Supporting Information

ansa-Metallocene Based Cyclic[2]pyrroles:

S. Ramakrishnan, Alagar Srinivasan*

Photosciences and Photonics Section, Chemical Sciences and Technology Division, National Institute for Interdisciplinary Science and Technology (NIIST – CSIR), Thiruvananthapuram – 695 019, Kerala.

E-mail address: indiansrini@gmail.com

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Synthesis of 6:

1, 1'-Bis (diphenylhydroxymethyl)ferrocene (1) (1.5 g, 2.73 mmol) and pyrrole (7.6 ml, 109.1 mmol) were stirred under nitrogen atmosphere for 10 min at room temperature. BF₃.OEt₂ (0.04 ml, 0.273 mmol) was added to the above mixture and the solution was stirred at room temperature for further 1h under dark conditions. Then it is quenched with 75 ml dichloromethane. The product was filtered and thoroughly washed with pet ether to remove excess pyrrole and yellow residue was identified as **6**. Yield: 80% (1.42 g); m.p: 230° C; ¹H NMR (300 MHz, CDCl₃): δ = 8.66 (brs, 2H, pyrrole NH), 7.19-7.14 (m, 12H, phenyl CH), 6.95-6.92 (m, 8H, phenyl CH), 6.69 (s, 2H, α-pyrrole CH), 6.10 (q, 2H, β-pyrrole CH), 5.75 (s, 2H, β-pyrrole CH), 3.88 (s, 4H, ferrocenyl CH); ¹³C NMR (400 MHz, CDCl₃): δ = 54.70, 70.61, 70.85, 97.48, 108.03, 108.47, 109.98, 116.44, 117.98, 126.85, 127.65, 129.64, 136.64, 148.00; FT-IR (CH₂Cl₂) 3428.57, 2917.58, 1637.87, 1486.99, 1262.03, 1019.20 cm⁻¹; FAB-MS (m/z): 648.

Synthesis of 7:

1, 1'-Bis (methyl-phenylhydroxymethyl)ferrocene (2) (0.5 g, 1.2 mmol) and pyrrole (3.3 ml, 47 mmol) were stirred under nitrogen atmosphere for 10 min at room temperature. BF₃.OEt₂ (0.015 ml, 0.12 mmol) was added to the above mixture and the solution was stirred at room temperature for further 1h under dark conditions. Then it is quenched with 75 ml dichloromethane. The work up was done using NaOH solution, water and the organic layer was extracted with dichloromethane and dried over anhydrous sodium sulphate. After removal of the solvent, the crude product was purified by silica gel column chromatography (100 – 200 mesh). A yellow fraction eluted with ethyl acetate: petroleum ether (3 : 97) gave a pale yellow solid identified as **7**. Yield: 65% (0.4 g); m.p: 164° C; ¹H NMR (300 MHz, CDCl₃): δ = 7.64 (brs, 2H, pyrrole NH), 7.18 (m, 10H, phenyl CH), 6.99 (m, 4H, phenyl CH), 6.61 (s, 2H, α-pyrrole CH), 6.13 (t, 2H, β-pyrrole CH), 5.96 (s, 2H, β-pyrrole CH), 3.99 (s, 4H, ferrocenyl CH), 3.90 (t, 3H, ferrocenyl CH), 3.82 (s, 1H, ferrocenyl CH), 1.94 (s, 6H, CH₃); ¹³C NMR (400 MHz, CDCl₃): δ = 28.77, 43.29, 68.30, 68.35, 68.41, 68.47, 68.86, 69.02, 98.12, 105.66, 107.46, 115.99, 126.09, 139.64, 148.80; FT-IR (CH₂Cl₂) 3436.18, 2979.86, 1593.98, 1492.47, 1444.84, 1366.28, 1265.28, 1025.07 cm⁻¹; FAB-MS (m/z): 524.06.

Synthesis of 8:

1, 1'-Bis(cyclohexylhydroxymethyl)ferrocene (**3**) (0.8 g, 2.1 mmol) and pyrrole (5.8 ml, 84 mmol) were stirred under nitrogen atmosphere for 10 min at room temperature. BF₃.OEt₂ (0.03 ml, 0.21 mmol) was added to the above mixture and the solution was stirred at room temperature for further 1h under dark conditions. Then it is quenched with 75 ml dichloromethane. After removal of the solvent, the crude product was purified by silica gel column chromatography (100 – 200 mesh). A yellow fraction eluted with ethyl acetate : petroleum ether (3.5 : 96.5) gave a pale yellow solid identified as **8**. Yield: 80% (0.8 g); m.p: 145° C; ¹H NMR (300 MHz, CDCl₃) δ = 7.93 (brs, 2H, pyrrole NH), 6.67 (s, 2H, α-pyrrole CH), 6.15 (d, J = 2.73 Hz, 2H, β-pyrrole CH), 6.02 (s, 2H, β-pyrrole CH), 3.97 (s, 4H, ferrocenyl CH), 3.84 (s, 4H, ferrocenyl CH), 2.23 (d, J : 13.3 Hz, 4H, cyclohexyl CH), 1.77 (m, 5H, cyclohexyl CH), 1.42 (m, 6H, cyclohexyl CH), 1.23 (m, 3H, cyclohexyl CH), 0.88 (m, 2H, cyclohexyl CH); ¹³C NMR (400 MHz, CDCl₃): $\delta = 22.46$, 22.86, 23.45, 26.75, 27.85, 28.45, 29.50, 35.15, 36.17, 38.26, 42.57,

64.75, 65.17, 67.17, 68.19, 69.15, 128.45, 137.75, 147.56; FT-IR (CH₂Cl₂) 3428.94, 3093.40, 2929.48, 1591.23, 1443.09, 1264.77, 1031.13 cm⁻¹; FAB-MS (m/z): 480.75. **Synthesis of 9**:

1, 1'-Bis(diethylhydroxymethyl)ferrocene (**4**) (0.8 g, 2.2 mmol) and pyrrole (6.2 ml, 88.8 mmol) were stirred under nitrogen atmosphere for 10 min at room temperature. BF₃.OEt₂ (0.03 ml, 0.22 mmol) was added to the above mixture and the solution was stirred at room temperature for further 1h under dark conditions. Then it is quenched with 75 ml dichloromethane. After removal of the solvent, the crude product was purified by silica gel column chromatography (100 – 200 mesh). A yellow fraction eluted with ethyl acetate: petroleum ether (3.5 : 96.5) gave a pale yellow solid identified as **9**. Yield: 68% (0.679 g); m.p: 122° C; ¹H NMR (300 MHz, CDCl₃) δ = 8.97 (brs, 2H, pyrrole NH), 6.74 (s, 2H, α-pyrrole CH), 6.17 (d, J : 2.45 Hz, 2H, β-pyrrole CH), 6.02 (s, 2H, β-pyrrole CH), 4.15 (s, 4H, ferrocenyl CH), 3.93 (s, 4H, ferrocenyl CH) , 1.87 (m, 8H, ethyl CH), 0.7 (t, 12H, ethyl CH); ¹³C NMR (400 MHz, CDCl₃): δ = 9.21, 32.07, 41.99, 67.92, 68.68, 106.28, 107.45, 115.86, 136.31; FT-IR (CH₂Cl₂) 3426.77, 3097.84, 2967.48, 2877.34, 1556.44, 1463.89, 1082.47, 1036.45, 827.36, 786.10 cm⁻¹; FAB-MS (m/z): 456.74.

