

Thermochromism of Polydiacetylenes in the Solid State and in Solution by the Self-Organization of Polymer Chains Containing No Polar Group

Satoshi Dei, Akinori Matsumoto, and Akikazu Matsumoto*

Department of Applied Chemistry, Graduate School of Engineering, Osaka City University,
Sugimoto, Sumiyoshi-ku, Osaka 558-8585, Japan

E-mail: matsumoto@a-chem.eng.osaka-cu.ac.jp

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Spectral Data of Monomers

4-Methoxybenzyl 10,12-pentacosadiynoate (**DA-1**): mp 41–42 °C; ^1H NMR (400 MHz, CD_3OD) δ 0.88 (t, $J = 7.2$ Hz, CH_3 , 3H), 1.26–1.62 (m, CH_2 , 32H), 2.24 (t, $J = 7.2$ Hz, $\text{CH}_2\text{--C}\equiv\text{C--C}\equiv\text{C--CH}_2$, 4H), 2.32 (t, $J = 7.2$ Hz, CH_2CO_2 , 2H), 3.81 (s, $\text{C}_6\text{H}_4\text{OCH}_3$, 3H), 5.04 (s, $\text{CH}_2\text{C}_6\text{H}_4$, 2H), 6.88–7.31 (m, C_6H_4 , 4H); ^{13}C NMR (100 MHz, CDCl_3) δ 14.14 (CH_3), 19.18, 19.21, 22.69, 24.91, 28.30, 28.35, 28.74, 28.87, 28.88, 29.04, 29.06, 29.10, 29.35, 29.48, 29.62, 29.63, 31.93, and 34.33 (CH_2), 55.27 (OCH_3), 65.20 and 65.29 ($\text{C}\equiv\text{C--C}\equiv\text{C}$), 65.89 (CO_2CH_2), 77.45 and 77.60 ($\text{C}\equiv\text{C--C}\equiv\text{C}$), 113.90, 128.25, 130.05, and 159.56 (phenyl), 173.73 (C=O); IR (KBr) 1721 cm^{-1} ($\nu_{\text{C=O}}$).

4-Methoxybenzyl 10,12-triacosadiynoate (**DA-2**): mp 33–34 °C; ^1H NMR (400 MHz, CD_3OD) δ 0.88 (t, $J = 7.2$ Hz, CH_3 , 3H), 1.26–1.63 (m, CH_2 , 28H), 2.24 (t, $J = 7.2$ Hz, $\text{CH}_2\text{--C}\equiv\text{C--C}\equiv\text{C--CH}_2$, 4H), 2.32 (t, $J = 7.2$ Hz, CH_2CO_2 , 2H), 3.81 (s, $\text{C}_6\text{H}_4\text{OCH}_3$, 3H), 5.05 (s, $\text{CH}_2\text{C}_6\text{H}_4$, 2H), 6.87–7.31 (m, C_6H_4 , 4H). ^{13}C NMR (100 MHz, CD_3OD) δ 14.13 (CH_3), 19.18, 19.21, 22.69, 24.91, 28.29, 28.35, 28.74, 28.86, 28.88, 29.04, 29.06, 29.10, 29.31, 29.48, 29.57, 31.90, and 34.33 (CH_2), 55.28 (OCH_3), 65.20 and 65.29 ($\text{C}\equiv\text{C--C}\equiv\text{C}$), 65.89 (CO_2CH_2), 77.45 and 77.61 ($\text{C}\equiv\text{C--C}\equiv\text{C}$), 113.90, 128.24, 130.05, and 159.56 (phenyl), 173.75 (C=O).

4-Methoxybenzyl 10,12-nonacosadiynoate (**DA-3**): mp 53–54 °C; ^1H NMR (400 MHz, CD_3OD) δ 0.88 (t, $J = 7.2$ Hz, CH_3 , 3H), 1.25–1.55 (m, CH_2 , 40H), 2.24 (t, $J = 7.2$ Hz, $\text{CH}_2\text{--C}\equiv\text{C--C}\equiv\text{C--CH}_2$, 4H), 2.32 (t, $J = 7.2$ Hz, CH_2CO_2 , 2H), 3.81 (s, $\text{C}_6\text{H}_4\text{OCH}_3$, 3H), 5.05 (s, $\text{CH}_2\text{C}_6\text{H}_4$, 2H), 6.88–7.31 (m, C_6H_4 , 4H).

