

SUPPORTING INFORMATION

Manuscript Title

A MODEL FOR THE PRESENCE OF POLYCHLORINATED BIPHENYLS (PCBs) IN THE WILLAMETTE RIVER BASIN (OREGON)

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TABLE S1. Key physicochemical properties of selected congeners.

VARIABLE	SYMBOL	CONGENER [CASRN]			Ref
		PCB-077 [32598-13-3]	PCB-118 [31508-00-6]	PCB-169 [32774-16-6]	
Chemical molecular weight (g mol ⁻¹)	MW	291.99	326.43	360.88	(1)
Air-side evaporation MTC parameter a (unitless) ^(a)	VEWa	0.4032	0.3942	0.3860	(2)
Air-side evaporation MTC parameter b (unitless) ^(a)	VEWb	0.0311	0.0311	0.0311	(2)
Water-side evaporation MTC parameter a (unitless) ^(a)	VEAa	416.43	406.00	395.40	(2)
Water-side evaporation MTC parameter b (unitless) ^(a)	VEAb	0.0035	0.0036	0.0038	(2)
Water-to-sediment mass transfer coefficient (m d ⁻¹)	VD	0.0012 ^(b)	0.0024	0.0048 ^(c)	(3)
Octanol-water partition coefficient (Log ₁₀ , unitless)	KOW	6.36	6.69	7.42	(4,5,6)
Half-life, water (d)	HLF _{water}	417 ^(d)	1292	2292 ^(c)	(37)
Half-life, sediment (d)	HLF _{sed}	2292 ^(d)	2292	7083 ^(c)	(37)
Half-life, soil (d)	HLF _{soil}	708 ^(d)	4167	22917 ^(c)	(37)
Half-life, leaf (d) ^(e)	HLF _{leaf}	708 ^(d)	4167	22917 ^(c)	(37)
Half-life, particles-on-leaf (d) ^(e)	HLF _{POL}	708 ^(d)	4167	22917 ^(c)	(37)
Particle scavenging ratio (unitless)	W _P	360000 ^(f)	330000	240000 ^(h)	(7)
Vapor scavenging ratio (unitless)	W _G	12000 ^(f)	25000 ^(g)	53000 ^(h)	(7)
Reaction rate constant (cm ³ molc ⁻¹ s ⁻¹)	kA _{ref}	5.90×10 ⁻¹³ ^(d)	3.00×10 ⁻¹³	1.60×10 ⁻¹³ ^(c)	(37)
Activation energy in air (J mol ⁻¹)	Ea _{air}	10460	12920	15380 ^(c)	(8)
Activation energy in other media (J mol ⁻¹)	Ea _{other}	30000	30000	30000	(37)
Henry's Law temperature coefficient (kJ mol ⁻¹)	ΔH _H	39.229	49.237	163.616	(1)
Henry's Law temperature coefficient (kJ mol ⁻¹ K ⁻¹)	ΔS _H	0.09	0.13	0.51	(1)
Henry's Law constant (Pa m ⁻³ mol ⁻¹) @ 25° C	H	16.7	36.3	23.4	(1)

^(a) Assumes a wind speed of 4.5 m s⁻¹

^(b) Value for PCB-066

^(c) Value for PCB-153

^(d) Value for PCB-052

^(e) Assumed same as soil per Wania (34)

^(f) Value for PCB-074

^(g) Value for PCB-132

^(h) Value for PCB-174

⁽ⁱ⁾ Value for PCB-101

Table S2. Equations used for calculation of rate constants, fluxes, and compartment masses.

Equation	Equation Description (Units)	Ref
RATE CONSTANT CALCULATIONS		
{01} Sediment Burial		
$RATE_{01} = SA_{sed} \cdot VB \cdot (1 - FDS) / VS$	Burial rate constant (d ⁻¹)	(3)
$FDS = 1 / (1 + (CSS \cdot OC_{sed} \cdot KOW \cdot \psi))$	Fraction of chemical freely dissolved in sediment (unitless)	(3) ^(a)
{02} Diffusion from Sediment to Water		
$RATE_{02} = SA_{sed} \cdot VD \cdot FDS / VS$	Sediment-water diffusion rate constant (d ⁻¹)	(3)
{03} Re-Suspension from Sediment to Water		
$RATE_{03} = (RFlux \cdot CSS) \cdot (1 - FDS) / (1000 \cdot VS)$	Re-suspension rate constant (d ⁻¹)	(3)
$BFlux = 1000 \cdot CSS \cdot SA_{sed} \cdot VB$	Burial flux (kg d ⁻¹)	(3)
$RFlux = Sflux - BFlux$	Re-suspension flux (kg d ⁻¹)	(3)
$SFlux = 1000 \cdot CPW \cdot SA_{water} \cdot VSS$	Settling flux (kg d ⁻¹)	(3)
{04} Degradation in Sediment		
$RATE_{04} = (0.693 / HLF_{sed}) \cdot \exp\left[\left(\frac{Ea_{other}}{8.314}\right) \cdot \left(\frac{1}{T_{ref}} - \frac{1}{TA}\right)\right]$	Temperature-corrected sediment degradation rate constant (d ⁻¹)	(37)
{05} Diffusion from Water to Sediment		
$RATE_{05} = SA_{sed} \cdot VD \cdot FDW / VW$	Water-sediment diffusion rate constant (d ⁻¹)	(3)
{06} Deposition from Water to Sediment		
$RATE_{06} = (VSS \cdot SA_{water}) \cdot (1 - FDW) / VW$	Water-sediment deposition rate constant (d ⁻¹)	(3)
$FDW = 1 / (1 + (CPW \cdot OC_{sps} \cdot KOW \cdot \psi))$	Fraction of chemical freely dissolved in water (unitless)	(3) ^(a)
$CPW = TSS / 1 \times 10^6$	Particle concentration in water (kg L ⁻¹)	---

Table S2. Equations used for calculation of rate constants, fluxes, and compartment masses.

Equation	Equation Description (Units)	Ref
$TSS = 7.2693 \cdot \exp(2 \times 10^{-5} \cdot Q)$	Total suspended solids concentration (mg L ⁻¹)	---
{07} River Flow		
$RATE_{07} = VW / (Q \cdot 2446.5)$	Water residence time (d)	---
$VW = LEN \cdot XSA$	Water volume (m ³)	---
$XSA = (A1 \cdot Q^{A2} + A0) \cdot 0.0929$	Cross-section area (m ²)	(11)
$SA_{water} = LEN \cdot WID$	Water surface area (m ²)	---
$WID = (W1 \cdot Q^{W2}) \cdot 0.3048$	River width (m)	(11)
$VS = AD \cdot LEN \cdot WID$	Sediment volume (m ³)	---
$SA_{sed} = SA_{water}$	Sediment surface area (m ²)	---
{08} Volatilization from Water to Air		
$RATE_{08} = SA_{water} \cdot FDW \cdot VE / VW$	Water-air volatilization rate constant (d ⁻¹)	(3)
$VE = 1 / (1 / VEW + 1 / (HTD_{water} \cdot VEA))$	Volatilization mass transfer coefficient (m d ⁻¹)	(3)
$VEW = VEWa \cdot \exp(VEWb \cdot TW)$	Water-side volatilization MTC (m d ⁻¹)	(3)
$VEA = VEa \cdot \exp(VEAb \cdot TW)$	Air-side evaporation MTV (m d ⁻¹)	(3)
$HTD_{water} = \exp(-\Delta H_H / (8.314 \cdot TW) + (\Delta S_H / 8.314))$	Temperature-dependent dimensionless Henry's law constant, water (unitless)	(1)
{09} Degradation in Water		
$RATE_{09} = (0.693 / HLF_{water}) \cdot \exp\left[\left(\frac{Ea_{other}}{8.314}\right) \cdot \left(\frac{1}{T_{ref}} - \frac{1}{TA}\right)\right]$	Temperature-corrected water degradation rate constant (d ⁻¹)	(37)

Table S2. Equations used for calculation of rate constants, fluxes, and compartment masses.

