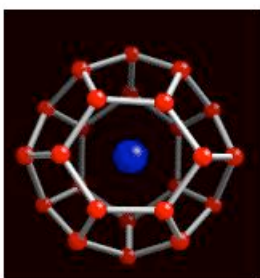
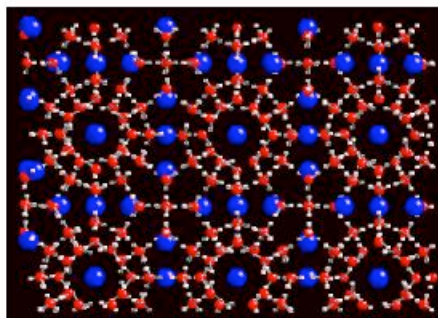
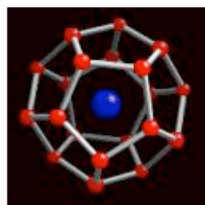


Supporting Information

Methane hydrate structure sl:

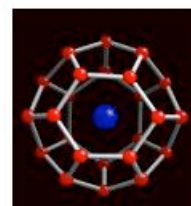
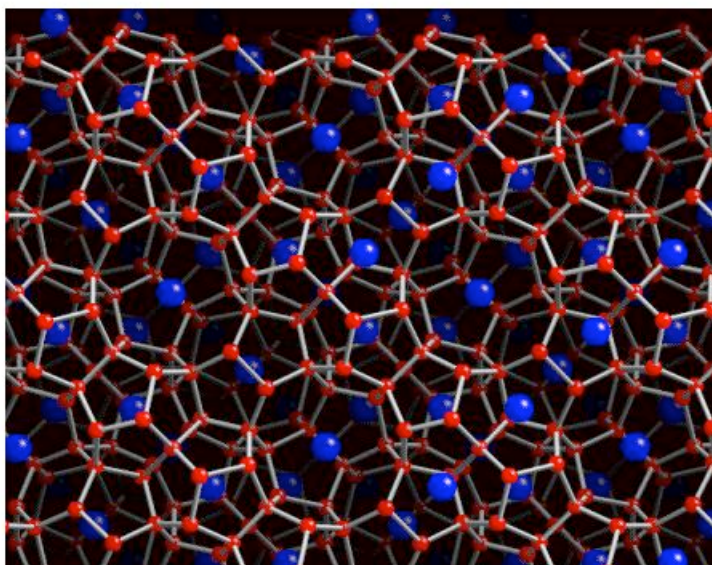


large (5¹²6²) cages

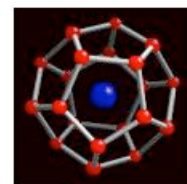


small (5¹²) cages

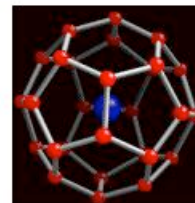
Methane hydrate structure sII:



