

## Supporting Information

### **Cyanide-Bridged Mn(III)-Fe(III) Bimetallic Complexes Based on the Pentacyano(1-methylimidazole)ferrate(III) Building Block: Structure and Magnetic Characterizations**

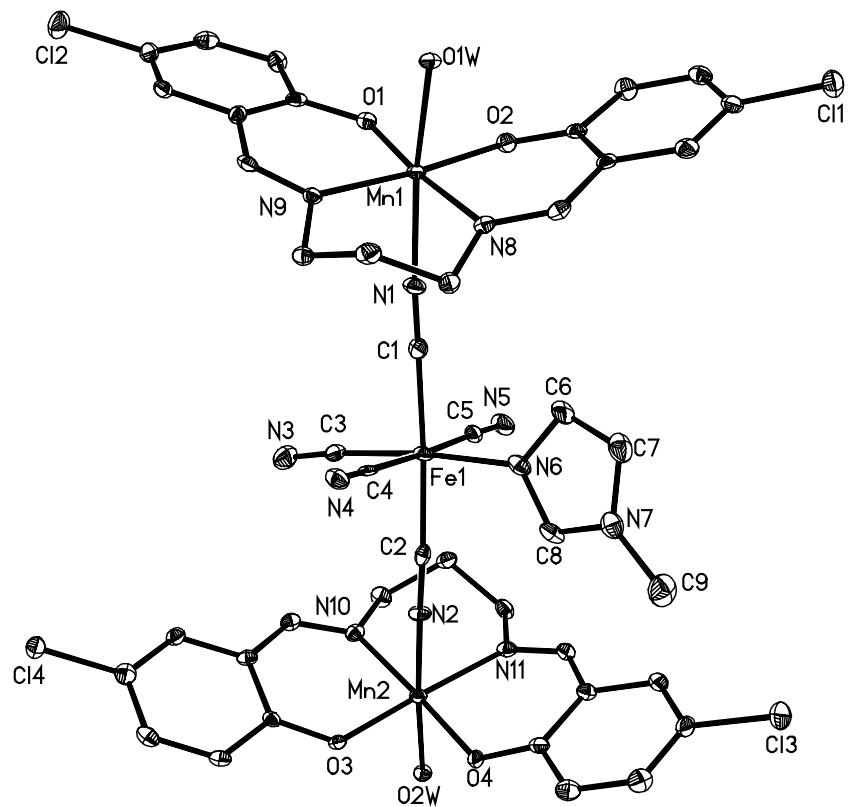
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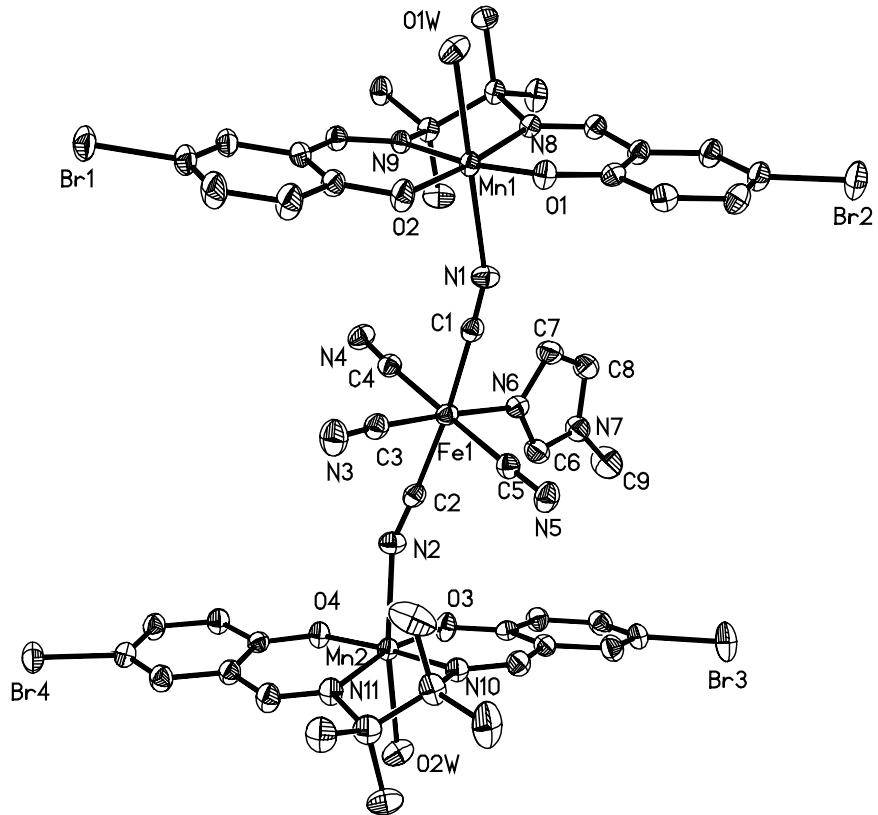
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**Table S1.** Selected Bond Distances ( $\text{\AA}$ ) and Angles (deg) of Complex **3**.