Synthesis of 10:

1, 1'-Bis(dimethylhydroxymethyl)ferrocene (5) (0.88 g, 2.91 mmol) and pyrrole (8 ml, 116.6 mmol) were stirred under nitrogen atmosphere for 10 min at room temperature. BF₃.OEt₂ (0.04 ml, 0.291 mmol) was added to the above mixture and the solution was stirred at room temperature for further 1h under dark conditions. Then it is quenched with 75 ml dichloromethane. After removal of the solvent, the crude product was purified by silica gel column chromatography (100 – 200 mesh). A yellow fraction

eluted with ethyl acetate : petroleum ether (3 : 97) gave a pale yellow solid identified as **10**. Yield: 65% (0.76 g); m.p: 108° C; ¹H NMR (300 MHz, CDCl₃) δ = 7.76 (brs, 2H, pyrrole NH), 6.48 (q, 2H, α-pyrrole CH), 6.01 (q, 2H, β-pyrrole CH), 5.85 (s, 2H, βpyrrole CH), 4.09 (s, 4H, ferrocenyl CH), 4.02 (s, 4H, ferrocenyl CH), 1.56 (s, 12H, methyl CH); ¹³C NMR (400 MHz, CDCl₃): δ = 29.98, 34.24, 68.25, 68.32, 99.37, 102.71, 107.77, 115.57, 140.86; FT-IR (CH₂Cl₂) 3417.58, 2917.58, 1594.01, 1262.03, 1119.38, 1023.36 cm⁻¹; FAB-MS (m/z): 400.18.

Synthesis of 24:

1, 1'-Bis(diphenylhydroxymethyl)ruthenocene (**23**) (1 g, 1.7 mmol) and pyrrole (5 ml, 68 mmol) were stirred under nitrogen atmosphere for 10 min at room temperature. BF₃.OEt₂ (0.02 ml, 0.17 mmol) was added to the above mixture and the solution was stirred at room temperature for further 1h under dark conditions. Then it is quenched with 75 ml dichloromethane. The product was filtered and thoroughly washed with pet ether to remove excess pyrrole and white residue was identified as **24**. Yield: 81% (0.95 g); m.p: 194° C; ¹H NMR (300 MHz, CDCl₃) δ = 9.23 (brs, 2H, pyrrole NH), 7.21 (m, 12H, phenyl CH), 7.01 (m, 8H, phenyl CH), 6.70 (t, 2H, α-pyrrole CH), 6.10 (t, 2H, β-pyrrole CH), 5.58 (t, 2H, β-pyrrole CH), 4.39 (s, 4H, ferrocenyl CH), 4.24 (s, 4H, ferrocenyl CH); ¹³C NMR (400 MHz, CDCl₃): δ = 73.66, 74.27, 100.76, 107.54, 109.51, 115.81, 126.45, 127.24, 129.23, 147.89; FT-IR (CH₂Cl₂) 3433.77, 2923.07, 1593.98, 1259.29, 1116.63, 1027.70 cm⁻¹; FAB-MS (m/z): 694.10.

Synthesis of 11:

6 (0.5 g, 0.77 mmol) and acetone (60 ml) were stirred under nitrogen atmosphere for 10 min at room temperature. Trifluoroacetic acid (0.006 ml, 0.077 mmol) was added

to the above mixture and the solution was stirred at room temperature for 2 days. After removal of the solvent, the crude product was purified by silica gel column chromatography (100 – 200 mesh). A yellow fraction eluted with Ethyl acetate : petroleum ether (1 : 99) gave a pale yellow solid identified as **11**. Yield: 48 % (0.26 g); m.p: 145° C; ¹H NMR (300 MHz, CDCl₃) δ = 9.24 (brs, 2H, pyrrole NH), 7.31 (m, 16H, phenyl CH), 6.58 (d, J = 9.2 Hz, 4H, phenyl CH), 6.02 (s, 2H, β-pyrrole CH), 5.07 (s, 2H, β-pyrrole CH), 4.10 (d, J = 11.8 Hz, 4H, ferrocenyl CH), 2.94 (s, 2H, ferrocenyl CH), 2.81 (s, 2H, ferrocenyl CH), 1.79 (s, 6H, methyl CH); ¹³C NMR (400 MHz, CDCl₃): δ = 29.29, 33.45, 42.45, 55.80, 66.5, 71.80, 72.45, 99.58, 110.04, 111.57, 112.19, 119.55, 119.78, 124.76, 128.76, 130.75, 135.78; FT-IR (CH₂Cl₂) 3431.36, 3065.93, 2924.00, 2853.25, 1597.45, 1492.09, 1443.09, 1415.66, 1275.75, 1220.88, 1048.43, 842.30 cm⁻¹; FAB-MS (m/z): 689.11.

The above macrocycle was also be obtained by the following method. **1** (0.2 g, 0.36 mmol) and **19** (0.063 g, 0.36 mmol) were allowed to stir under nitrogen atmosphere in 30 ml dry dichloromethane for 10 min. Then *p*-TSA (0.017 g, 0.09 mmol) were added to the above reaction mixture and allowed to stir for 2 days. Then the purification was done as above.

Synthesis of 12:

7 (0.26 g, 0.5 mmol) and acetone (37 ml) were stirred under nitrogen atmosphere for 10 min at room temperature. Trifluoroacetic acid (0.004 ml, 0.05 mmol) was added to the above mixture and the solution was stirred at room temperature for 45 min. After removal of the solvent, the crude product was purified by silica gel column chromatography (100 – 200 mesh). A yellow fraction eluted with Ethyl acetate : petroleum ether (1 : 99) gave a pale yellow solid identified as **12**.Yield: 19% (0.05 g); m.p: 140° C; ¹H NMR (300 MHz, CDCl₃) δ = 7.75 (brs, 2H, pyrrole NH), 7.13 (d, J = 6.765 Hz, 6H, phenyl CH), 6.90 (d, J = 7.836 Hz, 4H, phenyl CH), 5.96 (t, 2H, β-pyrrole CH), 5.46 (t, 2H, β-pyrrole CH), 4.25 (s, 4H, ferrocenyl CH), 4.17 (s, 2H, ferrocenyl CH), 3.18 (s, 2H, ferrocenyl CH), 1.97 (s, 6H, methyl CH), 1.65 (s, 6H, methyl CH); ¹³C NMR (400 MHz, CDCl₃): δ = 27.45, 29.47, 34.5, 41.5, 68.40, 68.69, 69.45, 69.76, 70.15, 99.15, 106.70, 107.05, 113.25, 127.11, 138.75; FT-IR (CH₂Cl₂) 3428.57, 3313.18, 2917.58, 2851.64, 1596.43, 1445.84, 1119.38, 1086.46 cm⁻¹; FAB-MS (m/z): 564.37.

The above macrocycle was also be obtained by the following method.2 (0.2 g, 0.47 mmol) and **19** (0.082 g, 0.47 mmol) were allowed to stir under nitrogen atmosphere in 30 ml dry dichloromethane for 10 min. Then *p*-TSA (0.022 g, 0.12 mmol) were added to the above reaction mixture and allowed to stir for 45 min. Then the purification was done as above.

Synthesis of 13:

8 (0.2 g, 0.42 mmol) and acetone (31 ml) were stirred under nitrogen atmosphere for 10 min at room temperature. Trifluoroacetic acid (0.003 ml, 0.042 mmol) was added to the above mixture and the solution was stirred at room temperature for 45 min. After removal of the solvent, the crude product was purified by silica gel column chromatography (100 – 200 mesh). A yellow fraction eluted with Ethyl acetate : petroleum ether (1 : 99) gave a pale yellow solid identified as **13**. Yield: 22% (0.047g); m.p: 123° C; ¹H NMR (300 MHz, CDCl₃) δ = 7.70 (brs, 2H, pyrrole NH), 6.12 (t, 2H, βpyrrole CH), 5.88 (q, 2H, β-pyrrole CH), 4.06 (m, 6H, ferrocenyl CH), 2.86 (d, J = 0.897 Hz, 2H, ferrocenyl CH), 2.15 (m, 5H, cyclohexyl CH), 1.74 (m, 5H, cyclohexyl CH), 1.6 (s, 6H, methyl CH), 1.38 (m, 6H, cyclohexyl CH), 0.87 (m, 4H, cyclohexyl CH); ¹³C NMR (400 MHz, CDCl₃): δ = 21.15, 22.75, 23.16, 26.79, 28.95, 29.05, 29.75, 36.15, 36.27, 39.45, 42.75, 65.66, 66.17, 67.18, 68.25, 69.16, 127.15, 138.45; FT-IR (CH₂Cl₂) 3456.04, 3093.40, 2925.44, 2854.50, 1574.77, 1446.97, 1415.66, 1264.77, 1187.96, 1042.56, 930.08 cm⁻¹; FAB-MS (m/z): 520.33.

