

Supporting Information
for
Isotope Effects as Probe of *Constrictive* and *Intrinsic*
Binding in Hemicarceplexes

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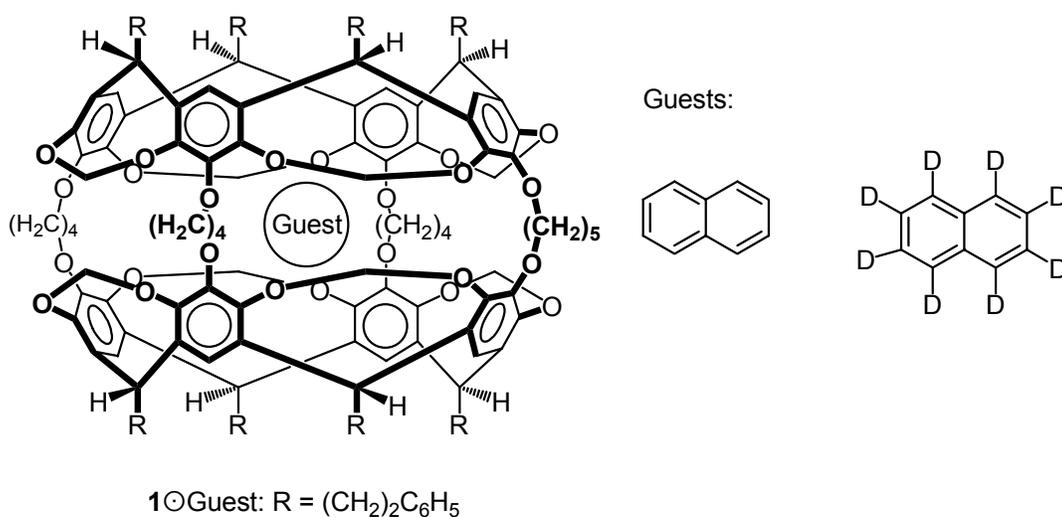
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Experimental Section

1. General. Nitrobenzene was distilled before use. ^1H and ^{13}C NMR spectra in CDCl_3 were referenced to residual CHCl_3 at 7.26 ppm and CDCl_3 at 77.0 ppm. MALDI-TOF mass spectra were recorded on an Applied Biosystems Voyager DE-Pro mass spectrometer in the reflector mode. 2, 4, 6-Trihydroxyacetophenone (THAP) containing silver trifluoroacetate was used as matrix.

2. Synthesis of isotomeric hemicarceplexes 1O naphthalene



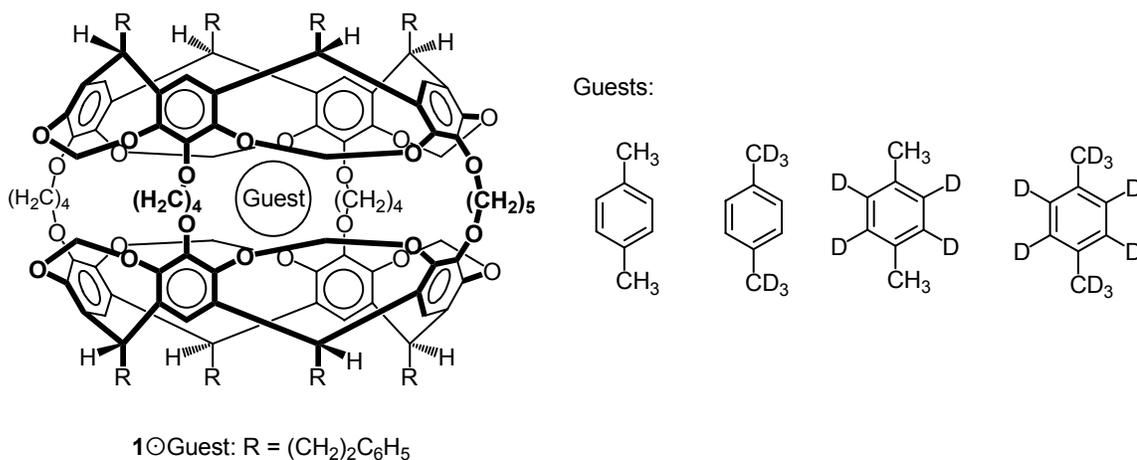
Hemicarceplexes 1O naphthalene and 1O naphthalene- d_8 were synthesized according to a literature procedure [1].

Hemicarceplex 1O naphthalene (61 % yield): ^1H NMR (400 MHz, CDCl_3 ; 22 °C) δ_{H} 7.30-7.21 (m, 24H); 7.15-7.21 (m, 20H); 7.13 (s, 2H); 7.11 (s, 2H); 6.95 (s, 4H); 5.60 (d, 4H, $J = 7.0$ Hz); 5.59 (d, 4H, $J = 7.0$ Hz); 4.88 (t, 4H, $J = 7.8$ Hz); 4.88 (t, 4H, $J = 7.8$ Hz); 4.32 (d, 4H, $J = 7.0$ Hz); 4.30 (d, 4H, $J = 7.0$ Hz); 4.22-4.16 (m, 4H); 3.35 (m, 4H Guest); 3.26-3.11 (m, 8H); 2.78-2.66 (m, 16H), 2.64-2.52 (m, 16H), 2.2 (sb, 4H); 2.12-

2.02 (m, 2H); 1.99-1.88 (m, 4H); 1.42-1.26 (m, 8H). MALDI-MS m/z 2497.94 ($[M+Ag]^+$ calc. 2497.92).

Hemicarceplex $1\ominus$ naphthalene- d_8 (60 % yield): 1H NMR (400 MHz, $CDCl_3$; 22 °C) δ_H 7.30-7.21 (m, 24H); 7.15-7.21 (m, 16H); 7.13 (s, 2H); 7.11 (s, 2H); 6.95 (s, 4H); 5.60 (d, 4H, $J = 7.0$ Hz); 5.59 (d, 4H, $J = 7.0$ Hz); 4.88 (t, 4H, $J = 7.8$ Hz); 4.88 (t, 4H, $J = 7.8$ Hz); 4.32 (d, 4H, $J = 7.0$ Hz); 4.30 (d, 4H, $J = 7.0$ Hz); 4.22-4.16 (m, 4H); 2.6-3.11 (m, 8H); 2.78-2.66 (m, 16H), 2.64-2.52 (m, 16H), 2.2 (sb, 4H); 2.12-2.02 (m, 2H); 1.99-1.88 (m, 4H); 1.42-1.26 (m, 8H). MALDI-MS m/z 2506.01 ($[M+Ag]^+$ calc. 2505.98).

3. Synthesis of isotopomeric hemicarceplexes $1\ominus p$ -xylene



Hemicarceplex **1** (39 mg, 0.017 mmol) [1] and *p*-xylene (0.5 mL) were sealed in a glass ampoule and heated to 150 °C for 18 hrs. The ampoule was opened and the solvent removed at a rotavaporator. The crude product was purified by preparative TLC (silica gel; $CHCl_3$ /ethyl acetate (200:1)), which gave the product as a white powder.

Hemicarceplex $1\ominus p$ -xylene (77 % yield): 1H NMR (300 MHz, $CDCl_3$; 22 °C) δ_H 7.15-7.25 (m, 40H); 6.91 (s, 6H); 6.88 (s, 2H); 5.94 (s, 4H, Guest); 5.71 (d, 4H, $J = 7.1$ Hz); 5.70 (d, 4H, $J = 7.1$ Hz); 4.88 (t, 8H, $J = 7.8$ Hz); 4.14 (d, 4H, $J = 7.1$ Hz); 4.13 (d,

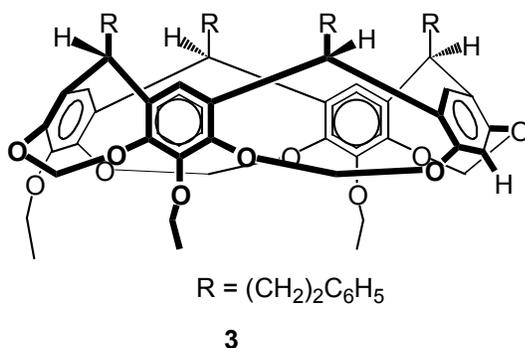
4H, $J = 7.1$ Hz); 3.75-3.95 (m, 16H); 2.45-2.78 (m, 32H), 1.7-2.0 (m, 18H); -1.94 (s, 6H, Guest). ^{13}C NMR (100 MHz, CDCl_3 ; 22 °C) δ_{C} 149.19; 148.98; 148.96; 148.95; 145.56; 144.46; 144.43; 141.83; 138.78; 138.77; 138.71; 138.65; 132.99 (Guest); 128.58; 128.43; 127.49 (Guest); 126.01; 114.28; 114.23; 114.18; 98.98; 98.96; 72.84; 72.10; 71.77; 27.14; 34.54; 32.32; 29.95; 27.47; 27.36; 22.3; 13.52 (Guest). MALDI-MS m/z 2475.94 ($[\text{M}+\text{Ag}]^+$ calc. 2475.94).

