[2+2] Cycloaddition Reactions with a Tungsten-Stabilized 2H-Phenol

Michael A. Todd, Michal Sabat, William H. Myers[#], W. Dean Harman

Contribution from the Department of Chemistry, University of Virginia Charlottesville, VA 22904-4319

Department of Chemistry, University of Richmond, Richmond, Virginia 23173 wdh5z@virginia.edu

Supplementary Material

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Experimental Details.

General Methods. All NMR spectra were obtained on either a 300 or 500 MHz Varian INOVA spectrometer, or on a 300 or 500 MHz Bruker Avance spectrometer. All chemical shifts are reported in ppm versus tetramethylsilane using residual shifts of the deuterated solvent as the internal standard. All coupling constants (J) are reported in All ³¹P NMR data is reported versus an external standard in acetone hertz (Hz). (Trimethylphosphate, δ -16.58 ppm). Infrared spectra were obtained on a MIDAC Prospect Series spectrometer as a glaze on a Horizontal Attenuated Total Reflectance (HATR) cell from Pike Industries. Electrochemical measurements were taken under a nitrogen atmosphere using a BAS Epsilon EC-2000 potentiostat. Cyclic voltammetry data were obtained in a three electrode cell from +1.7 to -1.7 V, with a glassy carbon working electrode, a platinum wire auxiliary electrode and a platinum wire reference electrode. All data were obtained using a 100 mv/s scan rate with tetrabutylammonium hexafluorphosphate (TBAH) as the electrolyte in N,N-dimethylacetamide (DMA) unless otherwise noted. All potentials were reported versus the Normal Hydrogen Electrode (NHE) using cobaltocenium hexafluorophosphate ($E_{1/2} = -0.78$ V) as an internal standard. For reversible waves the peak to peak separation was less than 100 mV. All HRMS data was obtained using a 1:1 water: acteonitrile solution with sodium trifluoroacetate as a standard. The data for metal complexes is reported using the five most intense peaks from the isotopic envelope, and is listed as m/z with the intensity relative to the most abundant peak of the isotopic envelope given in parentheses for both the calculated and observed peaks. The difference between calculated and observed peaks is reported in ppm. For isolated organic products, the monoisotopic peak is reported, with observed, calculated, and ppm difference again provided. Thin layer chromatography was performed on a Uniplate silica gel GF from Analtech Inc. Methylene chloride and benzene were all run down a column packed with activated alumina and purged with nitrogen prior to use. All other solvents and chemicals were used as received from Sigma-Aldrich, Acros Chemicals or Fischer Scientific. Compound 1 was previously reported.¹

TpW(NO)(PMe₃)(η^2 -phenol) (1b)

In an Erlenmeyer flask with a ground glass joint 40 mL methanol was purged for 10 min with argon. TpW(NO)(PMe₃)(η^2 -phenol) (**1a,b**) (4.068g, 6.814 mmol) and a stir bar were added. The reaction was purged for 5 min and DBU (0.249g, 1.596 mmol) was added. The flask was sealed with a glass stopper. After 3 d the reaction was filtered and washed with 2 mL methanol and dried *in vacuo*. A light yellow precipitate was collected **1b** (3.298g, 81% yield.) Since ketenes react with methanol the solid was reprecipitated using methylene chloride (10 mL) and hexanes (400 mL) to remove any remaining methanol prior to the synthesis of **2a,b**, **3a,b**, **4a,b** and **5**.

TpW(NO)(PMe₃)(η^2 -4,5-(7-chlorobicyclo[4.2.0]oct-4-ene-3,8-dione)) (2a,b)

In a glovebox under a dinitrogen atmosphere, using flame dried glassware, a solution of chloroacetylchloride (0.140g, 1.24 mmol) in 75 mL CH_2Cl_2 was slowly added drop-wise to a well stirred solution of **1b** (0.500g, 0.838 mmol) and DIEA (0.167g, 1.29

