

## **Supporting Information**

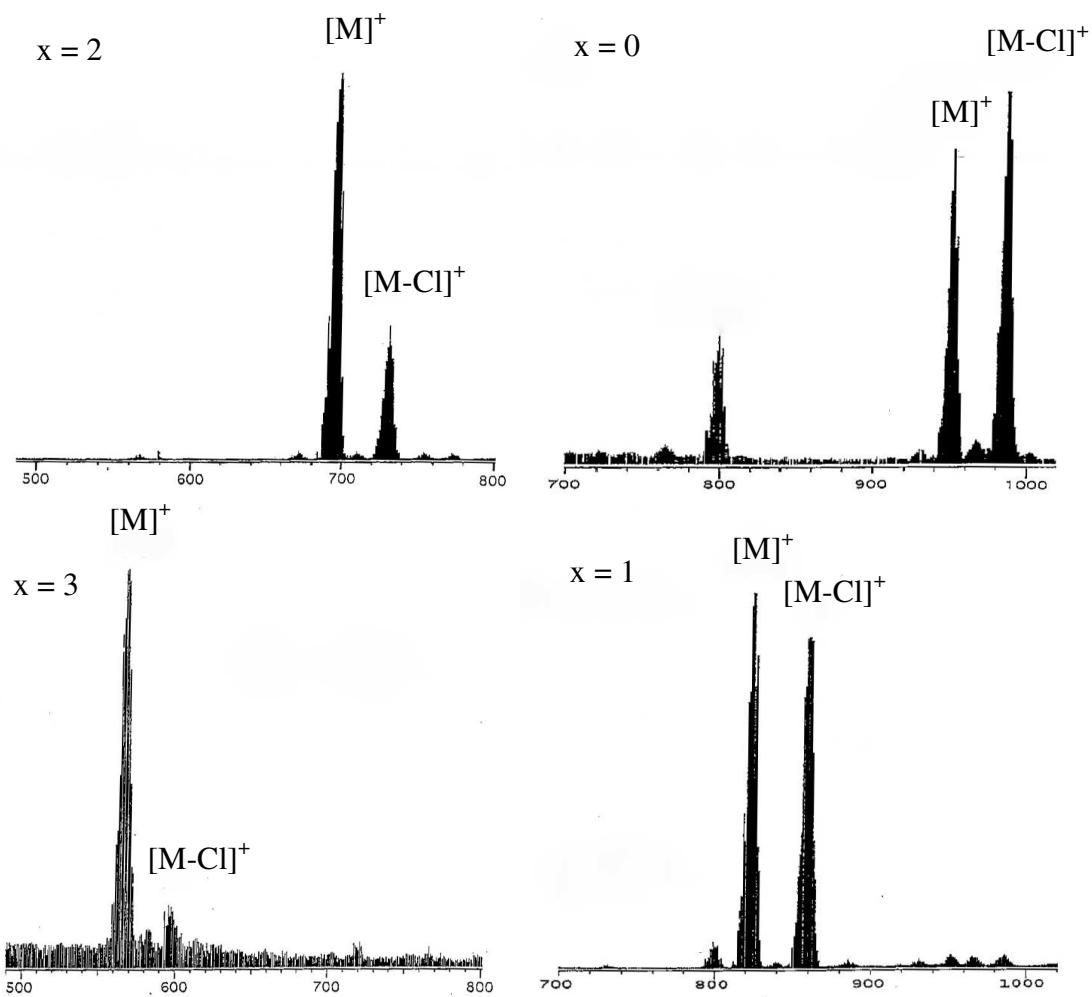
### **Electrochemical and Spectroscopic Characterization of a Series of Mixed-Ligand Diruthenium Compounds**

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**Figure S1.** Mass Spectra of  $\text{Ru}_2(\text{CH}_3\text{CO}_2)_x(\text{Fap})_{4-x}\text{Cl}$  ( $x = 0, 1, 2$  or  $3$ ).  $[\text{M}-\text{Cl}]^+$  = molecular ion peak.

**S2. Half-Wave Potentials for Redox Reactions of  $\text{Ru}_2(\text{CH}_3\text{CO}_2)_x(\text{Fap})_{4-x}\text{Cl}$  in MeCN and PhCN containing 0.1 M TBAP.**

solvent	# bridging ligand		$E_{1/2}$ (V vs SCE)		$E_{1/2}$ (V vs Fc/Fc <sup>+</sup> )			
	(CH <sub>3</sub> CO <sub>2</sub> ) <sub>3</sub> ,(x)	Fap	Ru <sub>2</sub> <sup>6+/5+</sup> Ru <sub>2</sub> <sup>4+/3+</sup>	Ru <sub>2</sub> <sup>5+/4+</sup>	Ru <sub>2</sub> <sup>6+/5+</sup> Ru <sub>2</sub> <sup>4+/3+</sup>	Ru <sub>2</sub> <sup>5+/4+</sup>		
MeCN	3	1	1.08	-0.41	-1.68 <sup>a</sup>	0.66	-0.81	-2.15 <sup>a</sup>
	2	2	0.78	-0.55		0.29	-1.00	
	1	3	0.55	-0.71		0.06	-1.20	
	0	4	0.47	-0.75		-0.03	-1.25	
PhCN	3	1	1.11	-0.43	-1.63 <sup>a</sup>	0.62	-0.92	-2.10 <sup>a</sup>
	2	2	0.75	-0.57	-1.55 <sup>a</sup>	0.27	-1.06	-2.02 <sup>a</sup>
	1	3	0.56	-0.63	-1.48 <sup>a</sup>	0.07	-1.11	-1.92 <sup>a</sup>
	0	4	0.49 <sup>b</sup>	-0.65 <sup>b</sup>		-0.01 <sup>b</sup>	-1.15 <sup>b</sup>	

<sup>a</sup>= peak potential,  $E_{pc}$ , at a scan rate of 0.1 V/s