Hemicarceplex **1**⊙*p*-xylene- d_4 (71 % yield): ^1H NMR (300 MHz, CDCl_3 ; 22 °C) δ_{H} 7.15-7.25 (m, 40H); 6.91 (s, 6H); 6.87 (s, 2H); 5.70 (d, 4H, $J = 7.0$ Hz); 5.69 (d, 4H, $J = 7.0$ Hz); 4.88 (t, 8H, $J = 7.8$ Hz); 4.13 (d, 4H, $J = 7.0$ Hz); 4.12 (d, 4H, $J = 7.0$ Hz); 3.75-3.95 (m, 16H); 2.43-2.78 (m, 32H), 1.7-2.0 (m, 18H); -2.03 (s, 6H, Guest). MALDI-MS m/z 2479.98 ($[\text{M}+\text{Ag}]^+$ calc. 2479.97).

Hemicarceplex **1**⊙*p*-xylene- d_6 (67 % yield): ^1H NMR (300 MHz, CDCl_3 ; 22 °C) δ_{H} 7.15-7.25 (m, 40H); 6.90 (s, 6H); 6.87 (s, 2H); 5.95 (s, 4H, Guest); 5.70 (d, 4H, $J = 7.1$ Hz); 5.69 (d, 4H, $J = 7.1$ Hz); 4.87 (t, 8H, $J = 7.8$ Hz); 4.14 (d, 4H, $J = 7.1$ Hz); 4.13 (d, 4H, $J = 7.1$ Hz); 3.75-3.95 (m, 16H); 2.43-2.78 (m, 32H), 1.7-2.0 (m, 18H). MALDI-MS m/z 2481.97 ($[\text{M}+\text{Ag}]^+$ calc. 2481.98).

Hemicarceplex **1**⊙*p*-xylene- d_{10} (73 % yield): ^1H NMR (300 MHz, CDCl_3 ; 22 °C) δ_{H} 7.15-7.25 (m, 40H); 6.90 (s, 6H); 6.88 (s, 2H); 5.71 (d, 4H, $J = 7.1$ Hz); 5.70 (d, 4H, $J = 7.1$ Hz); 4.87 (t, 8H, $J = 7.8$ Hz); 4.13 (d, 4H, $J = 7.1$ Hz); 3.75-3.95 (m, 16H); 2.43-2.78 (m, 32H), 1.7-2.0 (m, 18H). MALDI-MS m/z 2486.02 ($[\text{M}+\text{Ag}]^+$ calc. 2486.00).

4. Synthesis of standard 3



Trihydroxycavitand [2] (400 mg, 0.4 mmol) was dissolved in degassed DMF (25 mL) under argon. After addition of anhydrous K_2CO_3 (2 g) and iodoethane (6 mL, 75 mmol), the reaction mixture was stirred 20 hrs at 80 °C. The reaction mixture was filtered and poured into water (200 mL). The precipitated product was filtered off and dried at high vacuum. Column chromatography on silica gel (CHCl_3 ; EtOAc 0 – 5%) gave triethoxy cavitand **3** as white solid (300 mg, 70% yield)

^1H NMR (360 MHz, CDCl_3 ; 22 °C) δ_{H} 7.15-7.25 (m, 40H); 7.10 (s, 1H); 6.84 (s, 3H); 6.56 (s, 1H); 5.86 (d, 2H, $J = 7.2$ Hz); 5.81 (d, 2H, $J = 7.2$ Hz); 4.83 (t, 4H, $J = 7.9$ Hz); 4.46 (d, 2H, $J = 7.2$ Hz); 4.39 (d, 2H, $J = 7.2$ Hz); 3.97-4.04 (m, 6H); 2.45-2.70 (m, 16H), 1.25-1.33 (m, 9H); MALDI-MS m/z 1191.31 ($[\text{M}+\text{Ag}]^+$ calc. 1191.38).

5. Decomplexation of hemicarceplexes

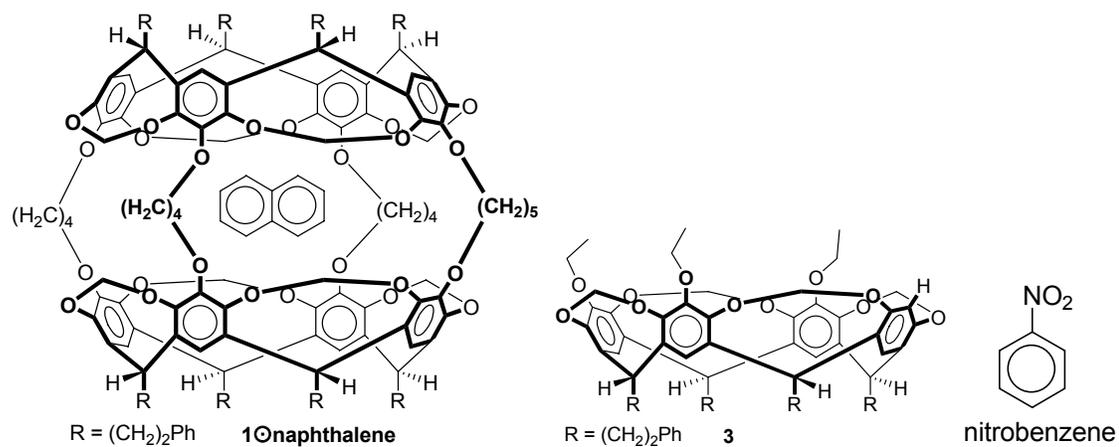
All decomplexation reactions were conducted in a thermostated silicon oil bath. The temperature variation were less than 0.1 °C. Approximately 1 mg of hemicarceplex and 0.5 mg of standard **3** were added to 0.8 ml of nitrobenzene in a 1 mL sample vial with teflon coated septum cap. The sample was sonicated until the solution was clear. To eliminate temperature difference between samples, decomplexation experiments involving isotopomeric hemicarceplexes were run in parallel. All sample vials were held

tightly together in a basket, which was fixed to a small stand inside the oil bath such that the silicon oil surface was just below the top of the septum cap. The sample vials were placed in the oil bath for 30 minutes to reach temperature equilibrium. 25 μl of samples were taken at 10 min time interval, diluted with 100 μl dichloromethane and subjected for HPLC analysis.

6. HPLC analysis

Two HPLC methods were developed For hemicarceplex 1 \odot naphthalene decomplexation reaction mixtures, the method conditions were as follows: column, Phenomenax Luna Si, 15.0 cm x 4.6 mm, 5 μm ; flow rate, 1 mL/min; temperature, 40 $^{\circ}\text{C}$; mobile phase, A, CH_2Cl_2 , B, 20% ether in CH_2Cl_2 ; gradient, A/B, 90/10 to A/B, 75/25 in 8 min, hold for 2 min then to 100% B in 2 min and hold for 10 min; UV detection, 227 nm; injection volume 20 μl , post time 15 minutes (Figure S1). For hemicarceplex 1 \odot *p*-xylene decomplexation reaction mixtures the method conditions are as follows: column, Phenomenax Luna Si, 15.0 cm x 4.6 mm, 5 μm ; flow rate, 1 mL/min; temperature, 40 $^{\circ}\text{C}$; mobile phase, A, CH_2Cl_2 , B, 20% ether in CH_2Cl_2 ; gradient, A/B, 95/5 for 5 min then to A/B, 90/10 in 8 min, hold for 2 min then to 100% B in 2 min and hold for 10 min; UV detection, 227 nm; injection volume 20 μl , post time 15 minutes (Figure S2). Three injections were made for each sample concentration measurement. The rsd% of concentration measurement among the three injections for all the samples was within 1.0%.

