## **Supporting Information**

C. S. Zalazar, M. L. Satuf, O. M. Alfano and A. E. Cassano

## COMPARISON OF H<sub>2</sub>O<sub>2</sub>/UV AND TiO<sub>2</sub>/UV PROCESSES FOR THE DEGRADATION OF DICHLOROACETIC ACID IN WATER

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The RTE valid for heterogeneous and homogeneous systems is

$$\frac{dI_{\lambda,\Omega}(\mathbf{x},t)}{ds} = \underbrace{-\kappa_{\lambda}(\mathbf{x},t)I_{\lambda,\Omega}(\mathbf{x},t)}_{ABSORPTION} - \underbrace{\sigma_{\lambda}(\mathbf{x},t)I_{\lambda,\Omega}(\mathbf{x},t)}_{OUT-SCATTERING} + \underbrace{j_{\lambda}^{e}(\mathbf{x},t)}_{EMISSION} + \underbrace{\frac{j_{\lambda}^{e}(\mathbf{x},t)}{4\pi}\int_{\Omega^{e}}I_{\lambda,\Omega^{e}}(\mathbf{x},t)p_{\lambda}(\Omega' \to \Omega)d\Omega}_{IN-SCATTERING}$$
(S1)

This is a tri-dimensional, pseudo-homogeneous radiative transfer equation along the direction of propagation  $\Omega$ , having "s" as the parameter of the directional derivative, valid for monochromatic radiation, with elastic, multiple and independent scattering.  $I_{\lambda,\Omega}(\mathbf{x},t)$  is the specific spectral intensity.  $\kappa_{\lambda}(\mathbf{x},t)$  is the linear, volumetric absorption coefficient and  $\sigma_{\lambda}(\mathbf{x},t)$  is the linear, volumetric scattering coefficient.  $\Omega$  is a unit vector in the direction of radiation propagation,  $\Omega'$  represents all the directions different from  $\Omega$  that contribute to in-scattering into the  $\Omega$  direction, and  $p_{\lambda}(\Omega' \rightarrow \Omega)$  is the phase function for elastic scattering, that gives the photon distribution function of scattering in the tri-dimensional space. Since radiation propagates at the speed of the light, the transient term has been neglected. However  $I_{\lambda,\Omega}(\mathbf{x},t)$  is still a function of time because all the parameters in the RTE can be a function of time. At normal "ambient" temperatures, internal emission is negligible  $[j_{\lambda}^{e}(\mathbf{x},t) \cong 0]$ . Equation S1 indicates that the specific spectral intensity is a function of six variables: three spatial variables  $(x_1, x_2, x_3)$ , two angular variables for the direction of radiation propagation  $(\theta, \phi)$ , and the wavelength distribution of the radiation under analysis  $(\lambda)$ .

The solution of eq S1 renders the values of specific intensity for every spatial point **x** and every direction  $\Omega$  in the reactor space. Then, from  $I_{\lambda,\Omega}(\mathbf{x},t)$ , we can calculate the spectral incident radiation  $G_{\lambda}$ , defined as

$$G_{\lambda}(\mathbf{x},t) = \int_{\Omega} I_{\lambda,\Omega}(\mathbf{x},t) d\Omega$$
(S2)

Finally, the LVRPA, required to compute the quantum efficiencies, is given by

$$\mathbf{e}_{\lambda}^{a}(\mathbf{x},t) = \kappa_{\lambda}(\mathbf{x},t) \mathbf{G}_{\lambda}(\mathbf{x},t)$$
(S3)

When polichromatic radiation is employed, the LVRPA averaged over the wavelength range can be calculated with the following equation:

$$\mathbf{e}_{\Sigma\lambda}^{a}\left(\mathbf{x},t\right) = \int_{\lambda} \kappa_{\lambda}\left(\mathbf{x},t\right) \mathbf{G}_{\lambda}\left(\mathbf{x},t\right) d\lambda \tag{S4}$$