mmol) in 25 mL CH₂Cl₂. After about 1 hour the addition was complete and the reaction was removed from the glovebox. HCl_(aq) (1M, 100 mL) was added and the CH₂Cl₂ layer was extracted, dried with MgSO₄, filtered and evaporated by rotatory evaporation. The residue was dissolved in 1 mL CH₂Cl₂. Diethyl ether (10 mL) was slowly added and the resulting precipitate was filtered, washed with five 10 mL portions of ether and dried *in vacuo*. The light beige precipitate was isolated as a mixture of two diastereomers (2a,b, 1.4:1 dr, 0.517g, 92 %). ¹H NMR (CDCl₃): δ 1.12 (d, 9H, J=8.4 Hz, PMe₃ A), 1.17 (d, 9H, J=8.7 Hz, PMe₃B), 2.01 (d, 1H, J=9.6 Hz, H4 A), 2.18 (d, 1H, J=9.6 Hz, H4 B), 2.58 (d, 1H, J=18.0, H2 B), 2.64 (d, 1H, J=16.8, H2 A), 2.87 (m, 2H, H2 A and B), 3.26 (dd, 1H, J=13.2, 9.6 Hz, H5 A), 3.35 (dd, 1H, J=9.3, 6.9 Hz, H6 B), 3.46 (dd, 1H, J=12.6, 9.6 Hz, H5 B), 3.69 (br t, J=9.3 Hz, H1 A), 3.87 (m, 2H, H6 A, H1 B) 4.77 (dd, J=6.9, 3.0 Hz, H7 B), 5.22 (dd, J=9.0, 2.1 Hz, H7 A), 6.09 (t, 1H, J=2.4 Hz, Tp), 6.11 (t, 1H, J=2.4 Hz, Tp), 6.22 (t, 1H, J=2.4 Hz, Tp), 6.24 (t, 1H, J=2.4 Hz, Tp), 6.34 (t, 1H, J=2.4 Hz, Tp), 6.35 (t, 1H, J=2.4 Hz, Tp), 7.31 (d, 1H, J=2.4 Hz, Tp), 7.36 (d, 1H, J=2.4 Hz, Tp), 7.42 (d, 1H, J=2.4 Hz, Tp), 7.45 (d, 1H, J=2.4 Hz, Tp), 7.54 (d, 1H, J=2.4 Hz, Tp), 7.68 (d, 1H, J=2.4 Hz, Tp), 7.70 (d, 1H, J=2.4 Hz, Tp), 7.76 (d, 1H, J=2.4 Hz, Tp), 8.09 (d, 1H, J=2.4 Hz, Tp). 13 C NMR (CDCl₃): δ 12.5 (d, J=32 Hz, PMe₃ A), 12.9 (d, J=30 Hz, PMe₃ B), 31.8 (s, C2 B), 32.3 (s, C2 A), 32.9 (d, J=4.0 Hz, C6 A) 41.4 (d, J=3.0 Hz, C6 B), 54.7(s, C1 B), 55.1 (s, C1 A), 57.3 (d, J=12 Hz, C5 A), 59.4 (s, C4 B), 60.6 (s, C4 A), 65.1 (d, J=13, C5 B), 68.5 (s, C7 A), 73.1 (s, C7 B), 106.1 (s, Tp), 106.2 (s, Tp), 106.8 (s, Tp), 106.9 (s, Tp), 107.3 (s, Tp), 107.4 (s, Tp), 136.3 (s, Tp), 136.4 (s, Tp), 136.9 (s, Tp), 137.2 (s, 2Tp), 139.9 (s, Tp), 140.0 (s, Tp), 142.7 (s, Tp), 143.1 (s, 2Tp), 143.6 (s, 2Tp), 202.0 (s, C8 A), 203.8 (s, C3 B), 204.2 (s, C8 A), 205.7 (s, C3 B). ³¹P NMR (CDCl₃): δ -8.0 (A), -8.8 (B). IR: $v_{NO} = 1567$, $v_{CO} = 1621$, $v_{CO} = 1787$ cm⁻¹. CV: $E_{p,a} = +1.14$ V. Anal Calc'd for C₂₀H₂₆BCIN7O₃PW[·]CH₂Cl₂: C, 33.25; H, 3.72; N, 12.93. Found: C, 33.72; H, 3.78; N, 13.00. HRMS: [M + H] obs'd (%), calc'd (%), diff. in ppm: 672.11771 (65.9), 672.11807 (67.1), -0.5; 673.11983 (62.5), 673.12046 (67.0), -0.9; 674.11835 (100.0), 674.11931 (100.0), -1.4; 675.12128 (49.3), 675.12207 (52.8), -1.2; 676.12037 (89.6), 676.12200 (91.4), -2.4.