7. HPLC Traces



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Acquired Wednesday, August 16, 2006 7:35:28 AM

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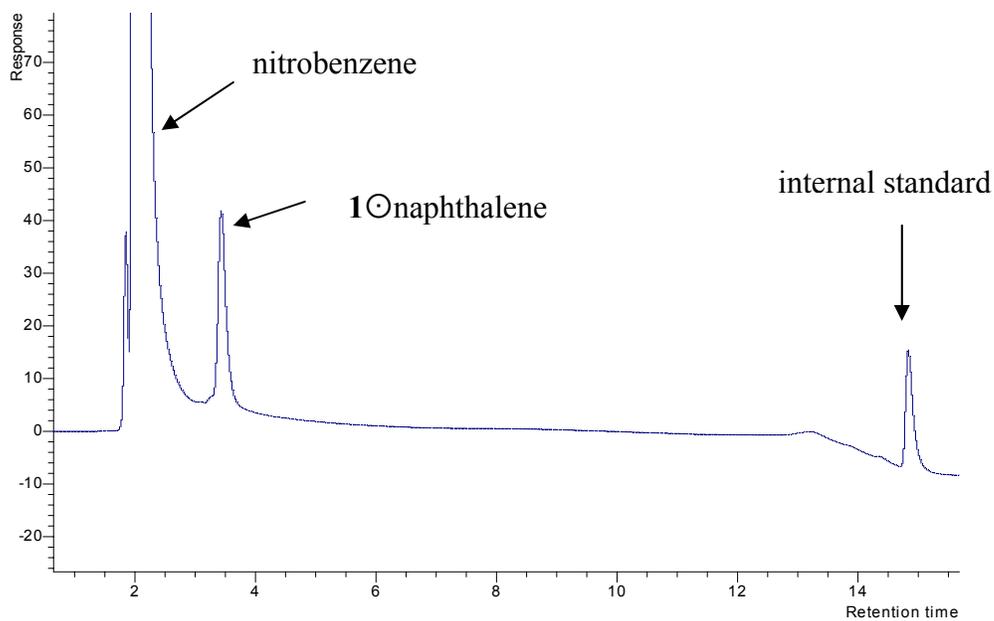
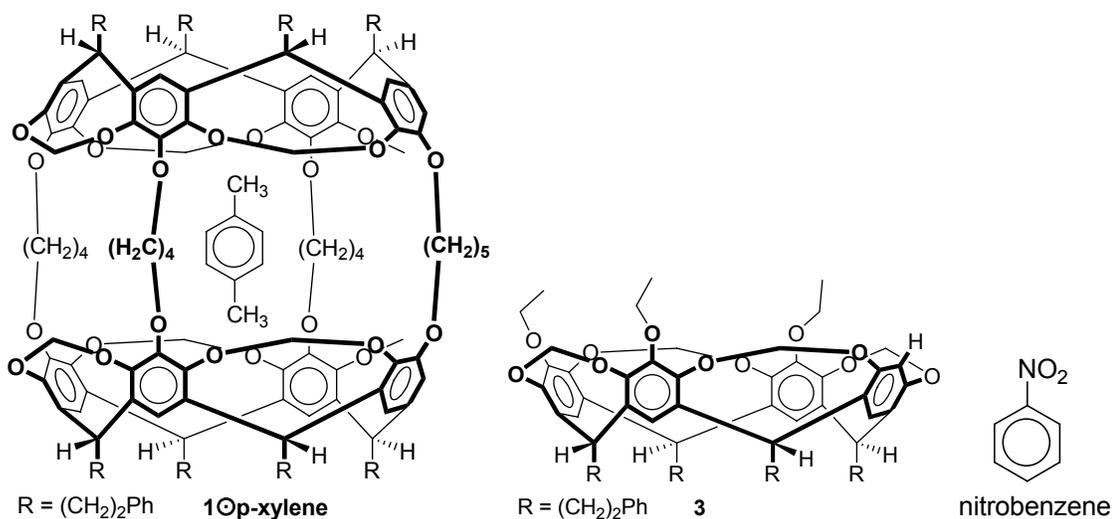


Figure S 1. NPLC separation of **1⊖naphthalene**, cavitand **3** (internal standard) and nitrobenzene.



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Acquired Sunday, October 08, 2006 4:37:28 AM

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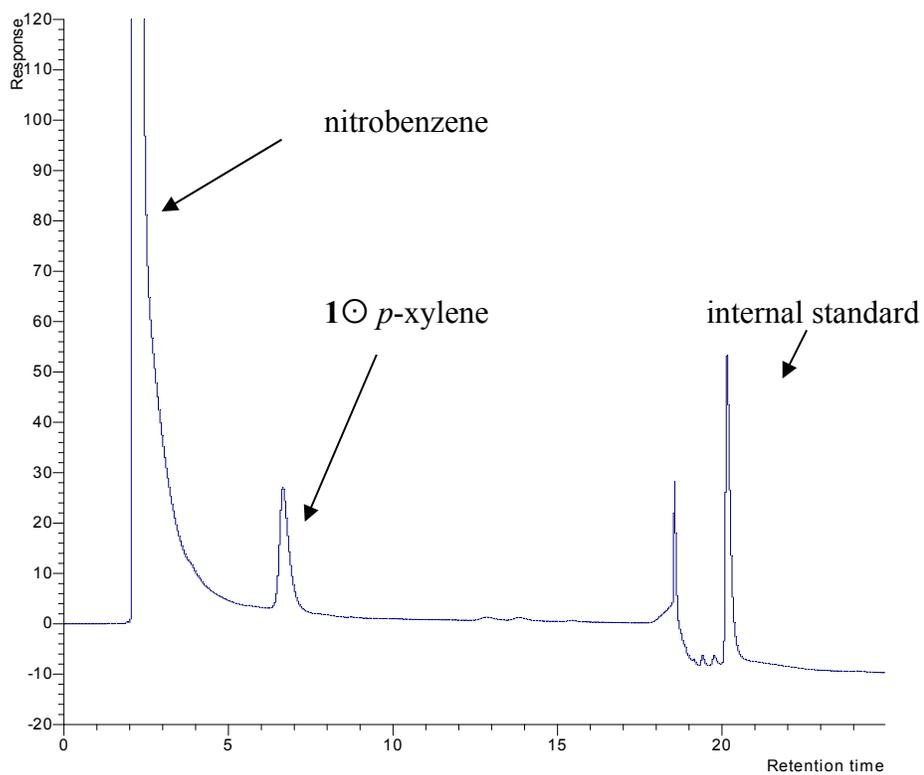


Figure S 2. NPLC separation of 1 \ominus p-xylene, cavitant **3**(internal standard) and nitrobenzene.

8. Spectra

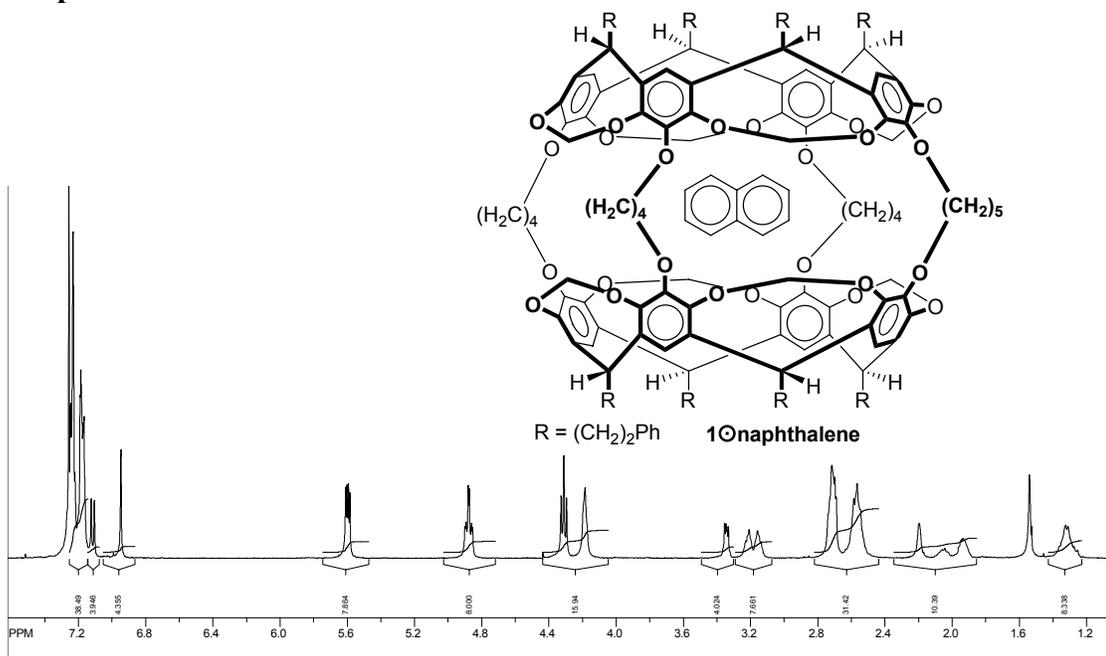


Figure S 3. ^1H NMR spectrum (400 MHz, CDCl_3 ; 22 $^\circ\text{C}$) of hemicarceplex **1O-naphthalene**