The above macrocycle was also be obtained by the following method.3 (0.2 g, 0.52 mmol) and **19** (0.091 g, 0.52 mmol) were allowed to stir under nitrogen atmosphere in 30 ml dry dichloromethane for 10 min. Then *p*-TSA (0.025 g, 0.13 mmol) were added to the above reaction mixture and allowed to stir for 45 min. Then the purification was done as above.

Synthesis of 14:

9 (0.45 g, 1 mmol) and acetone (73 ml) were stirred under nitrogen atmosphere for 10 min at room temperature. Trifluoroacetic acid (0.007 ml, 0.1 mmol) was added to the above mixture and the solution was stirred at room temperature for 45 min. After removal of the solvent, the crude product was purified by silica gel column chromatography (100 – 200 mesh). A yellow fraction eluted with Ethyl acetate : petroleum ether (1 : 99) gave a pale yellow solid identified as **14**.Yield: 19% (0.094 g); m.p: 120° C; ¹H NMR (300 MHz, CDCl₃) δ = 8.23 (brs, 2H, pyrrole NH), 6.06 (t, 2H, βpyrrole CH), 5.90 (t, 2H, β-pyrrole CH), 4.19 (d, J = 1.164 Hz, 2H, ferrocenyl CH), 4.10 (m, 4H, ferrocenyl CH), 3.02 (s, 2H, ferrocenyl CH), 1.94 (m, 8H, ethyl CH), 1.66 (s, 6H, methyl CH), 0.79 (t, 6H, ethyl CH), 0.71 (t, 6H, ethyl CH); ¹³C NMR (400 MHz, CDCl₃) δ = 9.18, 30.58, 31.51, 33.45, 36.23, 41.24, 65.57, 66.45, 67.45, 69.34, 102.59, 103.20, 105.84, 134.62, 136.94; FT-IR (CH₂Cl₂) 3434.06, 3329.67, 2922.90, 2851.64, 1742.12, 1585.75, 1259.29, 1119.38, 1018.49 cm⁻¹; FAB-MS (m/z): 496.60.

The above macrocycle was also be obtained by the following method. 4 (0.2 g, 0.56 mmol) and **19** (0.097 g, 0.56 mmol) were allowed to stir under nitrogen atmosphere in 30 ml dry dichloromethane for 10 min. Then *p*-TSA (0.027 g, 0.14 mmol) were added to the above reaction mixture and allowed to stir for 45 min. Then the purification was done as above.

Synthesis of 15:

10 (0.5 g, 1.25 mmol) and acetone (60 ml) were stirred under nitrogen atmosphere for 10 min at room temperature. Trifluoroacetic acid (0.01 ml, 0.125 mmol) was added to the above mixture and the solution was stirred at room temperature for 45 min. After removal of the solvent, the crude product was purified by silica gel column chromatography (100 – 200 mesh). A yellow fraction eluted with Ethyl acetate : petroleum ether (1 : 99) gave a pale yellow solid identified as **15**. Yield: 25% (0.13 g); m.p: 115° C; ¹H NMR (300 MHz, CDCl₃) δ = 7.88 (brs, 2H, pyrrole NH), 6.02 (s, 2H, βpyrrole CH), 5.82 (s, 2H, β-pyrrole CH), 4.07 (s, 2H, ferrocenyl CH), 3.97 (d, J = 11.3 Hz, 4H), 2.90 (s, 2H, ferrocenyl CH), 1.6 (s, 18H, methyl CH); ¹³C NMR (400 MHz, CDCl₃): δ = 24.56, 26.75, 27.17, 29.15, 35.47, 68.15, 68.42, 98.45, 113.45, 114.55, 126.77, 137.11; FT-IR (CH₂Cl₂) 3457.95, 3087.91, 2924.14, 2854.34, 1651.59, 1580.26, 1463.16, 1377.42, 1259.29, 1041.33, 828.58 cm⁻¹; FAB-MS (m/z): 439.98.

The above macrocycle was also be obtained by the following method.5 (0.2 g, 0.66 mmol) and **19** (0.12 g, 0.66 mmol) were allowed to stir under nitrogen atmosphere in 30 ml dry dichloromethane for 10 min. Then *p*-TSA (0.031 g, 0.17 mmol) were added

to the above reaction mixture and allowed to stir for 45 min. Then the purification was done as above.

Synthesis of 16:

7 (0.235 g, 0.45 mmol) and acetophenone (0.05ml, 0.45 mmol) were dissolved in 30 ml dry dichloromethane and stirred under nitrogen atmosphere for 10 min at room temperature. Trifluoroacetic acid (0.003 ml, 0.045 mmol) was added to the above mixture and the solution was stirred at room temperature for 45 min. After removal of the solvent, the crude product was purified by silica gel column chromatography (100 – 200 mesh). A yellow fraction eluted with Ethyl acetate : petroleum ether (1 : 99) gave a pale yellow solid identified as **16**. Yield: 23% (0.063 g); ¹H NMR (300 MHz, CDCl₃) δ = 8.03 (brs, 2H, pyrrole NH), 7.38 (d, J = 6.81 Hz, 8H, phenyl CH), 6.95 (d, J = 6.36 Hz, 3H, phenyl CH), 6.8 (d, J = 6.95 Hz, 4H, phenyl CH), 6.03 (t, 2H, β-pyrrole CH), 5.95 (t, 2H, βpyrrole CH), 4.03 (s, 4H, ferrocenyl CH), 3.38 (s, 2H, ferrocenyl CH), 3.04 (s, 2H, ferrocenyl CH), 1.91 (s, 9H, methyl CH); ¹³C NMR (400 MHz, CDCl₃): δ = 25.15, 26.17, 27.15, 35.15, 40.17, 66.70, 67.15, 67.45, 68.56, 69.15, 69.98, 98.15, 107.17, 108.05, 114.15, 128.11, 139.34; FT-IR (CH₂Cl₂) 3438.11, 2923.44, 2846.15, 1593.06, 1465.04, 1264.77, 1215.39, 1122.12, 1031.59 cm⁻¹; FAB-MS (m/z): 626.39.

The above macrocycle was also be obtained by the following method.2 (0.2 g, 0.47 mmol) and **20** (0.11 g, 0.47 mmol) were allowed to stir under nitrogen atmosphere in 30 ml dry dichloromethane for 10 min. Then *p*-TSA (0.022 g, 0.12 mmol) were added to the above reaction mixture and allowed to stir for 45 min. Then the purification was done as above.

Synthesis of 17:

8 (0.2g, 0.42 mmol) and cyclohexanone (0.04ml, 0.42 mmol) were dissolved in 30 ml dry dichloromethane and stirred under nitrogen atmosphere for 10 min at room temperature. Trifluoroacetic acid (0.003 ml, 0.042 mmol) was added to the above mixture and the solution was stirred at room temperature for 45 min. After removal of the solvent, the crude product was purified by silica gel column chromatography (100 - 200 mesh). A vellow fraction eluted with Ethyl acetate : petroleum ether (1 : 99) gave a pale vellow solid identified as 17. Yield: 25% (0.058g); m.p: 205° C: ¹H NMR (300 MHz, CDCl₃): δ = 7.50 (brs, 2H, pyrrole NH), 6.20 (t, 2H, β -pyrrole CH), 5.87 (t, 2H, β -pyrrole CH), 4.08 (s, 2H, ferrocenyl CH), 4.04 (s, 2H, ferrocenyl CH), 3.99 (s, 2H, ferrocenyl CH), 2.97 (s, 2H, ferrocenyl CH), 2.28 (m, 3H, cyclohexyl CH), 2.08 (t, 6H, cyclohexyl CH), 1.83 (m, 10H, cyclohexyl CH), 1.26 (m, 9H, cyclohexyl CH), 0.88 (m, 2H, cyclohexyl CH); ¹³C NMR (400 MHz, CDCl₃) $\delta = 22.59$, 22.78, 22.95, 23.15, 23.36, 25.94, 26.06, 26.14, 29.67, 37.88, 38.73, 39.01, 39.36, 39.78, 40.29, 40.84, 63.23, 65.10, 65.98, 66.16, 67.09, 68.55, 105.14, 105.32, 105.43, 106.37, 133.85, 136.44; FT-IR (CH₂Cl₂) 3450.54, 2924.11, 2851.64, 1640.61, 1443.09, 1215.39, 1119.38, 1042.56, 930.08 cm⁻¹; FAB-MS (m/z): 560.06.

