

Supporting Information

Single Hydrogen Bond Donation from Flavin N₅ to Proximal Asparagine Ensures FAD Reduction in DNA Photolyase

I M. Mahaputra Wijaya,[§] Tatiana Domratcheva,^{±,} Tatsuya Iwata,^{§,†} Elizabeth D. Getzoff,^δ and Hideki Kandori^{§,†,*}*

[§]Department of Frontier Materials, and [†]OptoBio-Technology Research Center, Nagoya Institute of Technology, Japan.

[±] Department of Biomolecular Mechanisms, Max Planck Institute for Medical Research, Jahnstrasse 29, Heidelberg 69120 Germany

^δDepartment of Integrative Structural and Computational Biology and The Skaggs Institute for Chemical Biology, The Scripps Research Institute, USA

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Expression of Enzymes. The unlabeled, uniform ^{13}C -labeled, and Asn reverse-labeled *E. coli* CPD photolyase enzyme was made from the E109A mutant which does not carry MTHF, used as WT enzyme, and prepared as N-terminal His₆-tagged protein using an expression system in *E. coli* strain BL21(DE3) as previously reported with slight modifications.^{16,31} For the uniform ^{13}C labeled enzyme, *E. coli* with the CPD photolyase gene was grown in M9 minimum medium with ^{13}C glucose (Cambridge Isotope Laboratories; 99.0% purity). The ^{12}C asparagine in ^{13}C enzyme, termed Asn reverse-labeled enzyme, was prepared as follows. The ^{12}C (unlabeled) Asn (100 mg/L) was added 30 min prior to the addition of IPTG into *E. coli* CPD photolyase grown in ^{13}C M9 minimum medium. Cells were grown at 24 °C until the OD₆₆₀ reached 0.4, IPTG was added to a final concentration of 1 mM, and the culture was incubated at 18 °C for 18 h, then harvested by centrifugation at 8,000 rpm for 5 min.

Purification of Enzymes. The harvested cells were resuspended in lysis buffer [50 mM NaH₂PO₄, 300 mM NaCl, 20 mM imidazole, and 5% (v/v) glycerol (pH 8.0)]. Using a French press to break the cells, the insoluble fraction was removed by centrifugation (17,000 g) for 60 min. The supernatant was loaded onto a Co-NTA column (TALON Metal Affinity Resin, Clontech) which was then washed in a 20-bed volume wash buffer [50 mM NaH₂PO₄, 300 mM NaCl, and 20 mM imidazole (pH 8.0)]. The fusion protein was eluted with elution buffer [50 mM NaH₂PO₄, 300 mM NaCl, and 250 mM imidazole (pH 8.0)]. For the enzyme resting in a semiquinoid form in an unphotolyzed state, the purified protein in elution buffer was then exchanged into storage buffer with an electron donor [50 mM HEPES, 100 mM NaCl, 25 mM β-mercaptoethanol, and 20% (v/v) glycerol (pH 7.0)] by dilution and ultrafiltration (Amicon Ultra 30K device, Millipore) prior to being stored at -80 °C prior to use. Meanwhile, for oxidized enzyme in an unphotolyzed state, to induce

oxidation, the purified enzyme was exchanged with storage buffer without an electron donor [50 mM HEPES, 100 mM NaCl, and 20% (v/v) glycerol (pH 7.0)] stored at 4 °C and aerated for several days until the green enzyme visibly turned yellow. The presence of ^{12}C Asparagine in ^{13}C enzyme does not change or alter the stability of the enzyme, since both enzyme has the same stability in room temperature upon sample preparation for spectroscopic measurement. From 1 L of *E. coli* culture, approximately 6 mg of E109A (WT) and 3 mg of uniform ^{13}C -labeled and Asn reverse-labeled enzyme were obtained from each liter of LB and M9 media, respectively. Isotope-labeling is believed to have yielded >90% of enzyme (see Results section).