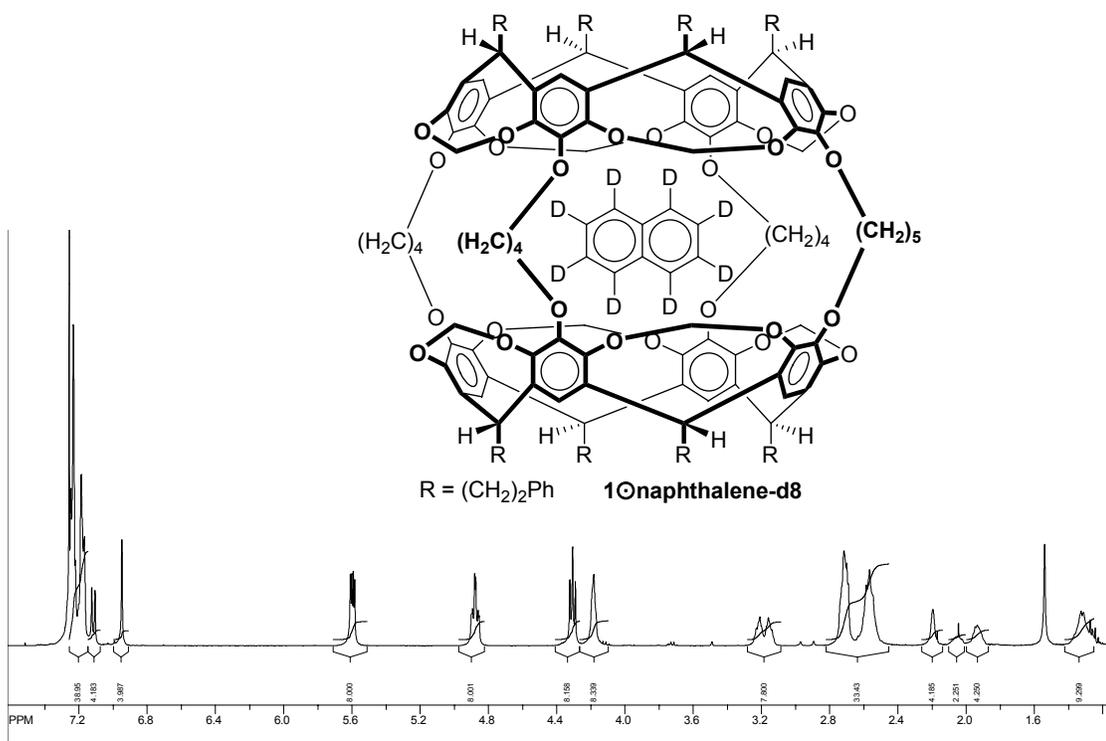


Figure S 4. ^1H NMR spectrum (400 MHz, CDCl_3 ; 22 $^\circ\text{C}$) of hemicarceplex **1O-naphthalene- d_8**

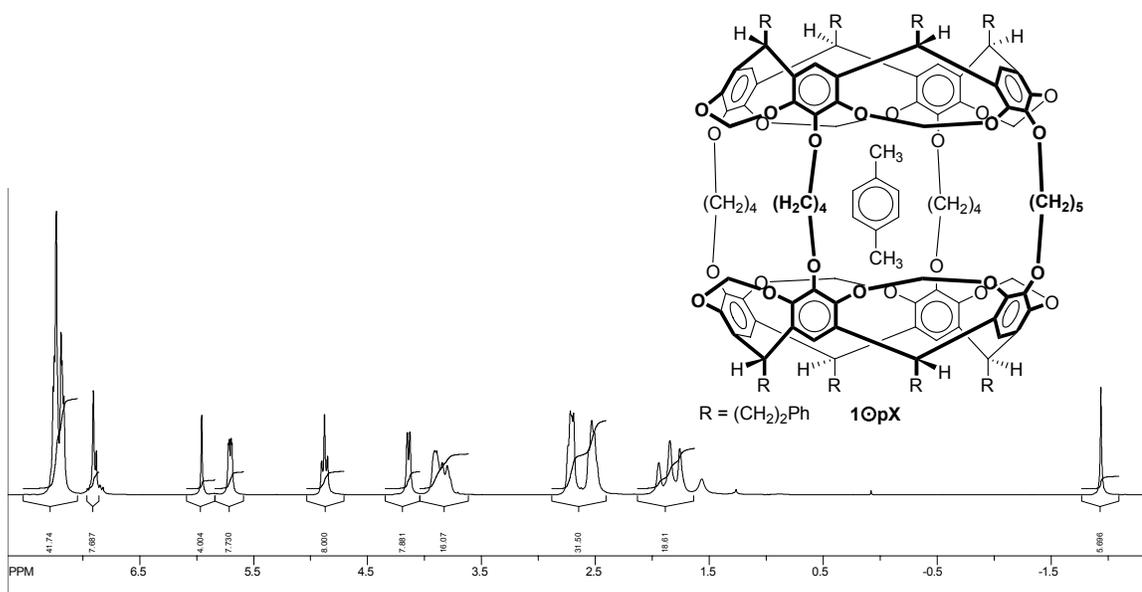


Figure S 5. ^1H NMR spectrum (300 MHz, CDCl_3 ; 22 $^\circ\text{C}$) of hemicarceplex 1Op -xylene

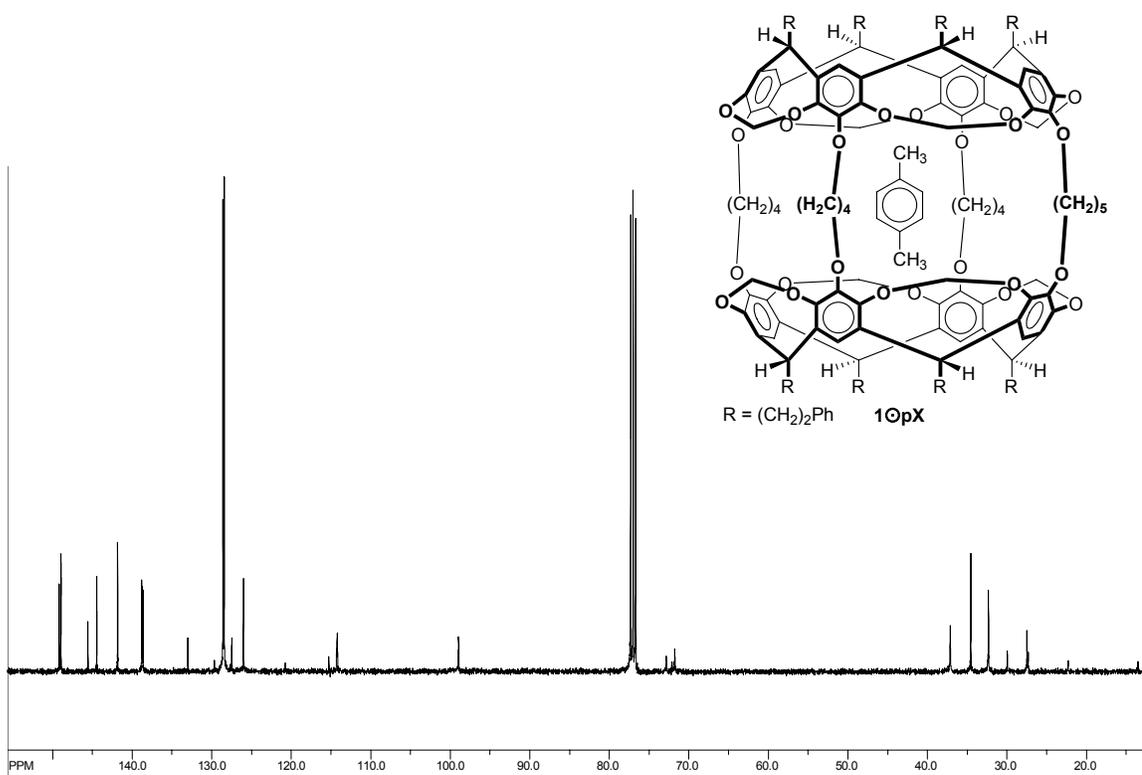


Figure S 6. ^{13}C NMR spectrum (100 MHz, CDCl_3 ; 22 $^\circ\text{C}$) of hemicarceplex 1Op -xylene

