Supporting Information for:

## Peptide-Mediated Reduction of Silver Ions on Engineered Biological Scaffolds

Ki Tae Nam<sup>1\*</sup>, Yun Jung Lee<sup>1\*</sup>, Eric M. Krauland<sup>2</sup>, Stephen T. Kottmann<sup>3</sup>, Angela M.

Belcher<sup>1,2 †</sup>

<sup>1</sup>Department of Materials Science & Engineering, <sup>2</sup>Departmet of Biological Engineering <sup>3</sup>Department of Chemistry, Massachusetts Institute of Technology, Cambridge, Massachusetts 02139, USA

\* These authors contributed equally to this work

<sup>†</sup>e-mail: belcher@mit.edu



Figure S1. UV-Vis absorption spectra of the  $E_6$  yeast with and without UV-light blocking polymer coatings under white fluorescent light in dark room. Yeast solutions are incubated in 1 mM AgOOCCH<sub>3</sub> solution.



Figure S2. Silver nanoparticles synthesized by un-induced yeast solution. Particles are aggregated and not bound to yeast surface peptides.



Figure S3. Average conformations of (a) hexa-aspartic  $acids(D_6)$  and (b) hexa-glutamic  $acids(E_6)$  from Monte Carlo simulations. Atom colors are: grey – carbon, white – hydrogen, red – oxygen, blue – nitrogen. Blue spheres indicate counter ions (Z = +1) placed in simulation to balance charge during the simulation.



Figure S4. UV-Vis absorption analysis of the silver reduction by soluble hexa-glutamic acid peptide. (a) Various peptide concentrations. Concentration of AgOOCCH<sub>3</sub> is 8 mM. (b) Various AgOOCCH<sub>3</sub> solution concentrations. Concentration of hexa-glutamic acid solution is 1 mM.



Figure S5. Silver nanoparticles produced by wild-type M13 viruses. Large and irregular silver particles are generated.