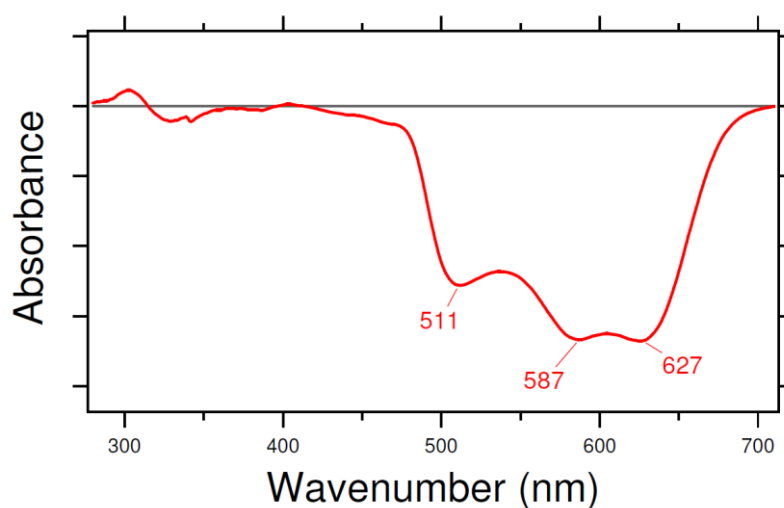


Figure S1. Light-induced difference UV-visible spectra of FADH• into FADH⁻ of unlabeled enzyme by illumination of >550 nm light for 2 min; one division is 10⁻² absorbance units. Peak close to 300 nm is due to sample absorbance exceeded detector limitation.

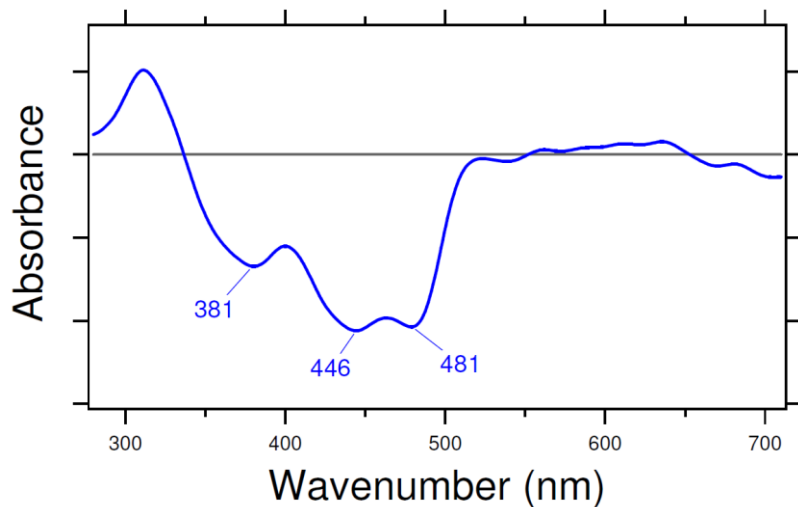


Figure S2. Light-induced difference UV-visible spectra from FAD^{ox} into FADH^- of unlabeled enzyme by illumination of >390 nm light for 2 min followed by >550 nm for 30 s; one division is 10^{-3} absorbance units. Peak close to 300 nm is due to sample absorbance exceeded detector limitation.

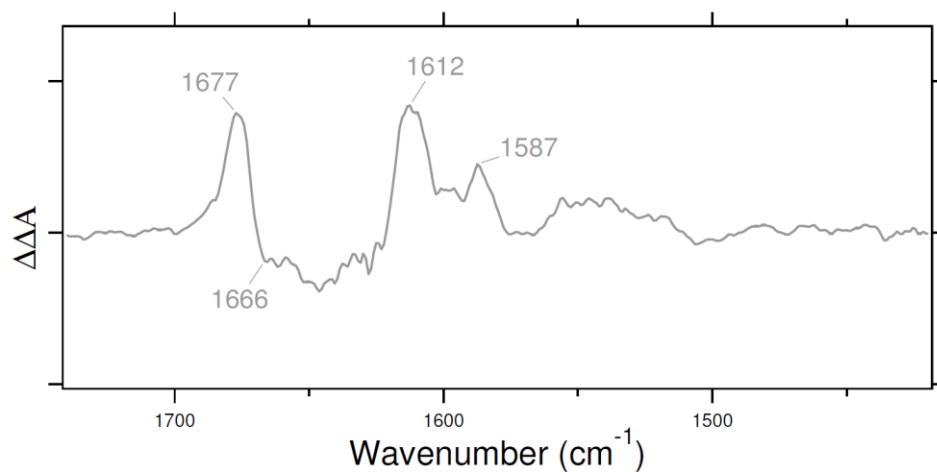


Figure S3. Double difference spectra of Asn reverse-labeled (blue) *minus* Uniformly ^{13}C -labeled (red) in Figure 1b, where one division equals 0.003 absorbance unit.