TpW(NO)(PMe₃)(η^2 -4,5-(7-(3-methoxyphenyl)bicyclo[4.2.0]oct-4-ene-3,8-dione)) (3a,b)

In a glovebox under a dinitrogen atmosphere, using flame dried glassware, a solution of oxalyl chloride (1.608g, 12.66 mmol) in 1 mL benzene was added to 3-methoxyphenylacetic acid (0.128g, 0.771 mmol). After 4 h the solvent was removed via evacuation. 1 mL benzene was added, evacuated to remove any remaining oxalyl chloride. The resulting oil was dissolved in 20 mL CH₂Cl₂ and slowly added dropwise to a well stirred solution of **1b** (0.199g, 0.333 mmol) and DIEA (0.101g, 0.783 mmol) in 10 mL CH₂Cl₂. After about 30 minutes the addition was complete and the reaction was removed from the glovebox. $HCl_{(aq)}$ (1M, 40 mL) was added and the CH₂Cl₂ layer was extracted, dried with MgSO₄, filtered and evaporated using rotatory evaporation. Acetone (15 mL) was added to the residue and the mixture was vigorously stirred. After 15 minutes the reaction was filtered and washed with 1 mL acetone, yielding an off-white solid (**3a**, 0.072 g, 29 %yield). ¹H NMR (CDCl₃): δ 1.09 (d, 9H, J=8.7 Hz, PMe₃), 2.26 (d, 1H, J=9.3 Hz, H4), 2.72 (d, 1H, J=18.0 Hz, H2), 2.98 (dd, 1H, J=18.0, 12.0 Hz, H2),

3.51 (m, 2H, H6, H5), 3.80 (m buried, H1), 3.80 (s, 3H, OMe), 4.41 (dd, 1H, J=7.2, 2.4 Hz, H7), 6.16 (d, 1H, J=2.4 Hz, Tp), 6.23 (d, 1H, J=2.4 Hz, Tp), 6.37 (d, 1H, J=2.4 Hz, Tp), 6.78-6.88 (m, 3H), 7.27 (m, 1H), 7.33 (t, 1H, J=2.4 Hz, Tp), 7.53 (t, 1H, J=2.4 Hz, Tp), 7.56 (t, 1H, J=2.4 Hz, Tp), 7.70 (t, 1H, J=2.4 Hz, Tp), 7.77 (t, 1H, J=2.4 Hz, Tp), 8.13 (t, 1H, J=2.4 Hz, Tp). ¹³C NMR (CDCl₃): δ 12.7 (d, J=29 Hz), 32.7 (s), 37.8 (s), 55.5 (s), 56.5 (s), 60.2 (s), 67.9 (d, J=13 Hz), 78.3 (s), 106.4 (s), 106.7 (s), 107.4 (s), 112.6 (s), 113.3 (s), 119.6 (s), 130.2 (s), 136.2 (s), 137.0 (s), 137.1 (s), 138.5 (s), 140.0 (s), 143.0 (s), 143.6 (s), 160.2 (s), 206.7 (s), 210.8 (s).

The filtrate was added to 50 mL hexane and the resulting precipitate was collected by filtration (**3b**, 0.128g, 52% yield.) ¹H NMR (CDCl₃): δ 0.97 (d, 9H, J=8.4 Hz, PMe₃), 2.04 (d, 1H, J=9.6 Hz, H4), 2.66 (dd, 1H, J=13.8, 9.6 Hz, H5), 2.80 (d, 1H, J=17.1 Hz, H2), 2.98 (dd, 1H, J=17.1, 7.5 Hz, H2), 3.72 (s, 3H, OMe), 3.92 (m, 2H, H1, H6), 4.90 (dd, 1H, J=9.0, 3.0 Hz, H7), 6.11 (t, 1H, J=2.1 Hz, Tp), 6.14 (t, 1H, J=2.1 Hz, Tp), 6.35 (t, 1H, J=2.1 Hz, Tp), 6.65 (d, 1H, J=2.1 Hz, Tp), 6.83-6.92 (m, 3H), 7.24 (m, 1H), 7.52 (d, 1H, J=2.1 Hz, Tp), 7.53 (d, 1H, J=2.1 Hz, Tp), 7.62 (d, 1H, J=2.1 Hz, Tp), 7.74 (d, 1H, J=2.1 Hz, Tp), 8.11 (d, 1H, J=2.1 Hz, Tp). ¹³C NMR (CDCl₃): δ 12.7 (d, J=29 Hz, PMe₃), 32.9 (s, C2), 33.3 (d, J=4 Hz, C6), 55.4 (s, OMe), 57.1 (s, C1), 61.2 (s, C4), 61.5 (d, J=12.0, C5), 71.4 (s, C7), 106.0 (s, Tp), 106.7 (s, Tp), 107.3 (s, Tp), 114.2 (s. Ph), 114.8 (s, Ph), 122.6 (s, Ph), 129.4 (s, Ph), 135.2 (s, Ph), 136.2 (s, Tp), 136.8 (s, Tp), 137.1 (s, Tp), 139.2 (s, Tp), 143.1 (s, Tp), 143.6 (s, Tp), 159.5 (s, Ph), 205.4 (s, C3), 210.1 (s, C8). IR: $v_{NO} = 1565$, $v_{CO} = 1600$, $v_{CO} = 1766$ cm⁻¹. CV: $E_{p,a} = +0.94$ V. Anal Calc'd for $C_{27}H_{33}BN_7O_4PW \cdot CH_2Cl_2$: C, 40.51; H, 4.25; N, 11.91. Found: C, 40.82; H, 4.25; N, 11.81.