Benzyl 10,12-pentacosadiynoate (**DA-4**): Yield 82%, mp 36 °C; ^1H NMR (400 MHz, CDCl_3) δ 0.88 (t, $J = 7.2$ Hz, CH_3 , 3H), 1.26–1.65 (m, CH_2 , 32H), 2.24 (t, $J = 7.2$ Hz, $\text{CH}_2\text{--C}\equiv\text{C--C}\equiv\text{C--CH}_2$, 4H), 2.35 (t, $J = 7.2$ Hz, CH_2CO_2 , 2H), 5.12 (s, $\text{CH}_2\text{C}_6\text{H}_5$, 2H), 7.31–7.39 (m, C_6H_5 , 5H); ^{13}C NMR (100 MHz, CDCl_3) δ 14.15 (CH_3), 19.19, 19.21, 22.70, 24.91, 28.29, 28.35, 28.74, 28.87, 28.89, 29.05, 29.11, 29.36, 29.49, 29.62, 29.64, 29.66, 31.93, and 34.30 (CH_2), 65.19 and 65.27 ($\text{C}\equiv\text{C--C}\equiv\text{C}$), 66.08 (CO_2CH_2), 77.45 and 77.63 ($\text{C}\equiv\text{C--C}\equiv\text{C}$) 128.19, 128.54, and 136.11 (phenyl), 173.67 (C=O); IR (KBr) 1726 cm^{-1} ($\nu_{\text{C=O}}$).

3-Methoxybenzyl 10,12-pentacosadiynoate (**DA-5**): mp 35–36 °C; ^1H NMR (400 MHz, CDCl_3) δ 0.88 (t, $J = 7.2$ Hz, CH_3 , 3H), 1.26–1.66 (m, CH_2 , 32H), 2.24 (t, $J = 7.2$ Hz, $\text{CH}_2\text{--C}\equiv\text{C--C}\equiv\text{C--CH}_2$, 4H), 2.36 (t, $J = 7.2$ Hz, CH_2CO_2 , 2H), 3.82 (s, $\text{C}_6\text{H}_4\text{OCH}_3$, 3H), 5.09 (s, $\text{CH}_2\text{C}_6\text{H}_4$, 2H), 6.85–7.30 (m, C_6H_4 , 4H); ^{13}C NMR (100 MHz, CDCl_3) δ 14.15 (CH_3), 19.19, 19.21, 22.71, 24.92, 28.30, 28.35, 28.75, 28.87, 28.90, 29.07, 29.11, 29.37, 29.49, 29.62, 29.65, 29.67, 31.93, and 34.30 (CH_2), 55.24 (OCH_3), 65.20 and 65.29 ($\text{C}\equiv\text{C--C}\equiv\text{C}$), 65.93 (CO_2CH_2) 77.45 and 77.62 ($\text{C}\equiv\text{C--C}\equiv\text{C}$), 113.57, 113.64, 120.33, 129.62, 137.64, and 159.71 (phenyl), 173.63 (C=O); IR (KBr) 1727 cm^{-1} ($\nu_{\text{C=O}}$).

4-Chlorobenzyl 10,12-pentacosadiynoate (**DA-6**): Yield 86%, mp 34–35 °C; ^1H NMR (400 MHz, CDCl_3) δ 0.88 (t, $J = 7.2$ Hz, CH_3 , 3H), 1.26–1.64 (m, CH_2 , 32H), 2.24 (t, $J = 7.2$ Hz, $\text{CH}_2\text{--C}\equiv\text{C--C}\equiv\text{C--CH}_2$, 4H), 2.34 (t, $J = 7.2$ Hz, CH_2CO_2 , 2H), 5.07 (s, $\text{CH}_2\text{C}_6\text{H}_4$, 2H), 7.28–7.35 (m, C_6H_4 , 4H); IR (KBr) 1725 cm^{-1} ($\nu_{\text{C=O}}$).

4-Bromobenzyl 10,12-pentacosadiynoate (**DA-7**): Yield 85%, mp 40–41 °C; ^1H NMR (400 MHz, CDCl_3) δ 0.88 (t, $J = 7.2$ Hz, CH_3 , 3H), 1.26–1.64 (m, CH_2 , 32H), 2.24 (t, $J = 7.2$ Hz, $\text{CH}_2\text{--C}\equiv\text{C--C}\equiv\text{C--CH}_2$, 4H), 2.34 (t, $J = 7.2$ Hz, CH_2CO_2 , 2H), 5.06 (s, $\text{CH}_2\text{C}_6\text{H}_4$, 2H), 7.22–7.50 (m, C_6H_4 , 4H); IR (KBr) 1725 cm^{-1} ($\nu_{\text{C=O}}$).