The above macrocycle was also be obtained by the following method.3 (0.2 g, 0.52 mmol) and **21** (0.11 g, 0.52 mmol) were allowed to stir under nitrogen atmosphere in 30 ml dry dichloromethane for 10 min. Then *p*-TSA (0.025 g, 0.13 mmol) were added to the above reaction mixture and allowed to stir for 45 min. Then the purification was done as above.

Synthesis of 18:

9 (0.2 g, 0.42 mmol) and 3-pentanone (0.05 ml, 0.42 mmol) were dissolved in 30 ml dry dichloromethane and stirred under nitrogen atmosphere for 10 min at room temperature. Trifluoroacetic acid (0.003 ml, 0.042 mmol) was added to the above mixture and the solution was stirred at room temperature for 45 min. After removal of the solvent, the crude product was purified by silica gel column chromatography (100 – 200 mesh). A yellow fraction eluted with Ethyl acetate : petroleum ether (1 : 99) gave a pale yellow solid identified as **18**. Yield: 17% (0.040g); m.p: 115° C; ¹H NMR (300 MHz, CDCl₃): δ = 8.33 (brs, 2H, pyrrole NH), 5.98 (t, 2H, β-pyrrole CH), 5.93 (t, 2H, β-pyrrole CH), 4.20 (s, 2H, ferrocenyl CH), 4.09 (d, J = 7.5 Hz, 4H, ferrocenyl CH), 2.99 (s, 2H, ferrocenyl CH), 1.97 (m, 12H, ethyl CH), 0.76 (t, 18 H, ethyl CH); ¹³C NMR (400 MHz, CDCl₃): δ = 9.15, 10.15, 15.16, 31.15, 32.41, 35.15, 42.15, 66.78, 67.15, 68.11, 69.15, 104.15, 105.16, 133.15, 135.78; FT-IR (CH₂Cl₂) 3429.52, 3321.25, 2947.16, 1676.87, 1560.08, 1260.00, 1085.15, 1040.87, 819.56 cm⁻¹; FAB-MS (m/z): 524.33.

The above macrocycle was also be obtained by the following method.4 (0.2 g, 0.56 mmol) and 22 (0.113 g, 0.56 mmol) were allowed to stir under nitrogen atmosphere in 30 ml dry dichloromethane for 10 min. Then *p*-TSA (0.026 g, 0.14 mmol) were added to the above reaction mixture and allowed to stir for 45 min Then the purification was done as above.

Synthesis of 25:

24 (0.2 g, 0.29 mmol) and acetone (21 ml) were stirred under nitrogen atmosphere for 10 min at room temperature. Trifluoroacetic acid (0.002 ml, 0.029 mmol) was added to the above mixture and the solution was stirred at room temperature for 2 days. After

removal of the solvent, the crude product was purified by silica gel column chromatography (100 – 200 mesh). The colourless fraction eluted with Ethyl acetate : petroleum ether (1 : 99) gave a pale yellow solid identified as **25**. Yield: 23.5 % (0.05 g); m.p: 135° C; ¹H NMR (300 MHz, CDCl₃) δ = 8.86 (brs, 2H, pyrrole NH), 7.31 (m, 16H, phenyl CH), 6.91 (m, 4H, phenyl CH), 5.96 (t, 2H, β-pyrrole CH), 5.17 (t, 2H, β-pyrrole CH), 4.44 (s, 2H, ferrocenyl CH), 4.38 (s, 2H, ferrocenyl CH), 3.66 (s, 2H, ferrocenyl CH), 3.43 (s, 2H, ferrocenyl CH); 1.72 (s, 6H, methyl CH); ¹³C NMR (400 MHz, CDCl₃): δ = 25.45, 26.17, 28.17, 42.35, 69.15, 71.15, 72.46, 100.15, 108.91, 116.45, 127.15, 136.15; FT-IR (CH₂Cl₂) 3372.49, 3049.45, 2920.67, 2846.15, 1596.87, 1486.99, 1443.09, 1273.00, 1217.96, 1037.07, 814.86 cm⁻¹; FAB-MS (m/z): 734.22.

The above macrocycle was also be obtained by the following method.23 (0.2 g, 0.34 mmol) and 19 (0.059 g, 0.34 mmol) were allowed to stir under nitrogen atmosphere in 30 ml dry dichloromethane for 10 min. Then *p*-TSA (0.016 g, 0.09 mmol) were added to the above reaction mixture and allowed to stir for 2 days. Then the purification was done as above.

Figure – 1: Mass Spectrum of 6

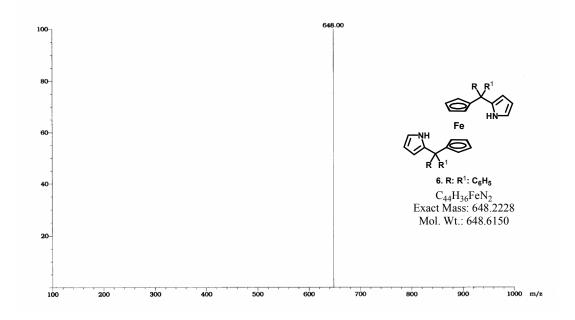


Figure – 1: Mass Spectrum of 7

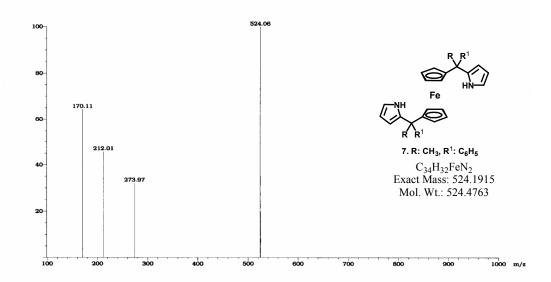


Figure – 1: Mass Spectrum of 8

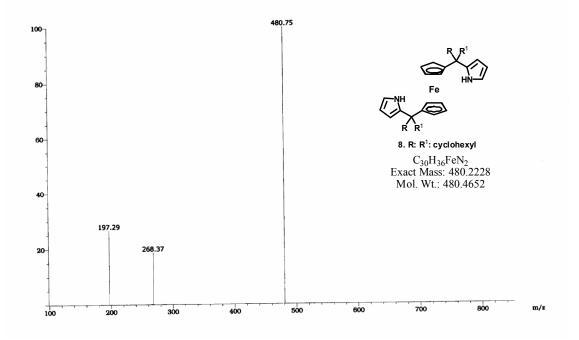
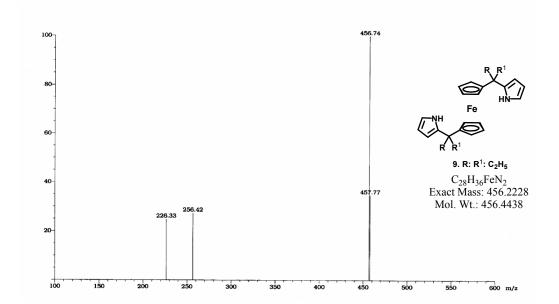


