Supporting Information for

Photoaddition Reactions of 1,2-Diketones with Silyl Ketene Acetals. Formation of β -Hydroxy- γ -ketoesters.

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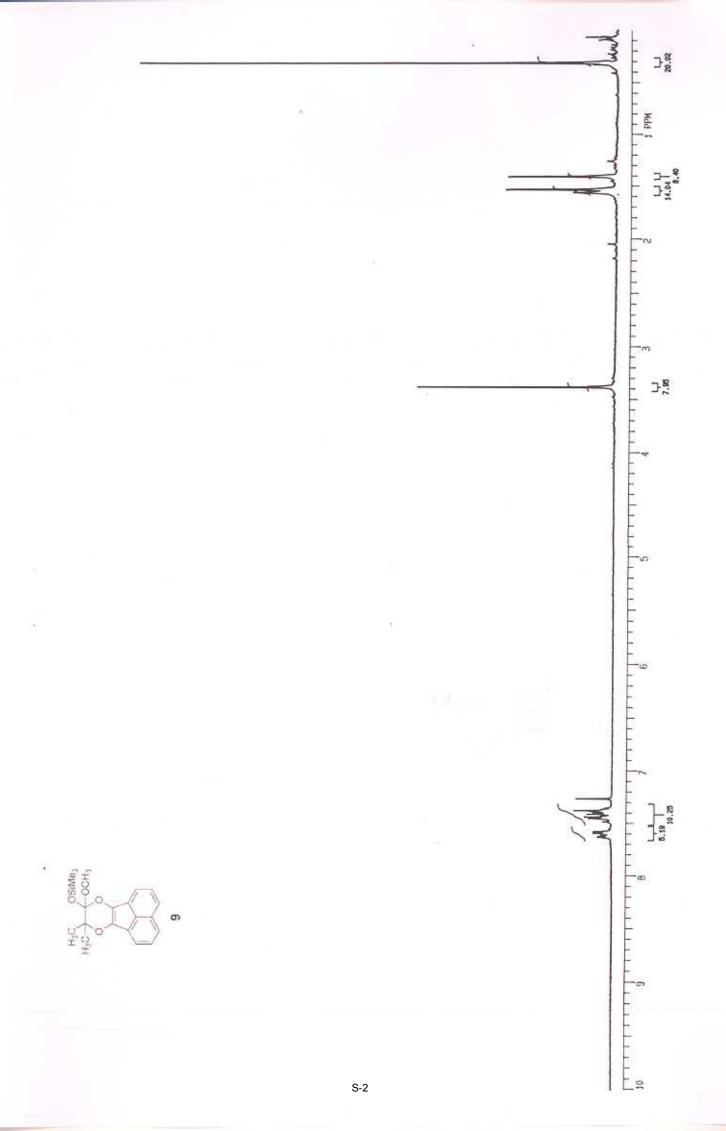
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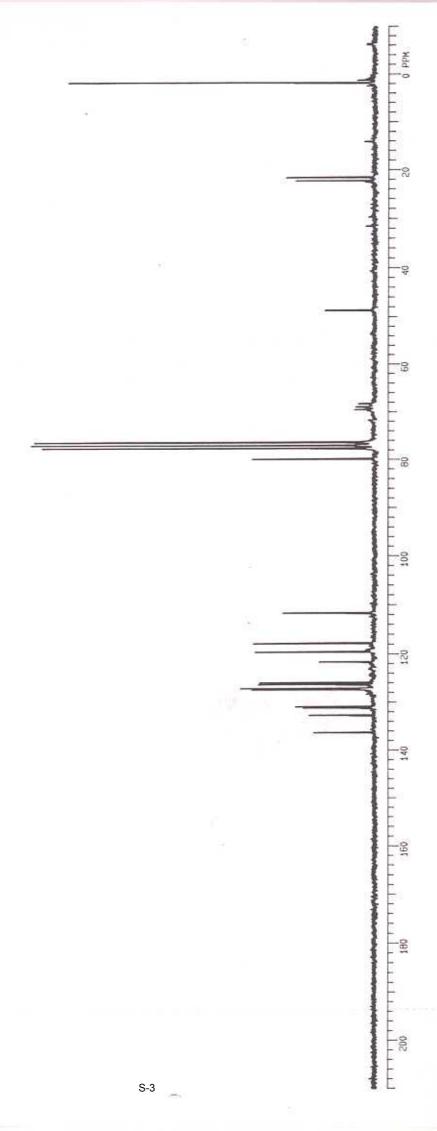
Content page

General procedure S-1NMR Spectra of Compounds $S-2 \sim S-77$

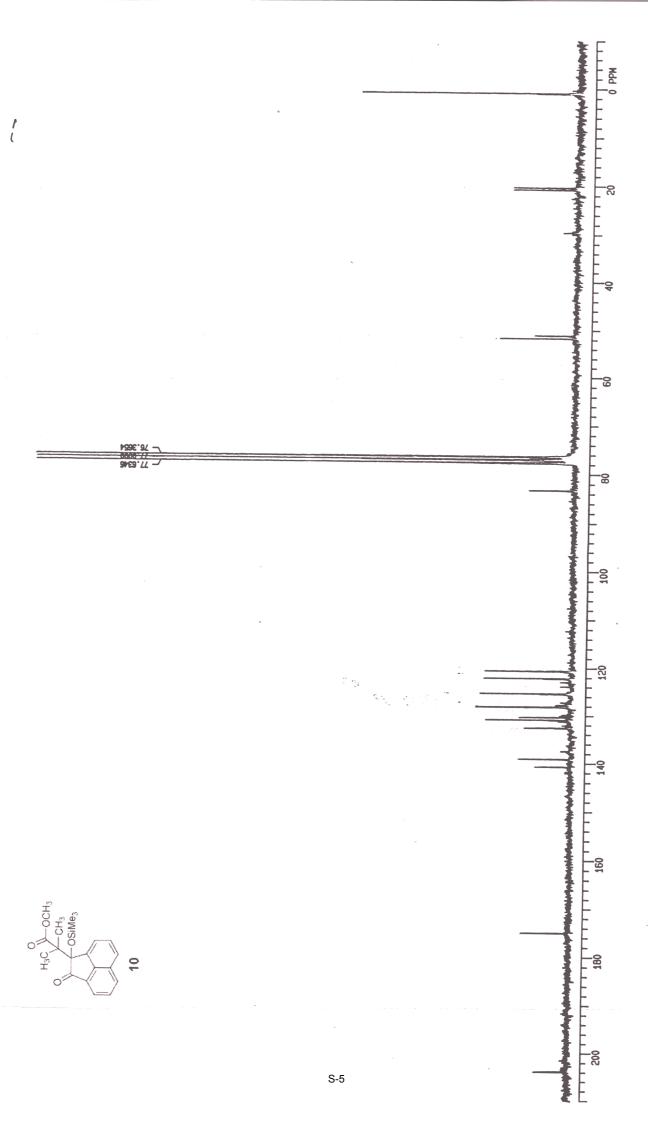
General Procedure

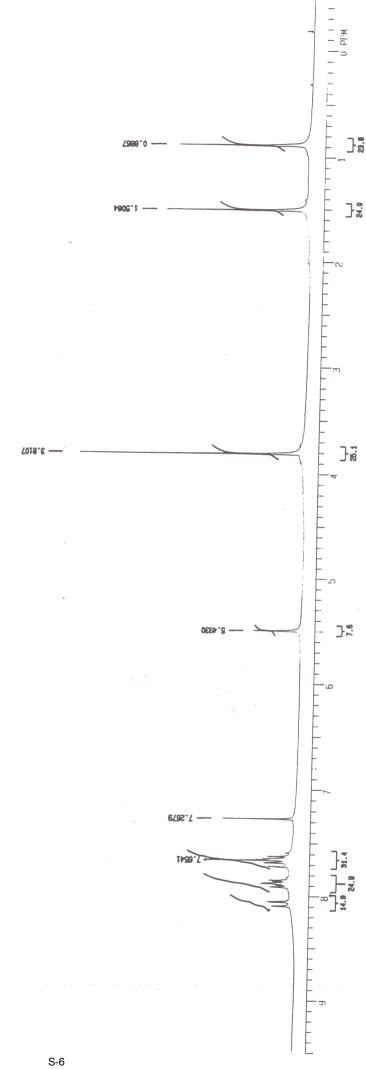
The Chemical shifts of resonances in the ¹H- and ¹³C-NMR (300 *MHz*) spectra were recorded on CDCl₃ solutions and are reported in parts per million relative to CHCl₃ peak (7.24 ppm) as an internal standard. IR spectral bands are reported in cm⁻¹. Preparative photochemical reactions were conducted with an apparatus consisting of a 450 W Hanovia medium vapor pressure mercury lamp surrounded by a Uranium glass filter in a water-cooled quartz immersion well surrounded by the solution being irradiated. The photolysis solutions were purged with nitrogen before and during irradiations. The photolysates were concentrated under reduced pressure giving residues which were subjected to silica gel column chromatography. High resolution (HRMS) mass spectra were obtained by use of electron impact ionization unless otherwise noted. All starting materials used in the photoreactions derived from commercial sources. All new compounds described are isolated as oils in >90% purity (by NMR analysis) unless noted otherwise.