4-Carboxybenzyl 10,12-pentacosadiynoate (**DA-8**): ^1H NMR (400 MHz, CDCl_3) δ 0.88 (t, $J = 7.2$ Hz, CH_3 , 3H), 1.25–1.50 (m, CH_2 , 32H), 2.23 (t, $J = 7.2$ Hz, $\text{CH}_2\text{--C}\equiv\text{C--C}\equiv\text{C--CH}_2$, 4H), 2.37 (t, $J = 7.2$ Hz, CH_2CO_2 , 2H), 5.42 (s, $\text{CH}_2\text{C}_6\text{H}_4$, 2H), 7.50–8.10 (m, C_6H_4 , 4H)

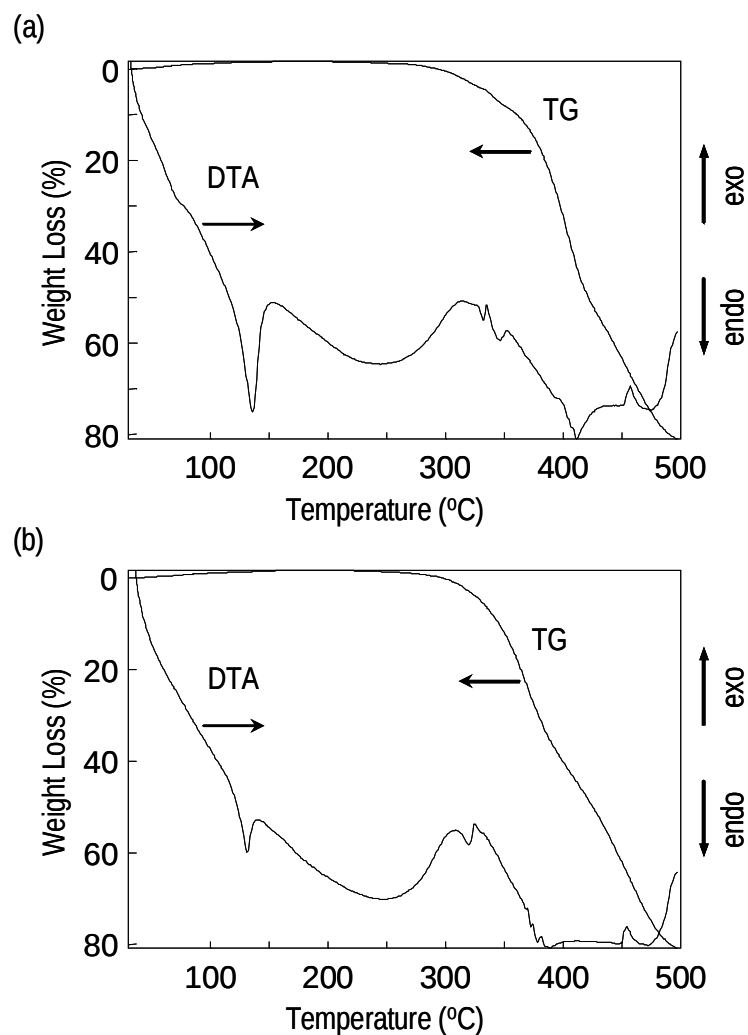


Figure S1. TG/DTA curves of (a) crystalline and (b) partly crystalline **PDA-1s**, which were obtained by solid-state polymerization and the subsequent dissolution and reprecipitation processes, respectively. Measured in a nitrogen stream at the heating rate of 10 °C/min.

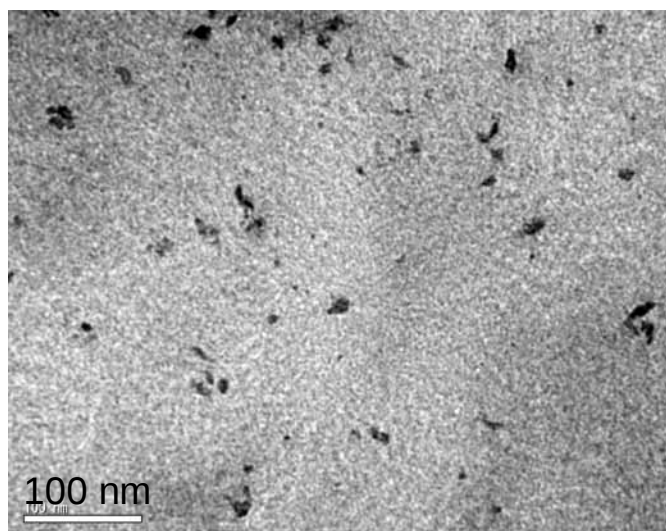


Figure S2. TEM photograph of **PDA-1** embedded poly(vinyl alcohol) film.

