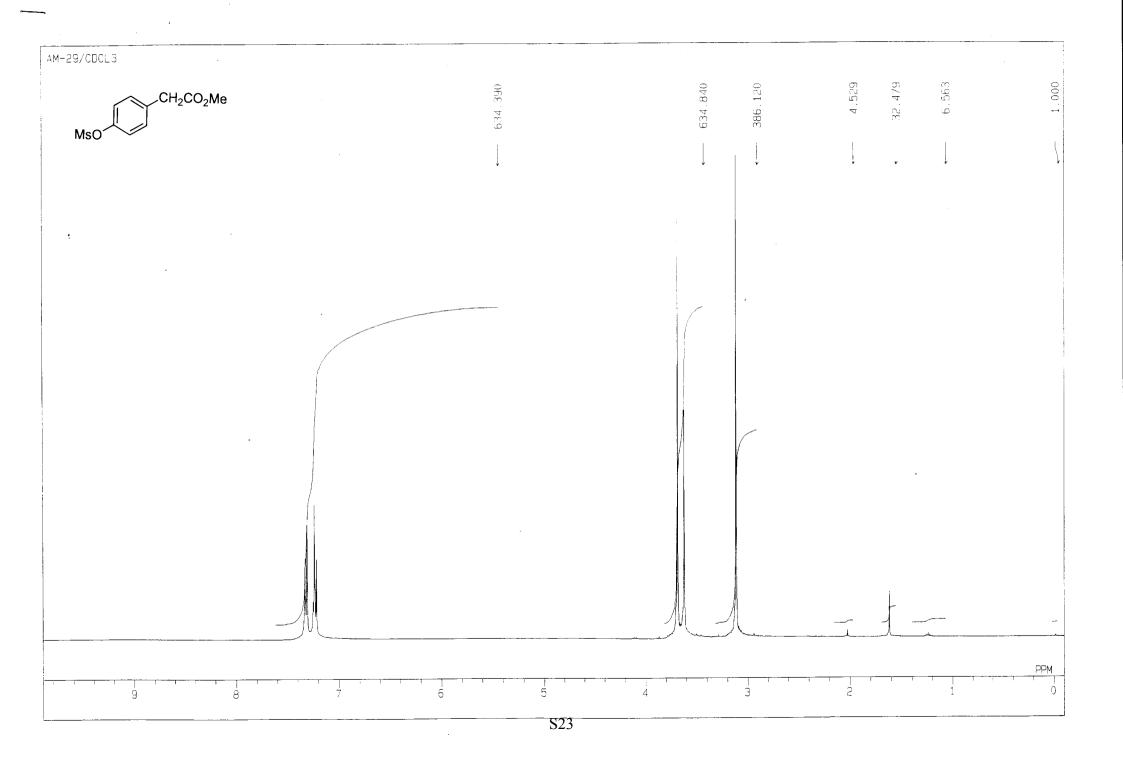


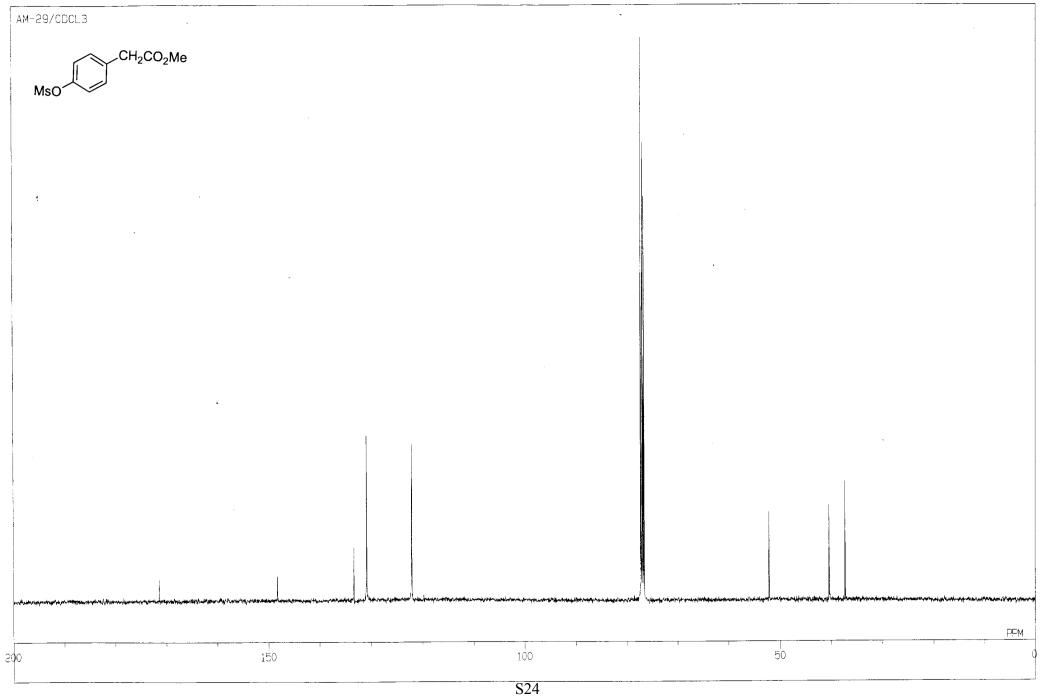


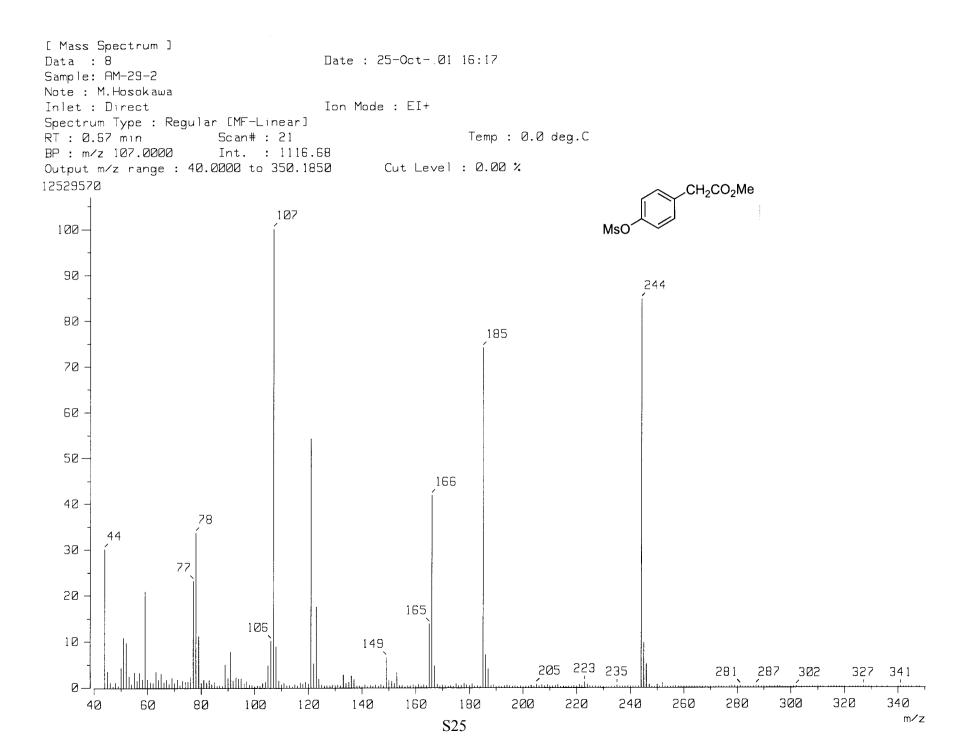