Equation	Equation Description (Units)	Ref
{10} Advective Inflows & Outflows		
$FLUX10_{in} = ((CTA \cdot u(d) \cdot 86400)/1 \times 10^{18}) \cdot LEN \cdot AirHeight$	Advective inflow (kg d ⁻¹)	(27)
$FLUX10_{out} = AIR/(VA/(u(d) \cdot LEN \cdot AirHeight \cdot 86400))$	Advective outflow (kg d ⁻¹)	(1)
{11} Diffusion from Air to Water		
$RATE_{11} = SA_{water}/VA \cdot (VE/HTD_{water}) \cdot (1 - \phi)$	Air-water diffusion rate constant (d ⁻¹)	(27)
$\phi = K_p \cdot TSP/(1 - K_p \cdot TSP)$	Fraction of chemical bound to dust particles, (unitless)	(13)
$K_p = (KOW/HTD_{air}) \cdot 0.25 \times 10^{-12}$ {Aerosol Organic Fraction = 20%}	Particle-gas partition coefficient (m ³ μg ⁻¹)	(16)
$HTD_{air} = \exp(-\Delta H_H/(8.315 \cdot TA) + (\Delta S_H/8.314))$	Temperature-dependent dimensionless Henry's law constant, air (unitless)	(1)
{12, 13, 14} Deposition from Air to Water		
$RATE_{12} = ((CTA/1 \times 10^{15}) \cdot \phi \cdot SA_{water} \cdot UD)/1000$	Dry particulate deposition rate constant (d ⁻¹)	(12)
$RATE_{13} = (UR \cdot 0.0254 \cdot SA_{water} \cdot W_P \cdot \phi \cdot (CTA/1 \times 10^{15}))/1000$	Wet particulate deposition rate constant (d ⁻¹)	(12)
$RATE_{14} = (UR \cdot 0.0254 \cdot SA_{water} \cdot W_G \cdot (1 - \phi) \cdot (CTA/1 \times 10^{15}))/1000$	Wet gaseous deposition rate constant (d ⁻¹)	(12)
{15} Diffusion from Air to Soil		
$RATE_{15} = ks_{diff} \cdot (CTA \cdot 1 \times 10^{-18})$	Air-soil diffusion rate constant (d ⁻¹)	---
$ks_{diff} = ((D_a \cdot SA_{land} \cdot \theta_v)/Z_d) \cdot 86400 \cdot 1 \times 10^{-4}$	Diffusive exchange volume (m ³ d ⁻¹)	(14)
$D_a = 1.9/(MW)^{2/3}$	Diffusivity of chemical in air (cm ² s ⁻¹)	(15)
{16, 17, 18} Deposition from Air to Soil		
$RATE_{16} = (CTA/1 \times 10^{15}) \cdot \phi \cdot (SA_{nonforest} + SA_{forest} \cdot (1 - I_{dry})) \cdot UD/1000$	Dry particulate deposition rate constant (d ⁻¹)	(27)
$I_{dry} = 1 - \exp[(1 - fW_{leaf}) \cdot (-\alpha VAF \cdot \rho_{area})]$	Fraction of dry-depositing chemical intercepted by	(27)

Table S2. Equations used for calculation of rate constants, fluxes, and compartment masses.

Equation	Equation Description (Units)	Ref
	canopy (unitless)	
$RATE_{17} = (UR \cdot 0.0254 \cdot SA_{\text{nonforest}} + SA_{\text{forest}} \cdot (1 - I_{\text{wet}}) \cdot W_P \cdot \phi \cdot (CTA/1 \times 10^{15})) / 1000$	Wet particulate deposition rate constant (d^{-1})	(27)
$RATE_{18} = (UR \cdot 0.0254 \cdot SA_{\text{nonforest}} + SA_{\text{forest}} \cdot (1 - I_{\text{wet}}) \cdot W_G \cdot (1 - \phi) \cdot (CTA/1 \times 10^{15})) / 1000$	Wet gaseous deposition rate constant (d^{-1})	(27)
{19, 20} Deposition from Air to POL		
$RATE_{19} = ((V_{\text{dry}} \cdot I_{\text{dry}} \cdot SA_{\text{forest}}) / VA) \cdot (\phi / (DL / \rho P))$	Dry particulate deposition rate constant (d^{-1})	(27)
$VA = SA_{\text{total}} \cdot \text{AirHeight}$	Basin air volume (m^3)	---
$V_{\text{dry}} = UD \cdot (DL / \rho P)$	Volumetric dry deposition of particles ($m^3 m^{-2}$)	(27)
$DL = TSP / 1 \times 10^9$	Dust load in air compartment ($kg m^{-3}$)	(27)
$RATE_{20} = ((V_{\text{wet}} \cdot I_{\text{wet}} \cdot SA_{\text{forest}}) / VA) \cdot (\phi / (DL / \rho P))$	Wet particulate deposition rate constant (d^{-1})	(27)
$V_{\text{wet}} = W_P \cdot UR \cdot 0.0254 \cdot (DL / \rho P)$	Volumetric wet deposition of particles ($m^3 m^{-2}$)	(27)
{21} Diffusion from Air to Leaf		
$RATE_{21} = ((2 \cdot LAI \cdot SA_{\text{forest}} \cdot gC + LAI \cdot SA_{\text{forest}} \cdot gS) \cdot 1 / VA)$	Air-leaf diffusion rate constant (d^{-1})	(27)
$gB = (D_{\text{air}} / 1000) / LAP$	Conductance of the air boundary layer ($m d^{-1}$)	(27)
$gC = ((1/gB) + (1/g_{\text{cuticle}}))^{-1}$	Total conductance of the cuticular path ($m d^{-1}$)	(27)
$g_{\text{cuticle}} = (P_{\text{cuticle}} / (Z_{\text{air}} / Z_{\text{water}})) \cdot 86400$	Conductance of the cuticle ($m d^{-1}$)	(27)
$gS = ((1/g_{\text{stomata}}) + (1/gB))^{-1}$	Total conductance of the stomatal pathway ($m d^{-1}$)	(27)
$g_{\text{stomata}} = (D_{\text{air}} / 10000) \cdot Sn$	Conductance of the cuticle ($m s^{-1}$)	(27)
$P_{\text{cuticle}} = 10^{(0.704 \cdot \log KOW - 11.2)}$	Permeance of the cuticle ($m s^{-1}$)	(27)
$Z_{\text{water}} = 1/H$	Fugacity capacity in air vapor ($mol Pa^{-1} m^{-3}$)	(27)

Table S2. Equations used for calculation of rate constants, fluxes, and compartment masses.

Equation	Equation Description (Units)	Ref
$H = \text{HTD}_{\text{air}} \cdot 8.314 \cdot (273.15 + \text{TA})$	Dimensionless Henry's Law constant (unitless)	---
$Z_{\text{air}} = 1/(8.314 \cdot (273.15 + \text{TA}))$	Fugacity capacity in total water ($\text{mol Pa}^{-1} \text{m}^{-3}$)	(27)
{22} Deposition from Air to Leaf		
$\text{RATE}_{22} = (\text{SA}_{\text{forest}}/\text{VA}) \cdot \text{UR} \cdot 0.0254 \cdot \text{I}_{\text{wet}} \cdot (Z_{\text{air}}/Z_{\text{TotalAir}})$	Wet gaseous deposition rate constant (d^{-1})	(27)
{23} Degradation in Air		
$\text{RATE}_{23} = 0.693/(1/k_A \cdot 1/86400)$	Temperature-corrected atmosphere degradation rate constant (d^{-1})	---
$k_A = k_{A_{\text{ref}}} \cdot \text{OH} \cdot \exp\left[\left(\frac{E_{a_{\text{air}}}}{8.314}\right) \cdot \left(\frac{1}{T_{\text{ref}}} - \frac{1}{\text{TA}}\right)\right]$	Reaction rate constant, air (s^{-1})	(37)
{24} Volatilization from Soil to Air		
$\text{RATE}_{24} = \left[\frac{86400}{Z_d \cdot \text{KOC}_{\text{soil}} \cdot \text{BD}_{\text{soil}}}\right] \cdot \text{HTD}_{\text{soil}} \cdot \left(\frac{D_a}{Z_d}\right) \cdot \left[1 - \left(\frac{\text{BD}_{\text{soil}}}{\rho_{\text{soil}}}\right) - \theta_W\right]$	Soil-air volatilization rate constant (d^{-1})	(15)
$\text{HTD}_{\text{soil}} = \exp(-\Delta H_H/(8.315 \cdot \text{TS}) + (\Delta S_H/8.314))$	Temperature-dependent dimensionless Henry's Law constant, soil (unitless)	(1)
$\text{KOC}_{\text{soil}} = \text{KOW} \cdot \text{OC}_{\text{soil}} \cdot \psi$	Organic carbon-water partition coefficient ($\text{cm}^3 \text{g}^{-1}$)	(16)
{25} Erosion from Soil to Water		
$\text{RATE}_{25} = ((\text{XE} \cdot \text{SD} \cdot \text{ER}) \cdot \text{FIP}_{\text{soil}} \cdot 1 \times 10^{-7})/(\text{BD} \cdot Z_d)$	Soil-water erosion rate constant (d^{-1})	(15)
$\text{SD}_k = 0.6 \cdot \text{SA}_{\text{land}}^{-0.125}$	Sediment delivery ratio (unitless)	(15)
$\text{FIP}_{\text{soil}} = (\text{KOC}_{\text{soil}} \cdot \text{BD}_{\text{soil}})/(\text{HTD}_{\text{soil}} - \theta_V + \theta_W + \text{KOC}_{\text{soil}} \cdot \text{BD}_{\text{soil}})$	Fraction of chemical on soil solids (unitless)	(14)
{26} Runoff from Soil to Water		

Table S2. Equations used for calculation of rate constants, fluxes, and compartment masses.