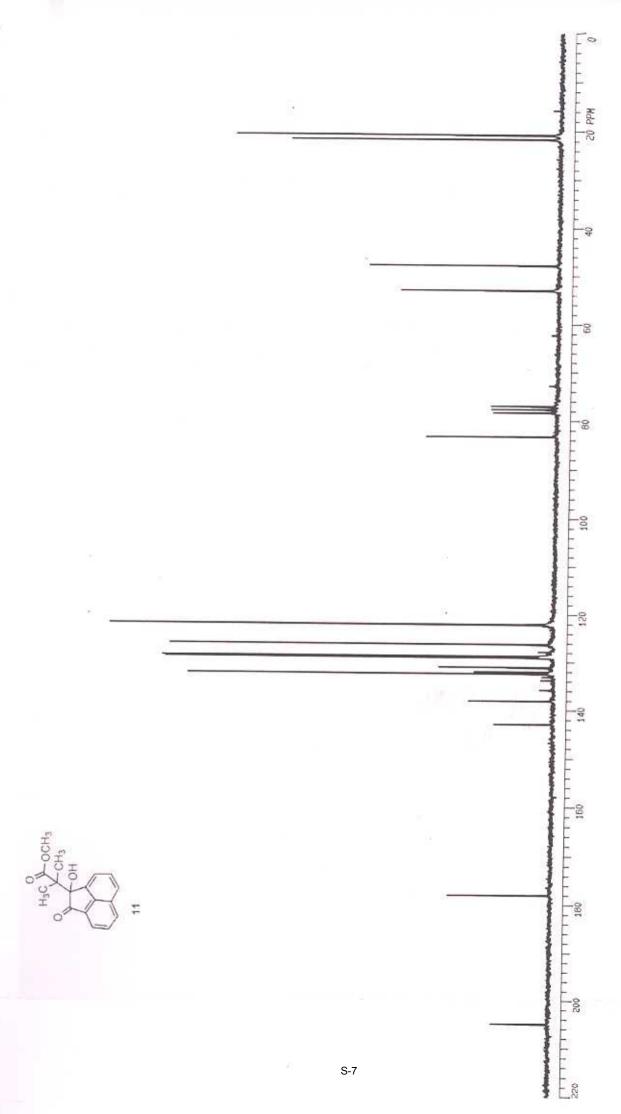


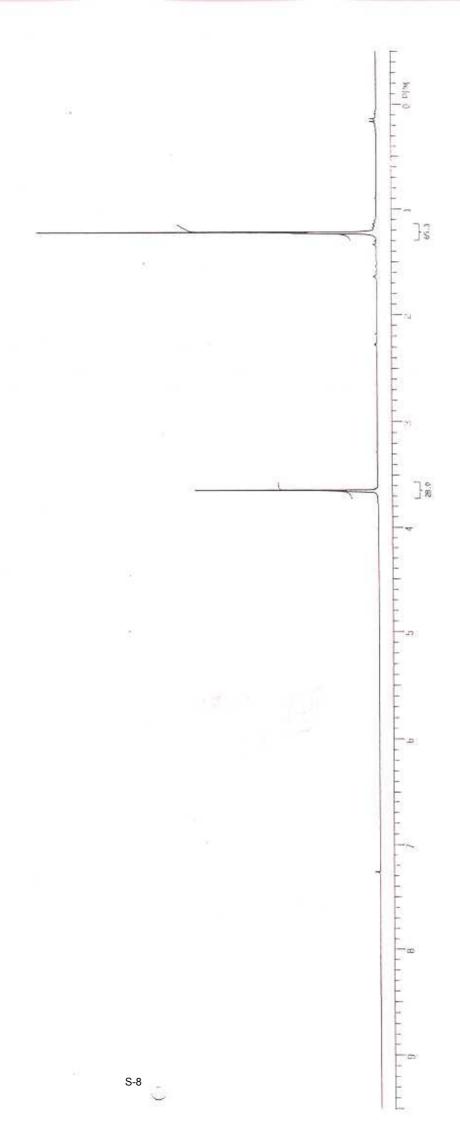


H₃C OCH₃ OSIMe₃

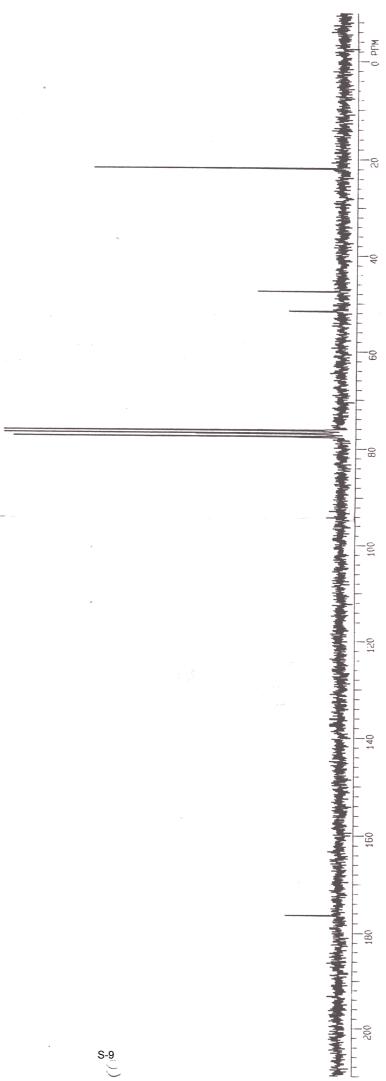


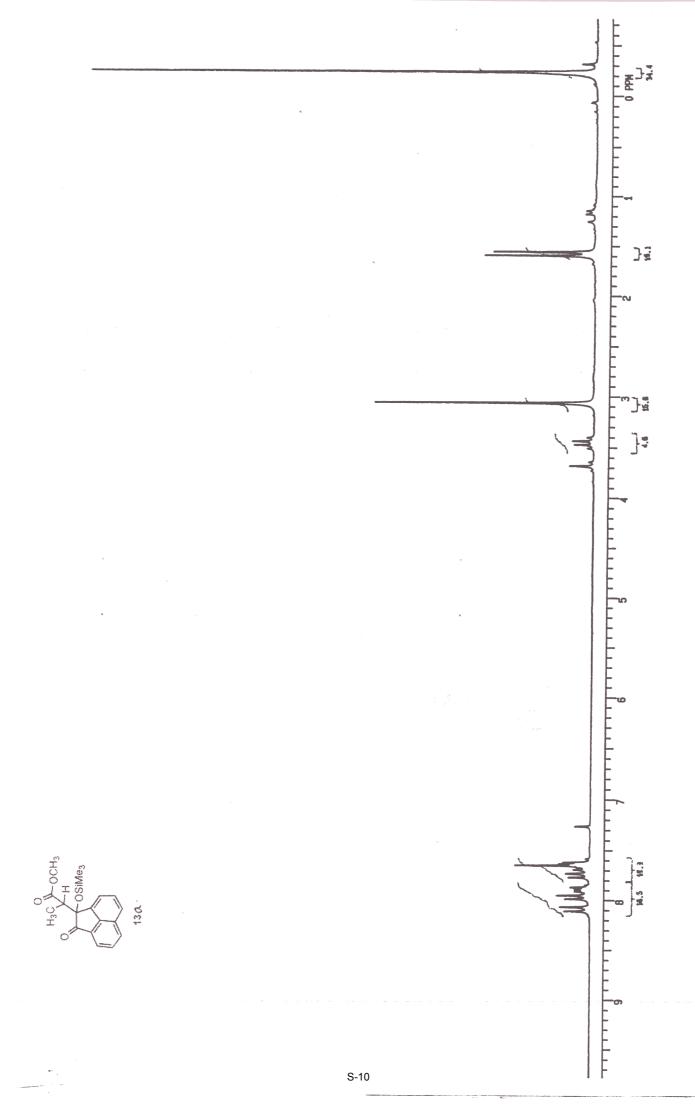


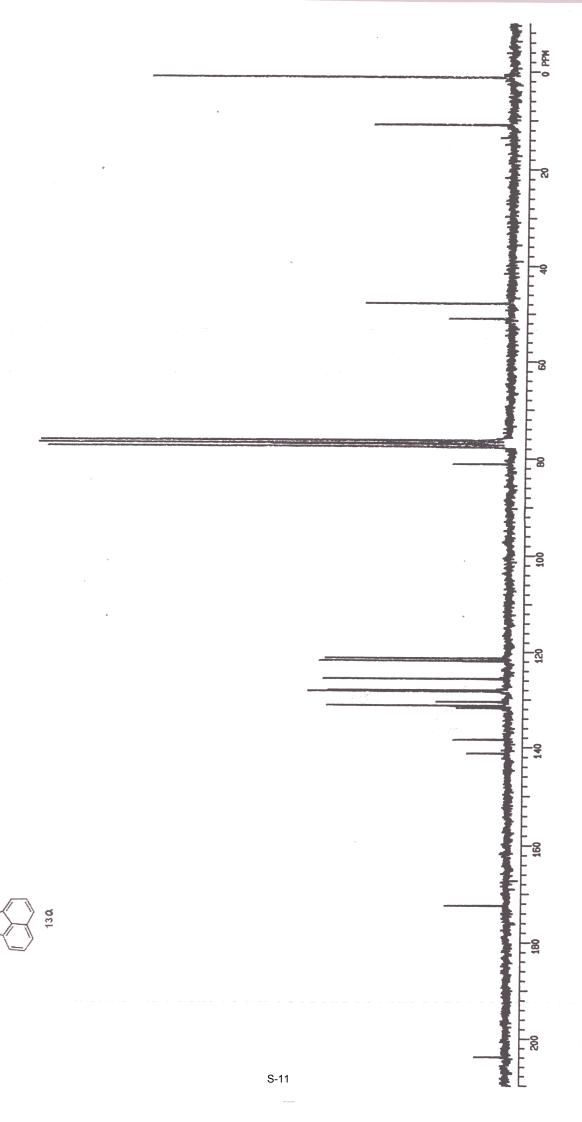


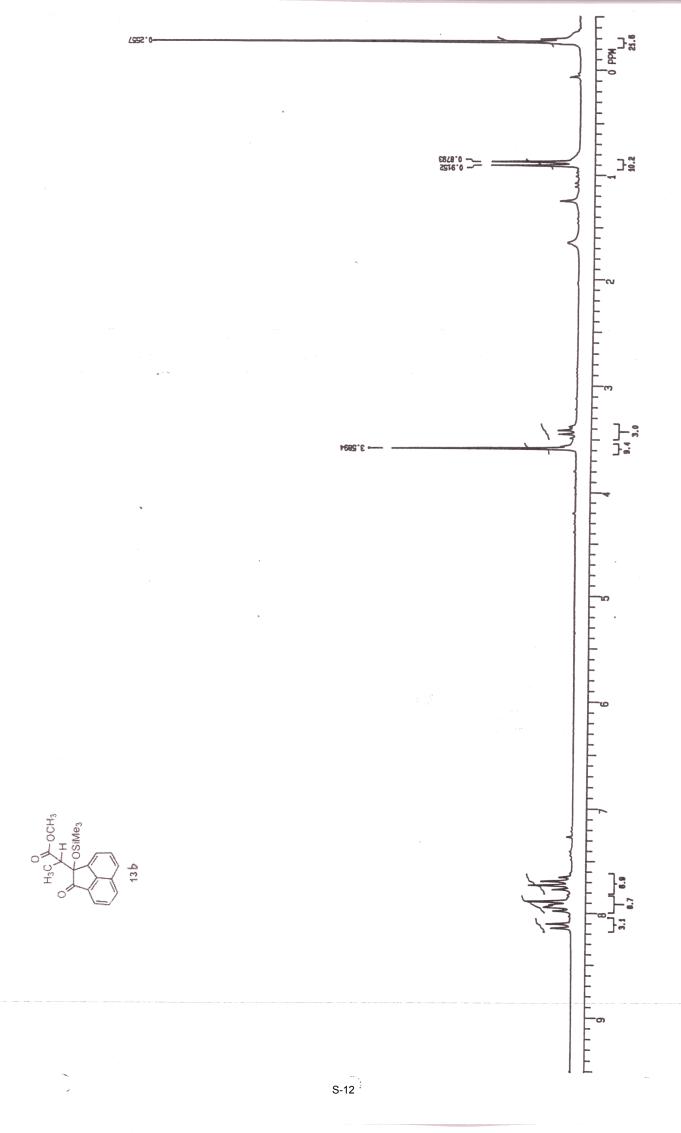


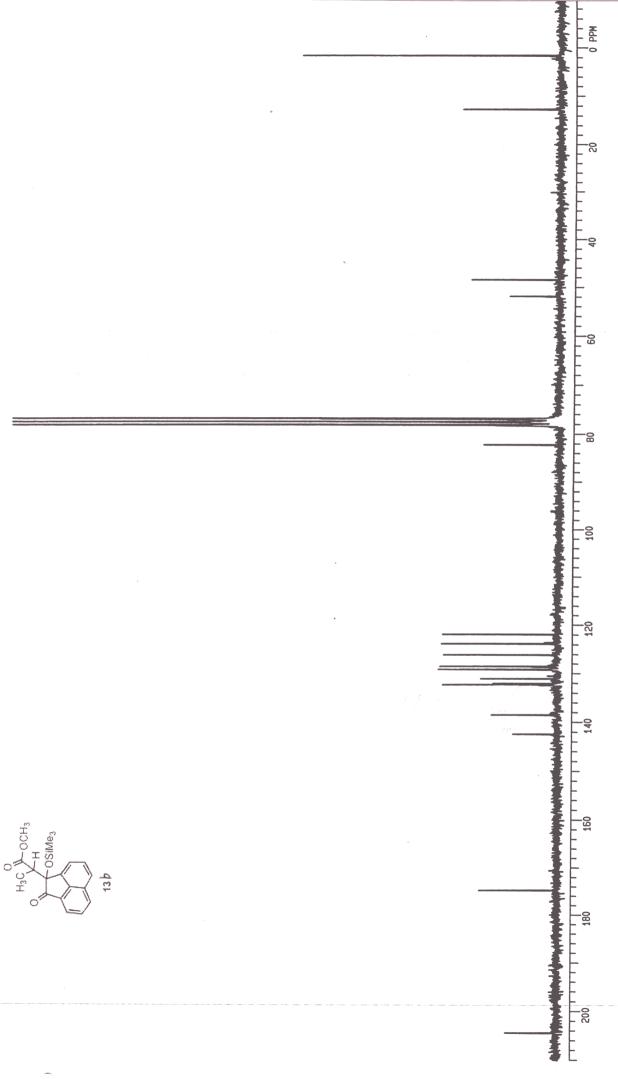
H₃CO CH₃ 2

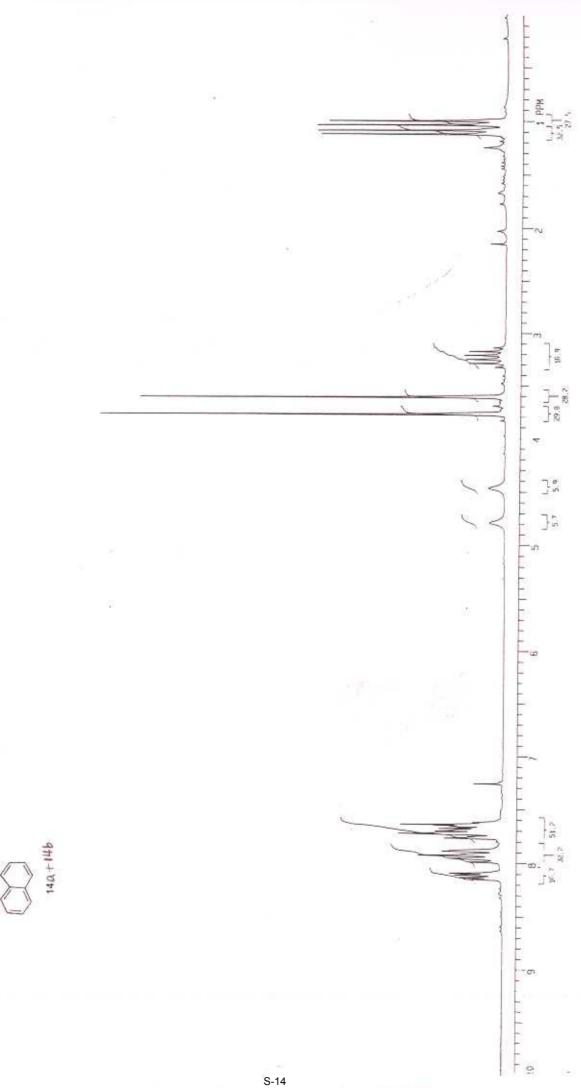


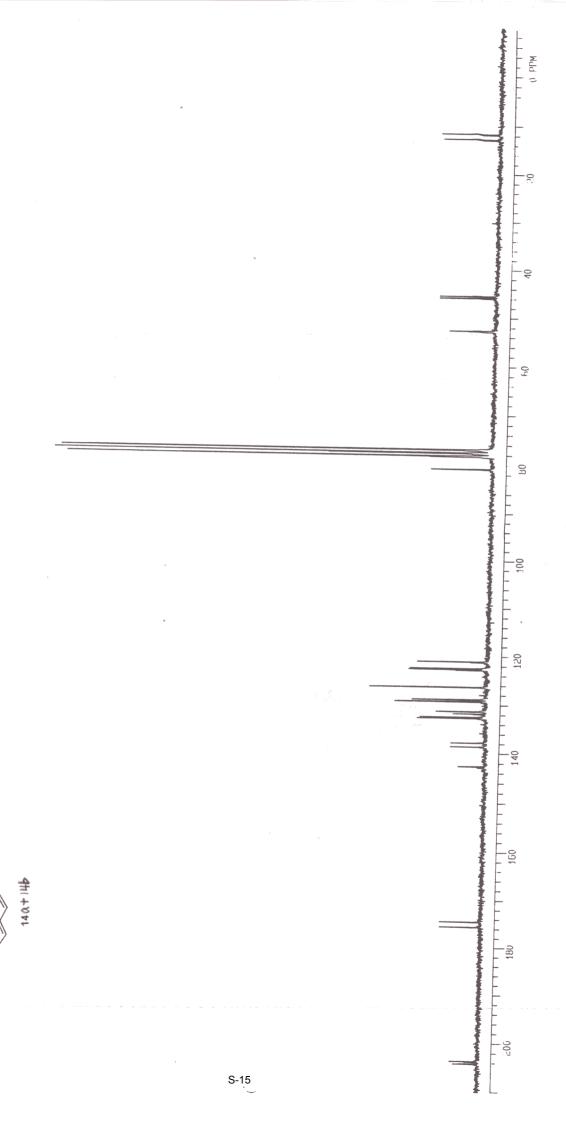


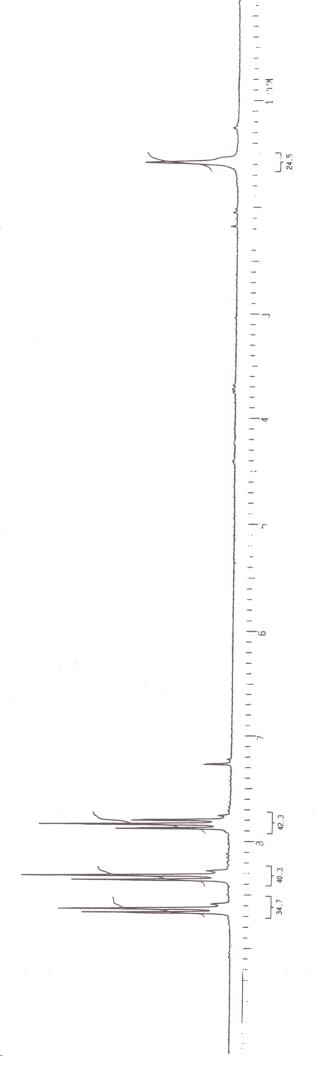


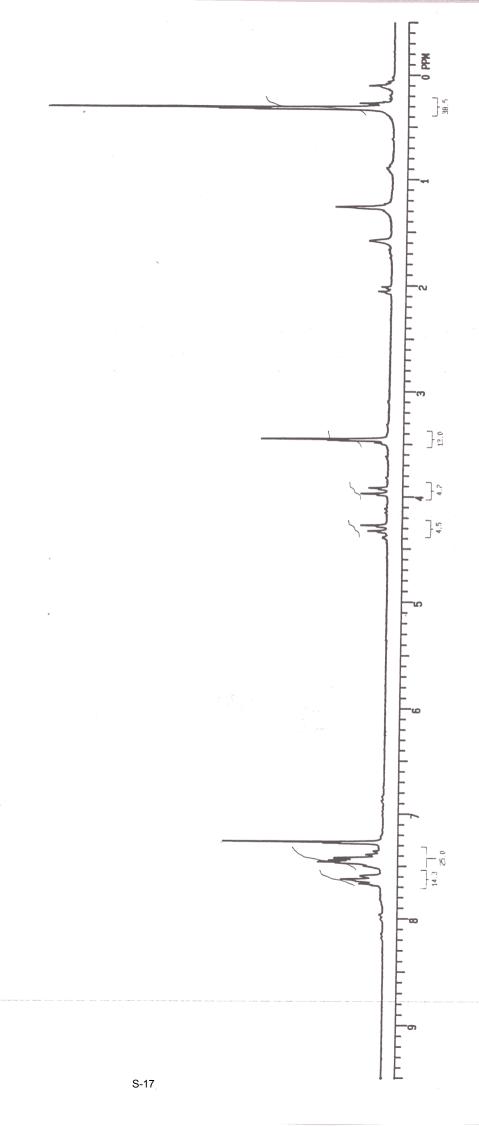