(5¹²6²) cages

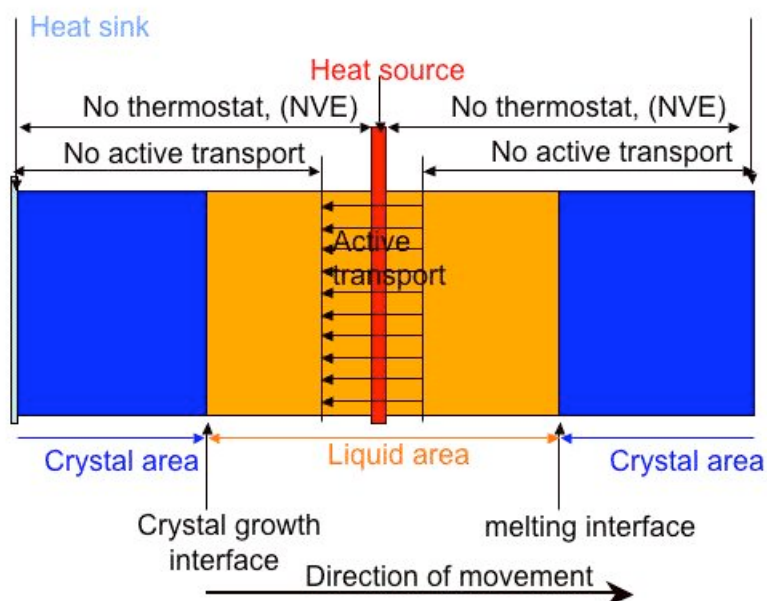


(5¹²) cages

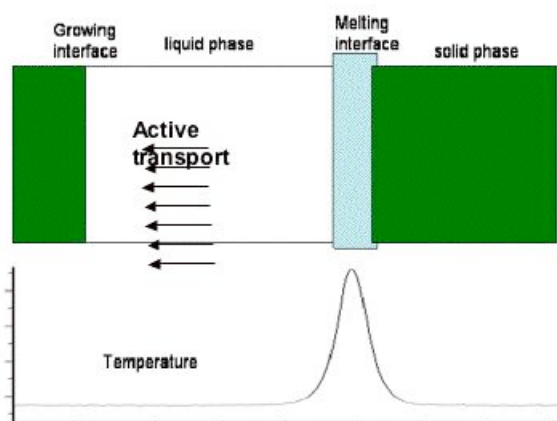


(5¹²6⁴) cages

MD setup 1: with local thermostats and temperature gradient. For details see Gulam Razul et al., *Mol. Phys.*, **2005**, 103, 1929.



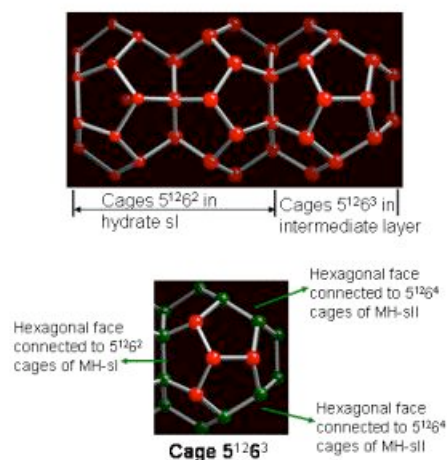
MD setup 2: without temperature gradients. For details see Vatananu J., Kusalik P. G., *submitted, J. Chem. Phys.*



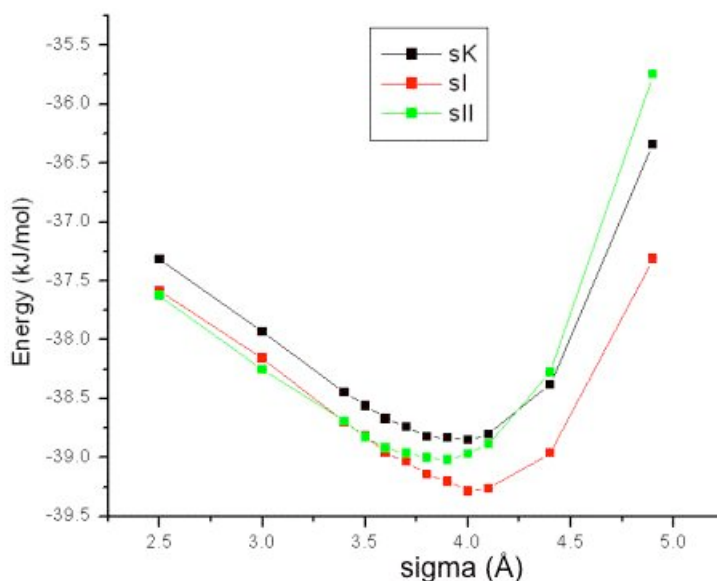
It is sufficient here to note that the measured interfacial properties, as well as the observed growth rates are independent of the choice of setup 1 or 2 (within our uncertainties). Here we have chosen to use setup 2 as it affords a superior environment from which to observe growth of hydrate crystals.

A configuration showing the $5^{12}6^2$ cages of MH-sl connecting to an intermediate layer consisting of $5^{12}6^3$ cages. The $5^{12}6^3$ cages provide the right geometry to connect MH sl and sII.

This image clarifies how it is possible to form a polycrystal consisting of MH-sl and MH-sII [as reported in Vatamanu J., and Kusalik P. G., *J. Phys. Chem. B.*, **110**, 15896 (2006)].



The total potential energies for different model hydrate crystal structures versus the LJ sigma of the guest molecule at 100 atm and 245 K. MH-sl (red line), MH-sII (green line) and MH-sK (black line). We point out that for molecular sizes bigger than 4.1 Å, the newly identified MH-sK appears energetically preferred to MH-sII.