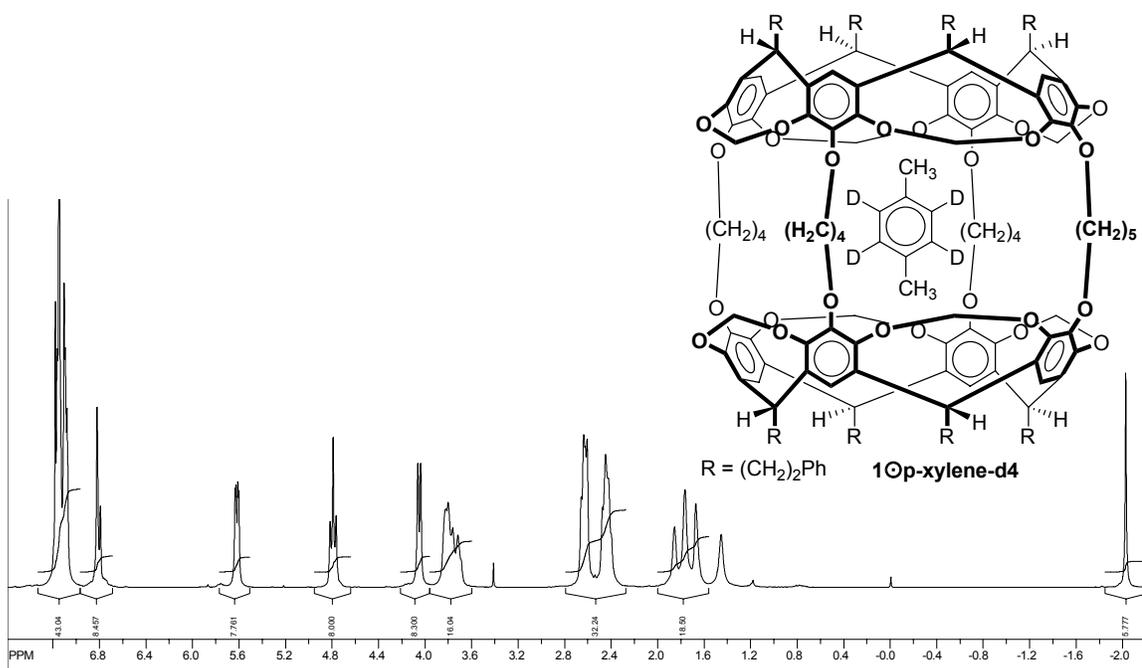


Figure S 7. ^1H NMR spectrum (300 MHz, CDCl_3 ; 22 $^\circ\text{C}$) of hemicarceplex $1@p\text{-xylene-}d_4$

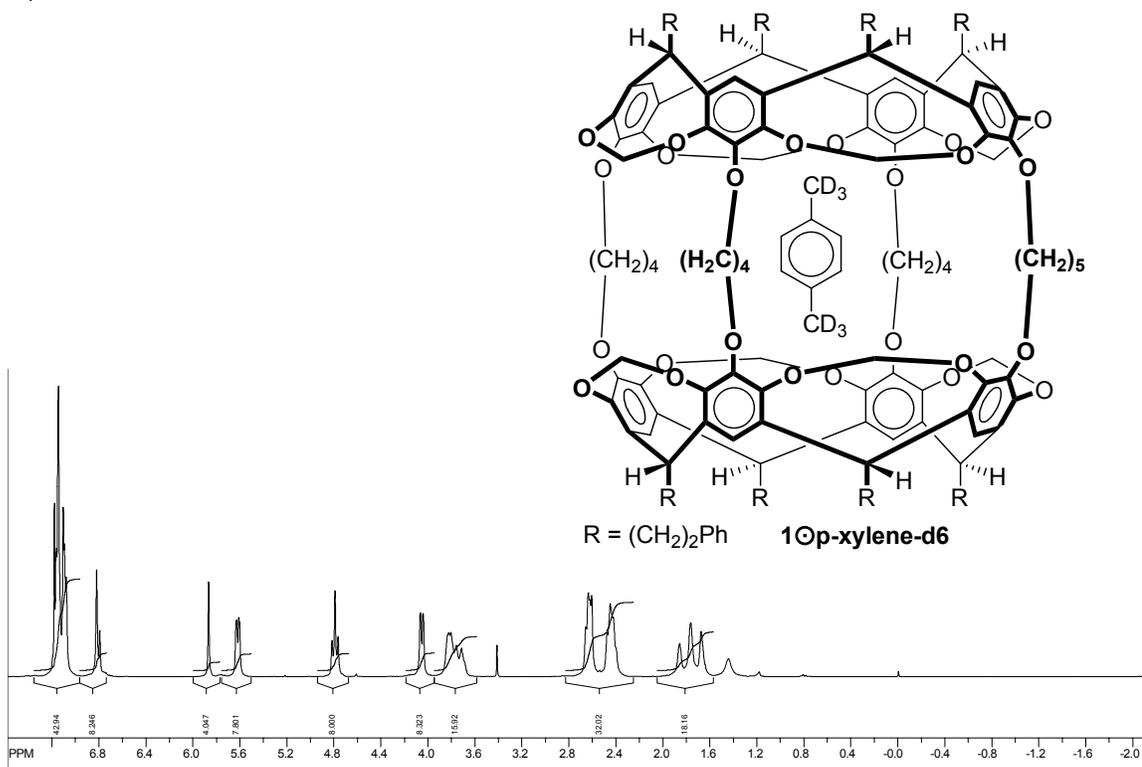


Figure S 8. ^1H NMR spectrum (300 MHz, CDCl_3 ; 22 $^\circ\text{C}$) of hemicarceplex $1@p\text{-xylene-}d_6$

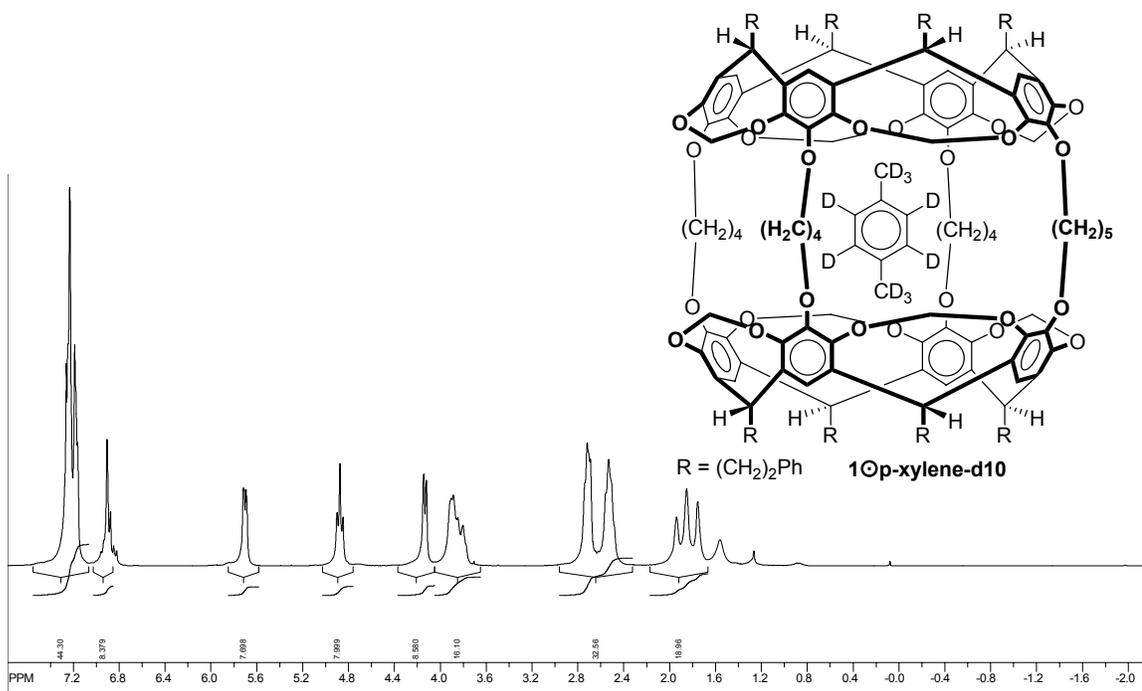


Figure S 9. ^1H NMR spectrum (300 MHz, CDCl_3 ; 22 $^\circ\text{C}$) of hemicarceplex $1@p\text{-xylene-}d_{10}$