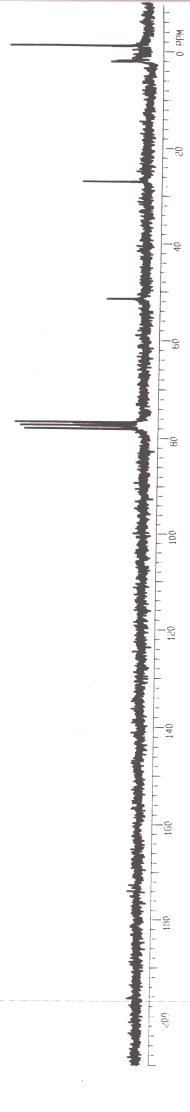


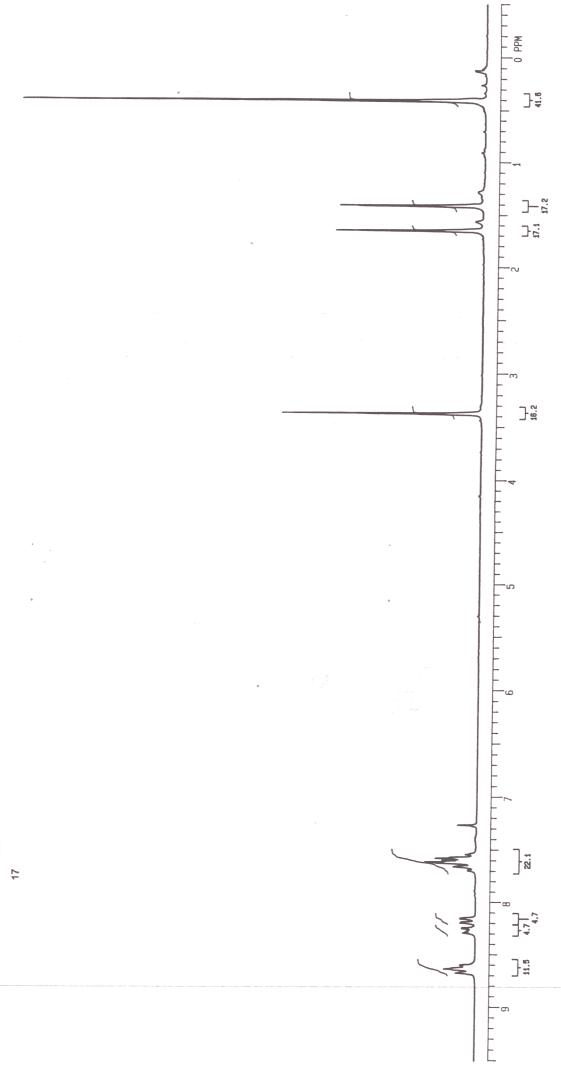




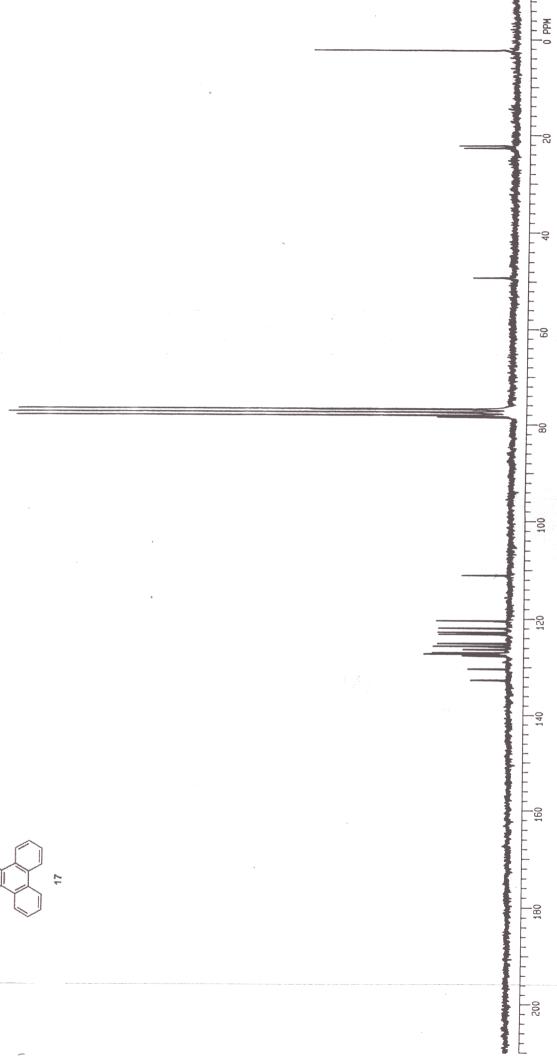


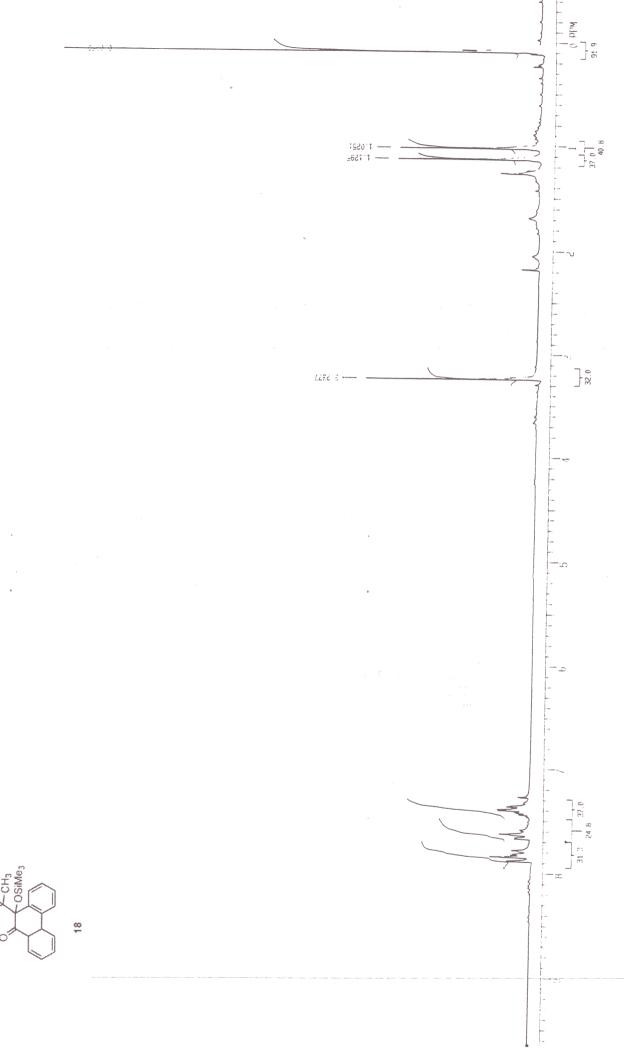


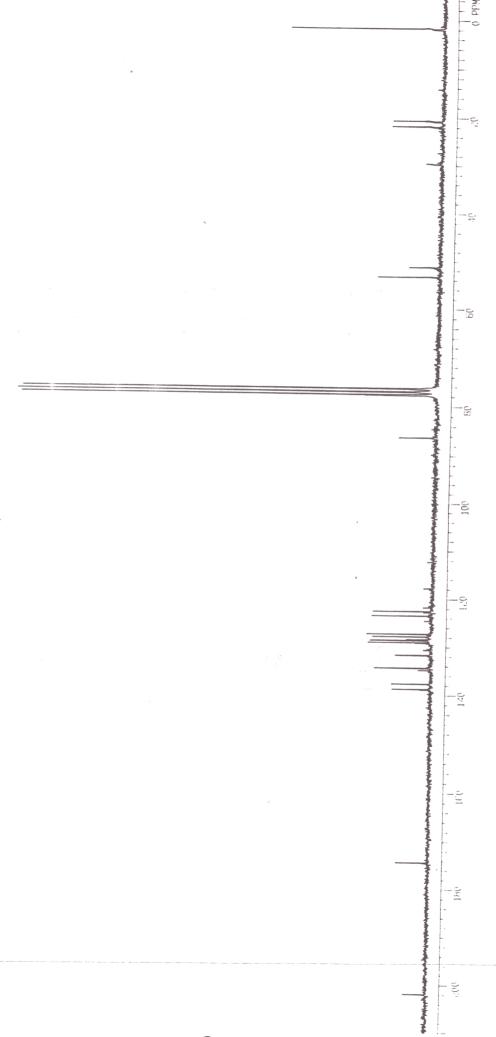


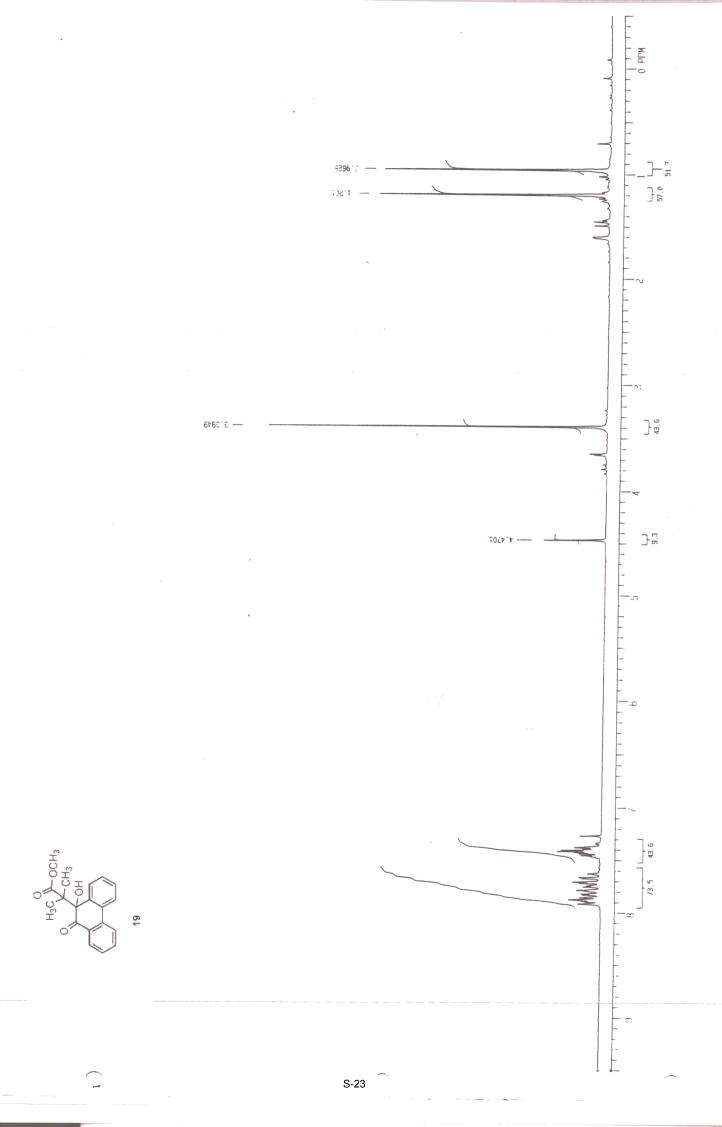


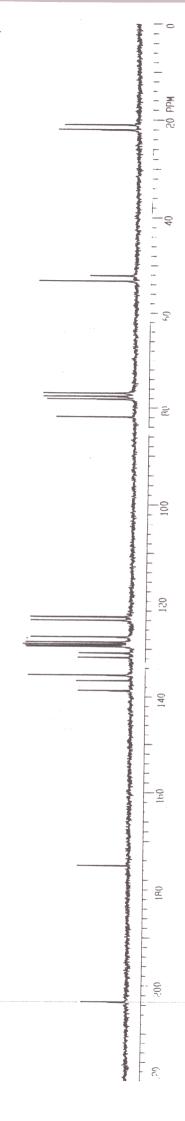
 H_3C OSIMe₃ H_3C \leftarrow OCH₃

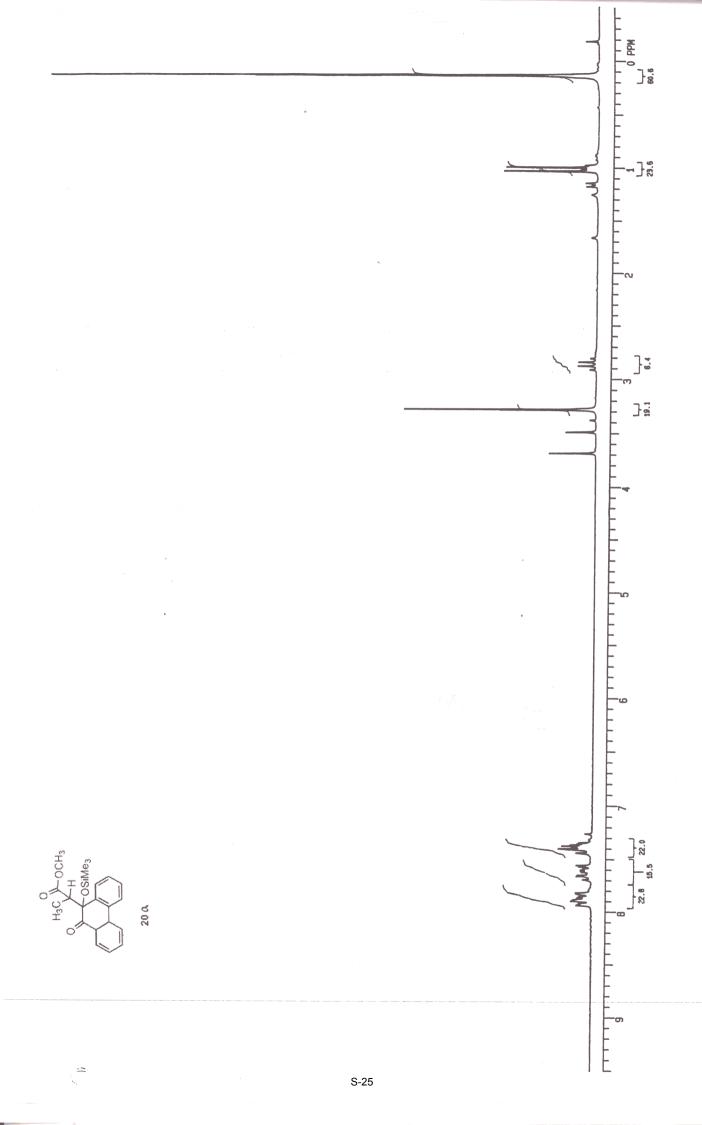


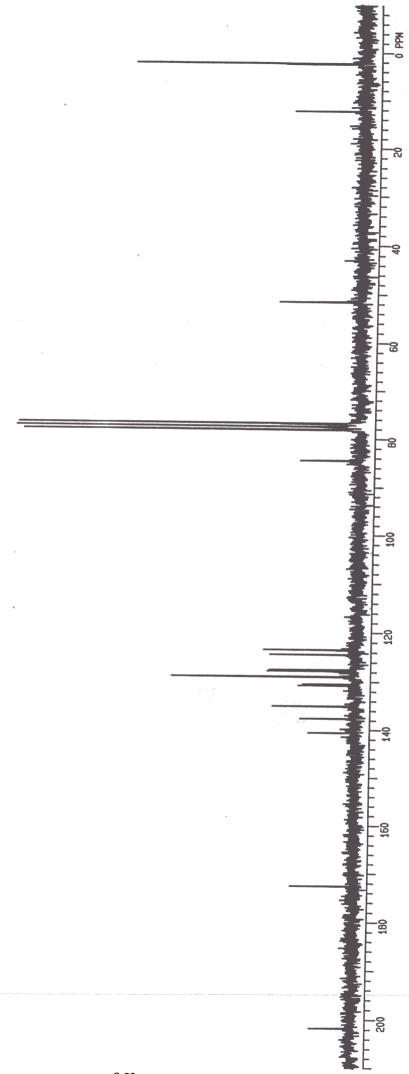




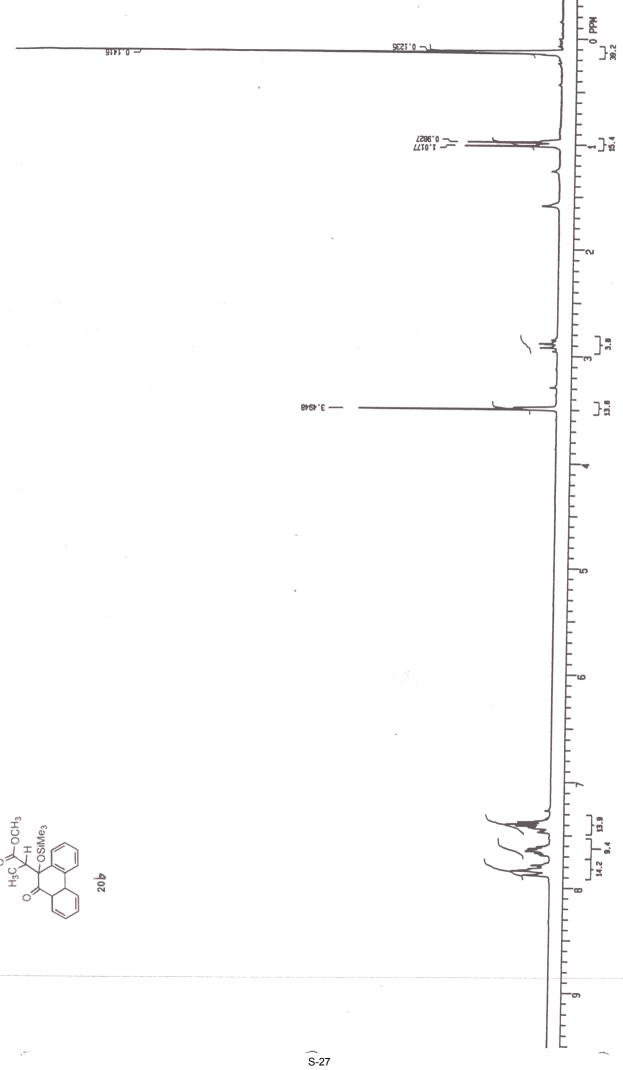


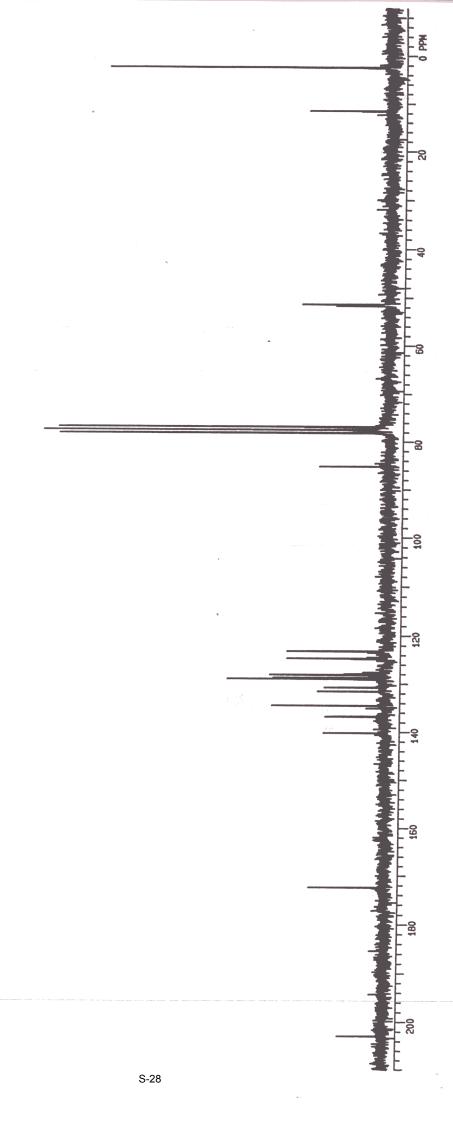