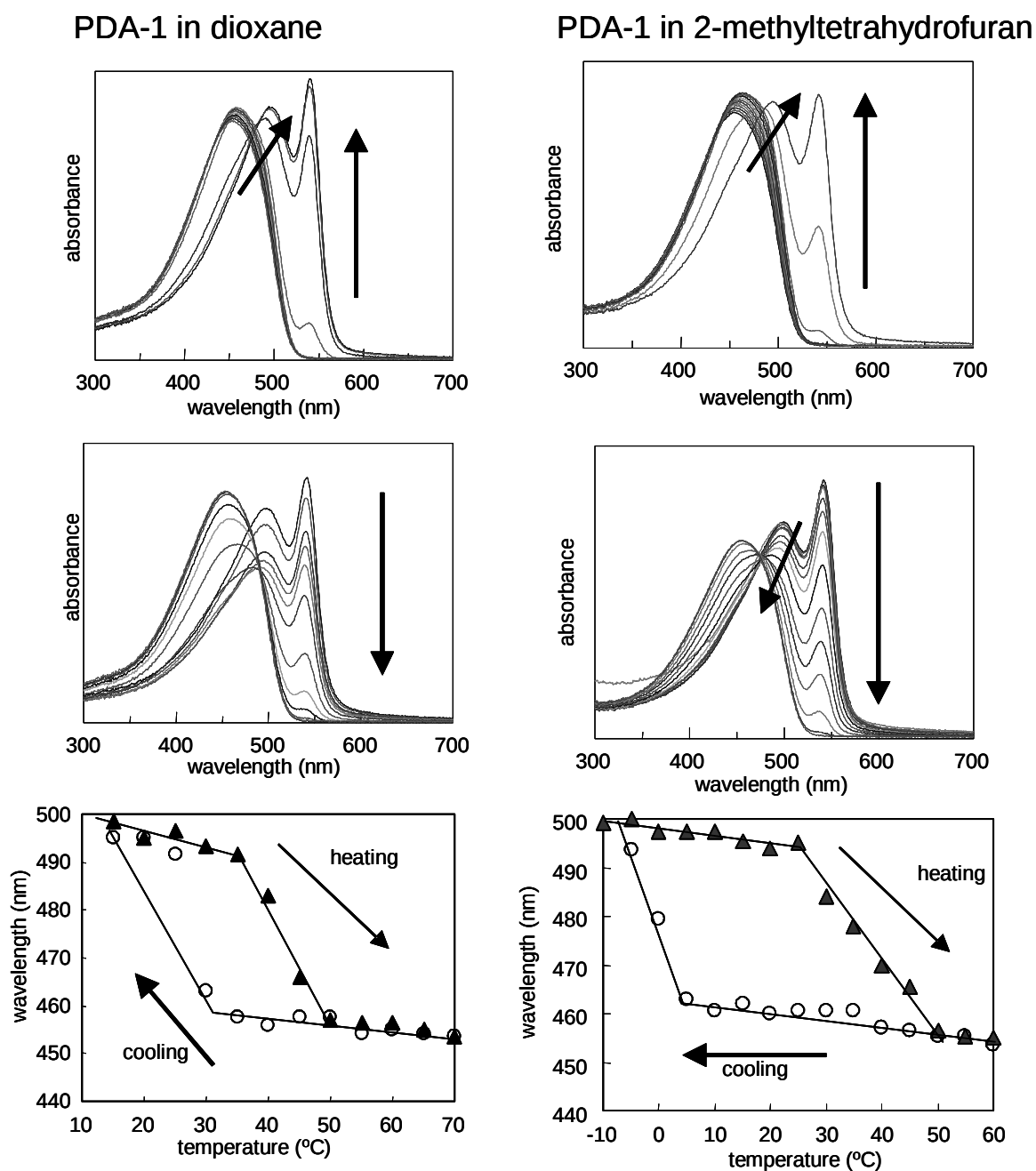


Figure S3. Change in the absorption spectra of **PDA-1** in 1,4-dioxane and 2-methyltetrahydrofuran during cooling and heating processes with each 5 °C step in the temperature range of 10 to 70 °C and –10 to 60 °C, respectively.

