Supporting Information

Organic Dyes Containing Furan Moiety for High-Performance Dye-Sensitized Solar Cells

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Experimental Methods

General Information. Unless otherwise specified, all the reactions were performed under nitrogen atmosphere using standard Schlenk techniques. All solvents used were purified by standard procedures, or purged with nitrogen before use. ¹H NMR spectra were recorded on a Bruker 400-MHz or 500-MHz spectrometer. Absorption spectra were recorded on a Cary 50 probe UV-Vis spectrophotometer. All chromatographic separations were carried out on silica gel (60M, 230-400 mesh). Mass spectra (FAB) were recorded on a VG70-250S mass spectrometer. Elementary analyses were performed on a Perkin-Elmer 2400 CHN analyzer. The preparation of TiO₂ precursor and the electrode fabrication were carried out based on previous report¹ with an autoclaved temperature of 240 \Box . The film thickness measured by a profilometer (Dektak3, Veeco/Sloan Instruments Inc., USA) was about 15 µm. The TiO₂ electrode with a 0.25 cm² geometric area was immersed in a acetonitrile/tert-butanol mixture (volume ratio 1:1) containing 3×10^{-4} M cis-di(thiocyanato)bis(2,2'-bipyridyl-4,4'-dicarboxylato) ruthenium(II) bis(tetrabutylammonium) (N719, Solaronix S.A., Switzerland) or in the THF solution containing 3×10^{-4} M organic sensitizers for overnight. A platinized FTO was used as a counter electrode and was controlled to have an active area of 0.36 cm^2 by adhered polyester tape with thickness of 60 μ m. After rinsing with CH₃CN or THF, the photoanode was placed on top of the counter

electrode and tightly clipping them together to form a cell. Electrolyte was then injected into the space and then sealing the cell with the Torr Seal cement (Varian, MA, USA). The electrolyte was composed of 0.5 M lithium iodide (LiI), 0.05 M iodine (I₂), and 0.5 M 4-tert-butylpyridine (TBP) dissolved in acetonitrile. A 0.6×0.6 cm² cardboard mask was clipped onto the device to constrain the illumination area. The photoelectrochemical characterizations on the solar cells were carried out by using a modified light source, 300 W Xe lamp (Oriel 6258) equipped with a water-based IR filter and AM 1.5 filter (Oriel 81080 kit). Photocurrent-voltage characteristics of the DSSCs were recorded with a potentiostat/galvanostat (CHI650B, CH Instruments, Inc., USA) at a light intensity of 100 mW/cm² measured by a thermopile probe (Oriel 71964). The light intensity was further calibrated by an Oriel reference solar cell (Oriel 91150) and adjusted to be 1.0 sun. The monochromatic quantum efficiency was recorded through a monochromator (Oriel 74100) at short circuit condition. The intensity of each wavelength was in the range of 1 to 3 mW/cm^2 . The photovoltage transients of assembled devices were recorded with a digital oscilloscope (LeCroy, WaveSurfer 24Xs). Pulsed laser excitation was applied by a Q-switched Nd:YAG laser (Continuum, model Minilite II) with 1 Hz repetition rate at 532 nm and a 5 ns pulse width at half-height. The beam size was slightly larger than 0.5×0.5 cm² to cover the area of the device with an incident energy of 1 mJ/cm². The

recombination lifetime of photoinjected electrons with oxidized dyes was measured by transient photovoltages at open circuit with the presence of LiI electrolyte (0.5 M). The average electron lifetime can be estimated approximately by fitting a decay of the open circuit voltage transient with $\exp(-t/\tau_R)$, where t is time and τ_R is an average time constant before recombination.

Synthesis of materials

(4-(Diphenylamino)benzyl)triphenylphosphonium bromide $(1a)^2$, 4-(diphenylamino)phenylboronic acid $(2a)^3$, 9,9-diethyl-*N*,*N*-diphenyl-7-(5-(tributylstannyl)thiophen-2-yl)-9*H*-fluoren-2-amine $(3a)^4$, ((7-(diphenylamino)-9,9-diethyl-9*H*-fluoren-2-yl)methyl)triphenylphosphonium bromide $(4a)^2$ were prepared according to the literature procedure.

