

Mechanistic studies on the formation of η^2 -diphosphine (η^6 -*p*-cymene)ruthenium(II) compounds

Adrian B. Chaplin, Céline Fellay, Gábor Laurenczy and Paul J. Dyson*

Institut des Sciences et Ingénierie Chimiques, Ecole Polytechnique Fédérale de Lausanne (EPFL), CH-1015 Lausanne, Switzerland. E-mail: paul.dyson@epfl.ch

SUPPLEMENTARY MATERIAL

1. Simulation analysis of the ^{31}P -NMR spectrum of **[2c]PF₆** pg. S2
2. Solid state structures of **[2a]PF₆** and **[2d]PF₆**, pg. S3
3. Kinetic studies of **4a** and **4b** pg. S4
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1. Simulation analysis of the ^{31}P -NMR spectrum of [2c]PF₆

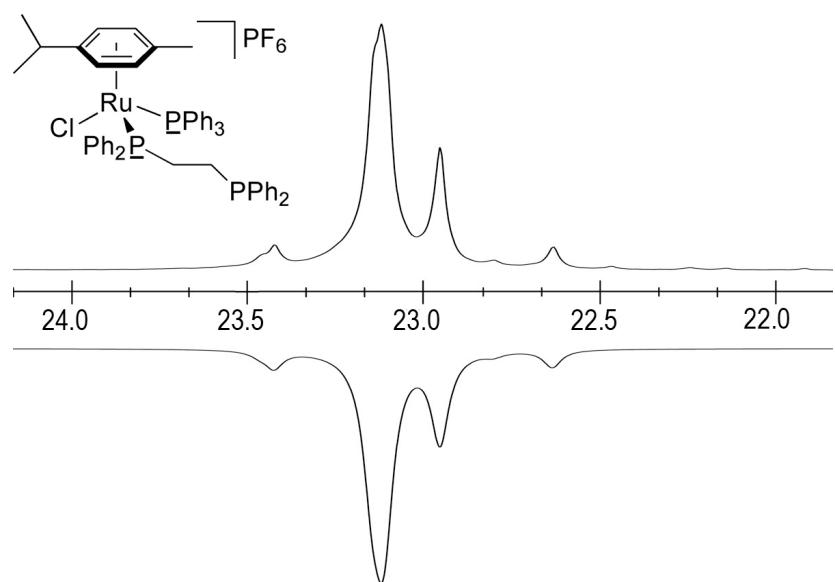


Figure S1: Observed (above) and simulated (below) ^{31}P -NMR spectrum of [2c]PF₆.

2. Solid state structures of [2a]PF₆ and [2d]PF₆

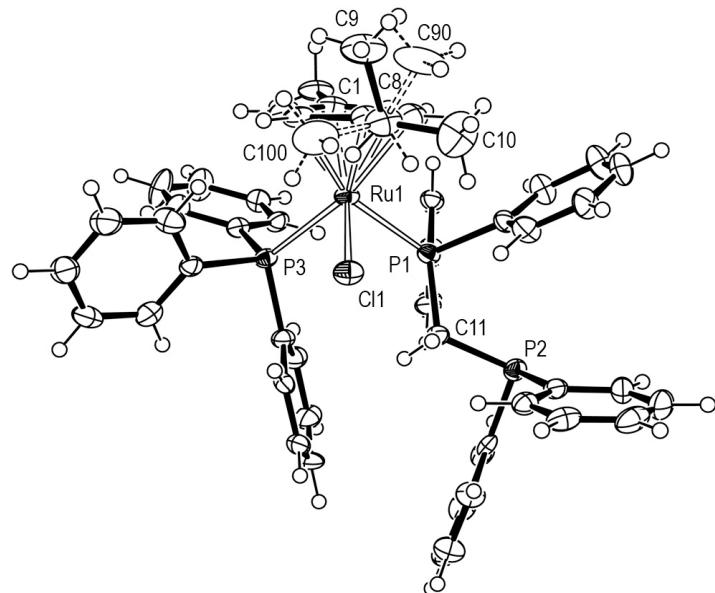


Figure S2: ORTEP representation of [2a]PF₆ thermal ellipsoids are drawn at 50% probability level. Counter ion and solvent molecules omitted for clarity. The minor disordered component is shown with dashed bonds (C90 and C100). Relevant bond parameters are given in Table 2 (manuscript).

