Supporting Information for the Manuscript entitled Sterically Hindered Aluminum Alkyls: Practical Scavenging Agents of Use in Olefin Polymerization

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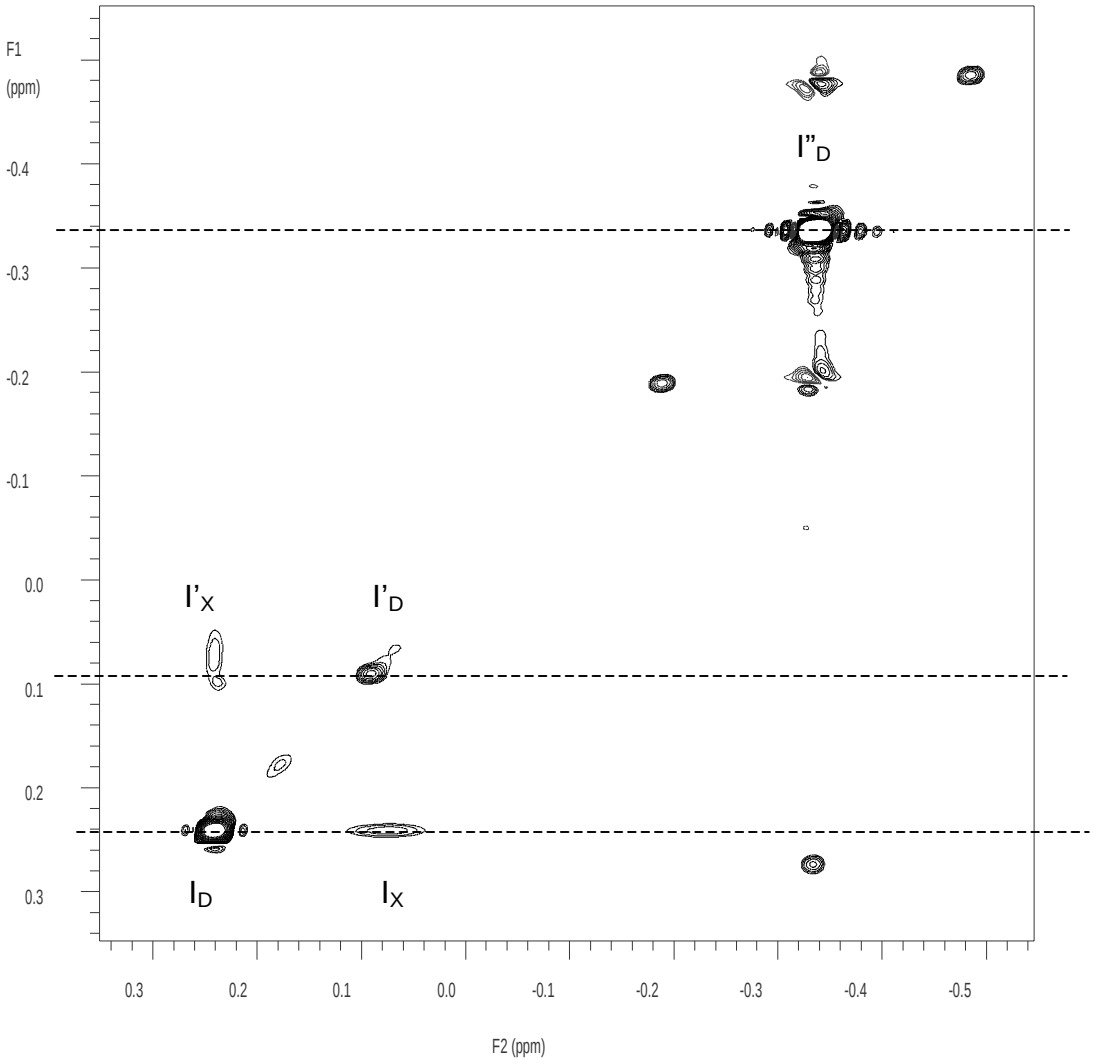
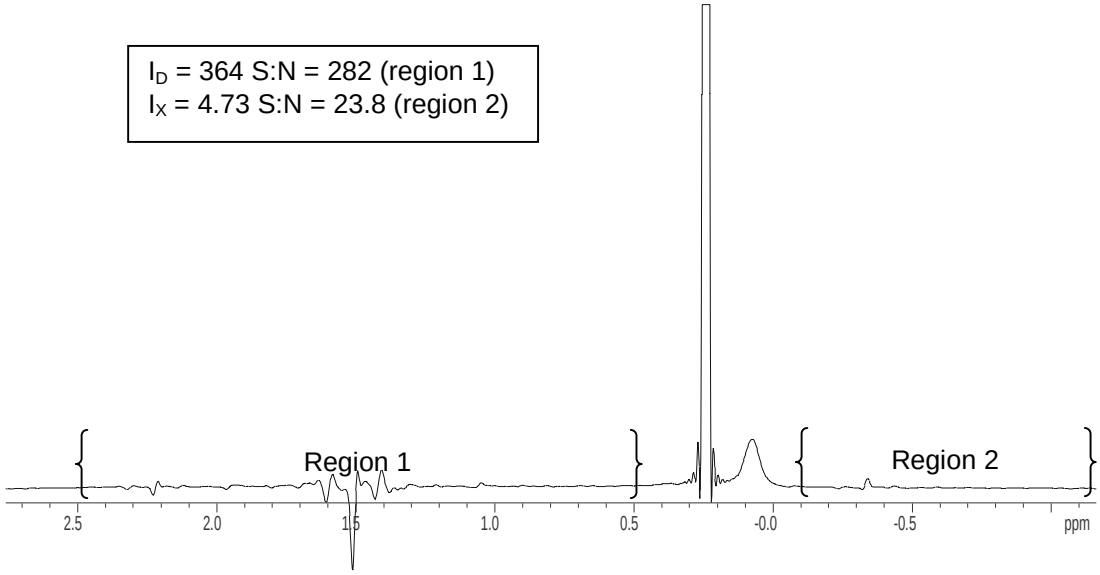