Figure – 1: Mass Spectrum of 9



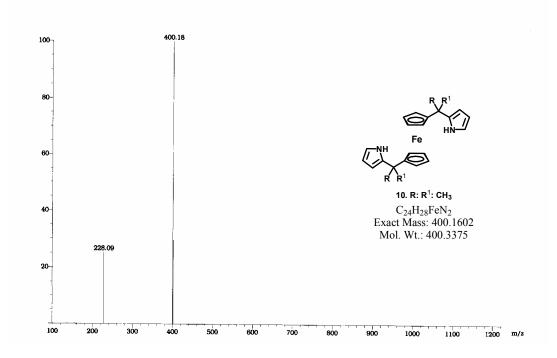
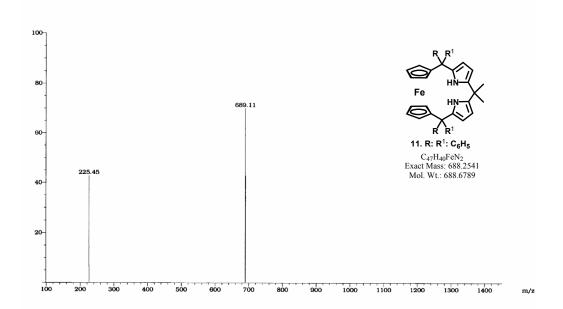


Figure – 2: Mass Spectrum of 11



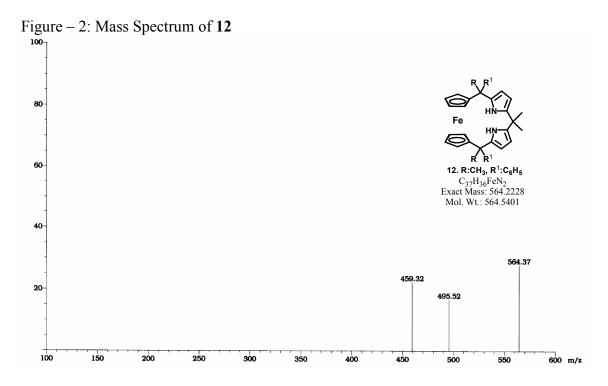


Figure – 2: Mass Spectrum of 13

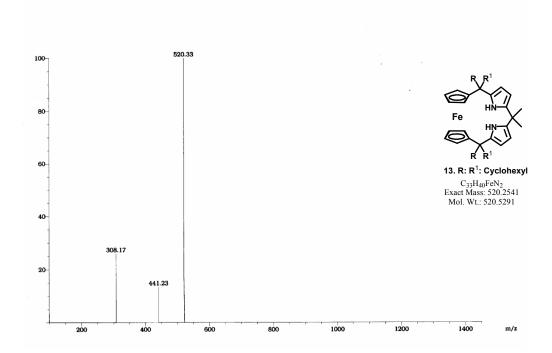


Figure – 2: Mass Spectrum of 14

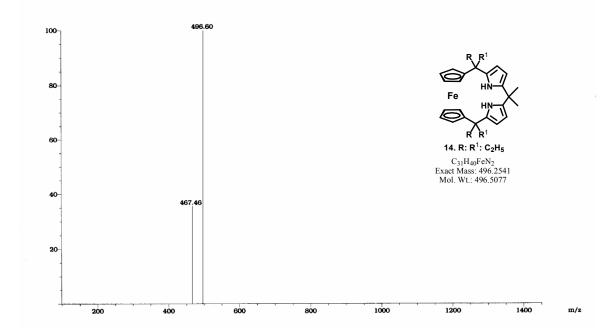


Figure – 2: Mass Spectrum of 15

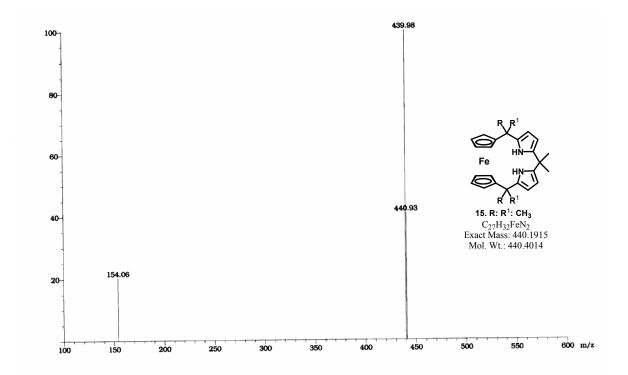


Figure – 2: Mass Spectrum of 16

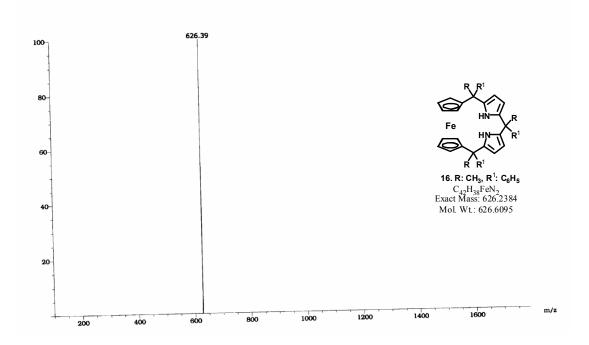


Figure – 2: Mass Spectrum of 17

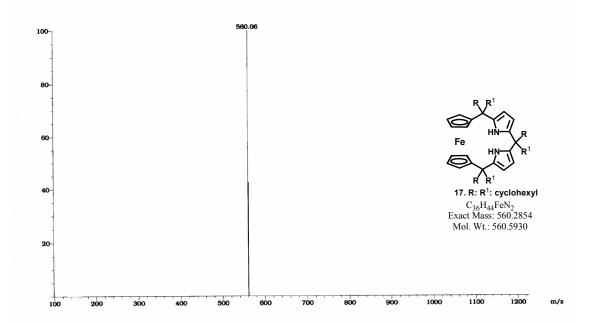


Figure – 2: Mass Spectrum of 18

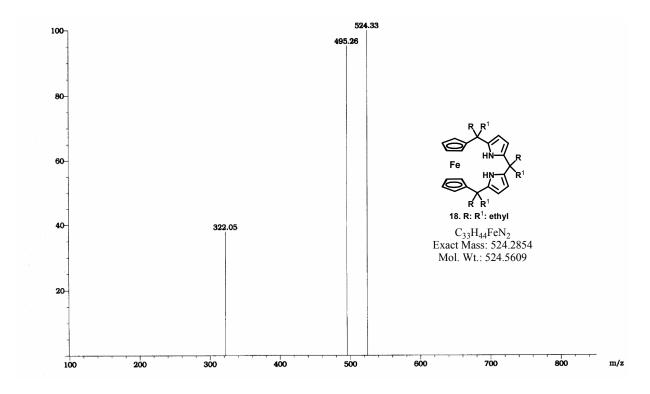


Figure – 1: Mass Spectrum of 24

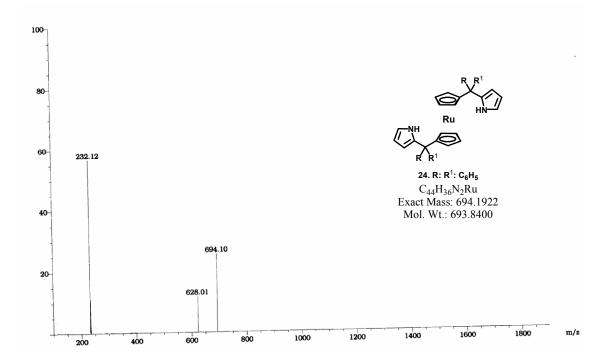


Figure – 2: Mass Spectrum of 25

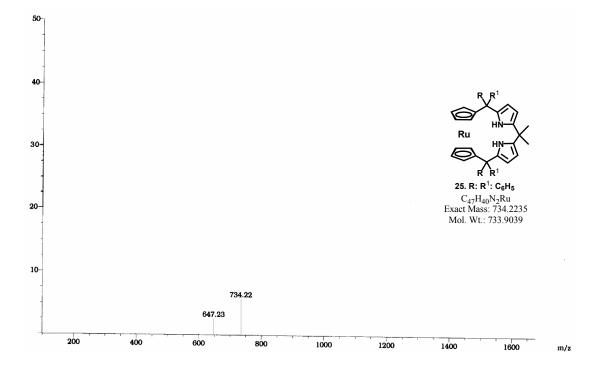


Figure – 3: Mass Spectrum of Expanded derivative of **15** (2 : 2 product):