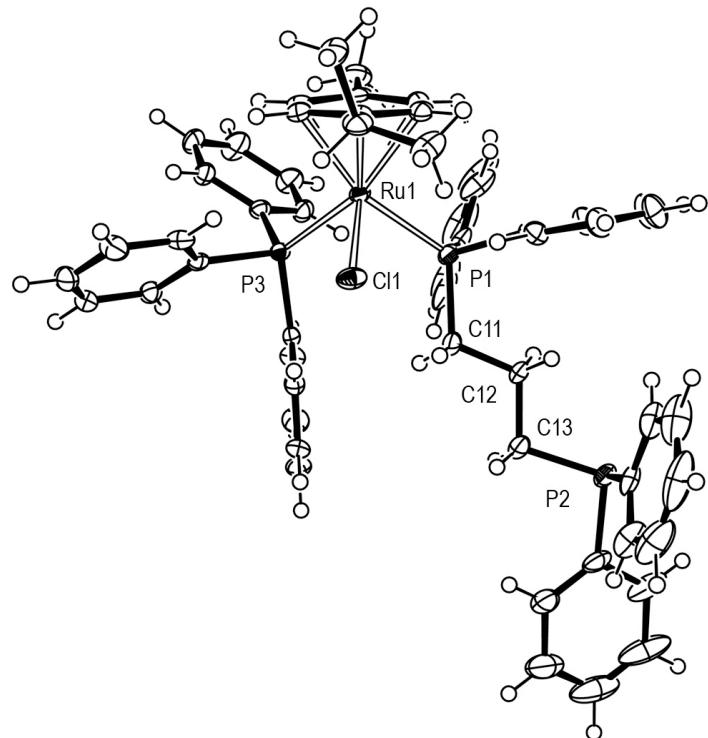


Figure S3: ORTEP representation of [2d]PF₆; thermal ellipsoids are drawn at 50% probability level. Counter ion and solvent molecules omitted for clarity. Relevant bond parameters are given in Table 2 (manuscript).

3. Kinetic studies of **4a** and **4b**

Table S1: Rate constants for chelation reactions of **4a** and **4b** at different temperatures.^a

		ligand	solvent (v/v)	T / K	$10^4 \cdot k_{\text{obs}} / \text{s}^{-1}$	R^2_{fit}
1	4a→[5a]Cl	dppm	1:1 MeOH-CH ₂ ClCH ₂ Cl	280.4	0.144 ± 0.006	0.998
2		dppm	1:1 MeOH-CH ₂ ClCH ₂ Cl	297.6	1.45 ± 0.08	0.997
3		dppm	1:1 MeOH-CH ₂ ClCH ₂ Cl	303.4	3.1 ± 0.2	0.994
4		dppm	1:1 MeOH-CH ₂ ClCH ₂ Cl	312.6	9.7 ± 1.3	0.987
5	4b→[5b]Cl	dppv	1:1 MeOH-CH ₂ ClCH ₂ Cl	265.2	0.113 ± 0.011	0.992
6		dppv	1:1 MeOH-CH ₂ ClCH ₂ Cl	281.9	0.90 ± 0.06	0.996
7		dppv	1:1 MeOH-CH ₂ ClCH ₂ Cl	297.6	6.4 ± 0.7	0.993
8		dppv	1:1 MeOH-CH ₂ ClCH ₂ Cl	303.4	11.4 ± 1.1	0.992
9		dppv	CH ₂ ClCH ₂ Cl	315.4	0.190 ± 0.014	0.999
10		dppv	CH ₂ ClCH ₂ Cl	326.2	0.69 ± 0.14	0.998
11		dppv	CH ₂ ClCH ₂ Cl	338.2	2.7 ± 0.2	0.997
12		dppv	CH ₂ ClCH ₂ Cl	349.2	7.2 ± 0.4	0.999

^a Errors quoted are the standard errors calculated from the exponential fitting.

