

Supplementary Material

The Effect of Different Divalent Cations on the Kinetics and Fidelity of RB69 DNA Polymerase.

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Supplemental Figures S1-S6.

Supporting Information

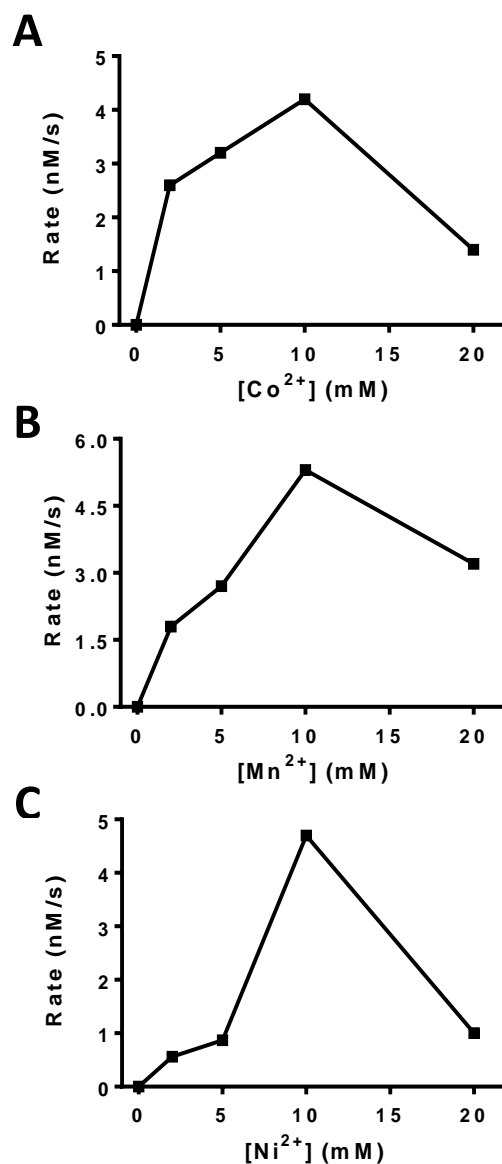


Figure S1. Determination of optimum divalent cation concentrations. RB69pol (40 nM) was pre-incubated with DNA_{13A} (100 nM) in reaction buffer containing increasing concentrations of the divalent cation (Co²⁺, Mn²⁺, or Ni²⁺) [0, 2, 5, 10, and 20 mM]. Reactions were initiated by adding 500 μ M dTTP and subsequently quenched with 0.5 M EDTA (pH 8.0) at various times ranging from 10-100 s. All data were obtained at 23 °C. **(A)** Plot of the rate of DNA product formation as a function of [Co²⁺]. **(B)** Plot of the rate of DNA product formation as a function of [Mn²⁺]. **(C)** Plot of the rate of DNA product formation as a function of [Ni²⁺].