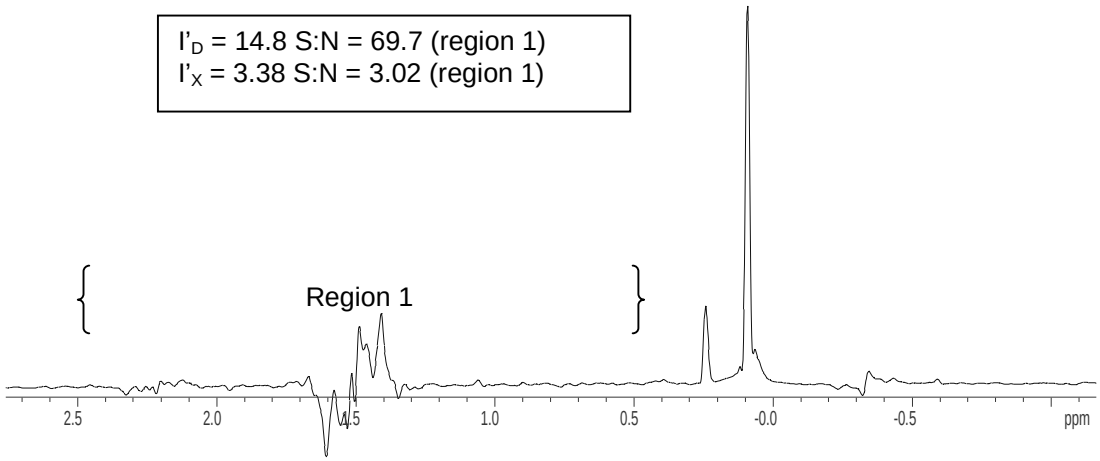
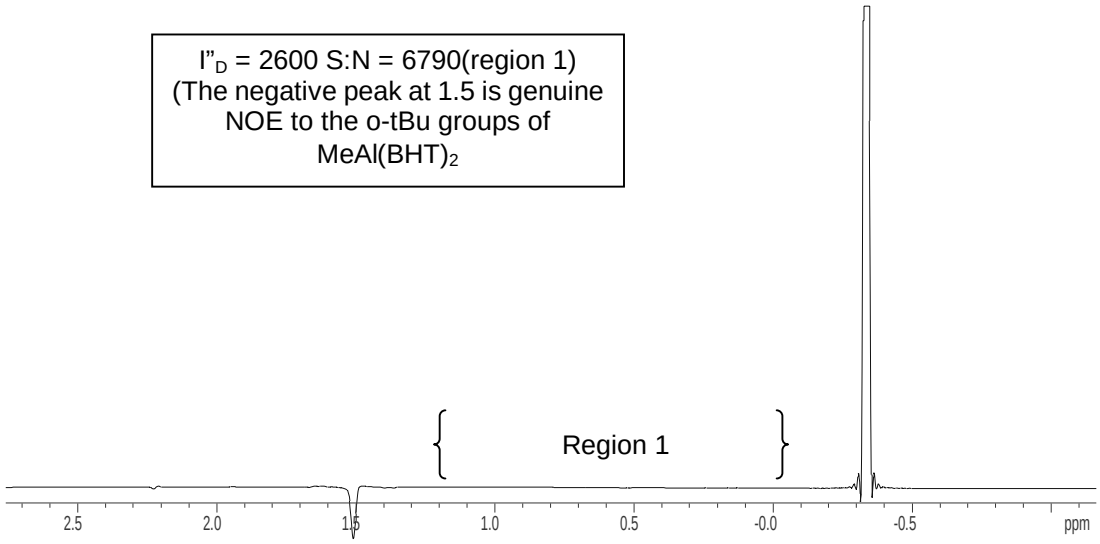
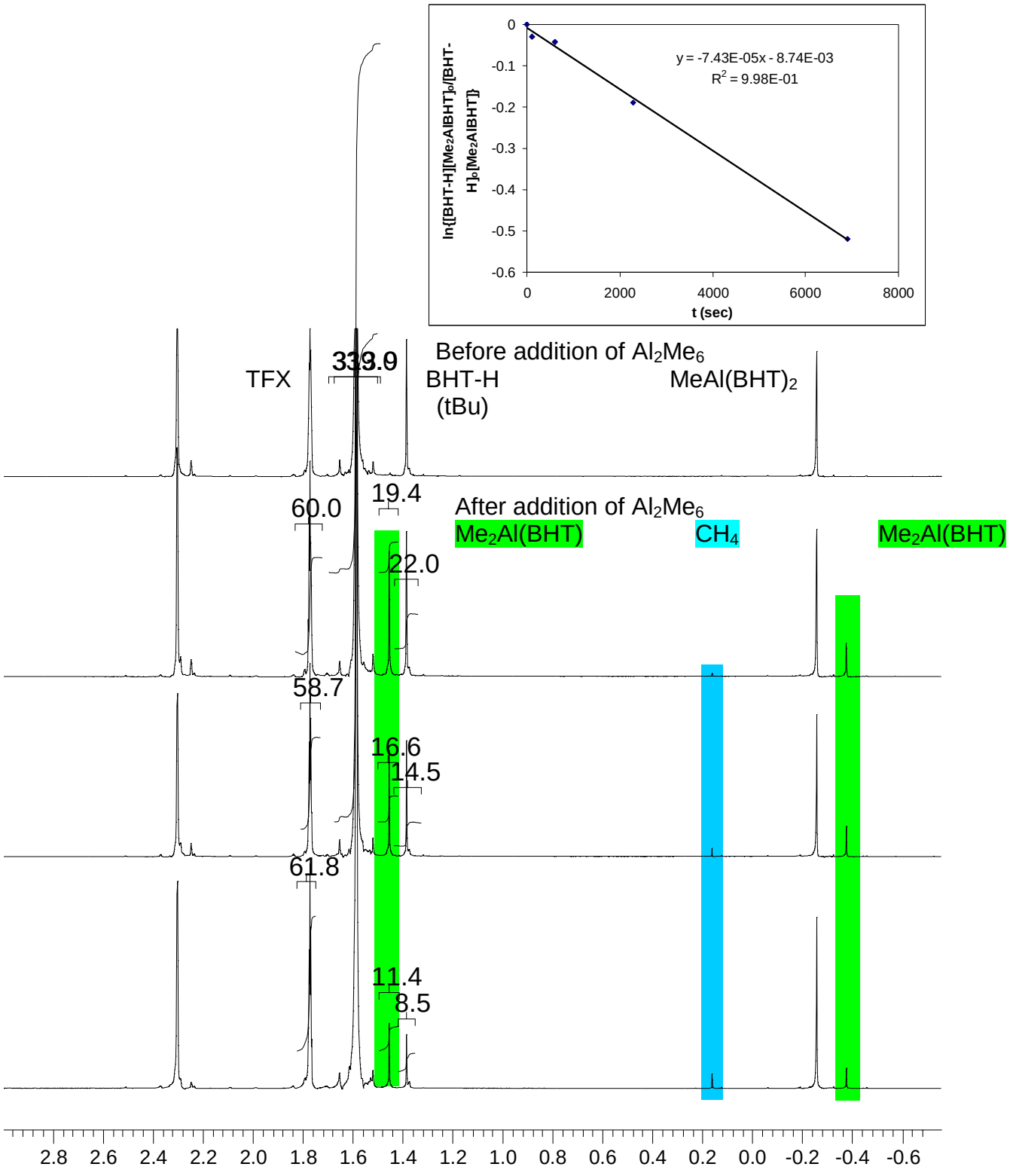
Item Page #