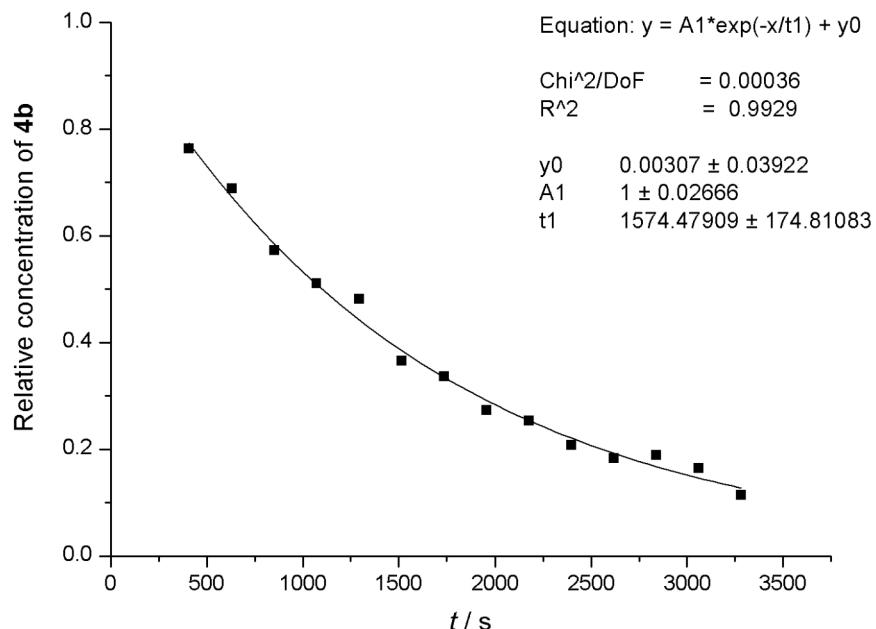


Figure S4: Typical exponential fit; corresponds to entry 7 in Table S1.