PDA-1 in toluene

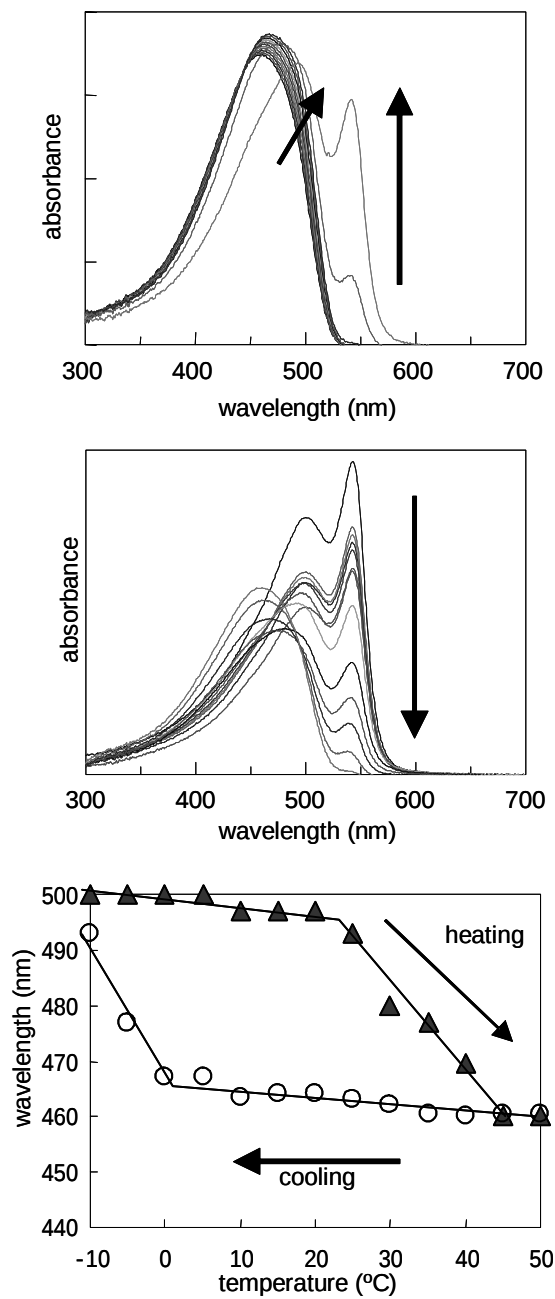


Figure S4. Change in the absorption spectrum of **PDA-1** in toluene during cooling and heating processes with each 5 °C step in the temperature range of –10 to 50 °C.