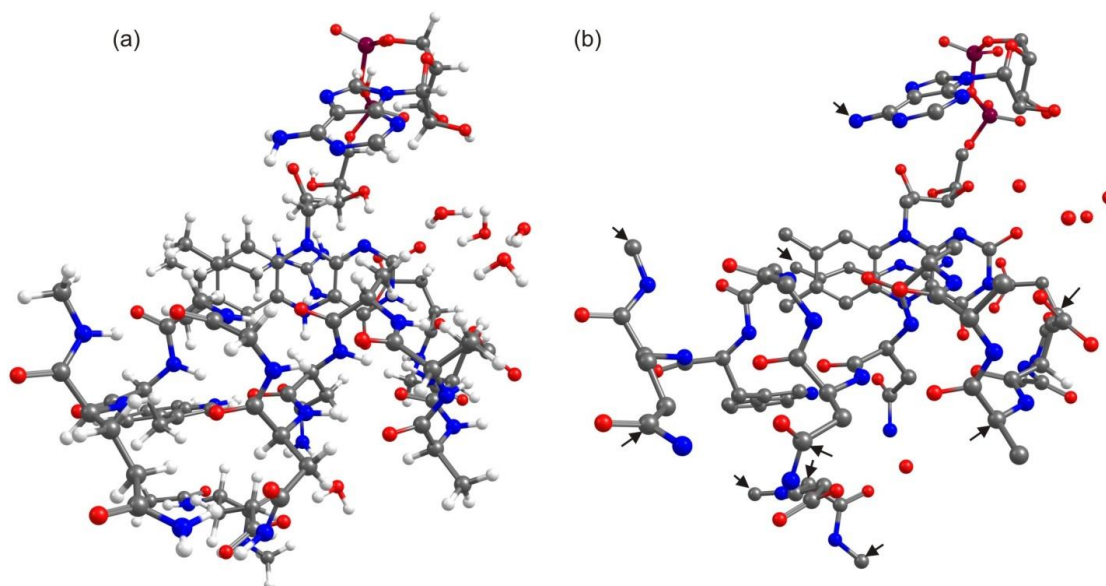


Figure S4. The computational model of the *E. coli* photolyase active site. (a) All atoms of the model in the FADH⁻ state and (b) the heavy atoms of the model shown in the panel (a). The black arrows point to the atoms with the Cartesian coordinates fixed to the respective PDB values during geometry optimization.