Equation	Equation Description (Units)	Ref
$RATE_{26} = (UR \cdot ROfrac \cdot 2.54) \cdot FIW_{soil} / (\theta_W \cdot Z_d)$	Soil-water dissolved runoff rate constant (d ⁻¹)	(15)
$FIW_{soil} = \theta_W / (HTD_{soil} \cdot \theta_V) + \theta_W + KOC_{soil} \cdot BD_{soil}$	Fraction in soil interstitial water (unitless)	(14)
{27} Degradation in Soil		
$RATE_{27} = (0.693/HLF_{soil}) \cdot \exp\left[\left(\frac{Ea_{other}}{8.314}\right) \cdot \left(\frac{1}{T_{ref}} - \frac{1}{TA}\right)\right]$	Temperature-corrected soil degradation rate constant (d ⁻¹)	(28)
{28} Wash-Off from POL to Soil		
$RATE_{28} = (V_{wet} \cdot I_{wet} \cdot SA_{forest}) / V_{leaf} P$	Wash-off rate constant (d ⁻¹)	(27)
$V_{leaf} P = V_{leaf} \cdot 1 \times 10^9$	POL volume (m ³)	(27)
{29} Litter-Fall from POL to Soil		
$RATE_{29} = K_{litter}$	POL loss via litterfall rate constant (d ⁻¹)	(27)
{30} Diffusion from POL to Leaf		
$RATE_{30} = K_{POL-leaf}$	POL-leaf diffusion rate constant (d ⁻¹)	(27)
{31} Blow-Off from POL to Air		
$RATE_{31} = ((V_{dry} \cdot I_{dryt} \cdot SA_{forest}) / V_{leaf} P) \cdot RainEvent$	Blow-off rate constant (d ⁻¹)	(27)
$RainEvent \langle \text{if } UR > 0 \text{ then } 1 \text{ else } 0 \rangle$		(27)
{32} Degradation in POL		
$RATE_{32} = (0.693/HLF_{POL}) \cdot \exp\left[\left(\frac{Ea_{other}}{8.314}\right) \cdot \left(\frac{1}{T_{ref}} - \frac{1}{TA}\right)\right]$	Temperature-corrected POL degradation rate constant (d ⁻¹)	(27, 37)
{33} Litter-Fall from Leaf to Soil		

Table S2. Equations used for calculation of rate constants, fluxes, and compartment masses.

Equation	Equation Description (Units)	Ref
$RATE_{33} = K_{litter}$	Leaf litterfall rate constant (d^{-1})	(27)
{34} Diffusion from Leaf to POL		
$RATE_{34} = 0.01 \cdot K_{POL-leaf}$	Leaf-POL diffusion rate constant (d^{-1})	(27)
{35} Diffusion from Leaf to Air		
$RATE_{35} = (2 \cdot LAI \cdot SA_{forest} \cdot gC + LAI \cdot SA_{forest} \cdot gS) \cdot 1/V_{leaf} \cdot Z_{air}/Z_{leaf}$	Leaf-air diffusion rate constant (d^{-1})	(27)
$V_{leaf} = (SA_{forest} \cdot LAI \cdot \rho_{area})/820$	Volume of leaves (m^3)	(27)
$Z_{leaf} = 0.18 \cdot Z_{air} + Z_{water} + 0.02 \cdot Z_{water} \cdot KOW$	Fugacity capacity in total leaf ($mol\ pa^{-1}\ m^{-3}$)	(27)
$Z_{TotalAir} = Z_{air} \cdot (1 - DL/\rho P) + Z_{solid} \cdot (DL/\rho P)$	Fugacity capacity in total air ($mol\ pa^{-1}\ m^{-3}$)	(27)
$Z_{solid} = Z_{air} \cdot (\phi \cdot (1 - DL/\rho P)) / ((1 - \phi) \cdot (DL/\rho P))$	Fugacity capacity in particulates ($mol\ pa^{-1}\ m^{-3}$)	(27)
{36} Degradation in Leaf		
$RATE_{36} = (0.693/HLF_{leaf}) \cdot \exp\left[\left(\frac{Ea_{other}}{8.314}\right) \cdot \left(\frac{1}{T_{ref}} - \frac{1}{TA}\right)\right]$	Temperature-corrected leaf degradation rate constant (d^{-1})	(27, 37)
COMPARTMENT FLUX AND MASS CALCULATIONS		
Air Compartment		
$AIR = AIR_{t-dt} + \left(\begin{array}{l} FLUX_{08} + FLUX_{24} + FLUX_{35} + FLUX_{31} + FLUX_{10} - FLUX_{23} \\ - (FLUX_{11-14} + FLUX_{15-18} + FLUX_{19-20} + FLUX_{21-22}) \end{array} \right) \cdot dt$	Mass in air compartment (kg)	---
$FLUX_{08} = WATER \cdot RATE_{08}$	Water-air volatilization flux ($kg\ d^{-1}$)	---
$FLUX_{11-14} = AIR \cdot (RATE_{11} + RATE_{12} + RATE_{13} + RATE_{14})$	Air-water diffusion and deposition flux ($kg\ d^{-1}$)	---
$FLUX_{15-18} = AIR \cdot (RATE_{16} + RATE_{17} + RATE_{18}) + RATE_{15}$	Air-soil diffusion and deposition flux ($kg\ d^{-1}$)	---
$FLUX_{19-20} = AIR \cdot (RATE_{19} + RATE_{20})$	Air-POL deposition flux ($kg\ d^{-1}$)	---

Table S2. Equations used for calculation of rate constants, fluxes, and compartment masses.

Equation	Equation Description (Units)	Ref
$FLUX_{21-22} = AIR \cdot (RATE_{21} + RATE_{22})$	Air-leaf diffusion and deposition flux (kg d ⁻¹)	---
$FLUX_{23} = AIR \cdot RATE_{23}$	Air degradation flux (kg d ⁻¹)	---
$FLUX_{24} = SOIL \cdot RATE_{24}$	Soil-air volatilization flux (kg d ⁻¹)	---
$FLUX_{31} = POL \cdot RATE_{31}$	POL-air blow-off flux (kg d ⁻¹)	---
$FLUX_{35} = LEAF \cdot RATE_{35}$	Leaf-air diffusion flux (kg d ⁻¹)	---
Fish Compartment		
$FISH = FISH_{t-dt} + (FLUX_{37} - FLUX_{38}) \cdot dt$	Mass in fish compartment, single fish (kg)	---
$FLUX_{37} = (CWT \cdot BSF \cdot k1 \cdot WT_{fish} + C_{diet} \cdot kD \cdot WT_{fish}) \cdot 1 \times 10^{-12}$	Flux into fish (kg d ⁻¹)	(10)
$C_{diet} = CWT \cdot BSF \cdot KOW \cdot L_{diet} \cdot \beta_{fish}$	Concentration in fish diet (ng kg ⁻¹)	(10)
$k1 = 1 / ((0.01 + (1/KOW)) \cdot (WT_{fish}^{0.4}))$	Gill uptake rate constant (L kg ⁻¹ d ⁻¹)	(10)
$kD = (0.02 \cdot WT_{fish}^{-0.15} \cdot \exp(0.06 \cdot TW)) / (5.1 \times 10^{-8} \cdot KOW + 2)$	Dietary uptake rate constant (kg kg ⁻¹ d ⁻¹)	(10)
$BSF = 1 / (1 + (POC \cdot 1 \times 10^{-6} \cdot \psi_2 \cdot KOW) + (DOC \cdot 1 \times 10^{-6} \cdot \psi_3 \cdot KOW))$	Bioavailable fraction of chemical (unitless)	(17)
$POC = PD_{ratio} \cdot DOC$	Concentration of particulate organic carbon (mg L ⁻¹)	(10)
$FLUX_{38} = FISH \cdot (k2 + kE + kG)$	Flux out of fish (kg d ⁻¹)	(10)
$k2 = k1 / (L_{fish} \cdot KOW)$	Elimination rate constant (d ⁻¹)	(10)
$kE = 0.125 \cdot kD$	Fecal egestion rate constant (d ⁻¹)	(10)
$kG = 5.02 \times 10^{-4} \cdot WT_{fish}^{-0.2}$	Growth rate constant (d ⁻¹)	(10)

Leaf Compartment

Table S2. Equations used for calculation of rate constants, fluxes, and compartment masses.