Coordinates of the 94 molecules (80 water and 14 methane) making up a repeating unit.

The dimensions of the orthorhombic box are (12.116, 12.116, 20.824). All units in Å.

O	0.7558	-4.0752	-7.3860	O	0.7861	1.1235	9.3082	O	-5.4286	5.6631	5.1313
O	-0.6294	1.8843	-9.1690	O	0.7436	-1.7036	9.4106	O	-0.8247	3.4836	1.2168
O	0.6861	4.2881	-0.9607	O	-3.0076	-4.0594	-9.9690	O	2.9037	-1.8399	1.4455
O	-5.2974	-2.6406	-9.2702	O	-0.6934	-0.2909	-5.3110	O	0.7181	-2.6842	2.9490
O	-0.7387	-4.1916	-3.1946	O	-3.0622	-2.8302	-2.4230	O	5.1141	-2.6598	2.8988
O	5.2354	1.1253	9.4328	O	0.7156	-4.9541	-0.9667	O	2.9247	1.0040	1.4930
O	-5.3023	3.3530	-3.2032	O	5.2012	-2.7164	-4.9696	O	-3.1520	2.1229	0.4170
O	-5.4002	-4.1957	-3.1064	O	-0.8226	3.4228	-3.1238	O	-0.8723	5.7131	5.1012
O	0.7565	-2.6053	-4.9159	O	-2.9550	5.5787	-8.6357	O	-3.2061	-0.3071	1.7277
O	-2.9863	3.1664	-9.9875	O	0.7863	2.0240	-4.8844	O	-3.0774	3.2576	8.0455
O	-1.7191	5.6527	-4.4633	O	5.1597	-1.7402	9.4459	O	-3.1827	-2.7274	0.3442
O	2.9163	-3.2474	-1.0249	O	-4.4768	5.6696	-4.4560	C	-3.0153	2.6317	-6.1764
O	0.8536	3.3879	-7.3905	O	0.6628	1.8503	2.9147	C	2.9708	-0.3499	-7.9132
O	5.0708	-5.0523	-1.0179	O	-5.4100	-2.5787	7.2495	C	-3.0623	-0.3489	9.4070
O	5.2964	3.3721	-7.3758	O	-0.8783	-0.4288	3.2804	C	-3.0239	-3.4080	-6.2063
O	-2.9868	-0.4121	-3.7693	O	-5.4651	3.4225	1.2333	C	5.9418	5.7024	9.4379
O	-5.2555	5.6526	-7.0828	O	-0.7571	1.9542	7.1597	C	5.9040	-0.3648	-1.0160
O	5.2316	4.1622	-1.0362	O	-5.4053	1.9401	7.1991	C	-0.0256	-0.3562	-0.9877
O	-3.0776	1.9392	-2.3261	O	0.5475	-4.0418	5.5161	C	2.9903	5.6890	-4.4324
O	2.9287	-3.4434	9.4615	O	-5.4470	-4.1865	1.1199	C	-3.1082	5.6712	-0.9739
O	-1.6170	-0.3622	-7.8980	O	0.7153	3.4602	5.3360	C	-3.1518	-3.3558	4.2208
O	5.3310	-4.1307	-7.4535	O	-0.8453	-2.4855	7.3013	C	2.9401	5.6721	2.4583
O	-4.4075	-0.3822	-7.9179	O	-5.4800	-0.3657	3.3403	C	-3.1311	2.6835	4.2013
O	2.9694	2.8719	9.3849	O	-3.1250	5.7063	6.7105	C	2.9488	-0.3635	5.9360
O	3.0337	1.1457	-3.4558	O	5.2004	-4.0786	5.4150	C	0.0300	5.7054	9.3863
O	3.0379	-1.7667	-3.5056	O	-3.1248	-3.9589	8.0765				
O	3.0508	4.2530	-8.8741	O	-1.7649	-0.2941	5.9026				
O	-5.3173	-0.4735	-5.3353	O	-0.8761	-4.0634	1.1697				
O	-5.3208	1.8650	-9.1797	O	-1.7540	5.7488	2.4921				
O	-0.7459	-2.6259	-9.2055	O	5.0789	1.9549	3.0035				
O	2.9701	-4.9722	-8.7993	O	-4.5424	5.6881	2.5396				
O	-0.6755	5.6703	-7.0424	O	2.8988	-4.7569	6.8604				
O	2.8947	2.6207	-0.9866	O	2.9565	4.3826	6.8457				
O	5.2763	1.8252	-4.9815								