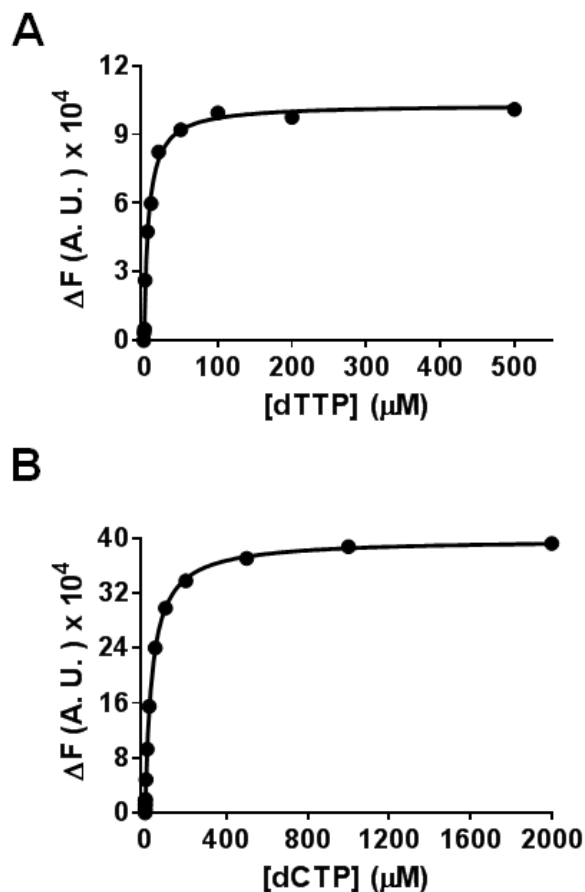


Figure S2. Equilibrium fluorescence titration plots of the RB69pol-ddP/T complex fluorescence quenching with increasing [dTTP] (or [dCTP]). The concentration of DNA_{Pdd} was 200 nM and that of RB69pol was 1 μM . **(A)** Plot showing the change in fluorescence quenching as a function of [dTTP] in the presence of Co^{2+} . The concentrations of dTTP used were 0, 0.5, 1, 2, 5, 10, 20, 50, 100, 200, and 500 μM . Fluorescence intensities at 365 nm were fitted to a hyperbolic equation. Titration of dTTP vs. 2AP in the presence of 10 mM Co^{2+} gives a $K_{d,g} = 5.8 \pm 0.7 \mu\text{M}$. **(B)** Plot showing the change in fluorescence quenching as a function of [dCTP] in the presence of Mn^{2+} . The concentrations of dCTP used were 0, 0.02, 0.04, 0.1, 0.25, 0.5, 1, 2, 5, 10, 20, 50, 100, 200, 500, 1000, and 2000 μM . Fluorescence intensities at 365 nm were fitted to a hyperbolic equation to obtain a $K_{d,g} = 33 \pm 6 \mu\text{M}$. (ΔF) represents the change in fluorescence in the direction of quenching and ΔF increases with an increase in [dTTP] (or [dCTP]).