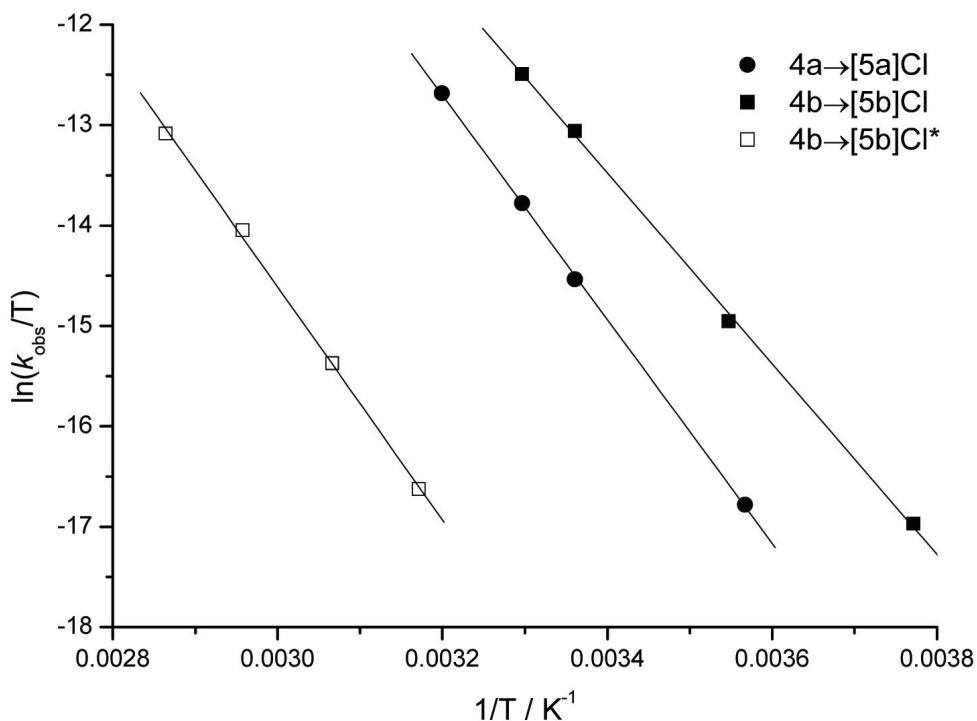


Figure S5: Eyring plot for chelation of reactions of **4a** ($R^2 = 1.000$) and **4b** ($R^2 = 0.999$) in 1:1 MeOH-CH₂ClCH₂Cl and **4b** ($R^2 = 0.999$) in CH₂ClCH₂Cl (indicated by asterisk).

Table S2: Rate constants for chelation reactions of **4a** and **4b** at different pressures.^a

		ligand	solvent (v/v)	P / bar	$10^4 \cdot k_{\text{obs}} / \text{s}^{-1}$	R^2_{fit}
1	4a → [5a]Cl	dppm	1:1 MeOH-CH ₂ ClCH ₂ Cl	1	1.353 ± 0.002	1.000
2		dppm	1:1 MeOH-CH ₂ ClCH ₂ Cl	504	1.310 ± 0.003	1.000
3		dppm	1:1 MeOH-CH ₂ ClCH ₂ Cl	1003	1.239 ± 0.003	1.000
4		dppm	1:1 MeOH-CH ₂ ClCH ₂ Cl	1507	1.150 ± 0.003	1.000
5	4b → [5b]Cl	dppv	1:1 MeOH-CH ₂ ClCH ₂ Cl	1	5.774 ± 0.009	1.000
6		dppv	1:1 MeOH-CH ₂ ClCH ₂ Cl	248	6.300 ± 0.012	1.000
7		dppv	1:1 MeOH-CH ₂ ClCH ₂ Cl	486	6.43 ± 0.02	1.000
8		dppv	1:1 MeOH-CH ₂ ClCH ₂ Cl	748	6.873 ± 0.012	1.000
9		dppv	1:1 MeOH-CH ₂ ClCH ₂ Cl	997	6.88 ± 0.03	1.000
10		dppv	1:1 MeOH-CH ₂ ClCH ₂ Cl	1250	7.18 ± 0.03	1.000

^a Errors quoted are the standard errors calculated from the exponential fitting.

4. Kinetic studies of [2a-e]PF₆

Table S3: Rate constants for chelation reactions of [2a-e]PF₆ at different temperatures.^a