TpW(NO)(PMe₃)(η^2 -4,5-(7,7-dichlorobicyclo[4.2.0] oct-4-ene-3,8-dione)) (4)

Diisopropylethylamine (0.043g, 0.34 mmol) was dissolved in 10 mL CH₂Cl₂, added to **1b** (0.100g, 0.168 mmol) and stirred. Dichloroacetylchloride (0.049g, 0.34 mmol) was dissolved in 10 mL CH₂Cl₂ and slowly added dropwise via a separatory funnel over about 20 minutes. The reaction was removed from the glovebox and $HCl_{(aq)}$ (1M, 20 mL) was added. The reaction was extracted, dried with MgSO₄ and evaporated using rotatory evaporation. 2 mL EtOAc was added and the reaction was stirred. After 5 minutes the reaction was filtered, washed twice with 1 mL EtOAc, dried in vacuo. An light beige precipitate was isolated (4, 0.077g, 65% yield). ¹H NMR (CDCl₃): δ 1.18 (d, 9H, J=8.4 Hz, PMe₃), 2.18 (d, 1H, J=9.6 Hz, H4), 2.74 (d, 1H, J=17.5 Hz, H2), 2.89 (dd, 1H, J=17.5, 10.4 Hz, H2), 3.34 (dd, 1H, J= 12.9, 9.6 Hz, H5), 3.95 (d, 1H, J=10.4 Hz, H6), 4.27 (t, 1H, J=10.4 Hz, H1), 6.14 (t, J=2.1 Hz, Tp), 6.28 (t, J=2.1 Hz, Tp), 6.39 (t, J=2.1 Hz, Tp), 7.37 (d, J=2.1 Hz, Tp), 7.43 (d, J=2.1 Hz, Tp), 7.56 (d, J=2.1 Hz, Tp), 7.72 (d, J=2.1 Hz, Tp), 7.79 (d, J=2.1 Hz, Tp), 8.13 (d, J=2.1 Hz, Tp). ¹³C NMR (CDCl₃): $\delta 12.6$ (d, J=29 Hz), 32.0 (s), 49.3 (d, J=4 Hz), 54.9 (s), 59.5 (d, J=12 Hz), 60.1(s), 93.3(s), 106.3(s), 107.0 (s), 107.5 (s), 136.5 (s), 137.1 (s), 137.3 (s), 139.9 (s), 143.2 (s), 143.7 (s), 197.1 (s), 203.8 (s). ³¹P NMR (CDCl₃): δ -10.1. IR: v_{NO} = 1563, v_{CO} = 1621, $v_{CO} = 1798 \text{ cm}^{-1}$. CV: $E_{p,a} = +1.18 \text{ V}$. Anal Calc'd for $C_{20}H_{25}BCl_2N_7O_3PW$: C, 33.93; H, 3.56; N, 13.85. Found: C, 33.65; H, 3.61; N, 13.36.

