Supporting Information for:

Three-Coordinate Iron(II) Expanded Ring N-Heterocyclic Carbene Complexes

Jay J. Dunsford,[†] David J. Evans,[‡] Thomas Pugh,[†] Sachin N. Shah, [‡] Nicholas F. Chilton,^{*,†} and Michael J. Ingleson^{*,†}

> [†] School of Chemistry, University of Manchester, Brunswick Street, M13 9PL, UK. [‡] Department of Chemistry, University of Hull, Cottingham Road, Hull, HU6 7RX, UK.

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1. X-Ray Crystallography

Data for compounds 2.C₆D₆, 2.C₇H₈, 3, 4, 5 and 6 were recorded on an Agilent Supernova diffractometer, with Mo K α radiation (mirror monochromator, $\lambda = 0.7107$). The CrysAlisPro^[1] software package was used for data collection, cell refinement and data reduction. Data for 1 was recorded on a Bruker APEX-II diffractometer, with Cu K α radiation (graphite monochromator, $\lambda = 1.5418$ Å). The Bruker APEX2 software package was used for cell refinement and data reduction. For all data sets the CrysAlisPro^[1] software package was used for cell refinement and data reduction. For all data sets the CrysAlisPro software package was used for empirical absorption corrections, which were applied using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm. All structures were solved using direct methods and refined against F2 using the OLEX2^[2] software package. Non-hydrogen atoms were refined anisotropically. Hydrogen atoms were all located in a difference map and repositioned geometrically.

Crystallographic data have been deposited with the Cambridge Crystallographic Data Centre under references: 1435131, 1446871, 1446872, 1435128, 1435129, 1446021, and 1435130 for **1**, **2.C₆D₆**, **2.C₇H₈**, **3**, **4**, **5** and **6**, respectively. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

	1	$2.C_6D_6$	2.C ₇ H ₈
Empirical Formula	$C_{29}H_{42}Br_2FeN_2$	C ₃₅ H ₄₈ Cl ₂ FeN ₂	$C_{32.5}H_{42}Cl_2FeN_2$
Fw/g mol ⁻¹	634.31	623.50	587.43
Cryst syst, Space Group	Monoclinic, P21/n	Monoclinic, P2 ₁ /c	Monoclinic, P2 ₁ /c
a/Å	9.43883(12)	17.3888(11)	11.155(2)
b/Å	24.9246(3)	9.6923(4)	19.056(4)
c/Å	12.43441(14)	21.2098(12)	15.342(3)
α/deg	90	90	90
β/deg	95.9310(11)	112.052(7)	96.520(9)
γ/deg	90	90	90
Vol/Å ³	2909.64(6)	3313.1(3)	3240.2(11)
Z, calc density (Mg m ⁻³)	4, 1.448	4, 1.250	4, 1.204
abs coeff (mm ⁻¹)	7.503	0.642	0.652
F(000)	1304.0	1328.0	1240.0
cryst, colour	Plate, Colourless	Block, colourless	Block, colourless
cryst dimens /mm ³	0.2 imes 0.2 imes 0.1	0.4 imes 0.4 imes 0.2	0.4 imes 0.3 imes 0.2
θ range (deg)	7.094 to 133.19	6.576 to 50.048	6.058 to 50.054
Refins collected /unique	35861 / 5080	12492 / 5844	17089 / 5688
R _{int}	0.0259	0.0295	0.0774
Data/restraints/ parameters	5080 / 0 / 315	5844 / 0 / 369	5688 / 0 / 360
final R indicies $[F^2 < 2\theta(F^2)]$:	0.0204, 0.0503	0.0498, 0.1043	0.0802, 0.1486
R indicies (all data): R1, wR2	0.0211, 0.0506	0.0795, 0.1143	0.1497, 0.1864
Largest diff peak and hole/e Å ⁻³	0.30 / -0.45	0.41 / -0.40	0.51/-0.58