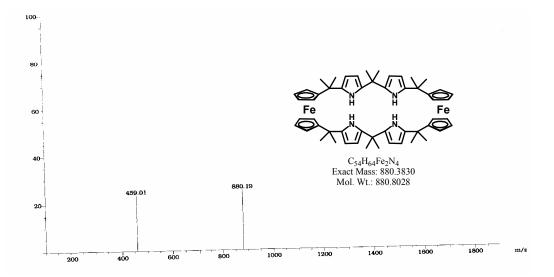


Figure – 3: Mass Spectrum of Expanded derivative of **15** (3 : 3 product):

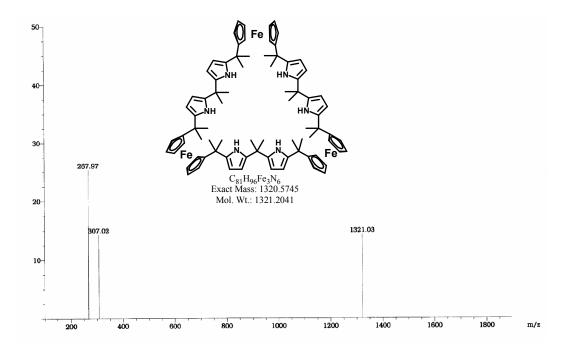
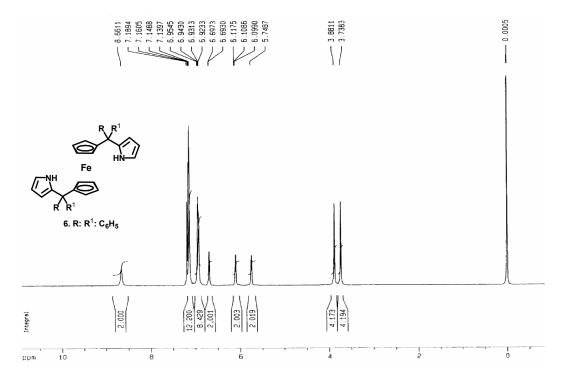


Figure – 4: ¹H NMR spectrum of 6



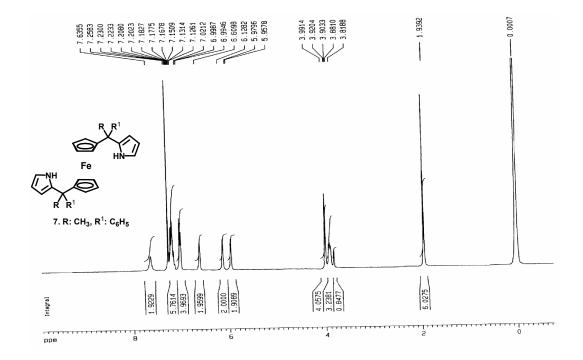


Figure – 4: ¹H NMR spectrum of $\mathbf{8}$

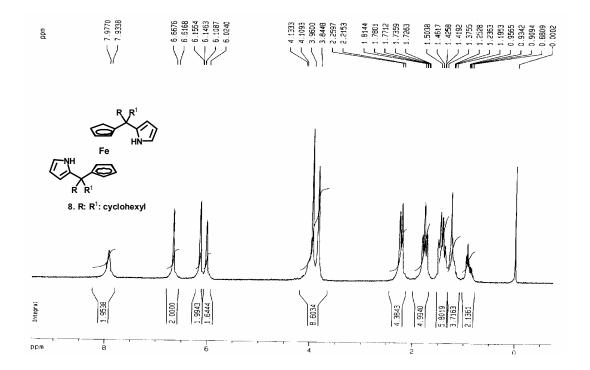


Figure – 4: ¹H NMR spectrum of 9

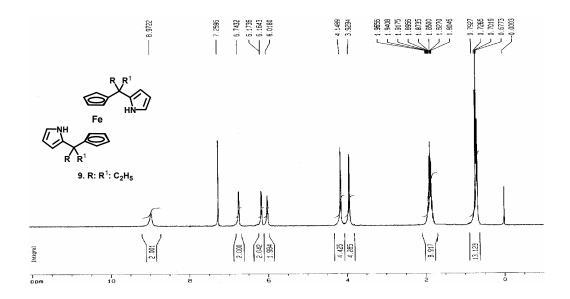
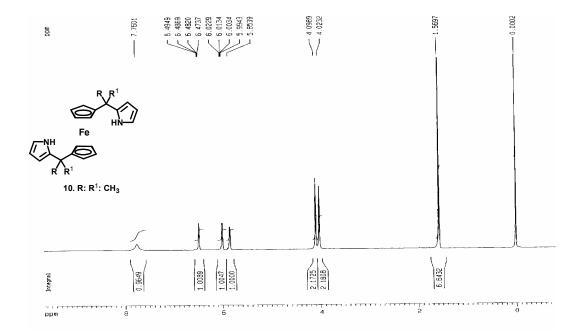


Figure – 4: ¹H NMR spectrum of 10





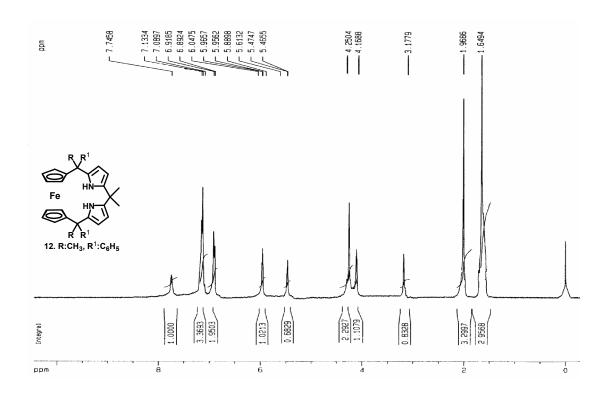


Figure – 5: ¹H NMR spectrum of **13**

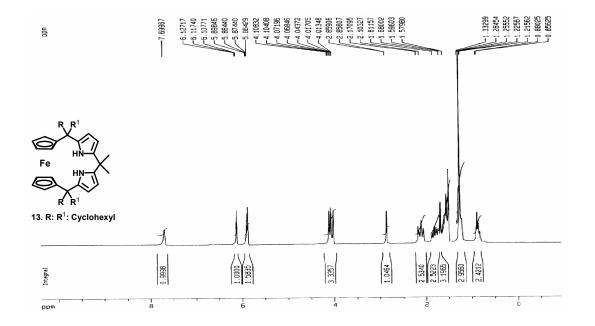


Figure – 5: ¹H NMR spectrum of **14**

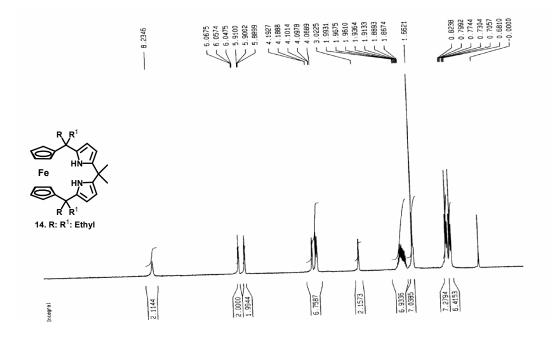
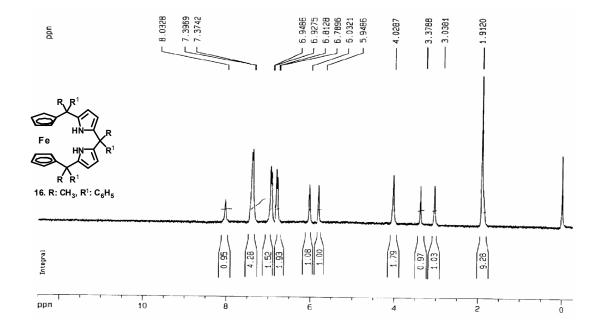