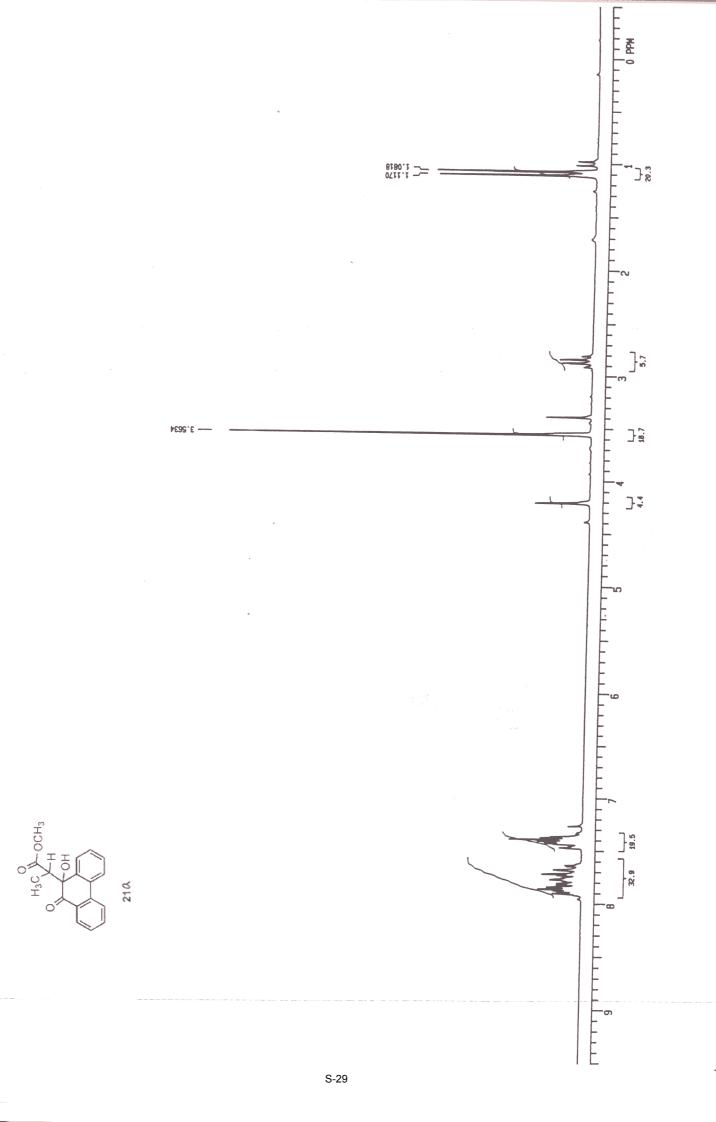


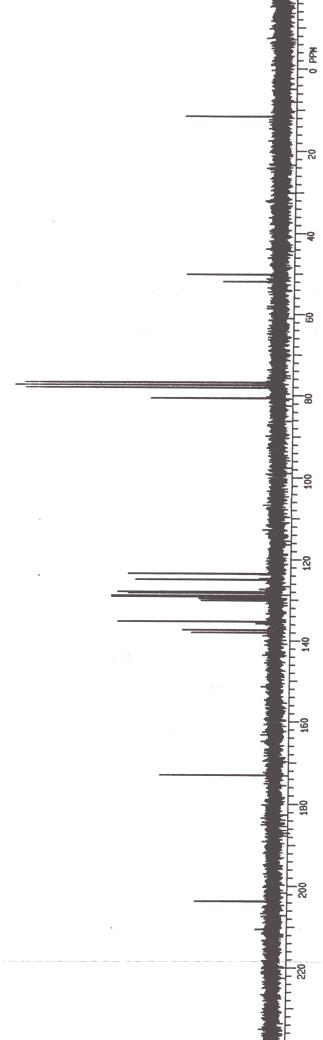


20 A

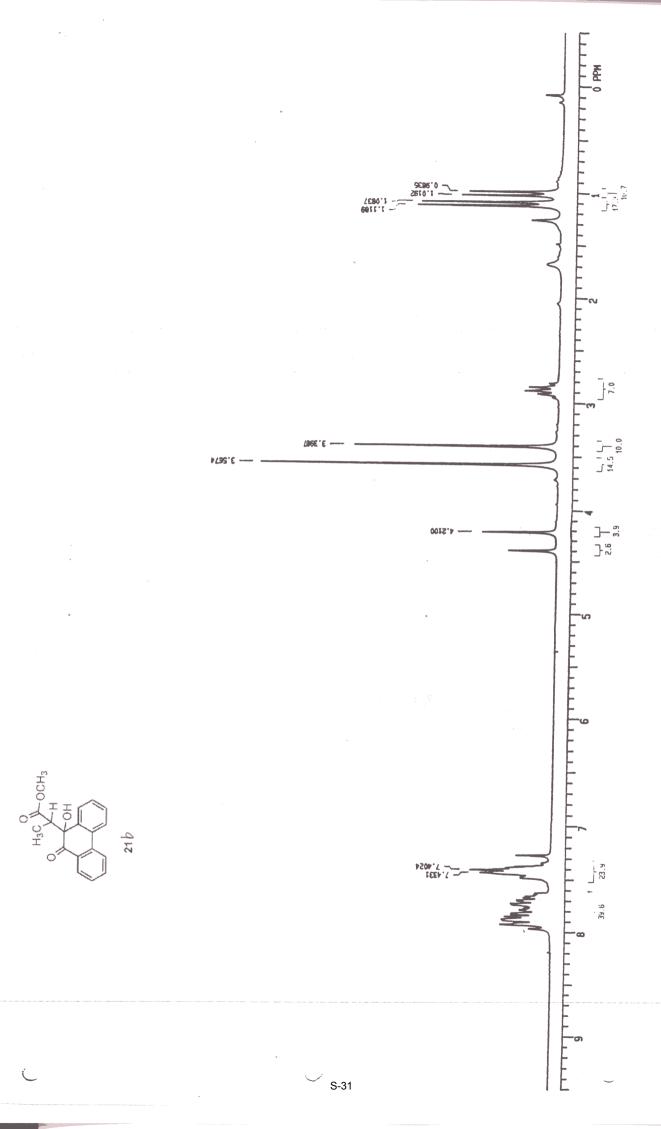


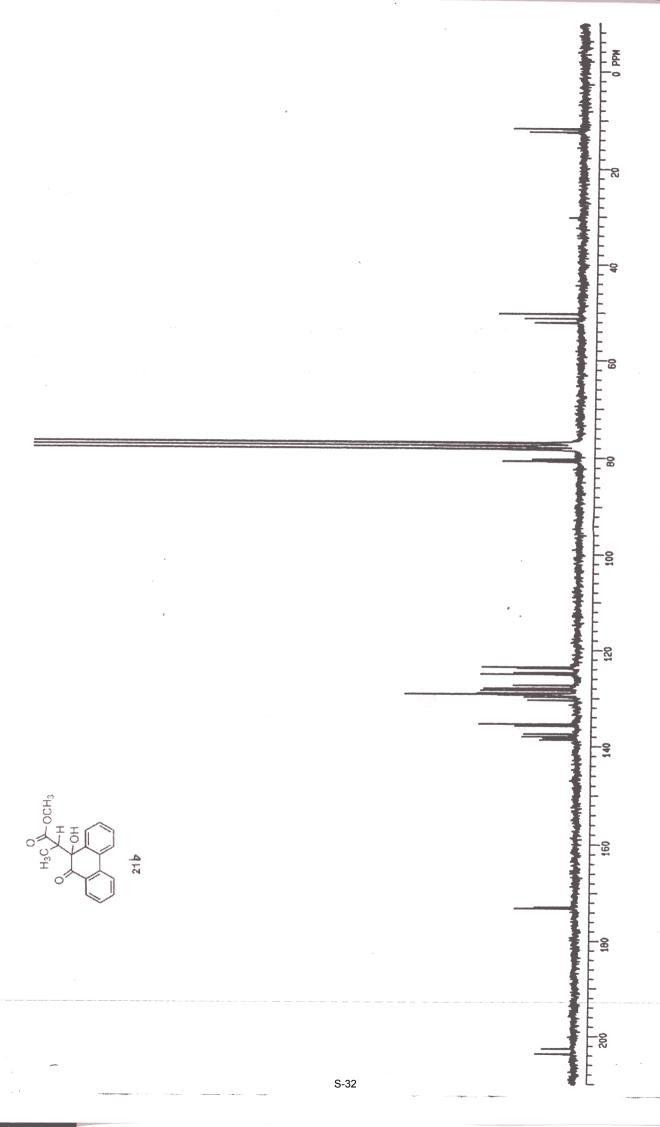


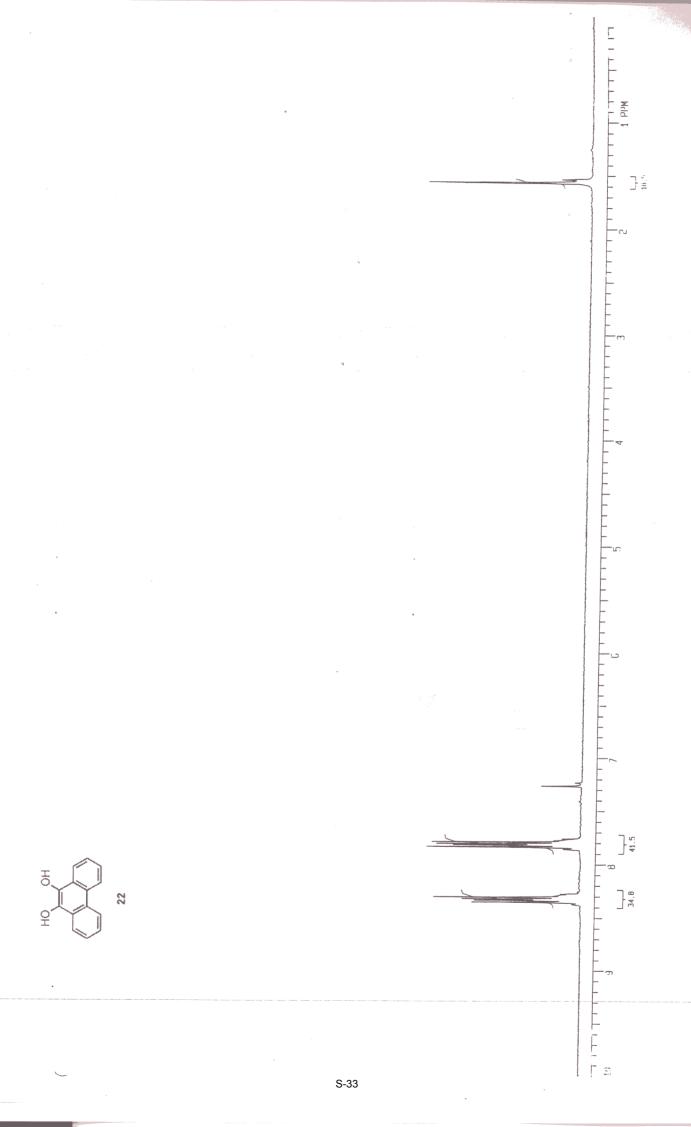


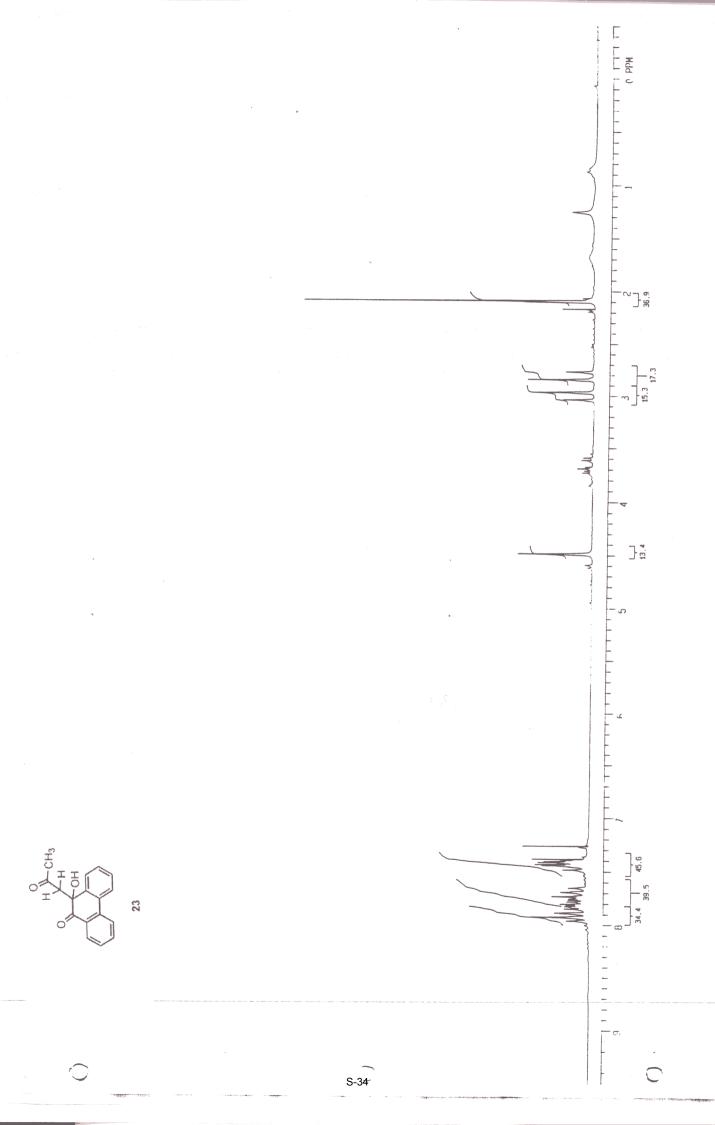


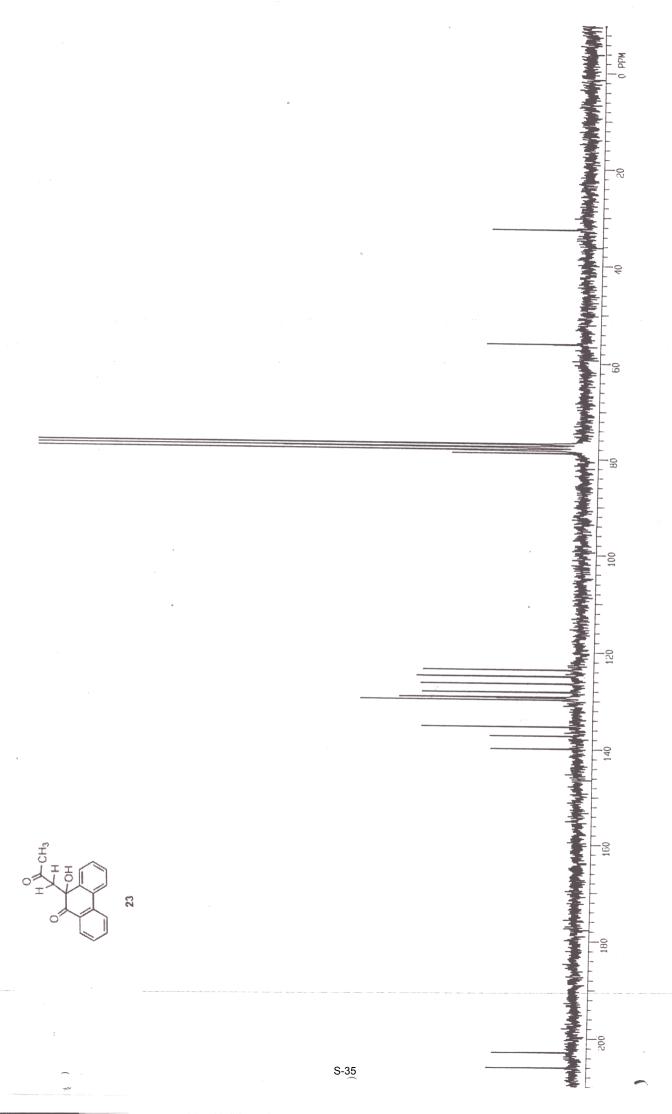
21a







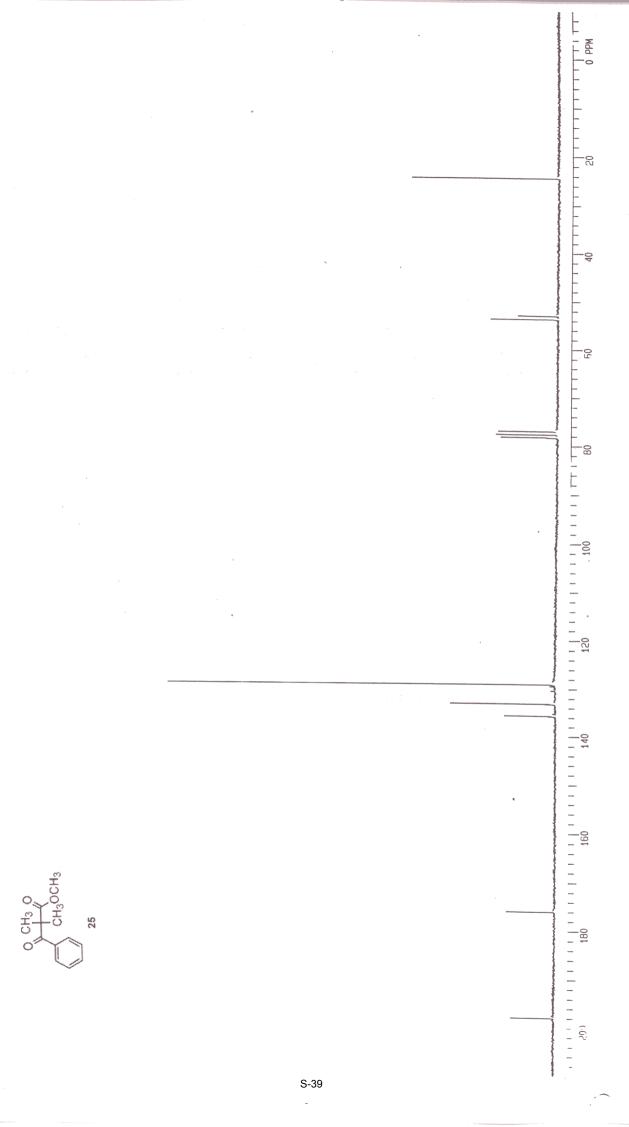


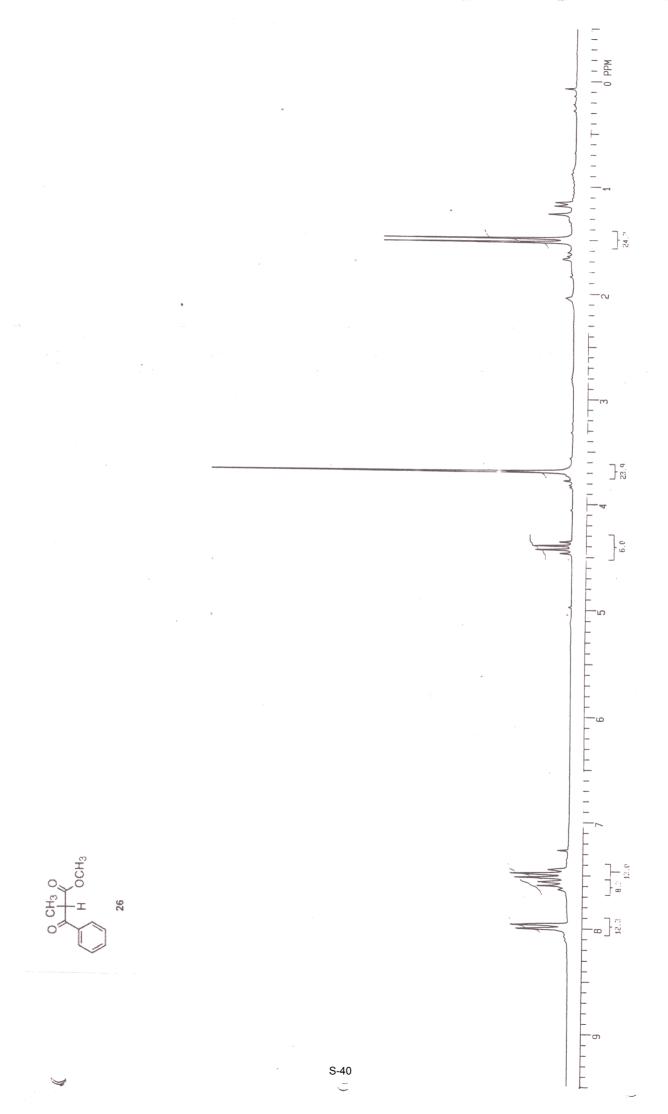


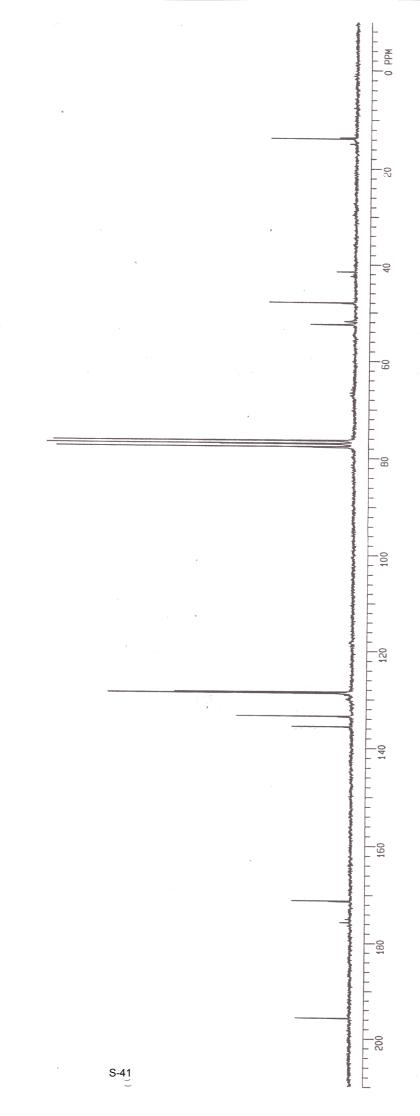
7595.0 -S-36

Wind o