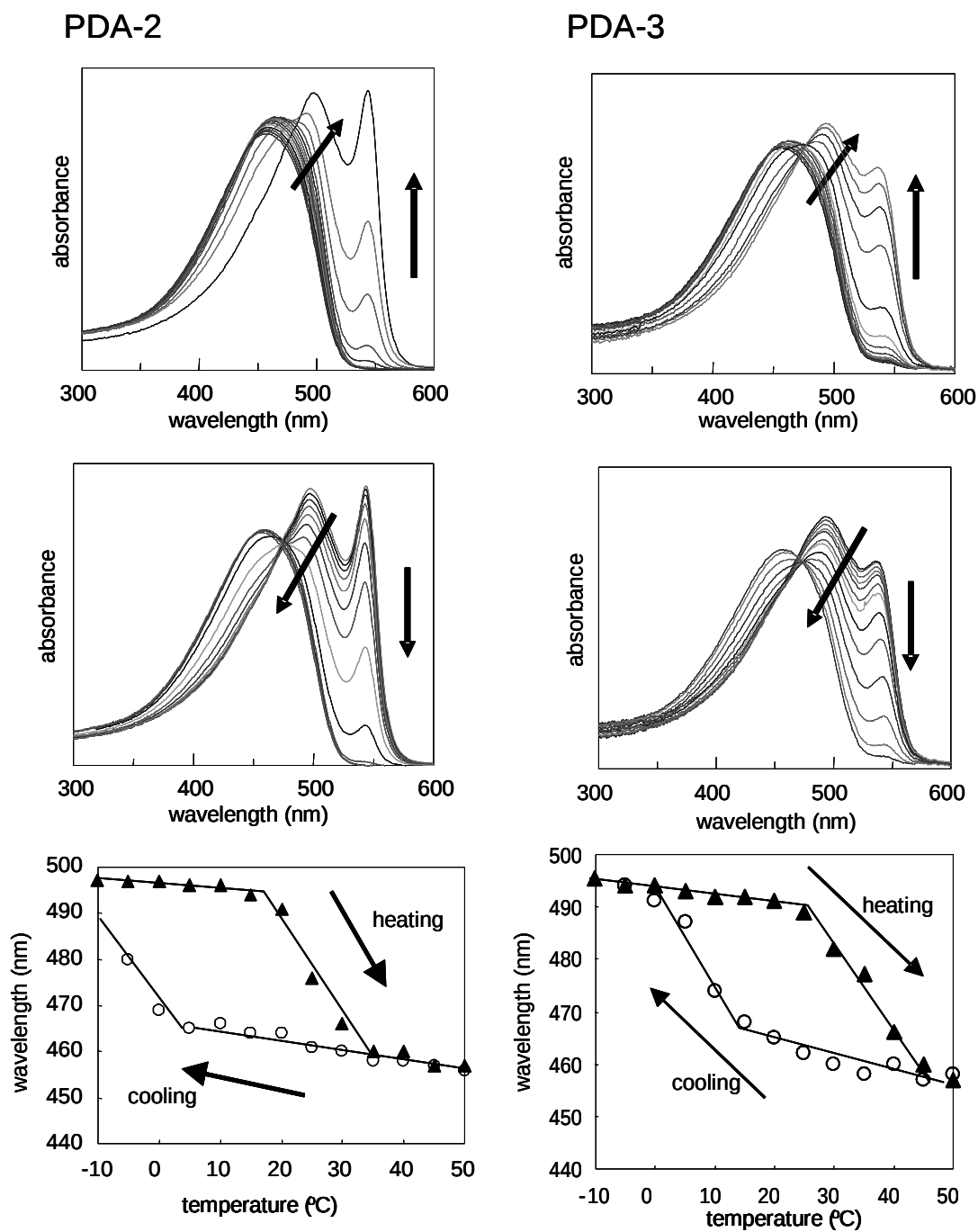
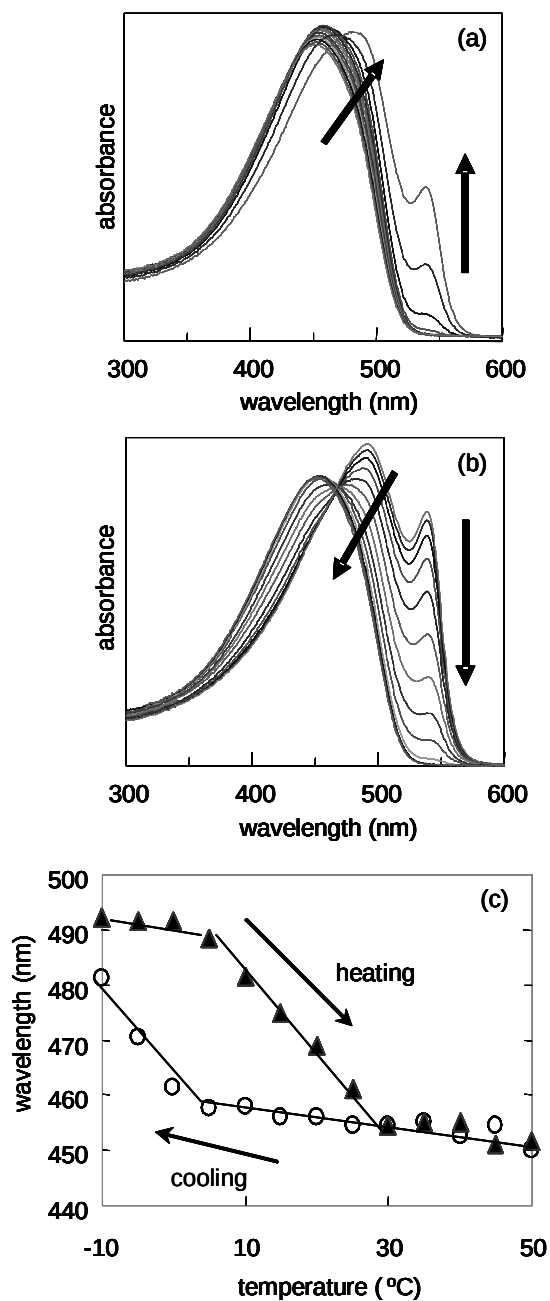


Figure S5. Change in the absorption spectra of **PDA-2** and **PDA-3** in THF during cooling and heating processes with each 5 °C step in the temperature range of –10 to 50 °C.