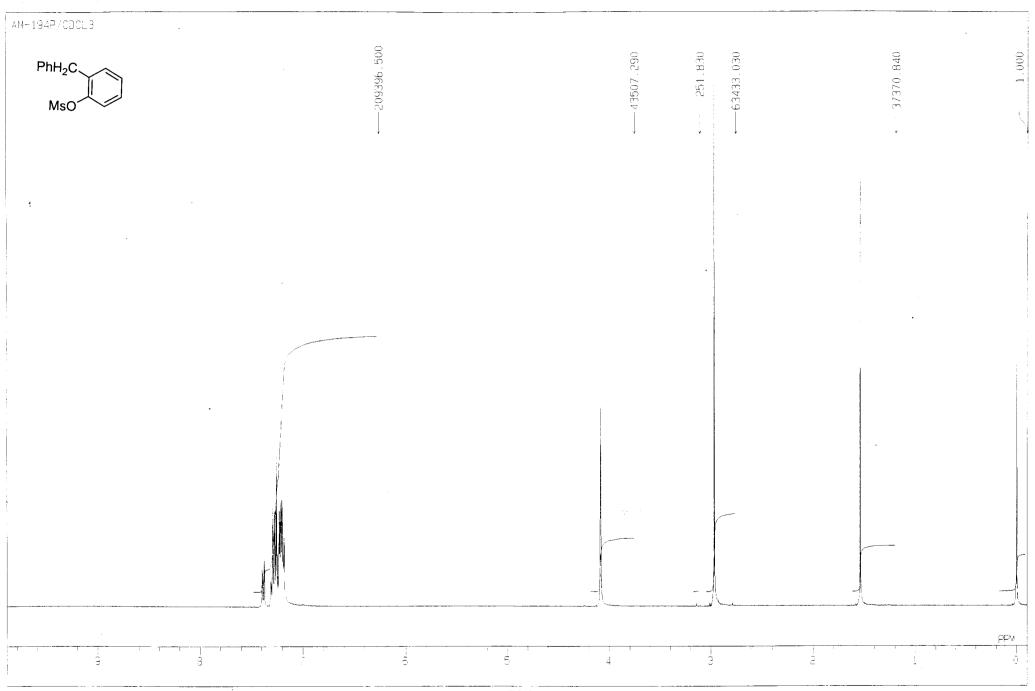
Date Run: 8-8-2005 (Time Run: 10:35:34) Ionization mode: EI+

