

**SUPPORTING INFORMATION**

Characterization of ZnE (E = S, Se, Te)  
Materials Synthesized Using  
Silylated–Chalcogen Reagents in Mesoporous  
MCM-41

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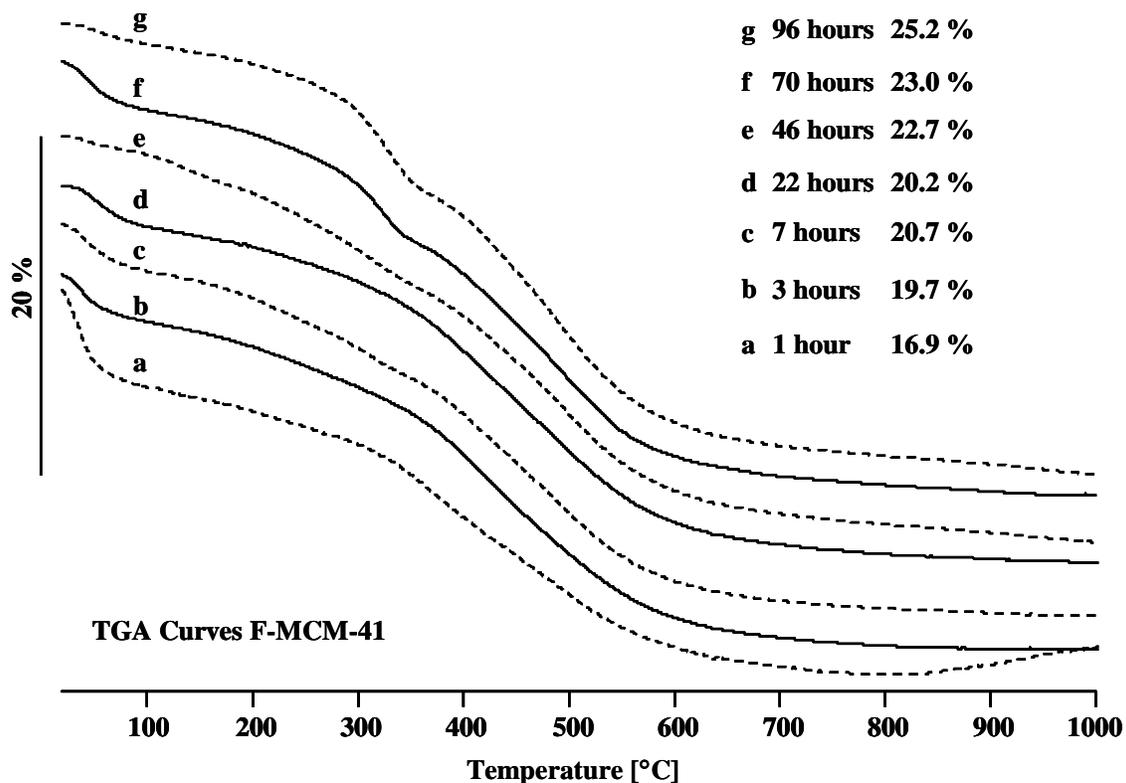
### Supplementary Material.

**Figure S1.** TGA curves for functionalized MCM-41 monitored over a 96 hour time period. Percentage indicates percent loss during heating.

**Figure S2.** TGA curves for Zn-MCM-41 monitored over a 96 hour time period. Percentage indicates percent loss during heating.

**Figure S3.** Raman spectra of TPED, F-MCM-41 and Zn-MCM-41. C–H deformation of ethylene diamine ( $\times$ ).

**Table S1.** Peak area ratios between  $S(\text{SiMe}_3)_2$ ,  $\text{Me}_3\text{SiOAc}$  and  $\text{O}(\text{SiMe}_3)_2$  and the internal standard (toluene) taken at specified time intervals during the synthesis of ZnS-MCM-41. Shown are the ratios observed when an excess of  $S(\text{SiMe}_3)_2$  and an approximately stoichiometric (1:0.5) amount of  $S(\text{SiMe}_3)_2$  are used in the synthesis.



