

Supporting Information

Hollow PbWO_4 Nanospindles via a Facile Sonochemical Route

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1. Comparative experimental results under vigorous electric stirring

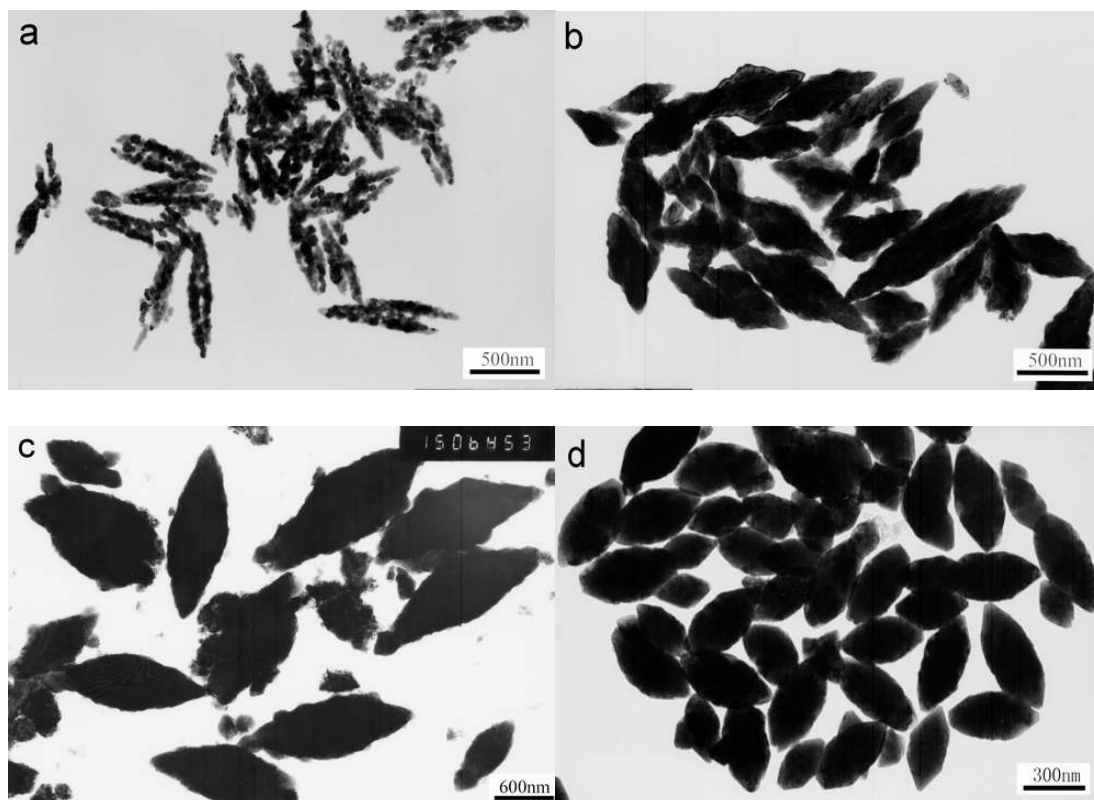


Figure S1. TEM images of PbWO_4 sample prepared under vigorous electric stirring in different P123 concentration systems: (a) $2 \text{ g}\cdot\text{L}^{-1}$, (b) $3 \text{ g}\cdot\text{L}^{-1}$, (c) $4 \text{ g}\cdot\text{L}^{-1}$, and (d) $5 \text{ g}\cdot\text{L}^{-1}$.

2. Effect of sonication time on the final products

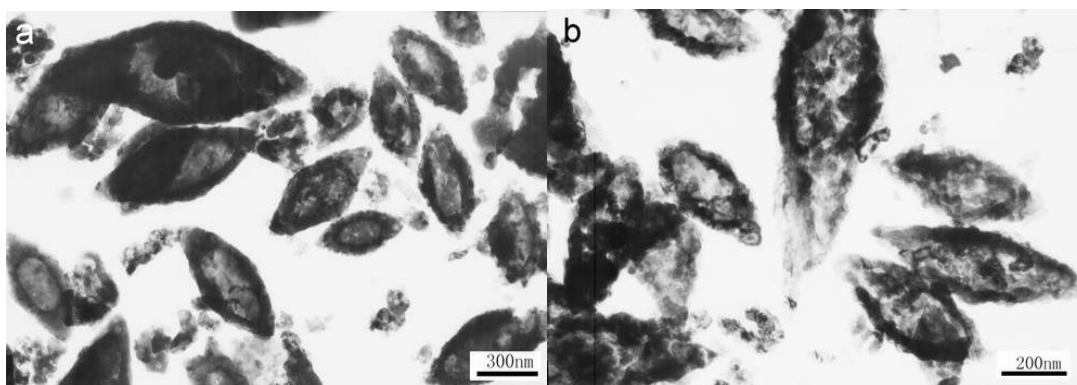


Figure S2. TEM images of samples obtained after sonication for (a) 60 min, and (b) 120 min.

3. Effect of concentrations of $\text{Pb}(\text{CH}_3\text{COO})_2$ and Na_2WO_4

The experimental results show that the uniform hollow spindles could only be obtained at the concentration of 0.5-1 M. When the concentration of $\text{Pb}(\text{CH}_3\text{COO})_2$ and Na_2WO_4 were changed to 0.1 M (1/5 of optimal concentration), the product was a mixture of hollow spindles and nanorods, and the yield was much lower (Figure S3a). When the concentration of $\text{Pb}(\text{CH}_3\text{COO})_2$ and Na_2WO_4 were changed to 2 M (4 times of optimal concentration), the morphology changed to be small nanoparticles together with irregular aggregation (Figure S3b). We think when the concentrations of $\text{Pb}(\text{CH}_3\text{COO})_2$ and Na_2WO_4 were relatively low (the P123 concentration was relatively higher), the copolymer might also work as capping reagent to restrict the growth of PbWO_4 crystals resulting the formation of nanorods. While under higher reactant concentration, nucleation occurred at an outburst speed right after the reaction solutions were mixed, leading to a large quantity of PbWO_4 precipitation, so small nanoparticles with their aggregation were obtained under this situation.

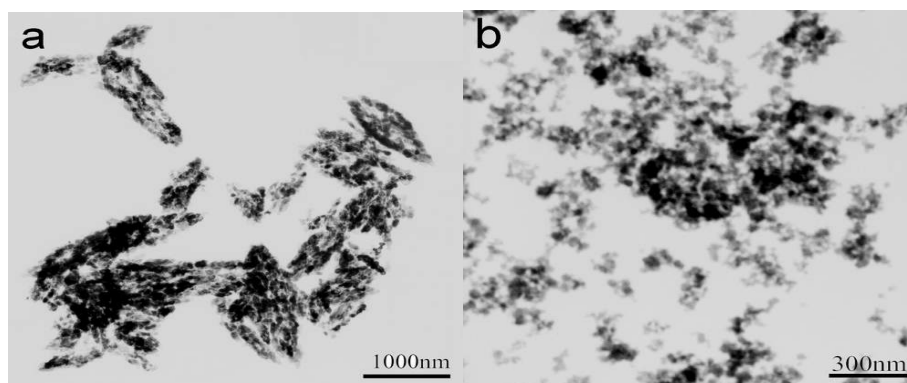


Figure S3. TEM images of samples obtained with the concentration of $\text{Pb}(\text{CH}_3\text{COO})_2$ and Na_2WO_4 of (a) 0.1 M, and (b) 2 M.