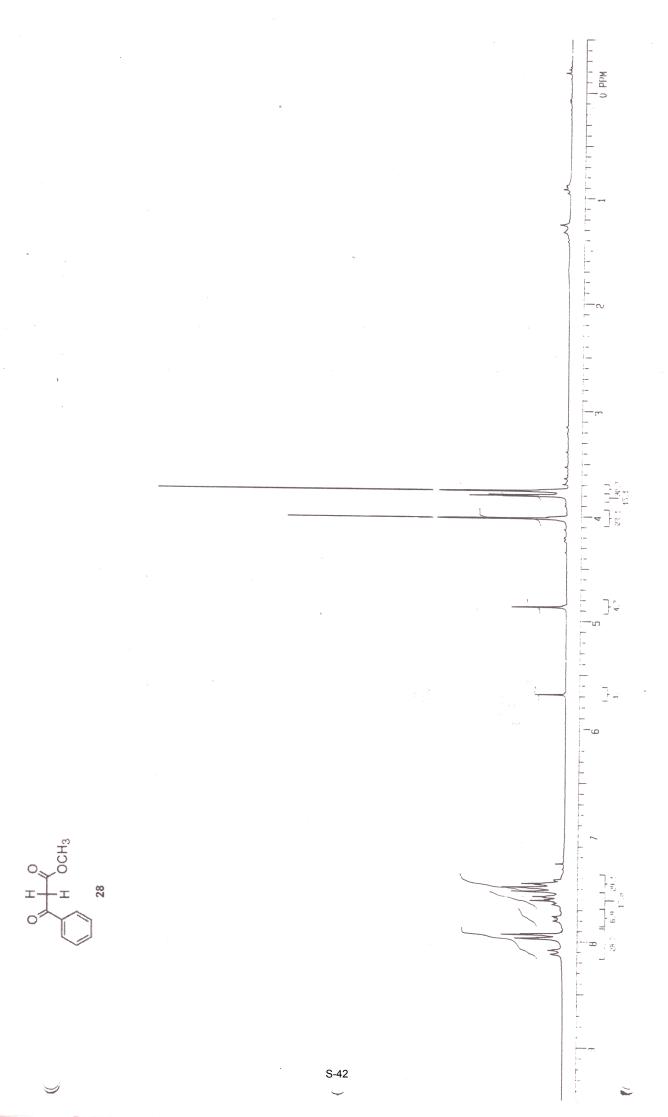
75.0 31.2

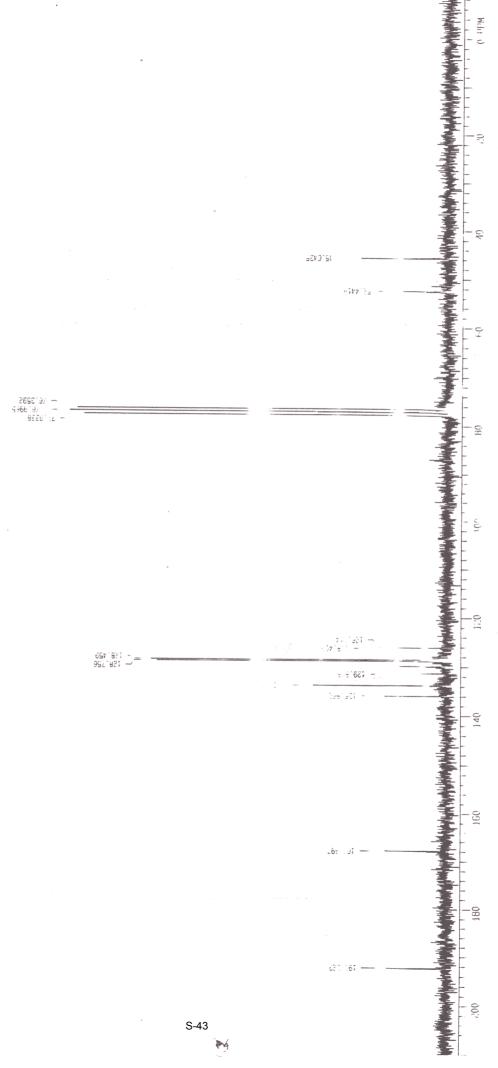




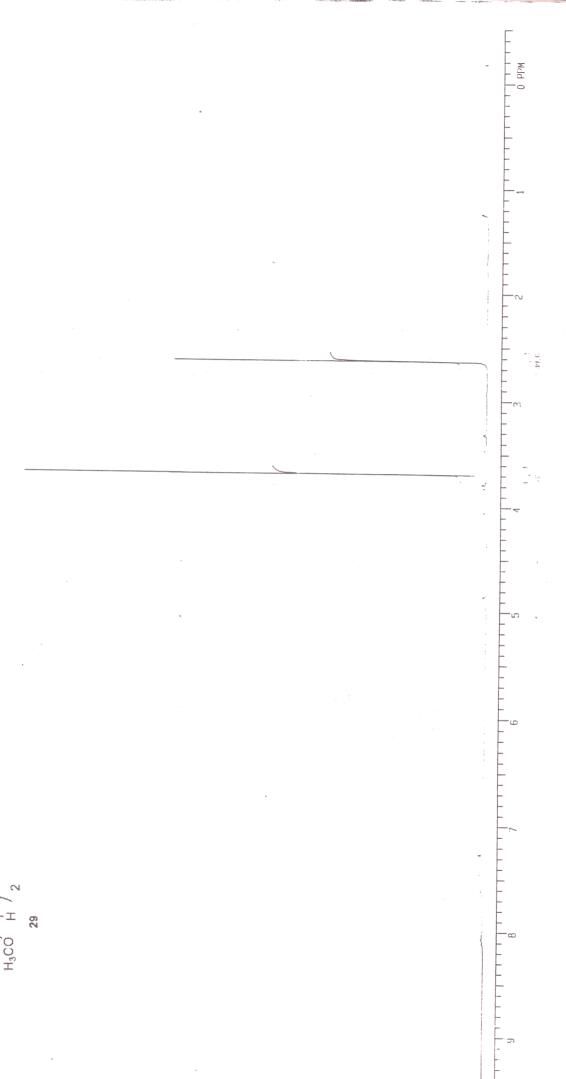


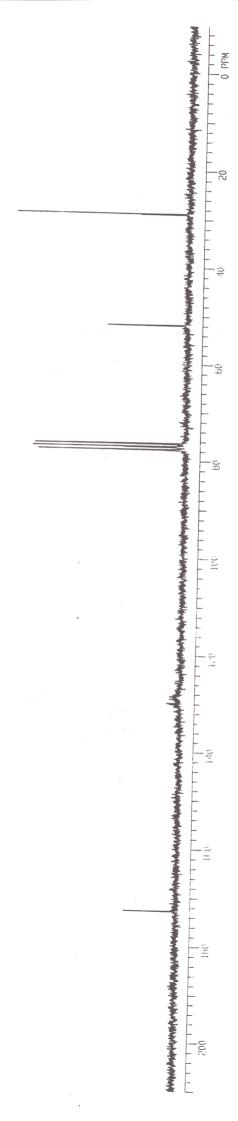
Z T CH3

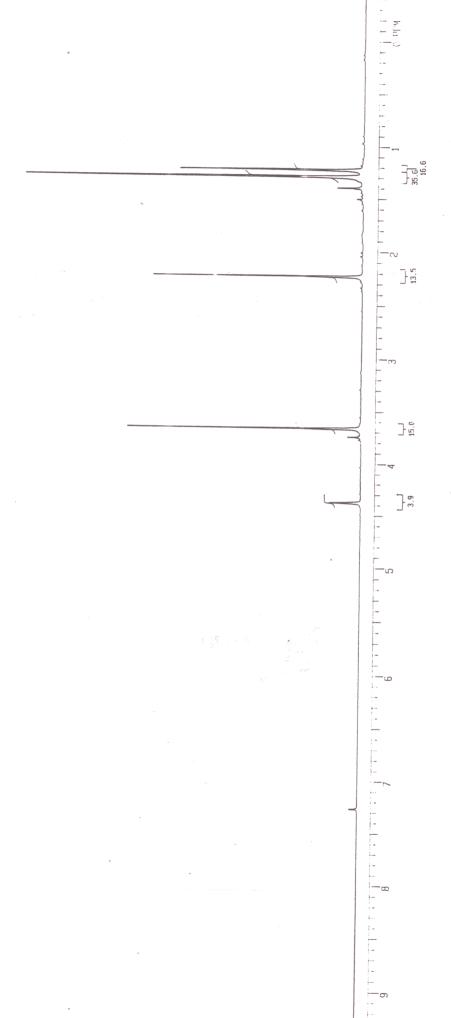




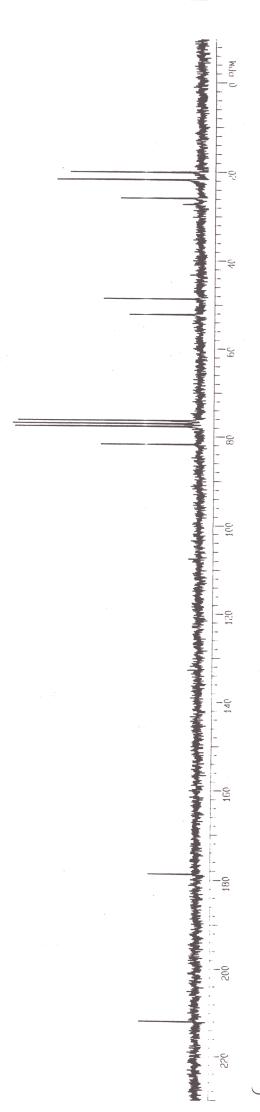
S-43





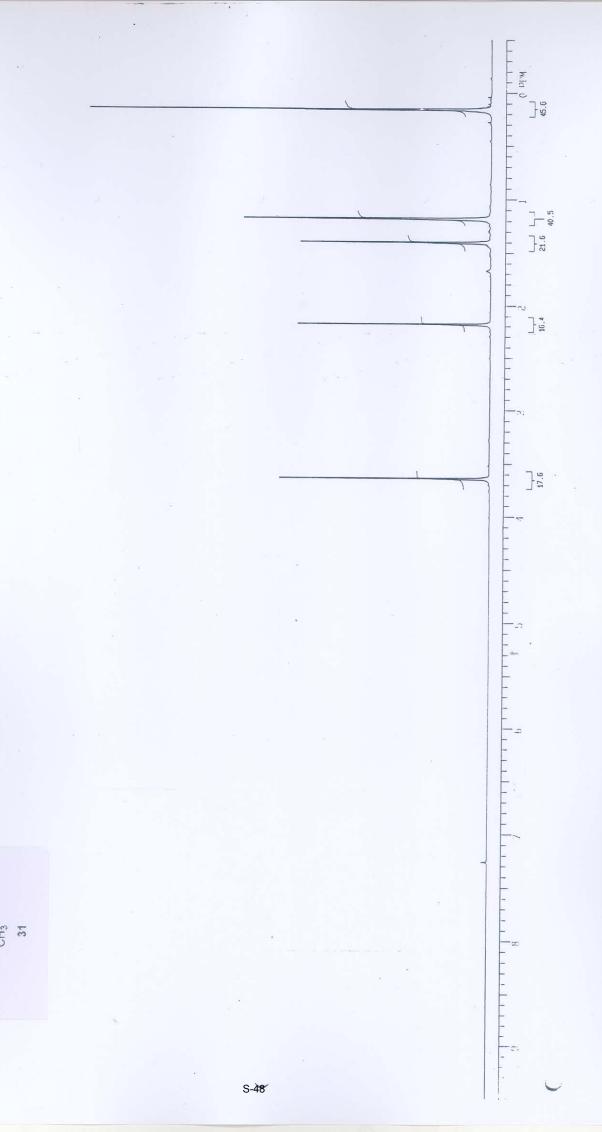


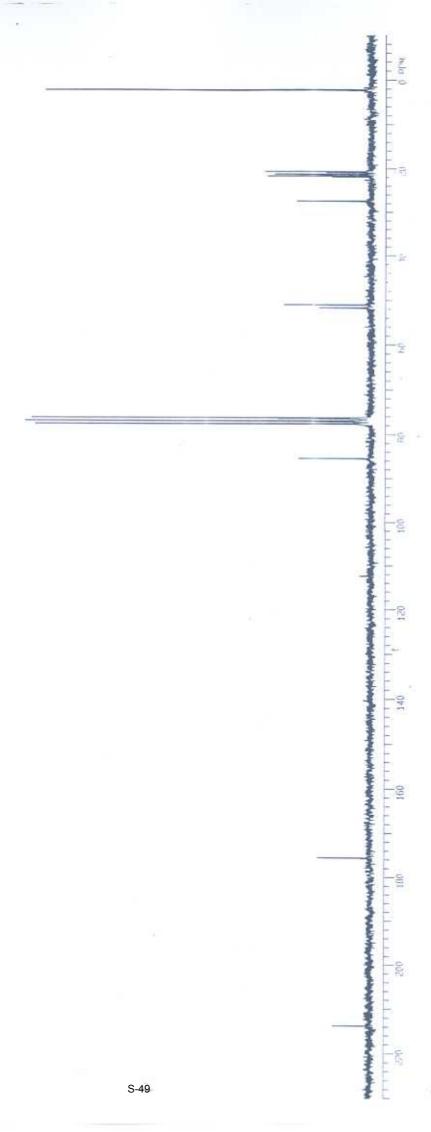
H₃C CH₃ CH₃ OC

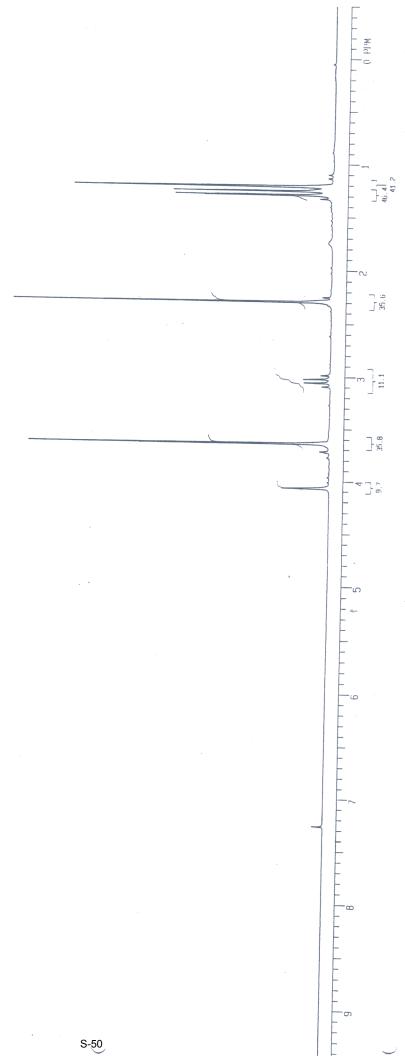


H₃C CH₃ OCH₃

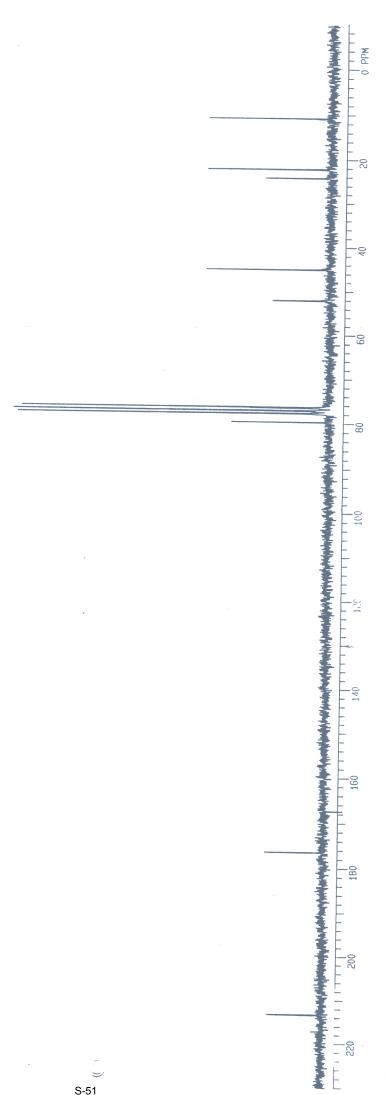
S-47