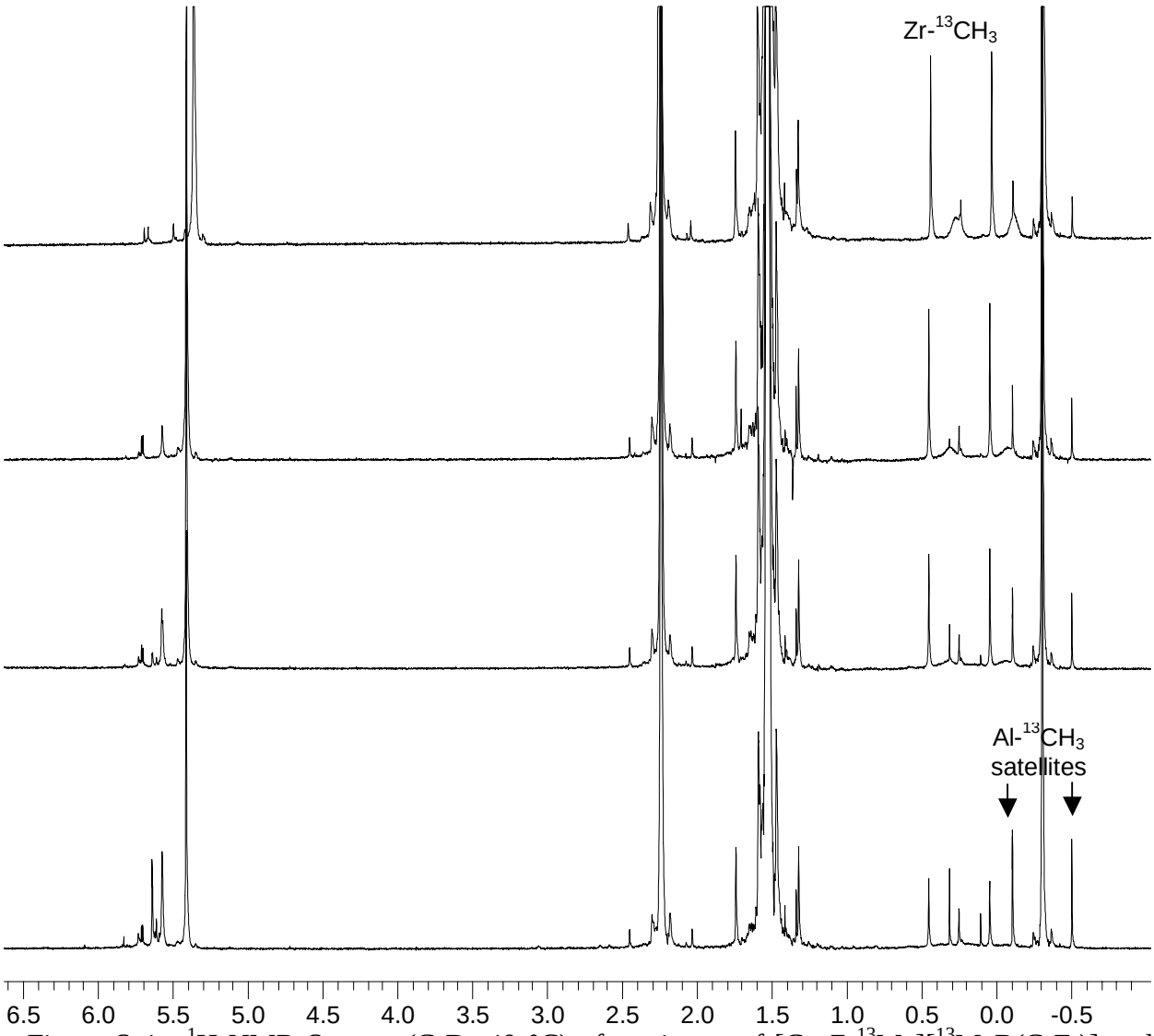

F1
(ppm)
-0.4
-0.3
-0.2
-0.1
0.0
0.1
0.2
0.3
I"D
I'X
I'D
ID
IX
0.3
0.2
0.1
0.0
-0.1
-0.2
-0.3
-0.4
-0.5
F2 (ppm)


I_D = 364 S:N = 282 (region 1)
I_X = 4.73 S:N = 23.8 (region 2)
Region 1
Region 2
2.5
2.0
1.5
1.0
0.5
-0.0
-0.5
ppm


I'_D = 14.8 S:N = 69.7 (region 1)
I'_X = 3.38 S:N = 3.02 (region 1)
Region 1
2.5
2.0
1.5
1.0
0.5
-0.0
-0.5
ppm


I''_D = 2600 S:N = 6790(region 1)
(The negative peak at 1.5 is genuine NOE to the o-tBu groups of $MeAl(BHT)_2$
Region 1
2.5
2.0
1.5
1.0
0.5
-0.0
-0.5
ppm


y = -7.43E-05x - 8.74E-03
R2 = 9.98E-01
