# Theoretical Investigation of the Mechanism of the Selective Catalytic Reduction of Nitric Oxide with Ammonia on H-form Zeolites 

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## Figure Captions

Figure 1S: Part 1 of the overview of the reaction network involving $\mathrm{N}_{2} \mathrm{O}_{3}$. Stated energies are only electronic energies without zero-point correction and are with respect to the gas-phase and the empty zeolite framework. Dotted arrows correspond to two structures that are only slightly different and thus, no transition state was considered between them.

Figure 2S: Structures corresponding to the overview given in Figure 1S. Structures that were presented in the main article are not shown again.

Figure 3S: Part 2 of the overview of the reaction network involving $\mathrm{N}_{2} \mathrm{O}_{3}$. Stated energies are only electronic energies without zero-point correction and are with respect to the gas-phase and the empty zeolite framework. Dotted arrows correspond to two structures that are only slightly different and thus, no transition state was considered between them.

Figure 4S: Structures corresponding to the overview given in Figure 3S. Structures that were presented in the main article are not shown again. Starting configurations that are identical with those in part 1 ("a") are also not shown again.

Figure 5S: Overview of the reaction network involving $\mathrm{N}_{2} \mathrm{O}_{4}$. Stated energies are only electronic energies without zero-point correction and are with respect to the gas-phase and the the empty zeolite framework. Dotted arrows correspond to two structures that are only slightly different and thus, no transition state was considered between them.

Figure 6S: Structures corresponding to the overview given in Figure 5S. Structures that were presented in the main article are not shown again.

Figure S1


Figure S2


TS 7a


ST 10a



Figure S3


Figure S4




Figure 55


Figure S6


TS 3

ST 7

ST 9


ST9


TS 5




TS 7


TS 10



ST 20