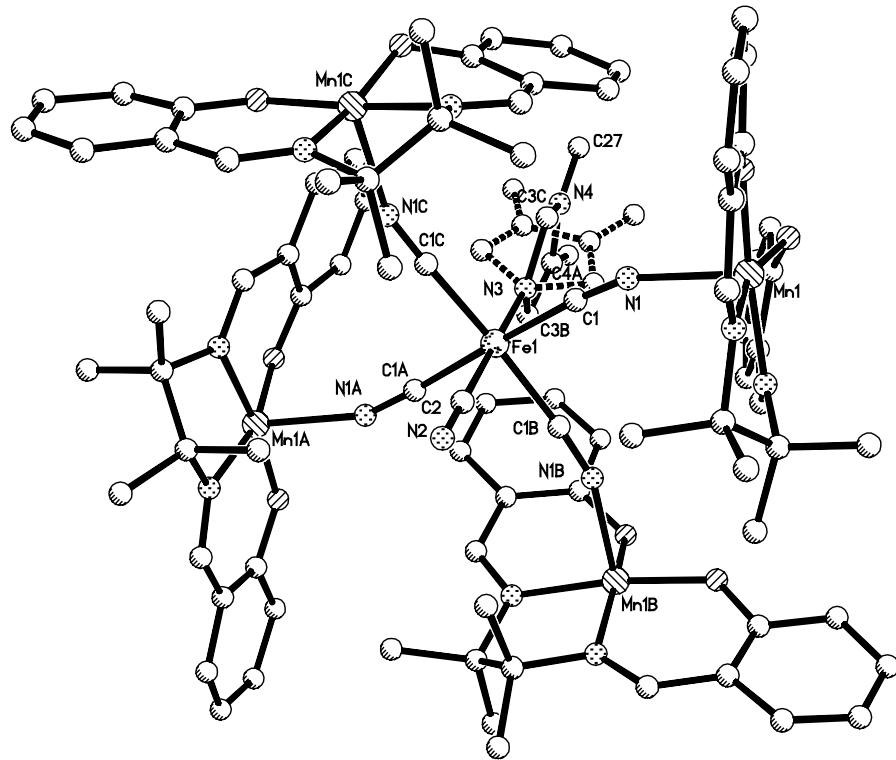
Fe(1)-C(1)	1.946(6)	Fe(1)-C(2)	1.920(6)
Fe(1)-C(3)	1.921(7)	Fe(1)-C(4)	1.944(6)
Fe(1)-C(5)	1.945(7)	Fe(1)-N(6)	2.010(6)
Fe(2)-C(43)	1.921(7)	Fe(2)-C(44)	1.937(6)
Fe(2)-C(45)	1.969(6)	Fe(2)-C(46)	1.956(6)
Fe(2)-C(47)	1.947(6)	Fe(2)-N(16)	1.990(5)
Mn(1)-N(1)	2.221(5)	Mn(1)-O(1W)	2.246(4)
Mn(1)-O(1)	1.908(4)	Mn(1)-N(8)	2.098(5)
Mn(1)-O(2)	1.907(4)	Mn(1)-N(9)	2.042(5)
Mn(2)-N(2)	2.249(5)	Mn(2)-O(2W)	2.233(4)
Mn(2)-O(3)	1.883(4)	Mn(2)-N(10)	2.051(5)
Mn(2)-O(4)	1.918(4)	Mn(2)-N(11)	2.059(5)
Mn(3)-N(13)	2.229(5)	Mn(3)-O(3W)	2.234(4)
Mn(3)-O(5)	1.907(4)	Mn(3)-N(18)	2.049(5)
Mn(3)-O(6)	1.907(4)	Mn(3)-N(19)	2.070(5)
Mn(4)-N(12)	2.298(5)	Mn(4)-O(4W)	2.222(4)
Mn(4)-O(7)	1.875(4)	Mn(4)-N(20)	2.029(5)
Mn(4)-O(8)	1.892(4)	Mn(4)-N(21)	2.050(5)
Mn(1)-N(1)-C(1)	165.2(6)	N(1)-C(1)-Fe(1)	176.8(6)
Mn(2)-N(2)-C(2)	156.9(5)	N(2)-C(2)-Fe(1)	174.8(6)
Mn(3)-N(13)-C(46)	152.8(5)	N(13)-C(46)-Fe(2)	174.5(6)
Mn(4)-N(12)-C(44)	147.8(5)	N(12)-C(44)-Fe(2)	177.1(6)



**Figure S1.** One of the two independent *trans*-trinuclear units in complex **3** with atom labeling scheme showing 30% probability thermal ellipsoids (hydrogen atoms are omitted for clarity).