Figure – 5: ¹H NMR spectrum of 16



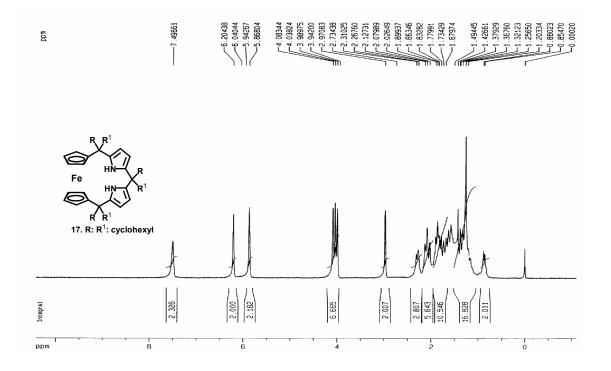
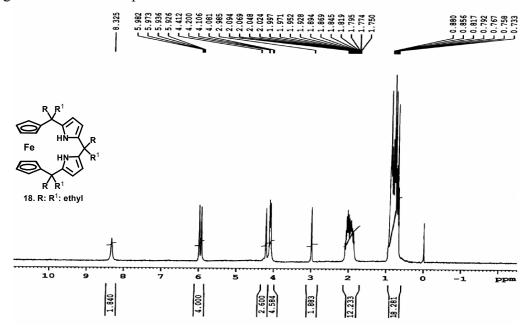


Figure – 5: ¹H NMR spectrum of **18**



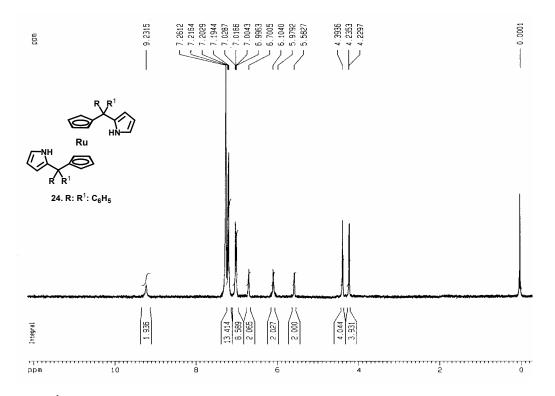


Figure – 5: ¹H NMR spectrum of **25**

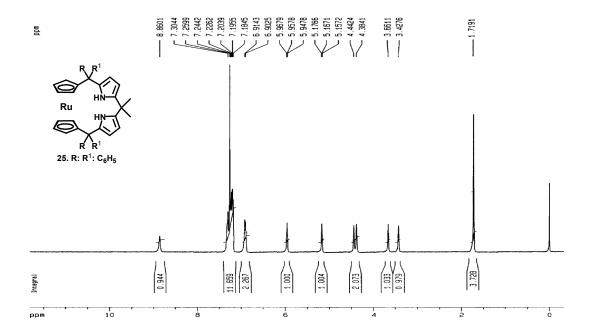


Figure – 6: 13 C Spectrum of **6**

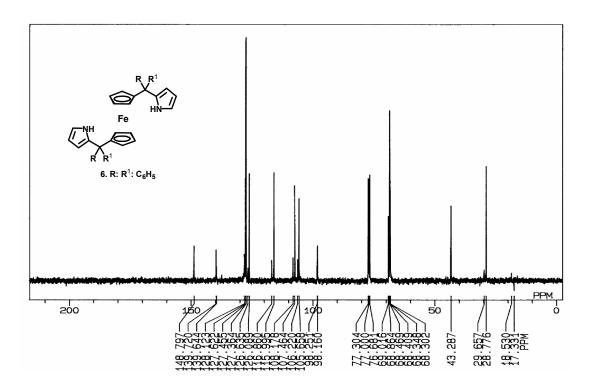


Figure – 6: 13 C Spectrum of 7

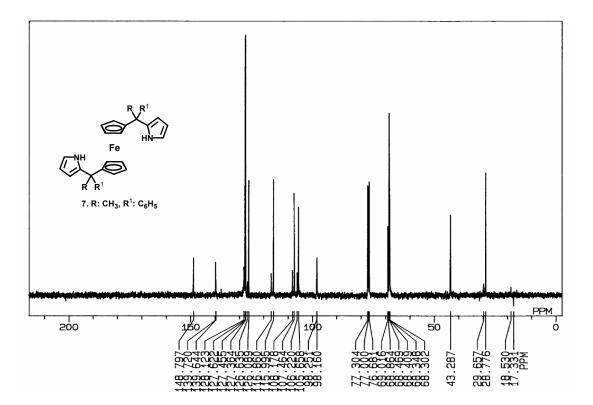


Figure – 6: 13 C Spectrum of **9**

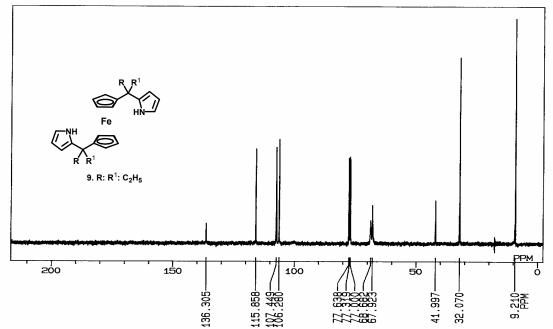


Figure – 6: 13 C Spectrum of **10**

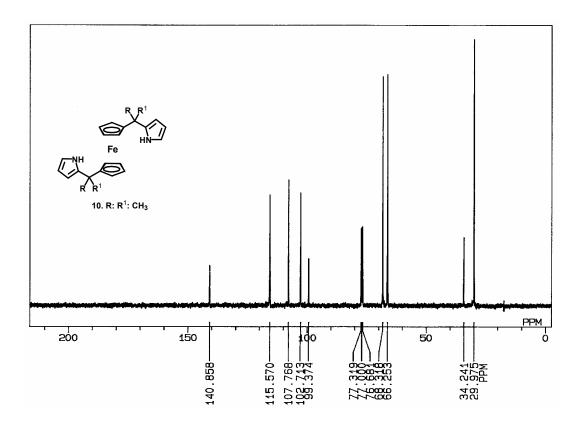


Figure – 7: ¹³C Spectrum of **14**

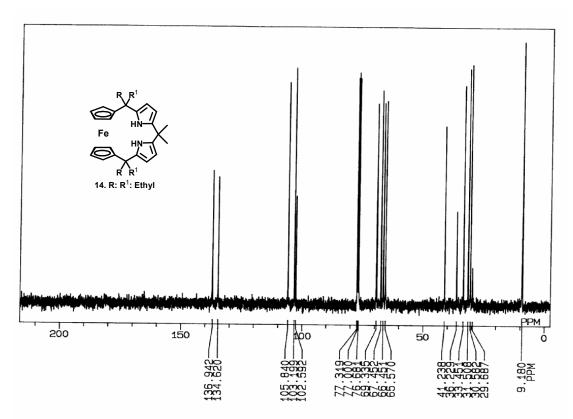


Figure – 7: ¹³C Spectrum of **17**

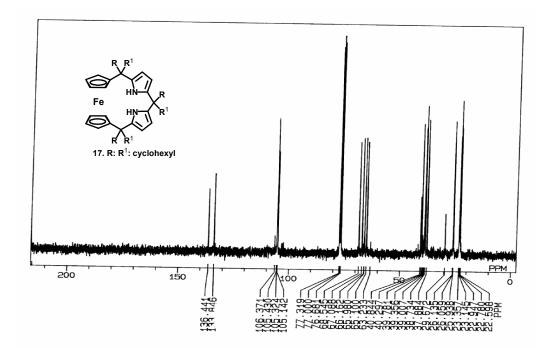


Figure – 8: ^{1}H – ^{1}H COSY spectrum of 11

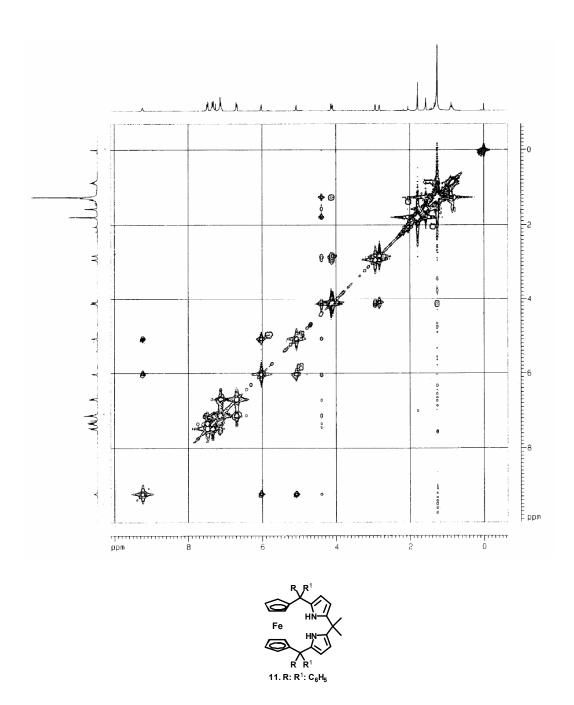


Figure -8: ${}^{1}H - {}^{1}H COSY$ spectrum of **11** shows the correlation between pyrrolic NH