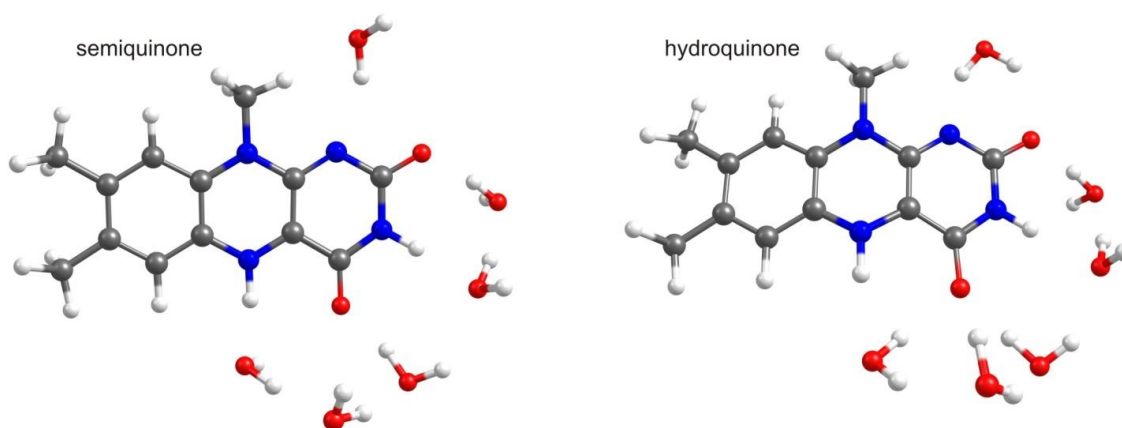


Figure S5. Structure of the lumiflavin-water complex LfH-6H₂O in the neutral semiquinone and anionic hydroquinone states. Geometry optimization was carried out with the (U)B3LYP/6-31G* method.

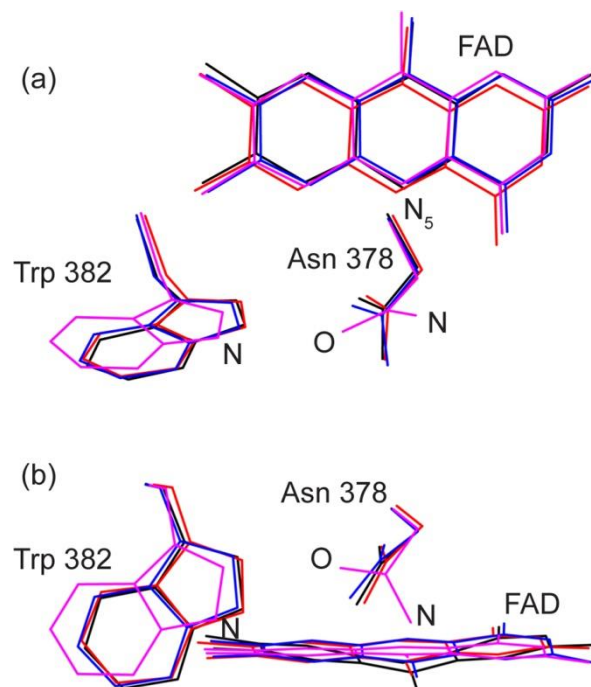


Figure S6. Geometry of the model in the FAD^{ox} state with rotated Asn378 (magenta) in comparison to the geometries of the models presented in Figure 4. The N and O atoms are indicated for rotated Asn378.

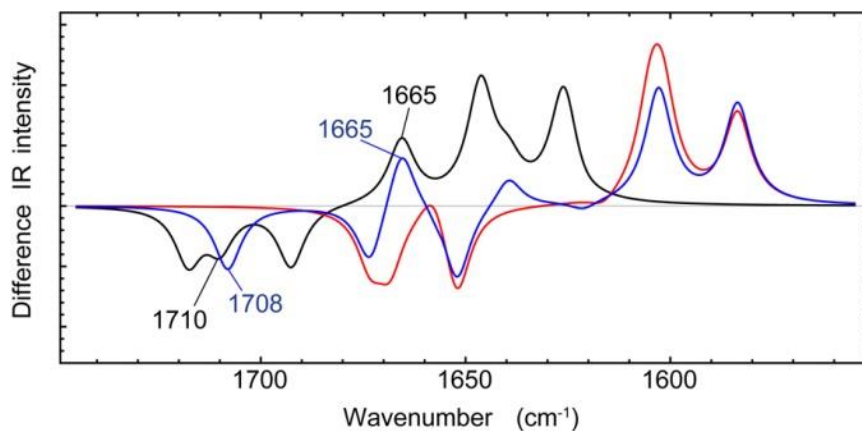


Figure S7. Computed infrared difference spectra $\text{FADH}^- \text{ minus } \text{FAD}^{\text{ox}}$. The model with FAD^{ox} contains rotated Asn378. Unlabeled (black), uniformly ^{13}C -labeled (red) and Asn reverse-labeled (blue) spectra. The numbers indicate the assignment of the Asn378 frequencies. Assignments of

other frequencies in the region of the double bond stretches are presented in Table S1. IR intensity is in Debye/AMU·Å².

Table S1. Assignments of the computed infrared spectra in the region of the double bond stretches.