Equation	Equation Description (Units)	Ref
$LEAF = LEAF_{t-dt} + (FLUX_{30} + FLUX_{21-22} - FLUX_{34} - FLUX_{33} - FLUX_{35} - FLUX_{36}) \cdot dt$	Mass in leaf compartment (kg)	---
$FLUX_{30} = POL \cdot RATE_{30}$	POL-leaf diffusion flux (kg d ⁻¹)	---
$FLUX_{33} = LEAF \cdot RATE_{33}$	Leaf-soil litterfall flux (kg d ⁻¹)	---
$FLUX_{34} = LEAF \cdot RATE_{34}$	Leaf-POL diffusion flux (kg d ⁻¹)	---
$FLUX_{36} = LEAF \cdot RATE_{36}$	Leaf degradation flux (kg d ⁻¹)	---
Particle-on-Leaf (POL) Compartment		
$POL = POL_{t-dt} + (FLUX_{34} + FLUX_{19-20} - FLUX_{30} - FLUX_{31} - FLUX_{28-29} - FLUX_{32}) \cdot dt$	Mass in POL compartment (kg)	---
$FLUX_{28-29} = POL \cdot (RATE_{28} + RATE_{29})$	Leaf-soil wash-off and litterfall flux (kg d ⁻¹)	---
$FLUX_{31} = POL \cdot RATE_{31}$	POL-soil blow-off flux (kg d ⁻¹)	---
$FLUX_{32} = POL \cdot RATE_{32}$	POL degradation flux (kg d ⁻¹)	---
Sediment Compartment		
$SED = SED_{t-dt} + (FLUX_{05-06} - FLUX_{01} - FLUX_{02-03} - FLUX_{04}) \cdot dt$	Mass in sediment compartment (kg)	---
$FLUX_{01} = SED \cdot RATE_{01}$	Burial flux (kg d ⁻¹)	---
$FLUX_{02-03} = SED \cdot (RATE_{02} + RATE_{03})$	Sediment-water diffusion and re-suspension flux (kg d ⁻¹)	---
$FLUX_{04} = SED \cdot RATE_{04}$	Sediment degradation flux (kg d ⁻¹)	---
$FLUX_{05-06} = WATER \cdot (RATE_{05} + RATE_{06})$	Water-sediment diffusion and deposition flux (kg d ⁻¹)	---
Soil Compartment		
$SOIL = SOIL_{t-dt} + (FLUX_{15-18} - FLUX_{33} + FLUX_{28-29} - FLUX_{25-26} - FLUX_{24} - FLUX_{27}) \cdot dt$	Mass in soil compartment (kg)	---
$FLUX_{25-26} = SOIL \cdot (RATE_{02} + RATE_{03})$	Soil-water erosion and runoff flux (kg d ⁻¹)	---

Table S2. Equations used for calculation of rate constants, fluxes, and compartment masses.

Equation	Equation Description (Units)	Ref
$FLUX_{27} = SOIL \cdot RATE_{27}$	Soil degradation flux (kg d ⁻¹)	---
Water Compartment		
$WAT = WAT_{1-dt} + \left(\begin{array}{l} FLUX_{02-03} + FLUX_{11-14} + FLUX_{25-26} + FLUX_{38} \\ - FLUX_{05-06} - FLUX_{08} - FLUX_{09} - FLUX_{07} - FLUX_{37} \end{array} \right) \cdot dt$	Mass in water compartment (kg)	---
$FLUX_{07} = (WATER/RATE_{07})$	River flow flux (kg d ⁻¹)	---
$FLUX_{09} = WATER \cdot RATE_{09}$	Water degradation flux (kg d ⁻¹)	---
MISCELLANEOUS CALCULATIONS		
$CWT = (WATER/VW) \cdot 1 \times 10^9$	Total concentration in water (ng L ⁻¹)	---
$CTA = (AIR/VA) \cdot 1 \times 10^{18}$	Total concentration in air (fg m ⁻³)	---
$CST = SED/(VS \cdot CSS \cdot 1000) \cdot 1 \times 10^9$	Total concentration in sediment (μg kg ⁻¹)	---
$CSD = (CST/OC_{sed}/(KOW \cdot \psi))$	Concentration in sediment pore water (μg L ⁻¹)	---
$C_{soil} = (SOIL/(SA_{land} \cdot BD_{soil} \cdot 1000 \cdot (Zd/100))) \cdot 1 \times 10^{12}$	Total concentration in surface soil (ng kg ⁻¹)	---
$CB_{fish} = (FISH/WT_{fish}) \cdot 1 \times 10^9$	Concentration in fish (μg kg ⁻¹)	(10)
$CTV = ((LEAF/V_{leaf})/\rho_{leaf}) \cdot 1 \times 10^{15}$	Concentration in leaves (pg kg ⁻¹)	---
$SA_{forest} = SA_{land} \cdot ForestFrac$	Surface area of forested land (m ²)	---
$SA_{land} = SA_{total} - SA_{water}$	Surface area of land (m ²)	---
$SA_{nonforest} = SA_{land} \cdot (1 - ForestFrac)$	Surface area of non-forested land (m ²)	---

^(a) Equation corrected per conversation with the author.

Table S3. Summary of model input variables.

Variable	Description (units)	Value	Note / Ref
ψ	Organic carbon-water partition coefficient (unitless)	0.41 ^(a)	(16)
ψ_2	Particulate organic carbon proportionality constant (unitless)	0.35 ^(a)	Eq. 4 (17)
ψ_3	Dissolved organic carbon proportionality constant (unitless)	0.08	Eq. 4 (17), (18)
ρ_{area}	Above-ground vegetation biomass inventory (kg m ⁻²)	2	Eq. 7-2 (27)
ρ_{leaf}	Density of leaves (kg m ⁻³)	820	(27)
β_{fish}	Overall food web biomagnification factor (unitless)	PCB-077: 2.75 PCB-118: 6.0 PCB-169: 2.75	Calibrated with the region-specific Gobas-type food web model (10)
ρ_p	Dust particle density (kg m ⁻³)	1400	Eq. 7-2 (27)
ρ_{soil}	Solids particle density (g cm ⁻³)	2.7	Table B-2-6 (15)
θ_v	Soil void fraction (unitless)	0.3	(14)
αVAF	Vegetation attenuation factor (m ² kg ⁻¹)	2.9	Eq. 7-2 (27)
θ_w	Soil volumetric water content (ml cm ⁻³)	0.2	Table B-2-6 (15)
A0	Cross section area coefficient (unitless)	0	(11)
A1	Cross section area coefficient (unitless)	5.86	(11)
A2	Cross section area coefficient (unitless)	0.69	(11)
AD	Active sediment depth (m)	0.03	Assumed maximum depth of chemically active zone
BD _{soil}	Soil bulk density (kg L ⁻¹)	1.5	Willamette soils data (22)
CSS	Concentration of solids in sediment (kg L ⁻¹)	0.4	Assumes a mean porosity of 0.83 and mean sediment density 2350 kg m ⁻³
DOC	Dissolved organic carbon concentration in water (mg L ⁻¹)	Graphical function	Time series water quality data from USGS gage 14211000 (RM 12.8)
ER	Particle enrichment ratio (unitless)	3	Table B-2-3 (15)
ForFrac	Fraction of land area covered by coniferous forest	0.7	NRCS data for the Willamette Valley (32)
fW _{leaf}	Water content of leaf (kg kg ⁻¹)	0.8	Eq. 7-2 (27)
AirHeight	Mixing height of air column (m)	5000	(27)
I _{wet}	Plant interception fraction of wet deposition (unitless)	0.2	(27)

Table S3. Summary of model input variables.

Variable	Description (units)	Value	Note / Ref
K_{litter}	Litterfall rate constant, conifers (d^{-1})	0.0021	(27)
$K_{\text{POL-leaf}}$	Transfer factor, POL to leaf compartment (d^{-1})	0.3	(27)
LAI	Leaf-area index ($\text{m}^2 \text{m}^{-2}$)	9.3	(29, 30)
LAP	Thickness of air boundary layer (m)	0.001	Eq. 7-12 (27)
L_{diet}	Lipid content of fish diet (unitless)	0.01	(10)
LEN	River (main stem) length (m)	225,000	USGS watershed data
L_{fish}	Lipid content of fish (unitless)	0.05	Mean from (31)
OC_{sed}	Organic carbon content of bottom sediment (unitless)	0.0037	Willamette River Mercury TMDL (www.deq.state.or.us/lab/wqm/wb/mercurystudy.htm, accessed June 2007)
OC_{soil}	Fraction of organic carbon in soil (unitless)	U(0.01, 0.03) ^(b)	(22)
OC_{sps}	Organic carbon content of suspended solids (unitless)	0.03	Appendix A (15)
OH	Hydroxyl (OH) radical concentration (molec cm^{-3})	Winter: 2×10^6 Spring: 0.8×10^6 Summer: 0.09×10^6 Fall: 0.8×10^6	Mean value in 2000 m surface layer at 45°N (9)
Q	Mean of daily mean flow ($\text{ft}^3 \text{s}^{-1}$)	Graphical function	Time series flow data from USGS gauge at Corvallis, Oregon
PD_{ratio}	Ratio of POC to DOC in water (unitless)	U(0.17, 0.36) ^(b)	Measured ratio of POC to DOC at Willamette RM 12.8
RO_{frac}	Fraction of precipitation becoming runoff (unitless)	0.4	(24)
SA_{total}	Land surface area (m^2)	2.90×10^{10}	USGS watershed data
Sn	Stomatal area (m^{-1})	200	Eq. 7-14 (27)
TA	Mean daily air temperature ($^{\circ}\text{C}$)	Graphical function	Time series air temperature data from Corvallis, Oregon (Oregon State University), 1953-2006
TS	Mean daily maximum soil temperature at 2" depth ($^{\circ}\text{C}$)	Graphical function	Time series soil temperature data from Hyslop Farm, Oregon, 1973-1984 (25)
TSP	Total suspended particulates ($\mu\text{g m}^{-3}$)	Graphical function	Time series data for PM2.5 from ODEQ monitors in the Willamette Basin, 2003- 2006

Table S3. Summary of model input variables.