		ligand	solvent	T / K	10 ⁴ ·k _{obs} / s ⁻¹	R ² _{fit}
1	[2a]PF ₆ →[5a]PF ₆	dppm	CH ₂ ClCH ₂ Cl	326.7	0.141 ± 0.005	0.998
2		dppm	CH ₂ ClCH ₂ Cl	331.6	0.240 ± 0.009	0.998
3		dppm ^b	CH ₂ ClCH ₂ Cl	343.4	0.93 ± 0.03	0.999
4		dppm	CH ₂ ClCH ₂ Cl	349.1	2.39 ± 0.03	0.998
5		dppm	CH ₂ ClCH ₂ Cl	353.0	3.7 ± 0.3	0.994
6	[2b]PF ₆ →[5b]PF ₆	dppv	CH ₂ ClCH ₂ Cl	327.2	0.147 ± 0.008	0.997
7		dppv	CH ₂ ClCH ₂ Cl	333.5	0.33 ± 0.02	0.997
8		dppv	CH ₂ ClCH ₂ Cl	343.7	1.13 ± 0.10	0.994
9		dppv	CH ₂ ClCH ₂ Cl	352.6	3.8 ± 0.3	0.997
10	[2c]PF ₆ →[5c]PF ₆	dppe	CH ₂ ClCH ₂ Cl	332.7	0.215 ± 0.009	0.999
11		dppe	CH ₂ ClCH ₂ Cl	343.5	1.15 ± 0.07	0.997
12		dppe	CH ₂ ClCH ₂ Cl	354.3	4.6 ± 0.2	0.998
13	[2d]PF ₆ →[5d]PF ₆	dppp	CH ₂ ClCH ₂ Cl	332.1	0.28 ± 0.02	0.998
14		dppp	CH ₂ ClCH ₂ Cl	343.1	1.31 ± 0.06	0.998
15		dppp	CH ₂ ClCH ₂ Cl	354.3	6.1 ± 0.2	0.999
16	[2e]PF ₆ →[5e]PF ₆	dppf	CH ₂ ClCH ₂ Cl	295.0	0.027 ± 0.007	0.999
17		dppf	CH ₂ ClCH ₂ Cl	308.9	0.31 ± 0.02	0.999
18		dppf	CH ₂ ClCH ₂ Cl	320.0	1.9 ± 0.2	0.996
19		dppf	CH ₂ ClCH ₂ Cl	332.5	9.5 ± 0.7	0.996

^a Errors quoted are the standard errors calculated from the exponential fitting. ^b Excluded from Erying fit.

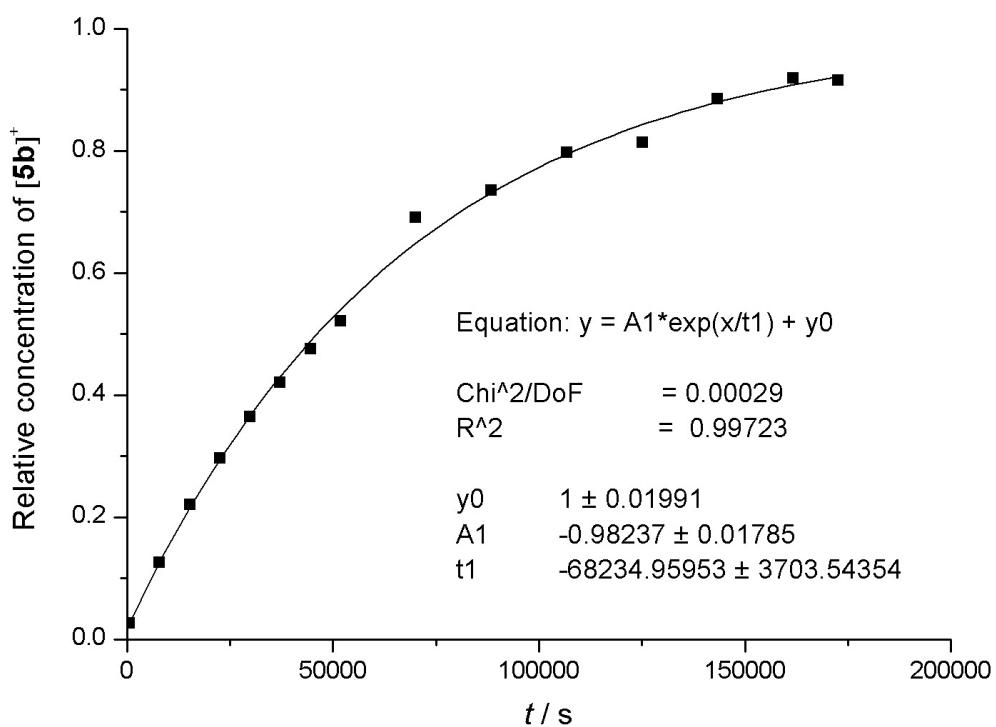


Figure S6: Typical exponential fit; corresponds to entry 6 in Table S3.