Table S1. Crystal Structure Refinement Data for 1, 2.C₆D₆ and 2.C₇H₈

	3	4	5	6
Empirical Formula	C ₂₇ H ₃₈ Br ₂ FeN ₂ O	C ₃₅ H ₄₂ Br ₂ FeN ₂ .2C ₆ H ₆	$C_{78}H_{100}Br_4Fe_2N_4$	C ₃₁ H ₄₆ Br ₂ FeN ₂ O
Fw/g mol ⁻¹	622.26	706.37	1524.95	678.37
Cryst syst, space	Triclinic, P-1	Monoclinic, P21/c	Triclinic, P-1	Orthorhombic,
Group				Pbca
a/Å	8.4392(4)	12.2689(4)	10.6547(5)	15.5289(4)
b/Å	12.4618(5)	16.2289(3)	12.2902(8)	18.4203(6)
c/Å	13.7286(6)	18.0297(5)	16.2704(13)	22.0688(6)
α/deg	78.655(4)	90	70.448(7)	90
β/deg	80.812(4)	108.813(3)	88.517(5)	90
γ/deg	80.569(4)	90	67.228(5)	90
Vol/Å ³	1384.29(11)	3398.10(17)	1838.0(2)	6312.7(3)
Z, calc density (Mg m ^{3})	2, 1.493	4, 1.381	2, 1.378	8, 1.428
abs coeff (mm ⁻¹)	3.452	2.820	2.612	3.035
F(000)	636.0	1448.0	788.0	2800.0
cryst, colour	Block, Colourless	Block, Colourless	Block, Colourless	Block, Colourless
crvst dimens /mm ³	$0.4 \times 0.3 \times 0.3$	0.3 imes 0.3 imes 0.2	0.7 imes 0.3 imes 0.2	0.5 imes 0.2 imes 0.2
θ range (deg)	6.732 to 58.058	6.776 to 50.052	6.716 to 52.744	6.788 to 50.048
Reflns collected /	11179 / 6298	13336 / 5989	13192 / 7489	15419 / 5552
unique				
R _{int}	0.0211	0.0195	0.0361	0.0274
Data/restraints/	6298 / 0 / 304	5989 / 0 / 429	7489 / 0 / 393	5552 / 0 / 342
parameters				
final R indicies $[F^2 < 2\theta(F^2)]$.	0.0318, 0.0692	0.0331, 0.0862	0.0537, 0.1193	0.0307, 0.0631
R indicies (all data):	0.0409, 0.0730	0.0423, 0.0908	0.0812, 0.1336	0.0432, 0.0677
R1, wR2				
Largest diff peak and hole/e Å ⁻³	0.67 / -0.78	0.36 / -0.36	1.08 / -0.83	0.59 / -0.59

Tal	ble	S2.	Crystal	Structure	Refinement	Data	for	3, 4,	, 5 a	and	6
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2. Mössbauer Spectroscopy Data



Figure S1. Mössbauer spectra of [Fe(7-DiPP)Br₂], 1 recorded at 80K.



Figure S2. Mössbauer spectra of [Fe(7-Mes)Br₂(THF)], 3 recorded at 80K.



Figure S3. Mössbauer spectra of [Fe(7-Mes)Br₂], 4 recorded at 80K.



Figure S4. Mössbauer spectra of [Fe(SDiPP)Br₂]₂, **5** recorded at 80K. This spectra was best modelled as two iron(II) species in 77% and 23% respectively. We postulate the major component (77%) to be [Fe(SDiPP)Br₂]₂. *Multiple independently synthesised batches which returned satisfactory microanalyses always contained the same observed minor species*.

3. SQUID Magnetometry Data

The magnetic properties of polycrystalline samples of **1** and **3** were measured using a Quantum Design MPMS-7 SQUID magnetometer at temperatures in the range 1.8-300 K. The sample was prepared in a glove box by lightly crushing polycrystalline samples of **1** and **3** prior to transfer into an NMR tube, being restrained in eicosane (to prevent sample torquing) and flame sealed under vacuum.