TpW(NO)(PMe₃)(η^2 -4,5-(8-hydroxy-7-(3-methoxyphenyl)bicyclo[4.2.0] oct-4-ene-3-one)) (5)

A solution of **3a** (0.050g, 0.067 mmol) in 0.5 mL of 1:1 CH₂Cl₂: MeOH was added to NaBH₄. After 5 minutes 1 mL H₂O was added. The CH₂Cl₂ layer was extracted, dried over MgSO₄ and evaporated by rotatory evaporation. The residue was dissolved in 0.25 mL CH₂Cl₂, precipitated into 50 mL hexanes, filtered and dried in *vacuo*. The precipitate was isolated as an off-white solid (5, 0.33g, 66%). ¹H NMR (CDCl₃): 80.87 (d, 9H, J=8.1 Hz, PMe₃), 2.23 (d, 1H, J=9.6 Hz, H4), 2.41 (d, 1H, J=2.7 Hz, OH), 2.50 (d, 1H, J=17.1 Hz, H2), 3.00 (dd, 1H, J=17.1, 7.8 Hz, H2), 3.09 (dd, 1H, J=14.1, 9.6 Hz, H5), 3.27 (q, 1H, J= 7.8 Hz, H1), 3.74 (s, OMe), 3.93 (m, 2H, H6, H7), 4.79 (m, 1H, H8), 6.14 (m, 2H), 6.32 (t, 1H, J=2.1 Hz), 6.83 (m, 1H), 6.91 (d, 1H, J=2.1 Hz), 7.18 (m, 1H), 7.52 (m, 2H), 7.62 (m, 2H), 7.72 (d, J=2.1 Hz), 8.08 (d, J=2.1 Hz), ¹³C NMR (CDCl₃): δ12.7 (d, J=29 Hz, PMe₃), 33.5 (s, C2), 39.6 (s, C3), 39.8 (d, J=4 Hz, C4), 51.7 (s, C8), 55.4 (s, OMe), 61.6 (s, C6), 63.9 (d, J=11 Hz, C5), 73.5 (s, C7), 106.2 (s,Tp), 106.6 (s,Tp), 107.1 (s,Tp), 113.7 (s), 116.3 (s), 124.0 (s), 128.8 (s, 2C), 135.9 (s), 136.6 (s), 136.8 (s), 143.6(s, 3C), 159.0 (s), 210.6 (s). IR: $v_{NO} = 1563$, $v_{CO} = 1595$, $v_{OH} = 1563$ 3498 cm⁻¹. CV: $E_{p,a} = +0.74$ V. Anal Calc'd for $C_{27}H_{35}BN_7O_4PW$: C, 43.40; H, 4.72; N, 13.12. Found: C, 43.02; H, 4.66; N, 12.83.

8-hydroxy-7-(3-methoxyphenyl)bicyclo[4.2.0] oct-4-ene-3-one (6)

A solution of N-Bromosuccinimide (0.019g, 0.11 mmol) in 0.5 mL CH₃CN and 0.5 mL CH₂Cl₂ was added to added to **6** (0.040g, 0.054 mmol). After 10 minutes the solvent was removed by rotatory evaporation. The reaction was chromatographed using a 20 x 20, 500 μ M preparatory TLC plate using 1:1 EtOAc:hexanes as the eluent. Compound **7** (0.0063g, 48%, R_f= 0.24) was isolated as a colorless oil. The recovery of the metal complex was not attempted. ¹H NMR (CDCl₃): δ 1.95 (br s, 1H), 2.49 (dd, 1H, J=18.0, 7.8 Hz), 2.64 (dd, 1H, J=18.0, 1.8 Hz), 3.25 (q, 1H, J=6.3 Hz), 3.49 (m, 1H), 3.79 (s, 3H), 4.00 (dd, 1H, J=9.3, 5.7 Hz), 4.71 (td, 1H, J=5.7, 2.7 Hz), 6.18 (dd, 1H, J=10.5, 2.1 Hz), 6.80-6.95 (m, 4H), 7.24 (t, 1H, J=8.1 Hz). ¹³C NMR (CDCl₃): δ 33.8 (s), 35.1 (s), 38.4 (s), 49.5 (s), 55.4 (s), 74.7 (s), 112.5 (s), 115.7 (s), 122.1 (s), 129.6 (s), 130.5 (s), 137.3 (s), 149.7 (s), 159.8 (s), 198.4 (s). IR: ν_{CO} = 1662, ν_{OH} = 3415 cm⁻¹. HRMS:[C₁₅H₁₆O₃ + Na⁺] obs'd (%), calc'd (%), diff. in ppm: 267.09938 (100), 267.09917 (100), 0.8.