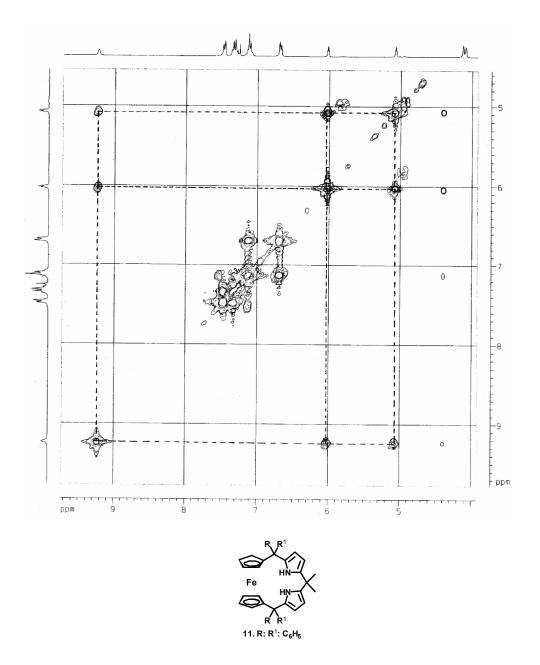


Figure – 8: ${}^{1}H$ – ${}^{1}H$ COSY spectrum of **11** shows correlation in the Ferrocenyl-CH

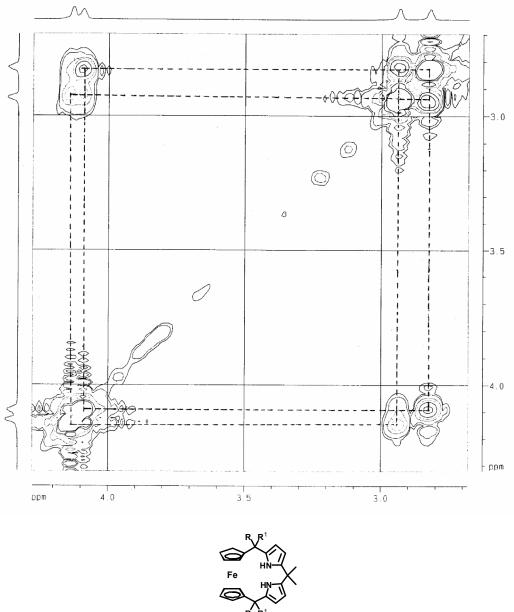
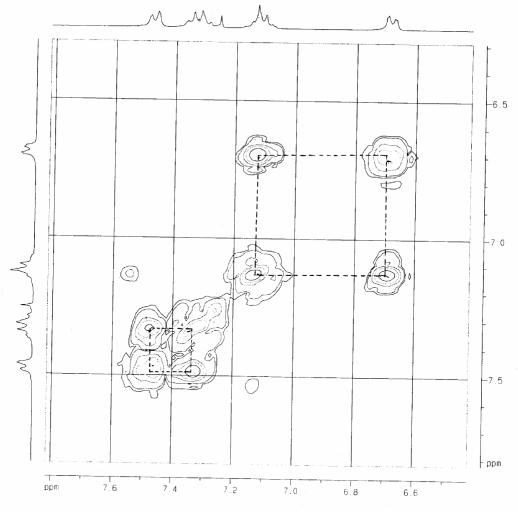
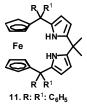




Figure – 8: ${}^{1}H$ – ${}^{1}H$ COSY spectrum of **11** shows correlation in the phenyl-CH protons





No.	Pyrrolic β-CH [*]	Ferrocenyl – CH
11 – 6	5.55 - 5.93: (-0.38)	3.28 - 3.81: (-0.53)
12 – 7	5.71-6.05: (-0.34)	3.86 - 3.91: (-0.05)
13 – 8	6.00 - 6.09: (-0.09)	3.46 - 3.9: (-0.44)
14 – 9	5.98 - 6.10: (-0.12)	3.77 - 4.04: (-0.27)
15 – 10	5.92 - 5.93: (-0.01)	3.65 - 4.06: (-0.41)
16 – 7	5.99 - 6.05: (-0.06)	3.48 - 3.91: (-0.43)
17 – 8	6.04 - 6.09: (-0.05)	3.77 - 3.90: (-0.13)
18 – 9	5.95 - 6.10: (-0.15)	3.76 - 4.04: (-0.28)
25 - 24	5.57 - 5.84: (-0.27)	3.98 - 4.32: (-0.34)

Table – 1: The shift difference observed in the pyrrolic β -CH and Ferrocenyl CH protons of macrocycle and dipyrromethane.

^{*}For calculation, pyrrolic α -CH protons of **6** – **10** and **24** are not included.

Figure – 9: Single crystal X-ray Structure of **6** (top view):

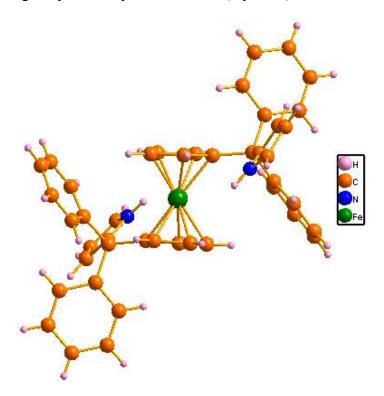


Figure – 9: Single crystal X-ray Structure of **11** (top view):

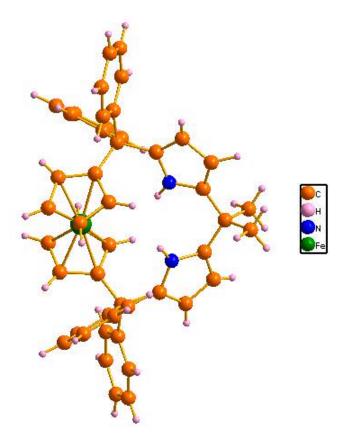
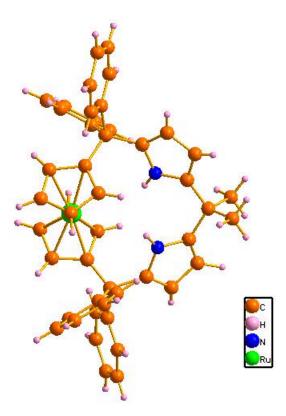


Figure – 9: Single crystal X-ray Structure of 25 (top view):



Supporting references for 6, 11 and 25:

- 1. Altormare, A., Gascarano, G., Giacovazzo, C., Guagliardi, A. (1993), SIR92. J. Appl. Cryst. 26, 343-350.
- 2. Blessing, R. (1995). Acta Cryst. A51, 33-38.
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- 4. Bruker (1999). SADABS, Bruker AXS Inc., Madison, Wisconsin, USA.
- 5. Farrugia, L. J. (1999). WinGX. J. Appl. Cryst. 32, 837-838.
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- 7. Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- 8. Spek, A. L. (1990) Acta Cryst. A46, C34.
- 9. Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.