Variable	Description (units)	Value	Note / Ref
TSS	Total suspended solids in water (mg L ⁻¹)	Function of flow	Relationship derived from USGS data collected at RM 12.8
TW	Water temperature (C°)	Graphical function	Time series water temperature data from USGS gauge at RM 84.1
UD	Dry deposition velocity (m d ⁻¹)	500	Default for dioxin in (27)
<i>u</i> (d)	Wind speed, average annual (m s ⁻¹)	3.81	Oregon Climate Service (Oregon State University): 1985 - 2005
UR	Daily mean rainfall (in)	Graphical function	Time series precipitation data from Corvallis, Oregon (Oregon State University), 1889-2006
VB	Sediment burial mass transfer coefficient (m d ⁻¹)	0.0	Assumption that burial in deep sediment unlikely in an active river
VD	Water-to-sediment mass transfer coefficient (m d ⁻¹)	0.0024	(3)
VSS	Suspended solids settling rate (m d ⁻¹)	1.1	Estimated assuming a mean sediment particle diameter of 4 μm and a mean sediment density of 2.65 g cm ⁻³
W1	River width coefficient 1 (unitless)	18.1	(11)
W2	River width coefficient 2 (unitless)	0.374	(11)
WT _{fish}	Body weight of fish (kg)	0.65	Mean from (31)
XE	Unit soil loss (kg km ⁻² d ⁻¹)	Graphical function	Value varies seasonally around a basin-wide annual average of ≈ 1480, based on observations in (32, 35, 36)
Z _d	Soil depth (cm)	2	Table B-2-3 (15)

^(a) These values should be the same (see Karichoff (39) or Seth (40)). Their disparity reflects that between the original publications cited.

^(b) U(LB, UB) = Lower (LB) and upper (UB) bounds of a uniform random variable. When data suggest a range of values, but no distributional information is provided, a uniform distribution is the most parsimonious way to represent this range.

Table S4. Summary of available Willamette Basin congener-specific PCB data, and data sources, by media type.

Congener	ENVIRONMENTAL MEDIA						
	Air ^(a)	Surface Soil ^(b)	Terrestrial Vegetation	Surface Water ^(c,g)	Bed Sediment ^(c,g)	Suspended Sediment ^(d)	Fish ^(e,f,g)
008	---	ub	---	ph	ub	lb	---
018	---	ub	---	ph ^{018/030}	ub, ph	lb	ph ^{018/030}
077	ub	ub	---	ph	ph	lb	lb, ph
101	---	ub	---	---	ub	lb	lb, ph
105	ub	ub	---	ph	ph	lb	lb, ph
110	---	ub	---	ph ^{110/115}	ub, ph	lb	ph ^{110/115}
118	ub	ub	---	ph	ph ^{106/118}	lb	---
126	ub	ub	---	ph	ph	---	ph, lb
128	---	ub	---	ph ^{128/166}	ub, ph ^{128/162}	---	lb, ph ^{128/166}
153	---	ub	---	ph ^{153/168}	ub, ph	lb	lb, ph
156	ub	ub	---	---	ph	---	lb, ph
157	ub	ub	---	---	ph	---	---
169	ub	ub	---	---	ph	---	lb, ph
ΣPCB	---	---	---	ub	---	---	lb, ph

Sampling Locations

ub Upper Basin (> RM 26.5)

lb Lower Basin (< RM 26.5, ~RM 12.8)

ph Portland Harbor Superfund site (RM 3.5-9.5)

^(a) Cleverly et al. (19); ^(b) U.S. EPA (20); ^(c) Villeneuve et al. (21); ^(d) Morace (26); ^(e) Sethajintanin et al. (33); ^(f) Henny et al. (31); ^(g) LWG (38)