TpW(NO)(PMe₃)(η^2 -4,5-(3-oxobicyclo[4.1.0]hept-4-ene-7-carboxylic acid methyl ester)) (7)

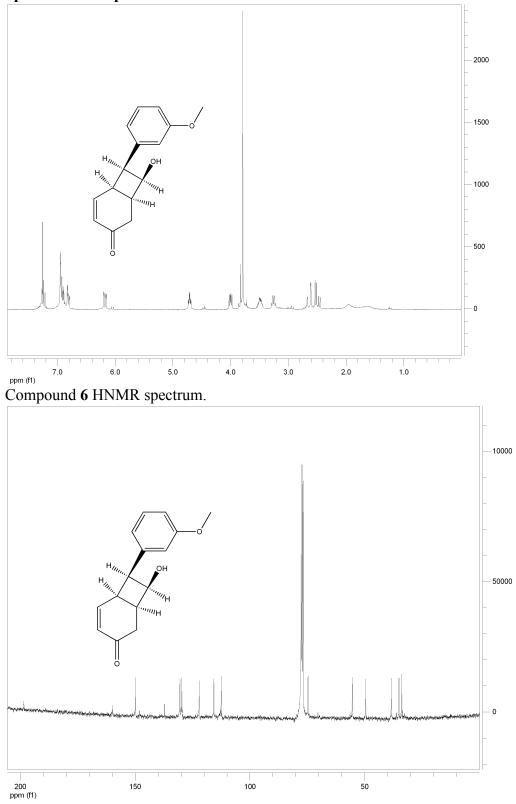
A mixture of LiOMe (0.005g, 0.13 mmol) in 0.25 mL methanol was added to a solution of **2a,b** (0.050g, 0.074 mmol) in 0.25 mL CH₂Cl₂. After 3 hours the solvent was removed by rotatory evaporation. The residue was dissolved in 1 mL CH₂Cl₂ and extracted with 3 mL water. The organic layer was dried with MgSO₄, filtered and removed by rotatory evaporation. The residue was dissolved in 0.25 mL methylene chloride, then 2 mL Et₂O and 4 mL hexanes were added. The reaction was filtered and dried *in vacuo*. An off-white precipitate (**7**, 0.029g, 58% yield) was isolated. ¹H NMR (CDCl₃): δ 1.21 (d, 9H, J=8.7 Hz, PMe₃), 1.65 (dd, 1H, J=4.5, 3.6 Hz, H7), 1.97 (d, 1H, J=9.6 Hz, H4), 2.08 (m,

1H, H1), 2.21 (dd, 1H, J=9.3, 3.6 Hz, H6), 2.68 (d, 1H, J=18.0 Hz, H2), 3.01 (dd, 1H, J=18.0, 5.4 Hz, H2), 3.47 (dd, 1H, J=12.0, 9.6 Hz, H5), 3.68 (s, 3H, OMe), 6.12 (t, 1H, J=2.4 Hz, Tp), 6.21 (t, 1H, J=2.4 Hz, Tp), 6.36 (t, 1H, J=2.4 Hz, Tp), 7.28 (d, 1H, J=2.4 Hz, Tp), 7.53 (d, 2H, J=2.4 Hz, Tp), 7.68 (d, 1H, J=2.4 Hz, Tp), 7.75 (d, 1H, J=2.4 Hz, Tp), 8.15 (d, 1H, J=2.4 Hz, Tp). ¹³C NMR (CDCl₃): δ 13.1 (d, J=29 Hz, PMe₃), 24.7 (s, C1), 29.6 (d, J=3 Hz, C6), 31.9 (s, C7), 35.4 (c, C2), 51.7 (s, Me), 60.2 (s, C4), 63.9 (s, C5), 106.1 (s, Tp), 106.5 (s, Tp), 107.3 (s, Tp), 136.0 (s, Tp), 140.0 (s, Tp), 143.0 (s, Tp), 143.8 (s, Tp), 174.7 (s, C8), 207.2 (s, C3). IR: ν_{NO} = 1571, ν_{CO} = 1598 ν_{CO} = 1712cm⁻¹. CV: $E_{p,a}$ = +0.86 V. Anal Calc'd for C₂₁H₂₉BN₇O₄PW: C, 37.69; H, 4.37; N, 14.65. Found: C, 37.34; H, 4.49; N, 14.23.