4-(2-(Furan-2-yl)vinyl)-*N*,*N*-diphenylaniline (1b)

(4-(Diphenylamino)benzyl)triphenylphosphonium bromide (**1a**) (1.67 mmol, 1 g) THF (22 mL) and *t*-BuOK (1.69 mmol, 0.19 g) was stirred at ambient temperature under nitrogen atmosphere for 1 h. 2-furaldehyde (1.77 mmol, 0.17 g) was dissolved in THF and added dropwise to the solution, and the reaction mixture was stirred for 1 h at ambient temperature, whereupon the mixture was heated to reflux for 24 h. The reaction mixture was allowed to cool to ambient temperature and a molar excess of water was added. The organic layer was extracted with CH₂Cl₂, washed with brine. The organic extract was dried over anhydrous MgSO₄ and evaporated under vacuum.

The crude product was used in the next step without further purification.

5-(4-(diphenylamino)styryl)furan-2-carbaldehyde (1c)

A mixture of 1b (0.79 g, 2.34 mmol) and THF was cooled to -78 °C using acetone-liquid N2 bath. n-Butyl lithium (1.75 mL, 1.6 M solution in hexanes, 2.81 mmol) was added dropwise over 10 min with vigorous stirring. It was brought to 0 °C during 1 h and kept at this temperature for additional 1 h. Again the mixture was cooled to -78 °C and DMF was added at once. The solution was brought to room temperature and stirred overnight. The reaction was quenched by 1 N HCl and extracted with diethyl ether. The combined organic extract was dried over anhydrous MgSO₄ and filtered. The filtrate was evaporated to yield the crude product as an orange solid. It was purified by column chromatography on silica gel using hexane/CH₂Cl₂ mixture (1:1) as eluent to give 1c as a yellow-orange solid. ¹H NMR (400 MHz, acetone- d_6): δ 9.59 (s, 1H), 7.56 (d, J = 8.6 Hz, 2H), 7.45 (d, J = 3.7 Hz, 1H), 7.34 (m, 5H), 7.10 (m, 6H), 7.06 (d, J = 16.3 Hz, 1H), 7.00 (m, 2H), 6.72 (d, J = 3.7 Hz, 1H) MS (FAB) *m/z*: 365.1 (M⁺).

(E)-2-Cyano-3-(5-(4-(diphenylamino)styryl)furan-2-yl)acrylic acid (1)

To a flask containing a mixture of 1c (0.44 g, 1.20 mmol), cyanoacetic acid (0.12 g,

1.41mmol), ammonium acetate (0.030 g, 0.39 mmol) was added acetic acid (10 mL). The mixture was heated at 100 °C for 8 h. and allowed to cool to room temperature. The resulting solid was filtered and washed with distilled water, diethyl ether and methanol to give a dark brown solid. The pure product was obtained by silica gel chromatography to afford **1**. ¹H NMR (500 MHz, acetone- d_6): δ 7.99 (s, 1H), 7.51 (d, J = 8.6 Hz, 2H), 7.47 (d, J = 9.0 Hz, 1H), 7.45 (d, J = 3.4 Hz, 1H), 7.36 – 7.32 (m, 4H), 7.13 – 7.08 (m, 6H), 7.00 (d, J = 8.6 Hz, 2H), 6.78 (d, J = 3.7 Hz, 1H). ¹³C NMR (100 MHz, THF- d_8): δ 164.4, 160.4, 149.6, 149.1, 148.4, 137.9, 134.2, 131.1, 130.2, 129.1, 125.9, 125.2, 124.4, 123.5, 116.7, 113.8, 112.7, 112.6, 98.5. MS (FAB) m/z : 432.1 (M⁺). Anal. Calcd for C₂₈H₂₀N₂O₃: C, 77.76; H, 4.66; N, 6.48. Found: C, 77.46; H, 4.76; N, 5.98.

5-(4-(Diphenylamino)phenyl)furan-2-carbaldehyde (2b)

A stirred mixture of 4-(diphenylamino)phenylboronic acid (1.11 g, 3.83 mmol), 5-bromofuran-2-carbaldehyde (0.60 g, 3.45 mmol), $Pd(PPh_3)_4$ (0.044 g 0.038 mmol) in toluene/ethanol (50/5 mL) and 2M aqueous Na₂CO₃ solution (3 equiv.) was added via syringe. The reaction mixture was heated to 85 °C for 24h. The product was extracted with CH₂Cl₂, wash with water, and dried over MaSO₄. After evaporation of the solvent, the residue was purified by chromatography on silica gel by eluting with Hexane/dichloromethane to afford **2b**. H NMR (400 MHz, acetone-*d*₆): δ 9.61 (s, 1H), 7.77 (dd, *J* = 7.0, 2.0 Hz, 2H), 7.50 (d, *J* = 4.0 Hz, 1H), 7.38 – 7.34 (m, 4H), 7.15 – 7.12 (m, 6H), 7.06 (dd, *J* = 7.0, 2.0 Hz, 2H), 7.02 (d, *J* = 3.6 Hz, 1H). MS (FAB) *m/z*: 339.1 (M⁺).