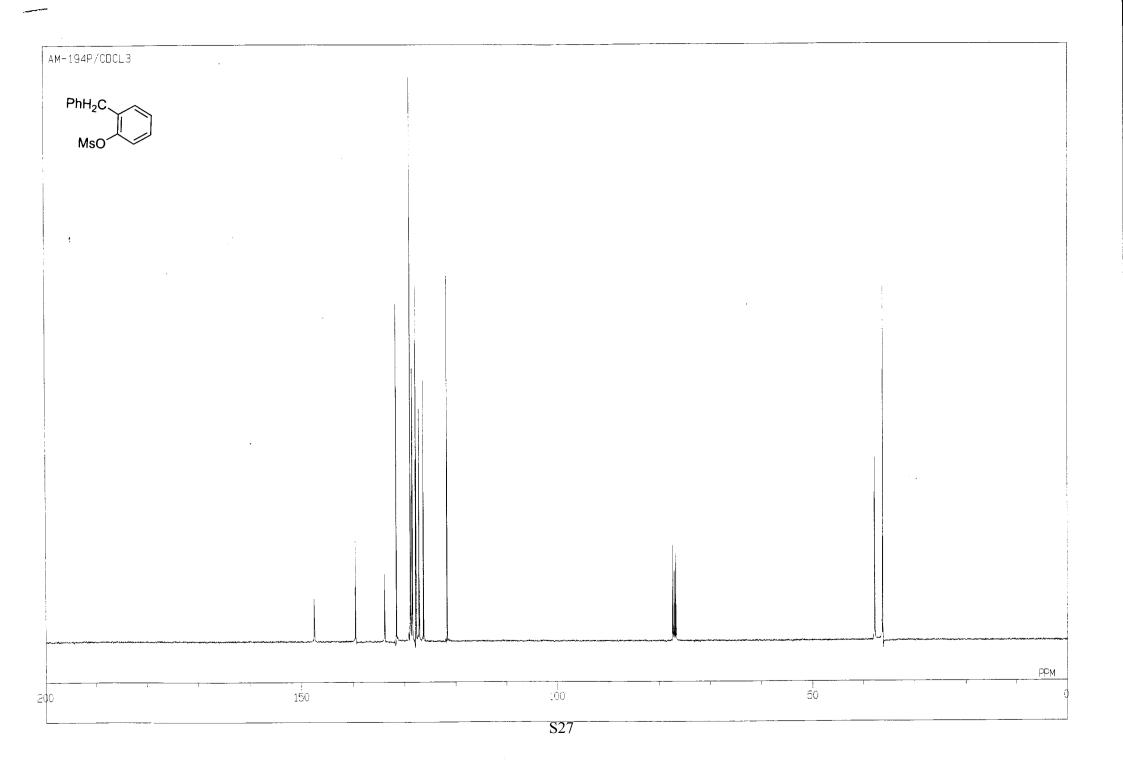


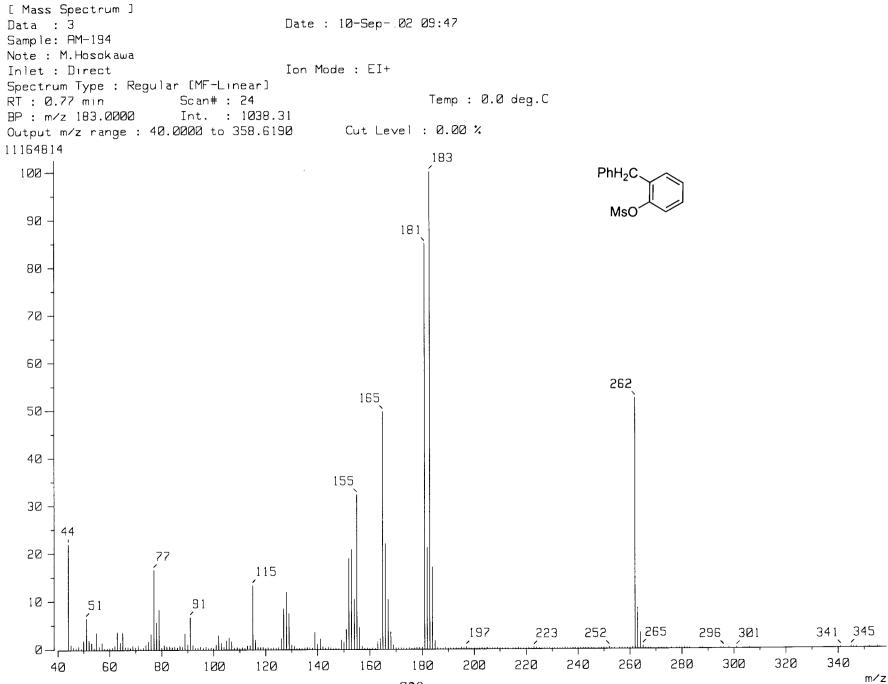