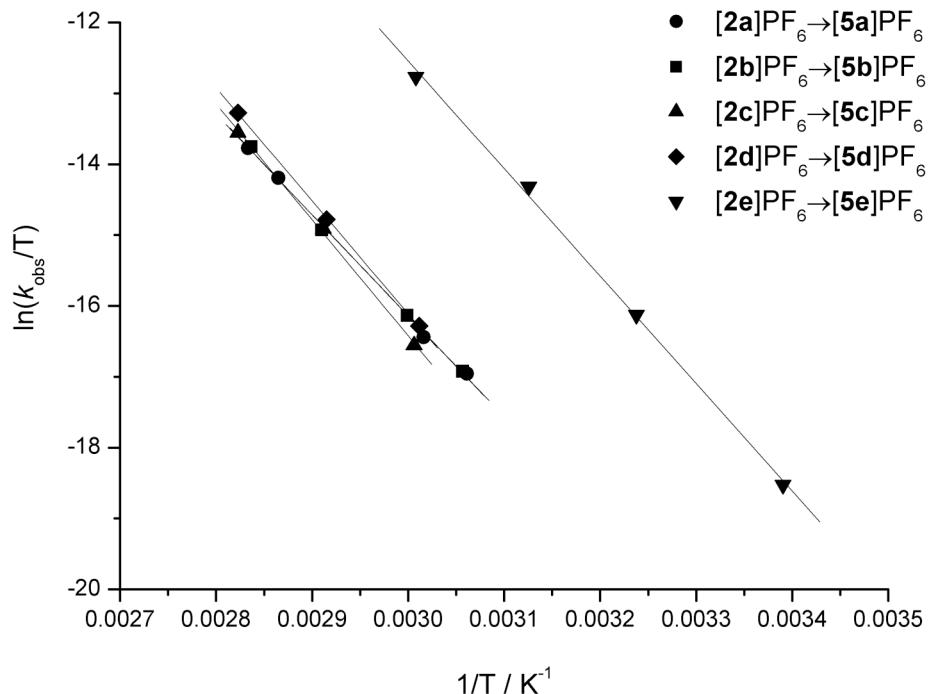


Figure S7: Eyring plot for the chelation of complexes $[2]\text{PF}_6$ in $\text{CH}_2\text{ClCH}_2\text{Cl}$ ($[2e]\text{PF}_6 R^2 = 0.998$).