ln{[BHT-H][Me2AlBHT]o/[BHT-H]o[Me2AlBHT]}
t (sec)
Before addition of Al2Me6
TFX
BHT-H
(tBu)
MeAl(BHT)2
After addition of Al2Me6
Me2Al(BHT)
CH4
Me2Al(BHT)
60.0
19.4
22.0
58.7
16.6
14.5
61.8
11.4
8.5



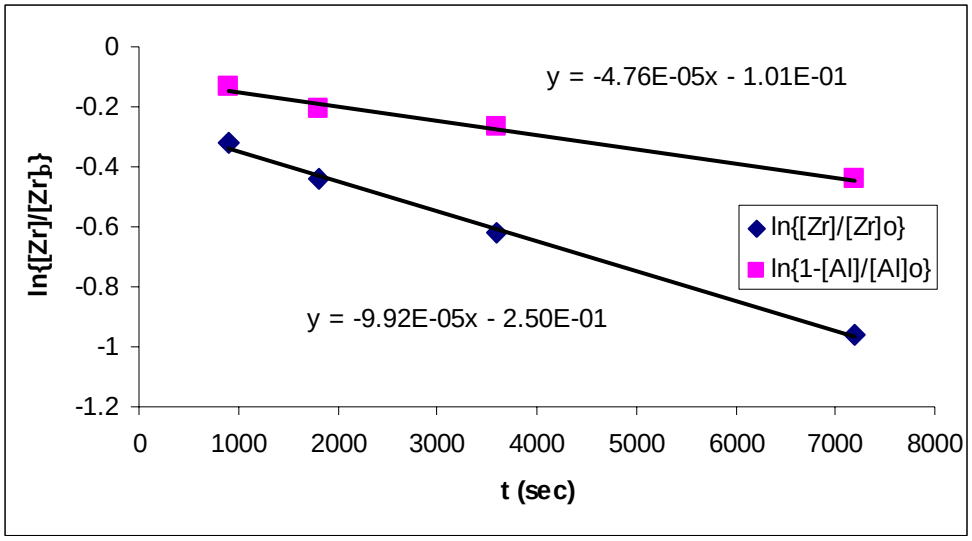


Figure S-4 - ^{1}H NMR Spectra (C_6D_6 40 °C) of a mixture of [$Cp_2Zr^{13}Me$][$^{13}MeB(C_6F_5)$] and $MeAl(BHT)_2$ as a function of time. Plot of the rate of disappearance of Zr-$^{13}CH_3$ and appearance of Al-^{13}C satellites as a function of time.

1. Chen, M.-C.; Roberts J. A. S.; Marks T. J. *J. Am. Chem. Soc.* **2004**, *126*, 4605-25.