4. CASSSCF Calculation Information

CASSCF calculations were performed with MOLCAS 8.0.^[3] The active space consisted of six electrons in the five 3d orbitals. The 5 S = 2, 45 S = 1 and 50 S = 0 configurations were considered both in the orbital optimisation (RASSCF) and the spin-orbit mixing (RASSI) procedures, and the ZFS of the S = 2 ground state was extracted from the spin-orbit mixed states (SINGLE_ANISO). Given orbital energies are the energies of the five S = 2 states calculated in the CASSCF procedure. Basis sets from the ANO-RCC library of VTZP (Fe), VDZP (Br, C) and VDZ (H) quality were employed,^[4] along with Cholesky decomposition of the two-electron integrals. All parameters took their default values. Matrix elements of orbital angular momentum between the spin-free states of the S = 2 configurations taken from the output of the RASSI routine.

Structures for the deformation of torsion angles can be found in the accompanying .xyz file.



Figure S5. K_i parameter for 4 as the bromide ions are deformed along the coordinate δ .

$$K_{j} = \sqrt{\sum_{\alpha,\beta \in x,y,z} (\langle \varphi_{1} | \hat{L}_{\alpha} | \varphi_{j} \rangle \langle \varphi_{j} | \hat{L}_{\beta} | \varphi_{1} \rangle)^{2}}$$

4. References

[1] CrysAlisPro, Agilent Technologies, Version 1.171.35.19 (release 27-10-2011 CrysAlis171 .NET) (compiled Oct 27 2011,15:02:11)

[2] Olex2.refine (O. V. Dolomanov, L. J. Bourhis, R. J. Gildea, J. A. K. Howard and H. Puschmann, J. Appl. Cryst. (2009). 42, 339-341).

[3] Aquilante, F.; Autschbach, J.; Carlson, R. K.; Chibotaru, L. F.; Delcey, M. G.; De Vico, L.; Fdez. Galván, I.; Ferré, N.; Frutos, L. M.; Gagliardi, L.; Garavelli, M.; Giussani, A.; Hoyer, C. E.; Manni, G.; Lischka, H.; Ma, D.; Malmqvist, P. Å.; Müller, T.; Nenov, A.; Olivucci, M.; Pedersen, T. B.; Peng, D.; Plasser, F.; Pritchard, B.; Reiher, M.; Rivalta, I.; Schapiro, I.; Segarra-Martí, J.; Stenrup, M.; Truhlar, D. G.; Ungur, L.; Valentini, A.; Vancoillie, S.; Veryazov, V.; Vysotskiy, V. P.; Weingart, O.; Zapata, F.; Lindh, R. *Journal of Computational Chemistry*, **2015**, DOI: 10.1002/jcc.24221.

[4] Roos, B. O.; Lindh, R.; Malmqvist, P-Å.; Veryazov, V.; Widmark, P.-O. J. Phys. Chem. A, **2005**, 109, 6575–6579.

5. NMR Data for All Novel Compounds



Figure S6. ¹H NMR of [Fe(7-DiPP)Br₂], **1** (THF-*d*₈, 400 MHz, 298K).



Figure S7. ¹H NMR of [Fe(7-DiPP)Cl₂], **2** (*protio*-THF, 400 MHz, 298K).



Figure S8. ¹H NMR of [Fe(7-Mes)Br₂(THF)], $3 (C_6D_6, 400 \text{ MHz}, 298\text{K})$.



Figure S9. ¹H NMR of [Fe(7-Mes)Br₂], **4** (C₆D₆, 400 MHz, 298K).



Figure S10. ¹H NMR of [Fe(SDiPP)Br₂]₂, **5** (C₆D₆, 400 MHz, 298K).



Figure S11. ¹H NMR of [Fe(SDiPP)Br₂(THF-*d*₈)], **6** (THF-*d*₈, 400 MHz, 298K).

6. Additional NMR Experiments

6a. The addition of excess d_8 -THF to a C₆D₆ solution of [Fe(7-DiPP)Br₂], 1



Figure S13. ¹H NMR (C_6D_6 , 400 MHz, 298K) of: [Fe(7-DiPP)Br₂], **1** in C_6D_6 with an excess of d_8 -THF.



Figure S14. Collected ¹H NMR (C_6D_6 , 400 MHz, 298K) of: [Fe(7-Mes)Br₂], **4** (blue); and [Fe(7-Mes)Br₂], **4** spiked with excess d_8 -THF (red).