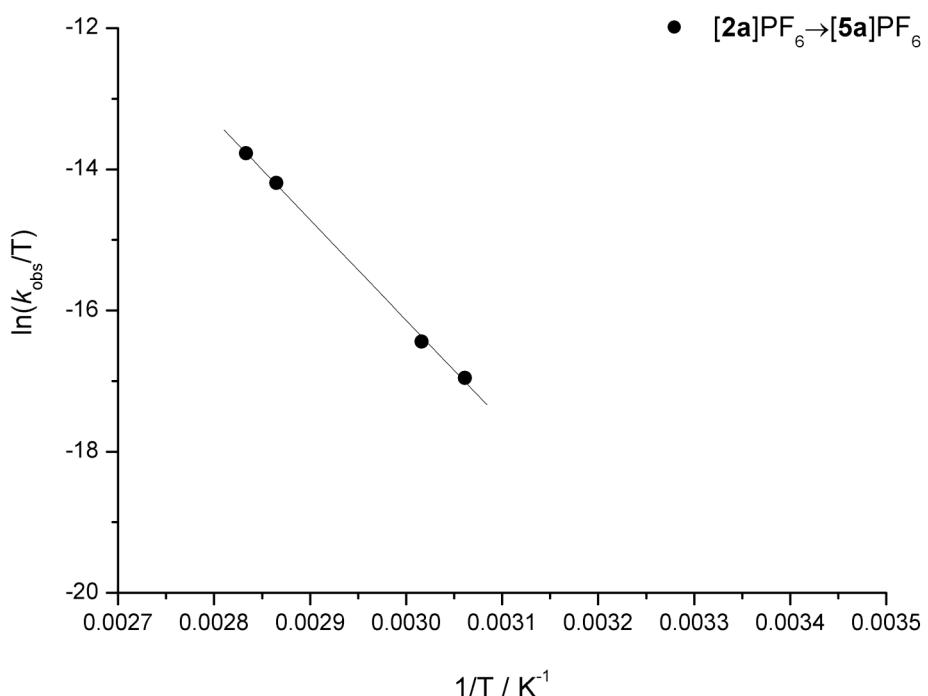


Figure S8: Eyring plot for the chelation of [2a]PF₆ in CH₂ClCH₂Cl ($R^2 = 0.999$).

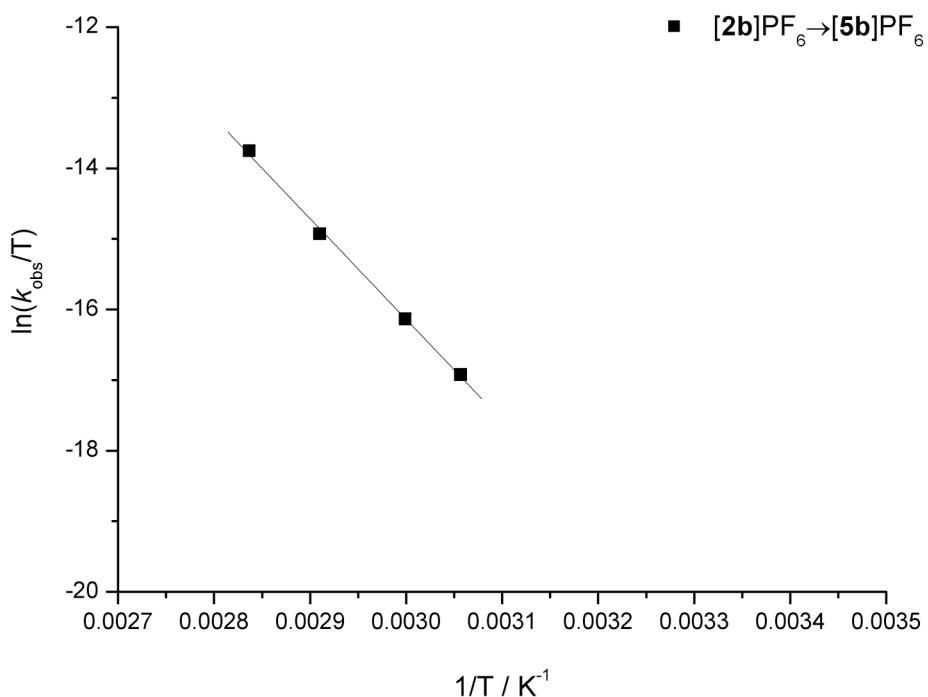


Figure S9: Eyring plot for the chelation of [2b]PF₆ in CH₂ClCH₂Cl ($R^2 = 0.998$).