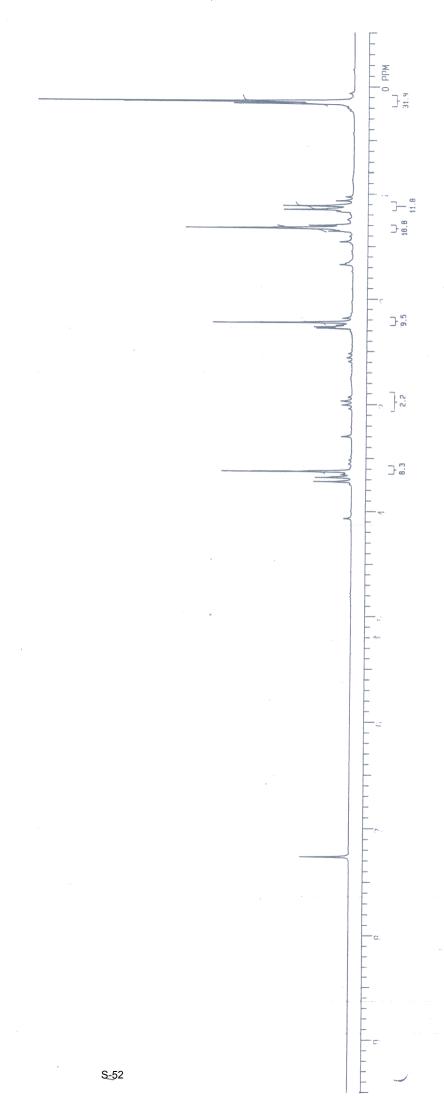




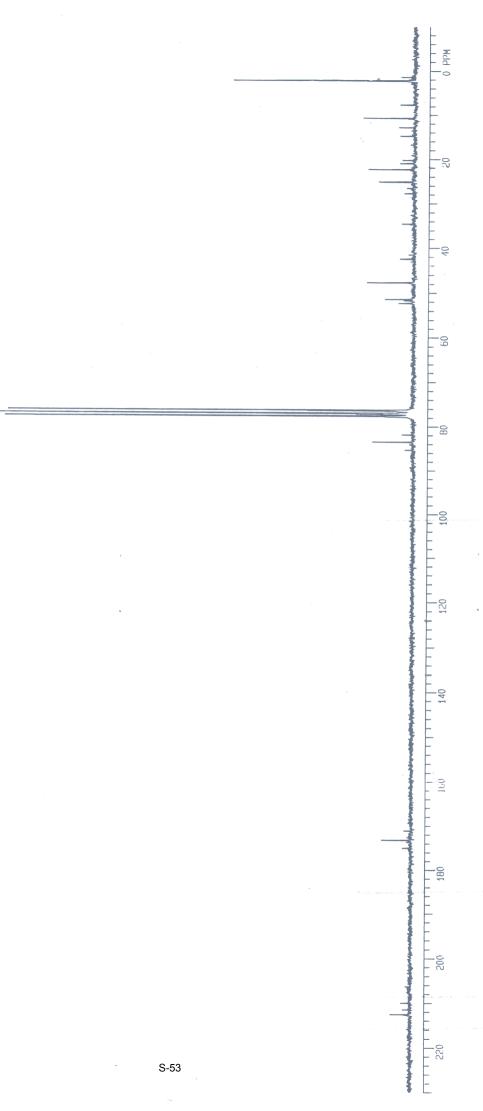
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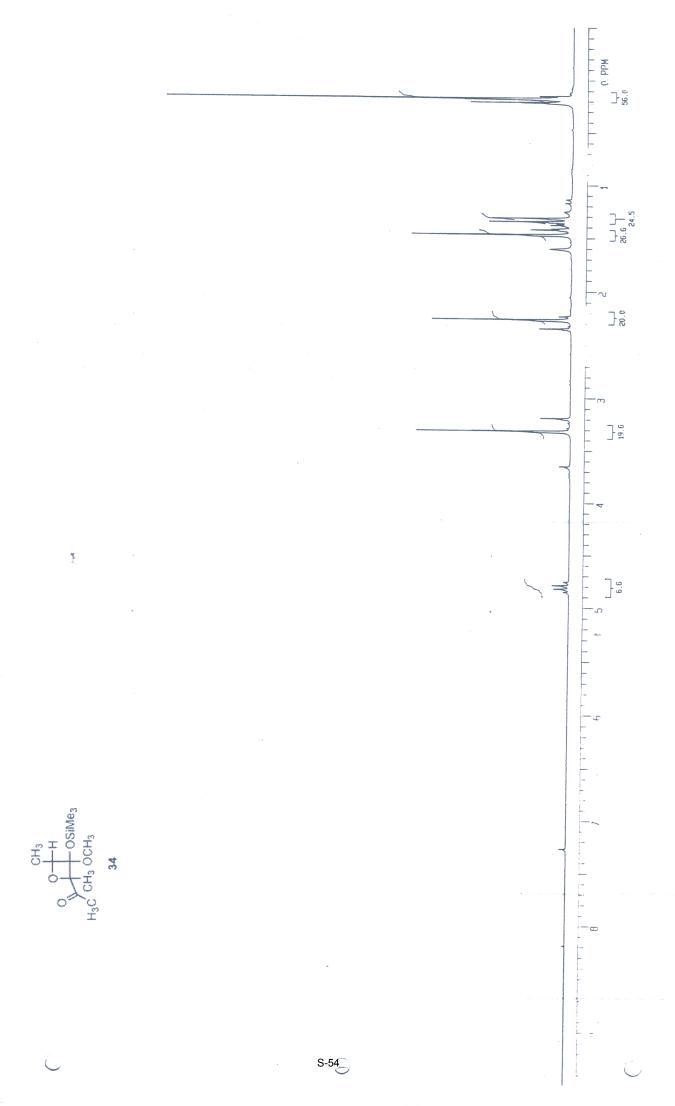


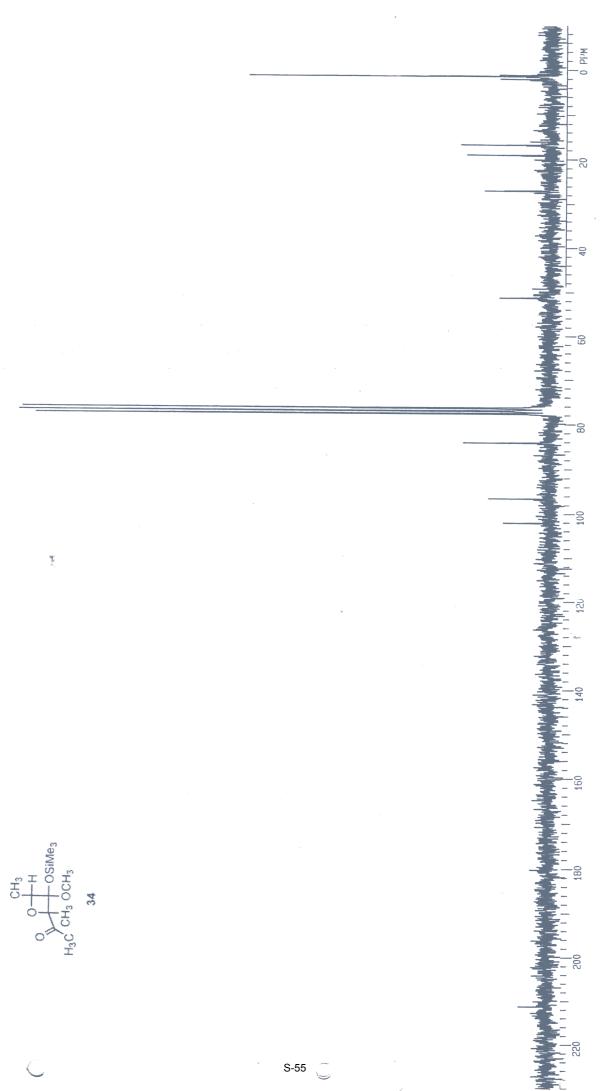
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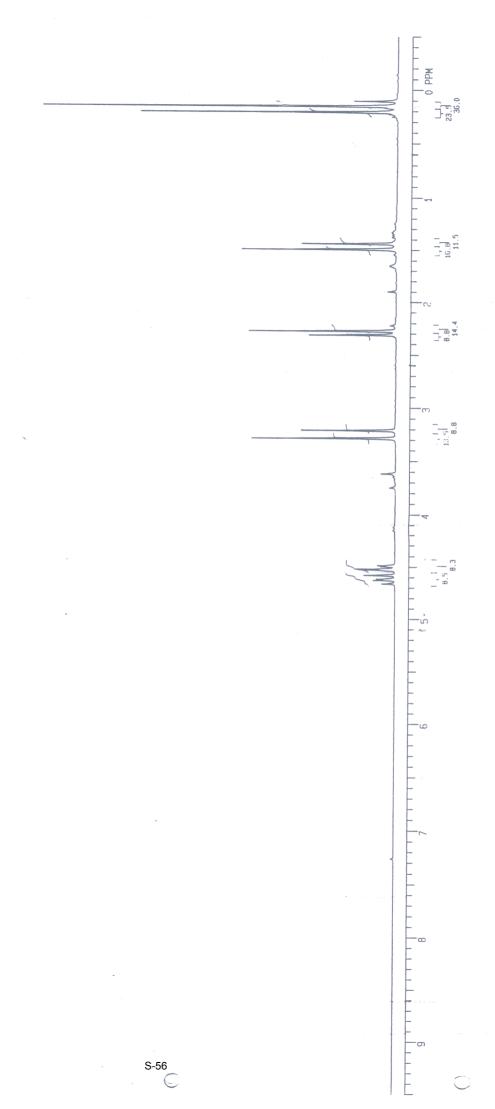
Me₃SiOH





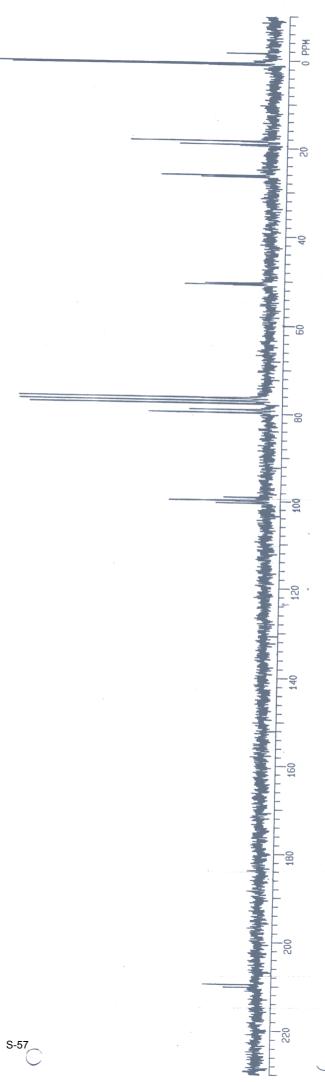


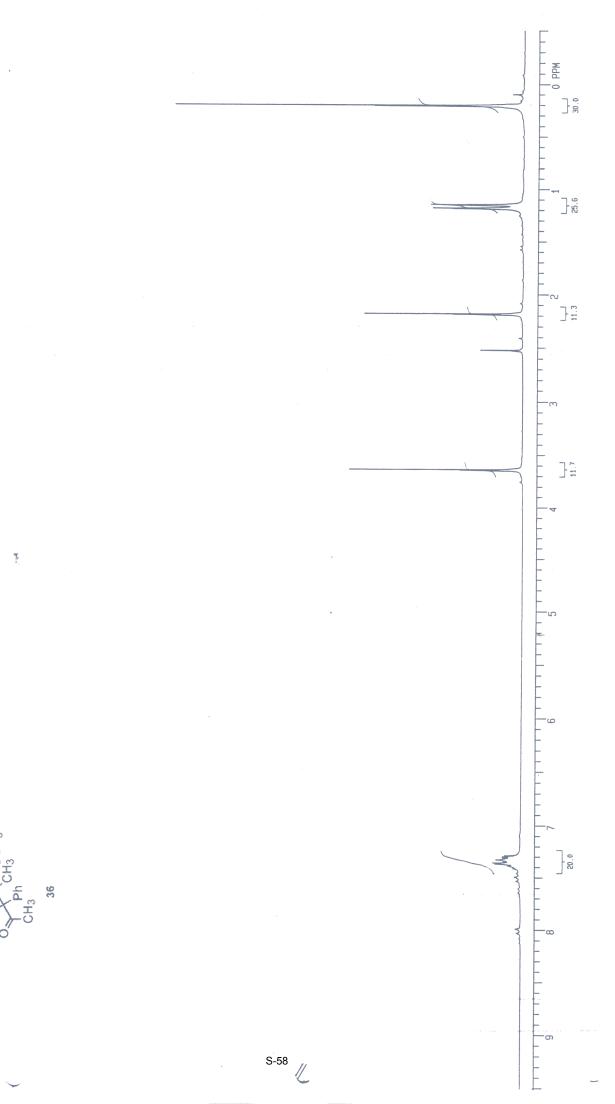
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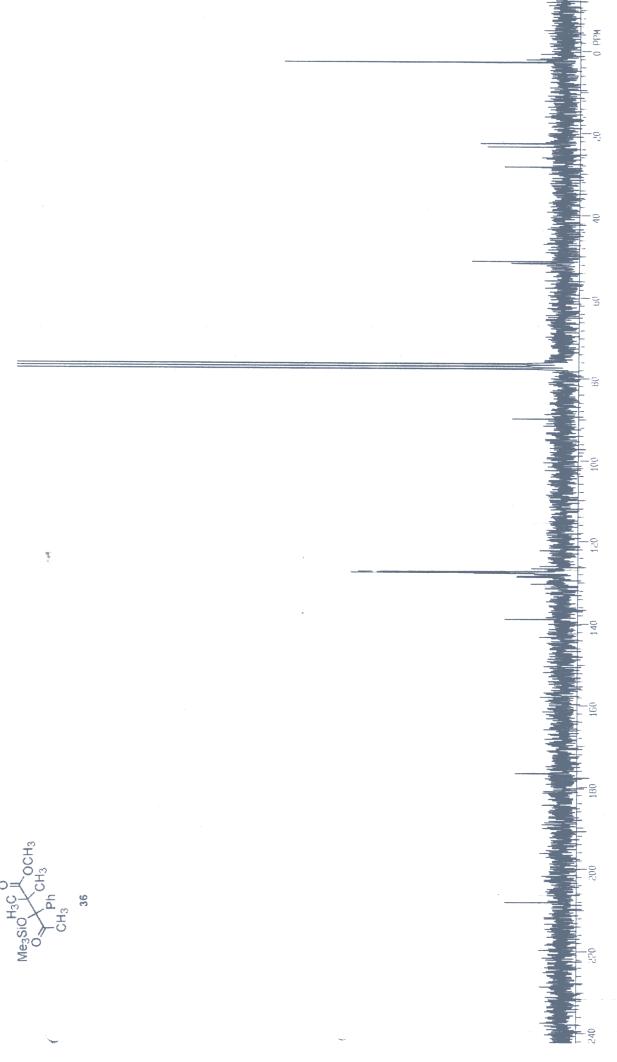