Figure S1

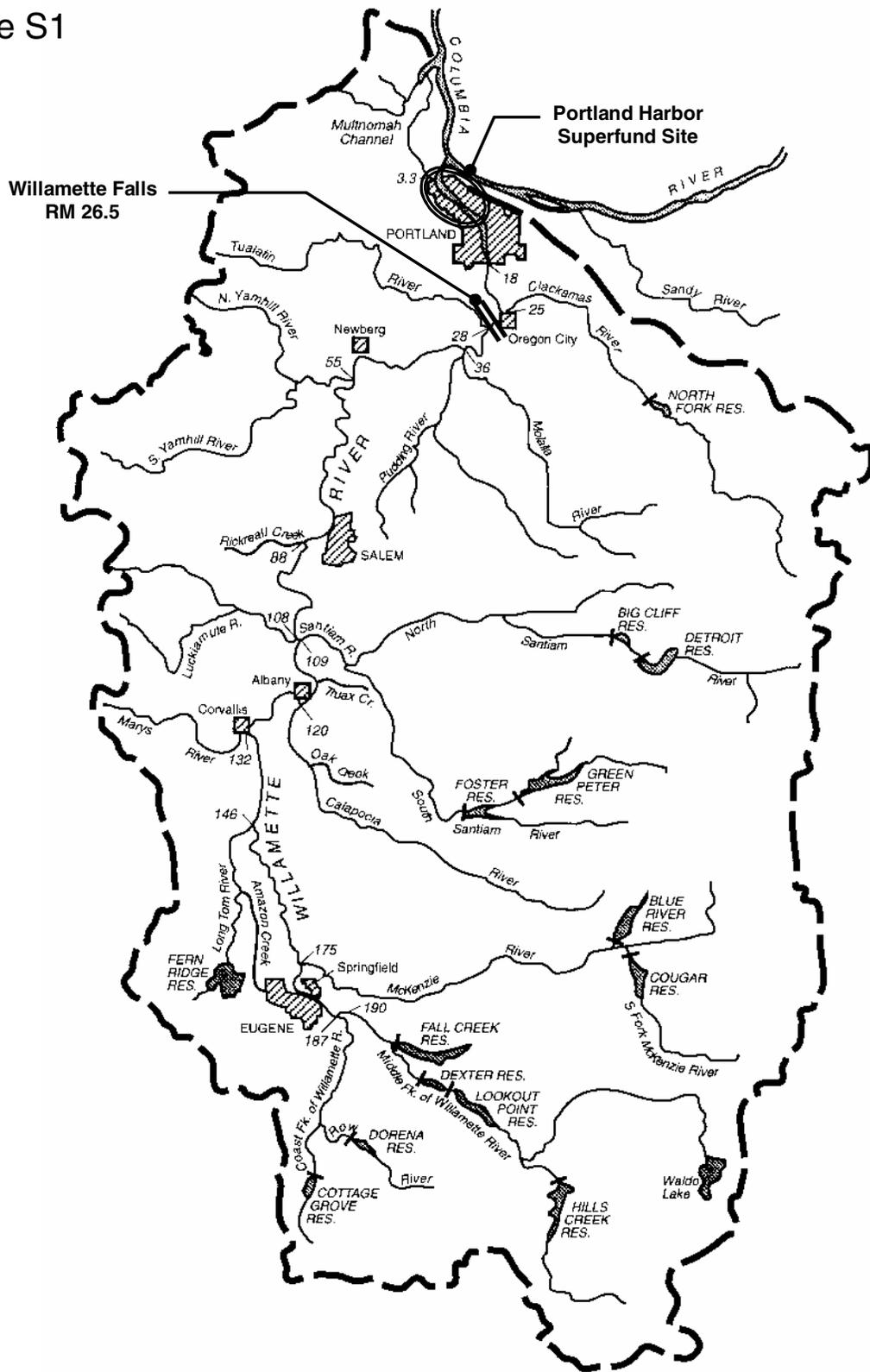


Figure S2

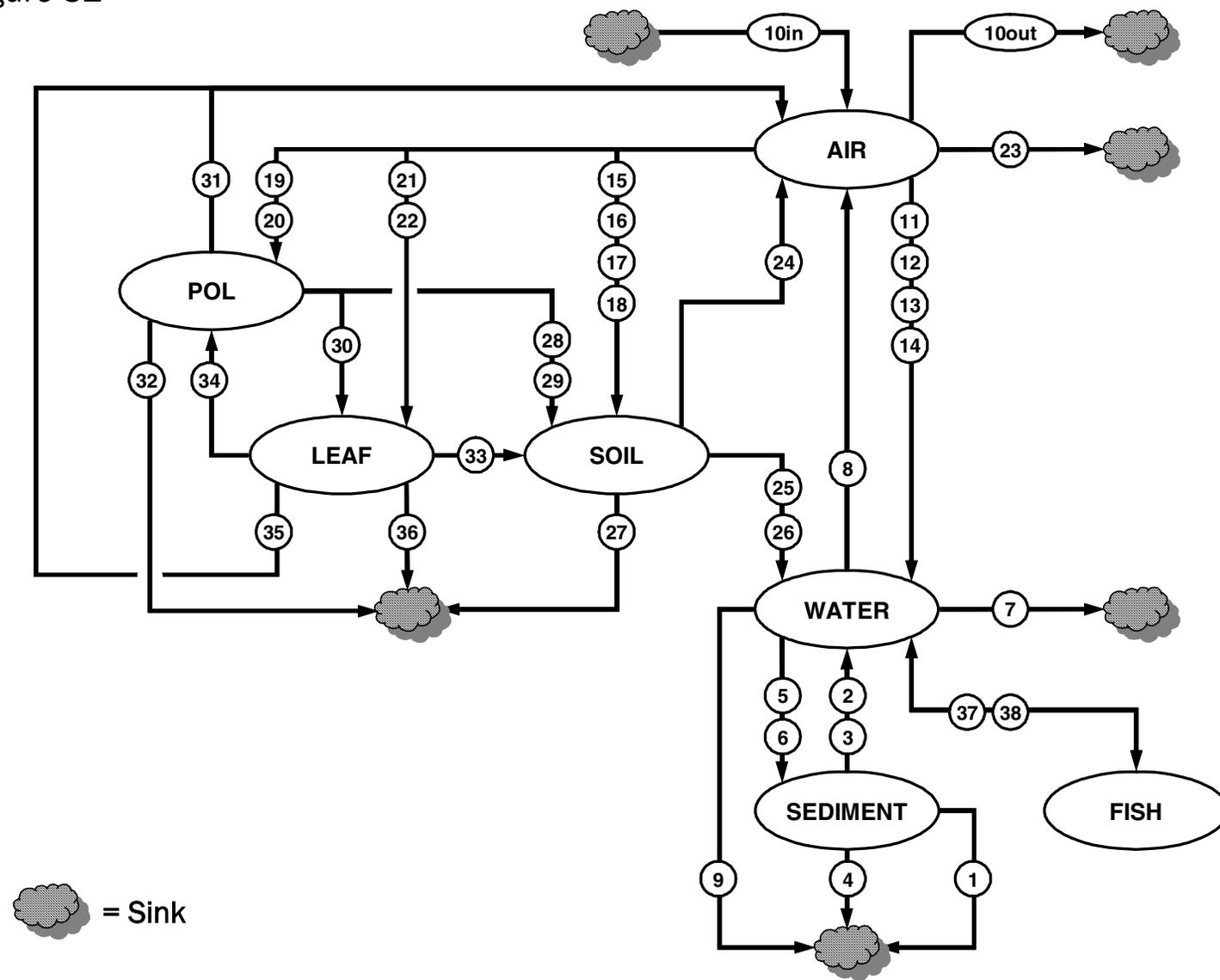


Figure S2. Schematic representation of the environmental compartments and chemical fate processes in the model and list of fluxes.

- [1] Sediment Burial
- [2] Sediment to Water via Diffusion
- [3] Sediment to Water via Resuspension
- [4] Sediment Degradation
- [5] Water to Sediment via Diffusion
- [6] Water to Sediment via Deposition
- [7] River Flow
- [8] Water to Air via Volatilization
- [9] Water Degradation
- [10in] Advective Inflow
- [10out] Advective Outflow
- [11] Air to Water via Diffusion
- [12] Air to Water via Dry Particulate Deposition
- [13] Air to Water via Wet Particulate Deposition
- [14] Air to Water via Wet Gaseous Deposition
- [15] Air to Soil via Diffusion
- [16] Air to Soil via Dry Particulate Deposition
- [17] Air to Soil via Wet Particulate Deposition
- [18] Air to Soil via Wet Gaseous Deposition
- [19] Air to Particles-on-Leaf (POL) via Dry Particulate Deposition
- [20] Air to POL via Wet Particulate Deposition
- [21] Air to Leaf via Diffusion
- [22] Air to Leaf via Wet Gaseous Deposition
- [23] Air Degradation
- [24] Soil to Air via Volatilization
- [25] Soil to Water via Particulate Erosion
- [26] Soil to Water via Dissolved Runoff
- [27] Soil Degradation
- [28] POL to Soil via Wash-off
- [29] POL to Soil via Litter-fall
- [30] POL to Leaf via Diffusion
- [31] POL to Air via Blow-off
- [32] POL Degradation
- [33] Leaf to Soil via Litter-fall
- [34] Leaf to POL via Diffusion
- [35] Leaf to Air via Diffusion
- [36] Leaf Degradation
- [37] Water to Fish via Uptake
- [38] Fish to Water via Excretion