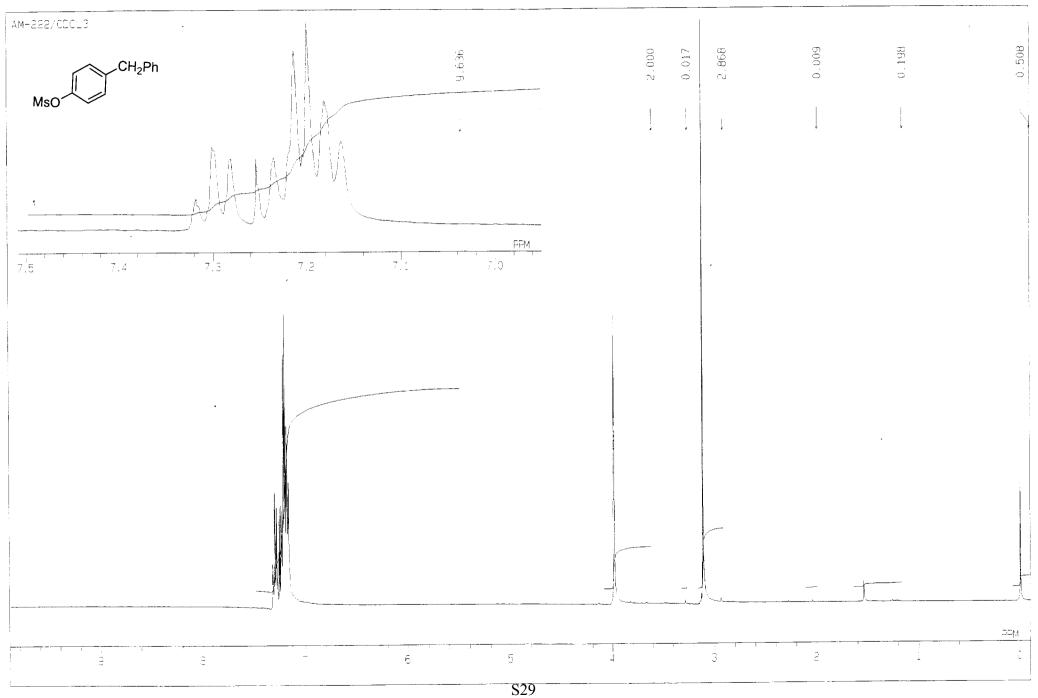


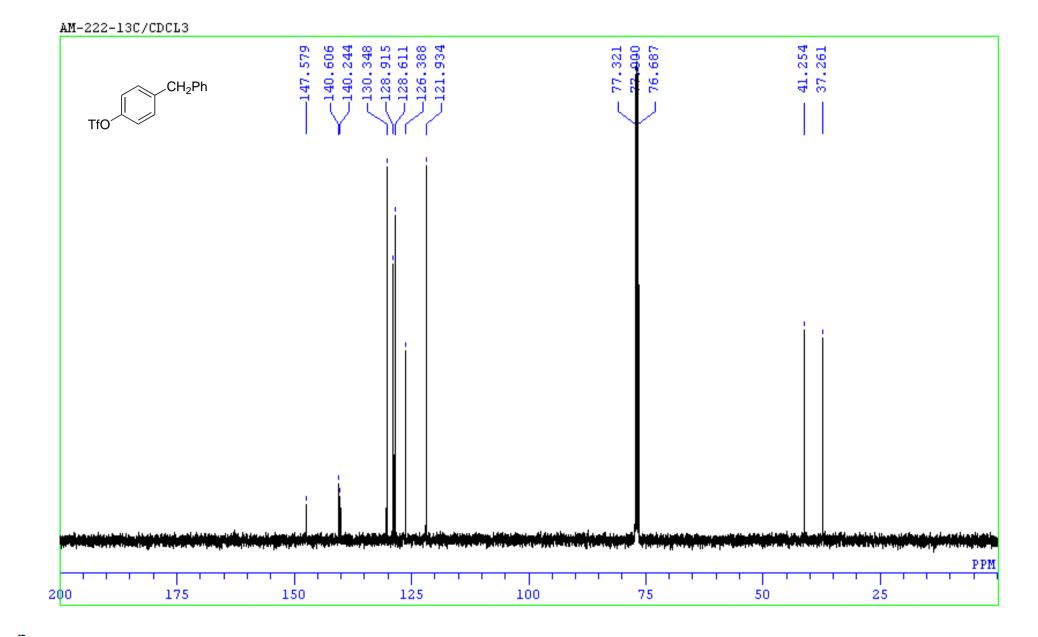