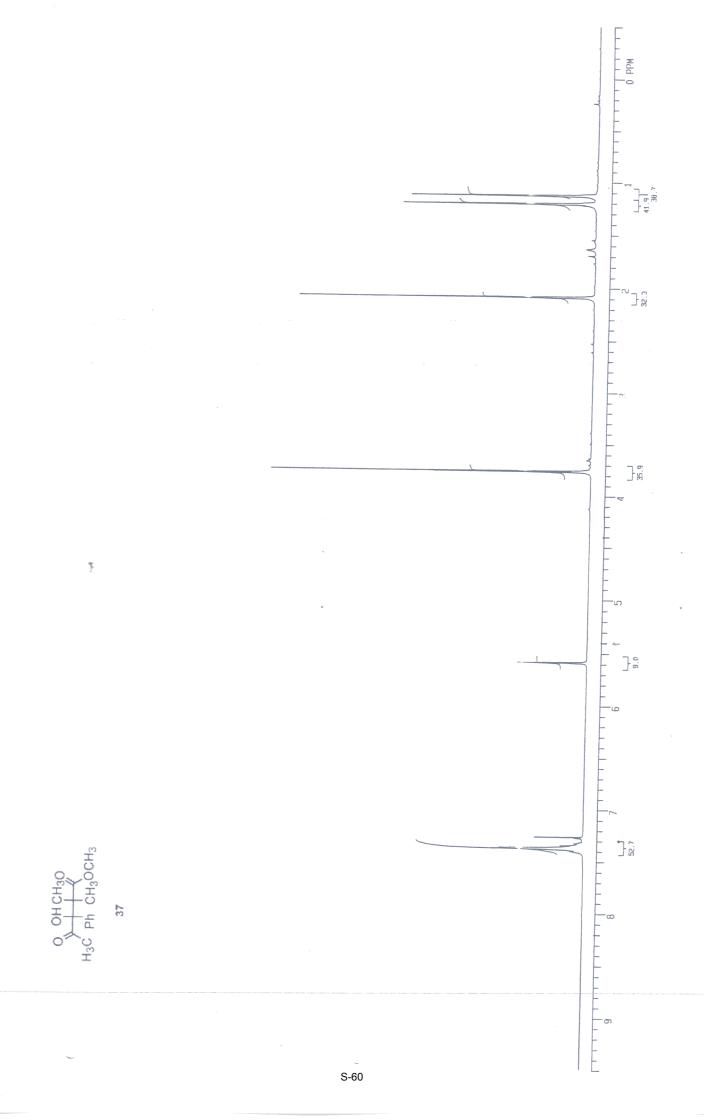
H₃C CH₃ OCH₃

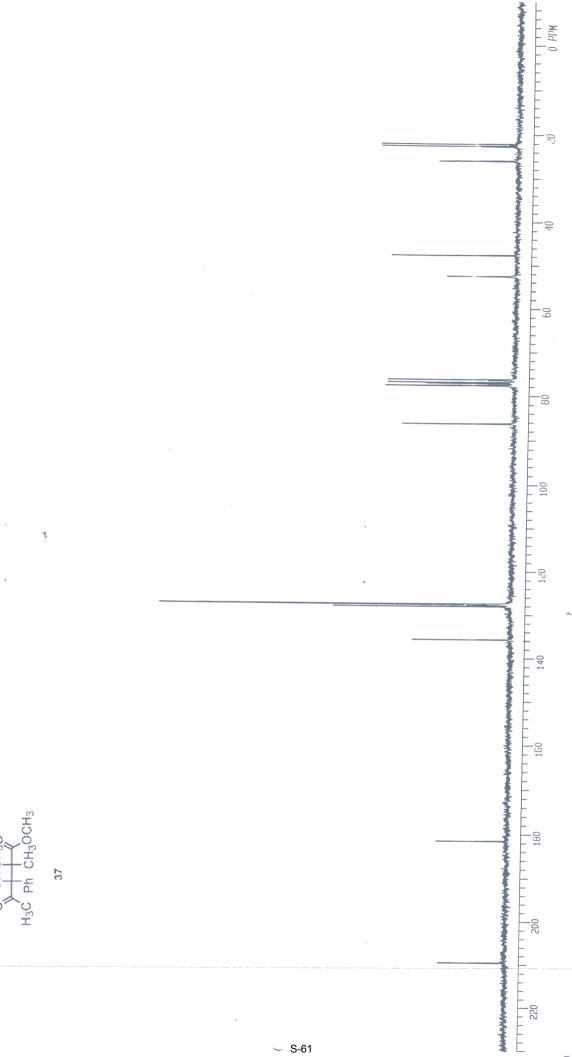
350+356

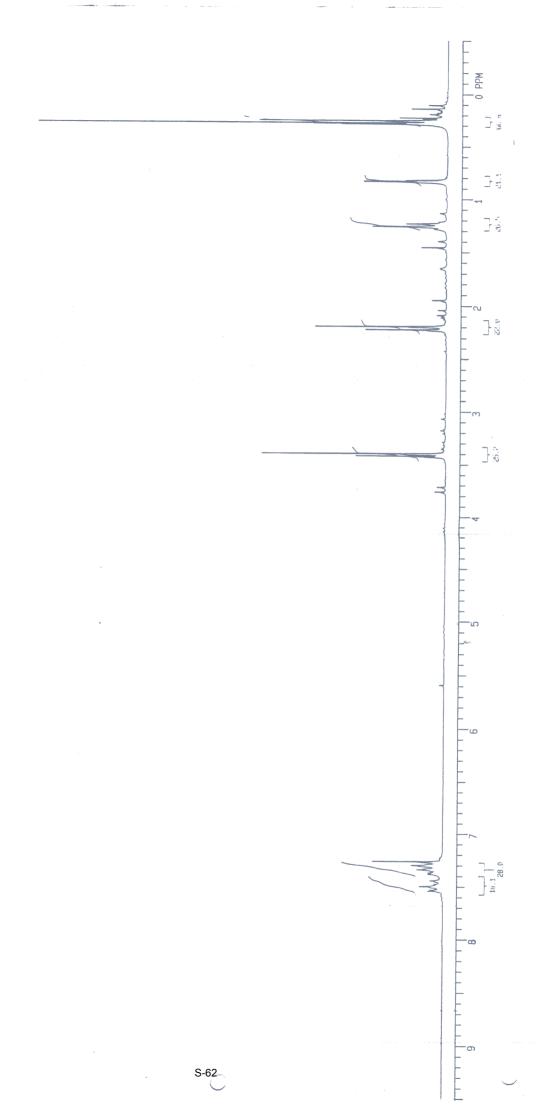






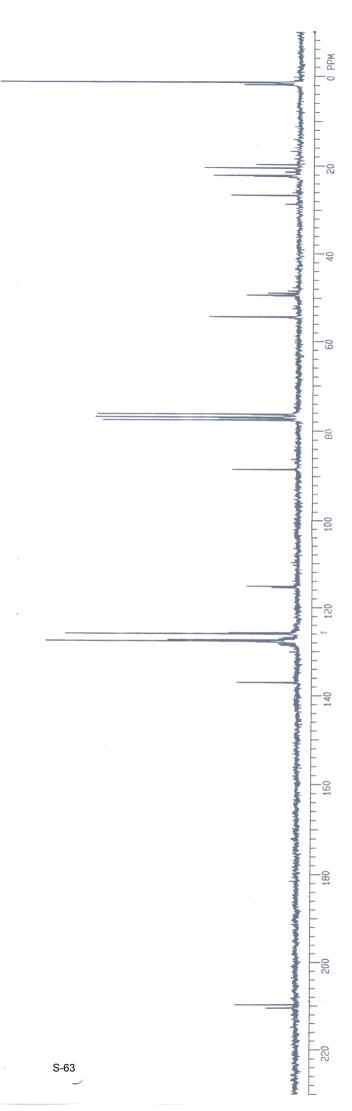




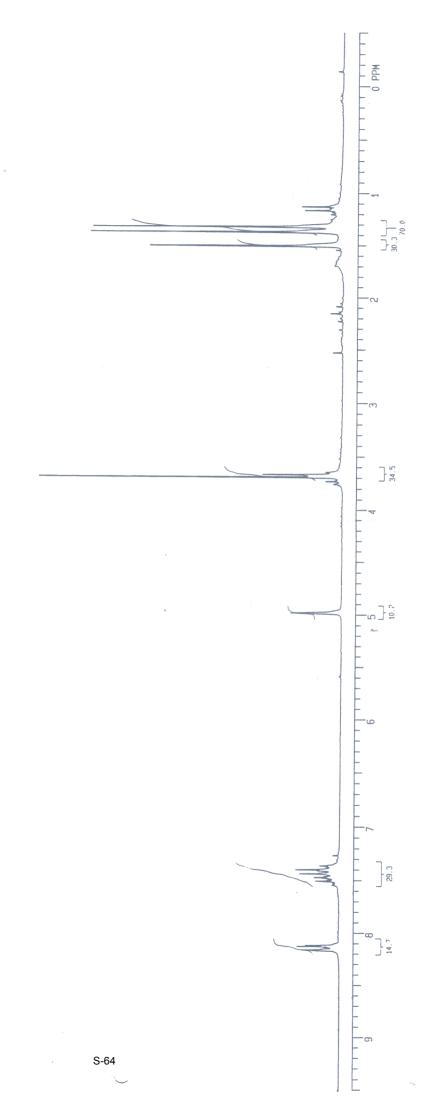


OSiMe₃
OOCH₃
H₃C Ph CH₃

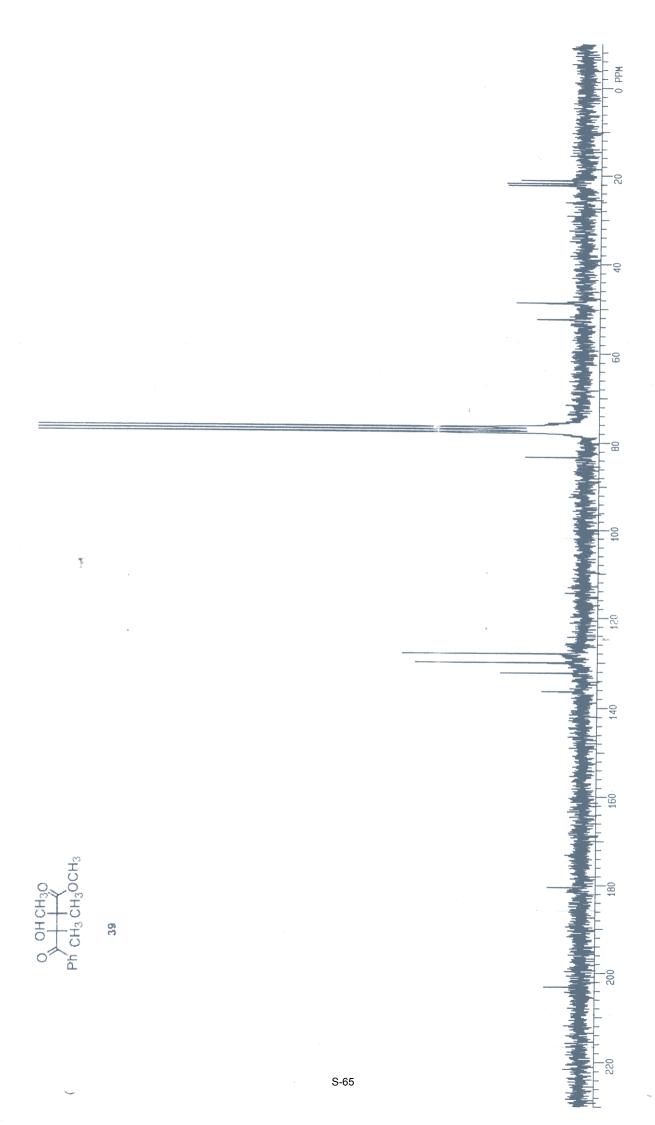
38 a + 38b

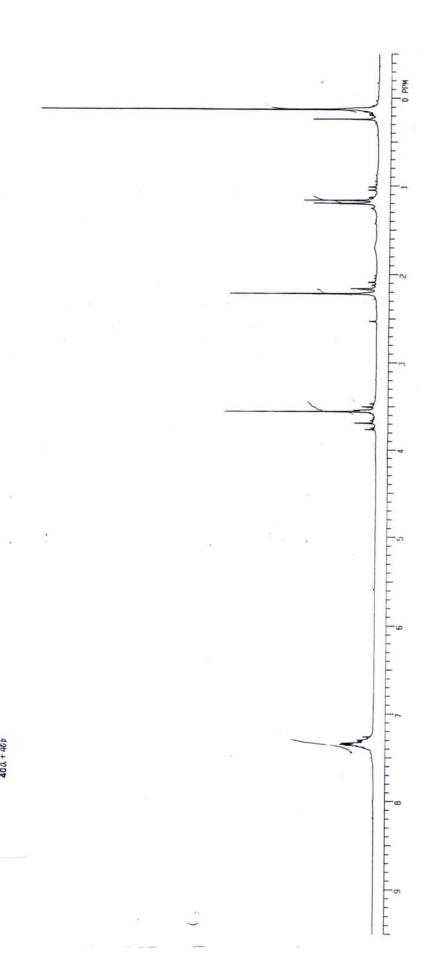


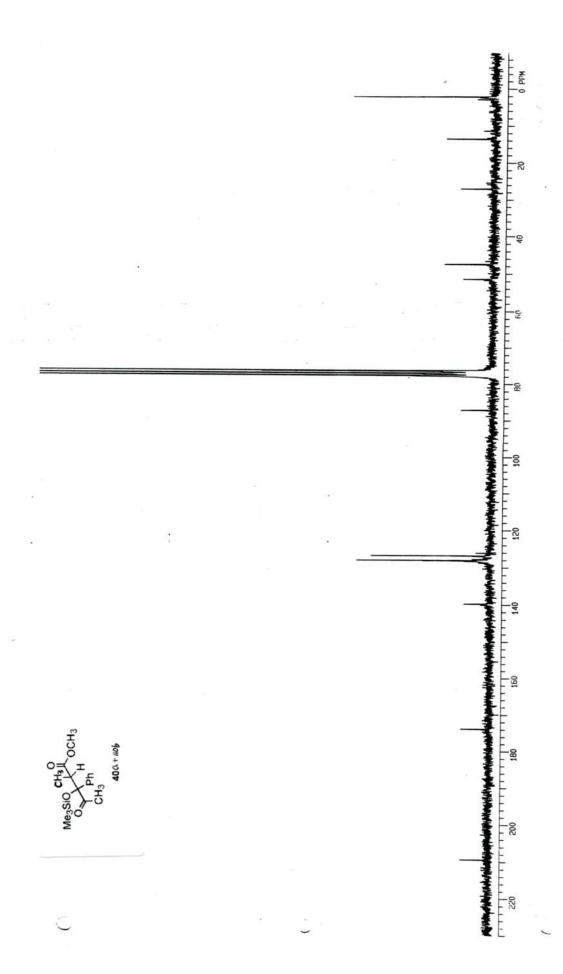
38a+38b

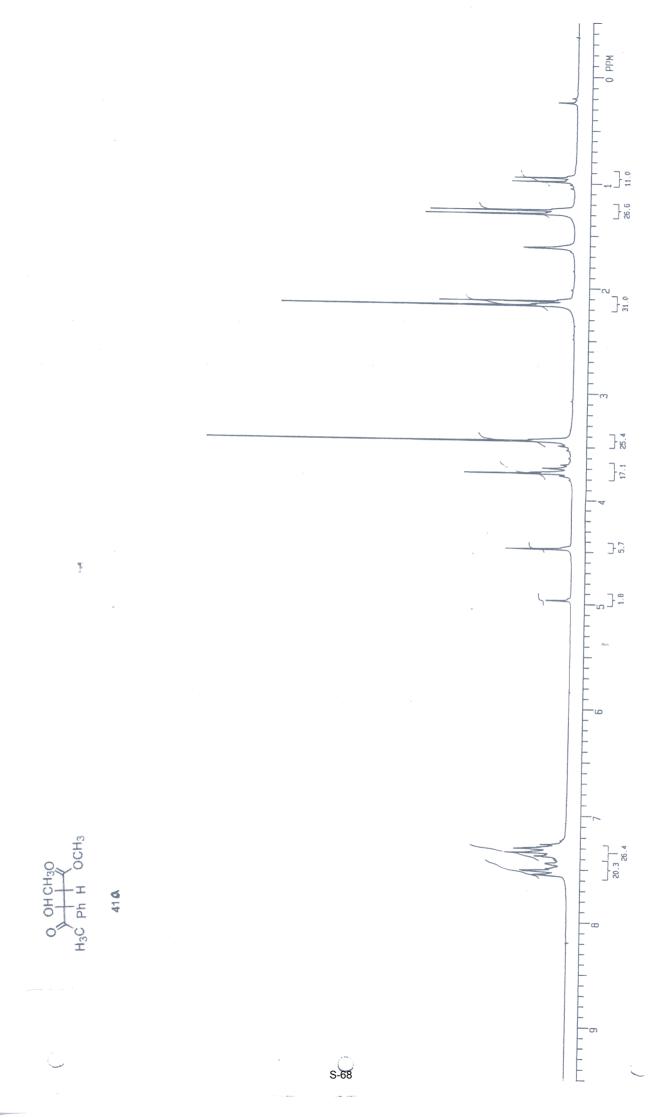


