## Supporting Information

# Facile Synthesis of Highly Biocompatible Poly(2methacryloyloxyethyl phosphorylcholine)-Coated Gold Nanoparticles 

Jian-Jun Yuan, Andreas Schmid and Steven P. Armes *<br>Department of Chemistry, The University of Sheffield, Brook Hill, Sheffield S3 7HF, United Kingdom

Andrew L. Lewis

Biocompatibles Ltd., Chapman House, Farnham Business Park, Weydon Lane, Farnham, Surrey, GU9 8QL, UK.
email: s.p.armes@sheffield.ac.uk


Figure S1. Particle size distribution of gold nanoparticles synthesized using PDMA $_{102}$ homopolymer (calculated by counting approximately 200 particles from TEM images such as that shown in Figure 2B). The synthesis conditions are the same as those described in the caption for Figure 2.


Figure S2. Gold nanoparticles synthesized using the $\mathrm{PEO}_{45}-\mathrm{PDMA}_{30}$ block copolymer: (A) a representative TEM image; (B) UV-visible absorption spectrum. The nanoparticles were prepared by mixing 0.50 mL of an aqueous solution of the copolymer ([DMA] $=11.1 \times 10^{-5}$ moles) with 3.0 mL of an aqueous solution of 4 mM $\mathrm{HAuCl}_{4}$. This reaction mixture was stirred for 24 h at $20^{\circ} \mathrm{C}$. The $[\mathrm{DMA}] /\left[\mathrm{HAuCl}_{4}\right]$ molar ratio was 9.25 and $\left[\mathrm{HAuCl}_{4}\right]=1.2 \times 10^{-5}$ moles.



Figure S3. Aggregation of gold nanoparticles at around their isoelectric point (approximately pH 8.3 ). The gold sol was synthesized using $\mathrm{PDMA}_{102}$ homopolymer using the same conditions as those described in the caption for Figure 2: (A) UVvisible absorption spectra recorded for the gold sol at $\mathrm{pH} 7.0,8.3$ and 11.2; (B) is a representative TEM image depicting the aggregated particles obtained at pH 8.3. The inset shows the primary particles at higher magnification.


Figure S4. FT-IR spectra recorded for: (a) the $\mathrm{PMPC}_{30}-\mathrm{PDMA}_{30}$ diblock copolymerstabilized gold nanoparticles (b) the $\mathrm{PMPC}_{30}-\mathrm{PDMA}_{30}$ diblock copolymer. The arrow indicates the weak ester carbonyl stretch due to the minor copolymer component.


Figure S5. Particle size distributions obtained for the gold nanoparticles shown in (a) Figure 2C, (b) Figure 4A, (c) Figure 4B and (d) Figure 4C.


Figure S6. UV-visible absorption spectra recorded for gold sols synthesized using the $\mathrm{PDMA}_{102}$ homopolymer with varying [DMA]:[HAuCl $]$ molar ratios at $20^{\circ} \mathrm{C}$ for 24 h with continuous stirring. These sols were synthesized by mixing 0.50 mL of a 1.0 wt. \% aqueous solution of $\mathrm{PDMA}_{102}$ with [DMA]:[ $\mathrm{HAuCl}_{4}$ ] molar ratios of 79.5, 39.8, $31.8,26.5,19.9$ and 13.3, respectively.