PDA-4



PDA-5

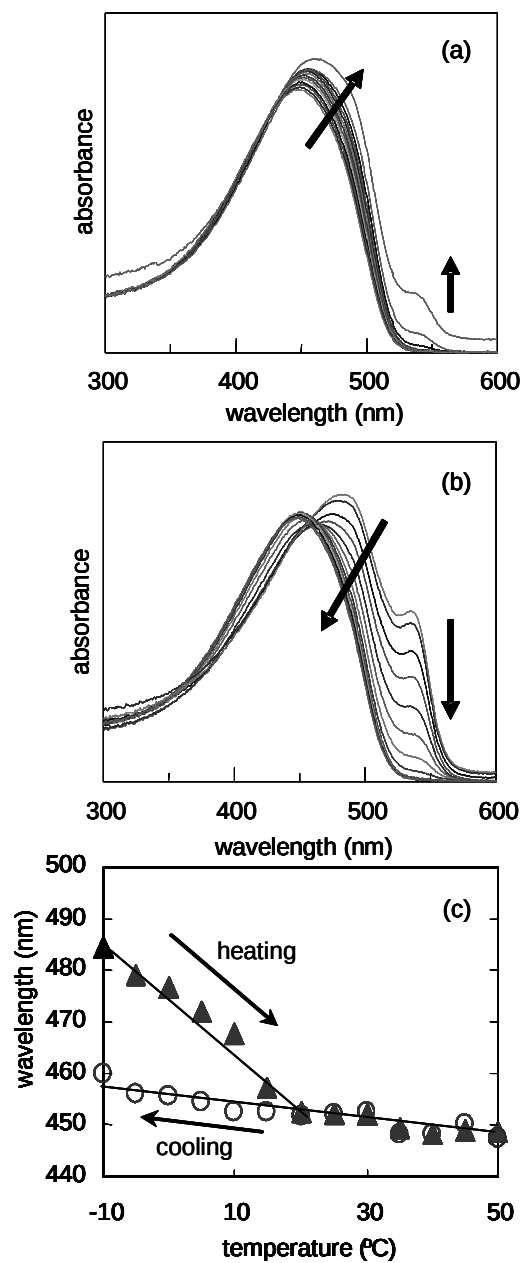


Figure S6. Change in the absorption spectra of **PDA-4** and **PDA-5** in THF during cooling and heating processes with each 5 °C step in the temperature range of -10 to 50 °C.