Figure S3

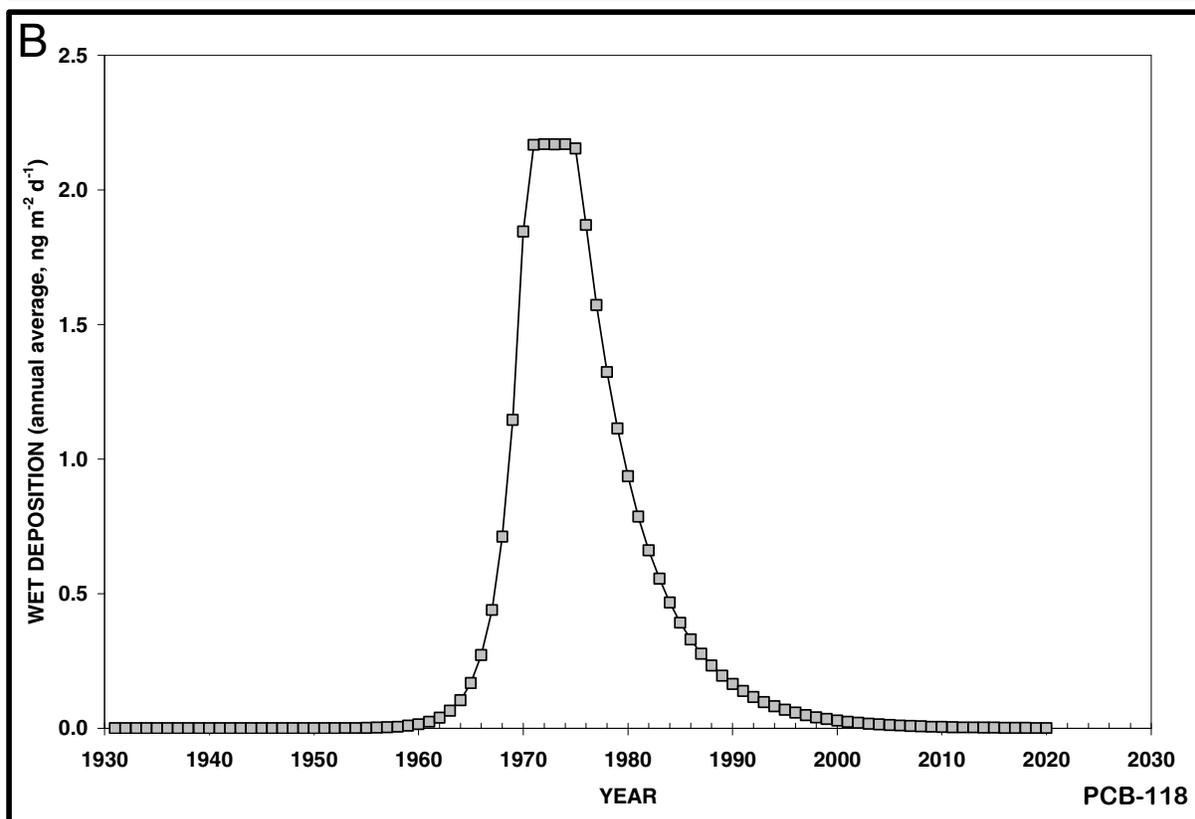
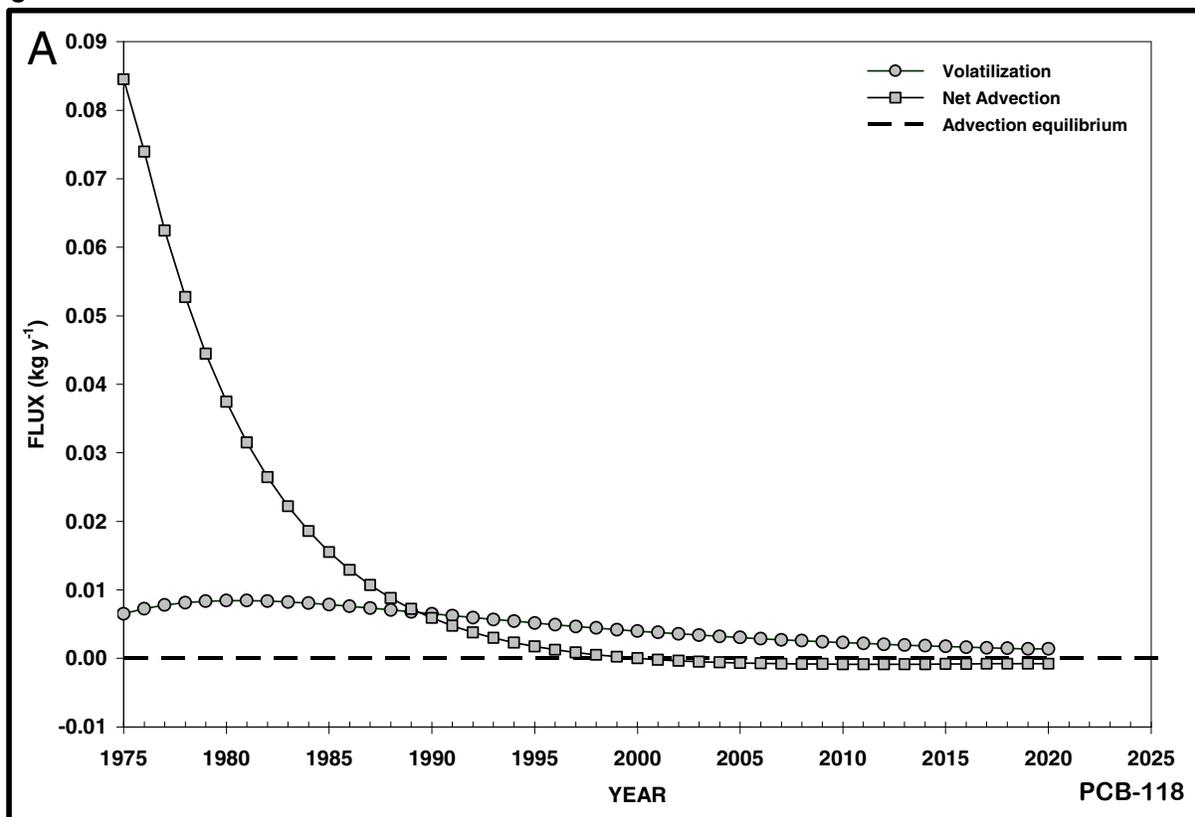
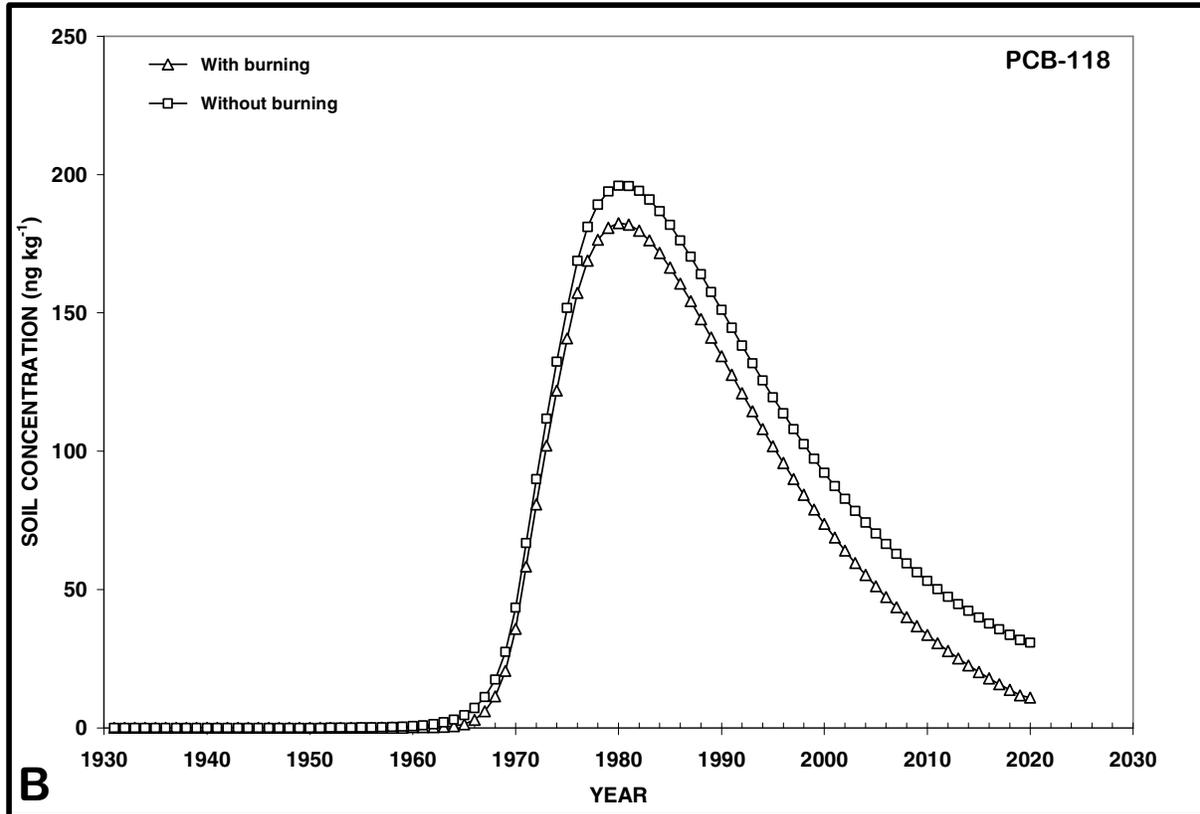
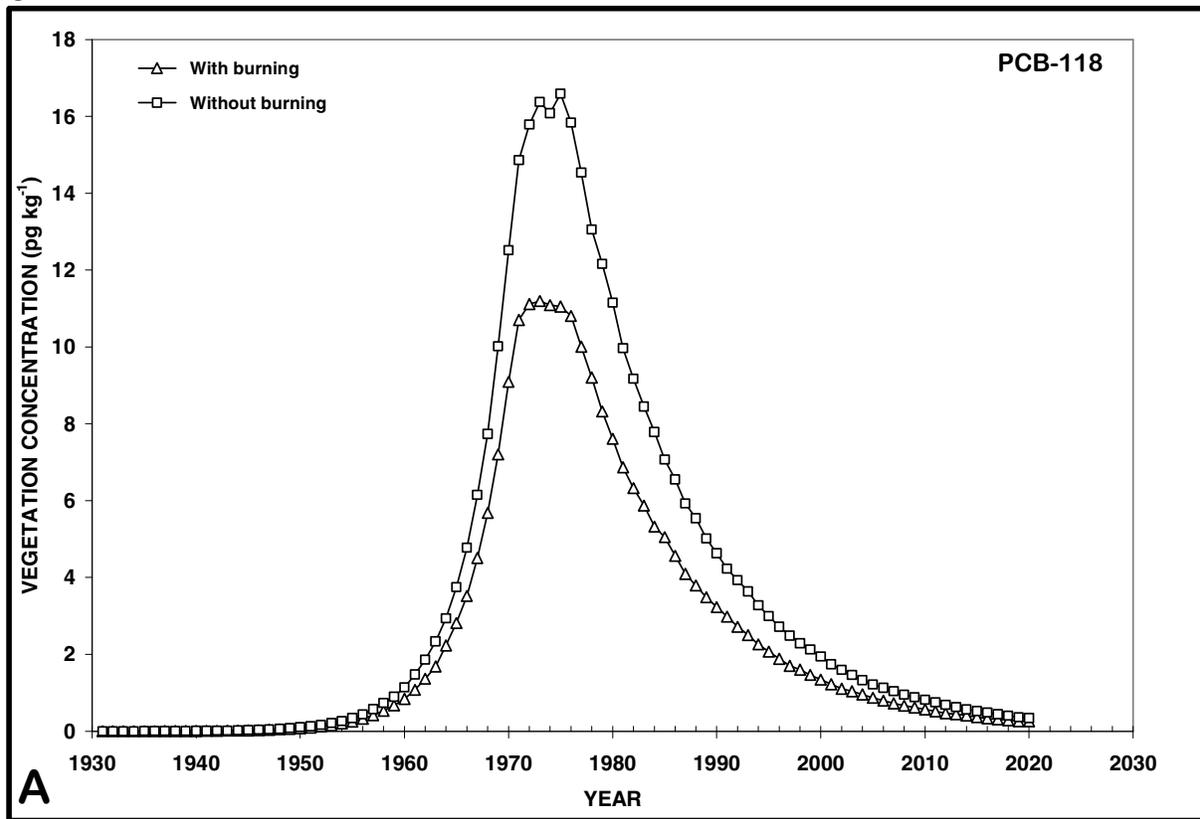