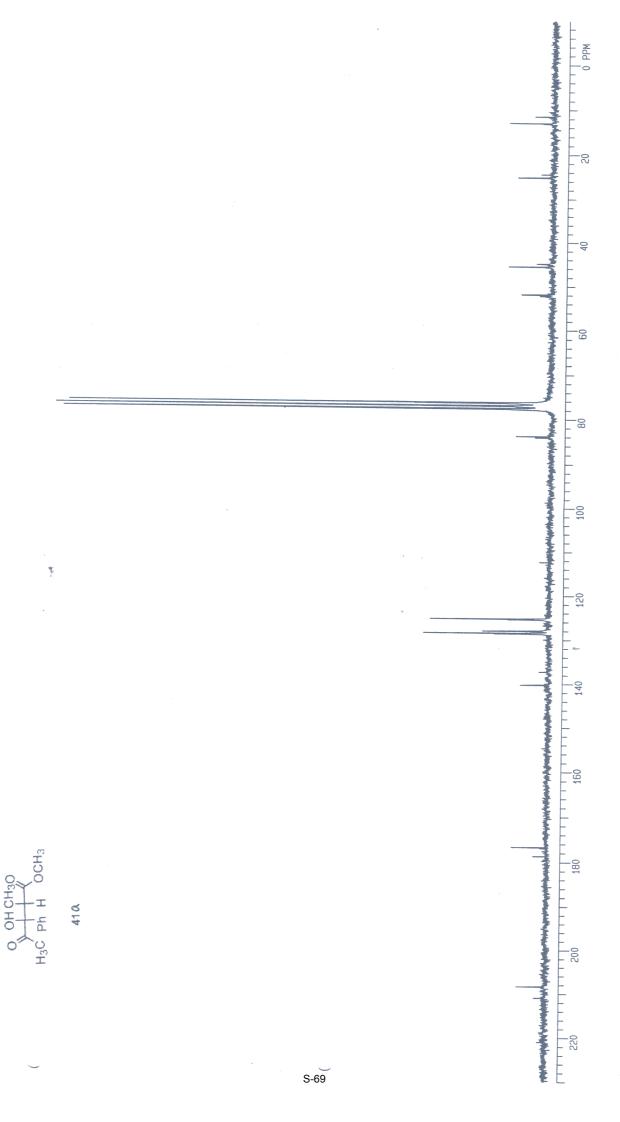
O OH CH₃O Ph CH₃ CH₃OCH

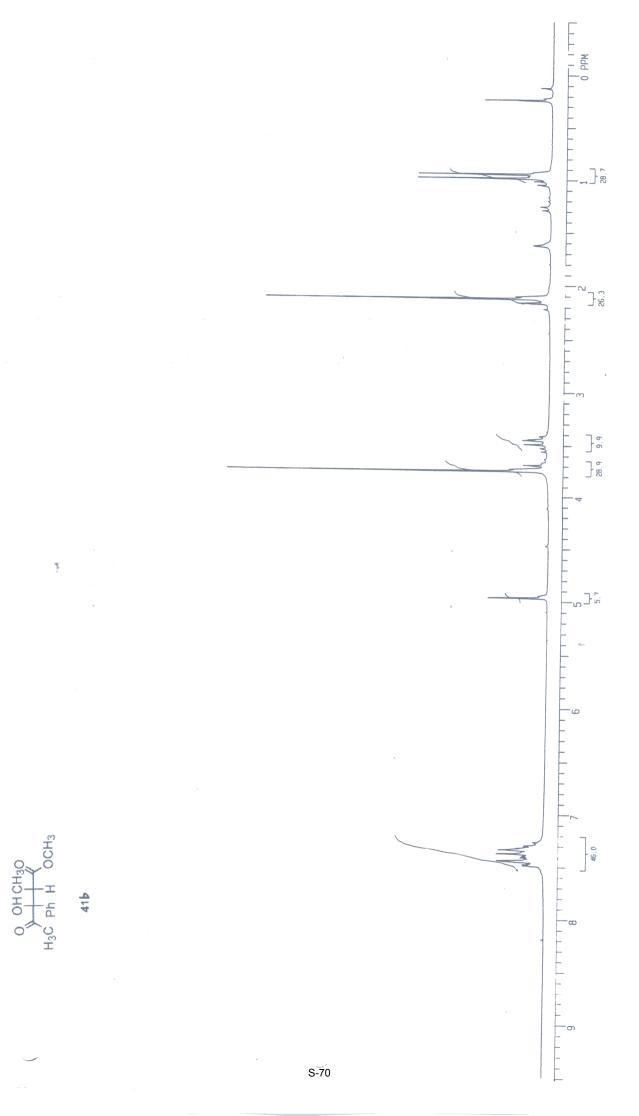


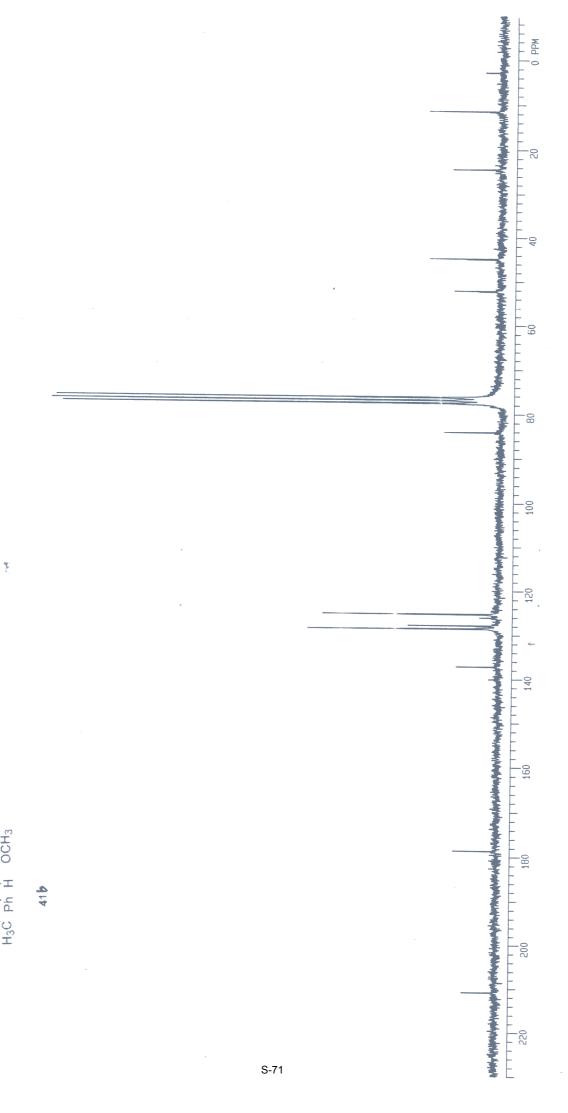


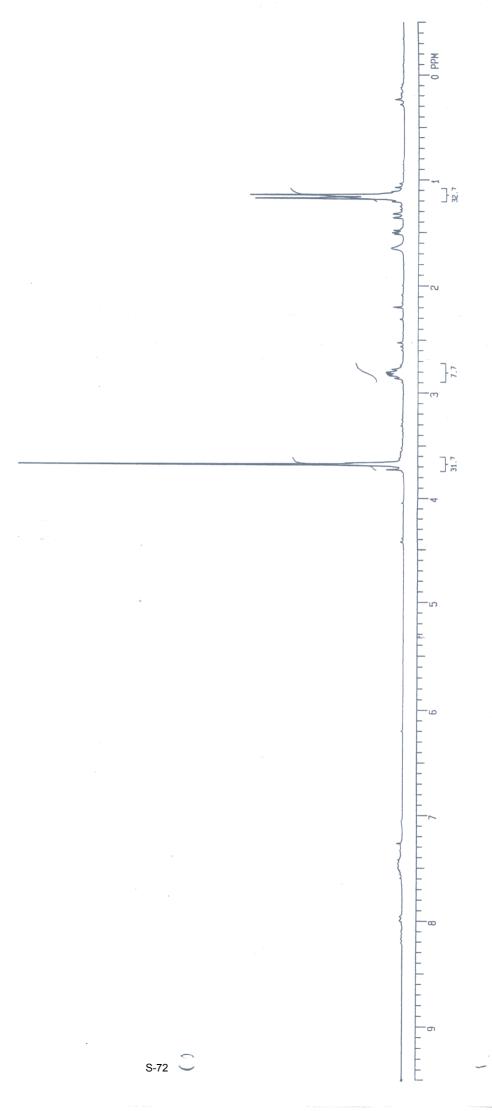


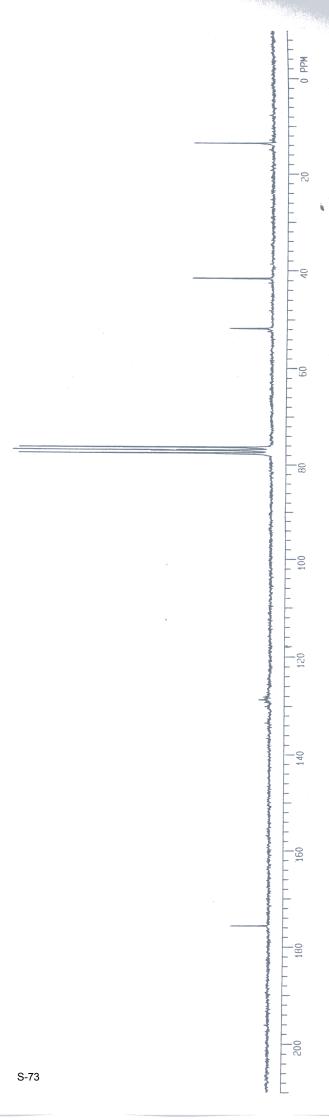




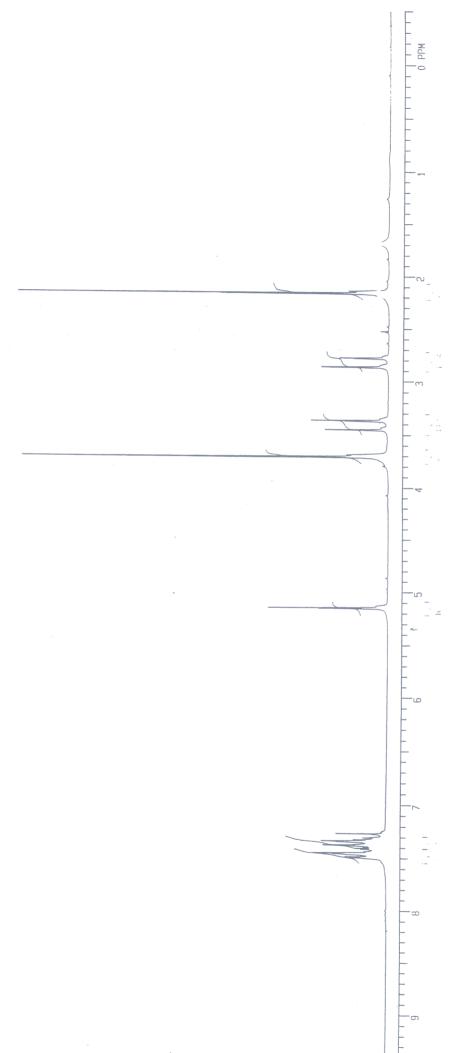




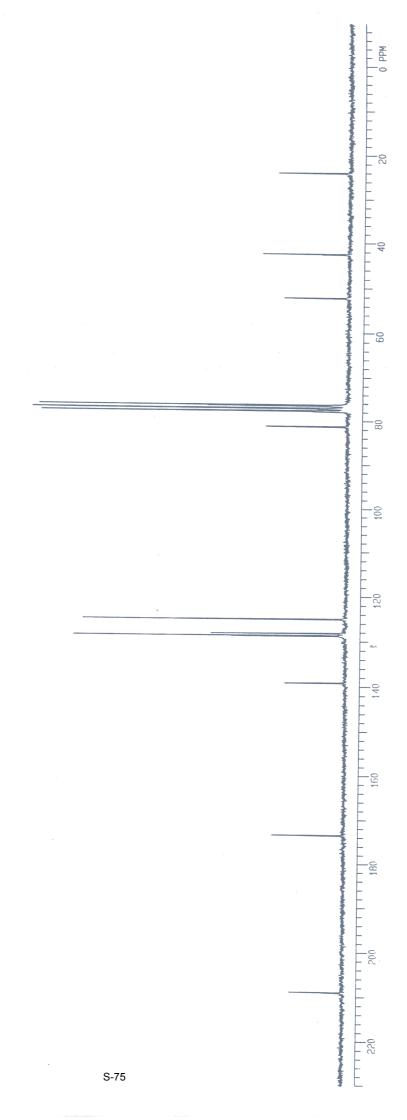


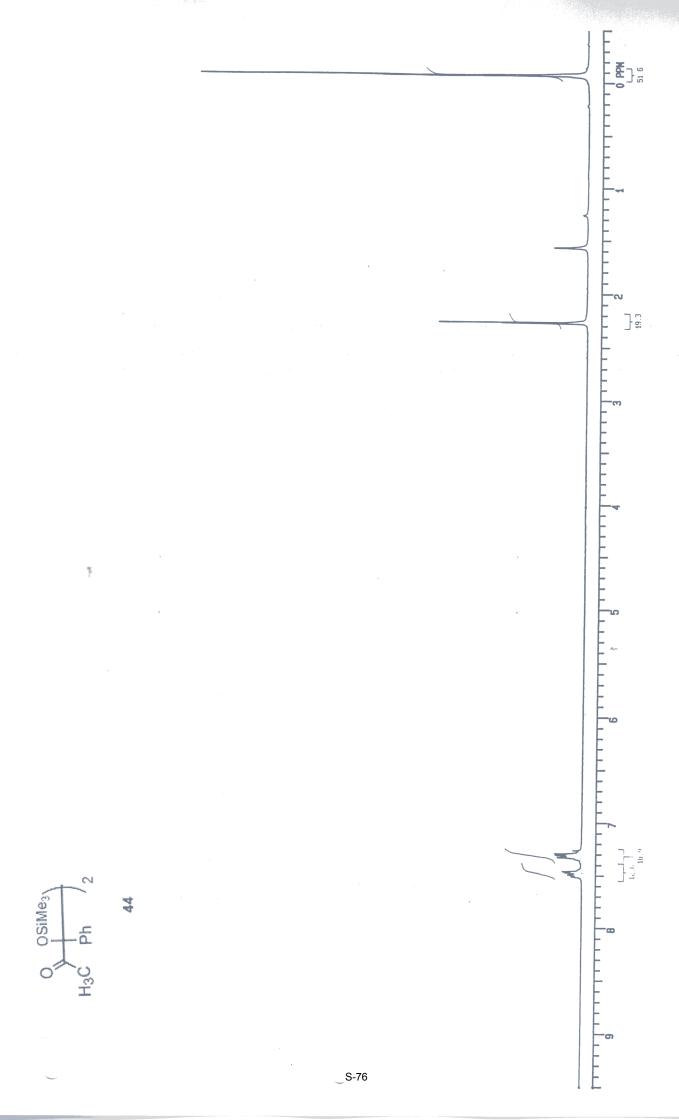


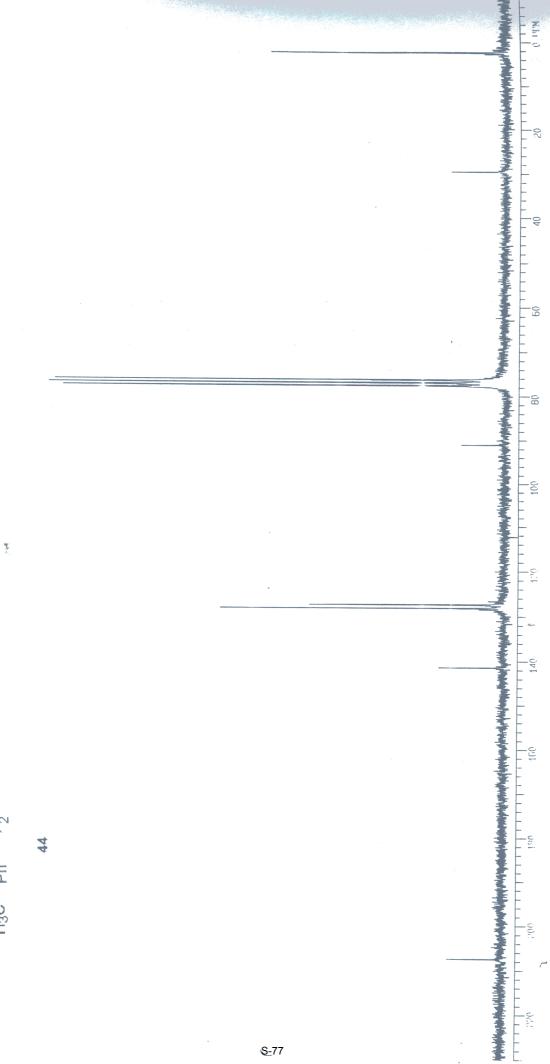
H₃CO H 2



H₃C Ph H OCH







ŲSiMe_{3√}