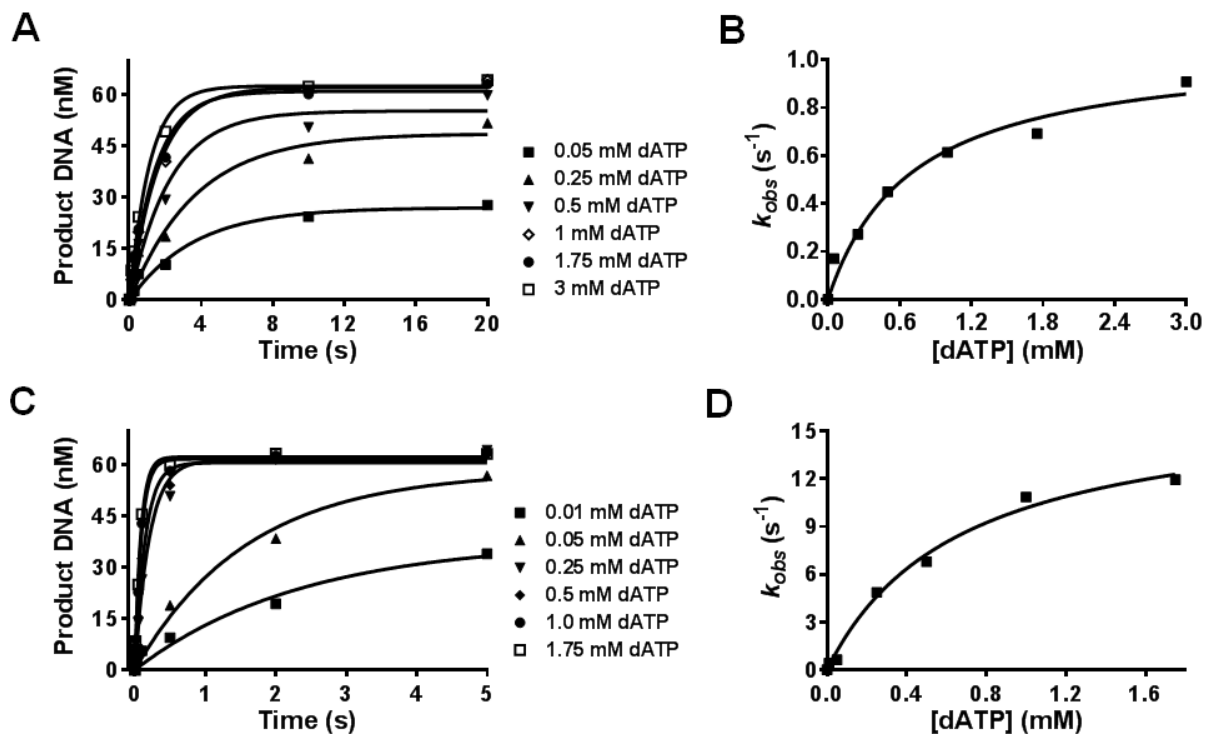


Figure S3. Concentration dependence of the rate of dATP incorporation opposite dT past DNA containing dA/dC mismatch at the primer terminus. RB69pol (1 μM) was pre-incubated with DNA_{ACMM} (80 nM) in reaction buffer and was mixed with increasing concentrations of dATP [0.05, 0.25, 0.5, 1, 1.75, and 3 mM] containing 10 mM Mg^{2+} . Reactions were quenched with 0.5 M EDTA (pH 8.0) at various times ranging from 0.1-20 s. All data were obtained at 23 °C. **(A)** Plots of the amount of extended DNA product obtained as a function of time at various [dATP] in 10 mM Mg^{2+} . Points are experimental, while curves are based on a fit of the data to Eq.1. **(B)** The single exponential rates obtained were plotted as a function of [dATP] and fitted to Eq. 2 to obtain a k_{pol} of $1.1 \pm 0.1 \text{ s}^{-1}$ and a $K_{d,app}$ of $0.8 \pm 0.2 \text{ mM}$. RB69pol (1 μM) was pre-incubated with DNA_{ACMM} (80 nM) in reaction buffer and was mixed with increasing concentrations of dATP [0.01, 0.05, 0.25, 0.5, 1.0, and 1.75 mM] containing 10 mM Mn^{2+} . Reactions were quenched with 0.5 M EDTA (pH 8.0) at various times ranging from 0.04-5 s. **(C)** Plots of the amount of extended DNA product obtained as a function of time at various [dATP] in 10 mM Mn^{2+} . Points are experimental, while curves are based on a fit of the data to Eq.1. **(D)** The single exponential rates obtained were plotted as a function of [dATP] and fitted to Eq. 2 to obtain a k_{pol} of $17 \pm 1 \text{ s}^{-1}$ and a $K_{d,app}$ of $0.6 \pm 0.1 \text{ mM}$.

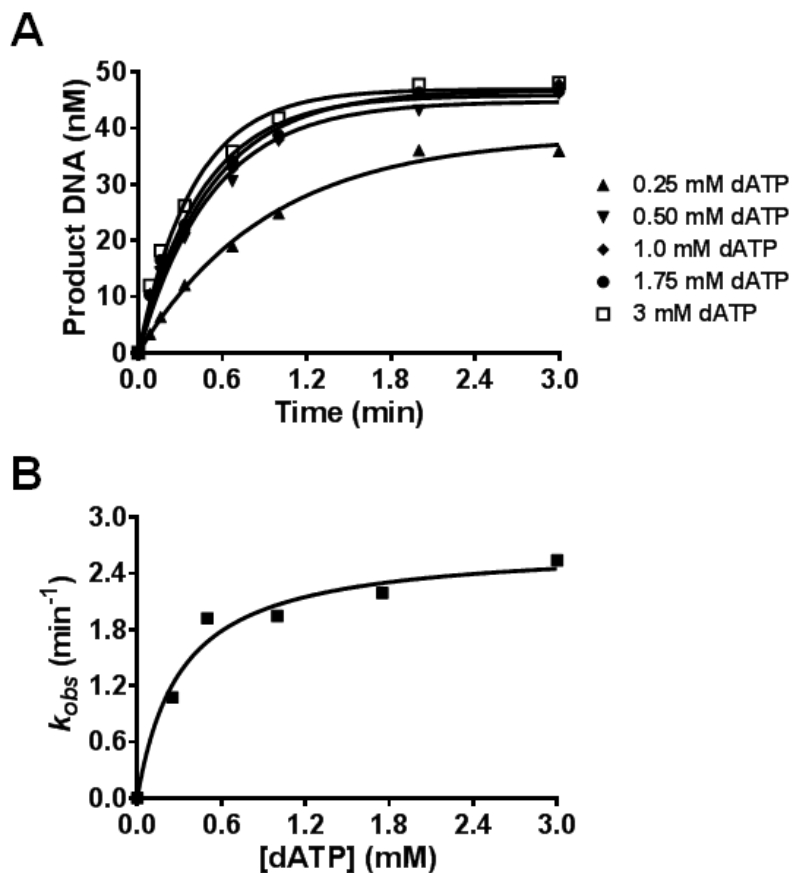


Figure S4. Concentration dependence of the rate of dATP incorporation opposite dT past DNA containing dA/dG mismatch at the primer terminus. RB69pol (1 μM) was pre-incubated with DNA_{AGMM} (80 nM) in reaction buffer and was mixed with increasing concentrations of dATP [0.25, 0.5, 1, 1.75, and 3 mM] containing 10 mM Co^{2+} . Reactions were quenched with 0.5 M EDTA (pH 8.0) at various times ranging from 0.08-3 min. All data were obtained at 23 °C. **(A)** Plots of the amount of extended DNA product obtained as a function of time at various [dATP] in 10 mM Co^{2+} . Points are experimental, while curves are based on a fit of the data to Eq. 1. **(B)** The single exponential rates obtained were plotted as a function of [dATP] and fitted to Eq. 2 to obtain a k_{pol} of $0.05 \pm 0.005 \text{ s}^{-1}$ and a $K_{\text{d,app}}$ of $0.31 \pm 0.08 \text{ mM}$.