Figure S4



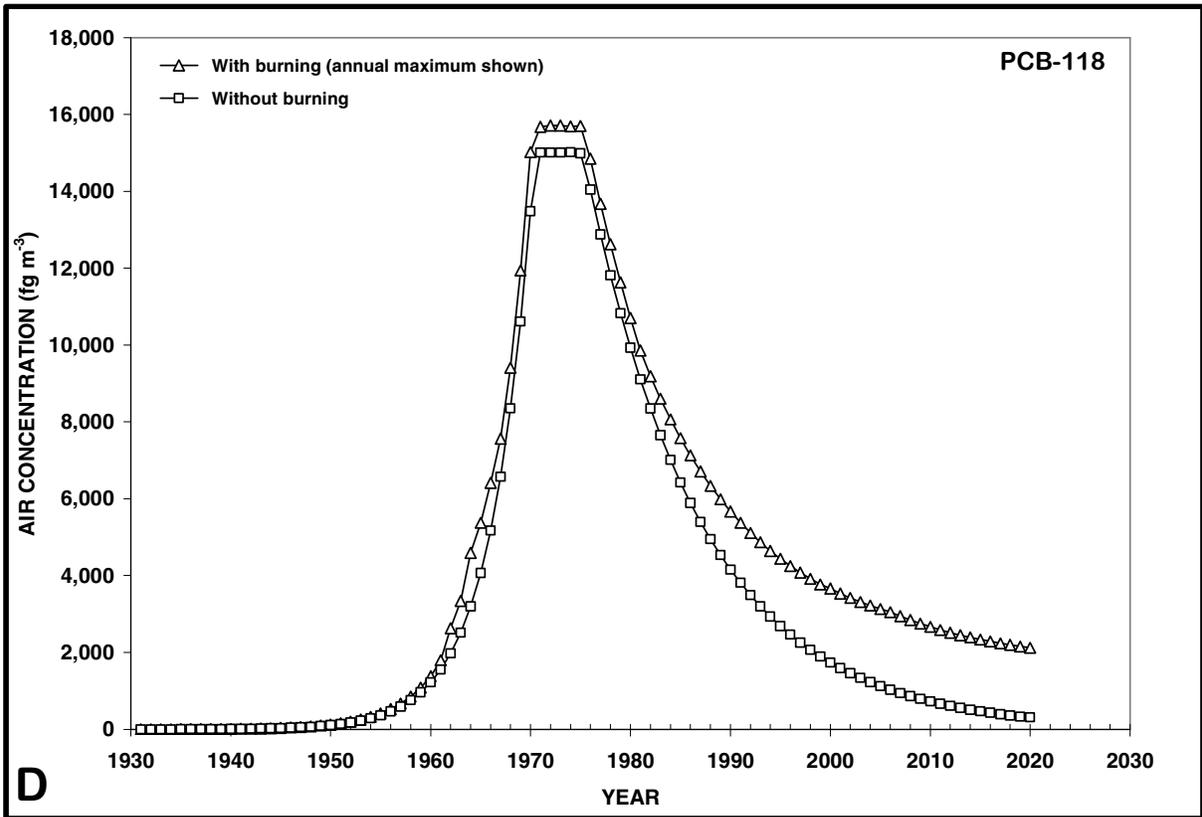
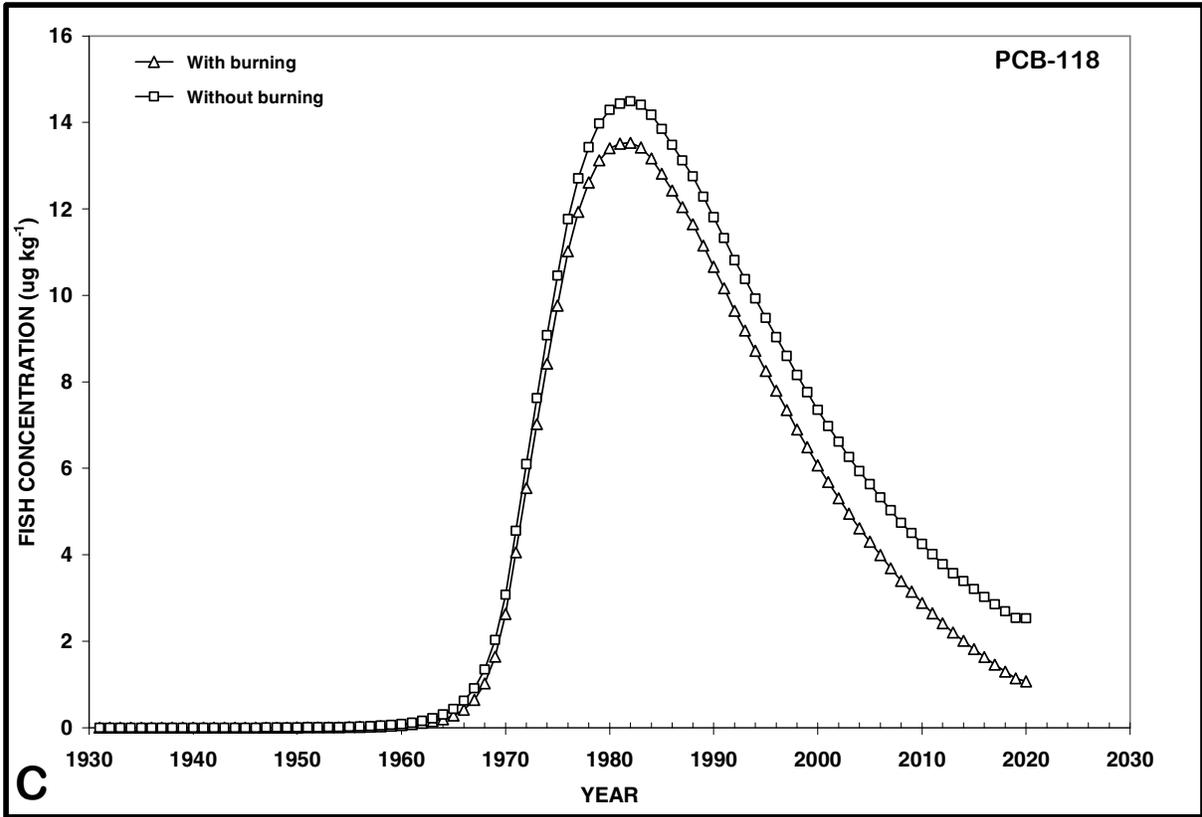
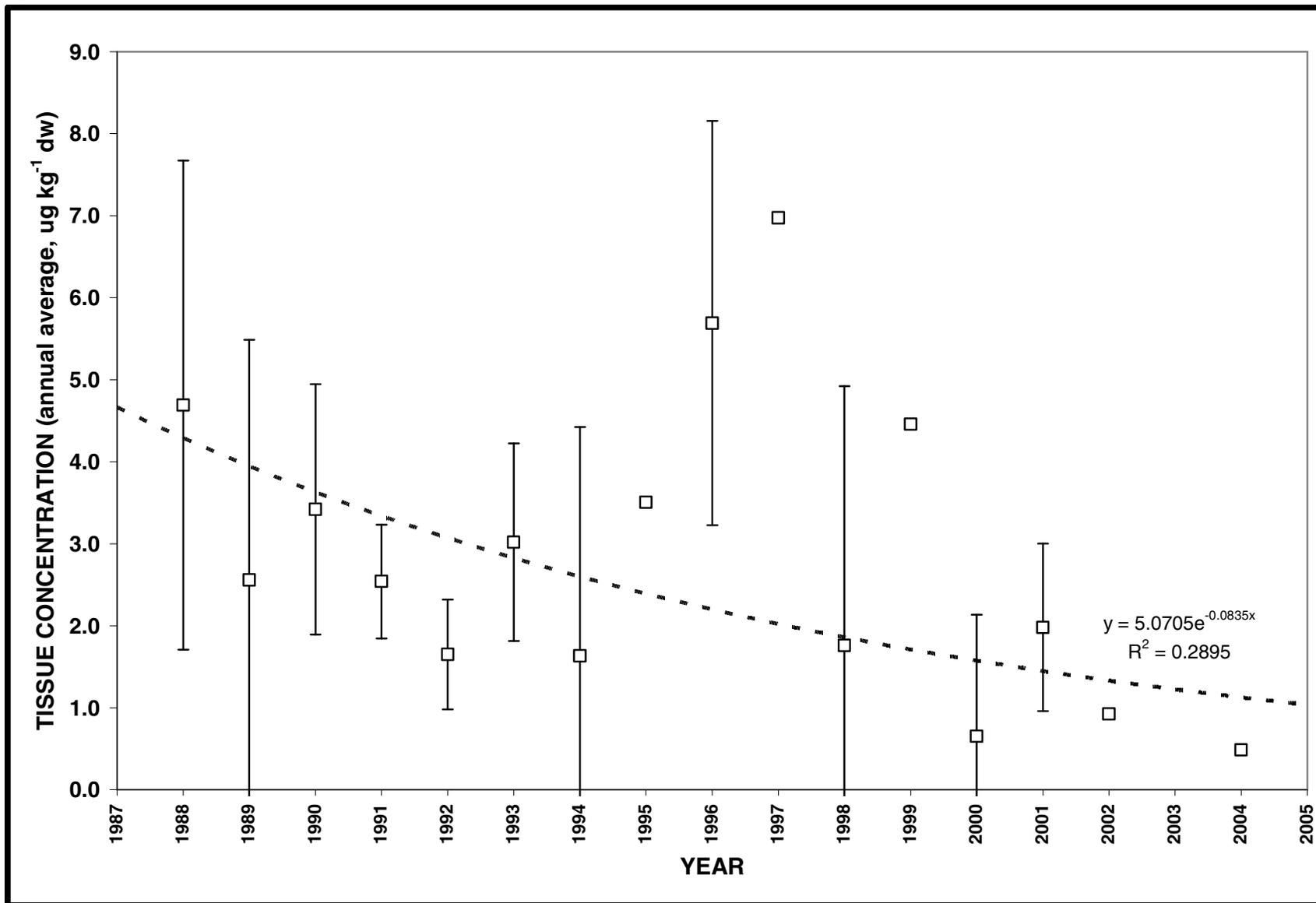


Figure S5



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