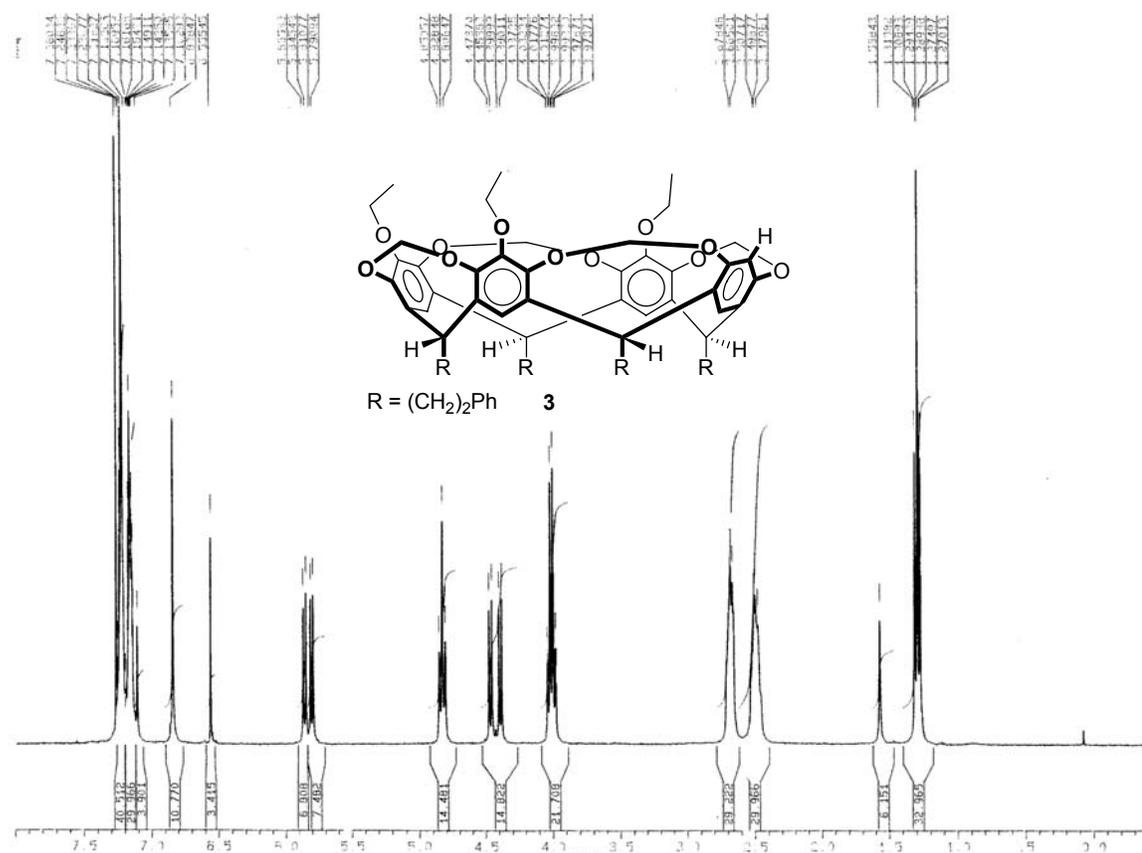


Figure S 10. ^1H NMR spectrum (360 MHz, CDCl_3 ; 22 $^\circ\text{C}$) of **3**.

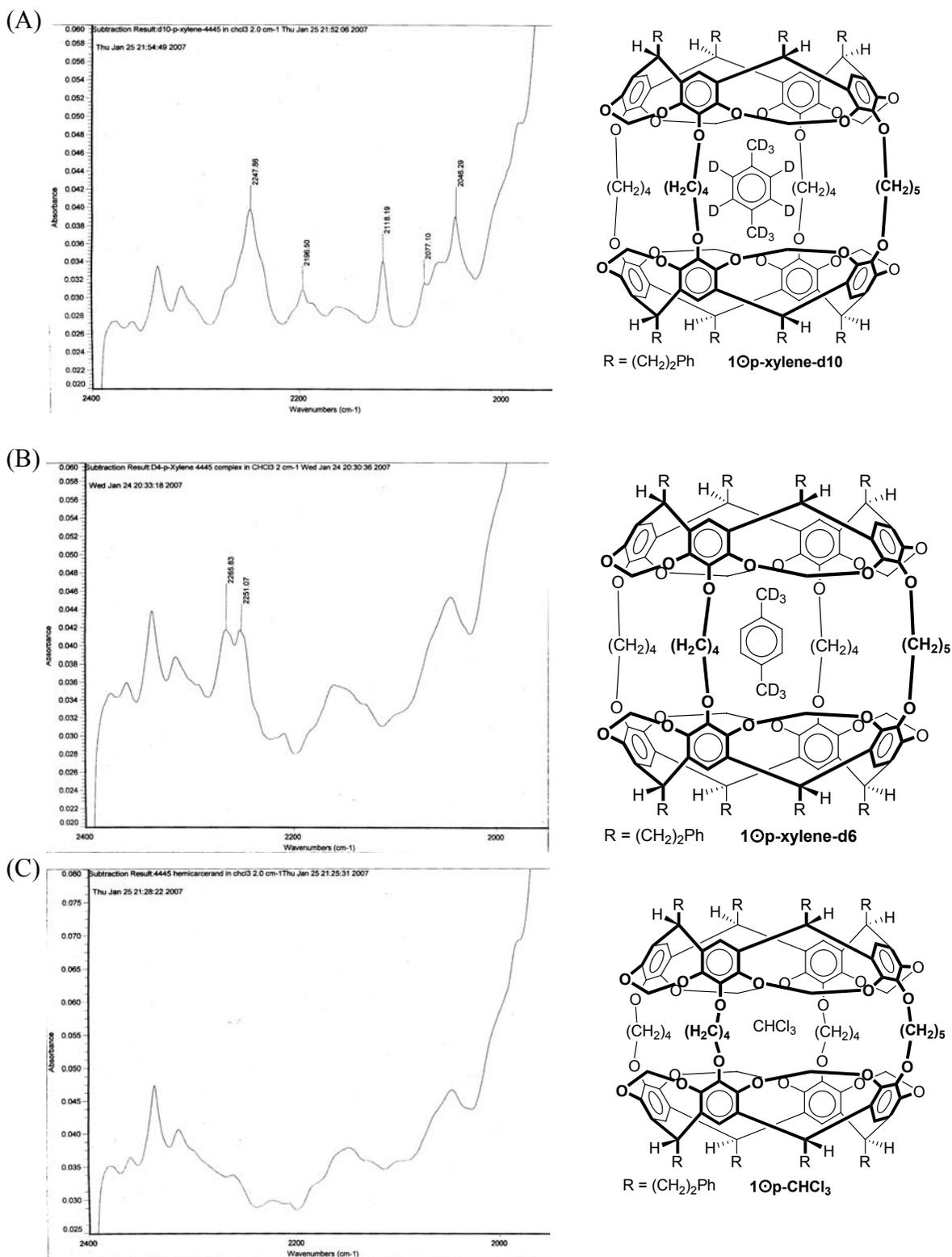


Figure S 11. Partial FT IR spectra of **1⊕p-xylene-d₁₀** (A), **1⊕p-xylene-d₆** (B) and **1⊕CHCl₃** (C) in CHCl₃