¹² C			¹³ C			Fl- ¹³ C, Asn- ¹² C		
Frq. ^{a)}	IR Int. ^{b)}	Assign. ^{c)}	Frq.	IR Int.	Assign.	Frq.	IR Int.	Assign.
FAD ^{ox}								
1730	9.79	Fl C=O ₂	1686	10.21	Fl C=O ₂	1686	9.98	Fl C=O ₂
1702	11.95	Fl C=O ₄	1665	3.8	Asn	1677	11.6	Asn
1677	10.23	Asn	1661	13.33	Fl C=O ₄	1662	9.52	Fl C=O ₄
1645	3.27	Asn	1613	6.29	Asn	1645	3.13	Asn
1613	1.03	Fl	1559	2.13	Fl	1559	2.11	Fl
1568	15.56	Fl	1523	15.28	Fl	1522	15.38	Fl
1530	16.68	Fl	1496	2.63	Fl	1495	2.64	Fl
1516	3.24	Fl	1489	0.43	Fl	1490	0.42	Fl
1488	0.65	Fl	1469	2.7	Fl	1469	2.71	Fl
FAD ^{ox} -(Asn ^{rot})								
1718	9.1	Fl C=O ₂	1673	8.31	Fl C=O ₂	1709	10.42	Asn
1710	6.46	Asn	1668	9.22	Asn	1673	10.28	Fl C=O ₂
1693	19.07	Fl C=O ₄	1652	14.95	Fl C=O ₄	1652	12.81	Fl C=O ₄
1621	1.4	Asn	1616	1.83	Asn	1621	1.44	Asn
1614	0.18	Fl	1559	0.51	Fl	1559	0.53	Fl
1565	11.21	Fl	1520	11.93	Fl	1520	11.98	Fl
1527	14.31	Fl	1494	5.05	Fl	1494	5.04	Fl
1508	0.09	Fl	1481	1.78	Fl	1481	1.78	Fl
1491	0.41	Fl	1468	0.33	Fl	1572	0.56	Fl
FADH [•]								
1688	16.79	Fl C=O ₂	1657	5.27	Asn	1663	9.28	Asn
1664	4.07	Asn	1644	19.0	Fl C=O ₂	1644	18.98	Fl C=O ₂
1659	20.78	Fl C=O ₄	1617	12.08	Fl C=O ₄	1626	12.23	Fl C=O ₄
1626	6.05	Asn	1587	8.20	Asn	1616	5.82	Asn
1603	0.17	Fl	1556	1.79	Fl	1557	1.43	Fl
1591	2.81	Fl	1546	0.94	Fl	1547	0.68	Fl
1530	16.26	Fl	1494	12.18	Fl	1494	12.64	Fl
1516	2.37	Fl	1491	1.65	Fl	1492	1.16	Fl
1502	0.78	Fl	1472	3.2	Fl	1472	3.19	Fl
FADH ⁻								
1666	10.48	Asn	1658	6.02	Asn	1666	10.94	Asn
1646	19.30	Fl C=O ₂	1603	14.66	Asn	1640	5.06	Asn
1640	5.02	Asn	1602	15.29	Fl C=O ₂	1603	19.01	Fl C=O ₂
1626	19.13	Fl C=O ₄	1584	14.61	Fl C=O ₄	1584	16.4	Fl C=O ₄
1599	0.28	Fl	1544	0.38	Fl	1544	0.35	Fl
1584	3.39	Fl	1530	2.57	Fl	1530	2.63	Fl
1555	3.64	Fl	1505	2.81	Fl	1505	2.8	Fl
1505	4.75	Fl	1482	0.85	Fl	1482	0.83	Fl
1493	1.36	Fl	1474	0.42	Fl	1474	0.42	Fl

^{a)} harmonic normal mode frequency, scaled by 0.965 (in cm⁻¹)

b) infrared intensity (in Debye/AMU·Å²)

c) assignments of the normal models to the C=O and NH₂ vibrations of Asn378 (indicated as "Asn"), C=O₂ and C=O₄ of flavin ("Fl C=O₂" and "Fl C=O₄") and C=C and C=N of flavin ("Fl").