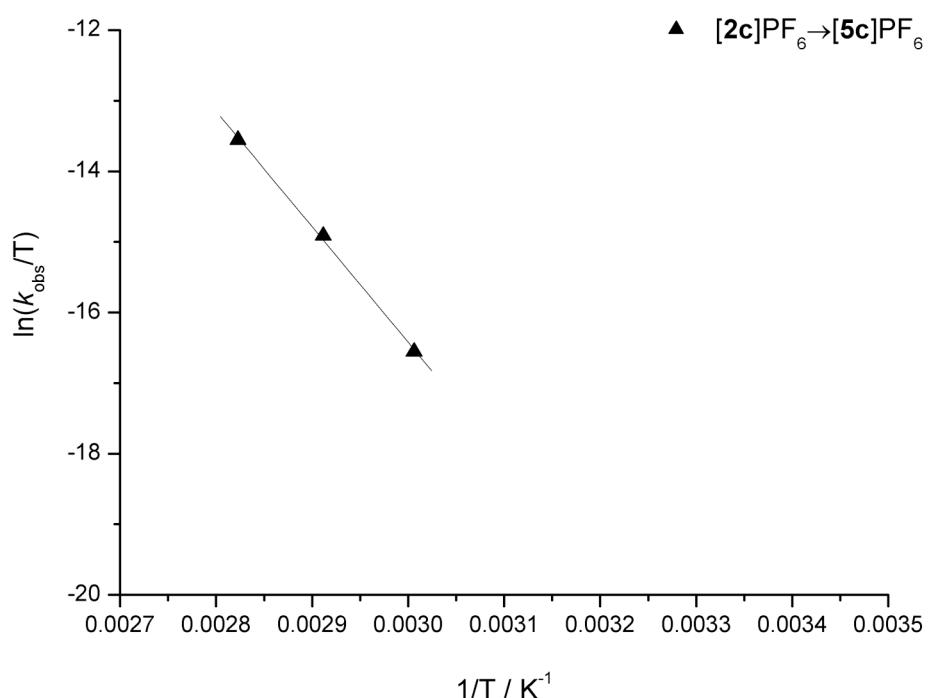


Figure S10: Eyring plot for the chelation of $[2\mathbf{c}]\text{PF}_6$ in $\text{CH}_2\text{ClCH}_2\text{Cl}$ ($R^2 = 0.999$).

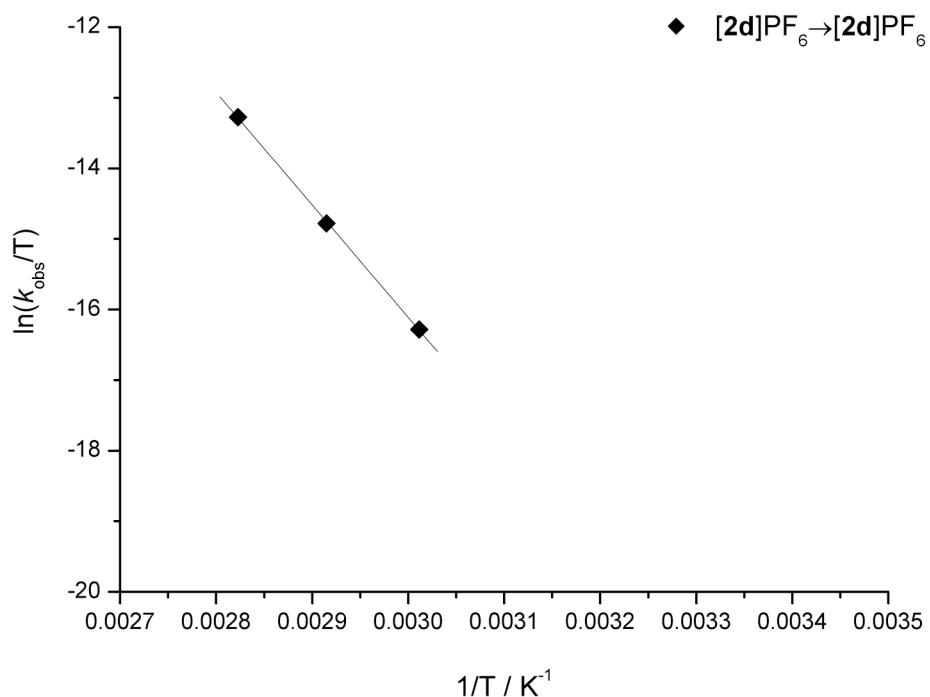


Figure S11: Eyring plot for the chelation of $[2\mathbf{d}]\text{PF}_6$ in $\text{CH}_2\text{ClCH}_2\text{Cl}$ ($R^2 = 1.000$).