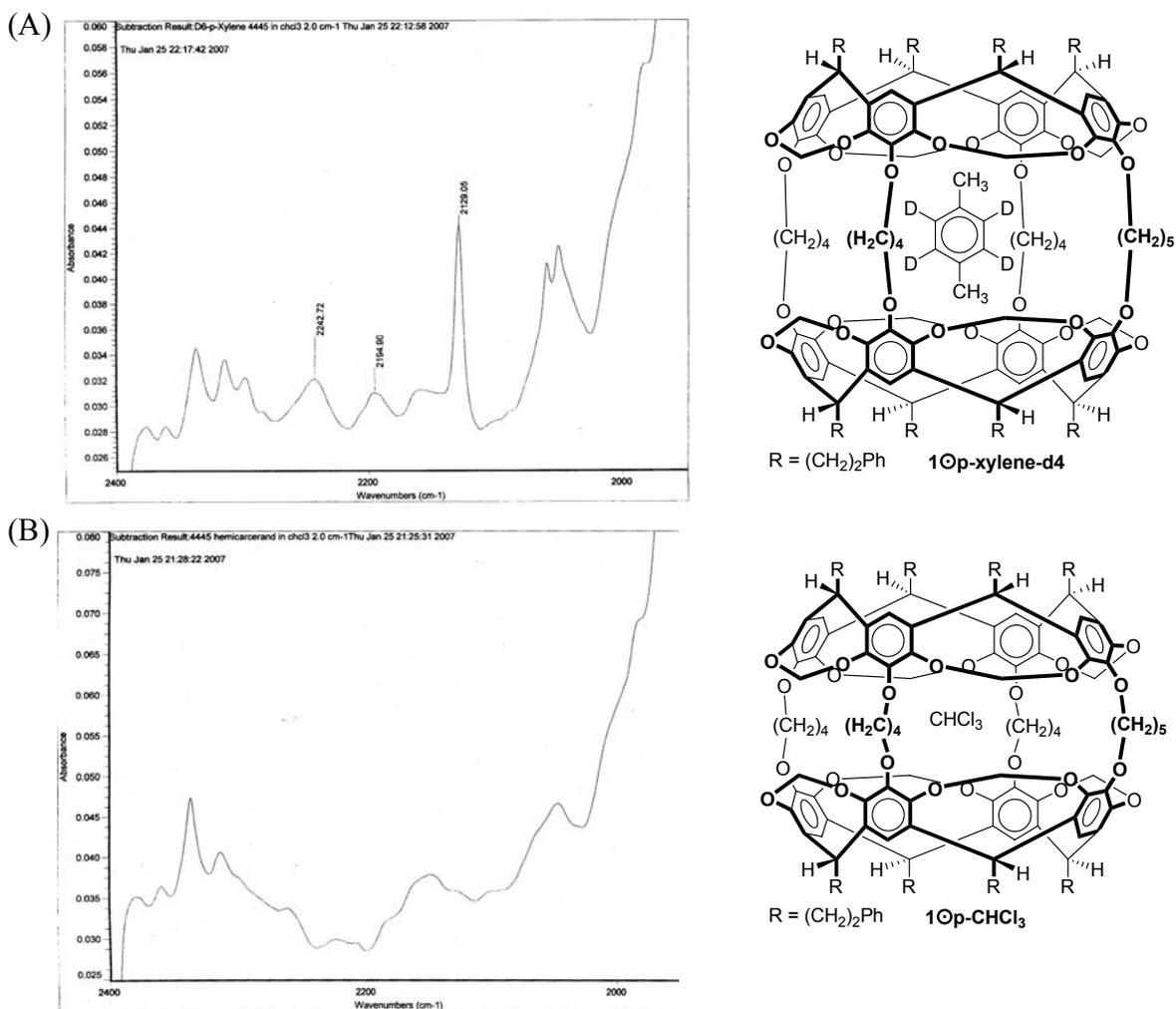


Figure S 12. Partial FT IR spectra of **1⊖p-xylene-d₄** (A), and **1⊖CHCl₃** (B) in CHCl₃

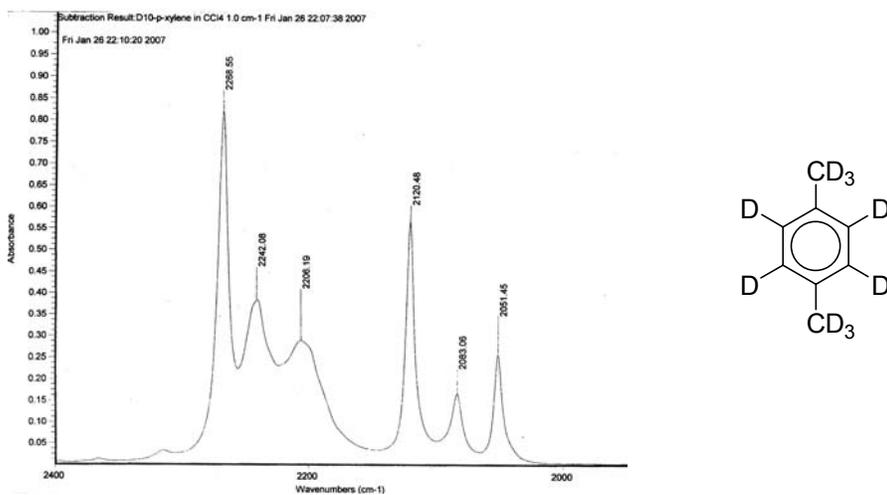


Figure S 13. Partial FT IR spectrum of *p*-xylen-d₁₀ in CCl₄

9. Kinetic Plots

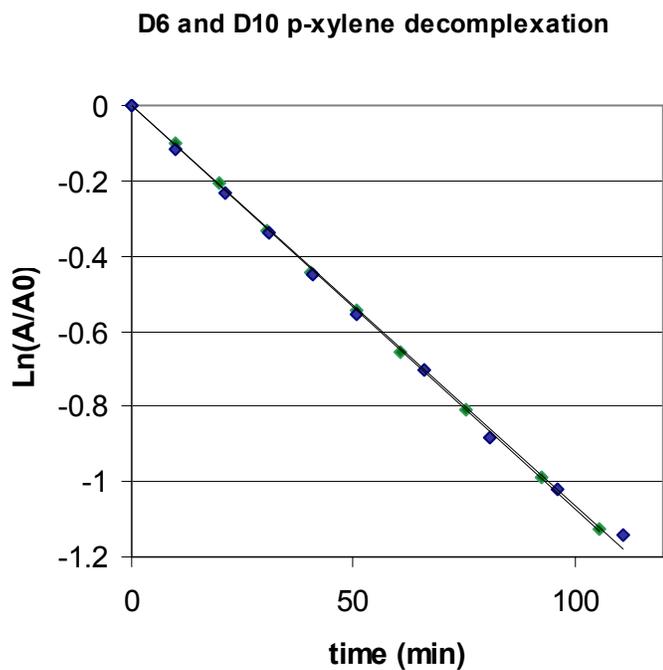


Figure S 14. Kinetic plots ($\ln(A/A_0)$ vers time) for the decomplexation of $1\text{-}p\text{-xylene-}d_{10}$ (dark blue squares) and $1\text{-}p\text{-xylene-}d_6$ (green squares) at 120 °C in nitrobenzene.

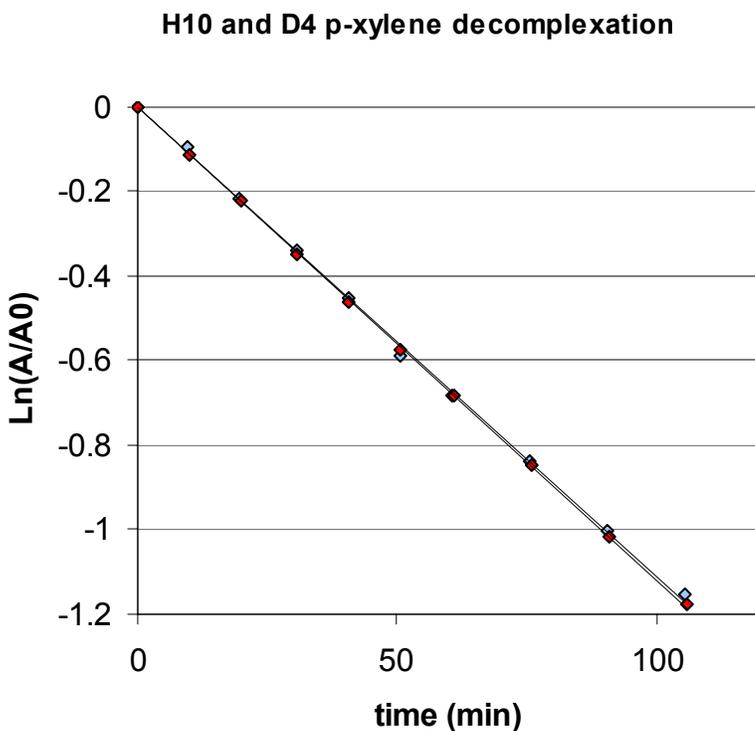


Figure S 15. Kinetic plots ($\ln(A/A_0)$ vers time) for the decomplexation of $1\text{-}p\text{-xylene}$ (red squares) and $1\text{-}p\text{-xylene-}d_4$ (light blue squares) at 120 °C in nitrobenzene.

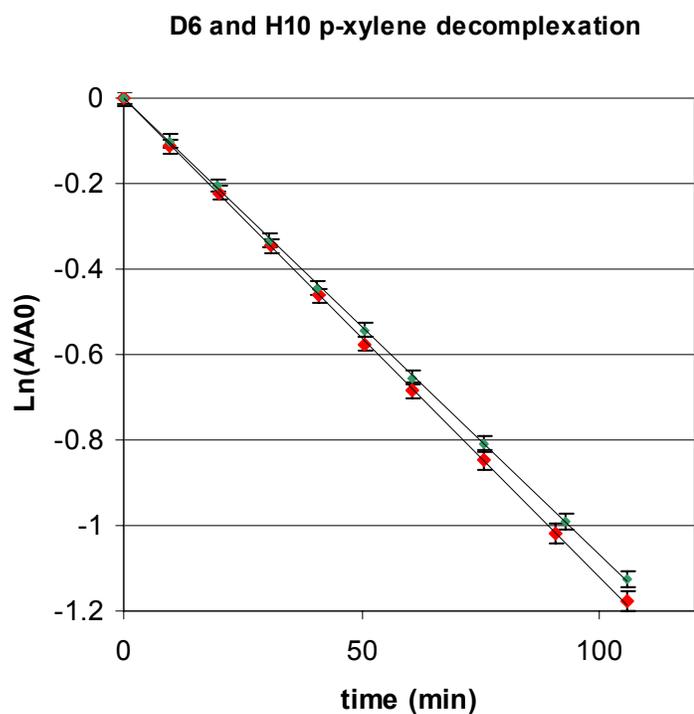


Figure S 16. Kinetic plots ($\ln(A/A_0)$ vers time) for the decomplexation of $1,4\text{-p-xylene}$ (red squares) and $1,4\text{-p-xylene-}d_6$ (green squares) at $120\text{ }^\circ\text{C}$ in nitrobenzene.