(E)-3-(5-(4-(Diphenylamino)phenyl)furan-2-yl)-2-cyanoacrylic acid (2)

It was synthesized by the same procedure as described above for **1**. ¹H NMR (400 MHz, acetone- d_6): δ 8.02 (s, 1H), 7.86 (m, 2H), 7.45 (d, J = 3.6 Hz, 1H), 7.37 (m, 4H), 7.15 (m, 6H), 7.13 (m, 1H), 7.06 (d, J = 8.8 Hz, 2H). ¹³C NMR (100 MHz, THF- d_8) δ 164.4, 160.8, 150.3, 148.8, 148.1, 138.1, 138.0, 130.4, 127.3, 126.2, 125.8, 124.8, 123.2, 122.9, 116.7, 108.7, 108.6, 97.9. MS (FAB) *m/z*: 406.1 (M⁺). Anal. Calcd for C₂₆H₁₈N₂O₃: C, 76.83; H, 4.46; N, 6.89. Found: C, 76.46; H, 4.50; N, 6.55.

5-(5-(7-(Diphenylamino)-9,9-diethyl-9*H*-fluoren-2-yl)thiophen-2-yl)furan-2-carba ldehyde (3b)

A mixture of 5-bromofuran-2-carbaldehyde (0.92 g, 5.26 mmol), 9,9-diethyl-*N*,*N*-diphenyl-7-(5-(tributylstannyl)thiophen-2-yl)-9*H*-fluoren-2-amine (**3a**) (4.00 g, 5.26 mmol), Pd(PPh₃)₄ (0.06 g, 0.053 mmol) and dry DMF were placed into two-necked flask under nitrogen atmosphere and stirred overnight at 100 °C. After cooling, it was quenched with aqueous KF and extracted with diethyl ether. The organic extracts were washed with brine solution and dried over anhydrous MgSO₄. The crude product was purified by column chromatography using a dichloromethane / hexane mixture as eluent to give **3b** as a yellow-orange solid. ¹H NMR (400 MHz, THF- d_8): δ 7.98 (s, 1H), 7.72-7.63 (m, 5H), 7.53 (d, J = 3.9 Hz, 1H), 7.46 (d, J = 3.7 Hz, 1H), 7.25-7.21 (m, 4H), 7.14 (d, J = 1.9 Hz, 1H), 7.09 (d, J = 7.6 Hz, 4H), 7.03-6.98 (m, 3H), 6.94 (d, J = 3.8 Hz, 1H), 2.04-1.95 (m, 4H), 0.38 (t, J = 7.3 Hz, 6H). MS (FAB) m/z: 565.2 (M⁺).

(*E*)-2-Cyano-3-(5-(5-(7-(diphenylamino)-9,9-diethyl-9*H*-fluoren-2-yl)-thiophen-2yl)furan-2-yl)acrylic acid (3)

It was synthesized by the same procedure as described above for **1**. ¹H NMR (400 MHz, THF-*d*₈): δ 7.98 (s, 1H), 7.72-7.63 (m, 5H), 7.53 (d, *J* = 3.6 Hz, 1H), 7.45 (d, *J* = 4.0 Hz, 1H), 7.26-7.22 (m, 4H), 7.14 (d, *J* = 2.0 Hz, 1H), 7.10-7.09 (m, 4H), 7.04-6.97 (m, 3H), 6.94 (d, *J* = 3.6 Hz, 1H), 2.08-1.90 (m, 4H), 0.38 (t, 6H). ¹³C NMR (100 MHz, THF-*d*₈): δ 164.2, 161.9, 161.3, 161.0, 155.5, 152.4, 151.7, 149.0, 148.9, 148.5, 142.9, 138.7, 137.1, 132.8, 131.0, 129.3, 127.8, 126.7, 126.1, 125.9, 125.7, 125.2, 124.3, 124.1, 123.6, 122.7, 122.4, 122.3, 121.4, 121.3, 120.7, 119.7, 119.2, 116.4, 110.6, 108.8, 98.7, 57.1, 33.3, 32.0, 9.6, 8.6, 8.3. MS (FAB) *m/z*: 632.2 (M⁺).