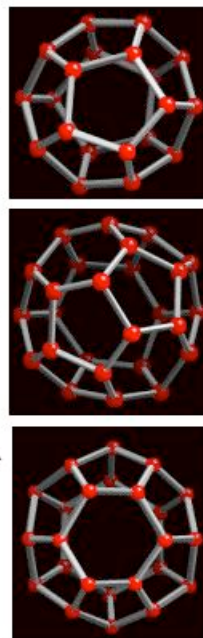
The new hydrate structure (sK) identified in this study.

14 Carbons
80 Oxygens

6 (5¹²) cages

4 'intermediate' (5¹²6³) cages

4 (5¹²6²) cages (as in sl hydrate)



[001] face of sK hydrate

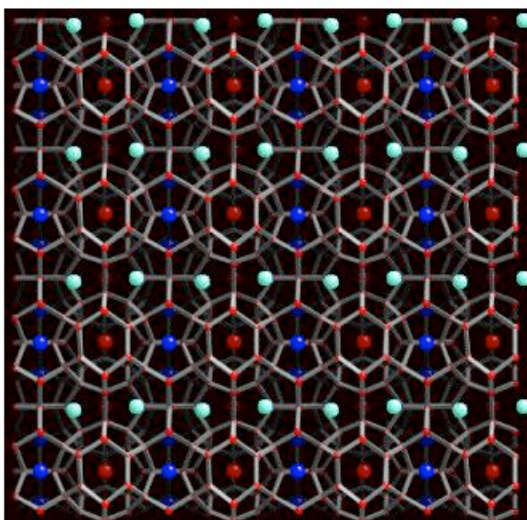
Oxygen atoms are shown as small connected red spheres.

The centers of each cage are labeled as follows:

blue: small 5^{12} cage

aqua: $5^{12}6^2$ cages

red: intermediate $5^{12}6^3$ cages



Note: depth shading has been applied

[010] face of sK hydrate

Oxygen atoms are shown as small connected red spheres.

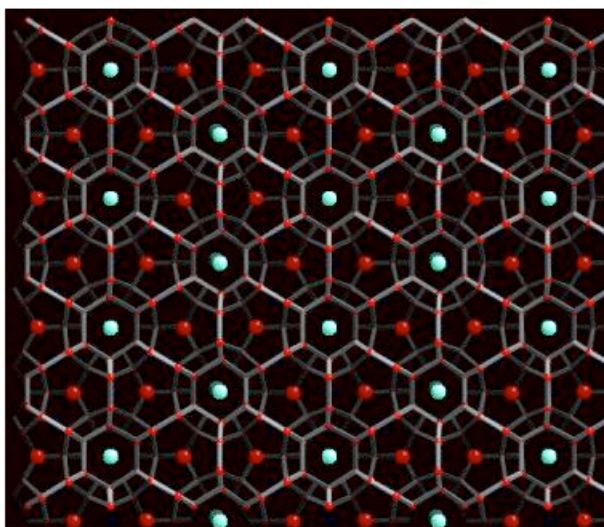
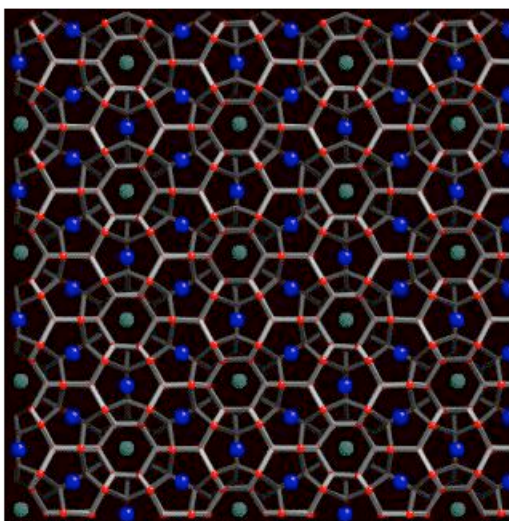
The centers of each cage are labeled as follows:

blue: small 5^{12} cage

aqua: $5^{12}6^2$ cages

red: intermediate $5^{12}6^3$ cages

Note: depth shading has been applied



The hexagonal symmetry of the crystal is apparent from these images.