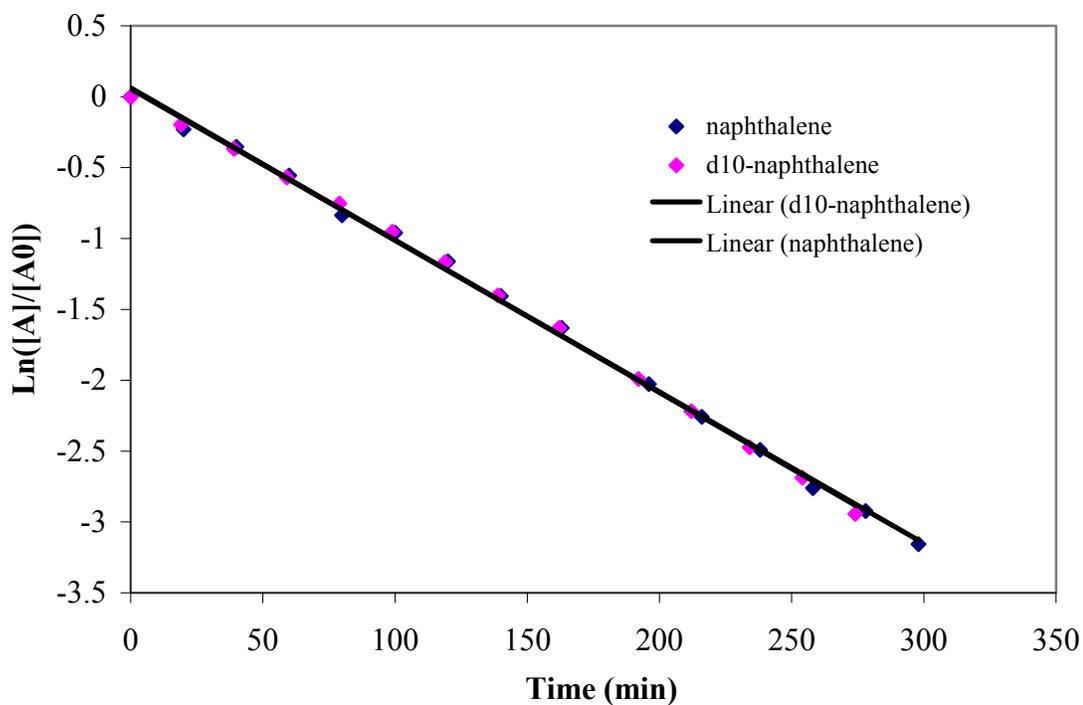


Figure S 17. Kinetic plots ($\ln(A/A_0)$ vers time) for the decomplexation of $1,4\text{-Naph}$ (blue) and $1,4\text{-Naph-}d_8$ (pink) at $169\text{ }^\circ\text{C}$ in nitrobenzene.

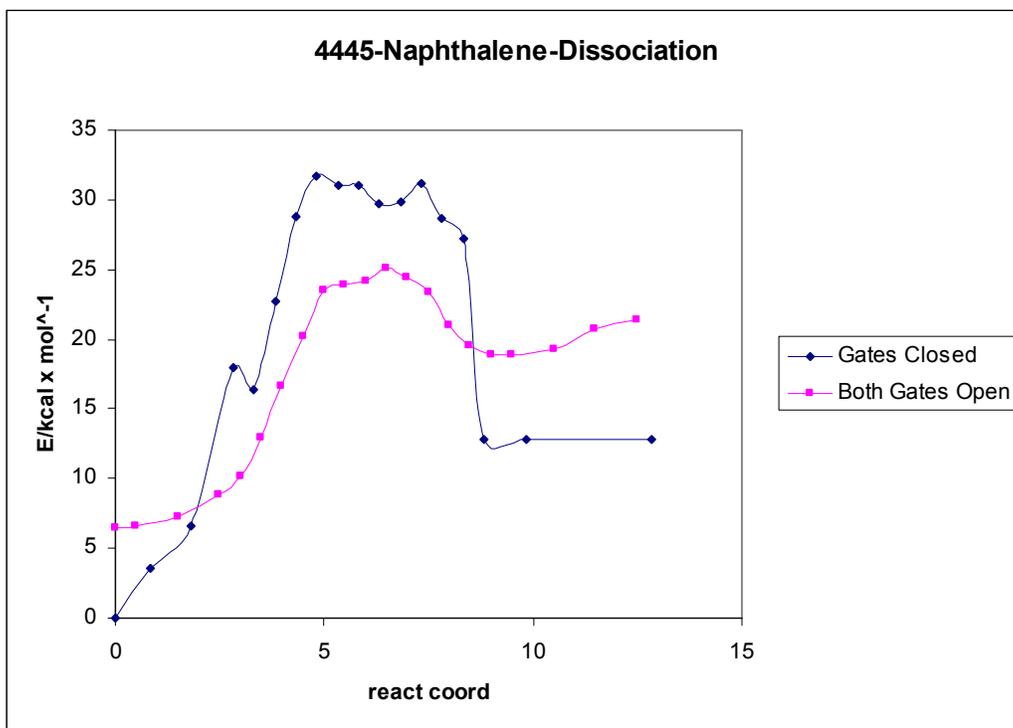


Figure S 18. Computed Energy surface (MM3) for decomplexation of $1\ominus\text{Naph}$ with (pink) and without (blue) *French door gating*. Computations were performed as described in reference [3]. No correction for bulk phase solvation was performed.

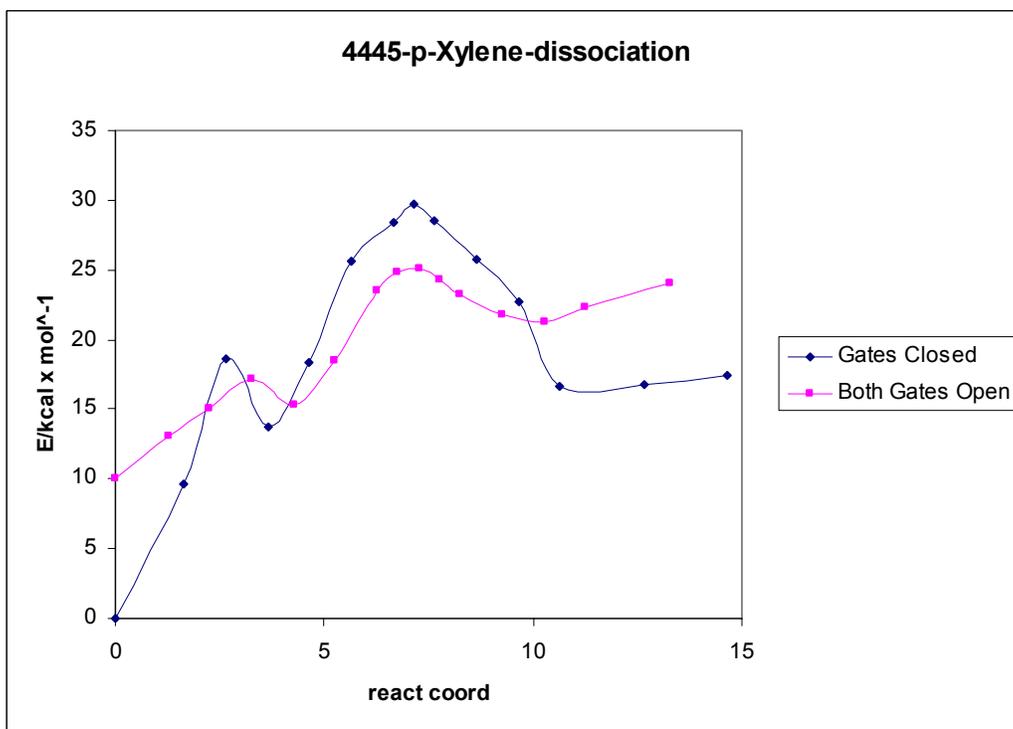


Figure S 19. Computed Energy surface (MM3) for decomplexation of $1\ominus\text{pX}$ with (pink) and without (blue) *French door gating*. Computations were performed as described in reference [3]. No correction for bulk phase solvation was performed.

References

- [1] Yoon, J.; Sheu, C.; Houk, K. N.; Knobler, C. B.; Cram, D. J. *J. Org. Chem.* **1996**, *61*, 9323-9339.
- [2] Cram, D. J.; Tanner, M. E.; Knobler, C. B. *J. Am. Chem. Soc.* **1991**, *113*, 7717-7727.
- [3] Sheu C., Houk, K. N. *J. Am. Chem. Soc.* **1996**, *118*, 8056-8070.