Anal. Calcd for C₄₁H₃₂N₂O₃S: C, 77.82; H, 5.10; N, 4.43; S, 5.07. Found: C, 77.42; H, 5.54; N, 4.07; S, 5.37.

7-(2-(5-Bromofuran-2-yl)vinyl)-9,9-diethyl-*N*,*N*-diphenyl-9*H*-fluoren-2-amine, mixture of *E*/*Z* (4b).

It was prepared from ((7-(diphenylamino)-9,9-diethyl-9*H*-fluoren-2-yl)methyl)triphenylphosphonium bromide (**4a**) and 5-bromofuran-2-carbaldehyde as described above for **1b**. The crude product was used in the next step without further purification.

5-(2-(7-(Diphenylamino)-9,9-diethyl-9*H*-fluoren-2-yl)vinyl)furan-2-carbaldehyde (4c).

It was synthesized by the same procedure as described above for **1c**. The cis-to-trans isomerization was achieved by reflux with iodine (50 mg) in toluene (10 mL). ¹H NMR (400 MHz, acetone- d_6): δ 9.61 (s, 1H), 7.76-7.72 (m, 3H), 7.63 (dd, J = 8.0, 1.7 Hz, 1H), 7.48 (d, J = 8.8 Hz, 1H), 7.46 (d, J = 3.9 Hz, 1H), 7.32-7.23 (m, 5H), 7.15 (d, J = 1.9 Hz, 1H), 7.11-7.09 (m, 4H), 7.07-7.01 (m, 3H), 6.77 (d, J = 3.7 Hz, 1H), 2.11-1.93 (m, 4H) , 0.36 (m, J = 7.3 Hz, 6H). MS (FAB) m/z: 509.2 (M⁺).

2-Cyano-3-(5-((*E*)-2-(7-(diphenylamino)-9,9-diethyl-9*H*-fluoren-2-yl)vinyl)furan-

2-yl)acrylic acid (4)

It was synthesized by the same procedure as described above for **1**. ¹H NMR (400 MHz, acetone-*d*₆): δ 7.98 (s, 1H), 7.77-7.70 (m, 3H), 7.59-7.55 (m, 2H), 7.42 (d, *J* = 3.7 Hz, 1H), 7.32-7.25 (m, 5H), 7.14-7.01 (m, 8H), 6.80 (d, *J* = 3.6 Hz, 1H). ¹³C NMR (100 MHz, THF-*d*₈): δ 164.1, 161.7, 161.4, 161.0, 155.6, 152.5, 151.7, 149.1, 149.0, 148.5, 143.0, 138.7, 137.1, 132.8, 131.1, 129.6, 129.4, 127.8, 126.7, 126.1, 126.0, 125.7, 125.2, 124.4, 124.1, 123.7, 122.8, 122.5, 122.3, 121.5, 121.3, 120.7, 119.8, 119.2, 116.4, 110.6, 108.8, 98.7, 57.1, 33.3, 32.1, 9,6, 8.6, 8.3. MS (FAB) *m/z*: 576.2 (M⁺).

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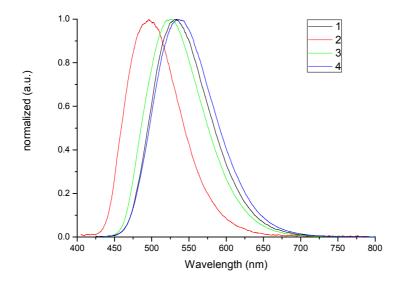
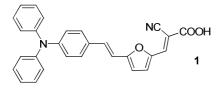


Figure S1. Emission spectra of dyes recorded in THF



2762219 0698219 8296619 9681012 824002
8885.7 22511.7 25511.40 82811.7 32811.7 82811.7 82812.7
-7.46038 -7.44607 -7.44607 -7.44607 -7.44607 -7.44607 -7.446038 -7.446038 -7.446038 -7.446038 -7.446038 -7.446038
790907 7-7.52549 7-7.505649 7-7.47830

wdd