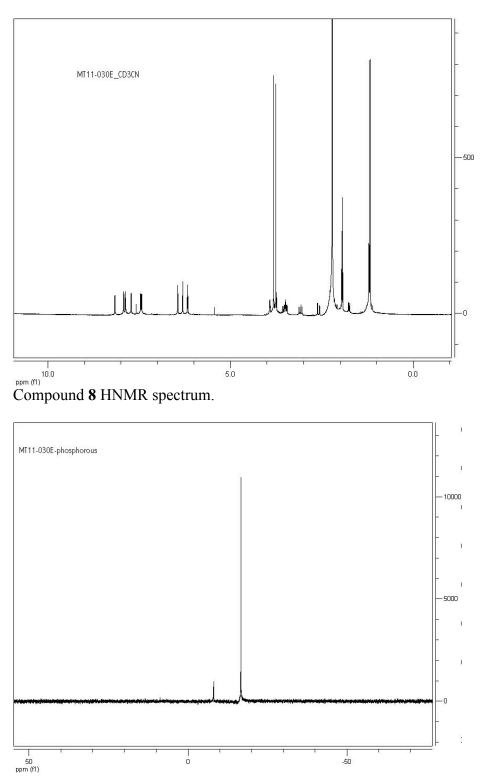
TpW(NO)(PMe₃)(η^2 -4,5-(3-Oxo-bicyclo[4.2.0]octa-4,7-diene-7,8-dicarboxylic acid dimethyl ester)) (8)

Dimethyl acetylenedicarboxylate (0.146g, 1.03 mmol) was dissolved in 0.25 mL CH₂Cl₂ and added to **1b** (0.050g, 0.084 mmol) and heated at 50 °C. After 6.5 h the solvent was removed by applying a N_2 stream under vacuum. When the residue was completely dry 0.25 mL CH₂Cl₂ was added and this soltuion was added to 50 mL hexanes. The yellow precipitate was collected by filtration and dried in vacuo. The precipitate (0.057g) contained mostly 9 with traces of several other very minor complexes. This mixture was purified by preparatory thin layed chromatography using a 20 x 20 cm, 500 μ M silica plate with ethyl actate as the mobile phase. 8 (0.0085g) was isolated from a band ($R_f = 0.49$) that contained a 40% paramagnetic impurity by electrochemistry. ¹H NMR (CD₃CN): δ1.19 (d, 9H, J=8.7 Hz, PMe₃), 1.75 (d, 1H, J=9.9 Hz), 2.59 (d, 1H, J=16.8 Hz), 3.08 (dd, 1H, J=16.8, 9.0 Hz), 3.47 (dd, 1H, J=9.0, 4.2 Hz), 3.53 (dd, 1H, J=12.9, 9.9 Hz), 3.76 (s, 3H), 3.82 (s, 3H), 3.92 (d, 1H, J=4.2), 6.18 (t, 1H, J=2.4 Hz, Tp), 6.31 (t, 1H, J=2.4 Hz, Tp), 6.45 (t, 1H, J=2.4 Hz, Tp), 7.44 (d, 1H, J=2.4 Hz, Tp), 7.46 (d, 1H, J=2.4 Hz, Tp), 7.73 (d, 1H, J=2.4 Hz, Tp), 7.88 (d, 1H, J=2.4 Hz, Tp), 7.93 (d, 1H, J=2.4 Hz, Tp), 8.17 (d, 1H, J=2.4 Hz, Tp). ¹³C NMR (CDCl₃): δ13.0 (d, J=29 Hz), 31.2 (s), 36.3 (s), 41.9 (s), 42.5 (s), 52.0 (s), 52.2 (s), 60.8 (s), 62.3 (d, J=12 Hz), 106.3 (s), 106.6 (s), 107.3 (s), 136.1 (s), 136.8 (s), 137.0 (s), 140.1 (s), 143.0 (s), 143.5 (s), 145.2 (s), 149.6 (s), 161.7 (s), 162.4 (s), 208.0 (s). ³¹P NMR (CD₃CN): δ-8.0. IR: $v_{NO} = 1567$, $v_{CO} = 1618$, $v_{CO} = 1713$ cm⁻¹. CV: $E_{p,a} = +1.03$ V. Ms (ESI): [MH+] = 740.

Spectra for compounds 6 and 8.



Compound 6¹³C NMR spectrum.



Compound 8³¹P NMR spectrum.