**Figure S1.**

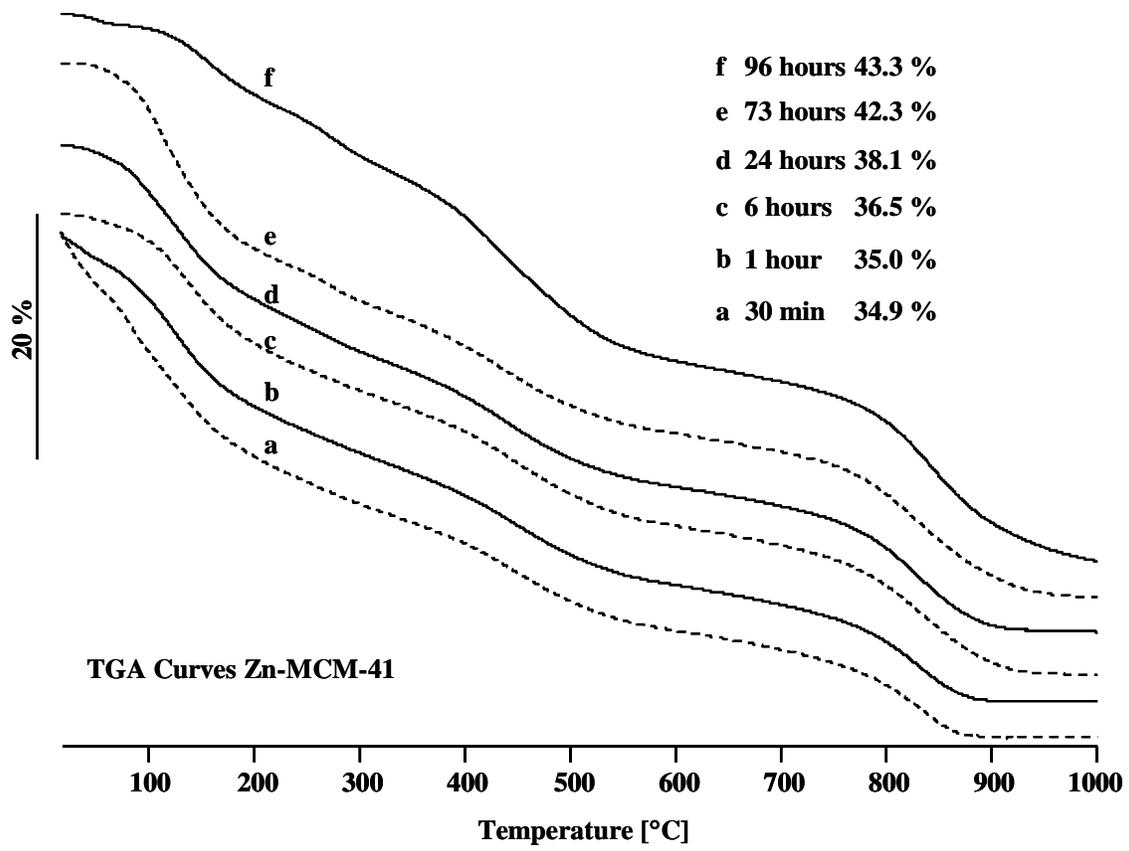


Figure S2.

## Raman Spectra

(×) C–H deformation of ethylene moiety

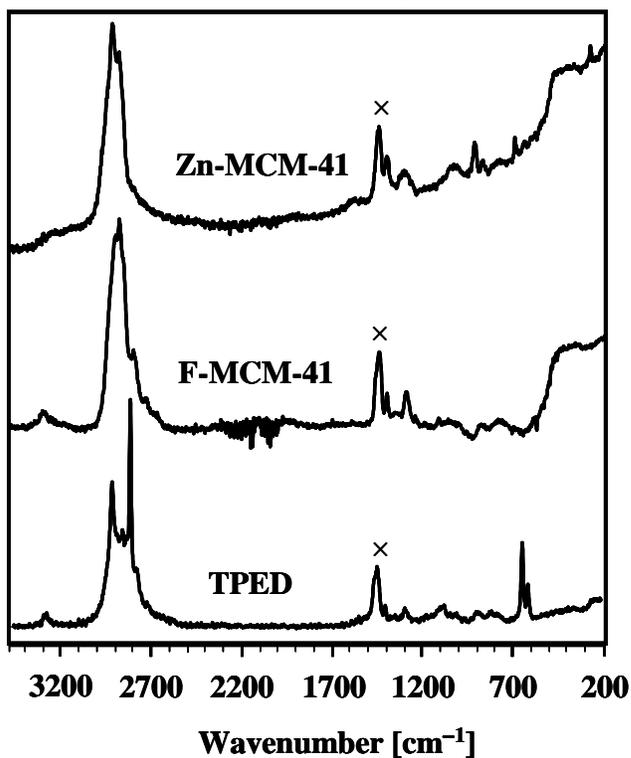


Figure S3.

Table S1.

Time [hours]	Peak Area Ratios Between Designated Product and Toluene Standard					
	S(SiMe <sub>3</sub> ) <sub>2</sub>		Me <sub>3</sub> SiOAc		O(SiMe <sub>3</sub> ) <sub>2</sub>	
	Excess	No-excess	Excess	No-excess	Excess	No-excess
2	1.5	0.3	0.4	0.3	0.0	0.0
4	1.3	0.2	0.4	0.4	0.0	0.0
22	1.1	0.0	0.4	0.4	0.2	0.2
70	1.0	0.0	0.4	0.4	0.5	0.3
90	0.9	0.0	0.4	0.27	0.9	0.3
120	0.8	0.0	0.4	0.11	1.1	0.3
160	0.4	0.0	0.4	0.0	1.8	0.4