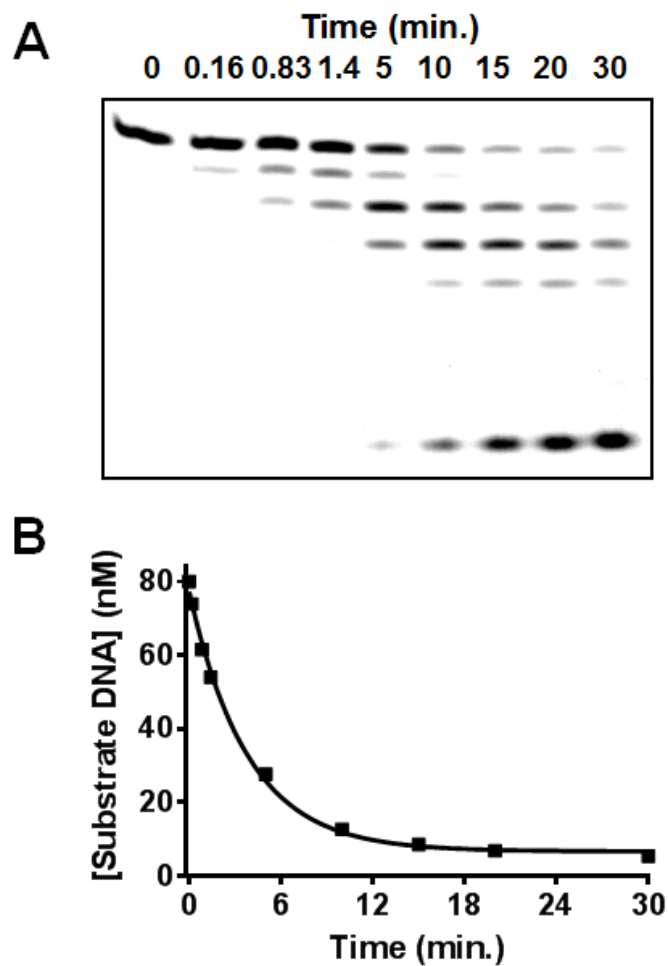


Figure S5. Exonuclease activity of RB69pol using Ni^{2+} . The reaction mixture contained 80 nM 5'-fluorescein labeled 13/18 mer substrate ($\text{DNA}_{13\text{A}}$), 1 μM RB69pol and 10 mM Ni^{2+} . **(A)** Digestion patterns visualized using FUJI scanner with fluorescein as the probe. **(B)** Plot of [Substrate DNA] remaining versus time fit to Eq. 4 to extract a k_{exo} of $0.24 \pm 0.02 \text{ s}^{-1}$.

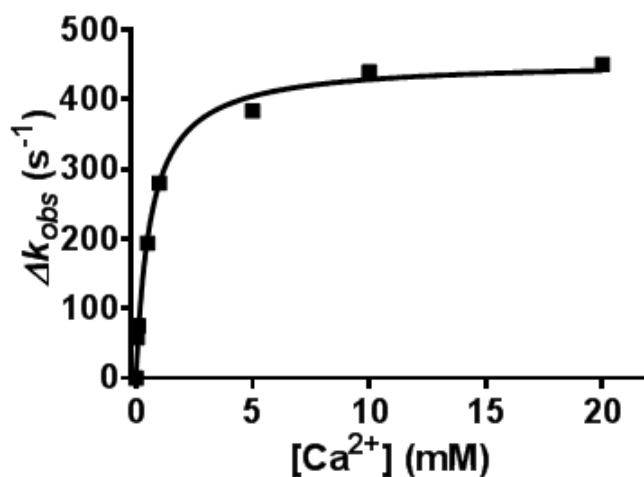


Figure S6. Competition between Ca^{2+} and Mg^{2+} for metal ions binding site. RB69pol (1 μM) was pre-incubated with DNA_{I3A} (80 nM) in reaction buffer and was mixed with dTTP (1 mM) containing Mg^{2+} (10 mM), and varying $[\text{Ca}^{2+}]$ (0.05-20 mM). Reactions were quenched with 0.5 M EDTA (pH 8.0) at various times ranging from 4-100 ms. All data were obtained at 23 °C. Plot of Δk_{obs} as a function of $[\text{Ca}^{2+}]$. Points are experimental, while the curve is based on a fit of the data to Eq. 7 to obtain a $K_{\text{d,Ca}}$ of $630 \pm 72 \mu\text{M}$. (Δk_{obs}) represents change in the rate of reaction as a function of increasing $[\text{Ca}^{2+}]$ and Δk_{obs} increases with an increase in $[\text{Ca}^{2+}]$.