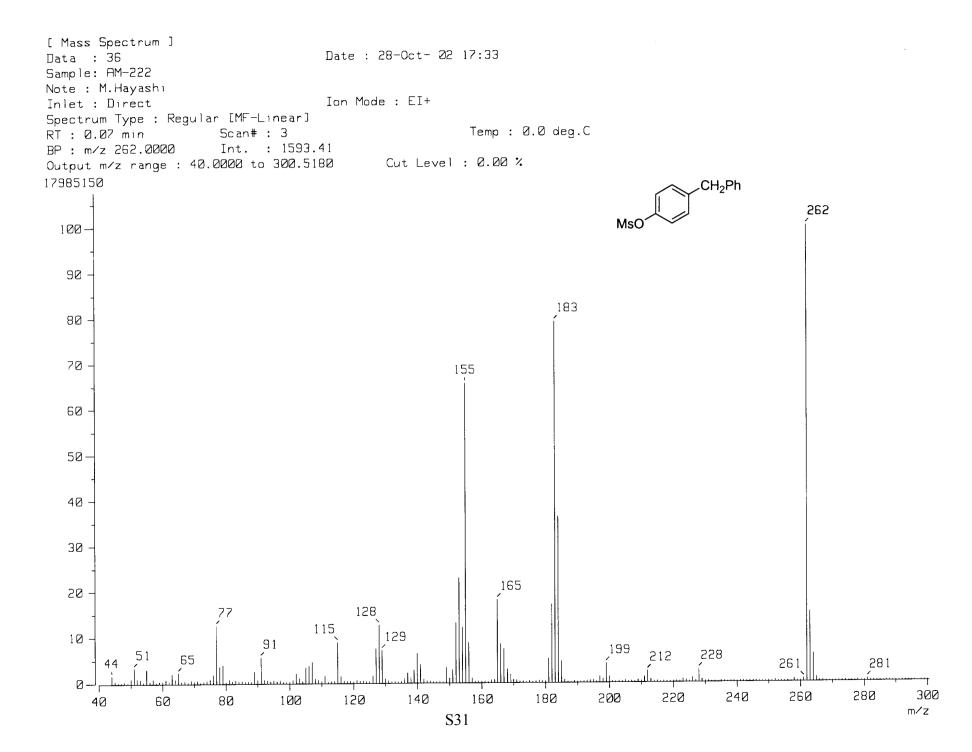












.