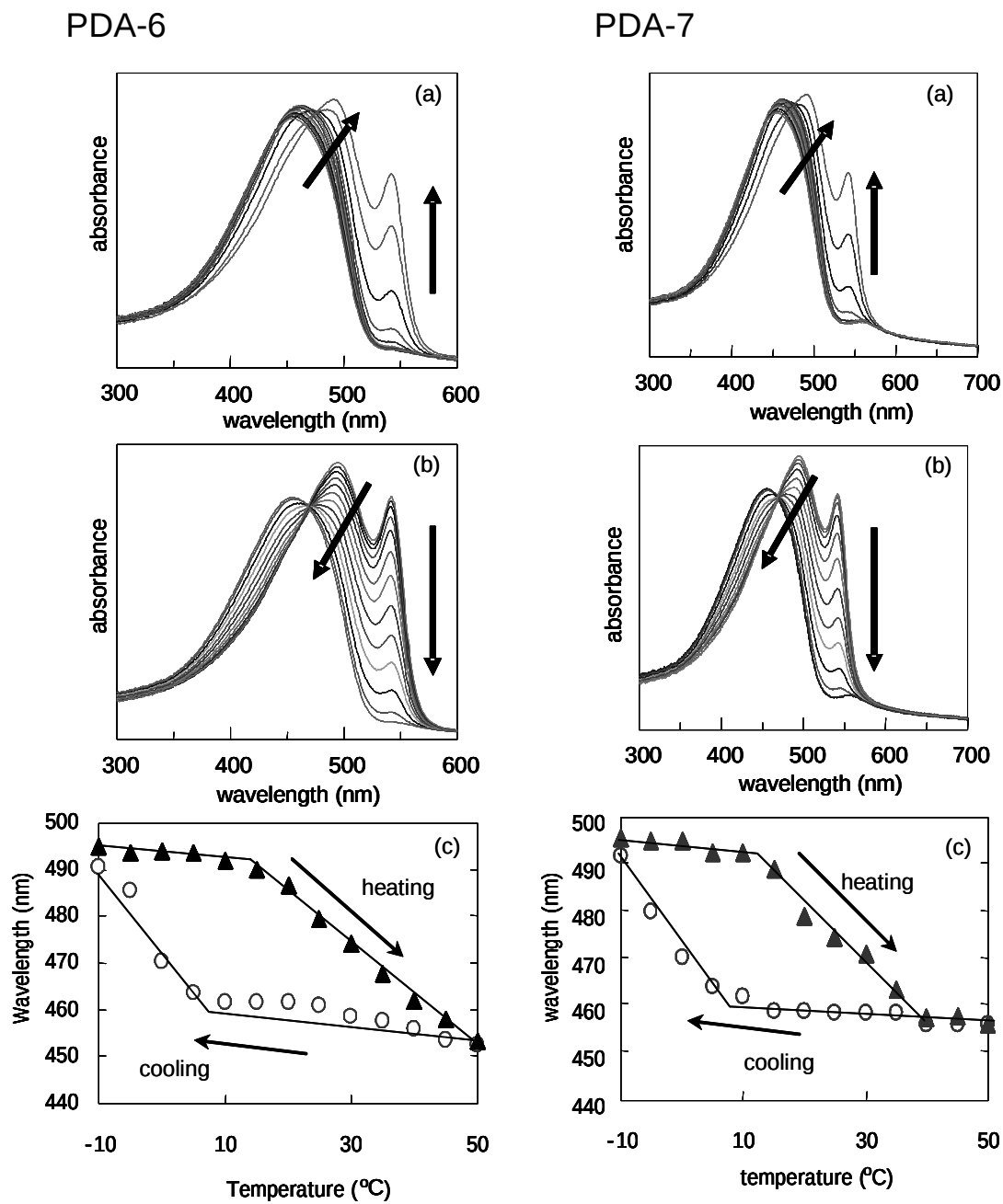


Figure S7. Change in the absorption spectra of **PDA-6** and **PDA-7** in THF during cooling and heating processes with each 5 °C step in the temperature range of –10 to 50 °C.

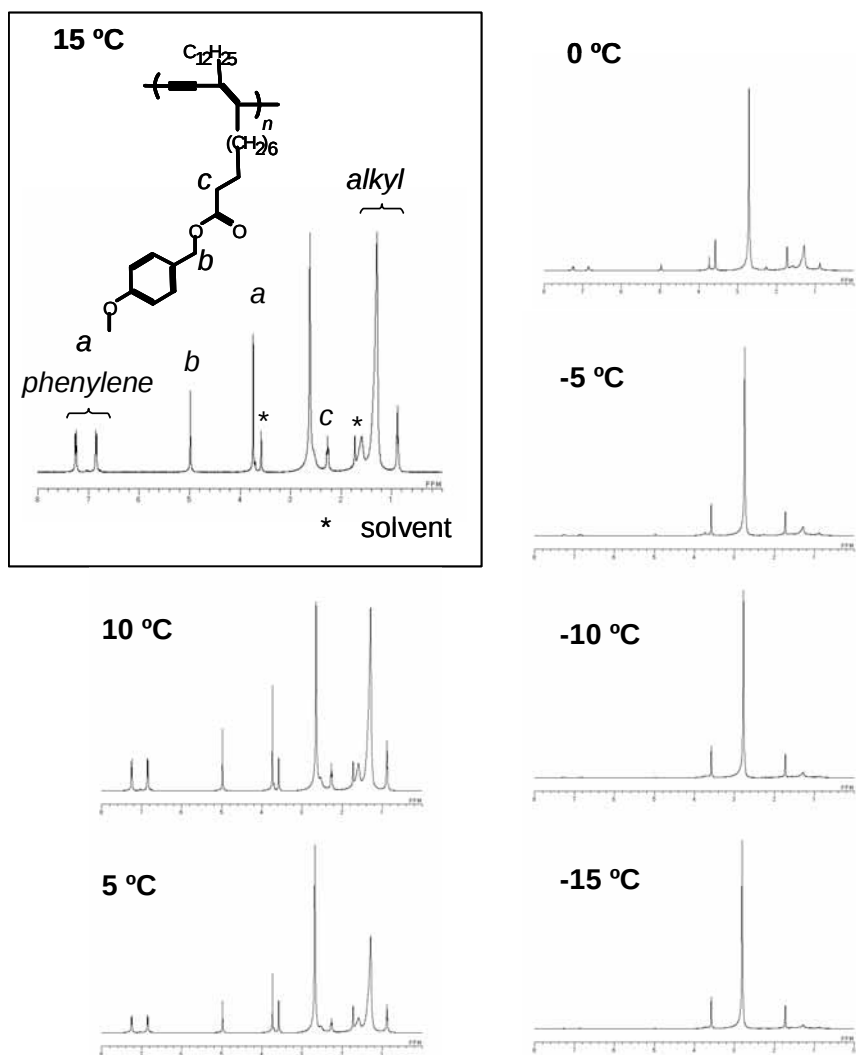


Figure S8. ^1H NMR spectrum of **PDA-1** in $\text{THF-}d_8$ at various temperatures.