**Figure S2.** Trinuclear structure of complex **5** with atom labeling scheme showing 30% probability thermal ellipsoids (hydrogen atoms and crystallized methanol molecules are omitted for clarity).



**Figure S3.** Structure of complex **7** (hydrogen atoms, perchlorate ions and water molecules are omitted for clarity). Dashed lines represent the disordered 1-methylimidazole groups. Selected bond distances ( $\text{\AA}$ ) and angles (deg): Fe(1)-C(1) 1.936(8); Fe(1)-C(2) 1.935(14); Fe(1)-N(3) 1.961(13); Mn(1)-N(1) 2.197(8); Mn(1)-O(1) 1.871(5); Mn(1)-O(2) 1.863(6); Mn(1)-N(5) 1.984(8); Mn(1)-N(6) 1.969(7); Fe(1)-C(1)-N(1) 176.0(7); C(1)-N(1)-Mn(1) 157.3(7).

Eq. 1:

$$\chi_t = \frac{Ng^2\beta^2}{kT} \left[ \frac{A}{B} \right] \quad (1)$$

in which  $A = 165 + 84\exp(-9J/kT) + 84\exp(-J/kT) + 35\exp(-8J/kT) + 35\exp(-2J/kT) + 10\exp(-7J/kT) + 10\exp(-3J/kT) + \exp(-6J/kT) + \exp(-4J/kT)$  and  $B = 20 + 16\exp(-9J/kT) + 16\exp(-J/kT) + 12\exp(-8J/kT) + 12\exp(-2J/kT) + 8\exp(-7J/kT) + 8\exp(-3J/kT) + 4\exp(-6J/kT) + 4\exp(-4J/kT)$ .

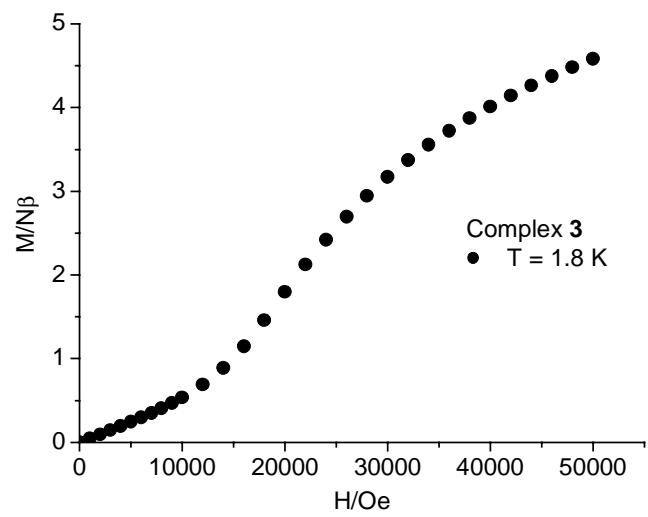
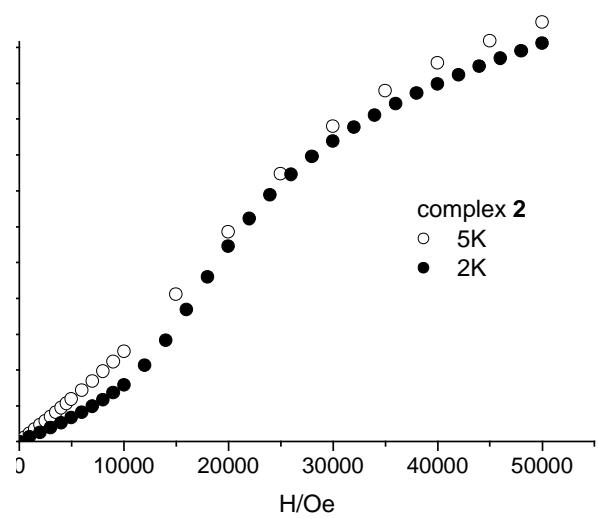
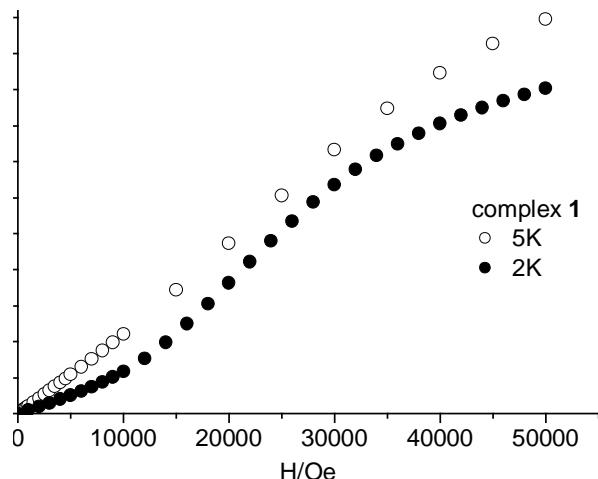
Eq. 2:

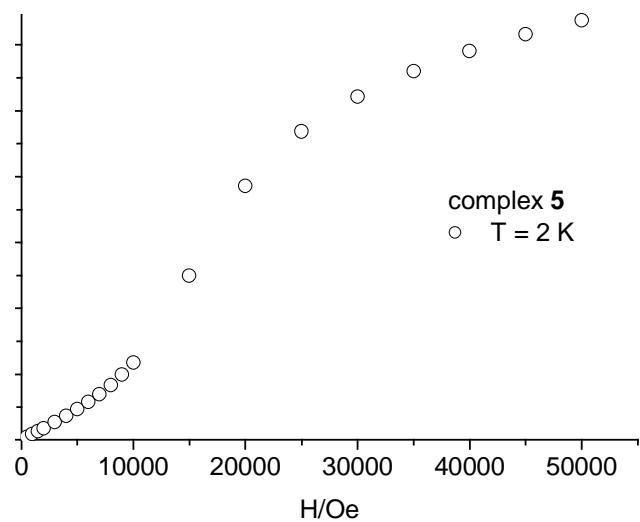
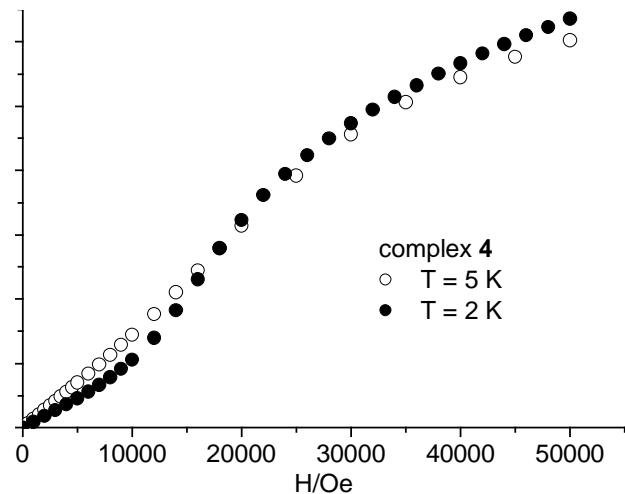
$$\chi_m = \frac{\chi_t}{1 - \chi_t(2zJ' / Ng^2\beta^2)} \quad (2)$$

Eq. 3:

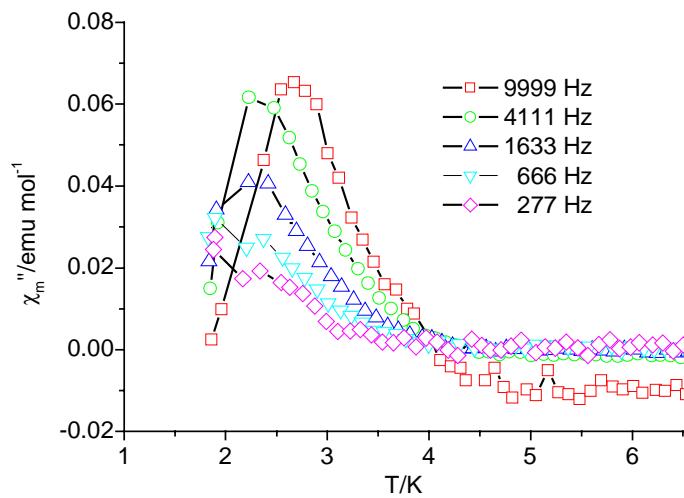
$$\chi_m = \frac{Ng^2\beta^2}{kT} \left[ \frac{A}{B} \right] \quad (3)$$

in which  $A = 1 + 9\exp(2D/kT) + 25\exp(6D/kT) + 49\exp(12D/kT) + 81\exp(20D/kT) + 84\exp(-9J/kT) + 84\exp(-J/kT) + 35\exp(-8J/kT) + 35\exp(-2J/kT) + 10\exp(-7J/kT) + 10\exp(-3J/kT) + \exp(-6J/kT) + \exp(-4J/kT)$  and  $B = 4 + 4\exp(2D/kT) + 4\exp(6D/kT) + 4\exp(12D/kT) + 4\exp(20D/kT) + 16\exp(-9J/kT) + 16\exp(-J/kT) + 12\exp(-8J/kT) + 12\exp(-2J/kT) + 8\exp(-7J/kT) + 8\exp(-3J/kT) + 4\exp(-6J/kT) + 4\exp(-4J/kT)$ .





**Figure S4.** Field dependence of magnetization for the trinuclear complexes.



**Figure S5.** Out-of-phase component of the zero-static ( $H_{DC} = 0$ ) AC susceptibility at different frequencies for **5**.